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# 1 **Chloroform (CHCl<sub>3</sub>) emissions from coastal Antarctic tundra**

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16 **Key points:**

- 17 1. Antarctic tundra is a natural source of CHCl<sub>3</sub>, emitting up to 0.1 Gg each year into  
18 the atmosphere.
- 19 2. Penguins input organic matter and marine halides into soil through excrement,  
20 which facilitate microbial-mediated CHCl<sub>3</sub> production.
- 21 3. The strength of CHCl<sub>3</sub> source will vary in response to changes in penguin  
22 population/colony size, and the extent of Antarctic warming.

23 **Abstract**

24 In this study, the first *in situ* static-chamber measurements were conducted at coastal  
25 Antarctica tundra for CHCl<sub>3</sub> fluxes, which showed that CHCl<sub>3</sub> was naturally emitted from  
26 the Antarctic tundra at  $35 \pm 27 \text{ nmol m}^{-2} \text{ d}^{-1}$ , comparable to other reported important natural  
27 sources. Significantly enhanced CHCl<sub>3</sub> emission rates ( $66 \pm 20 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) were observed  
28 from ornithogenic soil on the island populated with penguins, which was rich in organic  
29 matter and halides coming from penguin excrements. It is estimated that Antarctic tundra  
30 emits up to 0.1 Gg CHCl<sub>3</sub> per year, which is an important source for regional atmospheric  
31 CHCl<sub>3</sub>. Laboratory-based incubations suggested that organic carbon and chlorine inputs by  
32 penguins may stimulate O<sub>2</sub> dependent microbial-mediated CHCl<sub>3</sub> emission from the  
33 Antarctic tundra, and all tundra soils showed the maximum CHCl<sub>3</sub> emission at 4 °C. The  
34 strength of this CHCl<sub>3</sub> source is also expected to change in response to Antarctic warming.

35 **Plain language summary**

36 Chloroform (CHCl<sub>3</sub>) is the second largest natural carrier of atmospheric chlorine,  
37 which can catalyze stratospheric ozone depletion. Natural sources of CHCl<sub>3</sub> are believed  
38 to predominate over anthropogenic sources, accounting for 50-90% of global CHCl<sub>3</sub>

39 emissions. Among the natural sources, soils are the second largest source, after the ocean.  
40 This study conducted the first *in situ* static-chamber measurements and lab-based  
41 incubations on CHCl<sub>3</sub> emissions from Antarctic tundra, and found that it was an important  
42 regional source, emitting up to 0.1 Gg CHCl<sub>3</sub> each year (< 1% of natural terrestrial sources)  
43 into the atmosphere. Penguin activities deposited large amounts of excrement in colony  
44 tundra and enhanced organic matter and chlorine content in the soil, which promoted the  
45 production of CHCl<sub>3</sub> mediated by microbial activities. Temperature-controlled incubations  
46 indicated that tundra soils showed the maximum CHCl<sub>3</sub> emission at 4 °C, and temperature  
47 increase and freeze-thaw cycles might influence annual and seasonal CHCl<sub>3</sub> emissions  
48 from Antarctic tundra. This study suggested that the strength of CHCl<sub>3</sub> source will vary in  
49 response to changes in penguin population/colony size, and the extent of Antarctic  
50 warming.

## 51 **Introduction**

52 Chlorine-containing halocarbons, such as chloromethane (CH<sub>3</sub>Cl), chloroform (CHCl<sub>3</sub>),  
53 and chlorofluorocarbons (CFCs), can photo-dissociate in the stratosphere, releasing  
54 reactive chlorine radicals that catalyze ozone depletion. Among the naturally produced  
55 atmospheric chlorocarbons, CHCl<sub>3</sub> is the second largest natural carrier of chlorine after  
56 CH<sub>3</sub>Cl. With an average tropospheric lifetime of 149 days (Engel et al., 2018), it is  
57 categorized as a very short-lived substance (VSLS). Hence, the ozone-depleting capacity  
58 of CHCl<sub>3</sub> was thought to be minor and was not regulated by the Montreal Protocol.  
59 However, observations over the past decade showed its atmospheric molar fractions have  
60 been steadily increasing (Engel et al., 2018; Fang et al., 2019). Numerical model  
61 simulations also indicated the recovery of stratospheric ozone may be significantly delayed  
62 if atmospheric concentrations of VSLSs, including CHCl<sub>3</sub>, continue to grow (Fang et al.,  
63 2019).

64 Natural sources of CHCl<sub>3</sub> are believed to predominate over anthropogenic sources,  
65 accounting for 50-90% of global CHCl<sub>3</sub> emissions (McCulloch, 2003; Worton et al., 2006).  
66 The natural emissions of CHCl<sub>3</sub> are mainly from surface ocean (~360 Gg yr<sup>-1</sup>) and  
67 terrestrial soils (over 200 Gg yr<sup>-1</sup>). Rhew et al (2008a) first measured CHCl<sub>3</sub> fluxes from  
68 the Arctic tundra, and an extrapolation of these fluxes suggested that tundra globally could  
69 contribute to 1–2% of atmospheric CHCl<sub>3</sub>. Tundra in Greenland and Northern Scandinavia  
70 also showed large emissions of CHCl<sub>3</sub> (Johnsen et al., 2016; Macdonald et al., 2020). Ice-  
71 free tundra in the Antarctica is as large as  $7.2 \times 10^{10}$  m<sup>2</sup>, but no CHCl<sub>3</sub> flux measurement  
72 has ever been conducted due to the remote location and adverse weather conditions (Lee  
73 et al., 2017; Terauds & Lee, 2016).

74 Different from other terrestrial ecosystems, Antarctic tundra is often colonized by large  
75 numbers of sea animals, such as penguins (Croxall et al., 2002; Sun et al., 2000). Penguin

76 activities strongly alter local biogeochemical environments through depositing guano,  
77 promoting the formation of ornithogenic tundra soil, which is fertile in organic matter and  
78 marine elements (Michel et al., 2006; Schaefer et al., 2008). The activities of  
79 microorganisms are often higher in these ornithogenic soils, compared to normal tundra  
80 soil (Tscherko et al., 2003; Zdanowski et al., 2005). Significantly higher emission rates of  
81 greenhouse gases and other trace gases were observed from ornithogenic soil in penguin  
82 colonies (Zhu et al., 2006, 2008, 2009). Therefore, given the abundant organic carbon,  
83 chloride contents and microorganism populations in Antarctic colony tundra soil, the  
84 unique biogeochemical environment may facilitate the chlorination of organic matter to  
85 produce  $\text{CHCl}_3$ .

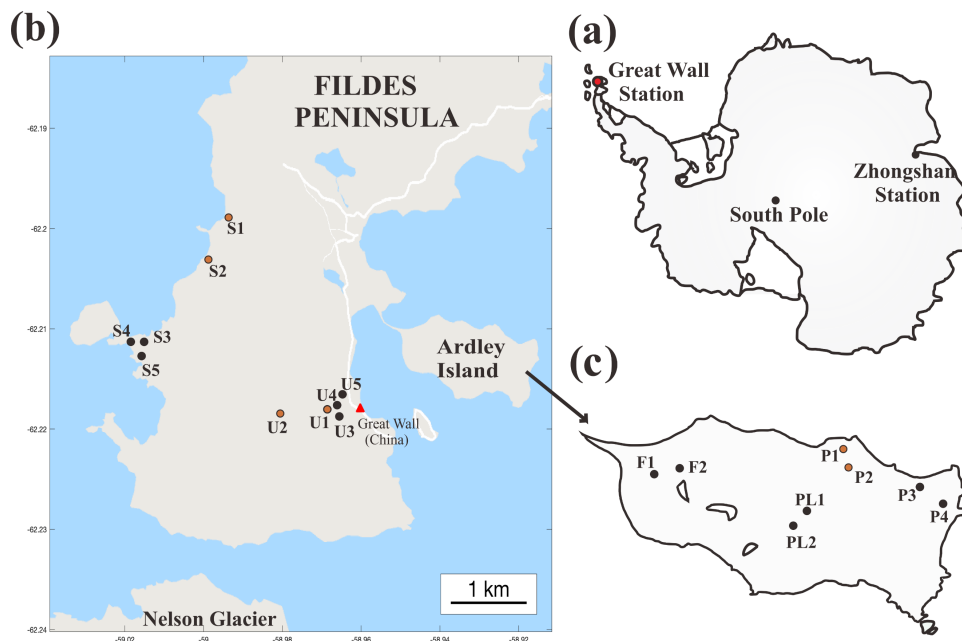
86 In this study, *in situ* static-chamber measurements of  $\text{CHCl}_3$  fluxes were conducted in  
87 Antarctic tundra during the 36<sup>th</sup> Chinese Antarctic Research Expedition (CHINARE-36).  
88 Soil samples were also collected from the tundra on Fildes Peninsula and Ardley Island,  
89 and then incubated in the laboratory (1) under simulated natural conditions to obtain  $\text{CHCl}_3$   
90 fluxes, and to compare with the *in situ* results; (2) in the range of -12 °C to 12 °C to quantify  
91  $\text{CHCl}_3$  emissions under simulated seasonal cycles and projected Antarctic warming; and  
92 (3) under different manipulated conditions (such as  $\text{N}_2$ -anoxic and thermal treatments) to  
93 explore the  $\text{CHCl}_3$  formation pathway in Antarctic tundra.

## 94 **Materials and methods**

### 95 **Study sites**

96 The study area is located on Fildes Peninsula (61°51'S-62°15'S, 57°30'W-59°00'W)  
97 and Ardley Island (62°13'S, 58°56'W) in West Antarctica (Figure 1). The annual average  
98 temperature is about -2.5 °C, in a range of -26.6~11.7 °C, and the annual average  
99 precipitation (mainly in snowfall) is about 630 mm (Zhu et al., 2013). Fildes Peninsula  
100 (Figure 1b) is the largest ice-free area in King George Island, the middle of the peninsula

101 is bedrock-exposed hills, and a well-developed tundra ecosystem is located along the  
102 coastline, with the lichens and mosses as the main vegetation types (Borchhardt et al.,  
103 2017). The western coast of the peninsula is inhabited by a variety of seals (Wang et al.,  
104 2019). The eastern side of Ardley Island is an important habitat for Adélie, gentoo and  
105 chinstrap penguins with a combined population over 10,000 (Zhu et al., 2013) (Figure 1c).  
106 The adjacent penguin-lingering tundra areas are covered by cushions of mosses, lichens,  
107 and algae (Borchhardt et al., 2017), and penguins occasionally wander here. The western  
108 side of the island, which is in the prevailing wind direction, is a lowland tundra marsh as  
109 well as an abandoned former penguin colony (Yang et al., 2019). Each year, large amounts  
110 of excrement are deposited on Ardley Island and leached into tundra soil by snowmelt from  
111 surrounding hills, and seawater is also brought inland by penguins. In general, sea animal  
112 colony tundra soils have higher gravimetric moisture and soil organic matter levels in  
113 comparison to normal tundra soils (soil properties are given in Table S1).



114

115 **Figure 1.** Maps of the study area and sampling sites. (a) Antarctic continent and  
116 the location of the Great Wall Antarctic Research Station (red dot); (b) Sampling  
117 sites on the Fildes Peninsula, including the seal colony tundra SCS (S1-5) and  
118 normal upland tundra UTS (U1-5); (c) Sampling sites on the Ardley Island,  
119 including the western abandoned former penguin colony FPS (F1, F2), the eastern  
120 active penguin colony tundra PCS (P1-4), and the adjacent penguin-lingering

121 tundra PLS (PL1, PL2). The orange and black dots represent *in situ* flux  
122 measurement sites and soil sample collection sites, respectively.

### 123 ***In situ* CHCl<sub>3</sub> flux measurements**

124 During the 36<sup>th</sup> Chinese National Antarctic Research Expedition, *in situ* static-chamber  
125 measurements (Figure S1) for CHCl<sub>3</sub> fluxes were conducted from seal colony tundra (S1,  
126 S2; Histic Cryosols), penguin colony tundra (P1, P2; Ornithogenic Cryosols) and normal  
127 upland tundra (U1, U2; soil combinations of Turbic Cryosols and Hyperskeletal Leptosols)  
128 (Figure 1, orange dots). Air samples within the chamber were drawn into pre-evacuated air  
129 flasks (1-litre; Entech, California, United States) at 1 min, 16 min and 31 min through a  
130 stainless-steel tube extended into the chamber center point. During sampling, the chamber  
131 was vented through a tube to allow for pressure equilibrium and to avoid over-sampling  
132 the soil pore air (additional information on sampling procedure is given in Text S1). Each  
133 location had two replicate static-chamber measurements.

134 CHCl<sub>3</sub> within the flask samples were cryotrapped and analyzed by a gas  
135 chromatograph-mass spectrometer (GC/MS, Agilent 7890B/5973N, Agilent Technologies,  
136 California, United States). The samples were calibrated against a commercial standard  
137 mixture of TO-15 compounds (Linde PLC, New Jersey, United States). The CHCl<sub>3</sub> molar  
138 fraction was reported in units of parts per trillion (p.p.t; the same hereafter). Based on daily  
139 standard measurements, the instrumental precision ( $1\sigma$ ) for CHCl<sub>3</sub> analysis was less than  
140 4%. Details on the cryotrapping procedure, capillary column selection and GC oven  
141 temperature setting are given in Supporting Information (Text S2).

### 142 **Laboratory soil incubation experiment**

143 Soil samples (top 5-10 cm) were collected from Ardley Island at penguin colony (P1-  
144 P4), penguin-lingering area (PL1, PL2), and abandoned former penguin colony (F1, F2).  
145 Soils were also collected on Fildes Peninsula from normal upland tundra (U1-U5) and seal



146 colony (S1-S5) (Figure 1). Each sample (25 grams) had at least two replicates ( $n \geq 2$ )  
147 incubated as described below.

148 **(i) Soil incubation under simulating summertime temperature.** Each soil sample  
149 (U3-U5; S3-S5; F1-F2; PL1-PL2; P3-P4) was enclosed in glass Mason jars, and then  
150 incubated overnight at 4 °C using a water/ethylene glycol bath (VWR Model 1180S,  
151 Pennsylvania, United States) to simulate Antarctic mean summer temperature. A stainless-  
152 steel lid to the glass jar was connected to a pre-evacuated stainless-steel tube ( $V \approx 25$  mL),  
153 which was further connected to the cryo-trapping system (Khan et al., 2012) of a gas  
154 chromatograph/mass spectrometer (GC/MS; Agilent 6890N/5973, Agilent Technologies,  
155 California, United States). The headspace air was drawn, cryofocused, and injected into  
156 the GC/MS at 1 min, 31 min and 61 min after sealing. Weekly calibration curves were  
157 constructed with a natural whole air standard collected and calibrated at the Scripps  
158 Institution of Oceanography, University of California at San Diego (SIO-98 calibration  
159 scale). The same standard was run before and after each batch of incubations to correct for  
160 the daily drift of the mass spectrometer signal.

161 **(ii) Incubation under anoxic condition and post thermal sterilization.** Two sets of  
162 incubation experiments were carried out at 4 °C under anoxic conditions and post thermal  
163 sterilization, respectively, with the soil samples (U3-U5; S3-S5; F1-F2; PL1-PL2; P3-P4)  
164 to explore the potential effects of oxygen levels and soil microorganisms. The anoxic  
165 condition was induced by flushing and filling the jar with ultra-pure nitrogen gas ( $\geq$   
166 99.999%, Praxair, Connecticut, United States); thermal sterilization was conducted by  
167 autoclaving the soil samples at 150 °C for 2 hours. The moisture was brought back by  
168 adding the same amount of distilled water as mass loss. The soil samples were incubated  
169 for  $\text{CHCl}_3$  flux following the same procedure above.

170 **(iii) Temperature-gradient incubation.** Soil samples (U3-U5; S3-S5; P3-P4) were  
171 also incubated along a temperature gradient between -4 and 12 °C at 4 °C intervals, in order  
172 to construct the relationship between CHCl<sub>3</sub> fluxes and soil temperature, and to explore  
173 projected CHCl<sub>3</sub> emissions under Antarctic warming.

174 **(iv) Freeze-thaw cycles (FTCs) incubation.** To simulate Antarctic seasonal  
175 temperature cycles, soil samples (U1-U2; S1-S2, S4; P1-P2, P4) were incubated over three  
176 consecutive FTCs. Within each FTC, soil sample was frozen at -12 °C (mean wintertime  
177 temperature) for 8 hours, and then 20 ml of headspace air was extracted and analyzed every  
178 2 hours for four times. The sample was then thawed at 4 °C (mean summertime temperature)  
179 for 8 hours, after which the headspace air was analyzed following the same procedure.  
180 After each sampling, the jar was flushed with the standard whole air to conserve constant  
181 pressure within the jar.

## 182 **Flux calculation**

183 For CHCl<sub>3</sub> productions, most of the flux measurements showed significant linear  
184 relationship ( $R^2 > 0.70$ ,  $p < 0.05$ ) between CHCl<sub>3</sub> concentrations in chamber/jar headspace  
185 and enclosure time. Therefore, net positive fluxes were calculated using the linear least  
186 squares fit between CHCl<sub>3</sub> molar fraction and time, multiplied by the number moles of air  
187 in the static-chamber or glass jar, and then normalized to the surface area of soil (Jiao et  
188 al., 2018).

$$189 \quad F = \frac{1}{A} \frac{dc}{dt} N_a \quad \dots\dots\dots \text{(Eq. 1)}$$

190 Where F represents the CHCl<sub>3</sub> flux from the soil [ $\text{nmol m}^{-2} \text{d}^{-1}$ ]; A represents soil  
191 surface area; and  $N_a$  is the number moles of air in the chamber or jar. For soil incubations,  
192 CHCl<sub>3</sub> emission rates are normalized to the dry mass of soil, and reported in the unit of

193 pmol g<sup>-1</sup> d<sup>-1</sup>. For linear regressions with low coefficients of determination (r<sup>2</sup><0.70), the  
194 curve fits were not considered robust and the fluxes were treated as zero.

195 CHCl<sub>3</sub> degradation in soil was assumed to follow first-order kinetics to the ambient  
196 CHCl<sub>3</sub> concentration (Khan et al., 2011).

197 
$$\frac{d[c]}{dt} = -k[c] \dots\dots\dots \text{(Eq. 2)}$$

198 Therefore, when uptake of CHCl<sub>3</sub> was observed, a linear least squares fit was applied  
199 to the natural log of headspace CHCl<sub>3</sub> concentration (ln[c]) over incubation time to get the  
200 first-order constant, k, which was then multiplied by background CHCl<sub>3</sub> concentration in  
201 Southern Hemisphere (Prinn et al., 2018) to obtain *in situ* uptake rates.

## 202 **Analysis of soil properties**

203 Soil pH, gravimetric moisture content (Mc), total organic carbon (TOC), total nitrogen  
204 (TN), total phosphorus (TP) and chlorine (Cl) contents were also analyzed. The  
205 methodology is given in Supporting Information (Text S3).

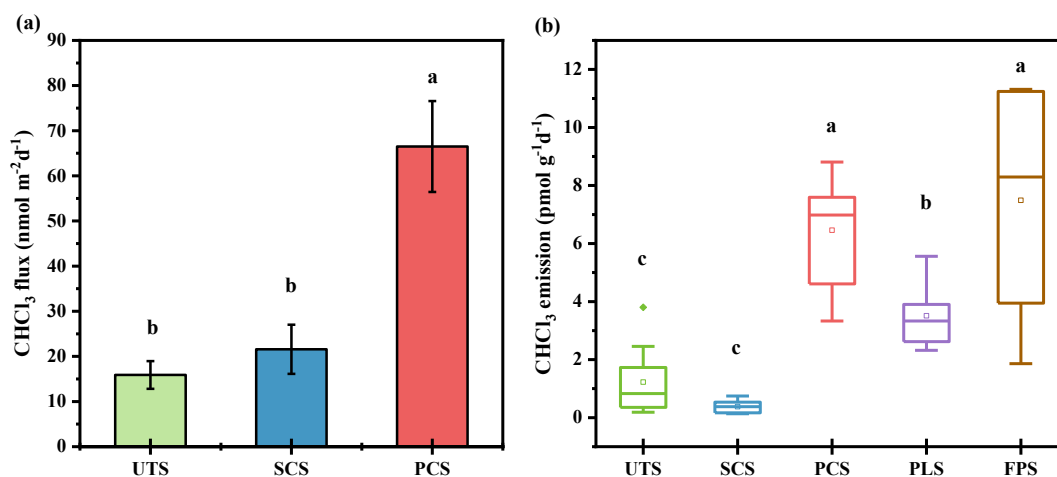
## 206 **Results and discussion**

### 207 **CHCl<sub>3</sub> Emission from Antarctic Tundra**

208 *In situ* static-chamber measurements showed that normal upland tundra (UTS), seal  
209 colony tundra (SCS), and active penguin colony tundra (PCS) were CHCl<sub>3</sub> sources, with  
210 emission rates of 16 ± 6 nmol m<sup>-2</sup> d<sup>-1</sup>, 22 ± 11 nmol m<sup>-2</sup> d<sup>-1</sup>, and 66 ± 20 nmol m<sup>-2</sup> d<sup>-1</sup>,  
211 respectively (Figure 2a). Laboratory-based incubations showed that all Antarctic tundra  
212 soils emitted CHCl<sub>3</sub>, with mean emission rates ranging from 0.4 to 7.5 pmol g<sup>-1</sup> d<sup>-1</sup> (Figure  
213 2b), further supporting the *in situ* finding that Antarctic tundra was a CHCl<sub>3</sub> source.

214 This study provides the first CHCl<sub>3</sub> flux measurements from the Antarctic tundra. The  
215 average CHCl<sub>3</sub> flux (35 ± 27 nmol m<sup>-2</sup> d<sup>-1</sup>) of different types of Antarctic tundra was within

216 the range of those from the Alaskan Arctic tundra ( $45 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Rhew et al., 2008a),  
 217 and Arctic glacial forefield tundra ( $74 \pm 33 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Macdonald et al., 2020); the flux  
 218 from penguin colony tundra ( $66 \pm 20 \text{ nmol m}^{-2} \text{ d}^{-1}$ ), where the highest emission rates were  
 219 observed in this study, was higher than or close to those from Arctic tundra.

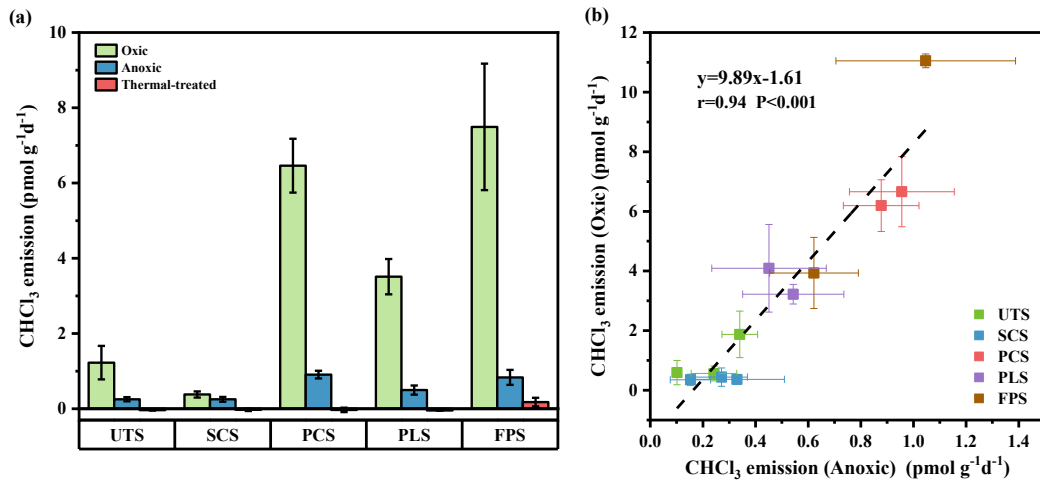


220

221 **Figure 2.** (a) Bar plots of *in situ* measurements on chloroform (CHCl<sub>3</sub>) fluxes  
 222 from normal upland tundra (UTS, n = 4), seal colony tundra (SCS, n = 4), and  
 223 active penguin colony tundra (PCS, n = 4) in coastal Antarctica. (b) Box plots of  
 224 laboratory-based measurements on chloroform (CHCl<sub>3</sub>) emission rates from the  
 225 soils of normal upland tundra (UTS, n = 8), seal colony tundra (SCS, n = 8), active  
 226 penguin colony tundra (PCS, n = 7), penguin-lingering tundra (PLS, n = 6) and  
 227 abandoned former penguin colony tundra (FPS, n = 6). The different lowercase  
 228 letters (a, b, or c) indicate significant differences between the means (LSD, p <  
 229 0.05).

230 The average CHCl<sub>3</sub> fluxes from Antarctic tundra were an order of magnitude smaller  
 231 than those from several mid- and low- latitudes ecosystems, such as temperate peatland  
 232 ( $258 \pm 288 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Khan et al., 2012), pasture ( $241 \pm 221 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Cox et al.,  
 233 2004), subtropical degraded forested wetland ( $209 \pm 183 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Jiao et al., 2018;  
 234 Wang et al., 2016), and tropical rice paddies ( $201 \text{ nmol m}^{-2} \text{ d}^{-1}$ ) (Khalil et al., 1998). These  
 235 results from such diverse ecosystems (Table S2) suggest that soils globally are important  
 236 and widespread sources for atmospheric CHCl<sub>3</sub>, even in coastal Antarctica.

237 **Exploring CHCl<sub>3</sub> production mechanism**



238

239 **Figure 3.** (a) Bar plots of chloroform (CHCl<sub>3</sub>) emission rates from normal upland  
 240 tundra (UTS), seal colony tundra (SCS), active penguin colony tundra (PCS),  
 241 penguin-lingering tundra (PLS) and former penguin colony (FPS) based upon  
 242 laboratory incubations under simulated natural conditions, N<sub>2</sub>-anoxic conditions  
 243 and post thermal treatment. (b) Linear correlation between chloroform (CHCl<sub>3</sub>)  
 244 emission rates from coastal Antarctic tundra soils under ambient oxidic conditions  
 245 and N<sub>2</sub>-anoxic conditions.

246 In this study, soil CHCl<sub>3</sub> emission rates significantly declined to near zero after thermal  
 247 sterilization (Figure 3a). A number of studies have shown that CHCl<sub>3</sub> formation within soil  
 248 may be primarily of biotic origins (Bastviken et al., 2009; Breider & Hunkeler, 2014a;  
 249 Macdonald et al., 2020; Ruecker et al., 2014, 2015). Chloroperoxidases (CPOs), commonly  
 250 found in soil fungi, such as *Caldariomyces fumago* and *Peniophora pseudopini*, are capable  
 251 to produce hypochlorous acid (HOCl), which can chlorinate soil organic matter to produce  
 252 intermediate organohalogenes (such as trichloroacetyl-containing compounds), and  
 253 subsequently cleave to yield CHCl<sub>3</sub> (Albers et al., 2017; Breider & Hunkeler, 2014a, 2014b;  
 254 Wever & Barnett, 2017). The enzymatic activities of various CPOs are all suppressed at  
 255 high temperatures. For example, the activity of iron-CPO in *Caldariomyces fumago* is  
 256 reduced to almost 0% after a 15-minute incubation at 60 °C (Pickard & Hashimoto, 1988).  
 257 Thus, the dramatic reduction of emissions post thermal sterilization suggests that the  
 258 observed CHCl<sub>3</sub> production in Antarctica tundra soils was predominantly by enzymatic or  
 259 microbial activities. However, on the other hand, thermal sterilization may have altered the

260 chemical composition of soil organic matter to some extent, such as by decomposing or  
261 volatilizing labile carbon compounds, and inhibited  $\text{CHCl}_3$  production potential of abiotic  
262 process which is usually catalyzed by  $\text{Fe}^{3+}$  (Huber et al., 2009).

263 The optimal pH for chlorinating activities of CPOs was around 3-4 (Asplund et al.,  
264 1993; van Schijndel et al., 1994), lower than the soil pH of Antarctic tundra (Table S1).  
265 Consistent with this, an inverse proportional relationship was observed between  $\text{CHCl}_3$   
266 fluxes and soil pH across different types of Antarctic tundra ( $r = -0.72$ ,  $p = 0.008$ , Figure  
267 S2, S3).

268  $\text{CHCl}_3$  emission rates from Antarctic tundra soils under anoxic conditions ( $0.5 \pm 0.4$   
269  $\text{pmol g}^{-1} \text{d}^{-1}$ ) declined to about 14.3 % of that under oxic conditions ( $3.5 \pm 3.4 \text{ pmol g}^{-1} \text{d}^{-1}$ ), but still conserved a strong linear correlation ( $r = 0.94$ ,  $p < 0.001$ ) with those under  
270 ambient conditions (Figure 3), suggesting that the production of  $\text{CHCl}_3$  can occur both  
271 aerobically and anaerobically, although the aerobic production mechanism dominates.  
272 Most of the CPOs are heme-dependent, or prosthetic groups of transition metals are in  
273 oxidized states, such as iron (III) or vanadium (V) (Breider & Hunkeler, 2014a; Wever &  
274 Barnett, 2017). An oxic environment is essential to keep these transition metals in  
275 oxidation states for soil  $\text{CHCl}_3$  production. Laboratory incubations of soil samples from a  
276 coniferous forest confirmed that  $\text{O}_2$  is required in the dominant CPO-mediated chlorination  
277 process (Bastviken et al., 2009). The anaerobic environment may have reduced the  
278 oxidation states of transition metals, and subsequently suppressed the enzymatic pathway  
279 of  $\text{CHCl}_3$  formation in Antarctic tundra soils.  
280

### 281 **Effects of penguin activities on tundra $\text{CHCl}_3$ emission**

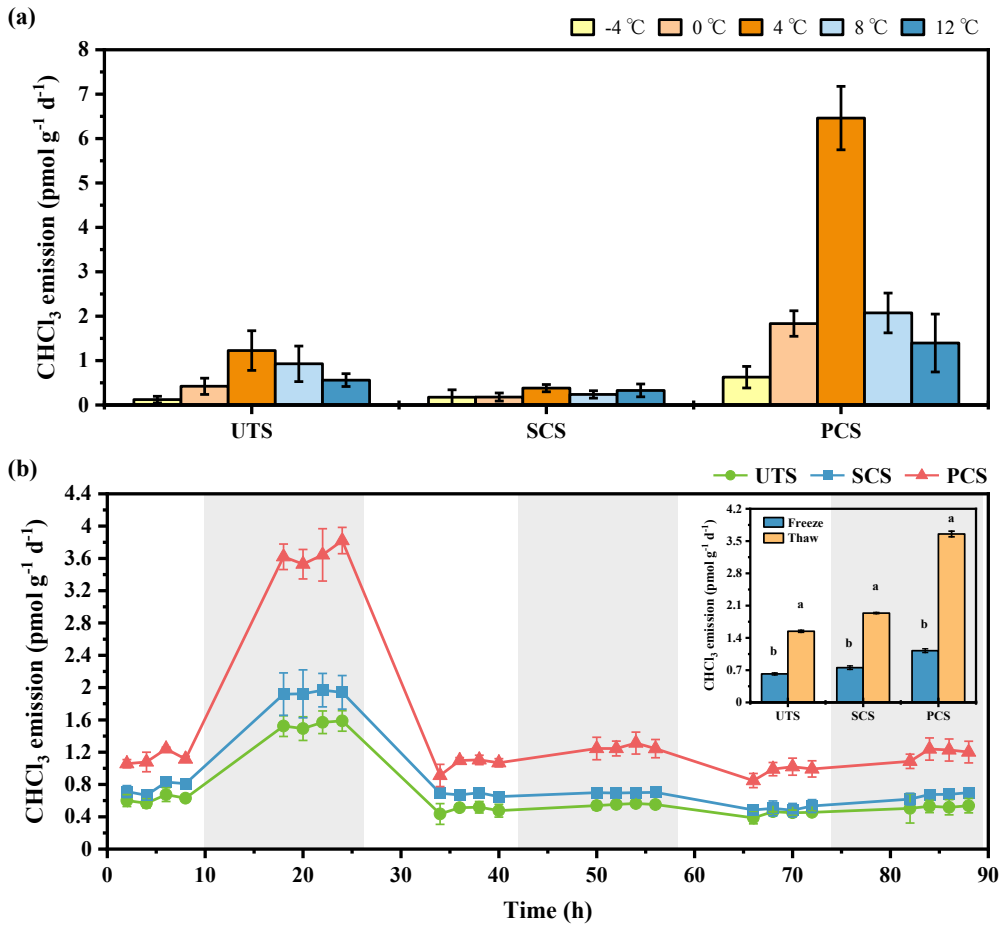
282 *In situ*  $\text{CHCl}_3$  fluxes showed that the average fluxes from penguin colony tundra were  
283 more than 3-4 times higher than those from seal colony tundra and normal upland tundra  
284 (Figure 2a). For lab-based incubations, the highest emission rate was observed from the

285 abandoned former penguin colony (FPS, mean  $7.5 \pm 4.1 \text{ pmol g}^{-1} \text{ d}^{-1}$ ) and the active  
286 penguin colony (PCS, mean  $6.5 \pm 1.9 \text{ pmol g}^{-1} \text{ d}^{-1}$ ), followed by the adjacent penguin-  
287 lingering tundra (PLS, mean  $3.5 \pm 1.2 \text{ pmol g}^{-1} \text{ d}^{-1}$ ). These values were about 3-6 times  
288 higher than normal upland tundra (UTS) ( $1.2 \pm 1.3 \text{ pmol g}^{-1} \text{ d}^{-1}$ ). These results indicated  
289 that penguin activities may have increased the  $\text{CHCl}_3$  emission from tundra soils.

290 Microorganisms, which are responsible for  $\text{CHCl}_3$  production, are often unevenly  
291 distributed and can lead to differences in  $\text{CHCl}_3$  forming activity (Albers et al., 2011).  
292 Ornithogenic soil on Ardley Island affected by penguin activities was more abundant in  
293 organic carbon (TOC), nitrogen (TN) and phosphorus (TP) than other tundra soils (Table  
294 S1), which were major factors influencing microbial populations in Antarctic soil  
295 (Tscherko et al., 2003; Wang et al., 2019; Zhu et al., 2015). Fungi abundance analysis of  
296 the Antarctic tundra soil samples showed that the majority of the microorganisms was in  
297 the phylum of *Ascomycota* and *Basidiomycota*, to which most of CPOs-related fungi  
298 belonged (Durán et al., 2019). Therefore, the higher  $\text{CHCl}_3$  emission rates from penguin-  
299 related tundra may be attributed to the colonization of CPO-related microorganisms in this  
300 organic matter rich soil.

301  $\text{CHCl}_3$  fluxes across all the types of tundra soils also showed significant correlations ( $p$   
302  $< 0.05$ ) with TOC, TP, Cl, and Mc (Figure S2, S3). Compared to seal colony soil and  
303 normal upland tundra soil, ornithogenic soil within the penguin colony and its adjacent  
304 tundra were much richer in organic carbon (Table S1), which can provide organic  
305 substrates for CPOs-catalyzed  $\text{CHCl}_3$  production. The high frequency transit of penguins  
306 between the sea and tundra brings seawater inland, and the sea breeze also blew in abundant  
307 salt aerosols onto Ardley Island (especially the western side), which increased soil chloride  
308 and moisture. All these factors induced by penguin activities and wind patterns potentially  
309 led to the enhanced  $\text{CHCl}_3$  emissions from tundra populated with penguins.

310 **Temperature effects on CHCl<sub>3</sub> emission**



311

312 **Figure 4.** Chloroform (CHCl<sub>3</sub>) emission rates based on laboratory soil incubations  
 313 **(a)** at different temperatures; and **(b)** during three consecutive freeze-thaw cycles  
 314 (FTCs) between -12 °C and 4 °C. The inserted plot showed the comparison of  
 315 CHCl<sub>3</sub> emission rates in the first FTC. The incubated samples were collected from  
 316 normal upland tundra (UTS), seal colony tundra (SCS), and active penguin colony  
 317 tundra (PCS).

318 The temperature-gradient incubation experiments showed that Antarctic tundra soils  
 319 were a source for CHCl<sub>3</sub> across the entire temperature range of -4 ~ 12 °C (Figure 4a). All  
 320 tundra soils showed maximum production rates of CHCl<sub>3</sub> at 4 °C (close to local mean air  
 321 temperature in summer); mean CHCl<sub>3</sub> emission rates decreased to close to zero at -4 °C  
 322 and increased significantly when temperature rose above zero. Compared to 4 °C, CHCl<sub>3</sub>  
 323 emission rates from tundra soils decreased at 8 °C and 12 °C, but they were still higher  
 324 than those at -4 °C. The Antarctic region is highly sensitive to climate change (Bockheim  
 325 et al., 2013; Schofield et al., 2010), and Western Antarctica, especially the Antarctic



326 Peninsula, has been experiencing the highest temperature increase (Bromwich et al., 2013;  
327 Vaughan et al., 2003). With additional warming, more days above freezing can be expected,  
328 which would influence the annual pattern of CHCl<sub>3</sub> emissions from Antarctic tundra.

329 Freeze-thaw cycles (FTCs) incubations showed that CHCl<sub>3</sub> emission rates were 3.3  
330 times higher during thaw than during freeze under the first FTC for PCS, while for UTS  
331 and SCS, the emission rates were about 2.5 times higher (Table S3). As permafrost thaws,  
332 the increase of soil moisture and temperature can enhanced soil microbial activity and  
333 organic material release for microorganisms to access (Ludwig et al., 2006; Schimel &  
334 Clein, 1996). However, CHCl<sub>3</sub> emission rates did not significantly differ between the  
335 freeze and thaw incubations in the two following FTCs, with rates remaining low, similar  
336 to those during the first freeze period, but CHCl<sub>3</sub> emission rates from PCS were  
337 consistently higher than those from SCS and UTS (Figure 4b, Table S3). The lack of  
338 significant CHCl<sub>3</sub> emission rates in the two following FTCs is not necessarily predicted. It  
339 might be attributed to the depletion of substrates in the soil, such as chloride ions, iron (III),  
340 etc. More likely, the inhibited CHCl<sub>3</sub> emission rates in the following cycles may be  
341 attributed to a rapid FTC-induced reduction of microbial populations. A single rapid FTC  
342 can kill up to half of the soil microorganisms, leading to either reduced decomposition rates  
343 or to immobilization of nutrients tied up in the microbial biomass (Larsen et al., 2002;  
344 Schimel & Clein, 1996). Hence, the seasonal pattern of CHCl<sub>3</sub> emission needs to be  
345 considered when quantifying the annual fluxes.

## 346 **Regional Assessment**

347 Given the spatial and temporal variabilities of CHCl<sub>3</sub> fluxes and the limited number of  
348 measurements, the attempt to extrapolate to the continental scale would introduce  
349 significant uncertainties. More *in situ* studies are needed to capture diurnal, seasonal, and  
350 spatial variations of CHCl<sub>3</sub> emissions. However, since no prior fluxes have been measured

351 from Antarctic environment, a simple quantitative upscaling of these first-reported CHCl<sub>3</sub>  
352 fluxes is still worthwhile in assessing its influence on regional atmospheric compositions  
353 and on global CHCl<sub>3</sub> inventory.

354 Assuming the CHCl<sub>3</sub> fluxes measured from this study are representative of the  $7.2 \times$   
355  $10^{10}$  m<sup>2</sup> ice-free tundra in Antarctic continent (Lee et al., 2017; Terauds & Lee, 2016),  
356 among which 30% is colonized or influenced by penguins or other sea animals (Zhu et al.,  
357 2008), during the austral summertime (assuming an active period of 100 days), and CHCl<sub>3</sub>  
358 flux in winter (the other 265 days) accounts for one-third of that based upon the FTCs  
359 experiments, it is estimated that Antarctic tundra emits  $0.08 \pm 0.02$  Gg of CHCl<sub>3</sub> per year  
360 naturally, which constitutes a small fraction (< 1%) of global CHCl<sub>3</sub> terrestrial sources.  
361 The size of this CHCl<sub>3</sub> source is minor compared to other sources, largely due to the  
362 relatively small area of ice-free tundra in the Antarctica. Therefore, it may not exert a  
363 significant impact on global atmospheric CHCl<sub>3</sub> concentrations, such as mid-/low-  
364 latitudes and the other hemisphere. However, it may have a relatively greater influence on  
365 local CHCl<sub>3</sub> concentrations, especially given the relatively small portion of the troposphere  
366 in the Antarctic region. On the other hand, the rapid warming of the Antarctic continent is  
367 expanding the area of ice-free tundra and influencing the size of penguin populations and  
368 colonies (Barbraud et al., 2020; Lee et al., 2017; Lyver et al., 2014). Hence, the relative  
369 contribution to atmospheric CHCl<sub>3</sub> from tundra ecosystems will vary in response to these  
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379 The data that supports the discussion and conclusion is all presented in manuscript and  
380 Supporting Information. A copy of the data is archived on Zenodo and available to  
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## 382 **References**

- 383 Albers, C. N., Jacobsen, O. S., Flores, É. M. M., Pereira, J. S. F., & Laier, T. (2011).  
384 Spatial variation in natural formation of chloroform in the soils of four coniferous  
385 forests. *Biogeochemistry*, *103*(1–3), 317–334. [https://doi.org/10.1007/s10533-](https://doi.org/10.1007/s10533-010-9467-9)  
386 [010-9467-9](https://doi.org/10.1007/s10533-010-9467-9)
- 387 Albers, C. N., Jacobsen, O. S., Flores, E. M. M., & Johnsen, A. R. (2017). Arctic and  
388 subarctic natural soils emit chloroform and brominated analogues by alkaline  
389 hydrolysis of trihaloacetyl compounds. *Environmental Science & Technology*,  
390 *51*(11), 6131–6138. <https://doi.org/10.1021/acs.est.7b00144>
- 391 Asplund, G., Christiansen, J. V., & Grimvall, A. (1993). A chloroperoxidase-like catalyst  
392 in soil: Detection and characterization of some properties. *Soil Biology and*  
393 *Biochemistry*, *25*(1), 41–46. [https://doi.org/10.1016/0038-0717\(93\)90239-8](https://doi.org/10.1016/0038-0717(93)90239-8)
- 394 Barbraud, C., Delord, K., Bost, C. A., Chaigne, A., Marteau, C., & Weimerskirch, H.  
395 (2020). Population trends of penguins in the French Southern Territories. *Polar*  
396 *Biology*, *43*(7), 835–850. <https://doi.org/10.1007/s00300-020-02691-6>
- 397 Bastviken, D., Svensson, T., Karlsson, S., Sandén, P., & Öberg, G. (2009). Temperature  
398 sensitivity indicates that chlorination of organic matter in forest soil is primarily  
399 biotic. *Environmental Science & Technology*, *43*(10), 3569–3573.  
400 <https://doi.org/10.1021/es8035779>

401 Bockheim, J., Vieira, G., Ramos, M., López-Martínez, J., Serrano, E., Guglielmin, M., et  
402 al. (2013). Climate warming and permafrost dynamics in the Antarctic Peninsula  
403 region. *Global and Planetary Change*, *100*, 215–223.  
404 <https://doi.org/10.1016/j.gloplacha.2012.10.018>

405 Borchhardt, N., Schiefelbein, U., Abarca, N., Boy, J., Mikhailyuk, T., Sipman, H. J. M.,  
406 & Karsten, U. (2017). Diversity of algae and lichens in biological soil crusts of  
407 Ardley and King George islands, Antarctica. *Antarctic Science*, *29*(3), 229–237.  
408 <https://doi.org/10.1017/S0954102016000638>

409 Breider, F., & Hunkeler, D. (2014a). Investigating chloroperoxidase-catalyzed formation  
410 of chloroform from humic substances using stable chlorine isotope analysis.  
411 *Environmental Science & Technology*, *48*(3), 1592–1600.  
412 <https://doi.org/10.1021/es403879e>

413 Breider, F., & Hunkeler, D. (2014b). Mechanistic insights into the formation of  
414 chloroform from natural organic matter using stable carbon isotope analysis.  
415 *Geochimica et Cosmochimica Acta*, *125*, 85–95.  
416 <https://doi.org/10.1016/j.gca.2013.09.028>

417 Bromwich, D. H., Nicolas, J. P., Monaghan, A. J., Lazzara, M. A., Keller, L. M.,  
418 Weidner, G. A., & Wilson, A. B. (2013). Central West Antarctica among the most  
419 rapidly warming regions on Earth. *Nature Geoscience*, *6*(2), 139–145.  
420 <https://doi.org/10.1038/ngeo1671>

421 Cox, M. L., Fraser, P. J., Sturrock, G. A., Siems, S. T., & Porter, L. W. (2004). Terrestrial  
422 sources and sinks of halomethanes near Cape Grim, Tasmania. *Atmospheric*  
423 *Environment*, *38*(23), 3839–3852. <https://doi.org/10.1016/j.atmosenv.2004.03.050>

424 Croxall, J. P., Trathan, P. N., & Murphy, E. J. (2002). Environmental Change and  
425 Antarctic Seabird Populations. *Science*, *297*(5586), 1510–1514.  
426 <https://doi.org/10.1126/science.1071987>

427 Durán, P., Barra, P. J., Jorquera, M. A., Viscardi, S., Fernandez, C., Paz, C., et al. (2019).  
428 Occurrence of soil fungi in Antarctic pristine environments. *Frontiers in*  
429 *Bioengineering and Biotechnology*, *7*. <https://doi.org/10.3389/fbioe.2019.00028>

430 Engel, A., Rigby, M., Burkholder, J. B., Fernandez, R. P., Froidevaux, L., Hall, B. D., et  
431 al. (2018). *Update on ozone-depleting substances (ODSs) and other gases of*  
432 *interest to the Montreal Protocol* (Chapter 1 in Scientific Assessment of Ozone  
433 Depletion: 2018, Global Ozone Research and Monitoring Project No. Report No.  
434 58). Geneva, Switzerland: World Meteorological Organization. Retrieved from  
435 <https://www.esrl.noaa.gov/csd/assessments/ozone/2018/citations.html>

436 Fang, X., Park, S., Saito, T., Tunnicliffe, R., Ganesan, A. L., Rigby, M., et al. (2019).  
437 Rapid increase in ozone-depleting chloroform emissions from China. *Nature*  
438 *Geoscience*, *12*(2), 89. <https://doi.org/10.1038/s41561-018-0278-2>

439 Jiao, Y., Ruecker, A., Deventer, M. J., Chow, A. T., & Rhew, R. C. (2018). Halocarbon  
440 emissions from a degraded forested wetland in coastal South Carolina impacted  
441 by sea level rise. *ACS Earth and Space Chemistry*, *2*(10), 955–967.  
442 <https://doi.org/10.1021/acsearthspacechem.8b00044>

443 Johnsen, A. R., Jacobsen, O. S., Gudmundsson, L., & Albers, C. N. (2016). Chloroform  
444 emissions from arctic and subarctic ecosystems in Greenland and Northern  
445 Scandinavia. *Biogeochemistry*, *130*(1–2), 53–65. [https://doi.org/10.1007/s10533-](https://doi.org/10.1007/s10533-016-0241-5)  
446 [016-0241-5](https://doi.org/10.1007/s10533-016-0241-5)

447 Huber, S. G., Kotte, K., Schöler, H. F., & Williams, J. (2009). Natural abiotic formation  
448 of trihalomethanes in soil: results from laboratory studies and field samples.

449 *Environmental Science & Technology*, 43(13), 4934-4939.  
450 <https://doi.org/10.1021/es8032605>

451 Khalil, M. A. K., Rasmussen, R. A., Shearer, M. J., Chen, Z.-L., Yao, H., & Yang, J.  
452 (1998). Emissions of methane, nitrous oxide, and other trace gases from rice  
453 fields in China. *Journal of Geophysical Research: Atmospheres*, 103(D19),  
454 25241–25250. <https://doi.org/10.1029/98JD01114>

455 Khan, M. A. H., Rhew, R. C., Whelan, M. E., Zhou, K., & Deverel, S. J. (2011). Methyl  
456 halide and chloroform emissions from a subsiding Sacramento–San Joaquin Delta  
457 island converted to rice fields. *Atmospheric Environment*, 45(4), 977–985.  
458 <https://doi.org/10.1016/j.atmosenv.2010.10.053>

459 Khan, M. A. H., Whelan, M. E., & Rhew, R. C. (2012). Effects of temperature and soil  
460 moisture on methyl halide and chloroform fluxes from drained peatland pasture  
461 soils. *Journal of Environmental Monitoring*, 14(1), 241–249.  
462 <https://doi.org/10.1039/C1EM10639B>

463 Larsen, K. S., Jonasson, S., & Michelsen, A. (2002). Repeated freeze–thaw cycles and  
464 their effects on biological processes in two arctic ecosystem types. *Applied Soil  
465 Ecology*, 21(3), 187–195. [https://doi.org/10.1016/S0929-1393\(02\)00093-8](https://doi.org/10.1016/S0929-1393(02)00093-8)

466 Lee, J. R., Raymond, B., Bracegirdle, T. J., Chadès, I., Fuller, R. A., Shaw, J. D., &  
467 Terauds, A. (2017). Climate change drives expansion of Antarctic ice-free habitat.  
468 *Nature*, 547(7661), 49–54. <https://doi.org/10.1038/nature22996>

469 Ludwig, B., Teepe, R., Lopes de Gerenyu, V., & Flessa, H. (2006). CO<sub>2</sub> and N<sub>2</sub>O  
470 emissions from gleyic soils in the Russian tundra and a German forest during  
471 freeze–thaw periods—a microcosm study. *Soil Biology and Biochemistry*, 38(12),  
472 3516–3519. <https://doi.org/10.1016/j.soilbio.2006.06.006>

473 Lyver, P. O., Barron, M., Barton, K. J., Ainley, D. G., Pollard, A., Gordon, S., et al.  
474 (2014). Trends in the breeding population of Adélie penguins in the Ross Sea,  
475 1981–2012: a coincidence of climate and resource extraction effects. *PLOS ONE*,  
476 9(3), e91188. <https://doi.org/10.1371/journal.pone.0091188>

477 Macdonald, M. L., Wadham, J. L., Young, D., Lunder, C. R., Hermansen, O., Lamarche-  
478 Gagnon, G., & O’Doherty, S. (2020). Consumption of CH<sub>3</sub>Cl, CH<sub>3</sub>Br, and CH<sub>3</sub>I  
479 and emission of CHCl<sub>3</sub>, CHBr<sub>3</sub>, and CH<sub>2</sub>Br<sub>2</sub> from the forefield of a retreating  
480 Arctic glacier. *Atmospheric Chemistry and Physics*, 20(12), 7243–7258.  
481 <https://doi.org/10.5194/acp-20-7243-2020>

482 McCulloch, A. (2003). Chloroform in the environment: occurrence, sources, sinks and  
483 effects. *Chemosphere*, 50(10), 1291–1308. [https://doi.org/10.1016/S0045-](https://doi.org/10.1016/S0045-6535(02)00697-5)  
484 [6535\(02\)00697-5](https://doi.org/10.1016/S0045-6535(02)00697-5)

485 Michel, R. F. M., Schaefer, C. E. G. R., Dias, L. E., Simas, F. N. B., Benites, V. de M., &  
486 Mendonça, E. de S. (2006). Ornithogenic gelsols (cryosols) from maritime  
487 Antarctica. *Soil Science Society of America Journal*, 70(4), 1370–1376.  
488 <https://doi.org/10.2136/sssaj2005.0178>

489 Pickard, M. A., & Hashimoto, A. (1988). Stability and carbohydrate composition of  
490 chloroperoxidase from *Caldariomyces fumago* grown in a fructose–salts medium.  
491 *Canadian Journal of Microbiology*, 34(8), 998–1002.  
492 <https://doi.org/10.1139/m88-175>

493 Prinn, R. G., Weiss, R. F., Arduini, J., Arnold, T., DeWitt, H. L., Fraser, P. J., et al.  
494 (2018). History of chemically and radiatively important atmospheric gases from  
495 the Advanced Global Atmospheric Gases Experiment (AGAGE). *Earth System*  
496 *Science Data*, 10(2), 985–1018. <https://doi.org/10.5194/essd-10-985-2018>

497 Rhew, R. C., Teh, Y. A., Abel, T., Atwood, A., & Mazéas, O. (2008a). Chloroform  
498 emissions from the Alaskan Arctic tundra. *Geophysical Research Letters*, 35(21).  
499 <https://doi.org/10.1029/2008GL035762>

500 Rhew, R. C., Miller, B. R., & Weiss, R. F. (2008b). Chloroform, carbon tetrachloride and  
501 methyl chloroform fluxes in southern California ecosystems. *Atmospheric*  
502 *Environment*, 42(30), 7135–7140. <https://doi.org/10.1016/j.atmosenv.2008.05.038>

503 Ruecker, A., Weigold, P., Behrens, S., Jochmann, M., Laaks, J., & Kappler, A. (2014).  
504 Predominance of biotic over abiotic formation of halogenated hydrocarbons in  
505 hypersaline sediments in Western Australia. *Environmental Science &*  
506 *Technology*, 48(16), 9170–9178. <https://doi.org/10.1021/es501810g>

507 Ruecker, A., Weigold, P., Behrens, S., Jochmann, M., Barajas, X. L. O., & Kappler, A.  
508 (2015). Halogenated hydrocarbon formation in a moderately acidic salt lake in  
509 Western Australia – role of abiotic and biotic processes. *Environmental*  
510 *Chemistry*, 12(4), 406–414. <https://doi.org/10.1071/EN14202>

511 Schaefer, C. E. G. R., Simas, F. N. B., Gilkes, R. J., Mathison, C., da Costa, L. M., &  
512 Albuquerque, M. A. (2008). Micromorphology and microchemistry of selected  
513 Cryosols from maritime Antarctica. *Geoderma*, 144(1), 104–115.  
514 <https://doi.org/10.1016/j.geoderma.2007.10.018>

515 Schimel, J. P., & Clein, J. S. (1996). Microbial response to freeze-thaw cycles in tundra  
516 and taiga soils. *Soil Biology and Biochemistry*, 28(8), 1061–1066.  
517 [https://doi.org/10.1016/0038-0717\(96\)00083-1](https://doi.org/10.1016/0038-0717(96)00083-1)

518 Schofield, O., Ducklow, H. W., Martinson, D. G., Meredith, M. P., Moline, M. A., &  
519 Fraser, W. R. (2010). How do polar marine ecosystems respond to rapid climate  
520 change? *Science*, 328(5985), 1520–1523. <https://doi.org/10.1126/science.1185779>



521 Sun, L., Xie, Z., & Zhao, J. (2000). A 3,000-year record of penguin populations. *Nature*,  
522 407(6806), 858. <https://doi.org/10.1038/35038163>

523 Terauds, A., & Lee, J. R. (2016). Antarctic biogeography revisited: updating the  
524 Antarctic conservation biogeographic regions. *Diversity and Distributions*, 22(8),  
525 836–840. <https://doi.org/10.1111/ddi.12453>

526 Tscherko, D., Bölter, M., Beyer, L., Chen, J., Elster, J., Kandeler, E., et al. (2003).  
527 Biomass and enzyme activity of two soil transects at King George Island,  
528 maritime Antarctica. *Arctic, Antarctic, and Alpine Research*, 35(1), 34–47.  
529 [https://doi.org/10.1657/1523-0430\(2003\)035\[0034:BAEAOT\]2.0.CO;2](https://doi.org/10.1657/1523-0430(2003)035[0034:BAEAOT]2.0.CO;2)

530 Van Schijndel, J. W., Barnett, P., Roelse, J., Vollenbroek, E. G., & Wever, R. (1994).  
531 The stability and steady-state kinetics of vanadium chloroperoxidase from the  
532 fungus *Curvularia inaequalis*. *European Journal of Biochemistry*, 225(1), 151–  
533 157. <https://doi.org/10.1111/j.1432-1033.1994.00151.x>

534 Vaughan, D. G., Marshall, G. J., Connolley, W. M., Parkinson, C., Mulvaney, R.,  
535 Hodgson, D. A., et al. (2003). Recent Rapid Regional Climate Warming on the  
536 Antarctic Peninsula. *Climatic Change*, 60(3), 243–274.  
537 <https://doi.org/10.1023/A:1026021217991>

538 Wang, J.-J., Jiao, Y., Rhew, R. C., & Chow, A. T. (2016). Haloform formation in coastal  
539 wetlands along a salinity gradient at South Carolina, United States. *Environmental*  
540 *Chemistry*, 13(4), 745–756. <https://doi.org/10.1071/EN15145>

541 Wang, Q., Zhu, R., Zheng, Y., Bao, T., & Hou, L. (2019). Effects of sea animal  
542 colonization on the coupling between dynamics and activity of soil ammonia-  
543 oxidizing bacteria and archaea in maritime Antarctica. *Biogeosciences*, 16(20),  
544 4113–4128. <https://doi.org/10.5194/bg-16-4113-2019>

545 Wever, R., & Barnett, P. (2017). Vanadium chloroperoxidases: the missing link in the  
546 formation of chlorinated compounds and chloroform in the terrestrial  
547 environment? *Chemistry – An Asian Journal*, 12(16), 1997–2007.  
548 <https://doi.org/10.1002/asia.201700420>

549 Worton, D. R., Sturges, W. T., Schwander, J., Mulvaney, R., Barnola, J.-M., &  
550 Chappellaz, J. (2006). 20th century trends and budget implications of chloroform  
551 and related tri- and dihalomethanes inferred from firn air. *Atmospheric Chemistry  
552 and Physics*, 6(10), 2847–2863. <https://doi.org/10.5194/acp-6-2847-2006>

553 Yang, L., Gao, Y., Sun, L., Xie, Z., Yang, W., Chu, Z., et al. (2019). Enhanced westerlies  
554 drove penguin movement at 1000 yr BP on Ardley Island, west Antarctic  
555 Peninsula. *Quaternary Science Reviews*, 214, 44–53.  
556 <https://doi.org/10.1016/j.quascirev.2019.04.026>

557 Zdanowski, M. K., Zmuda, M. J., & Zwolska, I. (2005). Bacterial role in the  
558 decomposition of marine-derived material (penguin guano) in the terrestrial  
559 maritime Antarctic. *Soil Biology and Biochemistry*, 37(3), 581–595.  
560 <https://doi.org/10.1016/j.soilbio.2004.08.020>

561 Zhu, R., Kong, D., Sun, L., Geng, J., Wang, X., & Glindemann, D. (2006). Tropospheric  
562 phosphine and its sources in coastal Antarctica. *Environmental Science &  
563 Technology*, 40(24), 7656–7661. <https://doi.org/10.1021/es061601e>

564 Zhu, R., Liu, Y., Xu, H., Ma, J., Zhao, S., & Sun, L. (2008). Nitrous oxide emissions  
565 from sea animal colonies in the maritime Antarctic. *Geophysical Research  
566 Letters*, 35(9). <https://doi.org/10.1029/2007GL032541>

567 Zhu, R., Liu, Y., Ma, E., Sun, J., Xu, H., & Sun, L. (2009). Greenhouse gas emissions  
568 from penguin guanos and ornithogenic soils in coastal Antarctica: Effects of

569 freezing–thawing cycles. *Atmospheric Environment*, 43(14), 2336–2347.  
570 <https://doi.org/10.1016/j.atmosenv.2009.01.027>

571 Zhu, R., Liu, Y., Xu, H., Ma, D., & Jiang, S. (2013). Marine animals significantly  
572 increase tundra N<sub>2</sub>O and CH<sub>4</sub> emissions in maritime Antarctica. *Journal of*  
573 *Geophysical Research: Biogeosciences*, 118(4), 1773–1792.  
574 <https://doi.org/10.1002/2013JG002398>

575 Zhu, R., Shi, Y., Ma, D., Wang, C., Xu, H., & Chu, H. (2015). Bacterial diversity is  
576 strongly associated with historical penguin activity in an Antarctic lake sediment  
577 profile. *Scientific Reports*, 5(1), 17231. <https://doi.org/10.1038/srep17231>