UCLA UCLA Previously Published Works

Title

Nazarov cyclization of 1,4-pentadien-3-ols: preparation of cyclopenta[b]indoles and spiro[indene-1,4'-quinoline]s

Permalink https://escholarship.org/uc/item/0rq1q1j7

Journal Chemical Communications, 52(13)

ISSN 1359-7345

Authors

Wang, Zhiming Xu, Xingzhu Gu, Zhanshou <u>et al.</u>

Publication Date 2016-02-14

DOI

10.1039/c5cc08596a

Peer reviewed



HHS Public Access

Chem Commun (Camb). Author manuscript; available in PMC 2017 August 17.

Published in final edited form as:

Author manuscript

Chem Commun (Camb). 2016 February 14; 52(13): 2811-2814. doi:10.1039/c5cc08596a.

Nazarov cyclization of 1,4-pentadien-3-ols: preparation of cyclopenta[*b*]indoles and spiro[indene-1,4[']-quinoline]s[†]

Zhiming Wang^a, Xingzhu Xu^a, Zhanshou Gu^a, Wei Feng^a, Houjun Qian^a, Zhengyi Li^a, Xiaoqiang Sun^a, and Ohyun Kwon^b

^aJiangsu Key Laboratory of Advanced Catalytic Materials and Technology, School of Petrochemical Engineering, Changzhou University, Changzhou, Jiangsu 213164, P. R. China.

^bDepartment of Chemistry and Biochemistry, University of California, Los Angeles, 607 Charles E. Young Drive East, Los Angeles, CA 90095-1569, USA.

Abstract

The first Lewis acid-catalyzed intramolecular interrupted Nazarov cyclization of 1,4-pentadien-3ols is described. Using FeBr₃ as the catalyst, a series of new substituted cyclopenta[*b*]indoles was prepared—through a sequence of Nazarov cyclization, nucleophilic amination, and isomerization —with good yields and high diastereo- and regioselectivities. A similar catalytic process was also developed for the synthesis of structurally interesting spiro[indene-1,4'-quinoline]s.

> Indoles and their derivatives, especially multicyclic indole compounds, are seemingly ubiquitous in natural products and pharmaceutical agents, acting as important key structural motifs in many bioactive molecules.¹ Among them, cyclopenta[b]indoles occupy a significant place in the fields of natural products and medicinal chemistry. There are numerous biologically active indole alkaloids and medicinally important compounds containing a cyclopenta[b]indole unit as a core structure, including yuehchukene,² fischerindoles,³ terpendoles,⁴ bruceollines,⁵ malasseziacitrin,⁶ and the drug laropiprant.⁷ Traditionally, cyclopenta[b]indoles have been constructed through Fischer indole synthesis from the corresponding phenylhydrazines and cyclopentanones.⁸ The major drawbacks of the classic Fischer indole synthesis, however, are the limited substrate scope and the lack of regioselectivity. Although intramolecular Friedel-Crafts cyclization reactions of indole with alkenes can also provide access to cyclopenta[b]indoles, this approach can require multistep preparations of the starting materials and harsh reaction conditions.⁹ Consequently, the development of new methodologies for the construction of the cyclopenta[b]indole scaffold under mild reaction conditions with high efficiency and selectivity remains an active research area.¹⁰ Methods involving both pyrrole and cyclopentane ring formation in a single step are particularly desirable.¹¹

The Nazarov cyclization, a primary tool for the preparation of cyclopentenone compounds, has been applied widely in the synthesis of natural products and bioactive molecules.¹² In

[†]Electronic supplementary information (ESI) available. CCDC 1414046–1414050 and 1414065. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c5cc08596a

Correspondence to: Zhiming Wang; Ohyun Kwon.

particular, the interrupted Nazarov cyclization has been developed extensively for rapid access to complex functionalized cyclopentenones.¹³ Although the Nazarov cyclization of divinyl ketones has been studied thoroughly, examples of Nazarov cyclization based on 1,4pentadien-3-ols are relatively scarce.¹⁴ In the limited examples, 1,4-pentadien-3-ols were treated with a Brønsted or a Lewis acid to form pentadienyl cations, which underwent 4π electrocylic ring closure to afford the cyclopentadiene products.^{14b,d,i} The broader application of this transformation has been hampered, however, by a lack of control with respect to the position of the double bonds.^{14a,b,i} For example, Hall reported recently that the arylboronic acid-catalyzed Nazarov reaction of a divinyl alcohol furnished a 1: 2.2 mixture of regioisomeric cyclopentadienes.^{14a} Moreover, to the best of our knowledge, the interrupted Nazarov cyclization of 1,4-pentadien-3-ols has not been reported previously. As a means of controlling the regiochemistry of the Nazarov cyclization of 1,4-pentadien-3-ols, we envisioned intramolecular trapping of the allylic cation by an appended aniline unit (Scheme 1). Accordingly, treatment of 3-(2-anilinyl)-1,4-pentadien-3-ol (1) with either a Brønsted or a Lewis acid should produce the pentadienyl cation 2, which, upon Nazarov cyclization, should form the allylic cation 3. Potentially, the steric bias endowed by the substitution around the cyclopentene cation would control its trapping by the aniline unit on the less-hindered side, selectively forming the single regioisomer 5, which is likely to isomerize to yield the indole 6.

To explore the feasibility of this transformation, we chose (E)-3-(2-tosylamidophenyl)-1phenylpenta-1,4-dien-3-ol (1a) as the model substrate (Table 1). Treatment of 1a with 30 mol% FeBr₃ in CHCl₃ at 50 °C gave 2-phenyl-4-tosyl-1,2,3,4-tetrahydrocyclopenta[b]indole (6a) as the sole product in 75% yield through the expected reaction sequence (entry 1). The structure of **6a** was confirmed through X-ray crystallographic analysis.¹⁵ Employing FeCl₃, AlCl₃, or ZrCl₄ as the catalyst resulted in a markedly lower yield (10-55%) of the desired product (entries 2-4). In contrast, changing the catalyst to AuCl₃ or HBr led to the production of 2-pentyl-1-tosyl-4-vinyl-1,2-dihydroquinoline (7), presumably through intramolecular allylic amination of the pentadienyl cation precursor of the Nazarov cyclization (entries 5 and 6).¹⁶ A lower yield of **6a** (60%) was afforded when using CH₂Cl₂ as the solvent (entry 7). An even less polar solvent (toluene) provided an even lower yield and worse selectivity of the product 6a, while THF and MeCN delivered the 1,2dihydroquinoline by-product 7 exclusively (entries 8–10). Increasing the catalyst loading to 40-100 mol% did not improve the outcome of the reaction (entries 11 and 12). Lowering the catalyst loading to 20-5 mol% provided only a moderate yield of the cyclopenta[b]indole **6a**, along with the by-product **7** (entries 13 and 14). A similar result was obtained when diluting the original mixture fivefold (entry 15). It seems that the reaction pathway is highly dependent on the concentration of the catalyst in the reaction system. Gratifyingly, this intramolecular interrupted Nazarov cyclization could be easily performed on the 4 mmol scale, giving the desired product **6a** in 73% yield (entry 16).

With the optimized conditions in hand, the scope of the reaction was examined using a variety of disubstituted 1,4-pentadien-3-ols (Table 2). *N*-Methanesulfonyl, *N*-benzenesulfonyl, *N*-(4-chloro)-benzenesulfonyl, and *N*-(4-methoxy)benzenesulfonyl (*E*)-3-(2-aminophenyl)-1-phenylpenta-1,4-dien-3-ols all functioned well in the cascade reaction,

As conjectured, the intramolecular amination of the allylic cation intermediate $\mathbf{3} \leftrightarrow \mathbf{4}$ occurred away from the sterically more demanding substituent. Encouraged by these results, several polysubstituted 1,4-pentadien-3-ols ($\mathbf{1s}$ - \mathbf{w}) were synthesized to examine the regioand diastereoselectivities of the reaction and the substituent effects of the 1,4-pentadien-3-ol (Scheme 2). The (*E*,*E*)-3-(2-tosylaminophenyl)-1,5-diarylpenta-1,4-dien-3-ols $\mathbf{1s}$ - \mathbf{u} reacted well, giving only their *trans* products $\mathbf{6s}$ - \mathbf{u} . The relative configuration of the *trans* isomer was established unequivocally through X-ray crystallographic analysis of compound $\mathbf{6t}$.^{15,17} The tetrasubstituted 1,4-pentadien-3-ol $\mathbf{1v}$ also gave a good yield of a single product ($\mathbf{6v}$). Again, the regioselectivity of the reaction appeared to be induced by steric bias. When (*E*)-3-(2-tosylaminophenyl)-1-methyl-1-phenylpenta-1,4-dien-3-ol ($\mathbf{1w}$) was applied, however, both the cyclopenta[*b*]indole $\mathbf{6w}$ (34%) and the spiro[indene-1,4'-quinoline] $\mathbf{8w}$ (24%) were isolated, with the latter formed presumably through Nazarov cyclization on the benzene ring and subsequent FeBr₃-catalyzed intramolecular hydroamination of the monosubstituted alkene moiety.^{18,19}

Because of their importance in natural product synthesis and pharmaceuticals, various methods have been developed for the preparation of quinoline derivatives.²⁰ Surprisingly, efficient synthetic pathways for the construction of structurally interesting spiro[indene-1,4'- quinoline] frameworks are very rare. We envisioned that changing the methyl group on the C-1 atom of **1w** to a phenyl group might stabilize the carbocation at the C-1 position of the substrate and improve the selectivity of the reaction to afford the spiro[indene-1,4'- quinoline] product. Gratifyingly, the use of various *N*-sulfonyl 3-(2-sulfonamidoaryl)-1,1- diphenylpenta-1,4-dien-3-ols **1x-ab** as substrates furnished several unusual spiro[indene-1,4'-quinoline] compounds **8x-ab** as the sole products in yields of 59–70% (Scheme 3). The Nazarov cyclization/hydroamination cascade leading to the spiro[indene-1,4'-quinoline] framework could accommodate *N*-tosyl, *N*-mesyl, *N*-4-chlorobenzenesulfonyl, and chloro groups on the substrates. The structures of compounds **6v** and **8y** were confirmed using X-ray crystallography.¹⁵

To further verify the reaction mechanism proposed in Scheme 1, three different sets of experiments were conducted (Scheme 4). When the reaction was run in the presence of three equivalents of EtOH, the putative carbocation underwent solvolysis to form the ethyl ether **9** in 26% yield. Moreover, the key reaction intermediate 2,3,3*a*,4- tetrahydrocyclopenta[*b*]indole **10** was isolated as a single *cis* diastereoisomer when the reaction was conducted at 30 °C. The ratio of the products **6a**, **7**, and **10** was highly dependent on the reaction time. After 15 min, **7** and **10** were obtained in yields of only 35 and 45%, respectively. As the reaction progressed, the yield of **6a** increased and the yield of

10 decreased, while the yield of **7** remained almost constant. When the indoline **10** was subjected to the optimized reaction conditions, the cyclopenta[*b*]indole **6a** was isolated in 90% yield. These results corroborate the notion that this cyclopenta[*b*]indole synthesis involves an intramolecular interrupted Nazarov cyclization and isomerization sequence, with the 2,3,3*a*,4-tetrahydrocyclopenta[*b*]indole **10** as the key reaction intermediate. The structures of compounds **9** and **10** were confirmed using X-ray crystallography.¹⁵

In conclusion, new Lewis acid-catalyzed cascade reactions based on Nazarov cyclization of 1,4-pentadien-3-ols have been developed, providing substituted cyclopenta[b]indoles and spiro[indene-1,4'-quinoline]s in good yields. The advantages of this approach are the use of inexpensive and environmentally friendly FeBr₃ as the catalyst, relatively mild reaction conditions, and exclusive regio- and diastereoselectivities. This facile and efficient methodology appears to be a useful tool for the synthesis of biologically important cyclopenta[b]indole and spiro[indene-1,4'-quinoline] derivatives.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

This study was supported by the NSF of China [grant 21372033 (to Z. W.)]; the Priority Academic Program Development of Jiangsu Higher Education Institutions; Jiangsu Key Laboratory of Advanced Catalytic Materials and Technology (BM2012110); the Qing Lan Project (to Z. W.); the NSF of the Jiangsu Higher Education Institutions of China [grants 12KJA150002 (to Z. L.) and 14KJA150002 (to Z. W.)]; and the NSF [CHE 7096481 (to O. K.)].

Notes and references

- For recent selected reviews on indoles, see:(a) Lancianesi S, Palmieri A, Petrini M. Chem Rev. 2014; 114:7108. [PubMed: 24905229] (b) Kochanowska-Karamyan AJ, Hamann MT. Chem Rev. 2010; 110:4489. [PubMed: 20380420] (c) Sharma V, Kumar P, Pathak D. J Heterocycl Chem. 2010; 47:491.
- 2. Kong YC, Cheng KF, Cambie RC, Waterman PG. J Chem Soc, Chem Commun. 1985; 47
- Stratmann K, Moore RE, Bonjouklian R, Deeter JB, Patterson GML, Shaffer S, Smith CD, Smitka TA. J Am Chem Soc. 1994; 116:9935.
- 4. Nakazawa J, Yajima J, Usui T, Ueki M, Takatsuki A, Imoto M, Toyoshima YY, Osada H. Chem Biol. 2005; 10:131.
- 5. Jordan JA, Gribble GW, Badenock JC. Tetrahedron Lett. 2011; 52:6772.
- 6. Irlinger B, Bartsch A, Krämer H-J, Mayser P, Steglich W. Helv Chim Acta. 2005; 88:1472.
- 7. Lai E, De Lepeleire I, Crumley TM, Liu F, Wenning LA, Michiels N, Vets E, O'Neill G, Wagner JA, Gottesdiener K. Clin Pharmacol Ther. 2007; 81:849. [PubMed: 17392721]
- (a) Ratni H, Blum-Kaelin D, Dehmlow H, Hartman P, Jablonski P, Masciadri R, Maugeais C, Patiny-Adam A, Panday N, Wright M. Bioorg Med Chem Lett. 2009; 19:1654. [PubMed: 19231176] (b) Shafiee A, Upadhyay V, Corley EG, Biba M, Zhao D, Marcoux JF, Campos KR, Journet M, King AO, Larsen RD, Grabowski EJJ, Volante RP, Tillyer RD. Tetrahedron: Asymmetry. 2005; 16:3094.
 (c) Lacoume B, Milcent G, Olivier A. Tetrahedron. 1972; 28:667.
- 9. (a) Yokosaka T, Nakayama H, Nemoto T, Hamada Y. Org Lett. 2013; 15:2978. [PubMed: 23745602]
 (b) Banwell MG, Ma X, Taylor RM, Willis AC. Org Lett. 2006; 8:4959. [PubMed: 17020346] (c) Baran PS, Richter JM. J Am Chem Soc. 2005; 127:15394. [PubMed: 16262402] (d) Harrison CA, Leineweber R, Moody CJ, Williams JMJ. Tetrahedron Lett. 1993; 34:8527.(e) Bonjouklian R, Moore RE, Patterson GML. J Org Chem. 1988; 53:5866.

- For recent selected examples of the synthesis of cyclopenta[b]indole scaffolds, see:(a) Lebée C, Kataja AO, Blanchard F, Masson G. Chem – Eur J. 2015; 21:8399. [PubMed: 25892287] (b) Kotha S, Ali R, Srinivas V, Krishna NG. Tetrahedron. 2015; 71:129.(c) Ivanova OA, Budynina EM, Skvortsov DA, Trushkov IV, Melnikov MY. Synlett. 2014:2289.(d) Li E, Li C, Wang J, Wang J, Dong L, Guo X, Song C, Chang J. Tetrahedron. 2014; 70:874.(e) Chu XQ, Zi Y, Lu XM, Wang SY, Ji SJ. Tetrahedron. 2014; 70:232.(f) Dong J, Pan L, Xu X, Liu Q. Chem Commun. 2014; 50:14797.(g) Tan W, Li X, Gong YX, Ge MD, Shi F. Chem Commun. 2014; 50:15901.(h) Zhang C, Zhang LX, Qiu Y, Xu B, Zong Y, Guo QX. RSC Adv. 2014; 4:6916.(i) Ma Y, You J, Song F. Chem – Eur J. 2013; 19:1189. [PubMed: 23238950] (j) Prasad B, Sreenivas BY, Krishna GR, Kapavarapu R, Pal M. Chem Commun. 2013; 49:6716.(k) Xu B, Guo ZL, Jin WY, Wang ZP, Peng YG, Guo QX. Angew Chem, Int Ed. 2012; 51:1059.(l) Zi W, Wu H, Toste FD. J Am Chem Soc. 2015; 137:3225. [PubMed: 25710515]
- (a) Dhiman S, Ramasastry SSV. Chem Commun. 2015; 51:557.(b) Xia G, Han X, Lu X. Org Lett. 2014; 16:2058. [PubMed: 24650163] (c) Saito K, Sogou H, Suga T, Kusama H, Iwasawa N. J Am Chem Soc. 2011; 133:689. [PubMed: 21171651] (d) Feldman KS, Hester DK II, Iyer MR, Munson P, López CS, Faza ON. J Org Chem. 2009; 74:4958. [PubMed: 19472993] (e) Feldman K, Hester SDK II, López CS, Faza ON. Org Lett. 2008; 10:1665. [PubMed: 18345683] (f) Feldman KS, Iyer MR, Hester DK II. Org Lett. 2006; 8:3113. [PubMed: 16805565] (g) Dhiman S, Ramasastry SSV. Org Lett. 2015; 17:5116. [PubMed: 26434732]
- For recent reviews on Nazarov cyclization, see:(a) Wenz DR, de Alaniz JR. Eur J Org Chem. 2015:23.(b) Tius MA. Chem Soc Rev. 2014; 43:2979. [PubMed: 24196585] (c) Spencer WT III, Vaidya T, Frontier AJ. Eur J Org Chem. 2013:3621.(d) Shimada N, Stewart C, Tius MA. Tetrahedron. 2011; 67:5851. [PubMed: 21857751] (e) Vaidya T, Eisenberg R, Frontier AJ. ChemCatChem. 2011; 3:1531.
- For a review on interrupted Nazarov cyclization, see:(a) Grant TN, Rieder CJ, West FG. Chem Commun. 2009:5676.For recent selected examples of interrupted Nazarov cyclization, see:(b) Kwon Y, Schatz DJ, West FG. Angew Chem, Int Ed. 2015; 54:9940.(c) Shenje R, Williams CW, Francois KM, France S. Org Lett. 2014; 16:6468. [PubMed: 25495709] (d) William R, Wang S, Ding F, Arviana EN, Liu XW. Angew Chem, Int Ed. 2014; 53:10742.(e) Riveira MJ, Mischne MP. J Org Chem. 2014; 79:8244. [PubMed: 25075431]
- 14. (a) Zheng H, Lejkowski M, Hall DG. Tetrahedron Lett. 2013; 54:91.(b) Hastings CJ, Backlund MP, Bergman RG, Raymond KN. Angew Chem, Int Ed. 2011; 50:10570.(c) Rieder CJ, Winberg KJ, West FG. J Org Chem. 2011; 76:50. [PubMed: 21105713] (d) Hastings CJ, Pluth MD, Bergman RG, Raymond KN. J Am Chem Soc. 2010; 132:6938. [PubMed: 20443566] (e) Harrington PE, Li L, Tius MA. J Org Chem. 1999; 64:4025.(f) Halterman RL, Tretyakov A. Tetrahedron. 1995; 51:4371.(g) Erker G. Pure Appl Chem. 1991; 63:797.(h) Threlkel RS, Bercaw JE, Seidler PF, Stryker JM, Bergman RG, Hill DE, White JD. Org Synth. 1987; 65:42.
- 15. Crystallographic data for 6a, 6t, 6v, 8y, 9, and 10 have been deposited with the CCDC 1414065, 1414046, 1414047, 1414050, 1414048, and 1414049, respectively.
- 16. Wang Z, Li S, Yu B, Wu H, Wang Y, Sun X. J Org Chem. 2012; 77:8615. [PubMed: 22934926]
- 17. (E,Z)-3-(2-Tosylaminophenyl)-1,5-diphenylpenta-1,4-dien-3-ol provided a messy reaction mixture, from which no desired cis product could be isolated.
- 18. Singh R, Panda G. Org Biomol Chem. 2011; 9:4782. [PubMed: 21347498]
- 19. Komeyama K, Morimoto T, Takaki K. Angew Chem, Int Ed. 2006; 45:2938.
- 20. (a) Basavaiah D, Reddy BS, Badsara SS. Chem Rev. 2010; 110:5447. [PubMed: 20735052] (b) Barluenga J, Rodríguez F, Fañanás F. Chem Asian J. 2009; 4:1036. [PubMed: 19360759] (c) Madapa S, Tusi Z, Batra S. Curr Org Chem. 2008; 12:1116.(d) Kouznetsov VV, Vargas Méndez LY, Meléndez Gómez CM. Curr Org Chem. 2005; 9:141.



6









Scheme 3.

Synthesis of spiro[indene-1,4'-quinoline]s.



Scheme 4. Mechanistic studies.

Table 1

Wang et al.

Optimization of reaction conditions^a

Hole Catt. Catt. Catt. Catt. Catt. Catt. Solvent Ga T Intro solvent solvent Temp. (°C) Time (h) Ga T Intro FeB ₁₃ Solvent Temp. (°C) Time (h) Ga T I FeB ₁₃ Solvent Temp. (°C) Time (h) Ga T I FeB ₁₃ Solvent Temp. (°C) Time (h) Ga T I FeB ₁₃ Solvent Temp. (°C) Time (h) Ga T I FeB ₁₃ CHCl ₃ Solvent T T T I FeB ₁₃ CHCl ₃ Solvent 0.5 T T T I FeB ₁₃ CHCl ₃ Solvence 0.5 T T T I FeB ₁₃ CHCl ₃ Solvence 0.5 T T T I FeB ₁₃ CHCl ₃ Solvence 0.5 T <th></th> <th></th> <th></th> <th></th> <th></th> <th></th> <th></th>							
Yielde (vio) Yielde (vio) Yielde (vio) FeBr3 CHCl3 Solvent Temp. (°C) Time (h) 6a 7 FeBr3 CHCl3 50 0.5 75 - FeCl3 CHCl3 50 0.5 75 - FeCl3 CHCl3 50 0.5 75 - AlCl3 CHCl3 50 0.5 75 - AlCl3 CHCl3 50 0.5 10 40 AlCl3 CHCl3 50 0.5 10 40 AuCl3 CHCl3 50 0.5 - 71 HBr CHCl3 50 0.5 - 71 HBr Thlr 50 0.5 - 40 FeBr3 THr 50 2 - 40 FeBr3 CHCl3 50 0.5 - 40 FeBr3 CHCl3 50 0.5 - 40	- // \\	ta the second	cat. solvent	6a Ts	<u>لا</u>	Z H	
htpCatalystSolventTemp. (°C)Time (h)637FeBr3CHCl3500.575FeCl3CHCl3500.555FeCl3CHCl3500.555AlCl3CHCl3500.510AlCl3CHCl3500.510AlCl3CHCl3500.50.510AlCl3CHCl3500.50.571AuCl3CHCl3500.50.570HBrCHCl3500.50.540FeBr3THF50266FeBr3CHCl3500.50.567FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568FeBr3dCHCl3500.50.568F						Yield	(%) ql
FeBr3 CHCl3 50 0.5 75 $-$ FeCl3 CHCl3 50 0.5 55 $-$ AICl3 CHCl3 50 0.5 55 $-$ AICl3 CHCl3 50 0.5 10 $-$ AuCl3 CHCl3 50 0.5 10 $-$ AuCl3 CHCl3 50 0.5 $-$ 71 HBr CHCl3 50 0.5 $-$ 71 HBr CHCl3 50 0.5 $-$ 71 FeBr3 THF 50 0.5 $-$ 40 FeBr3 THF 50 0.5 $-$ 68 I FeBr3 CHCl3 50 0.5 $-$ 68 FeBr3 CHCl3 50 0.5 $-$ 68 $-$ FeBr3 FeBr3 CHCl3 50 0.5 $-$ 68 FeBr3 FeBr3 S0	ntry	Catalyst	Solvent	Temp. (°C)	Time (h)	6a	٢
FeCl ₃ CHCl ₃ 50 0.5 55 -1 AlCl ₃ CHCl ₃ 50 0.5 10 -1 I ZrCl ₄ CHCl ₃ 50 0.5 10 -1 I ZrCl ₄ CHCl ₃ 50 0.5 10 -1 I AuCl ₃ CHCl ₃ 50 0.5 -1 -1 I HBr CHCl ₃ 50 0.5 -1 -1 I FeBr ₃ Three 50 0.5 -1 -40 I FeBr ₃ Three 50 0.5 -1 -46 I FeBr ₃ CHCl ₃ 50 0.5 -1 -1 I FeBr ₃ CHCl ₃ 50 0.5 -1 -1 I FeBr ₃ CHCl ₃ 50 0.5 -1 -1 I FeBr ₃ CHCl ₃ 50 0.5 -1 -1 <t< td=""><td></td><td>FeBr₃</td><td>CHC1₃</td><td>50</td><td>0.5</td><td>75</td><td> </td></t<>		FeBr ₃	CHC1 ₃	50	0.5	75	
$AICl_3$ $CHCl_3$ 50 0.5 10 -1 I $ZrCl_4$ $CHCl_3$ 50 0.5 10 40 T $AuCl_3$ $CHCl_3$ 50 0.5 10 40 T HBr $CHCl_3$ 50 0.5 -1 71 T HBr $CHCl_3$ 50 0.5 60 -1 40 T $FeBr_3$ $Thuene$ 50 0.5 60 -1 40 T $FeBr_3$ $Thuene$ 50 0.5 -1 40 10 $FeBr_3$ $CHCl_3$ 50 0.5 70 -1 11 $FeBr_3$ $CHCl_3$ 50 0.5 70 -1 12 $FeBr_3$ $CHCl_3$ 50 0.5 40 40 14 $FeBr_3$ $CHCl_3$ 50 0.5 40 40 5	0	FeCl ₃	CHC1 ₃	50	0.5	55	I
	~	AICI ₃	CHC1 ₃	50	0.5	10	I
λ LuCl ₃ CHCl ₃ 50 0.5 - 71 β HBr CHCl ₃ 50 0.5 - 71 γ FeBr ₃ CH2cl ₃ 50 0.5 - 40 γ FeBr ₃ CH ₂ Cl ₂ 40 0.5 60 - ϑ FeBr ₃ Thlre 50 2 30 40 ϑ FeBr ₃ CH ₃ CN 50 2 - 66 - 11 FeBr ₃ CH3CN 50 0.5 60 - 68 11 FeBr ₃ CHCl ₃ 50 0.5 67 - 68 12 FeBr ₃ CHCl ₃ 50 0.5 67 - 68 14 FeBr ₃ CHCl ₃ 50 0.5 - 68 - 14 FeBr ₃ CHCl ₃ 50 0.5 - 80 - 15 FeBr ₃ CHCl ₃		$ZrCl_4$	CHC1 ₃	50	0.5	10	40
	10	AuCl ₃	CHC1 ₃	50	0.5		71
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	2	HBr	CHC1 ₃	50	0.5		40
8 FeBr ₃ Toluene 50 2 30 40 0 FeBr ₃ THF 50 2 - 46 10 FeBr ₃ CH ₃ CN 50 1 - 46 11 FeBr ₃ CHCl ₃ 50 0.5 70 - 68 11 FeBr ₃ d CHCl ₃ 50 0.5 65 - 68 12 FeBr ₃ d CHCl ₃ 50 0.5 65 - 68 13 FeBr ₃ d CHCl ₃ 50 0.5 40 48 14 FeBr ₃ d CHCl ₃ 50 0.5 - 80 15 FeBr ₃ d CHCl ₃ 50 0.5 72 45 16 FeBr ₃ d CHCl ₃ 50 0.5 73 - 80	2	FeBr_3	CH_2Cl_2	40	0.5	60	
ϑ FeBr ₃ THF 50 2 $$ 46 10 FeBr ₃ CH ₃ CN 50 1 $$ 68 11 FeBr ₃ CHCl ₃ 50 0.5 70 $-$ 12 FeBr ₃ CHCl ₃ 50 0.5 65 $-$ 13 FeBr ₃ CHCl ₃ 50 0.5 65 $-$ 14 FeBr ₃ CHCl ₃ 50 0.5 40 48 15 FeBr ₃ CHCl ₃ 50 0.5 42 45 15 FeBr ₃ CHCl ₃ 50 0.5 73 $-$	~	FeBr_3	Toluene	50	2	30	40
	6	FeBr_3	THF	50	2		46
I1 $FeBr_3c$ CHCl ₃ 50 0.5 70 $-$ 12 $FeBr_3d$ CHCl ₃ 50 0.5 65 $-$ 13 $FeBr_3d$ CHCl ₃ 50 0.5 65 $-$ 14 $FeBr_3d$ CHCl ₃ 50 0.5 40 48 15 $FeBr_3d$ CHCl ₃ 50 0.5 42 45 16 $FeBr_3d$ CHCl ₃ 50 0.5 73 $-$	10	FeBr_3	CH ₃ CN	50	1		68
	Ξ	$\mathrm{FeBr}_3^\mathcal{C}$	CHC1 ₃	50	0.5	70	
13 $FeBr_3e$ CHCl ₃ 50 0.5 40 48 14 $FeBr_3f$ CHCl ₃ 50 0.5 80 15 $FeBr_3e$ CHCl ₃ 50 0.5 42 45 16 $FeBr_3h$ CHCl ₃ 50 0.5 73	12	FeBr_3^d	CHCl ₃	50	0.5	65	
$ \begin{array}{rrrrr} \mbox{I4} & \mbox{FeBr}_3 f & \mbox{CHCl}_3 & \mbox{50} & \mbox{0.5} & - & \mbox{80} \\ \mbox{I5} & \mbox{FeBr}_3 g & \mbox{CHCl}_3 & \mbox{50} & \mbox{0.5} & \mbox{42} & \mbox{45} \\ \mbox{I6} & \mbox{FeBr}_3 h & \mbox{CHCl}_3 & \mbox{50} & \mbox{0.5} & \mbox{73} & - \\ \end{array} $	13	$\mathrm{FeBr}_3^{\mathcal{O}}$	CHCl ₃	50	0.5	40	48
I5 FeBr ₃ e CHCl ₃ 50 0.5 42 45 16 FeBr ₃ h CHCl ₃ 50 0.5 73 -	14	${ m FeBr}_3^f$	CHC1 ₃	50	0.5		80
16 FeBr ₃ h CHCl ₃ 50 0.5 73 —	15	$\mathrm{FeBr}_3\mathcal{B}$	CHC1 ₃	50	0.5	42	45
	16	FeBr_{3}^{h}	CHC1 ₃	50	0.5	73	
	Isolated	yield.					

Chem Commun (Camb). Author manuscript; available in PMC 2017 August 17.

 $\mathcal{C}_{0.08}$ mmol of FeBr3 was used.

Author Manuscript	0.2 mmol of FeBr3 was used. 6.04 mmol of FeBr3 was used.	$f_{ m 0.01}$ mmol of FeBr3 was used.	^g 10 mL of CHCl3 was used.	$h_{ m Reaction}$ performed on the 4 mmol scale.
-------------------	---	---------------------------------------	---------------------------------------	--

Author Manuscript

Author Manuscript

Author Manuscript

Page 11

Table 2

Intramolecular interrupted Nazarov cyclizations of disubstituted 1,4-pentadien-3-ols^{a,b}



 $^a\mathbf{1}$ (0.2 mmol) and FeBr3 (0.06 mmol) were reacted in CHCl3 (2 mL) at 50 °C for 0.5 h.

^bIsolated yield.