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SUMMARY OF THE RESEARCH PROGRESS MEETING

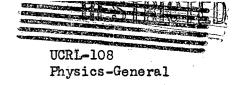
April 15, 1948

by Margaret Foss Folden

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SUMMARY OF THE RESEARCH PROGRESS MEETING

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Mass Spectrometer. F. L. Reynolds.

A spectrometer has been developed for the mass assignment of radioactive isotopes. Two slides, Figs. 1 and 2, of the instrument were shown and the parts explained.

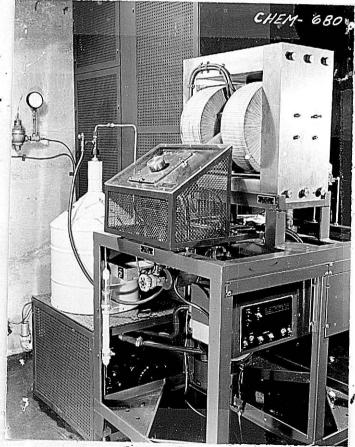


Figure 1

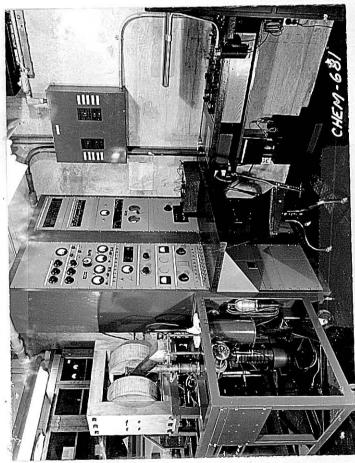


Figure 2

The source is excited by thermal means and consists of a 1 x 30 mil filament mounted on a plate, followed by a defining slit. Alternatively a bombardment source may be used for some elements. Generally the beam is recorded photographically; however, an arrangement employing an electrometer may also be used.

Measurements on the mass spectrometer give the following results:

Mass	Dispersion		
6-7	57 mm		
50-51	8 mm		
100-101	4 mm		
150-151	2.6		
200-201	2.0		
250-251	1.6		
1000-1001	.4 (Practical Resolution)		

Lines have been obtained from neodymium. However, with lead, no lines were seen from a thermal source. The instrument is not suited for abundance ratios but must be used for mass assignments.

The sample technique with the thermal source is important and the uniformity of sample on the filament has a good deal to do with uniformity of line. A photographic plate is superimposed on the initial undeveloped plate and finally both plates are developed. The stable isotopes give the location reference points for the new radioactive isotopes.

The following are the amounts and energies of various particles relative to the blackening of an Eastman III-O photographic plate:

1 x 10 ¹⁰	100 - 150 mass	8000 positive ions
1 x 10 ⁶	1 Mev	gamma rays
1 x 10 ⁵	l Mev	beta particles

Depending on the material, one positive ion will strike the photographic plate for every 10^2 to 10^6 atoms placed on the filament.

The rare earths, aluminum and magnesium are examples of usable materials on the thermal source.

Metabolism of Carrier-Free Beryllium. Kenneth Scott.

Experiments have been conducted to determine the toxic qualities of soluble salts of beryllium. These salts may be compared with mercuric chloride as to toxicity and are toxic at the milligram level. They are administered in the carrier-free state.

The experiments were done with Be 7 , prepared by the proton bombardment of spectrographically pure Li. The Be 7 had a half life of 52 days and a 4 per cent contamination with a 52 day β emitter.

Dosages were administered to rats in the left leg by intramuscular injection with the results listed below:

Present Dose ner Organ

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· · · · · · · · · · · · · · · · · · ·	1 day	4 days	8 days	16 days	32 days	64 days	
Liver	5	9	1.9	5.0	0.7	0.12	
Kidney	3	2	0.3	0.8	0.2	0.1	
Skeleton	29	32	32.0	28.0	32.0	29.0	
Urine	42	38	24.2	43.0	20.0	15.0	
Feces	11	10	37.8	16.0	40.0	54.0	
Unabsorbed Balance in Left Rear Leg	60	58	24.0	43.0	23.0	14.0	

Beryllium has been shown to cause bone tumors in rabbits.

Meson Production in the Synchrotron. L. Baumhoff.

Calculations for predicting meson production in the synchrotron have been made. The cross section for the production of mesons by 7 rays has been given by Bethe as follows:

$$4\pi \frac{e^2}{\pi c} \frac{g^2}{\pi c} \left(\frac{h}{\mu c}\right)^2 = 4.3 \times 10^{-28}$$

where v is the meson velocity.

This gives the distribution velocity of mesons produced by & rays.

The following gives the number of mesons per second produced by the Y's for a beam of one electron per sq. cm. per sec.

$$\frac{dc}{dE} = 4.3 \times 10^{-28} \sqrt{1 - \left(\frac{\mu c}{E}\right)^2} \cdot \frac{1}{E}$$

The average cross section over the range of gammas that produce mesons is

$$\overline{C} = 2.2 \times 10^{-28} \text{ cm}^2$$

The predicted cross section for electron-positron pair production is:

$$\frac{z^2}{(137)^3} \left(\frac{h}{mc}\right)^2 \qquad 2 \log \frac{E}{2 mc^2}$$

Averaged over & ray spectrum it is

$$\mathbf{z}^2$$
 (2.1 x 10⁻²⁶) per nucleus,

so that

Separation of electrons from mesons is accomplished by a magnetic field and a system of slits.

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