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Ir(III)-Based Agents for Monitoring the Cytochrome P450 3A4 Active Site Occupancy

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Abstract

Cytochromes P450 (CYPs) are a superfamily of enzymes responsible for biosynthesis and drug metabolism. Monitoring the activity of CYP3A4, the major human drug-metabolizing enzyme, is vital for assessing the metabolism of pharmaceuticals and identifying harmful drug–drug

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Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.inorgchem.2c02587. Experimental procedures for the synthesis and spectral data for **2–5**, procedures for photophysical, pharmacological, and biological studies, and X-ray crystallographic data (PDF)

interactions. Existing probes for CYP3A4 are irreversible turn-on substrates that monitor activity at specific time points in end-point assays. To provide a more dynamic approach, we designed, synthesized, and characterized emissive Ir(III) and Ru(II) complexes that allow monitoring of the CYP3A4 active-site occupancy in real time. In the bound state, probe emission is quenched by the active-site heme. Upon displacement from the active site by CYP3A4-specific inhibitors or substrates, these probes show high emission turn-on. Direct probe binding to the CYP3A4 active site was confirmed by X-ray crystallography. The lead Ir(III)-based probe has nanomolar K_d and high selectivity for CYP3A4, efficient cellular uptake, and low toxicity in CYP3A4-overexpressing HepG2 cells.

Cytochromes P450 (CYPs) are crucial enzymes responsible for biomolecule synthesis and drug metabolism. Among 57 human CYPs, CYP3A4 is the major drug-metabolizing enzyme responsible for oxidizing the majority of pharmaceuticals.¹ Because of high substrate promiscuity and plasticity of the active site, CYP3A4 is implicated in many drug– drug interactions that can cause drug toxicity.^{2–5} Additionally, CYP3A4 displays genetic polymorphism, where mutations facilitate or slow down drug metabolism, thereby affecting the therapeutic efficiency.^{6–8} These attributes make CYP3A4 an important target for activity monitoring, especially in complex systems such as liver microsomes and hepatocytes that model human drug metabolism *in vitro*. Current methods for monitoring CYP3A4 activity involve marker substrates, which require cumbersome and costly high-performance liquid chromatography analyses conducted over multiple time points, or irreversible turn-on reagents that make it difficult to monitor changes to CYP3A4 activity over time (Figure 1A).^{9–14} As an alternative approach to these classical methods, in this Communication, we report emissive Ir(III) and Ru(II) complexes that allow sensing of the occupancy of the CYP3A4 active site (Figure 1B).

We chose to examine Ir(III) and Ru(II) complexes as probes for CYP3A4 because they are powerful tools for monitoring biological activity.^{15–24} Probes of this class have long luminescence lifetimes, ranging from hundreds of nanoseconds up to ~100 μ s,^{15,16,25} which allows for time-resolved gating that can be used to exclude background emission from biomolecules and fluorogenic substrates. Thus, these compounds were expected to provide a distinct advantage over previous CYP3A4 probes containing organic-based fluorescent groups,²⁶ whose low nanosecond lifetimes preclude the measurement of CYP activity in human liver microsomes, the gold standard in drug metabolism.

Transition-metal-based probes were designed to interact with a hydrophobic surface within the substrate access channel of CYP3A4²⁷ and included a pyridyl side chain (see R₁ in Figure 2) to anchor the complex to the enzyme through direct heme iron coordination. Emissive sensors **2–5** were synthesized as racemic mixtures of and Λ isomers (Figure 2A). Ligand **1** was heated with the Ru(II) precursors *cis*-[Ru(L₁)₂Cl₂] (L₁ = 2,2[']-bipyridine or 1,10-phenanthroline), which gave compounds **2** and **3**. Alternatively, treating **1** with [Ir(μ -Cl)(C^N)₂]₂ [C^N = 2-phenylpyridine (ppy) or 2-phenylquinoline (pq)] gave complexes **4** and **5**. Complexes **2–5** were characterized by ¹H NMR, IR, and electronic absorption spectroscopies and electrospray ionization mass spectrometry. All data were consistent with the structures shown in Figure 2. Importantly, electronic absorption and emission spectra for

2–5 were in good agreement with data for the parent Ru(II) or Ir(III) complexes devoid of the R₁ side chain.^{28–30} All complexes emit brightly when excited with 435 nm light (Figure S7), with **4** having the highest emission quantum yield of 0.086(9) and a lifetime of 1.6 μ s, over twice as long as those of **2** and **3** (Table 1).

Equilibrium titration of CYP3A4 with 2–5 showed that all complexes exhibit type II binding, indicative of strong pyridine nitrogen coordination to the heme (Figure 3A,C–F). Spectral dissociation constants for 2–5 are listed in Table 1. Complexes 4 and 5 are far more potent than the Ru(II) inhibitors 2 and 3, indicating that CYP3A4 preferably binds monocationic over dicationic complexes. Importantly, attachment of the R₁ side chain dramatically increases the inhibitory potency, by nearly 100-fold. The control compound 6 shows type I binding (a blue shift in the Soret band) and is a weak inhibitor with a K_d value of $11.2 \pm 0.08 \,\mu$ M, whereas analogue 5 with the pyridyl-containing R₁ chain exhibits type II binding, with a stronger affinity of 130 ± 11 nM. Both the binding affinity determined from the equilibrium titrations and the IC₅₀ data indicated that Ir(III) sensors bind tighter and inhibit CYP3A4 more potently than Ru(II) compounds, with tunable K_d values as low as 70 ± 2 nM for 4.

Next, Ir(III) complexes **4** and **5** were cocrystallized with CYP3A4 (Figures 3C–F and S8). In both structures, the inhibitor's R_1 side chain curls above the heme and the terminal pyridine N ligates to the heme Fe (Fe–N distance of 2.20–2.23 Å). Hydrophobic residues Phe108, Phe220, Phe57, and Leu482 are in close contact with the ppy and pq groups of **4** and **5**, respectively. The electron density was well-defined for the heme-ligating pyridine, part of the tether, and the Ir(III) cores. The Ir ligands were poorly defined, which suggests that both the and Λ isomers of **4** and **5** were bound to the active site. The stereochemistry was not specified during structural refinement, but the Λ and isomers (shown in Figure 3C–F) were preferably selected for **4** and **5**, respectively, and fit into electron density maps by the refining program. Importantly, **4** and **5** are the first Ir complexes characterized to bind to a CYP enzyme.^{31–38}

To ensure that **4** binds to CYP3A4 more selectively than to other CYP isoforms, the IC₅₀ values of **4** against CYP3A4, CYP1A2, and CYP2C9 were determined using commercially available inhibitor screening kits (BioVision). Data from these kits versus the soluble reconstituted system in Table 1 cannot be compared directly because they were acquired under different conditions.²⁷ The derived IC₅₀ values were 2.8 ± 1.0 , >100, and 79 ± 6 μ M for CYP3A4, CYP2C9, and CYP1A2, respectively (Figure 3G). The 28- and >36-fold difference in IC₅₀ demonstrates the high selectivity and preferential binding of **4** to a larger and expandable active site of CYP3A4 (Figure S8). For comparison, the volume of the active-site cavity in ligand-free CYP3A4 is 1400 Å³ relative to 375 and 470 Å³ in CYP1A2 and CYP2C9, respectively.^{7,39,40}

With compound **4** identified as a lead, we evaluated its ability to act as an active-site photosensor by measuring changes in emission intensity upon the addition of ligand-free or substrate/inhibitor-bound CYP3A4 (Figure 3H). Strong luminescence quenching was observed when **4** (5 μ M) was mixed with ligand-free CYP3A4 (3 μ M), consistent with other emissive probes for P450 enzymes.^{41–45} The quenching was partial when CYP3A4

was bound to a substrate or inhibitor prior to the addition of **4**. Importantly, the emission levels were ligand-dependent and correlated with the ligands' binding affinity: the strongest CYP3A4 binder, ritonavir ($K_d = 19$ nM), was the most difficult to displace, whereas the weakly bound substrate, testosterone (K_d of 1.5 and 30 μ M for two binding sites), was expelled by the probe more easily.

To further substantiate the scope of our lead compound **4**, we assessed its inhibitory properties in HepG2 human hepatoma cells, where expression of most drug-metabolizing CYPs is negligible or absent. However, when HepG2 cells were stably engineered with vectors expressing CYPs, the protein levels reached those in primary human hepatocytes, which makes this model cell line a convenient *in vitro* tool to mimic drug metabolism in the liver.^{46–48} To determine the CYP3A4 inhibitory activity of **4**, HepG2 cells overexpressing CYP3A4 were used in conjuction with a bioluminiscent P450-Glo CYP3A4 assay. Importantly, a strong concentration-dependent decrease in activity was observed, with statistically significant inhibition at 300 nM (Figure 3I; ~20% inhibition and P < 0.05 vs control). These data confirm that **4** is able to efficiently penetrate HepG2-CYP3A4 cells and inhibit CYP3A4 activity at nanomolar concentrations.

Finally, to demonstrate that our photosensors can be visualized in cells, we employed fluorescence microscopy. HepG2-CYP3A4 cells were treated with **4** (5 μ M) for 1 h (Figure 3J), then rinsed with phosphate-buffered saline (PBS; pH 7.0), and imaged using the GFP channel. We found that **4** is cell-permeable and can be visually detected at concentrations as low as 5 μ M. Utilization of metal complexes at such low concentrations limits their cell toxicity. In fact, **4** is well-tolerated by HepG2-CYP3A4 cells (EC₅₀ > 50 μ M), as judged by a cellular viability assay (Figure 3K, MTT, 72 h). This result provides strong evidence that cell toxicity can be avoided or largely minimized when Ir(III) complexes are used as photosensors at low concentrations (<10 μ M).

In summary, Ir(III) compound **4** is a potent and specific inhibitor that serves as a photosensor for CYP3A4 active-site occupancy. The luminescence of **4** is quenched upon binding to CYP3A4 and recovers in a manner proportional to the binding affinity of CYP3A4 substrates and inhibitors. Furthermore, photosensor **4** penetrates and inhibits CYP3A4 in hepatic cells and emits brightly in the intracellular environment. This new class of photosensors is expected to provide a significant advantage over traditional end-point assays currently used for the detection of drug–drug interactions of CYP3A4 in cells. Another beneficial property of our photosensors is their prolonged luminescence lifetimes, which allow time-resolved emission measurements for excluding autofluorescence, a major problem in bioimaging that cannot be addressed with the current sensors. Studies are now underway in our laboratories to further develop this class of compounds and utilize Ir(III) photosensors for monitoring CYP3A4 active-site occupancy *in cellulo*.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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A. Irreversible monitoring of CYP3A4 due to oxidation by CYP.



B. Novel and reversible monitoring of CYP3A4 due to quenching by CYP.



Figure 1.

Emissive probes for monitoring metabolism by CYP3A4.







Figure 3.

(A) Spectral changes observed during equilibrium titration of CYP3A4 with **4**. The inset contains difference spectra. (B) Titration plot with the derived K_d values. (C and D) Crystal structure of the **4**–CYP3A4 complex at 2.78 Å resolution (PDB 7UAY). (E and F) Crystal structure of the **5**-CYP3A4 complex at 2.65 Å resolution (PDB 7UAZ). The blue and green meshes in panels C and E are $2F_o - F_c$ and polder-omit electron density maps contoured at the 1σ and 3σ levels, respectively. (G) Inhibition of CYP3A4, CYP1A4, and CYP2C9 activity by **4**. (H) Fluorescence spectra of **4** (5 μ M) in the absence and presence of CYP3A4 (3 μ M) bound to different substrates and inhibitors (10–20 μ M) showing ligand-dependent emission yields (0.1 M PBS, pH 7.4, 10% glycerol, and $\lambda_{ex} = 433$ nm). (I) CYP3A4 activity with **4** (0.3–10 μ M) determined by P450-Glo CYP3A4 assay or ketoconazole (1 μ M) as a positive control. Concentrations of 0.3–10 μ M are statistically significant from a control-containing vehicle: *, P < 0.05. (J) Fluorescence microscopy image (GFP filter) of HepG2-CYP3A4 cells treated with **4** (5 μ M). The inset is control fluorescence from vehicle-treated cells. (K) Cell viability at different concentrations of **4** (0.05–50 μ M) determined by a cellular viability assay (MTT, 72 h).

Table 1.

Dissociation Constants (Kd), IC50 Values for CYP3A4 (µM), and Emission Quantum Yields for Sensors 2-5

			H ₂ O	
compound	$K_{\rm d}^{\ a}(\mu{ m M})$	$\mathrm{IC}_{50}^{\ b}\left(\mu\mathrm{M} ight)$	$\Phi_{ m em}^{c}$	τ (μs)
2	53 ± 4	6.0 ± 0.5	0.046(3)	0.66
3	23 ± 2	3.1 ± 0.4	0.042(9)	0.75
4	$0.070 \pm 0.0.002$	0.25 ± 0.02	0.086(9)	1.6
5	0.130 ± 0.011	0.20 ± 0.01	0.007(1)	0.062
6	11.2 ± 0.8	1.02 ± 0.02	ND	ND

^aDetermined by spectrophotometric titration assay.

 b CYP3A4 activity assay with BFC, 293 ± 3 K, 0.2 μ M CYP3A4, 0.3 μ M cytochrome P450 reductase, versus DMSO control (100% activity), and standard error <10%.

^cEmission spectra of absorption matched solutions in Arsparged H₂O ($A_{435} \sim 0.07$), with $\lambda_{ex} = 435$ nm, a 455 nm long-pass filter, referenced to Ru(bpy)₃, and $\Phi_{em} = 0.042$.