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Mineral properties, microbes, transport, and plant-input profiles control vertical distribution and age of soil carbon stocks

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Highlights

We present a **B**iotic and **A**biotic **M**odel of **S**OM (BAMS1) to model SOM dynamics.

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The soil bulk specific surface area is the dominant control for the SOM stocks.

Plant-input profiles have negligible impacts on the bulk SOM profiles.

Plant-input profiles are important controls for the age of SOM at depth.

Abstract

The role of organo-mineral interactions, microbial dynamics, and vertical plant input profiles are hypothesized to be important in controlling soil organic matter (SOM) stocks and dynamics. To test this hypothesis, we enhanced and applied a model (Biotic and Abiotic Model of SOM – BAMS1) that represents microbial dynamics and organomineral interactions integrated with a multiphase <u>reactive transport</u> solver for variably saturated porous media. The model represents aqueous chemistry, aqueous advection and diffusion, gaseous diffusion, sorption processes, bacterial and fungal activity, and archetypal monomer- and polymer-carbon substrate groups, including dead cell wall material. This model structure is fundamentally different, and produces different SOM dynamics, than the pseudo-first order relationships that are used in most site- and global-scale terrestrial SOM models. We simulated two grasslands, a seven-site chronosequence (~3.9–240 Ky) in Northern California, and a Russian Chernozem site where soils were sampled 100 years apart. We calibrated the model's vertically-resolved soil bulk specific surface area (SBSSA) using observed bulk SOM content, and then tested the model against observed Δ^{14} C profiles. The modeled microbial processes, organo-mineral interactions, and vertical aqueous

transport produced realistic vertically-resolved predictions of bulk SOM content, Δ^{14} C values of SOM, <u>lignin</u> content, and fungi-to-aerobic bacteria biomass ratios. Using sensitivity analyses, we found that vertical carbon input profiles were important controls over the Δ^{14} C <u>depth distribution</u>. Shallower carbon input profiles lead to older carbon at depth. In addition, the SBSSA was the dominant control over the magnitude and vertical distribution of SOM stocks. The findings of this study demonstrate the value of explicitly incorporating <u>microbial activity</u>, sorption, and vertical transport into land models to predict SOM dynamics.

Keywords BAMS1 SOM dynamics Organo-mineral interactions

1. Introduction

Global <u>soil carbon</u> stocks are approximately three times as large as the carbon stocks in either the atmosphere or terrestrial vegetation (<u>Schmidt et al., 2011</u>). This carbon may be vulnerable to decomposition and release to the atmosphere due to environmental changes (e.g., <u>thermokarst</u>, warming). Predicting long-term changes in soil carbon storage and feedbacks with <u>climate change</u> requires understanding how <u>soil organic</u> <u>matter</u> (SOM) dynamics vary vertically within the <u>soil column</u> (<u>Schmidt et al.,</u> 2011, Koven et al., 2013, Riley et al., 2014).

Mineral surfaces interact chemically with <u>dissolved organic carbon</u> (DOC) and help stabilize soil organic carbon (SOC) (<u>Greenland, 1971, Jagadamma et al., 2012, Lutzow</u> <u>et al., 2006, Sibanda and Young, 1986, Sollins et al., 1996, Trumbore, 1997</u>). The proportion and persistence of organic compounds that are at least partially protected against decomposition on soil mineral surfaces depends on several factors, including the chemical composition of the substrate, texture, <u>mineralogy</u>, pH, temperature, <u>soil</u> <u>aggregates</u>, and hydrologic transport (<u>Conant et al., 2011</u>, <u>Gjettermann et al.,</u> 2008, Jardine et al., 1989, Jardine et al., 1990, Kalbitz et al., 2003, Kothawala and Moore, 2009, Moore and Turunen, 2004, Six et al., 2004). Organo-mineral interactions may be especially important below the active root zone (<u>Ahrens et al., 2015</u>, <u>Kalbitz</u> et al., 2000, Kalbitz et al., 2003, Masiello et al., 2004, Sanderman et al., 2008, Torn <u>et al., 1997a</u>) and in deeper horizons with iron, aluminum, and clay-rich minerals (Jardine et al., 1989, Jardine et al., 1990, Jardine et al., 2006, Kothawala et al., 2009). These findings may partially explain the observed wide range of SOM residence times among sites, but it is presumable that interactions with mineral surfaces introduce more complexity than currently represented in terrestrial <u>carbon cycle</u> models. Most land models apply a pseudo-first-order representation for soil organic matter (SOM) dynamics, with several SOM 'pools' with explicit turnover times that are dynamically affected by temperature and moisture (e.g., Century (Parton et al., 1987, Parton et al., 2010), RothC (Jenkinson and Coleman, 2008)). Some of these models also represent the role of mineral protection by modifying the proportional transfers between pools and the explicit turnover times as a function of mineralogy, usually clay content (Jenkinson and Coleman, 2008, Parton et al., 2010). To our knowledge, organo-mineral interactions are represented explicitly in only one global land model (Sulman et al., 2014). Sulman et al. (2014) argued that organo-mineral interactions help explain SOM responses in Free-Air CO₂ Enrichment (FACE) experiments and have global implications for future terrestrial carbon stocks. Recently, Ahrens et al. (2015) used a site-scale model (COMISSION) and observations from a site in Bavaria to demonstrate that SOC persistence is controlled by mineral surface stabilization and cyclic microbial assimilation and release. Our previous work with the BAMS1 model (Biotic and Abiotic Model of SOM; (Riley et al., 2014)) had similar conclusions. However, these models apply different approaches for sorption: Riley et al. (2014) applied separate adsorption and desorption rates to represent mineral-protected carbon, Sulman et al. (2014) described protected carbon with a first order turnover rate, and Ahrens et al. (2015) used a Langmuir sorption approach.

The level of model complexity required to accurately predict SOC dynamics under a changing climate remains unsettled (e.g., <u>Allison et al., 2010</u>, <u>Luo et al., 2015</u>, <u>Parton et al., 2010</u>). <u>Riley et al. (2014</u>) showed that representing modest microbial diversity, <u>organic chemistry</u>, and a reasonable combination of sorption parameters led to good matches with observed soil organic carbon profiles without resorting to an arbitrary depth-dependent decline in SOM turnover rates, as is often done in existing SOM models. Upon this premise, exploring this modeling philosophy may lead to valuable characterization of vertically-resolved soil organic matter, transport, mineral surface interaction, and microbial ecological interactions. In this work, we applied BAMS1, which includes a microbially mediated SOM reaction network with two microbial functional groups (heterotrophic aerobic bacteria and fungi) and several polymer and <u>monomer</u> species. Dynamic adsorption and desorption with a surface <u>complexation</u> model were added to the model for this work.

The specific objectives of this study are to (1) explore and quantify how bulk <u>mineral</u> <u>properties</u> affect carbon stocks and residence times of soil organic matter and (2) identify dominant controls for carbon stocks and inventory. We used sensitivity analyses and comparisons with published data to investigate controls on soil carbon stocks and turnover across a seven-site <u>marine-terrace grassland chronosequence</u> in California (Masiello et al., 2004) and in a <u>Chernozem steppe</u> grassland site in Russia (<u>Torn et al., 2002</u>).

2. Methodology and model inputs

2.1. Chronosequence terrace sites in Northern California

This chronosequence comprises seven terraces on the northern California coast (between 38° to 42° N and 122° to 125° W) that were formed on the same type of beach deposits but have soil development ages of 3.9, 29, 40, 118, 124, 124, and 240 ky, leading to significant differences in texture and mineralogical composition. We identify the Chronosequence Terrace (CT) sites as: CT-3.9ky, CT-29ky, CT-40ky, CT-118ky, CT-124ky-1, CT-124ky-2, and CT-240ky. Soil class ranges from mixed sandy soil (CT-3.9ky) to coarse loamy (CT-29ky) to fine loamy (CT-40ky to CT-240ky) along the chronosequence. Clay content increases in CT sites with the age; for example, soils on CT-3.9ky and CT-29ky have 5% and 15% maximum clay content, respectively. Soils on CT-40ky, CT-118ky, CT-124ky-1, CT-124ky-2, and CT-240ky have at least 25-30% maximum clay content (Merritts et al., 1991). Soils at the CT sites have been classified on the taxonomic soil order as Alfisols (Chadwick et al., 1990). The sites have a cool Mediterranean climate, with mean annual air temperature of 12 °C and mean annual precipitation of 1000 mm (Masiello et al., 2004, Merritts et al., 1991). These chronosequence terraces can be used to compare the effect of soil <u>mineralogy</u> and development while holding climate, vegetation, and other soil-forming factors relatively constant.

Merritts et al. (1991) described the soil development and mineralogy of these terraces in detail. The older sites are clay-rich and contain crystalline forms of iron-rich minerals. Total SOM content and SOM radiocarbon (Δ^{14} C) were measured in one <u>soil profile</u> in each of the seven terraces (Masiello et al., 2004). Measured bulk SOM content is the lowest in the CT-3.9ky site, intermediate in the CT-29ky and CT-40ky sites, and highest in the older sites (CT-118ky, CT-124ky-1, CT-124ky-2, and CT-240ky) (Masiello et al., 2004). Because the older sites have experienced more weathering and have higher clay content as compared to younger sites, they present more <u>mineral surface</u>area available to sorb <u>organic carbon</u> (Mayer, 1994).

2.2. Modern and 100-year-old-soil archives in Russian steppe sites

The Russian <u>steppe</u> soils are classified as <u>Mollisols</u> on the taxonomic soil order and have relatively large carbon stocks. The soil on these sites are characterized as clayloam and granular to sub-granular structure along the depth; clay content ranges from 24% to 30% in these sites (<u>Torn et al., 2002</u>). Several studies have been conducted in this region to better understand <u>soil carbon</u> stocks and turnover (<u>Guenet et al.,</u> <u>2013, Torn et al., 2002</u>). We used data from the Kammenaya Steppe Preserve (51° 02' N, 40° 43' E). The vegetation was typical steppe <u>prairie</u>, dominated by <u>herbaceous</u> perennials with some woody perennials. The sampling sites have a mean annual temperature of 6.4 [•]C and mean annual precipitation of 520 mm, with approximately 25% as snow. The snow cover typically lasts for slightly more than three months. SOM content and Δ^{14} C values were measured in three soil profiles (Archive, Pit1, and Pit2 soils) at this site. SOM content decreases from about 8% near the surface to less than 1% at 1.5 m depth.

2.3. Carbon decomposition reaction network

To investigate how organo-mineral interactions influence SOM storage and dynamics, we modified the BAMS1 model (Riley et al., 2014) by incorporating a surface <u>complexation</u> model and reducing the complexity of <u>monomer</u>speciation (Fig. 1; Table 1). The modified BAMS1 includes a mechanistic representation of sorption processes. The reaction network consisted of (1) three groups of organic polymers (i.e., <u>cellulose</u>, <u>hemicellulose</u>, and lignin); (2) one representative simple organic monomer (or <u>dissolved organic carbon(DOC)</u>); (3) fungal and bacterial biomass; and (4) Dead Cell Wall Material (DCWM). However, we have not considered differences in chemical composition of DCWM between fungi and bacteria at this point, but we acknowledge the potential importance of such a distinction. Above and belowground litter and root exudates are partitioned between polymers and monomers and represent the carbon inputs to the system (Table 1). The decomposition of 1 mol of organic monomer yields Y (g wet-biomass mol⁻¹) microbial biomass. The fraction of carbon present in the substrate that is assimilated by microbes is defined as the assimilation-torespiration ratio (e) (Table 1). The fraction (1-e) of organic monomers is respired as CO₂ (Riley et al., 2014).



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Fig. 1. Modified BAMS1 reaction Network Diagram. Bacteria and <u>Monomers</u> stand for heterotrophic aerobic bacteria and dissolved monomeric <u>organic carbon</u>, respectively; DCWM stands for "Dead Cell Wall Material".

Table 1. Parameters defining the <u>decomposition of organic matter</u> pools by bacteria and fungi for maximum specific consumption rate (μ), assimilation-to-respiration ratio (e), and yield (Y) (modified from (<u>Riley et al., 2014</u>)).

	//							
		Bact	eriaª	Y (mg		Fur	ngiª	Y (mg
	μ (s -1)	e (-)	Y (g wet- biomass mol⁻¹)	carbon mg⁻¹)	µ (s -1)	e (-)	Y (g wet- biomass mol ⁻¹)	carbon mg⁻¹)
Organic Monomers	2.32e- 7	0.08	67.2	0.3	_	-	_	_
Lignin	_	-	_	_	4.20e- 9	0.01 8	84.0	0.5
Cellulose	_	-	_	_	5.46e- 9	0.00 1	84.0	0.5
Hemicellulose	_	-	_	_	1.68e- 8	0.03 9	84.0	0.5
DCWM	3.98e- 12	0.01 8	48.0		_	_	-	_

а

Mortality of bacteria and fungi contribute to DCWM. Biomass death rate of bacteria is 2.32e-7 s⁻¹, and biomass death rate of fungi is 7e-9 s⁻¹ (Riley et al., 2014).

BAMS1 includes fungi and aerobic bacteria functional groups. Although substantially underrepresenting functional diversity in soils, this approach is practical given limits on observations for more detailed parameterization while maintaining the governing <u>ecological functions</u> in the ecosystem (<u>Goldfarb et al., 2011</u>). These microbial groups are assigned affinities to decompose <u>plant litter</u> and SOM (<u>DeAngelis et al., 2013</u>, <u>Neely et al., 1991</u>, <u>Romaní et al., 2006</u>, <u>Thevenot et al., 2010</u>) and on death produce dissolved organic monomer and DCWM (<u>Frostegard and Bààth, 1996</u>). The model assumes that aerobic heterotrophic bacteria (BA) use organic monomers as a source of carbon for biomass growth and maintenance, leading to biomass yield and CO₂ production (<u>Fig. 1</u>). Fungi are assumed to excrete <u>exoenzymes</u>, in proportion to their biomass, that depolymerize polymers into monomers for assimilation (<u>Ahrens</u> et al., 2015, <u>Sulman et al., 2014</u>). <u>Transporter enzyme</u> levels are also assumed proportional to biomass, leading to Michaelis-Menten-Monod growth kinetics, as described in Section <u>2.4</u>.

A set of aqueous primary (e.g., monomers, cellulose) and secondary (aqueous and surface complexes and gaseous) species were defined that participated in the biogeochemical reaction network (<u>Reed, 1982</u>, <u>Steefel and Lasaga, 1994</u>, <u>Yeh and</u> <u>Tripathi, 1991</u>). The relationship between primary and secondary species is expressed as:

(1)Pi=∑j=1NcvijPji=1,...,NR

where *P* represents chemical species, N_c is the number of primary species, *j* is the primary species index, *i* is the secondary species index, N_R is the number of reactions (one per each secondary species), and v_i is the stoichiometric coefficient of the *j*th primary species in the *i*th reaction.

2.4. Biological reaction rates

The microbial decomposition rate of substrate *C*^{*i*} is calculated using <u>Michaelis-Menten</u> <u>kinetics</u> (e.g., <u>German et al., 2012</u>, <u>Riley et al., 2014</u>). We note that Michaelis-Menten kinetics may be an inappropriate representation in complex substrate-consumer networks (<u>Tang and Riley, 2013</u>); this potential complication is discussed in Section <u>4.4</u>. <u>Kinetic reactions</u> among primary species, which include aqueous and sorption reaction kinetics and <u>biodegradation</u>, were represented using a general rate law:

$(2)ri=\sum s=1M[ki,s\times \prod j=1Ni(yjvi,jCjvi,j)\times \prod k=1Nm(Ci,kKMi,k+Ci,k)]$

where r_i is the reaction rate of the *i*th reaction with the number of mechanisms or pathways ranging from s=1toM; k, γ_i , and C_i denote the rate constant, <u>activity coefficient</u>,

and concentration of species *j*, respectively; and $v_{i,j}N_{i,j}N_{m,j}C_{i,k}$ represent the stoichiometric coefficient, number of reacting species in the forward rate term (product terms), number of Monod factors (Michaelis-Menten terms), and the concentration (mol L⁻¹) of the *k*th Monod species, respectively. The first [ki,s], second [\Box j=1Ni(yjvi,jCjvi,j)], and third [\Box k=1Nm(Ci,kKMi,k+Ci,k)] terms on the right-hand side of Equation (2) are referred to as the rate constants, product terms, and Michaelis-Menten terms, respectively (e.g., <u>Curtis, 2003</u>).

2.5. Surface complexation modeling

Adsorption and desorption of SOC depend on characteristics of the organic molecules, soil mineral surfaces, and aqueous chemistry (Dudal and Gérard, 2004, Goldberg et al., <u>2007</u>). There are various approaches for representing sorption processes, including kinetic and equilibrium adsorption isotherms(e.g., linear, Langmuir, Freundlich), and thermodynamic surface complexation models (e.g., Diffuse Layer Model) (Hiemstra et al., 1989, Sahai and Sverjensky, 1997a, Sahai and Sverjensky, 1997b). Several studies have found that DOC sorption onto mineral surfaces can be described by a Langmuir isotherm (Gu et al., 1994, Kothawala et al., 2008, Mayes et al., 2012). However, sorption models are parameterized only for the specific conditions for which they were derived, hence, they cannot account for out-of-sample (i.e., outside the parameterization data set) effects of chemical conditions on adsorption, desorption, and competitive sorption processes (e.g., Goldberg et al., 2007). In contrast, Surface Complexation Models (SCMs) consider variable biogeochemical conditions within a thermodynamic framework, by using mass action laws analogous to aqueous phase reactions. Since SCMs can account for changes in chemical speciation, competitive adsorption, and other multi-solute interactive chemical effects (Davis and Kent, 1990, Davis and Coston, 1998), we implemented an SCM in BAMS1 for this study. The sorption of solutes on solid surfaces can be described as a chemical reaction between aqueous species and surface sites (surface complexation) (Dzombak, 1990). (3)A+S ↔ A-S

where *A* is the <u>sorbate</u> molecule, *S* is the <u>sorbent</u> site (surface); and A–Srepresents the sorbed complex. At equilibrium, the sorption reactions can be described by a mass action law equation:

(4)KEq=[A-S][A][S]

where KEq is the equilibrium sorption constant, [A–S] represents the concentration of surface complexes (mol kg^{-1}_{water}), [A] is the concentration of sorbate molecule (mol kg^{-1}_{water}), and [S] is the concentration of sorbent sites (mol_{sites} kg^{-1}_{water}). In this study, we

investigated organo-mineral interactions in terms of the <u>soil bulk</u> specific surface area (SBSSA, $m^2 g_{soil}^{-1}$) and equilibrium constraints. The concentration of sorbent sites is calculated from input SBSSA and site density (mol_{sites} m⁻²), from which [S] is then computed as a function of <u>porosity</u> and liquid saturation.

In the model, the total amount of available sorption sites (mol_{sites} g_{soi}⁻¹) limits the soil's capacity to store carbon and is calculated as the product of the SBSSA and site density. We assumed the soil with a bulk generic mineral surface with a site density of 7.68×10^{-7} mol_{sites} m⁻², which represents a typical site density for clays and iron hydroxides (Davis and Coston, 1998) multiplied by an assumed weight fraction (taken as the clay content for each site) of such sorbing minerals in the soil. We assumed weight fractions for CT-3.9ky and CT-29ky as 0.05 and 0.15, respectively; for other sites (CT-40ky to CT-240ky and Russian sites), we assumed a weight fraction of 0.3. Monomers and DCWM are two primary species in the model that exist in the aqueous phase and undergo sorption. However, dissolved monomers and DCWM sorb onto bulk soil minerals differently (Riley et al., 2014). Therefore, BAMS1 represents monomers and DCWM sorption separately, but the sorption characteristics of individual minerals were not modeled. Instead, sorption onto soil minerals was modeled using an SBSSA without considering <u>electrostatic</u> interactions. <u>Wattel-Koekkoek and Buurman</u> (2004) showed that the mean residence time of SOM is related to the <u>cation-exchange</u> <u>capacity</u> and specific minerals (e.g., kaolinite-bound organic matter), and electrostatic interactions may be important in modeling Δ^{14} C values of SOM. However, several studies suggest that it is not possible to quantify the charge correctly for assemblages of mixed minerals (e.g., soils and sediments) (Davis and Coston, 1998, Westall et al., <u>1995</u>). Moreover, because of the heterogeneity of soil organic carbon molecules, it is reasonable to model SOM using a non-electrostatic approach (Westall et al., 1995). Detailed inclusion of mineral double-layer properties and spatially heterogeneous surface areas were deemed beyond the scope of this study, and beyond the level of first-order effects expected in our simulations. The SCM in this study was applied to simulations held at constant temperature and pH, but it could be used to investigate impacts of temperature and pH based on data availability.

The decadic <u>logarithm</u> of the equilibrium sorption coefficients (log(KEq)) of reactions of different functional groups of organic carbon with iron minerals has a wide range (8.25–13; (Lenhart and Honeyman, 1999, Murphy et al., 1999)). We assumed relatively smaller (9.08) and larger (11.25) log(KEq) values for monomers and DCWM, respectively, based on our original work in BAMS1 (<u>Riley et al., 2014</u>). We show below

with a sensitivity analysis (Sections 3.3 Sensitivity analysis for total SOM, 4.2 Sensitivity analyses) that log(KEq) does not substantially affect equilibrium SOM values. Several studies have reported that SBSSA increases with soil formation and age (up to a point) and decreases with increasing soil depth (Hodson et al., 1998, White et al., 1996, White et al., 2009). However, variation of SBSSA with depth and time is complex. Consistent with the idea that SBSSA increases as soils age, Merritts et al. (1991) reported an increase in accumulated mass of several forms of pedogenic iron and clay across the Chronosequence Terrace sites. Although texture and mineralogical composition have been used to develop various SBSSA pedotransfer functions (Whitfield and Reid, 2013, Feller et al., 1992, Mayer, 1994, White et al., 1996), it is difficult to accurately extrapolate SBSSA to other soils (Whitfield and Reid, 2013). Therefore, for simplicity, we consider SBSSA to be a generic bulk mineral surface area and calibrate its value at each of the ten sites, as described below. We then analyze relationships between SBSSA and observed clay content across the sites. Vanwalleghem et al. (2013) showed that the rate of soil formation may be described using an exponentially decreasing function with depth, since many physical factors that control weathering, such as temperature and soil moisture, are known to have exponentially decreasing depth profiles (Amenu et al., 2005). On this basis, we assumed that each CT and Russian site exhibits an exponentially decreasing SBSSA with depth (a), i.e., a=a0e-z/zm; where z is depth (m), z_m is a constant (m), and a_0 is the near-surface SBSSA. Because we did not have direct measurements of either a_0 or z_m , we estimated these parameters using SOM profiles observed at each site. During estimation, for each value of a_0 , we allowed z_m to range from 0 to 6 m (0.0, 0.25, 0.5, 0.75, 1, 1.25, 2, 3, 4, 5, 6). z_m values of 0.0 and 6 m imply no gradient and the steepest gradient, respectively. The best-fit values for a_0 and z_m were chosen for each site based on R² values (or coefficient of determination) and root mean squared error (RMSE) between simulated and observed SOM content. R² values are computed as

follows:

(5)R2=1-∑i=1n(Oi-Mi)2∑i=1n(Oi-O⁻)2

where O are the n observed SOM values; O^- is the mean of the observed SOM values; and M are the n estimated SOM values by the model. R^2 value 1 denotes the perfect match between modeled and observed values.

RMSE values are computed as follows:

(6)RMSE=∑i=1n(Oi−Mi)2n

RMSE values range from 0 to ∞ ; RMSE = 0 denotes the perfect match between modeled and observed values.

We selected all values of a_0 and z_m for each site when R² and RMSE values were in the 80th (top 20%) and 50th (top 50%) percentiles. High R² values ensure that predicted SOM values follow the qualitative SOM profile, and small RMSE values ensure that predicted SOM values are close to observations.

Estimated SBSSA (a_o) values ranged from 6.8 to 17.4 m² g_{sol}⁻¹ for the CT sites. For the Russian sites, the estimated SBSSA was 39.1 m² g_{sol}⁻¹. The CT values were consistent with SBSSA values reported for soils elsewhere in California under similar climatic conditions (White et al., 1996). Subsequently, we used these values for a_o and z_m in BAMS1 to predict Δ^{14} C values for each site that were estimated using SOM values.

2.6. Reactive transport solver

We integrated the modified reaction network (BAMS1) and surface complexation model in the one-dimensional version of the <u>multi-phase flow</u>and <u>reactive transport</u> simulator TOUGHREACT V2 (Xu et al., 2011). TOUGHREACT includes various processes to simulate flow and reactive transport in variably saturated, non-isothermal <u>porous media</u>, including <u>advection</u>; diffusion; representation of multiple competing microbial populations (Xu, 2008); adsorption and desorption; the soil <u>energy budget</u>; and equilibrium and kinetic chemical and biochemical reactions. The flow and transport are solved using integral finite difference spatial discretization. In addition, a sequential iteration approach as described in <u>Yeh and Tripathi (1991)</u> is used to solve <u>chemical</u> <u>transport</u> simulation. After solution of the <u>flow equations</u>, the fluid velocities and phase saturations are used for chemical transport simulation (Xu et al., 2011). The Darcy-Richards equation (<u>Pruess</u>, 1999) was used to model <u>water flow</u>, and the <u>van</u> <u>Genuchten (1980)</u> model was used to relate <u>matric potential</u> and <u>hydraulic</u> <u>conductivity</u> with soil water saturation. The overall mass balance equation was then solved in a one-dimensional (1D) vertical domain:

(7) $\partial Ci\partial t = -\partial \partial z [D\partial Ci\partial z + vCi] + \sum k \partial Ci\partial t | k$

where z (m) is the vertical dimension, *t* (s) is the time, *D* (m² s⁻¹) is the effective aqueous or gaseous <u>diffusivity</u>, and *v* (m s⁻¹) is the bulk aqueous or gaseous velocity. The last term denotes the sum of sinks or sources resulting from *k* reactive mechanisms. 2.7. Representing Δ^{14} C values of SOM

SOM $\Delta^{14}C$ data are useful benchmarks of soil biogeochemical models, because $\Delta^{14}C$ values are a function of the residence time of soil carbon since

its <u>photosynthetic</u> assimilation from the atmosphere (<u>Torn et al., 2009</u>, <u>Trumbore, 2000</u>). Based on Δ^{14} C observations, the residence time of SOM typically increases with depth (<u>Jenkinson and Coleman, 2008</u>, <u>Torn et al., 1997a</u>, <u>Torn et al., 2002</u>, <u>Trumbore</u>,

<u>2000, Trumbore, 2009</u>). We predicted SOM Δ¹⁴C values by duplicating the reaction network into one set of reactions in terms of ¹²C species, and a separate but identical set of reactions in terms of 14C species neglecting combinatorial effects in molecules with multiple C atoms (see GEBIK and GEBIF equations in Maggi and Riley (2010)). We then applied a first order decay term λ (s⁻¹) for ¹⁴C radioactive decay($\lambda = 1/8267$ y) to all ¹⁴C species in aqueous and sorbed phases. Predicted and observed Δ^{14} C values are reported in permil (‰) denoting parts per thousand (Stuiver and Polach, 1999). To account for the ¹⁴C bomb pulse, we imposed northern hemispheric Δ^{14} C values for 1950–1976 and 1977–2003 as reported by Levin and Hesshaimer (2000) and Levin and Kromer (2004), respectively, for the Northern California and Russian observations that were made in 2004 and 1997, respectively; the archived Russian soil was collected between 1895 and 1903, before ¹⁴C release from nuclear weapons tests. The observed Δ^{14} C values of CT soils decreased from about 0‰ to -710‰. Δ^{14} C values are often more depleted in the older soils (CT-118ky, CT-124ky-1, CT-124ky-2, and CT-240ky) than the younger soils at the same depth; however, there are slightly higher Δ^{14} C values in CT-240ky than in CT-118ky (Masiello et al., 2004). The Russian soils have Δ^{14} C data from the pre-bomb era (<u>Torn et al., 2002</u>). These soils were sampled 100 years apart; the Δ^{14} C values decreased from about 0% to -700%, indicating very old carbon at depth.

2.8. Climate forcing, boundary conditions, and initial conditions

For terraces in California, mean precipitation and <u>evapotranspiration</u> were imposed as 1000 mm yr⁻¹ and 300 mm yr⁻¹, respectively (<u>Masiello et al.</u>,

<u>2004</u>; <u>http://www.ntsg.umt.edu/project/mod16</u>). For Russian sites, mean precipitation and evapotranspiration were imposed as 520 mm yr⁻¹ and 460 mm yr⁻¹, respectively (<u>Frolov, 2014</u>, <u>Torn et al., 2002</u>).

Root profiles for the chronosequence terrace and

Central <u>Chernozem</u> Region <u>grasslands</u> were selected from the Global Root Distribution Profile database (<u>Schenk and Jackson, 2002</u>, <u>Schenk and Jackson, 2005</u>) (available in <u>ESA's</u>Electronic Data Archive: Ecological Archives M072-004-

A1: http://www.esapubs.org/archive/mono/M072/004/appendix-A.htm).

We imposed carbon inputs from leaf, wood, and root litter based on simulations using CLM4.0 (Lawrence et al., 2011) and partitioned into the soil using separate exponentially depth-distributed profiles. Baseline simulations with no soil carbon as an initial condition and constant <u>climate forcing</u> were run for 5000 years, except for the youngest site, which was run for 3900 years, corresponding to its estimated age. We

verified that 5000 years of simulations produced ¹²C and ¹⁴C values for all components that were within 0.1% of those after 10,000 years of simulations. Therefore, 5000 years of simulations were effectively at steady state; the 5000-year simulations were subsequently performed for computational tractability.

2.9. Model analyses

We simulated <u>carbon cycling</u> of the Chronosequence Terrace sites with unchanged vegetation, temperature, and precipitation conditions, but varying the SBSSA (<u>Table 2</u>). We performed analogous simulations for soils of the Central Chernozem Region of Russia. We performed sensitivity analyses of SOM and Δ^{14} C profiles to changes in SBSSA, sorption coefficients, and vertical carbon input profiles using the CT-29ky site. Model predictions were compared to observations using R² and RMSE values as described by Equations (<u>5</u>), (<u>6</u>).

Sites	Soil bulk specific surface area, a₀ (m² g₅₀i⁻¹)	Exponential Decay Constant, z _m (Depth Profile)
CT-3.9ky	7.6	0
CT-29ky	11.6	0.5
CT-40ky	6.8–9.5	0.5
CT-118ky	13.5–17.4	0.75
CT-124ky-1	6.8-17.4	0.75
CT-124ky-2	13.5–17.4	0.75
CT-240ky	13.5–14.8	0.75
Russian Steppe: Archive, Pit1. and Pit2	39.1	2.25

Table 2. Estimated <u>soil bulk</u> specific surface area ($m^2 g_{soil}^{-1}$) for the two studied sites.

3. Results

3.1. SOM predictions for the chronosequence terrace and chernozem sites

We performed simulations to estimate total SOM stocks for

the <u>Chronosequence</u> Terrace sites, where mineral composition varied significantly with site age along the chronosequence (<u>Merritts et al., 1991</u>). To determine whether observed patterns in carbon stock among the sites could be caused by changes in <u>mineralogy</u>, we imposed the SBSSA given in <u>Table 2</u>. For site CT-3.9ky, there was only one combination that met the R² and RMSE criteria. For site CT-29ky, there were three combinations of the SBSSA (m² g_{sol}⁻¹) [(a) 9.3, (b) 11.6, and (c) 13.9] and the corresponding exponential decreasing constants [(a) 0, (b) 0.5, and (c) 0.5], but two of

the best three RMSE values (a and c) were more than 30% and 50% greater than the smallest RMSE value. Additionally, three combinations had little difference in R² values [(a) 0.768 (b) 0.768 (c) 0.761 for site CT-29ky]. For all other sites, we chose the range of a_0 and z_m based on the smallest RMSE values. We also report the range of a_0 and z_m that had RMSE values within 10% of the smallest RMSE value for each site (<u>Table 2</u>). We found that a_0 and z_m have increasing trends along the chronosequice. The increasing trend of a_0 along the chronosequence is consistent with the idea that older sites experience more weathering and therefore have higher SBSSA. The larger z_m for older sites is consistent with weathering occurring over a longer time extent. The model predictions with the best-fit parameters broadly matched the observed total and vertical SOM profiles in these sites (Fig. 2). Total SOM (between 0 and 2 m depth) was accurately predicted across these seven Chronosequence Terrace sites ($R^2 = 0.81$). Good matches with observed SOM profiles were predicted in the three youngest sites (CT-3.9ky, $R^2 = 0.75$; CT-29ky, $R^2 = 0.76$, and CT-40ky, $R^2 = 0.94$). Although we report R² values for the comparison between predicted and observed SOM values across depth, we note that this statistic should be interpreted carefully, since there are strong correlations across depth of SOM stocks. For example, a model error near the surface can propagate to depth, where it may result in larger error, but that error will not be indicative of model process representation errors at that depth. The least weathered site (CT-3.9ky) had the least SOM in both observations and predictions. For the older sites (CT-118ky, CT-124ky-1, CT-124ky-2, and CT-240ky), the model relatively accurately predicted SOM above 0.50–1.0 m depth and over-predicted SOM below 1.0 m depth. We discuss these discrepancies in Section 4.1.



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Fig. 2. Predicted and observed SOM content at the seven <u>Chronosequence</u> Terrace sites south of Eureka, California.

We performed analogous simulations to estimate total SOM stocks at the three <u>Chernozem</u> sites (one archive and two modern soils) and compared the results to observations by <u>Torn et al. (2002)</u>. Simulation results for all three sites had good overall correlations with observations ($R^2 = 0.97$, 0.96, and 0.90 for the Archive, Pit1, and Pit2 soils, respectively; Fig. 3). However, the model slightly under-predicted SOM content between 0.2 and 0.5 m and over-predicted SOM content between 0.7 and 1.2 m for the Archive soils. The variation in the field observations gives an indication of the spatial

heterogeneity and sampling and analytical precision, since the three observational datasets are nearby in the same system.



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Fig. 3. Predicted and observed SOM content for 3 sites in the Central <u>Chernozem</u>Region of Russia.

Although we do not have observations of individual SOM components to compare to our predictions, the predicted <u>vertical profiles</u> highlight the processes that control bulk SOM content. For example, at the CT-29ky site (Fig. 2B), input polymers (lignin, <u>cellulose</u>, hemicellulose) were predicted to be in the top meter of soil at steady state, due to imposed vertical carbon input profiles and model assumption that polymers are not transported by <u>advection</u> (Fig. 4D, 4E, 4G). Lignin had the largest concentration among input polymers.



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Fig. 4. Predictions of modeled carbon components after 5000 years using the simulation from the CT-29ky site.

Because <u>monomers</u> can dissolve into the aqueous phase and move vertically by advection and diffusion, sorbed monomers had a deeper concentration profile (Fig. 4H and 4I) than did the input polymers (Fig. 4D, 4E, 4G). Predicted aerobic bacteria (Fig. 4A) comprised more than 90% of the total microbial biomass, while the fungal biomass (Fig. 4B) was largest at the surface and declined rapidly with depth. 3.2. Δ^{14} C predictions of total SOM

Simulations at the Chronosequence Terrace sites (Fig. 5) resulted in Δ^{14} C values with profiles qualitatively consistent with observations. The model under-predicted Δ^{14} C values in the top 0.2 m of the three younger sites (CT-3.9ky, CT-29ky, CT-40ky), but predictions were close (10‰–40‰) to observations between 0.20 m and 1 m depth in the CT-29ky and CT-40ky sites, respectively. Observed Δ^{14} C values in the CT-3.9ky site

did not follow a clear pattern with depth, and the model did not predict this high variability. In the older sites (CT-124ky-1, CT-124ky-2, CT-118ky, CT-240ky), predictions deeper than 0.2 m were consistently less ¹⁴C depleted or younger than the observations. We performed sensitivity studies (perturbing SBSSA, <u>sorption</u>coefficients, and carbon input profiles) to identify mechanisms potentially responsible for biases between the model predictions and observations.



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Fig. 5. Δ_{14} C values of total SOM at the end of 5000-year simulations (except for CT-3.9ky that is after 3900-year) at the seven <u>Chronosequence</u> Terrace sites south of Eureka, California.

At the Chernozem site, the model predictions qualitatively matched the Δ^{14} C value profile down to 1 m in the Archive and Pit2 soils, and accurately matched the two available data points in the <u>shallow soil</u> (at 0.2 m depth) in the Pit1 soil (Fig. 6). In the Archive and Pit1 soils, the predicted Δ^{14} C values were less depleted than observed below 1 m depth, while the deeper Pit2 soil Δ^{14} C observations were more accurately captured. Also, the model (1) over-estimated ¹⁴C depletion in the top 0.1 m in the Archive and Pit1/Pit2 observations; (2) predicted a larger incorporation of the bomb ¹⁴C signal between about 0.2 and 1 m than observed; and (3) about correctly captured the observed differences in the top 0.1 m Δ^{14} C values between the Archive and post-bomb Pits.



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Fig. 6. Δ_{14} C values of total SOM at the end of 5000-year simulations at three sites in the Central <u>Chernozem</u> Region of Russia.

Although observations of Δ^{14} C values for individual SOM components were not available for comparison with our predictions, vertical Δ^{14} C profiles for individual SOM

components may indicate the processes predicted to control bulk SOM content and Δ^{14} C values. For example, using simulations from the CT-29ky site (Fig. 5B), the microbial biomass (aerobic bacteria) and dissolved monomers Δ^{14} C values (Fig. 7) broadly follow the bulk SOM Δ^{14} C values (Fig. 5B), indicating rapid recycling of SOM between sorbed and dissolved phases, and the relatively fast turnover by aerobic bacteria. The more depleted Δ^{14} C values of fungal biomass below the carbon input zone (below about 0.7 m) occurred because of fungal decomposition of DCWM. The predicted Δ^{14} C values for DCWM, which was used in the model to represent microbial cell wall polymers resulting from microbial death, were relatively depleted even in the top 0.5 m of soil. The fraction of sorbed SOM that is DCWM increased monotonically with depth below 1 m (not shown), reaching 15% by 2 m depth from 8% at 1 m depth.



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Fig. 7. Predictions of Δ_{4} C values of modeled carbon components at the end of a 5000year simulation for the CT-29ky site.

3.3. Sensitivity analysis for total SOM

To better characterize the impact of organo-mineral interactions on SOM dynamics, we analyzed the sensitivity of SOM content to SBSSA, sorption coefficients, and carbon

input profiles. For these analyses, CT-29ky site was used as the baseline scenario (Fig. 2B).

We varied SBSSA by $\pm 10\%$, $\pm 20\%$, and $\pm 50\%$ from the baseline value for this site throughout the <u>soil column</u>. Resulting variations in SOM stocks were approximately proportional to these changes (Fig. 8A). Varying monomer and DCWM sorption coefficients (KEq) throughout the soil column by $\pm 10\%$, $\pm 20\%$, and $\pm 50\%$ from the baseline value had little impact on SOM stocks (Fig. 8B and 8C).



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Fig. 8. Sensitivity of SOM profiles due to changes in (A) <u>soil bulk</u> specific surface area, (B) <u>monomer sorption</u> coefficients, (C) DCWM sorption coefficients, and (D) carbon input profiles. Deep carbon input profile3 is the deepest plant-input profile, and shallow carbon input profile1 is the shallowest plant-input profile.

In order to test the sensitivity of the <u>depth distribution</u> of plant inputs on verticallyresolved SOM stocks, we varied z_m by ±10%, ±20%, and ±50% from the baseline values but held the total rate of carbon inputs unchanged. These carbon input profiles changes had little impact on SOM stocks in the soil column (Fig. 8D), primarily because more than 90% of available sorption sites were occupied (Fig. 9) by sorbed monomers and DCWM (Fig. 4, Fig. 9). The differences in carbon saturation associated with the different carbon input profiles were less than 5% in the top meter. Below 1 m, the differences in carbon saturation were within ±15% of the baseline simulation. However, the SOM content was less than 0.0001 kg_c kg⁻¹_{soil} below 1 m depth.



Percent saturation of all available sorption sites

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Fig. 9. Percent saturation of available <u>sorption</u> sites for various carbon input profiles. 3.4. Sensitivity analysis for SOM Δ^{14} C values

We conducted a <u>sensitivity analysis</u> of Δ^{14} C values with the same variations described for SOM stocks. Altering SBSSA and sorption coefficients throughout the column by ±10%, ±20%, and ±50% from the baseline values (CT-29ky) had very small effects on Δ^{14} C (Fig. 10A, 10B, and 10C). In contrast, changes in the vertical carbon input profiles had large impacts on Δ^{14} C (Fig. 10D), with the shallower carbon input profiles producing more depleted Δ^{14} C values below about 0.2 m depth. Enhanced depletion occurred because SOC in deeper soil was not replenished by new carbon as quickly in the case of shallower carbon input profiles because of retardation caused by transport, microbial consumption and release through mortality, and sorption. Moreover, even though bulk SOM stock was not sensitive to carbon input profile due to saturation of <u>soil</u> <u>bulk</u> specific surface area, the deeper carbon input profiles introduced new carbon deeper into the soil, resulting in less depleted Δ^{14} C values in the deeper carbon input profile scenario relative to the shallow carbon input scenario.



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Fig. 10. Sensitivity of Δ_{14} C values of total SOM to changes in the (A) <u>soil bulk</u> specific surface area, (B) <u>monomer sorption</u> coefficients, (C) DCWM sorption coefficients, and (D) carbon input profiles.

4. Discussion

4.1. Predicted SOM stocks and $\Delta^{14}C$ values

The modified BAMS1 model predicted SOM stocks for the seven <u>Chronosequence</u> Terrace sites in broad agreement with observed profiles (R²between 0.75 and 0.94) and for the three Russian <u>grassland</u> profiles in excellent agreement (R² between 0.90 and 0.97) with observed profiles. We note that it is possible to relatively accurately calibrate different types of SOM decomposition models against integrated near-surface bulk SOM observations (He et al., 2014, Luo et al., 2015). However, it is difficult to infer the relative importance of various decomposition mechanisms by using bulk SOM observations alone.

For comparison, the recent study by <u>Ahrens et al. (2015)</u> used more extensive measurements (total and Δ^{IA} C values of SOC and mineral-associated carbon and microbial biomass) at a single site to calibrate 11 parameters in their vertically-resolved SOM model. <u>Sulman et al. (2014)</u> used the responses of microbial biomass and decomposition of <u>monomers</u> to amendments of a simple substrate to calibrate microbial and decomposition parameters of their model. Our approach of calibrating only two parameters to predict total SOM, and then using SOM Δ^{IA} C values as independent tests for the calibration, is sufficiently robust to justify using the model to evaluate hypotheses relating mineral controls on SOC stabilization to SOM storage. However, we acknowledge that more measurements of system properties could improve model calibration, which we will pursue in future work. Specific examples of abiotic and biotic properties that would be valuable within a model structure like BAMS1 include for example, mineral adsorbed SOM, chemical properties of dissolved monomers, microbial biomass and activity, and <u>mineral surface</u>properties (Conant et al., 2011).

The secondary constraint of accurately representing vertically-resolved SOM Δ^{14} C values implies that BAMS1 is more likely accounting for the mechanisms important in establishing observed vertical profiles of soil organic matter. For instance, predictions of Δ^{14} C values of modeled carbon components (Fig. 7) showed that the fungal Δ^{14} C values were relatively enriched and close to atmospheric values, because fungi degrade input polymers (cellulose, hemicellulose, lignin), which was consistent with the relatively fast turnover of these compounds. In addition, the predicted Δ^{14} C values for DCWM were comparatively depleted even near the surface. This depletion occurred because we assumed that the cell wall material is relatively difficult to decompose when sorbed to mineral surfaces, and therefore accumulated over long periods. Altogether, our model results indicate that (1) relatively fast microbial SOM cycling (~o(0.1 to 1 y)), (2) vertical transport, (3) mineral stabilization, and (4) relatively slow decomposition of DCWM led

to predictions that were consistent with observed vertical profiles of SOM bulk content and Δ^{14} C values.

These mechanisms are consistent with conclusions from recent studies that emphasize the importance of interactions among substrate availability, microorganisms, mineral surfaces, and slow <u>depolymerization</u> of substrates like DCWM in determining SOM profiles (Dungait et al., 2012, Kögel-Knabner, 2002, Schmidt et al., 2011). The mechanisms represented in BAMS1 did not impose intrinsic long turnover times and are in contrast to the widely applied pseudo-first-order SOM decomposition models, which impose a range of intrinsic decomposition rates on modeled SOM pools (Braakhekke et al., 2011, Braakhekke et al., 2014, Koven et al., 2013, Parton et al., 1987). For example, RothPC-1 (Jenkinson and Coleman, 2008) and CLM4.5 (Koven et al., 2013) impose depth-dependent SOM decomposition rates, mimicking the reduced microbial activity at depth that is dynamically prognosed in BAMS1.

Although pseudo-first-order approaches (<u>Braakhekke et al., 2011</u>, <u>Jenkinson and</u> <u>Coleman, 2008</u>, <u>Koven et al., 2013</u>, <u>Tang et al., 2013</u>) have been calibrated to represent vertical gradients of SOM content, they will likely predict different perturbation responses (compared to BAMS1) to changes in temperature, moisture, pH, and carbon inputs expected under <u>climate change(Riley et al., 2014</u>). Along the same line, <u>Luo et al.</u> (2015) highlighted that pseudo-first-order models of SOM lead to large uncertainty in predicting <u>soil carbon</u> and perturbation responses, particularly with future expected warming and drying conditions. They further emphasized the need to represent the <u>microbial community</u> and its carbon use efficiency as a function of environmental changes for accurate predictions of SOM stocks. For example, an earlier perturbation analysis with BAMS1 (Riley et al., 2014</u>) showed that interactions between microbial activity and vertical transport can lead to rapid changes in decomposition rates (e.g., from microbial priming) that would not emerge in a pseudo-first-order model that does not explicitly represent microbial activity.

Schimel et al. (2011) observed that pulsed drying and wetting events led to respiration from relatively old SOM, indicating that this old SOM may not be inert because of its association with mineral surfaces. Xu et al. (2004) reported the occurrence of short-lived, but high-rate respiratory pulses after rain events in a California grassland site. Lawrence et al. (2009) found that accounting for explicit exoenzyme and microbial decomposition controls improved representation of pulsed rewetting responses, although the simpler pseudo-first-order models they applied were more accurate under steady-state conditions. Our, and the recent results described above, argue that explicitly

representing microbial activity and organo-mineral interactions in land models may improve <u>carbon cycle</u> predictions under dynamic <u>climate forcing</u> conditions.

4.2. Sensitivity analyses

Soil <u>surface properties</u> change during soil development due to the formation of clays and other <u>sorbent</u> minerals (e.g., amorphous or poorly crystalline aluminum, iron, and <u>manganese</u> hydroxides), which can store more SOM than ordered minerals associated with younger soils (<u>Masiello et al., 2004, Mayer, 1994, Merritts et al.,</u> <u>1991, Six et al., 2000</u>). There is evidence that SBSSA increases with age up to a certain point, and then decreases (e.g., in Oxisols) (<u>Hodson et al., 1998</u>). Moreover, <u>soil</u> <u>weathering</u> affects SBSSA and reactivity (<u>Maher et al., 2009</u>, <u>White et al., 1996</u>, <u>White</u> <u>et al., 2009</u>). To examine these interactions within our modeling framework, we performed sensitivity analyses around our baseline values of <u>soil bulk</u> specific surface area and <u>sorption</u> coefficients. SOM content was affected throughout the column approximately proportionally to the imposed changes in SBSSA (<u>Fig. 8</u>A). Thus, a fractional change in SBSSA led to a comparable fractional change in SOM storage, at equilibrium. These results are consistent with previous studies that noted a positive correlation between mineral content and <u>organic carbon</u> in soils (<u>Kaiser and</u> <u>Guggenberger, 2003, Mayer, 1994</u>).

In contrast, the imposed changes in monomer and DCWM sorption coefficients did not affect SOM stocks (Fig. 8B). If sorption sites are mostly saturated (as they are at equilibrium in our simulations), increasing the sorption coefficients does not change the amount of sorbed monomers or DCWM. In particular, changes in sorption coefficients at or near saturation of mineral surface sites had little impact on SOM content. Several studies have suggested that weathering-related controls on SOM content are likely related to the surface chemistry of primary and secondary minerals (Chorover et al., 2004, Guggenberger and Kaiser, 2003). In our sensitivity analyses, we found that SOM Δ¹⁴C values were largely unaffected by imposed changes in SBSSA and sorption coefficients (Fig. 10A, 10B, 10C). This result is consistent with the model structure we imposed in BAMS1, where an increasing sorption capacity resulted in an increased adsorbed carbon pool in equilibrium with an aqueous concentration that is similar to that predicted with the original sorption capacity. The aqueous carbon concentration is effectively set by microbial activity, which is relatively constant as a function of sorption capacity. The microbial activity sets the effective decomposition rate of carbon inputs, thus leading to similar Δ¹⁴C values and inferred turnover times. A recent study conducted by <u>Ahrens et al. (2015)</u> also concluded that microbial interactions and aqueous carbon

concentration control Δ^{I4} C at depth and that sorption acts as a buffer for aqueous carbon concentration in the <u>soil column</u>. Sorption processes suppress microbial activity by physically protecting substrate and maintaining equilibrium microbial activity by gradually releasing carbon in the soil column as suggested in <u>Riley et al. (2014)</u>. The BAMS1 receptor-controlled decomposition rate is in contrast to the donor-controlled decomposition mechanism imposed in multi-pool, pseudo-first-order carbon decomposition models. In those models, larger SOM content could be obtained by reducing the turnover times (e.g., of the 'slow' pool), and arguing that the turnover times were functions of soil <u>mineral properties</u>. The result of such a change would be longer residence times. Thus, BAMS1 prognoses different Δ^{I4} C patterns in response to differences in soil <u>mineralogy</u> relative to models that impose longer turnover times, creating and highlighting an observational metric to inform which type of model structure is more appropriate.

Leifeld et al. (2009) reported that the residence time inferred from Δ^{14} C measurements of mineral-associated organic matter was positively related to soil mineral surface area. In our simulations, SBSSA is strongly positively correlated with Δ^{14} C values across depth at the CT sites (except for the youngest site; CT-3.9ky), and positively correlated at a lower level at CT-3.9ky, CT-29ky, CT-118ky, and CT-124ky for Δ^{14} C integrated over depth. We also found low sensitivity of predicted Δ^{14} C values to SBSSA in our <u>sensitivity</u> <u>analysis</u> of the CT-29ky site, as described above. We therefore conclude that the observed and modeled correlation between SBSSA and Δ^{14} C values across depth is not causative, but rather emerges from the combination of microbial activity, vertical transport of monomers, and stabilization on mineral surfaces.

Changes in the vertical carbon input profiles had a negligible impact on the bulk SOM profiles, which is consistent with studies that have shown that mineral-associated SOM is a function of the carbon saturation deficit (<u>Castellano et al., 2015</u>, <u>Cotrufo et al., 2013</u>). In addition, <u>Holub et al. (2005)</u>found little impact of belowground organic carbon input profile on SOM stocks in two different <u>forest soils</u> after 4 and 11 years. Clearly, there is an upper limit to the equilibrium mineral soil carbon level associated with the soil carbon saturation capacity. <u>Stewart et al. (2009</u>) argued that carbon stabilization capacity can be quantified for different SOM pools. A potential reason for a small effect of carbon input profiles on SOM content is that sorption sites were saturated at steady state (Fig. 9). Several observational studies have also demonstrated that mineral-associated fractions exhibit carbon-saturation behavior (<u>Castellano et al., 2015</u>, <u>Hassink and Whitmore, 1997</u>). Most contemporary SOM models assume a strong linkage

between carbon input profiles (root, litter, and wood litter) and SOM accumulation, but soils have finite capacities to accumulate carbon (<u>Gottschalk et al., 2012</u>). Soils may become saturated and reach equilibrium because of different combinations of <u>soil</u> texture, mineralogy, and climatic regimes (<u>Chung et al., 2010</u>, <u>Six et al., 2002</u>, <u>Stewart et al., 2009</u>). Therefore, it is desirable to represent carbon-saturation behavior of soils in SOM models.

In contrast, vertical carbon input profiles had a significant impact on predicted SOM Δ^{14} C values (Fig. 10D). Sorbed monomers and DCWM dominated bulk SOM content below 0.7 m. SOM Δ^{14} C values reflect the carbon residence time, which varies with carbon input profile; for example, deeper carbon input profiles will add younger carbon to deeper soil organic matter. Similarly, some studies of soil carbon storage and turnover demonstrated that, at a particular soil depth, carbon residence time is associated with input rates at that depth (Trumbore, 2009, Wang et al., 2013). Since we perturbed vertical carbon input profiles while keeping the total carbon input equal, we conclude that variations in carbon input depth profiles can alter SOM residence time because of microbial and abiotic controls on SOM cycling.

4.3. SOM modeling approaches

Currently, there are two broad classes of SOM models capable of integration with climate-scale land models: (1) pseudo-first order multi-pool models and (2) models representing explicit microbial and physical protection processes. Each of these model structures can be calibrated to match current SOM observations, but they will likely predict very different responses to climate perturbations. Clearly, there are trade-offs between model structural complexity, parameterization, and computational tractability in representing carbon dynamics at larger scales. Although current analyses are insufficient to argue that the increased complexity of the more mechanistic models will increase predictive capability under dynamic climate forcing, we argue that further development is warranted because these models integrate known relevant mechanisms (e.g., microbial dynamics, SOM chemical composition, and mineral-surface versus aqueous SOM partitioning) and are therefore expected to provide an explanation to observations of these mechanisms. One possible approach to reduce the additional computational costs associated with more complex SOM representations is the development of reduced order or surrogate models that could capture much of the important dynamics at substantially lower computational costs than the full model (e.g., Pau et al. (2016)).

4.4. Caveats and future model development

We modeled chemical components that were not observed individually to test our model. Although, it is possible to achieve total SOM content with various combinations of parameters, we used various metrics to test the validity of our assumptions, e.g., lignin content, fungi-to-aerobic bacteria biomass ratios, and Δ^{14} C values of SOM. We have not tested model structural uncertainty because of unavailability of observations of each modeled component; however, our analysis provides insights into processes that are not possible with a pseudo-first order SOM modeling framework. As with all SOM modeling analyses, the complexity of SOM dynamics requires substantial simplification of process representation and interconnections between processes. Among the many potential resulting problems, an interesting complication in representing microbial SOM dynamics arises because of recent work suggesting that Michaelis-Menten kinetics (which we applied here and are commonly applied in most microbe-explicit SOM models) may be inappropriate to represent the complex consumer, substrate, and abiotic interactions that occur during SOM decomposition (Tang and Riley, 2015, Tang et al., 2013) as well as in combinatorial isotope speciation and fractionation (Maggi and Riley, 2010). We did not apply these concepts in the current analysis but acknowledge that uncertainties in these interactions may cause uncertainty in our model results.

We also did not account here for temporal frequency of climate forcing (<u>Arora et al.,</u> <u>2016</u>, <u>Grant et al., 2006</u>, <u>Lawrence et al., 2009</u>, <u>Schimel et al., 2011</u>), feedbacks with vegetation (<u>Turner et al., 2004</u>), and nutrient constraints on decomposition (<u>Parton</u> <u>et al., 1987</u>), despite their potential importance on emergent SOM profiles; future work will address these shortcomings.

The model introduced here represents a one-dimensional soil column, but <u>terrestrial</u> <u>ecosystems</u> have spatial horizontal heterogeneity that may affects SOM dynamics in non-uniform <u>soil formations</u> (Arora et al., 2011, Frei et al., 2012, Li et al., 2008, Molins <u>et al., 2012</u>). Moreover, <u>pore</u> and molecular-scale processes and their effects on carbon exchanges between land and the atmosphere are expected to vary because of the <u>temporal variability</u>associated with many factors, including climate (Koven et al., 2013), <u>lithology</u>(Jobbágy and Jackson, 2000), seasonal dynamics (<u>Grant et al., 2006</u>), interfacial chemistry (<u>Torn et al., 1997b</u>), edaphic properties (<u>Bird et al., 2002</u>), soil moisture (<u>Riley and Shen, 2014</u>, <u>Sivapalan, 2005</u>), temperature (<u>Davidson and</u> <u>Janssens, 2006</u>), and vegetation (<u>Turner et al., 2004</u>). To bridge gaps between molecular–, site–, regional-, and climate-scale modeling, interdisciplinary and multiscale approaches will be required to account for the complex feedbacks that occur between <u>hydrology</u>, plant processes, microbial degradation, and mineralogical controls on carbon stocks and residence time.

5. Conclusions

We modified BAMS1, by incorporating a surface <u>complexation</u> model and reducing the complexity of monomer speciation, and integrated it with a vertically-resolved, multiphase reactive transport solver (TOUGHREACT) to investigate the impact of soil mineralogy on SOM dynamics. By accounting for microbial activity, sorption, and vertical transport, the model reasonably predicted SOM stocks and Δ^{14} C values at seven sites along a soil <u>chronosequence</u> in Northern California and three <u>grassland</u> sites in Russia. Model predictions were consistent with expectations of fungi:aerobic bacteria ratios. Our sensitivity analyses indicated that SBSSA was the dominant control over organic carbon stocks at these sites and that sorption coefficients did not strongly affect total SOM stocks. We found that the observed and modeled correlation between SBSSA and Δ^{14} C values across depth at each site is not causative, but rather emerges from the combination of microbial activity, vertical transport of monomers (DOC), and stabilization on mineral surfaces. Our model sensitivity analyses at an intermediate age site (CT-29ky) highlighted the impact of sorption coefficients and SBSSA on SOM Δ^{14} C values (which is a proxy of SOM residence times). Predicted SOM Δ¹⁴C values showed little sensitivity to changes in sorption coefficients of monomers and dead cell wall material and to the SBSSA, but were very sensitive to vertical carbon input profiles. Our results demonstrate that a model structure with explicit microbial activity, individual SOM components, vertically-resolved aqueous transport, and mineral surface sorption can lead to realistic bulk SOM and Δ^{14} C values, and that the emergent system responses to model parameters and inputs will be different than those from traditional pseudo-first order models. We therefore conclude that integration of this type of more mechanistically explicit model structure should be considered in climate-scale land models.

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References

Ahrens et al., 2015

B. Ahrens, M. Braakhekke, G. Guggenberger, M. Schrumpf, M. Reichstein**Contribution of** sorption, DOC transport and microbial interactions to the 14C age of a soil organic carbon profile: insights from a calibrated process model

Soil Biology and Biochemistry, 88 (2015), pp. 390-402, 10.1016/j.soilbio.2015.06.008

ArticleDownload PDFView Record in Scopus

Allison et al., 2010

S.D. Allison, M.D. Wallenstein, M.A. Bradford**Soil-carbon response to warming dependent on microbial physiology**

Nature Geoscience, 3 (5) (2010), pp. 336-340, 10.1038/ngeo846

CrossRefView Record in Scopus

Amenu et al., 2005

G.G. Amenu, P. Kumar, X.-Z. LiangInterannual variability of deep-layer hydrologic memory and mechanisms of its influence on surface energy fluxes

Journal of Climate, 18 (23) (2005), pp. 5024-5045, 10.1175/JCLI3590.1

CrossRefView Record in Scopus

Arora et al., 2016

B. Arora, D. Dwivedi, S.S. Hubbard, C.I. Steefel, K.H. Williams**Identifying geochemical hot moments and their controls on a contaminated river floodplain system using wavelet and entropy approaches**

Environmental Modelling & Software, 85 (2016), pp. 27-41, <u>10.1016/j.envsoft.2016.08.005</u> ArticleDownload PDFView Record in Scopus

<u>Arora et al., 2011</u>

B. Arora, B.P. Mohanty, J.T. McGuireInverse estimation of parameters for multidomain flow models in soil columns with different macropore densities

Water Resources Research, 47 (2011), <u>10.1029/2010WR009451</u>

Bird et al., 2002

S.B. Bird, J.E. Herrick, M.M. Wander, S.F. Wright**Spatial heterogeneity of aggregate stability** and soil carbon in semi-arid rangeland

Environmental Pollution, 116 (2002), pp. 445-455

ArticleDownload PDFView Record in Scopus

Braakhekke et al., 2011

M.C. Braakhekke, C. Beer, M.R. Hoosbeek, M. Reichstein, B. Kruijt, M. Schrumpf, P. Kabat**Somp rof: a vertically explicit soil organic matter model**

Ecological Modelling, 222 (2011), pp. 1712-1730, 10.1016/j.ecolmodel.2011.02.015

ArticleDownload PDFView Record in Scopus

Braakhekke et al., 2014

M.C. Braakhekke, C. Beer, M. Schrumpf, A. Ekici, B. Ahrens, M.R.Hoosbeek, B. Kruijt, P. Kabat, M. Reichstein**The use of radiocarbon to constrain current and future soil organic matter turnover and transport in a temperate forest**

Journal of Geophysical Research: Biogeosciences, 119 (2014), pp. 372-391, <u>10.1002/2013JG002420</u>

CrossRefView Record in Scopus

Castellano et al., 2015

M.J. Castellano, K.E. Mueller, D.C. Olk, J.E. Sawyer, J. SixIntegrating plant litter quality, soil organic matter stabilization and the carbon saturation concept Global Change Biology, 21 (9) (2015), pp. 3200-3209, <u>10.1111/gcb.12982</u> CrossRefView Record in Scopus

Chadwick et al., 1990

O.A. Chadwick, G.H. Brimhall, D.M. Hendricks**From a black to a gray box - a mass balance** interpretation of pedogenesis

Geomorphology, 3 (3–4) (1990), pp. 369-390, <u>10.1016/0169-555X(90)90012-F</u> <u>ArticleDownload PDFView Record in Scopus</u>

Chorover et al., 2004

J. Chorover, M.K. Amistadi, O.A. Chadwick**Surface charge evolution of mineral-organic** complexes during pedogenesis in Hawaiian basalt Geochimica et Cosmochimica Acta, 68 (2004), pp. 4859-4876, <u>10.1016/j.gca.2004.06.005</u> <u>ArticleDownload PDFView Record in Scopus</u>

Chung et al.,

<u>2010</u>

H. Chung, K.J. Ngo, A. Plante, J. Six**Evidence for carbon saturation in a highly structured** and organic-matter-rich soil

Soil Science Society of America Journal, 74 (2010), p. 130, <u>10.2136/sssaj2009.0097</u> CrossRefView Record in Scopus

> <u>Conant</u> <u>et al.,</u>

<u>2011</u>

R.T. Conant, M.G. Ryan, G.I. Ågren, H.E. Birge, E.A. Davidson, P.E.Eliasson, S.E. Evans, S.D. Fr ey, C.P. Giardina, F.M. Hopkins, R. Hyvönen, M.U.F.Kirschbaum, J.M. Lavallee, J. Leifeld, W.J. P arton, J. Megan Steinweg, M.D.Wallenstein, J.Å. Martin Wetterstedt, M.A. Bradford**Temperature and soil organic matter decomposition rates - synthesis of current knowledge and a way forward**

Global Change Biology, 17 (11) (2011), pp. 3392-3404, <u>10.1111/j.1365-2486.2011.02496.x</u> <u>CrossRefView Record in Scopus</u>

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M.F. Cotrufo, M.D. Wallenstein, C.M. Boot, K. Denef, E. Paul The Microbial Efficiency-Matrix Stabilization (MEMS) framework integrates plant litter decomposition with soil organic matter stabilization: do labile plant inputs form stable soil organic matter? Global Change Biology, 19 (4) (2013), pp. 988-995, <u>10.1111/gcb.12113</u>	
CrossRefView Record in Scopus	
	<u>Curtis,</u>
	2003
G. CurtisP.: comparison of approaches for simulating reactive solute transport involving	
organic degradation reactions by multiple terminal electron acceptors	
Computers & Geosciences, 29 (3) (2003), pp. 319-329, <u>10.1016/S0098-3004(03)00008-6</u>	
ArticleDownload PDFView Record in Scopus	
	Davidson and
	Janssens, 2006
E.A. Davidson, I.A. JanssensTemperature sensitivity of soil carbon decomposition and	
feedbacks to climate change	
Nature, 440 (2006), pp. 165-173, <u>10.1038/nature04514</u>	
CrossRefView Record in Scopus	
	Davis and Kent,
J.A. Davis, D.B. KentSurface complexation modeling in aqueous geochemistry	
Reviews in Mineralogy and Geochemistry, 23 (1990), pp. 176-260	

[online] Available from: Volltext nur via Bestellung

View Record in Scopus

	Davis and Costo
J. Davis, J. CostonApplication of the surface complexation concept to complex mineral	
assemblages	
Environmental Science & Technology (1998), pp. 2820-2828	
[online] Available from:	
http://pubs.acs.org/doi/abs/10.1021/es980312q	
(Accessed 1 October 2014)	
CrossRefView Record in Scopus	
	DeAngelis et al.,
K.M. DeAngelis, D. Chivian, J.L. Fortney, A.P. Arkin, B. Simmons, T.C.Hazen, W.L. SilverChange	
s in microbial dynamics during long-term decomposition in tropical forests	
Soil Biology and Biochemistry, 66 (2013), pp. 60-68, <u>10.1016/j.soilbio.2013.06.010</u>	
ArticleDownload PDFView Record in Scopus	
	Dudal and Gérar
Y. Dudal, F. GérardAccounting for natural organic matter in aqueous chemical equilibrium	
models: a review of the theories and applications	
Earth-Science Reviews, 66 (2004), pp. 199-216, <u>10.1016/j.earscirev.2004.01.002</u>	
ArticleDownload PDFView Record in Scopus	
	Dungait et al., 20
J.A.J. Dungait, D.W. Hopkins, A.S. Gregory, A.P. WhitmoreSoil organic matter turnover is	
governed by accessibility not recalcitrance	
Global Change Biology, 18 (2012), pp. 1781-1796, <u>10.1111/j.1365-2486.2012.02665.x</u>	
CrossRefView Record in Scopus	
	<u>Dzombak, 1990</u>
D. DzombakA.: Surface Complexation Modeling: Hydrous Ferric Oxide	
John Wiley & Sons (1990)	
[online] Available from:	
http://books.google.com/books?hl=en&lr=&id=LGBtoHNUbWwC&pgis=1	
(Accessed 18 November 2014)	
	Feller et al., 1993
C. Feller, E. Schouller, F. Thomas, J. Rouiller, A.J. HerbillonN2-Bet specific surface areas of	
some low activity clay soils and their relationships with secondary constituents and	
organic matter contents	
Soil Science, 153 (4) (1992), pp. 293-299, <u>10.1097/00010694-199204000-00005</u>	
CrossRefView Record in Scopus	
	<u>Frei et al., 2012</u>

S. Frei, K.H. Knorr, S. Peiffer, J.H. Fleckenstein**Surface micro-topography causes hot spots of biogeochemical activity in wetland systems: a virtual modeling experiment**

Journal of Geophysical Research: Biogeosciences, 117 (2012), 10.1029/2012JG002012	
	Frolov, 2014
A. FrolovV: estimation of the statistical characteristics of long-term fluctuations in	
evaporation from large river catchments	
Doklady Earth Sciences, 458 (1) (2014), pp. 1183-1186, <u>10.1134/S1028334X1409027X</u>	
CrossRefView Record in Scopus	
	Frostegard and E
A. Frostegard, E. BààthThe use of phospholipid fatty acid analysis to estimate bacterial and	
fungal biomass in soil	
Biology and Fertility of Soils, 22 (1996), pp. 59-65, <u>10.1007/BF00384433</u>	
View Record in Scopus	
	van Genuchten,
M. van GenuchtenT.: a closed-form equation for predicting the hydraulic conductivity of	
unsaturated Soils1	
Soil Science Society of America Journal, 44 (1980),	
p. 892, <u>10.2136/sssaj1980.03615995004400050002x</u>	
CrossRefView Record in Scopus	
	<u>German et al., 2</u>
D.P. German, K.R.B. Marcelo, M.M. Stone, S.D. AllisonThe Michaelis-Menten kinetics of soil	
extracellular enzymes in response to temperature: a cross-latitudinal study	
Global Change Biology, 18 (4) (2012), pp. 1468-1479, <u>10.1111/j.1365-2486.2011.02615.x</u>	
CrossRefView Record in Scopus	
	<u>Gjettermann et a</u>
B. Gjettermann, M. Styczen, H.C.B. Hansen, F.P. Vinther, S. HansenChallenges in modelling	
dissolved organic matter dynamics in agricultural soil using DAISY	
Soil Biology and Biochemistry, 40 (2008), pp. 1506-1518, <u>10.1016/j.soilbio.2008.01.005</u>	
ArticleDownload PDFView Record in Scopus	
	Goldberg et al., 2
S. Goldberg, L.J. Criscenti, D.R. Turner, J. a. Davis, K.J. Cantrell: adsorption-desorption	
processes in subsurface reactive transport modeling	
Vadose Zone Journal, 6 (3) (2007), p. 407, <u>10.2136/vzj2006.0085</u>	
CrossRefView Record in Scopus	
	<u>Goldfarb et al., 2</u>
K.C. Goldfarb, U. Karaoz, C.A. Hanson, C.A. Santee, M.A. Bradford, K.K.Treseder, M.D. Wallenst	t
ein, E.L. BrodieDifferential growth responses of soil bacterial taxa to carbon substrates of	
varying chemical recalcitrance	
Frontier's in Microbiology, 2 (2011), 10.3389/fmicb.2011.00094	

Gottschalk et al.

P. Gottschalk, J.U. Smith, M. Wattenbach, J. Bellarby, E. Stehfest, N.Arnell, T.J. Osborn, C. Jones , P. SmithHow will organic carbon stocks in mineral soils evolve under future climate? Global projections using RothC for a range of climate change scenarios Biogeosciences, 9 (2012), pp. 3151-3171, 10.5194/bg-9-3151-2012 CrossRefView Record in Scopus

R.F. Grant, T.A. Black, D. Gaumont-Guay, N. Klujn, A.G. Barr, K.Morgenstern, Z. NesicNet ecosystem productivity of boreal aspen forests under drought and climate change: mathematical modelling with Ecosys

Agricultural and Forest Meteorology, 140 (2006), pp. 152-170, 10.1016/j.agrformet.2006.01.012 ArticleDownload PDFView Record in Scopus

Greenland, 1971

Grant et al., 200

D.J. GreenlandInteractions between humic and fulvic acids and clays Soil Science, 111 (1) (1971), pp. 34-41 [online] Available from: http://agris.fao.org/agris-search/search.do?recordID=US201302364487 (Accessed 1 October 2014) CrossRefView Record in Scopus

B.H. Gu, J. Schmitt, Z.H. Chen, L.Y. Liang, J.F. McCarthyAdsorption and desorption of natural organic-matter on iron-oxide - mechanisms and models Environmental Science & Technology, 28 (1994), pp. 38-46, 10.1021/es00050a007 CrossRefView Record in Scopus

B. Guenet, T. Eglin, N. Vasilyeva, P. Peylin, P. Ciais, C. ChenuThe relative importance of decomposition and transport mechanisms in accounting for soil organic carbon profiles Biogeosciences, 10 (4) (2013), pp. 2379-2392, 10.5194/bg-10-2379-2013 CrossRefView Record in Scopus

G. Guggenberger, K. Kaiser

Dissolved Organic Matter in Soil: Challenging the Paradigm of Sorptive Preservation, in Geoderma, vol. 113 (2003), pp. 293-310

ArticleDownload PDFView Record in Scopus

J. Hassink, A.P. Whitmore A model of the physical protection of organic matter in soils Soil Science Society of America Journal, 61 (1) (1997), p. 131, 10.2136/sssaj1997.03615995006100010020x

Gu et al., 1994

Guenet et al., 20

Guggenberger a

Hassink and Wh

CrossRefView Record in Scopus

Y. He, J. Yang, Q. Zhuang, A.D. McGuire, Q. Zhu, Y. Liu, R.O. Teskey**Uncertainty in the fate of** soil organic carbon: a comparison of three conceptually different decomposition models at a larch plantation Journal of Geophysical Research: Biogeosciences, 119 (9) (2014), pp. 1892-1905, <u>10.1002/2014JG002701</u>

CrossRefView Record in Scopus

T. Hiemstra, J.C. De Wit, W. Van RiemsdijkMultisite proton adsorption modeling at the solid/solution interface of (hydr)oxides: a new approach
Journal of Colloid and Interface Science, 133 (1989), pp. 105-117, <u>10.1016/0021-9797(89)90285-</u>
<u>3</u>
<u>ArticleDownload PDFView Record in Scopus</u>

M.E. Hodson, S.J. Langan, F.M. Kennedy, D.C. Bain**Variation in soil surface area in a** chronosequence of soils from Glen Feshie, Scotland and its implications for mineral weathering rate calculations Geoderma, 85 (1) (1998), pp. 1-18, <u>10.1016/S0016-7061(98)00013-5</u> ArticleDownload PDFView Record in Scopus

Holub et al., 200

S.M. Holub, K. Lajtha, J.D.H. Spears, J.A. Tóth, S.E. Crow, B.A. Caldwell, M. Papp, P.T. Nagy**Org** anic matter manipulations have little effect on gross and net nitrogen transformations in two temperate forest mineral soils in the USA and central Europe Forest Ecology and Management, 214 (2005), pp. 320-330, <u>10.1016/j.foreco.2005.04.016</u> <u>ArticleDownload PDFView Record in Scopus</u>

Jagadamma et a

S. Jagadamma, M.A. Mayes, J.R. Phillips**Selective sorption of dissolved organic carbon compounds by temperate soils** PLoS One, 7 (2012), <u>10.1371/journal.pone.0050434</u>

Jardine et al., 20

P.M. Jardine, M.A. Mayes, P.J. Mulholland, P.J. Hanson, J.R. Tarver, R.J.Luxmoore, J.F. McCarthy , G.V. Wilson**Vadose zone flow and transport of dissolved organic carbon at multiple scales in humid regimes** Vadose Zone Journal, 5 (2006), p. 140, <u>10.2136/vzj2005.0036</u> CrossRefView Record in Scopus

Jardine et al., 19

P.M. Jardine, J.F. McCarthy, N.L. Weber**Mechanisms of dissolved organic carbon adsorption** on soil

<u>He et al., 2014</u>

Hiemstra et al., 2

Hodson et al., 19

Soil Science Society of America Journal, 53 (5) (1989),

p. 1378, <u>10.2136/sssaj1989.03615995005300050013x</u> <u>CrossRefView Record in Scopus</u>

Jardine et al., 19

P.M. Jardine, G.V. Wilson, J.F. McCarthy, R.J. Luxmoore, D.L. Taylor, L.W.Zelazny**Hydrogeoche** mical processes controlling the transport of dissolved organic carbon through a forested hillslope

Journal of Contaminant Hydrology, 6 (1990), pp. 3-19, <u>10.1016/0169-7722(90)90008-5</u> <u>ArticleDownload PDFView Record in Scopus</u>

Jenkinson and Coleman, 2008

D.S. Jenkinson, K. Coleman**The turnover of organic carbon in subsoils. Part 2. Modelling** carbon turnover

European Journal of Soil Science, 59 (2008), pp. 400-413, 10.1111/j.1365-2389.2008.01026.x

CrossRefView Record in Scopus

Jobbágy and Jackson, 2000

E.G. Jobbágy, R.B. JacksonThe vertical distribution of soil organic carbon and its relation to climate and vegetation

Ecological Applications, 10 (2) (2000), pp. 423-436, 10.1890/1051-

0761(2000)010[0423:TVDOSO]2.0.CO;2

CrossRefView Record in Scopus

Kaiser and Guggenberger, 2003

K. Kaiser, G. GuggenbergerMineral surfaces and soil organic matter

European Journal of Soil Science, 54 (2) (2003), pp. 219-236, <u>10.1046/j.1365-2389.2003.00544.x</u> <u>CrossRefView Record in Scopus</u>

Kalbitz et al., 2003

K. Kalbitz, J. Schmerwitz, D. Schwesig, E. Matzner

Biodegradation of soil-derived dissolved organic matter as related to its properties, in Geoderma, 113 (2003), pp. 273-291

ArticleDownload PDFView Record in Scopus

Kalbitz et al., 2000

K. Kalbitz, S. Solinger, J.-H. Park, B. Michalzik, E. Matzner**Controls on the dynamics of dissolved organic matter in soils: a review**

Soil Science, 165 (2000), pp. 277-304, <u>10.1097/00010694-200004000-00001</u>

CrossRefView Record in Scopus

Kögel-Knabner, 2002

I. Kögel-KnabnerThe macromolecular organic composition of plant and microbial residues as inputs to soil organic matter

Soil Biology and Biochemistry, 34 (2) (2002), pp. 139-162, <u>10.1016/S0038-0717(01)00158-4</u>

ArticleDownload PDFView Record in Scopus

Kothawala and Moore, 2009

D.N. Kothawala, T.R. Moore**Adsorption of dissolved nitrogen by forest mineral soils** Canadian Journal of Forest Research, 39 (2009), pp. 2381-2390, <u>10.1139/X09-147</u>

CrossRefView Record in Scopus

Kothawala et al., 2008

D.N. Kothawala, T.R. Moore, W.H. Hendershot**Adsorption of dissolved organic carbon to** mineral soils: a comparison of four isotherm approaches

Geoderma, 148 (2008), pp. 43-50, <u>10.1016/j.geoderma.2008.09.004</u>

ArticleDownload PDFView Record in Scopus

Kothawala et al., 2009

D.N. Kothawala, T.R. Moore, W.H. Hendershot**Soil properties controlling the adsorption of dissolved organic carbon to mineral soils**

Soil Science Society of America Journal, 73 (2009), p. 1831, <u>10.2136/sssaj2008.0254</u>

CrossRefView Record in Scopus

Koven et al., 2013

C.D. Koven, W.J. Riley, Z.M. Subin, J.Y. Tang, M.S. Torn, W.D. Collins, G.B.Bonan, D.M. Lawrenc e, S.C. Swenson**The effect of vertically resolved soil biogeochemistry and alternate soil C** and N models on C dynamics of CLM4

Biogeosciences, 10 (11) (2013), pp. 7109-7131, <u>10.5194/bg-10-7109-2013</u>

CrossRefView Record in Scopus

Lawrence et al., 2009

C.R. Lawrence, J.C. Neff, J.P. Schimel**Does adding microbial mechanisms of decomposition** improve soil organic matter models? A comparison of four models using data from a pulsed rewetting experiment

Soil Biology and Biochemistry, 41 (9) (2009), pp. 1923-1934, <u>10.1016/j.soilbio.2009.06.016</u> <u>ArticleDownload PDFView Record in Scopus</u>

Lawrence et al., 2011

D.M. Lawrence, K.W. Oleson, M.G. Flanner, P.E. Thornton, S.C.Swenson, P.J. Lawrence, X. Zen g, Z.-L. Yang, S. Levis, K. Sakaguchi, G.B. Bonan, A.G. Slater**Parameterization improvements and functional and structural advances in version 4 of the community land model** Journal of Advances in Modeling Earth Systems, 3 (1) (2011), <u>10.1029/2011MS00045</u> n/a-n/a

Leifeld et al., 2009

J. Leifeld, M. Zimmermann, J. Fuhrer, F. Conen**Storage and turnover of carbon in grassland** soils along an elevation gradient in the Swiss Alps Global Change Biology, 15 (2009), pp. 668-679, <u>10.1111/j.1365-2486.2008.01782.x</u>

CrossRefView Record in Scopus

Lenhart and Honeyman, 1999

J.J. Lenhart, B.D. Honeyman**Uranium(VI) sorption to hematite in the presence of humic acid** Geochimica et Cosmochimica Acta, 63 (19–20) (1999), pp. 2891-2901, <u>10.1016/S0016-</u>

7037(99)00269-0

ArticleDownload PDFView Record in Scopus

Levin and Hesshaimer, 2000

I. Levin, V. Hesshaimer**Radiocarbon – a unique tracer of global carbon cycle dynamics** Radiocarbon, 42 (2000), pp. 69-80, <u>10.2458/azu js rc.42.3855</u>

CrossRefView Record in Scopus

Levin and Kromer, 2004

I. Levin, B. KromerThe tropospheric 14CO2 level in mid-latitudes of the northern

hemisphere (1959–2003)

Radiocarbon, 46 (2004), pp. 1261-1272

[online] Available from:

https://journals.uair.arizona.edu/index.php/radiocarbon/article/view/4181

CrossRefView Record in Scopus

Li et al., 2008

F. Li, W.P. Kustas, M.C. Anderson, J.H. Prueger, R.L. Scott**Effect of remote sensing spatial** resolution on interpreting tower-based flux observations

Remote Sensing of Environment, 112 (2008), pp. 337-349, 10.1016/j.rse.2006.11.032

ArticleDownload PDFView Record in Scopus

Luo et al., 2015

Z. Luo, E. Wang, H. Zheng, J.A. Baldock, O.J. Sun, Q. Shao**Convergent modelling of past soil** organic carbon stocks but divergent projections

Biogeosciences, 12 (14) (2015), pp. 4373-4383, 10.5194/bg-12-4373-2015

CrossRefView Record in Scopus

Lutzow et al., 2006

M. v. Lutzow, I. Kogel-

Knabner, K. Ekschmitt, E. Matzner, G.Guggenberger, B. Marschner, H. Flessa**Stabilization of** organic matter in temperate soils: mechanisms and their relevance under different soil conditions - a review

European Journal of Soil Science, 57 (4) (2006), pp. 426-445, <u>10.1111/j.1365-2389.2006.00809.x</u> <u>CrossRefView Record in Scopus</u>

Maggi and Riley, 2010

F. Maggi, W.J. RileyMathematical treatment of isotopologue and isotopomer speciation and fractionation in biochemical kinetics

Geochimica et Cosmochimica Acta, 74 (6) (2010), pp. 1823-1835, <u>10.1016/j.gca.2009.12.021</u> <u>ArticleDownload PDFView Record in Scopus</u>

Maher et al., 2009

K. Maher, C.I. Steefel, A.F. White, D.A. Stonestrom**The role of reaction affinity and secondary** minerals in regulating chemical weathering rates at the Santa Cruz Soil Chronosequence, California

Geochimica et Cosmochimica Acta, 73 (2009), pp. 2804-2831, <u>10.1016/j.gca.2009.01.030</u> <u>ArticleDownload PDFView Record in Scopus</u>

Masiello et al., 2004

C.A. Masiello, O.A. Chadwick, J. Southon, M.S. Torn, J.W. Harden**Weathering controls on** mechanisms of carbon storage in grassland soils

Global Biogeochemical Cycles, 18 (2004), pp. 1-9, <u>10.1029/2004GB002219</u>

View Record in Scopus

Mayer, 1994

L. MayerM.: relationships between mineral surfaces and organic carbon concentrations in soils and sediments

Chemical Geology, 114 (3–4) (1994), pp. 347-363, 10.1016/0009-2541(94)90063-9

ArticleDownload PDFView Record in Scopus

Mayes et al., 2012

M.A. Mayes, K.R. Heal, C.C. Brandt, J.R. Phillips, P.M. Jardine**Relation between soil order and** sorption of dissolved organic carbon in temperate subsoils

Soil Science Society of America Journal, 76 (2012), p. 1027, <u>10.2136/sssaj2011.0340</u>

CrossRefView Record in Scopus

Merritts et al., 1991

D.J. Merritts, O.A. Chadwick, D.M. HendricksRates and processes of soil evolution on uplifted marine terraces, northern California

Geoderma, 51 (1991), pp. 241-275, 10.1016/0016-7061(91)90073-3

ArticleDownload PDFView Record in Scopus

Molins et al., 2012

S. Molins, D. Trebotich, C.I. Steefel, C. ShenAn investigation of the effect of pore scale flow on average geochemical reaction rates using direct numerical simulation

Water Resources Research, 48 (2012), 10.1029/2011WR011404

Moore and Turunen, 2004

T.R. Moore, J. Turunen**Carbon accumulation and storage in mineral subsoil beneath peat** Soil Science Society of America Journal, 68 (2004), p. 690, <u>10.2136/sssaj2004.0690</u>

CrossRefView Record in Scopus

Murphy et al., 1999

R.J. Murphy, J.J. Lenhart, B.D. Honeyman**The sorption of thorium (IV) and uranium (VI) to** hematite in the presence of natural organic matter

Colloids and Surfaces A Physicochemical and Engineering Aspects, 157 (1-3) (1999), pp. 47-

62, <u>10.1016/S0927-7757(99)00115-6</u>

ArticleDownload PDFView Record in Scopus

<u>Neely et al., 1991</u>

C.L. Neely, M.H. Beare, W.L. Hargrove, D.C. Coleman**Relationships between fungal and bacterial substrate-induced respiration, biomass and plant residue decomposition** Soil Biology and Biochemistry, 23 (1991), pp. 947-954, <u>10.1016/0038-0717(91)90175-J</u> <u>ArticleDownload PDFView Record in Scopus</u>

Parton et al., 2010

W.J. Parton, P.J. Hanson, C. Swanston, M. Torn, S.E. Trumbore, W. Riley, R. KellyForCent
model development and testing using the Enriched Background Isotope Study experiment
Journal of Geophysical Research: Biogeosciences, 115 (2010), <u>10.1029/2009JG001193</u>

Parton et al., 1987

W.J. Parton, D.S. Schimel, C.V. Cole, D.S. OjimaAnalysis of factors controlling soil organic matter levels in great plains Grasslands1

Soil Science Society of America Journal, 51 (5) (1987),

p. 1173, <u>10.2136/sssaj1987.03615995005100050015x</u>

CrossRefView Record in Scopus

Pau et al., 2016

G.S.H. Pau, C. Shen, W.J. Riley, Y. Liu**Accurate and efficient prediction of fine-resolution hydrologic and carbon dynamic simulations from coarse-resolution models** Water Resources Research, 52 (2) (2016), pp. 791-812, <u>10.1002/2015WR017782</u>

CrossRefView Record in Scopus

Pruess, 1999

K. Pruess**TOUGH2 User's Guide Version 2**

Lawrence Berkeley Natl. Lab. (1999)

210 [online] Available from:

http://escholarship.org/uc/item/4df6700h.pdf

Reed, 1982

M. ReedH.: calculation of multicomponent chemical equilibria and reaction processes in systems involving minerals, gases and an aqueous phase

Geochimica et Cosmochimica Acta, 46 (1982), pp. 513-528, <u>10.1016/0016-7037(82)90155-7</u> ArticleDownload PDFView Record in Scopus

<u>Riley et al., 2014</u>

W.J. Riley, F. Maggi, M. Kleber, M.S. Torn, J.Y. Tang, D. Dwivedi, N. GuerryLong residence times of rapidly decomposable soil organic matter: application of a multi-phase, multi-component, and vertically resolved model (BAMS1) to soil carbon dynamics

Geoscientific Model Development, 7 (4) (2014), pp. 1335-1355, <u>10.5194/gmd-7-1335-2014</u>

CrossRefView Record in Scopus

Riley and Shen, 2014

W.J. Riley, C. Shen**Characterizing coarse-resolution watershed soil moisture heterogeneity** using fine-scale simulations

Hydrology and Earth System Sciences, 18 (7) (2014), pp. 2463-2483, <u>10.5194/hess-18-2463-</u> 2014

CrossRefView Record in Scopus

<u>Romaní et al., 2006</u>

A.M. Romaní, H. Fischer, C. Mille-Lindblom, L.J. TranvikInteractions of bacteria and fungi on decomposing litter: differential extracellular enzyme activities

Ecology, 87 (2006), pp. 2559-2569, <u>10.1890/0012-9658(2006)87[2559:IOBAFO]2.0.CO;2</u> CrossRefView Record in Scopus

Sahai and Sverjensky, 1997a

N. Sahai, D.A. SverjenskyEvaluation of internally consistent parameters for the triple-layer model by the systematic analysis of oxide surface titration data

Geochimica et Cosmochimica Acta, 61 (1997), pp. 2801-2826, <u>10.1016/S0016-7037(97)00128-2</u>

ArticleDownload PDFView Record in Scopus

Sahai and Sverjensky, 1997b

N. Sahai, D.A. SverjenskySolvation and electrostatic model for specific electrolyte

adsorption

Geochimica et Cosmochimica Acta, 61 (1997), pp. 2827-2848, <u>10.1016/S0016-7037(97)00127-0</u> ArticleDownload PDFView Record in Scopus

Sanderman et al., 2008

J. Sanderman, J.A. Baldock, R. Amundson**Dissolved organic carbon chemistry and dynamics** in contrasting forest and grassland soils

Biogeochemistry, 89 (2008), pp. 181-198, <u>10.1007/s10533-008-9211-x</u>

CrossRefView Record in Scopus

Schenk and Jackson, 2002

H.J. Schenk, R.B. JacksonThe global biogeography of roots

Ecological Monographs, 72 (2002), pp. 311-328, 10.1890/0012-

9615(2002)072[0311:TGBOR]2.0.CO;2

CrossRefView Record in Scopus

Schenk and Jackson, 2005

H.J. Schenk, R.B. JacksonMapping the global distribution of deep roots in relation to climate and soil characteristics

Geoderma, 126 (2005), pp. 129-140, 10.1016/j.geoderma.2004.11.018

ArticleDownload PDFView Record in Scopus

Schimel et al., 2011

J.P. Schimel, J.Å.M. Wetterstedt, P.A. Holden, S.E. Trumbore**Drying/rewetting cycles mobilize** old C from deep soils from a California annual grassland

Soil Biology and Biochemistry, 43 (5) (2011), pp. 1101-1103, 10.1016/j.soilbio.2011.01.008

ArticleDownload PDFView Record in Scopus

Schmidt et al., 2011

M.W.I. Schmidt, M.S. Torn, S. Abiven, T. Dittmar, G. Guggenberger, I.A.Janssens, M. Kleber, I. Kö gel-Knabner, J. Lehmann, D.A.C. Manning, P. Nannipieri, D.P. Rasse, S. Weiner, S.E. Trumbor e**Persistence of soil organic matter as an ecosystem property**

Nature, 478 (2011), pp. 49-56, 10.1038/nature10386

CrossRefView Record in Scopus

Sibanda and Young, 1986

H.M. Sibanda, S.D. YoungCompetitive adsorption of humus acids and phosphate on goethite, gibbsite and two tropical soils

Journal of Soil Science, 37 (2) (1986), pp. 197-204, <u>10.1111/j.1365-2389.1986.tb00020.x</u>

CrossRefView Record in Scopus

Sivapalan, 2005

M. SivapalanPattern, Process and Function: Elements of a Unified Theory of Hydrology at the Catchment Scale, in Encyclopedia of Hydrological Sciences

(2005), pp. 193-219

View Record in Scopus

Six et al., 2004

J. Six, H. Bossuyt, S. Degryze, K. DenefA history of research on the link between (micro)aggregates, soil biota, and soil organic matter dynamics

Soil and Tillage Research, 79 (2004), pp. 7-31, <u>10.1016/j.still.2004.03.008</u>

ArticleDownload PDFView Record in Scopus

Six et al., 2002

J. Six, R.T. Conant, E.A. Paul, K. Paustian**Stabilization mechanisms of soil organic matter:** implications for C-saturation of soils

Plant Soil, 241 (2002), pp. 155-176, 10.1023/A:1016125726789

CrossRefView Record in Scopus

<u>Six et al., 2000</u>

J. Six, E.T. Elliott, K. PaustianSoil structure and soil organic matter: II. A normalized stability index and the effect of mineralogy

Soil Science Society of America, 64 (2000), pp. 1042-1049, 10.2136/sssaj2000.6431042x

CrossRefView Record in Scopus

Sollins et al., 1996

P. Sollins, P. Homann, B.A. Caldwell**Stabilization and destabilization of soil organic matter:** mechanisms and controls

Geoderma, 74 (1–2) (1996), pp. 65-105, 10.1016/S0016-7061(96)00036-5

ArticleDownload PDFView Record in Scopus

Steefel and Lasaga, 1994

C.I. Steefel, A.C. Lasaga**A coupled model for transport of multiple chemical species and** kinetic precipitation/dissolution reactions with application to reactive flow in single phase hydrothermal systems American Journal of Science, 294 (1994), pp. 529-592, <u>10.2475/ajs.294.5.529</u> <u>CrossRefView Record in Scopus</u>

Stewart et al., 2009

C.E. Stewart, K. Paustian, R.T. Conant, A.F. Plante, J. Six**Soil carbon saturation: implications for measurable carbon pool dynamics in long-term incubations** Soil Biology and Biochemistry, 41 (2009), pp. 357-366, <u>10.1016/j.soilbio.2008.11.011</u> ArticleDownload PDFView Record in Scopus

Stuiver and Polach, 1999

M. Stuiver, H.A. Polach**Radiocarbon** (1999)

<u>Sulman et al., 2014</u>

B.N. Sulman, C.A. Oishi, R.P. Phillips, S. E., P. S.Microbe-driven turnover offsets mineralmediated storage of soil carbon under elevated CO2
Nature Climate Change, 4 (2014), pp. 1099-1102, <u>10.1038/nclimate2436</u>
<u>CrossRefView Record in Scopus</u>

Tang and Riley, 2013

J. Tang, W.J. RileyImpacts of a new bare-soil evaporation formulation on site, regional, and global surface energy and water budgets in CLM4 Journal of Advances in Modeling Earth Systems, 5 (3) (2013), pp. 558-571, <u>10.1002/jame.20034</u>

CrossRefView Record in Scopus

Tang and Riley, 2015

J. Tang, W.J. RileyWeaker carbon–climate feedbacks resulting from microbial and abiotic interactions

Nature Climate Change, 5.1 (2015), pp. 56-60, <u>10.1038/nclimate2438</u> <u>CrossRefView Record in Scopus</u>

Tang et al.,

<u>2013</u>

J.Y. Tang, W.J. Riley, C.D. Koven, Z.M. Subin**CLM4-BeTR, a generic biogeochemical transport** and reaction module for CLM4: model development, evaluation, and application Geoscientific Model Development, 6 (1) (2013), pp. 127-140, <u>10.5194/gmd-6-127-2013</u> <u>CrossRefView Record in Scopus</u>

> <u>Theven</u> ot et al., 2010

M. Thevenot, M.F. Dignac, C. Rumpel**Fate of lignins in soils: a review** Soil Biology and Biochemistry, 42 (2010), pp. 1200-1211, <u>10.1016/j.soilbio.2010.03.017</u> <u>ArticleDownload PDFView Record in Scopus</u>

Τ

CrossRefView Record in Scopus

			<u>Torn</u>
			<u>et al.,</u>
			<u>2009</u>

M.S. Torn, C.W. Swanston, C. Castanha, S. Trumbore

N. Senesi, B. Xing, P.M. Huang (Eds.), Storage and Turnover of Natural Organic Matter in Soil, IUPAC Seri. (2009)

[online] Available from:

http://www.wiley.com/WileyCDA/WileyTitle/productCd-047041300X.html (Accessed 11 June 2015)

Torn et al.,

<u>1997a</u>

M.S. Torn, S.E. Trumbore, O.A. Chadwick, P.M. Vitousek, D.M. Hendricks**Mineral control of soil** organic carbon storage and turnover Nature, 389 (1997), pp. 170-173, <u>10.1038/38260</u> <u>CrossRefView Record in Scopus</u>

Torn et al., 1997

M.S. Torn, S.E. Trumbore, O.A. Chadwick, P.M. Vitousek, D.M. Hendricks**Mineral control of soil** organic carbon storage and turnover

Nature, 389 (1997), pp. 170-173, <u>10.1038/38260</u> <u>CrossRefView Record in Scopus</u>

S. Trumbore**Age of soil organic matter and soil respiration: radiocarbon constraints on belowground C dynamics** Ecological Applications, 10 (2000), pp. 399-411, <u>10.1890/1051-</u> 0761(2000)010[0399:AOSOMA]2.0.CO;2 <u>CrossRefView Record in Scopus</u>

S. Trumbore**Radiocarbon and soil carbon dynamics** Annual Review of Earth and Planetary Sciences, 37 (2009), pp. 47-66, <u>10.1146/annurev.earth.36.031207.124300</u> <u>CrossRefView Record in Scopus</u>

S.E. Trumbore**Potential responses of soil organic carbon to global environmental change** Proceedings of the National Academy of Sciences of the United States of America, 94(1997), pp. 8284-8291, <u>10.1073/pnas.94.16.8284</u> <u>CrossRefView Record in Scopus</u>

D.P. Turner, S. Ollinger, M. -I. Smith, O. Krankina, M. Gregory**Scaling net primary production to a MODIS footprint in support of Earth observing system product validation** International Journal of Remote Sensing, 25 (2004), pp. 1961-1979, <u>10.1080/0143116031000150013</u> <u>CrossRefView Record in Scopus</u>

T. Vanwalleghem, U. Stockmann, B. Minasny, A.B. McBratneyA quantitative model for integrating landscape evolution and soil formation
Journal of Geophysical Research: Earth Surface, 118 (2) (2013), pp. 331-347, <u>10.1029/2011JF002296</u>
<u>CrossRefView Record in Scopus</u>

Wang et al., 201

Trumbore, 2000

Trumbore, 2009

Trumbore, 1997

Turner et al., 200

Vanwalleghem e

Y. Wang, Z. Wang, Y. Li

B. Bond-Lamberty (Ed.), Storage/Turnover rate of inorganic carbon and its dissolvable part in the profile of saline/alkaline soils, PLoS One, 8 (11) (2013),
p. e82029, <u>10.1371/journal.pone.0082029</u>
CrossRef

Wattel-Koekkoek

E.J.W. Wattel-Koekkoek, P. BuurmanMean residence time of kaolinite and smectite-bound
organic matter in mozambiquan soils
Soil Science Society of America Journal, 68 (2004), pp. 154-161, <u>10.2136/sssaj2004.1540</u>
CrossRefView Record in Scopus

Westall et al., 19

J.C. Westall, J.D. Jones, G.D. Turner, J.M. Zachara**Models for association of metal-ions with** heterogeneous environmental sorbents **.1.** Complexation of Co(li) by leonardite humic-acid as a function of ph and Naclo4 concentration Environmental Science & Technology, 29 (4) (1995), pp. 951-959, <u>10.1021/es00004a015</u>

CrossRefView Record in Scopus

White et al., 199

White et al., 200

A.F. White, A.E. Blum, M.S. Schulz, T.D. Bullen, J.W. Harden, M.L. Peterson**Chemical** weathering rates of a soil chronosequence on granitic alluvium: I. Quantification of mineralogical and surface area changes and calculation of primary silicate reaction rates Geochimica et Cosmochimica Acta, 60 (14) (1996), pp. 2533-2550, <u>10.1016/0016-</u> <u>7037(96)00106-8</u>

ArticleDownload PDFView Record in Scopus

A.F. White, M.S. Schulz, D.A. Stonestrom, D.V. Vivit, J. Fitzpatrick, T.D.Bullen, K. Maher, A.E. Blu mChemical weathering of a marine terrace chronosequence, Santa Cruz, California. Part II: solute profiles, gradients and the comparisons of contemporary and long-term weathering rates

Geochimica et Cosmochimica Acta, 73 (2009), pp. 2769-2803, <u>10.1016/j.gca.2009.01.029</u> <u>ArticleDownload PDFView Record in Scopus</u>

C.J. Whitfield, C. Reid**Predicting surface area of coarse-textured soils: implications for** weathering rates

Canadian Journal of Soil Science, 93 (5) (2013), pp. 621-630, <u>10.4141/cjss2012-009</u> <u>CrossRefView Record in Scopus</u>

Xu et al., 2004

Whitfield and Re

L. Xu, D.D. Baldocchi, J. Tang**How soil moisture, rain pulses, and growth alter the response** of ecosystem respiration to temperature Global Biogeochemical Cycles, 18 (2004), pp. 1-10, <u>10.1029/2004GB002281</u> <u>View Record in Scopus</u>

<u>Xu, 2008</u>

T. Xu**TOUGHREACT User's Guide: a Simulation Program for Non-isothermal Multiphase Reactive Geochemical Transport in Variably Saturated Geologic Media, V1.2.1** Lawrence Berkeley Natl. Lab. (2008) [online] Available from:

http://escholarship.org/uc/item/9r80098d

(Accessed 6 November 2015)

 Xu et al., 2011

 T. Xu, N. Spycher, E. Sonnenthal, G. Zhang, L. Zheng, K. PruessToughreact version 2.0: a

 simulator for subsurface reactive transport under non-isothermal multiphase flow

 conditions

 Computers & Geosciences, 37 (2011), pp. 763-774, 10.1016/j.cageo.2010.10.007

 ArticleDownload PDFView Record in Scopus

 Yeh and Tripathi.

G.T. Yeh, V.S. Tripathi**A model for simulating transport of reactive multispecies components: model development and demonstration** Water Resources Research, 27 (1991), pp. 3075-3094, <u>10.1029/91WR02028</u> <u>CrossRefView Record in Scopus</u>