## Title

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## Publication Date

2020
DOI
10.1016/bs.mie.2020.01.007

Peer reviewed

# Implementation of the riding hydrogen model in CCTBX to support the next generation of X-ray and neutron joint refinement in Phenix 

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#### Abstract

A fundamental prerequisite for implementing new procedures of atomic model refinement against neutron diffraction data is the efficient handling of hydrogen atoms. The riding hydrogen model, which constrains hydrogen atom parameters to those of the non-hydrogen atoms, is a plausible parameterization for refinements. This work describes the implementation of the riding hydrogen model in the Computational Crystallography Toolbox and in Phenix. Riding hydrogen atoms can be found in several different configurations that are characterized by specific geometries. For each configuration, the hydrogen atom parameterization and the expressions for the gradients of refinement target function with respect to non-hydrogen parameters are described.


## 1. Introduction

X-ray crystallography is the leading method for obtaining three-dimensional structures of macromolecules. Oftentimes, hydrogen $(\mathrm{H})$ atoms play an important biological role and may be the subject of a structural study, in particular when trying to understand enzyme mechanism. Accurate determination of the location of key H atoms therefore requires experimental information, but as hydrogen has a weak X-ray scattering cross section, its electron density is not observed unless very high quality, and very high-resolution data are available. Even then, it is not guaranteed that all H atoms will be observed: Petrova and Podjarny (2004) showed that only about half of the hydrogen atom positions can be determined experimentally with high resolution electron density maps of macromolecules. By contrast, neutron crystallography allows the observation of hydrogen atoms (or deuterium substitutes) at resolutions better than $3 \AA$ (Ostermann, Tanaka, Engler, Niimura, \& Parak, 2002; Shu, Ramakrishnan, \& Schoenborn, 2000).

[^0]The determination of macromolecular structures with neutron diffraction has two major complications. First, since H atoms have a similar scattering contribution as other atoms, they need to be refined individually in the atomic model. Also, as a result of partial deuteration, some H atom positions are shared with deuterium atoms (D) meaning that their relative occupancies need to be refined. Given that H atoms constitute about $50 \%$ of the atoms in a typical macromolecular structure, refining them individually adds a substantial number of refined parameters. Second, the experimental data for neutron diffraction data are often poorer than those of corresponding X-ray data: the low flux of available neutron beams and incoherent scattering of hydrogen (if present) lead to low signal-to-noise ratios, and the limited data-collection time on oversubscribed instruments result in low completeness, averaging about $80 \%$ (Liebschner, Afonine, Moriarty, Langan, \& Adams, 2018). In contrast, the completeness of a typical X-ray dataset is expected to be greater than $95 \%$ (Dauter \& Dauter, 2017).

With rare exceptions, X-ray data and the corresponding atomic models are almost always available prior to structure determination with neutrons. The availability of two sources of information, namely X-ray and neutron data, may help alleviate some of the aforementioned challenges of refinement against neutron data: the joint X-ray and neutron (joint XN) refinement method enables building more complete atomic models that include experimentally observed hydrogen and non-hydrogen atoms (Adams, Mustyakimov, Afonine, \& Langan, 2009; Afonine et al., 2010; Coppens, 1967; Wlodawer, 1980; Wlodawer \& Hendrickson, 1982; Wlodawer, Miller, \& Sjölin, 1983). The currently used joint XN refinement method optimizes a single atomic model simultaneously against two datasets using a combined refinement target function:

$$
\begin{equation*}
T=w_{x} \bullet T_{\mathrm{X}-\mathrm{ray}}+w_{n} \bullet T_{\text {neutron }}+T_{\text {restraints }} \tag{1}
\end{equation*}
$$

where $T_{\mathrm{X} \text {-ray }}$ and $T_{\text {neutron }}$ are target functions relating the model and respective data, $T_{\text {restraints }}$ is a restraint term that adds a priori information, and $w_{\mathrm{x}}$ and $w_{\mathrm{n}}$ are empirical scale factors.

Using the refinement target (1) with a single atomic model relies on the assumption that the X-ray and neutron crystals and respective structures are isomorphous. This hypothesis is only approximate, because:
(a) If different crystals are used for the neutron and X-ray experiments, the unit cell parameters of the crystals may be different.
(b) Ordered and semi-ordered parts of the structures may be different.
(c) X-rays are scattered by electrons while neutrons are scattered by atomic nuclei. The datasets therefore convey different information about covalent X-H bond lengths: the electron distribution of the H atom, which has only one valence electron and no core electron, is shifted along the $\mathrm{X}-\mathrm{H}$ bond away from the H atom nucleus toward atom X . Accordingly, the locations of H atom electron density peaks and nuclear scattering length density peaks are different and the X-H bond lengths appear about $10-20 \%$ shorter for Xrays (Allen, 1986; Allen \& Bruno, 2010).
(d) The X-ray and neutron datasets may be collected under different conditions. For example, X-ray data are typically collected at cryogenic temperatures to mitigate radiation damage while neutron diffraction experiments are carried out at room temperature because neutrons do not damage the sample. Atomic models of structures derived from roomtemperature experiments are likely to exhibit more variability due to thermal motion (Fraser et al., 2011).
(e) Data collection times are very different: X-ray data can be collected in minutes at synchrotron beamlines (Turkenburg \& McAuley, 2013) while neutron data collections typically last days or weeks (Chen \& Unkefer, 2017; Coates et al., 2015). Atomic models derived from diffraction data are averaged over space (unit cells of the crystal) and time (duration of the data collection). It may be that data collected over substantially different periods of time result in different models.

Analysis of atomic models derived from joint XN refinement (see Chapter "What are the current limits on determination of protonation state using neutron macromolecular crystallography?" by Liebschner et al.) shows that in practice there often are differences between the neutron and X-ray datasets (for the reasons just described). Therefore, it is desirable for a joint XN refinement algorithm to yield two models that are related to each other and contain H atoms; each model should fit their respective dataset and reflect its properties while at the same time it should benefit from the complementarity of the X-ray and neutron data. Implementation details of such an approach will be described elsewhere. In this chapter we describe a fundamental prerequisite for implementing such a procedure in the software package Phenix (Liebschner et al., 2019), namely efficient handling of H atoms.

The contribution of H atoms can be modeled in several ways. One possibility is to include the H atoms explicitly in the input model and to refine them independently, similar to the other non-H atoms. Alternatively, while also being modeled explicitly, their parameters can be constrained by those of the non-H atoms. This approach does not increase the number of parameters even when the model contains H atoms. This is because only the positions of the non- H atoms are refined explicitly while the positions of H atoms are recalculated based on geometric constraints. This riding hydrogen model (Busing \& Levy, 1964; Sheldrick \& Schneider, 1997) is almost universally used in X-ray refinements. It may also be a plausible parameterization for refinements against neutron data (Gruene, Hahn, Luebben, Meilleur, \& Sheldrick, 2014), especially for low-resolution or poorly complete datasets. It is also conceivable to mix different hydrogen modeling approaches, such as refining H atoms with geometric degrees of freedom individually while treating the remaining H atoms as riding. Below we describe algorithmic details of the implementation of the riding hydrogen model in the Computational Crystallography Toolbox (CCTBX; Grosse-Kunstleve, Sauter, Moriarty, \& Adams, 2002) and in Phenix.

## 2. Parameterizing the riding hydrogen atom model for typical geometrical configurations

### 2.1 Riding H definition

The concept of riding hydrogen model ('riding H' in what follows) relies on the fact that the coordinates of most (but not all) H atoms in proteins can be unambiguously expressed through the coordinates of their covalently connected non-H atoms. To define the riding H atom position, at least three neighboring non- H atoms are necessary. These are typically the 'parent' atom $\left(\mathrm{A}_{0}\right)$, to which the H atom is covalently bound, and two other atoms, that we note by $\mathrm{A}_{1}$ and $\mathrm{A}_{2}$, bound to the parent atom (neighbors). An example configuration of a hydrogen atom is shown in Fig. 1 (left). In this example, the H atom is in a plane with $\mathrm{A}_{0}$, $A_{1}$ and $A_{2}$. This geometry can be found in the peptide unit or in aromatic rings.

Riding H atoms can be found in several different configurations; for example, three non-H atoms may define simultaneously several H atoms; an example is the $\mathrm{H} \gamma 11-\mathrm{H} \gamma 12-\mathrm{H} \gamma 13$ group in valine residue; all of its H atoms are bound to the $\mathrm{C} \gamma$ atom. Other typical examples are shown in Fig. 2. In each configuration, the number of neighbors and their ideal bond and angle values may vary. The algorithmic challenge is to define the coordinates of the H atom from the coordinates of a group of its non-H neighbors for each configuration. It is desirable that the algorithm is independent of eventual imperfections in the geometry of non-H atoms as much as possible.

For each configuration the H atom position is parameterized using coordinates of (at least) three non-H neighbors and by using $\vec{r}_{H}=\vec{r}_{0}+\vec{r}_{H 0}=\vec{r}_{0}+d_{H} \bullet \vec{u}_{H 0}$, where $\vec{u}_{H 0}$ is the unit vector along the $\mathrm{A}_{0}-\mathrm{H}$ direction $\vec{r}_{H 0}$ and $d_{H}$ is the ideal $\mathrm{H}-\mathrm{A}_{0}$ bond length (Fig. 1). Since $d_{H}$ is known for each kind of models (and is different for the X-ray and neutron data), the main task is to define $\vec{u}_{H 0}$ from the set of non-H neighbors and ideal bond lengths and angles. A chain of transformations is developed for all possible configurations. Each step passes from a set of input parameters to a set of output parameters. This way, the first step in the chain of transformations uses the coordinates of atoms parameterizing the H atom and respective ideal bond lengths and angles. The last step of the chain of transformations provides $\vec{u}_{H 0}$. This way, it is easier to apply the chain rule for gradient calculations (see Section 3 and Appendix) and the procedure can be carried out in the most efficient way (Lunin \& Urzhumtsev, 1985).

We note that configurations described below are general and not specific to protein residues only.

### 2.2 Coplanar configuration

The first configuration describes the H atom linked to the N atom in the peptide unit (Fig. 2 A ) or the H atoms in aromatic rings. The theoretical configuration is represented in Fig. 1. In the ideal conformation, the unit vectors in the directions $\mathrm{A}_{0}-\mathrm{A}_{1}\left(\vec{u}_{10}\right)$ and $\mathrm{A}_{0}-\mathrm{A}_{2}\left(\vec{u}_{20}\right)$ form the angle $a_{0}$. Being coplanar, the unit vector $\vec{u}_{H 0}$ can be expressed through these two
vectors as $\vec{u}_{H 0}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}$ forms the ideal angles $a_{1}$ and $a_{2}$ with $\vec{u}_{10}$ and $\vec{u}_{20}$, respectively. We note $c_{0}=\cos a_{0}, c_{1}=\cos a_{1}$ and $c_{2}=\cos a_{2}$.

The coefficients $a$ and $b$ can be found from the conditions $a+b \cdot c_{0}=c_{1}, a \cdot c_{0}+b=c_{2}$ that give an analytic solution $a=\left(1-c_{0}^{2}\right)^{-1}\left(c_{1}-c_{0} \bullet c_{2}\right), b=\left(1-c_{0}^{2}\right)^{-1}\left(c_{2}-c_{0} \bullet c_{1}\right)$. Thus, the values of parameters $a$ and $b$ for the ideal coplanar conformation can be calculated in advance.

The chain of transformations to obtain the H -atom position is the following:
(a) $\vec{r}_{10}=\vec{r}_{1}-\vec{r}_{0}, \vec{r}_{20}=\vec{r}_{2}-\vec{r}_{0}$
(b) $\vec{u}_{10}$ is normalized $\vec{r}_{10}, \vec{u}_{20}$ is normalized $\vec{r}_{20}$
(c) $\vec{r}_{H 0}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}$
(d) $\vec{u}_{H 0}=\left\|\vec{r}_{H 0}\right\|^{-1} \vec{r}_{H 0}$ with $\left\|\vec{r}_{H 0}\right\|=\left(\vec{r}_{H 0} \bullet \vec{r}_{H 0}\right)^{1 / 2}$

This procedure positions the H atom in the plane and at a correct distance to its parent atom. We note that the input model does not necessarily possess an ideal geometry, which means that the angle between $\mathrm{A}_{0}-\mathrm{A}_{1}$ and $\mathrm{A}_{0}-\mathrm{A}_{2}$ is not necessarily equal to the ideal value $a_{0}$, and similarly for $a_{1}$ and $a_{2}$. However, the riding procedure uses the ideal angle values to calculate the coefficients $a$ and $b$. This ensures that the H atom geometry is regularized to match the ideal values.

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~A}_{2}, d_{H}, a, b$.

### 2.3 2H-tetrahedral configuration

Examples of this configuration are the Ha 1 and Ha 2 atoms bound to the glycine Ca atom (Fig. 2B). The configuration is represented schematically in Fig. 3. The atoms $\mathrm{H}_{1}$ and $\mathrm{H}_{2}$ are at the given distance $d_{H}$ from atom $\mathrm{A}_{0}$. In the ideal configuration, they are symmetric with respect to the plane $\mathrm{A}_{1}-\mathrm{A}_{0}-\mathrm{A}_{2}$ and form an angle $\mathrm{H}_{1}-\mathrm{A}_{0}-\mathrm{H}_{2}$ equal to $2 \delta$. The unit vector $\vec{u}_{H 1}$ in the direction $\mathrm{A}_{0}-\mathrm{H}_{1}$ forms angles $a_{1}$ and $a_{2}$ with $\mathrm{A}_{0}-\mathrm{A}_{1}\left(\vec{u}_{10}\right)$ and $\mathrm{A}_{0}-\mathrm{A}_{2}\left(\vec{u}_{20}\right)$, respectively. If the coefficients $a$ and $b$ are defined as in Section 2.2, the vector $\vec{d}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}$ is collinear with the orthogonal projection of $\mathrm{A}_{0}-\mathrm{H}_{1}$ and $\mathrm{A}_{0}-\mathrm{H}_{2}$ onto the plane $\mathrm{A}_{1}-\mathrm{A}_{0}-\mathrm{A}_{2}$.

Steps (a) and (b) are defined as in Section 2.2. The chain of additional transformations is as follows:
(c) $\vec{v}=\vec{u}_{10} \times \vec{u}_{20}$
(d) $\vec{v}_{0}$ is normalized $\vec{v}$
(e) $\vec{d}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}$
(f) $\vec{d}_{0}$ is normalized $\vec{d}$
(g) $\vec{u}_{H 1}=\cos \delta \bullet \vec{d}_{0}+\sin \delta \bullet \vec{v}_{0}, \vec{u}_{H 2}=\cos \delta \bullet \vec{d}_{0}-\sin \delta \bullet \vec{v}_{0}$

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~A}_{2}, d_{H}, a, b, \delta$.

### 2.4 1H-tetrahedral configuration

An example is the $\mathrm{H}_{\mathrm{a}}$ atom bound to the Ca atom of non-glycine residues (Fig. 2C). The configuration is represented schematically in Fig. 4. The H atom forms a tetrahedron with $\mathrm{A}_{1}, \mathrm{~A}_{2}$ and $\mathrm{A}_{3}$. The unit vectors $\vec{u}_{10}, \vec{u}_{20}$ and $\vec{u}_{30}$ are defined along the directions $\mathrm{A}_{0}-\mathrm{A}_{1}$, $\mathrm{A}_{0}-\mathrm{A}_{2}$ and $\mathrm{A}_{0}-\mathrm{A}_{3}$. The vector $\vec{r}_{H 0}$ forms angles $a_{1}, a_{2}, a_{3}$ with the vectors $\vec{u}_{10}, \vec{u}_{20}$ and $\vec{u}_{30}$, respectively. $\vec{r}_{H 0}$ can be expressed as a linear combination of these vectors, $\vec{r}_{H 0}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}+c \bullet \vec{u}_{30}$. The coefficients can be found from the conditions $\vec{r}_{H 0} \bullet \vec{u}_{10}=c_{1}, \vec{r}_{H 0} \bullet \vec{u}_{20}=c_{2}, \vec{r}_{H 0} \bullet \vec{u}_{30}=c_{3}$ where $c_{1}=\cos a_{1}, c_{2}=\cos a_{2}, c_{3}=\cos$ $a_{3}$. If $\omega_{12}, \omega_{23}$ and $\omega_{13}$ are angles between respective pairs of $\vec{u}_{10}, \vec{u}_{20}$ and $\vec{u}_{30}$, then these conditions become a system of linear equations,

$$
\left\{\begin{array}{l}
a+\cos \omega_{12} \bullet b+\cos \omega_{13} \bullet c=c_{1} \\
\cos \omega_{12} \bullet a+b+\cos \omega_{23} \bullet c=c_{2} \\
\cos \omega_{13} \bullet a+\cos \omega_{23} \bullet b+c=c_{3}
\end{array}\right.
$$

The system can be solved analytically (Cramer, 1750; Kosinski, 2001), yielding the parameters $a, b$ and $c$ that can be calculated using ideal angles $a_{i}$ and $\omega_{i j}$ :

This geometry is an example where four non- H atoms are used to parameterize the H atom. In principle, $\mathrm{A}_{0}$ and any two atoms of $\mathrm{A}_{1}, \mathrm{~A}_{2}, \mathrm{~A}_{3}$ are sufficient to determine the H atom position. However, if the conformation of the tetrahedron is distorted, this can lead to suboptimal H atom coordinates. The use of all three neighbors is therefore preferable.

The chain of transformations is the following:
(a) $\vec{r}_{10}=\vec{r}_{1}-\vec{r}_{0}, \vec{r}_{20}=\vec{r}_{2}-\vec{r}_{0}, \vec{r}_{30}=\vec{r}_{3}-\vec{r}_{0}$
(b) $\vec{u}_{10}, \vec{u}_{20}, \vec{u}_{30}$ are normalized vectors $\vec{r}_{10}, \vec{r}_{20}, \vec{r}_{30}$, respectively
(c) $\vec{r}_{H 0}=a \bullet \vec{u}_{10}+b \bullet \vec{u}_{20}+c \bullet \vec{u}_{30}$
(d) $\vec{u}_{H 0}=\left\|\vec{r}_{H 0}\right\|^{-1} \vec{r}_{H 0}$ with $\left\|\vec{r}_{H 0}\right\|=\left(\vec{r}_{H 0} \bullet \vec{r}_{H 0}\right)^{1 / 2}$

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~A}_{2}, \mathrm{~A}_{3}, d_{H}, a, b, c$

### 2.5 Single H with rotational degree of freedom

An example is the $\mathrm{H} \eta$ atom bound to the $\mathrm{O} \eta$ atom of tyrosine (Fig. 2D), represented schematically in Fig. 5. The ideal angle between $\mathrm{A}_{0}-\mathrm{H}$ and $\mathrm{A}_{0}-\mathrm{A}_{1}$ is $a$, and there is a
dihedral angle $\varphi$ between the planes $\mathrm{H}-\mathrm{A}_{0}-\mathrm{A}_{1}$ and $\mathrm{A}_{0}-\mathrm{A}_{1}-\mathrm{B}_{1}$. We note $a=\cos (\pi-a)=-\cos$ $a, b=\sin a \cos \varphi$ and $c=\sin a \sin \varphi$.

This configuration has one degree of freedom, the dihedral angle $\varphi$. Its value can be arbitrarily set to some a priori known value or according to the angle found in the input model, otherwise it can be optimized to minimize clashes or maximize hydrogen bonding interactions.

The chain of transformations is the following:
(a) $\vec{r}_{01}=\vec{r}_{0}-\vec{r}_{1}, \vec{r}_{B 1 A 1}=\vec{r}_{B 1}-\vec{r}_{1}$-for this configuration we calculate vectors with respect to $\mathrm{A}_{1}$
(b) $\vec{u}_{1}$ is normalized $\vec{r}_{01}$
(c) $\vec{v}=\vec{r}_{B 1 A 1}-\left(\vec{r}_{B 1 A 1} \bullet \vec{u}_{1}\right) \bullet \vec{u}_{1}$-vector in the plane $\mathrm{A}_{0}-\mathrm{A}_{1}-\mathrm{B}_{1}$ and normal to $\vec{u}_{1}$
(d) $\vec{u}_{2}$ is normalized $\vec{v}$
(e) $\vec{u}_{3}=\vec{u}_{1} \times \vec{u}_{2}$
(f) $\vec{u}_{H 0}=a \bullet \vec{u}_{1}+b \bullet \vec{u}_{2}+c \bullet \vec{u}_{3}$

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~B}_{1}, d_{H}, a, b, c$.

### 2.6 3H-propeller configuration

An example is the group of $\mathrm{H} \gamma 11, \mathrm{H} \gamma 12$ and $\mathrm{H} \gamma 13$ atoms bound to $\mathrm{C} \gamma 1$ of valine (Fig. 2E). In the ideal configuration, we note the hydrogen atoms $\mathrm{H}_{1}, \mathrm{H}_{2}, \mathrm{H}_{3}$. Each atom forms an ideal angle $a$ between $\mathrm{A}_{0}-\mathrm{H}$ and $\mathrm{A}_{0}-\mathrm{A}_{1}$. A dihedral angle $\varphi$ is defined between the planes $\mathrm{H}_{1}-\mathrm{A}_{0}-\mathrm{A}_{1}$ and $\mathrm{A}_{0}-\mathrm{A}_{1}-\mathrm{B}_{1}$. For $\mathrm{H}_{2}$ and $\mathrm{H}_{3}$ the dihedral angle is $\varphi+\frac{2 \pi}{3}$ and $\varphi+\frac{4 \pi}{3}$, respectively.

The chain of transformations and the schematic representation is similar to that for the freerotation configuration, Section 2.5 and Fig. 5. The parameters are determined for the Hatom, $\mathrm{H}_{1}$, that possesses the ideal dihedral angle. The parameters are identical for the other two H-atoms, with exception of the angle $\varphi$, which is $\varphi+n \bullet \frac{2 \pi}{3}(n=1,2)$ for the second and third H -atom. Steps (a)-(e) are defined as in Section 2.5, step (f) is given by:
(f) $\vec{u}_{H 1}=a \bullet \vec{u}_{1}+b_{1} \bullet \vec{u}_{2}+c_{1} \bullet \vec{u}_{3}, \vec{u}_{H 2}=a \bullet \vec{u}_{1}+b_{2} \bullet \vec{u}_{2}+c_{2} \bullet \vec{u}_{3}$, $\vec{u}_{H 3}=a \bullet \vec{u}_{1}+b_{3} \bullet \vec{u}_{2}+c_{3} \bullet \vec{u}_{3}$, where $a=-\cos a, b_{1}=\sin a \cos \varphi, b_{2}=\sin \alpha \cos \left(\varphi+\frac{2 \pi}{3}\right)$, $b_{3}=\sin \alpha \cos \left(\varphi+\frac{4 \pi}{3}\right)$ and $c_{1}=\sin a \sin \varphi, c_{2}=\sin \alpha \sin \left(\varphi+\frac{2 \pi}{3}\right), c_{3}=\sin \alpha \sin \left(\varphi+\frac{4 \pi}{3}\right)$.

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~B}_{1}, d_{H}, a, b, c$.

### 2.7 2H-planar configuration

An example is the $\mathrm{H} \varepsilon 21$ and $\mathrm{H} \varepsilon 22$ atoms bound to $\mathrm{N} \varepsilon 2$ of glutamine (Fig. 1F). Each of the two H -atoms forms an ideal angle a with $\mathrm{A}_{0}-\mathrm{A}_{1}$ and they are located in the plane formed by the neighboring non- H atoms. One of the H -atoms, $\mathrm{H}_{1}$, forms the dihedral angle $\varphi$ equal to 0 between the planes $\mathrm{H}_{1}-\mathrm{A}_{0}-\mathrm{A}_{1}$ and $\mathrm{A}_{0}-\mathrm{A}_{1}-\mathrm{B}_{1}$. For the second atom, $\mathrm{H}_{2}$, this angle is equal to $\pi$.

To describe the chain of transformations we reuse the algorithm for the free-rotation configuration (Section 2.5). The parameters are determined for the $\mathrm{H}_{1}$ atom with the ideal dihedral angle $\varphi=0$. Step (f) for calculating the H atom coordinates becomes:
(f) $\vec{u}_{H 1}=a \bullet \vec{u}_{1}+b_{1} \bullet \vec{u}_{2}+c_{1} \bullet \vec{u}_{3}, \vec{u}_{H 2}=a \bullet \vec{u}_{1}+b_{2} \bullet \vec{u}_{2}+c_{2} \bullet \vec{u}_{3}$, where $a=-\cos a$, $b_{1}=\sin a \cos \varphi, b_{2}=\sin a \cos (\varphi+\pi), c_{1}=\sin a \sin \varphi, c_{2}=\sin a \sin (\varphi+\pi)$

If the angle $\varphi$ is equal to zero, the coefficients can be simplified: $b_{1}=\sin a, b_{2}=-\sin a$ and $c_{1}=c_{2}=0$. However, to keep this configuration versatile to allow for rotations of the H atom group around $\mathrm{A}_{0}-\mathrm{A}_{1}$, the general expression with $\varphi$ is used.

In summary, this configuration requires the following parameters: $\mathrm{A}_{0}, \mathrm{~A}_{1}, \mathrm{~B}_{1}, d_{H}, a, b, c$.

### 2.8 H atoms with a single parent

Some hydrogen atoms have a parent atom that has no further non-hydrogen neighbors covalently bound to it. Water molecules represent an example of this situation, as each H atom has only one non-H neighbor. It follows that the H atom positions cannot be expressed from connected non-H atoms, although it may be sometimes possible to parameterize the H atom using other interactions, such as hydrogen bonds and packing.

Therefore, the approach discussed in Sections 2.1-2.7, where the hydrogen position are redefined from information about covalent bonds and angles, cannot be used for H atoms that possess only a parent or a parent and only one neighbor.

This leaves two options. The first possibility is to refine such hydrogens as free (restrained) atoms leading to an increased number of parameters: at least three coordinates per hydrogen, e.g., by six for a water molecule. Another option is to introduce constraints and refine such groups as rigid using relevant Phenix options (Afonine, Grosse-Kunstleve, Urzhumtsev, \& Adams, 2009). This increases the number of parameters by three for each rigid group regardless of the number of H atoms. In this work we focus on the parameterization of the non-water H atoms and do not discuss refinement of H atoms with a single parent here.

## 3. Riding H: Refinement targets and their gradients

The aim of refinement is to change the parameters of a model (such as the atomic positions, $\vec{r}_{0}, \vec{r}_{1}, \vec{r}, \vec{r}_{3}, \ldots, \vec{r}_{N}$ ) to optimize a function of a set of observations that may be experimental data, some statistical information, stereochemical prior knowledge, etc. The fit between model parameters and the observations is commonly expressed by a target function. The lower the value of this function, the better the fit, usually resulting in an improved
model. Explicit modeling of hydrogen positions $\vec{r}_{H_{0}}, \vec{r}_{H_{1}}, \vec{r}_{H_{2}}, \ldots, \vec{r}_{H_{M}}$, may improve the macromolecular model and reduce the value of a target expressed through coordinates of the whole set of atoms:

$$
\begin{equation*}
R\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}, \vec{r}_{3}, \ldots, \vec{r}_{N} ; \vec{r}_{H_{0}}, \vec{r}_{H_{1}}, \ldots, \vec{r}_{H_{M}} ; \text { other parameters } \rightarrow \min \right. \tag{2}
\end{equation*}
$$

For simplicity, in what follows we consider only atomic coordinates. Refining the coordinates of H atoms independently increases the number of parameters, which can compromise the optimization procedure, especially when the number of experimental observations is low. Therefore, constraints can be applied in the form of riding hydrogens. For example, considering hydrogen $H_{0}$ at position $\vec{r}_{H_{0}}$ in formula (2) in a coplanar configuration defined by atoms $\mathrm{A}_{0}, \mathrm{~A}_{1}$ and $\mathrm{A}_{2}$ in the positions $\vec{r}_{0}, \vec{r}_{1}$ and $\vec{r}_{2}$ (Fig. 1), and doing similarly for other H atoms, reduces the target (2) to a function of coordinates of only non-hydrogen atoms

$$
\begin{align*}
& R_{\text {riding }}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}, \vec{r}_{3}, \ldots, \vec{r}_{N}\right)=R\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}, \vec{r}_{3}, \ldots, \vec{r}_{N}\right. \\
& \left.\vec{r}_{H_{0}}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}\right), \vec{r}_{H_{1}}(\ldots) \ldots, \vec{r}_{H_{M}}(\ldots)\right) \tag{3}
\end{align*}
$$

In practice, target (3) is calculated in two steps:
(a) Determine the coordinates of each hydrogen atom through the coordinates of neighboring non-H atoms using one of the algorithms described above, e.g., that in Section 2.2:

$$
\begin{align*}
& \vec{r}_{H_{0}}=\vec{R}_{\text {coplanar }}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}\right)=\left(x_{H_{0}}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}\right) ; y_{H_{0}}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}\right)\right. \text {; } \\
& \left.z_{H_{0}}\left(\vec{r}_{0}, \vec{r}_{1}, \vec{r}_{2}\right)\right) \tag{4}
\end{align*}
$$

(b) Calculate target (2) using a full set of atoms, both non-H and H. Crystallographic programs minimize the target (3) by applying gradient methods (Booth, 1947; Cauchy, 1847). To calculate the gradient of the target efficiently, a two-step procedure is used that applies the chain rule and inverts the chain of transformations described in Section 2 (Lunin \& Urzhumtsev, 1985). First, the gradient of the target is calculated with respect to all coordinates, as if all atoms ( H and non- H ) were independent:

$$
\begin{align*}
& \frac{\partial R}{\partial x_{0}}, \frac{\partial R}{\partial y_{0}}, \frac{\partial R}{\partial z_{0}}, \frac{\partial R}{\partial x_{1}}, \frac{\partial R}{\partial y_{1}}, \frac{\partial R}{\partial z_{1}}, \ldots \frac{\partial R}{\partial z_{\mathrm{N}}} ; \frac{\partial R}{\partial x_{H_{0}}}, \frac{\partial R}{\partial y_{H_{0}}}, \frac{\partial R}{\partial z_{H_{0}}}, \ldots, \frac{\partial R}{\partial x_{H_{M}}}, \frac{\partial R}{\partial y_{H_{M}}}, \\
& \frac{\partial R}{\partial z_{H_{M}}} \tag{5}
\end{align*}
$$

Then the gradient is recalculated with respect to the independent parameters, for example, in the case of atom $\mathrm{H}_{0}$, expressed as in Eq. (4):

$$
\begin{equation*}
\frac{\partial R_{\text {riding }}}{\partial x_{0}}=\frac{\partial R}{\partial x_{0}}+\frac{\partial R}{\partial x_{H_{0}}} \frac{\partial x_{H_{0}}}{\partial x_{0}}+\frac{\partial R}{\partial y_{H_{0}}} \frac{\partial y_{H_{0}}}{\partial x_{0}}+\frac{\partial R}{\partial z_{H_{0}}} \frac{\partial z_{H_{0}}}{\partial x_{0}} \tag{6}
\end{equation*}
$$

and similarly for all other coordinates. Each H atom contributes to the gradients of the nonH atoms that are used to describe its position. If the hydrogen group contains several H atoms, for example, as described in Sections 2.6 and 2.7, the derivatives of $R_{\text {riding }}$ contain similar contributions from all of them:

$$
\begin{align*}
& \frac{\partial R_{\text {riding }}}{\partial x_{0}}=\frac{\partial R}{\partial x_{0}}+\frac{\partial R}{\partial x_{H 1}} \frac{\partial x_{H 1}}{\partial x_{0}}+\frac{\partial R}{\partial y_{H 1}} \frac{\partial y_{H 1}}{\partial x_{0}}+\frac{\partial R}{\partial z_{H 1}} \frac{\partial z_{H 1}}{\partial x_{0}}+\frac{\partial R}{\partial x_{H 2}} \frac{\partial x_{H 2}}{\partial x_{0}} \\
& +\frac{\partial R}{\partial y_{H 2}} \frac{\partial y_{H 2}}{\partial x_{0}}+\frac{\partial R}{\partial z_{H 2}} \frac{\partial z_{H 2}}{\partial x_{0}}+\ldots \tag{7}
\end{align*}
$$

Also, a non-H atom may be used to parameterize several hydrogen groups; then the gradient with respect to its coordinates is a sum of contributions from all these groups. The coordinates of the non- H atoms are refined; at each refinement iteration, the H atom position is updated from these new values of $x_{0}, y_{0}, z_{0}$, etc.

To simplify the calculation, approximations such as $\frac{\partial x_{H}}{\partial x_{0}}=\frac{\partial y_{H}}{\partial x_{0}}=\frac{\partial z_{H}}{\partial x_{0}}=1$ and $\frac{\partial x_{H}}{\partial x_{1}}=\frac{\partial y_{H}}{\partial x_{1}}=\frac{\partial z_{H}}{\partial x_{1}}=0$, can be made (Bourhis, Dolomanov, Gildea, Howard, \& Puschmann, 2015). In our work, such approximations are not used and actually are not required. The transformation of the gradients based on the different H atom configurations is summarized in the Appendix.

## 4. Constructing the riding H model in $C C T B X$

### 4.1 Internal data structure for model and connectivity information

The $C C T B X$ contains several entities describing an atomic model. One is the hierarchy (iotbx.pdb.hierarchy), which is a nested data structure describing the macromolecule at the model, chain, residue and atom level. The hierarchy is a container of atomic model information that makes it possible to access the atomic attributes of a model from files (such as mmCIF; Adams et al., 2019), such as chain identifiers, residue names, atom labels, coordinates and so on (Grosse-Kunstleve \& Adams, 2010). While the hierarchy contains as much information as is available in the corresponding model file, it lacks any notion of atomic connectivity and covalent geometry. This information is available in another representation of the atomic model: the cctbx.geometry_restraints class (Grosse-Kunstleve, Afonine, \& Adams, 2004). An object of this class is built from the hierarchy and from the information available in stereochemical libraries such as the Monomer Library (Engh \& Huber, 1991; Vagin et al., 2004) or GeoStd (Moriarty \& Adams, n.d., http://sourceforge.net/ projects/geostd). The cctbx.geometry_restraints class is aware of atomic connectivity, it stores ideal and actual values of geometric parameters of the model (such as bonds, angles, torsions, planes, chiral volumes, and non-bonded interactions) and calculates functions to support refinement, such as the restraints target function and its derivatives.

Together, the coordinates of non-hydrogen atoms as well as the hierarchy and cctbx.geometry_restraints objects can be used to describe the configuration of hydrogen atoms in the model and calculate their riding model parameters (Sections 2.2-2.8) using ideal bond lengths and angles.

### 4.2 Building riding H from known connectivity

The riding H algorithm operates in two stages. First, the procedure determines the connectivity of H atoms in the input model. This means that for each H atom the information about atoms linked to each other (as well as corresponding bond lengths and angles) is accumulated for all neighbors defining its position. In the second step, this information is used to parameterize each H atom. This involves computing the coefficients described in Section 2 and checking if the input geometry is consistent with the assigned configuration. For example, if one H atom is missing in a propeller group, the information in the connectivity will be the same as for a 2 H -planar group (such as in the ARG head group). Without checks, the riding H procedure would assume that the two H atoms are in a $2 \mathrm{H}-$ planar configuration. Such instances are therefore not parameterized as riding H because it is difficult to determine the actual configuration in this case.

When multiple conformations are present, determining the connectivity becomes complex. An H atom may then have several parents and neighbors, leading to a complicated network of connections. An example is shown in Fig. 6; residue i (right) adopts a double conformation and residue $i+1$ (left) has a single conformation. To maintain planarity of the peptide unit that connects the two residues, the amide H atoms are also in a double conformation while the amide N atom is in a single conformation. For example, the H atom in conformation A has the parent atom N , which in turn is connected to the H atom in conformation $B$, the $C$ atoms of both conformations $A$ and $B$, and atom $C_{a}$ in residue $i+1$. To parameterize this H atom in a coplanar configuration, the correct subset of connections is needed. In this example, the atoms in conformation B are irrelevant for the atom H in conformation A . The assignment of parent and neighbor atoms in the presence of multiple conformations is achieved by exploiting the hierarchy object, which interprets the atomic model in terms of alternates (Grosse-Kunstleve \& Adams, 2010).

The algorithms that build the connectivity of H atoms and determine the parameterization are written in Python. This step typically takes $10 \%$ of the computing time necessary to generate the hierarchy and cctbx.geometry_restraints objects. The algorithms that rebuild H atoms and calculate the gradients are written in C++. They are typically an order of magnitude faster than calculating the target function (less than 0.1 s in most cases).

## 5. Summary

This work outlines future directions of the synergetic use of neutron and X-ray diffraction data and describes the implementation of the riding hydrogen model as a fundamental prerequisite for it. We argue that the traditional joint X-ray and neutron refinement approach that yields one atomic model meant to fit both datasets is inadequate. Instead, we suggest refining two atomic models, each one fitting the respective dataset, while also benefitting from the expectation that these models should be similar.

Riding H atoms can be found in several different configurations that are not specific to protein residues. For each configuration, the H atom position is parameterized using coordinates from a set of non-H neighbors and ideal bond lengths and angles. We successfully implemented the riding hydrogen model for typical H atom geometries. The parameterization and the expressions for the gradients of refinement targets with respect to the independent parameters are described for each configuration.

The riding H option is available in $C C T B X$ and Phenix since release 1.15.

## Acknowledgments


#### Abstract

This work was supported by the US National Institutes of Health (NIH) (grant P01GM063210), the Phenix Industrial Consortium and the NIH-funded (grant R01GM071939) Macromolecular Neutron Consortium between Oak Ridge National Laboratory and Lawrence Berkeley National Laboratory. This work was supported in part by the US Department of Energy under Contract No. DE-AC02-05CH1123. A.U. acknowledges the support and the use of resources of the French Infrastructure for Integrated Structural Biology FRISBI ANR-10-INBS-05 and of Instruct-ERIC.


## Appendix.: Gradient calculation for riding-H atoms

## A. 1 Chain rule and common steps

## A.1.1 Gradient components

Each riding H group consists of one or several H atoms at the positions
$\vec{r}_{H m}=\left(x_{H m}, y_{H m}, z_{H m}\right), m=0,1, \ldots, M$, and several non-hydrogen atoms at the positions $\vec{r}_{k}=\left(x_{k}, y_{k}, z_{k}\right), k=0,1, \ldots, K$ including the 'parent' atom with $k=0$ to which the hydrogen atoms of the group are bound. We suppose that we know the components of the gradient of the target $R$ with respect to all atoms:

$$
\vec{\nabla}_{H m} R=\left(\frac{\partial R}{\partial x_{H m}}, \frac{\partial R}{\partial y_{H m}}, \frac{\partial R}{\partial z_{H m}}\right) ; \vec{\nabla}_{r k} R=\left(\frac{\partial R}{\partial x_{k}}, \frac{\partial R}{\partial y_{k}}, \frac{\partial R}{\partial z_{k}}\right)
$$

The gradient $\vec{\nabla}_{r k} R_{\text {riding }}$ of the target with respect to independent parameters (coordinates of non-hydrogen atoms) is a sum of $\vec{\nabla}_{r k} R$ with the contributions $\vec{\nabla}_{r k H} R$, recalculated from $\vec{\nabla}_{H m} R$, possibly from several H atoms and several H groups.

Calculating these contributions $\vec{\nabla}_{r k H} R$ requires inversion of the formulae used to express the riding H atoms. Below we present calculation steps for each H atom configuration. There are some common expressions, so they are elaborated once and later only referred to.

## A.1.2 Riding hydrogen position

The last step for all configurations consists in calculating the hydrogen position at the distance $d_{H}$ along the unit vector $\vec{u}_{H 0}=\left(x_{u H 0}, y_{u H 0}, z_{u H 0}\right)$ that starts from the parent non-H atom $\mathrm{A}_{0}$ with coordinates $\vec{r}_{0}=\left(x_{0}, y_{0}, z_{0}\right)$ :

$$
\vec{r}_{H}=\vec{r}_{0}+d_{H} \vec{u}_{H 0}
$$

For the gradient calculation, we start from the gradient components $\vec{\nabla}_{H} R$ supposed to be known. Then

$$
\begin{gathered}
\vec{\nabla}_{u H 0} R=d_{H} \vec{\nabla}_{H} R \\
\vec{\nabla}_{r 0 H} R=\vec{\nabla}_{H} R
\end{gathered}
$$

## A.1.3 Vector normalization

Normalization of a vector $\vec{r}=(x, y, z)$ gives a new vector

$$
\vec{u}=\left(x_{u}(x, y, z), y_{u}(x, y, z), z_{u}(x, y, z)\right)=\|\cdot\|^{-1}(x, y, z)
$$

where $\|\cdot\|=\left(x^{2}+y^{2}+z^{2}\right)^{1 / 2}$. Let $\vec{\nabla}_{u} R=\left(\frac{\partial R}{\partial x_{u}}, \frac{\partial R}{\partial y_{u}}, \frac{\partial R}{\partial z_{u}}\right)$ be known. Then the gradient $\vec{\nabla}_{r} R$ with respect to $(x, y, z)$ has the $x$-component

$$
\frac{\partial R}{\partial x}=\frac{\partial R}{\partial x_{u}} \frac{\partial x_{u}}{\partial x}+\frac{\partial R}{\partial y_{u}} \frac{\partial y_{u}}{\partial x}+\frac{\partial R}{\partial z_{u}} \frac{\partial y_{u}}{\partial x}=\|\cdot\|^{-1} \frac{\partial R}{\partial x_{u}}-x\|\cdot\|^{-3}\left(\vec{r} \bullet \vec{\nabla}_{u} R\right)
$$

and similar expressions for $\frac{\partial R}{\partial y}$ and $\frac{\partial R}{\partial z}$. Or in vector form:

$$
\vec{\nabla}_{r} R=\|\cdot\|^{-1} \vec{\nabla}_{u} R-\|\cdot\|^{-3}\left(\vec{r} \cdot \vec{\nabla}_{u} R\right) \vec{r}
$$

## A.1.4 Cross product

The vector product of $\vec{r}_{1}=\left(x_{1}, y_{1}, z_{1}\right)$ and $\vec{r}_{2}=\left(x_{2}, y_{2}, z_{2}\right)$ gives

$$
\vec{v}=\vec{r}_{1} \times \vec{r}_{2}=\left(x_{v}, y_{v}, z_{v}\right)=\left(y_{1} z_{2}-z_{1} y_{2}, z_{1} x_{2}-x_{1} z_{2}, x_{1} y_{2}-y_{1} x_{2}\right)
$$

Let $\vec{\nabla}_{v} R=\left(\frac{\partial R}{\partial x_{v}}, \frac{\partial R}{\partial y_{v}}, \frac{\partial R}{\partial z_{v}}\right)$ be known. Then

$$
\vec{\nabla}_{r 1} R=\vec{r}_{2} \times \vec{\nabla}_{v} R,
$$

$$
\vec{\nabla}_{r 2} R=\vec{\nabla}_{v} R \times \vec{r}_{1}
$$

## A. 2 Gradient for particular riding H configurations

The next sections explain how to calculate the gradient of the target function with respect to independent coordinates, starting from the values given in Section A.1.1. For each configuration, we explicitly list the steps to which we apply the chain rule. Generally, we invert the steps described for each geometry, starting from the last and moving backward.

## A.2.1 Coplanar configuration

For definition of parameters, see Section 2.2.
We start from the last step (d) and move backward to (a) after getting $\vec{\nabla}_{u H 0} R$ and $\vec{\nabla}_{r 0 H} R$ from $\vec{\nabla}_{H} R$ according to Section A.1.2.
(d) obtain $\vec{\nabla}_{r H 0} R$ from $\vec{\nabla}_{u H 0} R$ according to Section A.1.3
(c) $\vec{\nabla}_{u 10} R=a \vec{\nabla}_{r H 0} R, \vec{\nabla}_{u 20} R=b \vec{\nabla}_{r H 0} R$
(b) obtain $\vec{\nabla}_{r 10} R$ from $\vec{\nabla}_{u 10} R$ and $\vec{\nabla}_{r 20} R$ from $\vec{\nabla}_{u 20} R$ according to Section A.1.3
(a) $\vec{\nabla}_{r 1} R_{\text {riding }}=\vec{\nabla}_{r 10} R, \vec{\nabla}_{r 2} R_{\text {riding }}=\vec{\nabla}_{r 20} R \vec{\nabla}_{r 0} R_{\text {riding }}=\vec{\nabla}_{r 0 H} R-\vec{\nabla}_{r 10} R-\vec{\nabla}_{r 20} R$

## A.2.2 2H-tetrahedral configuration

For definition of parameters, see Section 2.3.
We start from getting $\vec{\nabla}_{u H 0} R$ and $\vec{\nabla}_{r 0 H} R$ from $\vec{\nabla}_{H} R$ according to Section A.1.2.
(g) $\vec{\nabla}_{d 0} R=(\cos \delta) \vec{\nabla}_{d 0} R, \vec{\nabla}_{\nu 0} R=(\sin \delta) \vec{\nabla}_{d 0} R$
(f) $\vec{\nabla}_{d} R$ from $\vec{\nabla}_{d 0} R$ according to Section A.1.3
(e) $\vec{\nabla}_{u 10}^{(1)} R=a \vec{\nabla}_{d} R$ and $\vec{\nabla}_{u 20}^{(1)} R=b \vec{\nabla}_{d} R$
(d) obtain $\vec{\nabla}_{v} R$ from $\vec{\nabla}_{v 0} R$ according to Section A.1.3
(c) obtain $\vec{\nabla}_{u 10}^{(2)} R$ and $\vec{\nabla}_{u 20}^{(2)}$ from $\vec{\nabla}_{v} R$ according to Section A.1.4 and then calculate $\vec{\nabla}_{u 10} R=\vec{\nabla}_{u 10}^{(1)} R+\vec{\nabla}_{u 10}^{(2)} R$ and $\vec{\nabla}_{u 20} R=\vec{\nabla}_{u 20}^{(1)} R+\vec{\nabla}_{u 20}^{(2)} R$
(b) obtain $\vec{\nabla}_{r 10} R$ from $\vec{\nabla}_{u 10} R$ and $\vec{\nabla}_{r 20} R$ from $\vec{\nabla}_{u 20} R$ according to Section A.1.3

$$
\text { (a) } \vec{\nabla}_{r 1} R_{\text {riding }}=\vec{\nabla}_{r 10} R, \vec{\nabla}_{r 2} R_{\text {riding }}=\vec{\nabla}_{r 20} R \vec{\nabla}_{r 0} R_{\text {riding }}=\vec{\nabla}_{r 0 H} R-\vec{\nabla}_{r 10} R-\vec{\nabla}_{r 20} R
$$

## A.2.3 1H-tetrahedral configuration

For definition of parameters, see Section 2.4.
We start from getting $\vec{\nabla}_{u H 0} R$ and $\vec{\nabla}_{r 0 H} R$ from $\vec{\nabla}_{H} R$ according to Section A.1.2.
(d) obtain $\vec{\nabla}_{r H 0} R$ from $\vec{\nabla}_{u H 0} R$ according to Section A.1.3
(c) $\vec{\nabla}_{u 10} R=a \vec{\nabla}_{r H 0} R, \vec{\nabla}_{u 20} R=b \vec{\nabla}_{r H 0} R, \vec{\nabla}_{u 30} R=c \vec{\nabla}_{r H 0} R$
(b) obtain $\vec{\nabla}_{r 10} R$ from $\vec{\nabla}_{u 10} R, \vec{\nabla}_{r 20} R$, from $\vec{\nabla}_{u 20} R, \vec{\nabla}_{r 30} R$ from $\vec{\nabla}_{u 30} R$ according to Section A.1.3
(a) $\vec{\nabla}_{r 1} R_{\text {riding }}=\vec{\nabla}_{r 10} R, \vec{\nabla}_{r 2} R_{\text {riding }}=\vec{\nabla}_{r 20} R, \vec{\nabla}_{r 3} R_{\text {riding }}=\vec{\nabla}_{r 30} R$, $\vec{\nabla}_{r 0} R_{\text {riding }}=\vec{\nabla}_{r 0 H} R-\vec{\nabla}_{r 10} R-\vec{\nabla}_{r 20} R-\vec{\nabla}_{r 30} R$

## A.2.4 Single H with rotational degree of freedom

For definition of parameters, see Section 2.5.

We start from getting $\vec{\nabla}_{u H 0} R$ and $\vec{\nabla}_{r 0 H} R$ from $\vec{\nabla}_{H} R$ according to Section A.1.2.
(f) $\vec{\nabla}_{u 10}^{(1)} R=a \vec{\nabla}_{r H 0} R, \vec{\nabla}_{u 20}^{(1)} R=b \vec{\nabla}_{r H 0} R, \vec{\nabla}_{u 30} R=c \vec{\nabla}_{r H 0} R$
(e) obtain $\vec{\nabla}_{u 10}^{(2)} R$ and $\vec{\nabla}_{u 20}^{(2)} R$ from $\vec{\nabla}_{u 30} R$ according to Section A.1.4; superscript (2) is used to distinguish these derivatives from those calculated above $\vec{\nabla}_{u 20}^{(1)} R$ and $\vec{\nabla}^{(1)}{ }_{u 10} R$;
$\vec{\nabla}_{u 2} R=\vec{\nabla}_{u 20}^{(1)} R+\vec{\nabla}_{u 20}^{(2)} R ;$
(d) obtain $\vec{\nabla}_{v} R$ from $\vec{\nabla}_{u 2} R$ according to Section A.1.2
(c) $\vec{\nabla}_{r B A} R=\vec{\nabla}_{v} R-\vec{u}_{1}\left(\vec{u}_{1} \bullet \vec{\nabla}_{v} R\right) ; \vec{\nabla}_{u 1 v} R=-\vec{\nabla}_{v} R\left(\vec{r}_{B 1 A 1} \bullet \vec{u}_{1}\right)-\vec{r}_{B 1 A 1}\left(\vec{u}_{1} \bullet \vec{\nabla}_{v} R\right)$; $\vec{\nabla}_{u 1} R=\vec{\nabla}_{u 10}^{(1)} R+\vec{\nabla}_{u 10}^{(2)} R+\vec{\nabla}_{u 1 v} R$
(b) obtain $\vec{\nabla}_{r 01} R$ from $\vec{\nabla}_{u 1} R$ according to Section A.1.2
(a) $\vec{\nabla}_{r 1} R_{\text {riding }}=-\vec{\nabla}_{r 01} R-\vec{\nabla}_{r B A} R, \vec{\nabla}_{r B 1} R_{r i d i n g}=\vec{\nabla}_{r B 1} R, \vec{\nabla}_{r 0} R_{r i d i n g}=\vec{\nabla}_{r 0 H} R+\vec{\nabla}_{r 01} R$

## A.2.5 3H-propeller

For definition of parameters, see Section 2.6.

We start from getting $\vec{\nabla}_{u H 1} R, \vec{\nabla}_{r 0 H 1} R, \vec{\nabla}_{u H 2} R, \vec{\nabla}_{r 0 H 2} R$ and $\vec{\nabla}_{u H 3} R, \vec{\nabla}_{r 0 H 3} R$ from $\vec{\nabla}_{H 1} R, \vec{\nabla}_{H 2} R, \vec{\nabla}_{H 3} R$, respectively, according to Section A.1.2. Then
(f) $\vec{\nabla}_{u 10} R=a \sum_{m=1}^{3} \vec{\nabla}_{r 0 H m} R, \vec{\nabla}_{u 20} R=\sum_{m=1}^{3} b_{m} \vec{\nabla}_{r 0 H m} R, \vec{\nabla}_{u 30} R=\sum_{m=1}^{3} c_{m} \vec{\nabla}_{r 0 H m} R$

Steps (e)-(a) repeat the steps from Section A.2.4 except the last expression for $\vec{\nabla}_{r 0} R_{\text {riding }}$ that becomes

$$
\vec{\nabla}_{r 0} R_{\text {riding }}=\vec{\nabla}_{r 01} R+\sum_{m=1}^{3} \vec{\nabla}_{r 0 H m} R
$$

## A.2.6 2H-planar

For definition of parameters, see Section 2.7.
We start from getting $\vec{\nabla}_{u H 1} R, \vec{\nabla}_{r 0 H 1} R$ and $\vec{\nabla}_{u H 2} R, \vec{\nabla}_{r 0 H 2} R$ from $\vec{\nabla}_{H 1} R$ and $\vec{\nabla}_{H 2} R$, respectively, according to Section A.1.2. Then
(f) $\vec{\nabla}_{u 10} R=a\left(\vec{\nabla}_{r 0 H 1} R+\vec{\nabla}_{r 0 H 2} R\right), \vec{\nabla}_{u 20} R=b\left(\vec{\nabla}_{r 0 H 1} R+\vec{\nabla}_{r 0 H 2} R\right)$,
$\vec{\nabla}_{u 30} R=c\left(\vec{\nabla}_{r 0 H 1} R+\vec{\nabla}_{r 0 H 2} R\right)$
Steps (e)-(a) repeat the steps from Section A.2.4 except the last expression for $\vec{\nabla}_{r 0} R_{\text {riding }}$ that becomes

$$
\vec{\nabla}_{r 0} R_{\text {riding }}=\vec{\nabla}_{r 01} R+\vec{\nabla}_{r 0 H 1} R+\vec{\nabla}_{r 0 H 2} R
$$

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## Further reading

Blakeley MP, Ruiz F, Cachau R, Hazemann I, Meilleur F, Mitschler A, et al. (2008). Quantum model of catalysis based on a mobile proton revealed by subatomic x-ray and neutron diffraction studies of h-aldose reductase. Proceedings of the National Academy of Sciences of the United States of America, 105, 1844-1848. [PubMed: 18250329]
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Howard EI, Blakeley MP, Haertlein M, Haertlein IP, Mitschler A, Fisher SJ, et al. (2011). Neutron structure of type-III antifreeze protein allows the reconstruction of AFP-ice interface. Journal of Molecular Recognition, 24, 724-732. [PubMed: 21472814]
Howard EI, Guillot B, Blakeley MP, Haertlein M, Moulin M, Mitschler A, et al. (2016). Highresolution neutron and X-ray diffraction room-temperature studies of an H-FABP-oleic acid complex: Study of the internal water cluster and ligand binding by a transferred multipolar electron-density distribution. IUCrJ, 3, 115-126.


Fig. 1.
Example of a hydrogen atom H in a coplanar configuration. To improve visibility the parameters are shown in two figures. H is covalently bound to its parent atom $\mathrm{A}_{0}$, the neighbors are $\mathrm{A}_{1}$ and $\mathrm{A}_{2}$.


Fig. 2.
Examples of hydrogen atom configurations. While the examples shown in the figure refer to amino acid residues, the implemented riding-H algorithm is general and not specific to proteins only.


Fig. 3.
H -atoms in 2H-tetrahedral configuration. To improve visibility the parameters are shown on two figures. In the ideal configurations, the positions of $\mathrm{H}_{1}$ and $\mathrm{H}_{2}$ are symmetric with respect to the plane $A_{1}-A_{0}-A_{2}$. The vector $d_{0}$ is bisector of the angle $H_{1}-A_{0}-H_{2}$ and is in the plane $\mathrm{A}_{1}-\mathrm{A}_{0}-\mathrm{A}_{2}$.


Fig. 4.
$H$ atom in a $1 H$-tetrahedral position. The vector $A_{0}-H$ makes angles $a_{1}, a_{2}, a_{3}$ with vectors $A_{0}-A_{1}, A_{0}-A_{2}, A_{0}-A_{3}$, which form angles $\omega_{12}, \omega_{13}, \omega_{23}$ between themselves.


Fig. 5.
H -atom in the free-rotation configuration. Alternative positions are indicated with $\mathrm{H}_{2}$ and $H_{3}$. Only one atom among $B_{1}$ and $B_{2}$ defines a dihedral angle with $A_{1}-A_{0}-H_{1}$ (in this example: $\mathrm{B}_{1}$ ) and is therefore used to parameterize the H atom.


Fig. 6.
Illustration of connectivity when the H atom adopts a double conformation. Atom name labels are in orange, alternative conformation labels are in black.


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