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Embracing a new paradigm for temperature sensitivity of soil microbes

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- 23

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#### 26 Abstract

The temperature sensitivity of soil processes is of major interest, especially 27 28 in light of climate change. Originally formulated to explain the temperature 29 dependence of chemical reactions, the Arrhenius equation, and related  $Q_{10}$ 30 temperature coefficient, has a long history of application to soil biological 31 processes. However, empirical data indicate that Q<sub>10</sub> and Arrhenius model 32 are often poor metrics of temperature sensitivity in soils. In this opinion 33 piece, we aim to (1) review alternative approaches for characterizing 34 temperature sensitivity, focusing on Macromolecular Rate Theory (MMRT), 35 (2) provide strategies and tools for implementing a new temperature 36 sensitivity framework, (3) develop thermal adaptation hypotheses for the 37 MMRT framework, and (4) explore new questions and opportunities 38 stemming from this paradigm shift. Microbial ecologists should consider 39 developing and adopting MMRT as the basis for predicting biological rates as 40 a function of temperature. Improved understanding of temperature 41 sensitivity in soils is particularly pertinent as microbial response to 42 temperature has a large impact on global climate feedbacks.

43

#### 44 **1. Introduction**

45 The temperature sensitivity of soil biological processes under climate 46 change is of major interest because of the major consequences for soil 47 carbon dynamics (Bradford et al., 2016). Although definitions vary in the 48 literature, most simply, temperature sensitivity can be defined as the rate of 49 change with respect to temperature (Sierra, 2012); or mathematically, 50 temperature sensitivity is the first derivative of the temperature response. 51 Historically, the most common metrics for measuring temperature sensitivity 52 of soil processes have been the Arrhenius model or the Q<sub>10</sub> temperature 53 coefficient. It is important to note that  $Q_{10}$  is not the rate of change with respect to temperature, but rather the ratio between two rates. The 54 55 Arrhenius model and Q<sub>10</sub> temperature coefficient have been used to 56 characterize soil temperature sensitivity since the 1920s (Singh & Gupta, 57 1977). However, over the past several decades, a growing body of literature 58 now clearly demonstrates that Q<sub>10</sub> and Arrhenius models are ineffective and sometimes misleading models for characterizing temperature sensitivity in 59 60 soils (e.g. Lloyd & Taylor, 1994; Davidson et al., 2006; Hamdi et al., 2013; 61 Schipper et al., 2014; Tang & Riley, 2015; Alster et al., 2016a; Robinson et al., 2017). 62

63 We argue that the Q<sub>10</sub> temperature coefficient and the Arrhenius model 64 are ill suited for soil biological systems. First, the Arrhenius equation was not 65 originally intended for biological reactions, but instead to describe the 66 thermal dependence of reaction rates in physical chemistry. Applying it in

67 biological systems neglects enzyme catalysis of the reaction by assuming that enzyme tertiary structure is not temperature sensitive (DeLong et al., 68 69 2017). Second, these equations assume that biological reaction rates rise 70 monotonically with warming (i.e., only increase with increasing temperature). 71 In reality, these rates are typically unimodal—they peak at intermediate 72 temperatures, and decline at higher temperatures (Dell et al., 2011). Third, 73 Arrhenius and Q<sub>10</sub>-modeled rates are dependent on the temperature range 74 measured (Kirschbaum, 1995; Sierra, 2012; Schulte, 2015; Alster et al., 75 2016b; Pawar et al., 2016). Therefore, the same data fit to the  $Q_{10}$ 76 temperature coefficient and the Arrhenius model can yield different 77 parameter estimates for different temperature ranges, meaning that these 78 model parameters can be inconsistent metrics of temperature sensitivity. 79 Moreover, the parameters can be misleading when comparing results 80 between studies. Even more problematic with Q<sub>10</sub>, realistic values can be 81 generated when using randomly generated data as a consequence of the 82 mathematical formulation (Sierra, 2012). We therefore caution against the use of the Q10 temperature coefficient and the Arrhenius model in 83 84 biogeochemical modeling.

Here, we advocate for broader adoption of an alternative model of temperature sensitivity for soil microbial processes. Recent studies provide feasible alternatives to the Q<sub>10</sub> temperature coefficient and the Arrhenius model that provide technical advancement, empirical validation, and improved theoretical understanding of temperature sensitivity (Schipper *et* 

al., 2014; Pawar et al., 2016; Dobri & Bååth, 2018). Despite these advances, 90 91 of the papers published in *Global Change Biology* in 2017 and 2018 92 regarding temperature sensitivity in soil systems, 25 out of 31 (81%) only fit 93 data to the Q<sub>10</sub> temperature coefficient or Arrhenius model. To move beyond 94 these measures of temperature sensitivity, we provide (1) a review of 95 alternative approaches, focusing on Macromolecular Rate Theory (MMRT), (2) 96 strategies and tools to overcome potential barriers of transitioning to a new 97 temperature sensitivity framework, (3) hypotheses for incorporating MMRT 98 into thermal adaptation theory, and (4) exploration of new questions and 99 opportunities stemming from these new approaches.

100

#### 101 **2. Alternative approaches**

102 Over the past several decades, many alternative approaches have 103 been proposed to describe the temperature sensitivity of biological 104 processes. Some stem from enzyme biochemistry or microbiology and have 105 been applied to soils (Ratkowsky et al., 1982, 1983; Schipper et al., 2014), while others derive from empirical modifications of existing equations that fit 106 107 soil data (Lloyd & Taylor, 1994; Qi et al., 2002; Bååth, 2018). Most of these 108 approaches improve predictions of temperature sensitivity by modifying  $Q_{10}$ 109 or Arrhenius to account for residual variation in the data. However, most of 110 these modified models remain monotonic, so projected responses are not 111 necessarily representative of biological processes (Alster et al., 2016b).

112 Several non-monotonic, unimodal equations have been proposed to 113 describe temperature response in biological systems. Four of these have 114 been applied to soil processes: the Johnson and Lewin model (Jing et al., 115 2014), the square root model (Ratkowsky equation) (Pietikäinen et al., 2005; 116 Rinnan et al., 2009, 2011; Birgander et al., 2013; van Gestel et al., 2013; 117 Taylor et al., 2017; Duan et al., 2018), the equilibrium model (Menichetti et 118 al., 2015), and macromolecular rate theory (MMRT) (Schipper et al., 2014; 119 Alster et al., 2016a, 2016b; Robinson et al., 2017; Taylor et al., 2017; Duan 120 et al., 2018). Each has its own merits and shortcomings. However, all 121 improve upon the  $Q_{10}$  temperature coefficient and the Arrhenius model by 122 capturing the unimodality typical of biological enzymatic reactions. DeLong 123 et al. (2017) review the assumptions of these models from a thermodynamic 124 perspective (see Box 1 of DeLong et al., 2017). There are no studies directly 125 comparing all four approaches for soils, although Taylor et al. (2017) 126 compared the square root model and MMRT for nitrification by soil bacteria 127 and archaea. They noted that the two models did not differ in their 128 effectiveness (Taylor et al., 2017).

We propose the adoption of MMRT (Box 1) to represent temperature responses of soil biological systems for two reasons. First, in contrast to the square root model, MMRT is not strictly empirical but rather based on underlying thermodynamic theory. Second, in contrast to the equilibrium model, MMRT does not assume unlimited substrate supply (DeLong *et al.*, 2017). Since substrates for soil enzymatic reactions are typically limiting

135 (Schimel & Weintraub, 2003), temperature models that assume substrate136 saturation may not be as accurate.

137

#### **Box 1: Overview of Macromolecular Rate Theory**

Macromolecular Rate Theory (MMRT) was first proposed by Hobbs et al. (2013). They provided a model that better accounts for observed declines activity at temperatures below thermal enzyme denaturation in temperatures. They hypothesized that the curvature in biological temperature response curves is a function of the change in the heat capacity  $(\Delta C_{\rho}^{\dagger})$  between the enzyme-substrate complex and the enzymetransition state complex, not denaturation. Heat capacity describes how the temperature of an object increases with added energy. In the Arrhenius equation, the activation barrier is independent of temperature, which is generally true for reactions involving small molecules, such as water. biological reactions However, are typically mediated large by macromolecules, such as enzymes, which have large heat capacities (per mole), meaning that more energy is needed to raise their temperatures compared to their surrounding environment. Therefore, large  $\Delta C_{P}^{\sharp}$  values lead to temperature dependence of the activation energy. The assumption that the temperature sensitivity of activation energy is negligible is therefore not appropriate for biological reactions. Thus, biological reactions are expected to deviate from the Arrhenius model.

Macromolecular Rate Theory modifies the Arrhenius equation to account for the large change in heat capacity associated with the transition between the enzyme-substrate complex and the enzyme-transition state for macromolecules. The MMRT equation is:

$$\ln(k) = \ln\left(\frac{k_BT}{h}\right) - \frac{\Delta H_{T_0}^{\dagger} + \Delta C_P^{\dagger}(T - T_0)}{RT} \frac{+\Delta S_{T_0}^{\dagger} + \Delta C_P^{\dagger}(\ln T - \ln T_0)}{R},$$
(1)

where *k* is the rate, *T* is temperature,  $T_0$  is the reference temperature,  $k_B$  is Boltzmann's constant, *h* is Planck's constant, *R* is the universal gas constant, *H* is enthalpy, *S* is entropy, and ‡ indicates the transition state (Figure 1). With MMRT, the  $\Delta C_P^{\ddagger}$  of the enzyme controls the temperature response of the reaction. Minor mutations in isoenzymes can change the  $\Delta C_P^{\ddagger}$  and therefore result in reactions having different temperature responses (Hobbs *et al.*, 2013).



**Figure 1.** Example plot of k predicted by MMRT (solid, black line, *y*-axis, left hand side) and the first derivative of k (dotted, black line, *y*-axis, right hand side). The red line intersects dk/dT at the temperature optimum ( $T_{opt}$ ). The positive peak of dk/dT is equal to the point of maximum temperature sensitivity ( $TS_{max}$ ). Here, temperature dependence is the shape of the MMRT curve, while the temperature sensitivity corresponds to dk/dT of the MMRT curve.

Although originally intended to describe pure enzymatic response to temperature, MMRT has also been applied to soil microbial processes, at first by Schipper et al. (2014). They reported that MMRT is suitable for modeling soil enzymatic reactions and various ecosystem rates (i.e., respiration, nitrification, denitrification, and methane oxidation and production). Since then, MMRT has been further applied to soil biological reactions (Alster *et al.*, 2016a; Robinson *et al.*, 2017; Taylor *et al.*, 2017; Duan *et al.*, 2018; Liu *et al.*, 2018). Parameters from MMRT have also been used to define temperature response traits for microbes (Alster *et al.*, 2016b, 2018). These traits include  $\Delta C_{P}^{\dagger}$ , which describes the steepness of the temperature response curve,  $T_{opt}$ , which describes the point of maximum activity, and  $TS_{max}$ , which describes the point of greatest positive change in the reaction rate (i.e., point of maximum temperature sensitivity). While  $\Delta C_{P}^{\dagger}$  can be estimated directly by fitting data to equation 1,  $T_{opt}$  can be estimated by setting to zero the first derivative of that equation with respect to temperature and solving for *T* (*Arcus et al.*, 2016):

$$T_{opt} = \frac{\Delta H_{T_0}^{\dagger} - \Delta C_P^{\dagger} T_0}{-\Delta C_P^{\dagger} - R},$$
(2)

and  $TS_{max}$  (also known as  $T_{inf}$ ) can be estimated by setting to zero the second derivative of the MMRT equation with respect to temperature and solving for *T* (*Schipper et al.*, 2019):

$$TS_{max} = \frac{\Delta H_{T_0}^{\dagger} - \Delta C_P^{\dagger} T_0}{-\Delta C_P^{\dagger} \pm \sqrt{-\Delta C_P^{\dagger} R}}$$
(3)

These traits can replace activation energy or  $Q_{10}$  for describing and comparing the temperature response of different soil biological reactions. These traits also provide more intuitive and ecologically meaningful metrics for describing temperature response compared with the enthalpy and entropy parameters from MMRT.

138

#### **3.** Potential barriers and solutions

140 **3.1 Additional model parameters** 

141 One of the potential disadvantages of switching to MMRT is the 142 additional parameters in the model. More complex models can be prone to 143 overfitting and they require more data. The Arrhenius equation requires two 144 fitted parameters (i.e., activation energy and the pre-exponential factor), 145 while  $Q_{10}$  is based on a single slope parameter. More complex temperature models may involve three or four fitted parameters (DeLong et al., 2017). 146 147 Several studies comparing a three-parameter MMRT model and a two-148 parameter Arrhenius model found a superior goodness of fit for MMRT even 149 when accounting for the additional parameter (Alster et al., 2016a, 2016b; 150 Robinson et al., 2017; Liang et al., 2018). Additionally, with MMRT some 151 model parameters can be fixed using empirical information. The MMRT equation includes four parameters,  $T_0$ ,  $\Delta H_{T_0}^{\sharp}$ ,  $\Delta S_{T_0}^{\sharp}$ , and  $\Delta C_P^{\sharp}$ . However,  $T_0$  can 152 153 be set empirically to 4 to 10°C below the experimental  $T_{opt}$ , because  $T_0$  does 154 not strongly affect overall model fit (Schipper et al., 2014; Alster et al., 155 2016a). Due to the phenomena of enthalpy-entropy compensation (Sharp;

156 Chodera & Mobley, 2014),  $\Delta H_{T_0}^{\ddagger}$  and  $\Delta S_{T_0}^{\ddagger}$  are also typically interdependent 157 (Mills & Plotkin, 2015; Alster *et al.*, 2018). Therefore, the effective number of 158 model parameters is closer to two (Arcus *et al.*, 2016; Robinson *et al.*, 2017).

159 The MMRT traits provide a novel perspective on microbial and 160 enzymatic responses to temperature. Determining the temperature at which 161 the greatest change in rate occurs ( $TS_{max}$ ) could help identify climate 162 scenarios with large effects on nutrient cycling or greenhouse gas 163 production.  $TS_{max}$  is particularly interesting because it typically falls within 164 environmentally relevant temperature ranges (Alster *et al.*, 2016b).

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#### 166 **3.2 Comparing temperature response with prior studies**

Because researchers have long used  $Q_{10}$  and activation energy as measures of temperature sensitivity, there might be hesitation to adopt MMRT if its parameters are not comparable to previous models. Still, this hurdle could be overcome by fitting existing data to the MMRT model in a reanalysis. Here, we have included open source tools in our supplement to facilitate MMRT parameter fitting, including  $T_{opt}$  and  $TS_{max}$ .

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#### **3.3 Solutions for experimental limitations**

For optimal model parameterization, MMRT requires sufficient measurements across a broad temperature range. In an analysis of the sample size needed for fitting soil respiration data to the MMRT model, Robinson et al. (2017) demonstrate that model fits continue to improve up

179 until roughly 20 measurements at different temperatures. It is also important 180 to capture the  $T_{opt}$  within the range of temperatures measured (Alster *et al.*, 181 2018). These requirements (i.e., large number of independent temperature 182 measurements and a large temperature range) may be an obstacle to fitting 183 the MMRT model to existing empirical data or new data from small 184 experiments. To overcome this problem in new experiments, we suggest use 185 of multiple incubators or staggered time points to expand the range and 186 number of temperature points. Sample sizes need not increase, though. 187 Added temperature points across the experimental range could be offset by 188 decreasing replicate number at each point (O'Brien et al., 2009; Sefer et al., 189 2016). Total sample size could remain the same.

190 A temperature gradient block is another option. Common in 191 microbiology experiments, temperature gradient blocks are made of 192 aluminum set in a circulating water bath that is heated on one side and chilled on the other (detailed in Konishi et al., 2006). This approach has been 193 194 used to incubate soils and sediments at multiple temperatures with minimal 195 additional effort (Fey & Conrad, 2000; Yao & Conrad, 2000; Canion et al., 196 2014; Robinson et al., 2017). For experiments on soil microbial enzymes or 197 isolates, temperature gradients in thermal cyclers—often found in microbial 198 laboratories—are another option.

For field experiments, several solutions already exist to overcome these obstacles. Natural temperature and elevation gradients are commonly used for warming experiments (e.g., Bradford *et al.*, 2019; Dacal *et al.*, 2019)

202 and provide access to many temperature points. For manipulative field 203 experiments, measurements could be collected near and far from the 204 heating source because warming decreases with distance (Peterjohn et al., 205 1993). Field experiments also include temporal fluctuations in temperature. 206 Such variation can be used to fit the model, as well as compare between 207 warmed and control treatments (Carey et al., 2016; Li et al., 2019). 208 Additionally, variation in environmental temperatures poses an interesting 209 question for field studies: are organisms from more stable thermal 210 environments more or less temperature responsive? In other words, do they display more or less negative  $\Delta C_{\rho}^{\dagger}$ ? Overall, we urge scientists to consider 211 212 incorporating more temperature levels into future research projects.

213 While we strongly encourage incorporation of additional temperature 214 points, another logistical consideration is how temperature interacts with 215 moisture and oxygen availability. High temperatures typically decrease 216 moisture availability. The interaction of these three variables (i.e., temperature, moisture, and oxygen availability) may confound the 217 218 temperature-reaction rate relationship in soil systems (Sierra et al., 2017). 219 We therefore recommend controlling for these other variables when 220 estimating soil temperature response, for example by adding water, to 221 minimize misleading results.

222

#### 223 **3.4 Development of thermal adaptation theory**

Another potential barrier in applying MMRT is a lack of conceptual theory on thermal adaptation (Allison *et al.*, 2018), or how temperature response curves adapt to changes in temperature. Particularly, how should respiration rate, or enzymatic  $V_{max}$ , the maximum reaction velocity, adapt or acclimate to temperature change? Enzyme catalyzed reactions are typically characterized through Michaelis-Menten kinetics,

230

$$V = V_{max}[S]/(K_m + [S]), \qquad (4)$$

231

where V is velocity, S is substrate, and  $K_m$  is the half-saturation constant. 232  $V_{max}$  is thought to adapt to temperature through changes in the enzyme-233 234 substrate binding complex (Davidson & Janssens, 2006). According to the 235 Arrhenius theory, cold-adapted enzymes should have lower activation 236 energies to offset the lower kinetic energy of cooler systems, in comparison 237 to higher activation energies of warm-adapted enzymes. Therefore, 238 enzymatic adaptation to warming should entail increased activation energy 239 (Figure 2A), which is controlled by a change in the enthalpy of activation 240 (Wolfenden & Snider, 2001). A higher activation energy also implies a 241 greater temperature response (and  $Q_{10}$ ), because small changes in temperature have a larger effect on the reaction. Therefore,  $V_{max}$  is expected 242 243 to become more temperature responsive with soil warming.

244 Still, the empirical support for the Arrhenius-based theory of thermal 245 adaptation is contradictory. Some studies find that warm-adapted enzymes

246 are more temperature sensitive (Koch et al., 2007; Allison et al., 2018; Tang 247 et al., 2019), while other studies find the opposite (Koch et al., 2007; 248 Wallenstein et al., 2009; Brzostek & Finzi, 2012; Nottingham et al., 2016; 249 Razavi et al., 2017; Tang et al., 2019). This discrepancy could be due to 250 confusion in defining the term "temperature sensitivity." Some studies measure temperature sensitivity of  $V_{max}$ ,  $K_m$ , or both. Additionally, studies 251 252 can measure *intrinsic* versus *apparent* temperature sensitivity. These details 253 should be clearly reported.

254 These inconsistencies could also result from limitations in the 255 underlying reaction rate models. In particular, the assumption of a constant 256 activation energy for the transition state in Arrhenius-based theory may not 257 be valid, which could undermine thermal adaptation theory built on this 258 assumption. In addition, poor fits of the Arrhenius model to rate data may 259 lead to inaccurate estimates of the parameters used to test for thermal 260 adaptation. As an alternative, in the next section we propose new 261 hypotheses about microbial and enzymatic temperature adaption that follow from MMRT and focus on  $V_{max}$ . Interpreting the temperature sensitivity of  $K_m$ 262 263 is also important but requires more study of the thermal controls on enzyme-264 substrate binding.

265

#### **4. Hypotheses for temperature sensitivity and thermal adaptation**

267 To build a conceptual theory on how  $V_{max}$  should adapt to temperature 268 change under MMRT, we focus on two temperature response traits,  $\Delta C_{P}^{\dagger}$  and

 $\mathcal{T}_{opt}$ . A more negative  $\Delta C_{P}^{\sharp}$  corresponds to a steeper temperature response 269 curve whereas a less negative  $\Delta C_P^{\dagger}$  corresponds to a flatter curve. A steeper 270 271 temperature response curve would indicate larger changes in rate with 272 temperature compared with a flatter curve. Here, thermal adaptation refers to changes in temperature response traits (e.g.,  $\Delta C_P^{\dagger}$ ,  $T_{opt}$ ) of an enzyme in 273 274 response to shifts in the temperature or temperature regime experienced by 275 an organism. Below we propose three hypotheses derived from biochemical 276 and physiological mechanisms to describe how thermal adaptation might 277 occur within the MMRT framework (Figure 2B, C, and D).

Under the Enzyme Rigidity Hypothesis (Figure 2B), cooling causes  $\Delta C_P^{\ddagger}$ 278 to become more negative (i.e., steeper curve) and  ${\cal T}_{\it opt}$  decreases. A more 279 negative  $\Delta C_P^{\dagger}$  should be expected if cold-adapted enzymes have decreased 280 281 rigidity compared with warm-adapted enzymes (Fields & Somero, 1998; 282 Zavodszky et al., 1998; Fields, 2001). With cold-adapted enzymes, increasing 283 vibrations in the enzyme-substrate complex help compensate for declining 284 activity at lower temperatures (Wallenstein et al., 2011). A more negative  $\Delta C_P^{\sharp}$  (and a steeper curve) results when enzyme rigidity decreases through 285 286 an increased number of enzyme-transition-state species (Arcus et al., 2016). 287 With the Enzyme Rigidity Hypothesis, the entire temperature response curve 288 may also shift upwards with increasing temperature (Feller & Gerday, 2003; 289 Arcus et al., 2016). This directional shift and flattening of the temperature 290 response curve are in line with empirical findings of thermophilic versus

psychrophilic enzymes (Struvay & Feller, 2012; Arcus *et al.*, 2016), but this upward shift is not always observed (Struvay & Feller, 2012). Additionally, several studies have found that a more negative  $\Delta C_p^{\ddagger}$  corresponds to a lower  $T_{opt}$  (Hobbs *et al.*, 2013; Arcus *et al.*, 2016; Alster *et al.*, 2018), which is actually an expected consequence of the mathematics (Arcus *et al.*, 2016).

296 The second and third hypotheses, which we term the *Optimum Driven* 297 Hypothesis and the Thermal Breadth Hypothesis, respectively, take a 298 physiologically-driven approach (Figure 2C and D). For the Optimum Driven Hypothesis, the  $T_{opt}$  increases with warming to more closely match the new 299 thermal environment, but the  $\Delta C_{P}^{\dagger}$  remains the same (Figure 2C). We might 300 expect this type of response if  $\Delta C_{P}^{\dagger}$  is a highly conserved property of that 301 302 enzyme (Alster et al., 2016b). Furthermore, results from Alster et al. (2018) (see Figure 3C) suggests that multiple  $T_{opt}$  values are possible at the same 303  $\Delta C_P^{\dagger}$ , particularly at less negative  $\Delta C_P^{\dagger}$  values. However, it might be that this 304 relationship is not possible at more negative  $\Delta C_P^{\sharp}$  and that  $\Delta C_P^{\sharp}$  must increase 305 with increasing  $T_{opt}$  as is predicted in the Enzyme Rigidity Hypothesis. 306

Lastly, the Thermal Breadth Hypothesis predicts that changes in  $\Delta C_{P}^{\ddagger}$ are related to the temperature range of the environment (Figure 2D). Enzymes exposed to more temperature variation would have flatter temperature response curves (less negative  $\Delta C_{P}^{\ddagger}$ ) to maintain more constant rates across varying temperatures. We previously found that multiple  $\Delta C_{P}^{\ddagger}$  values are also possible with the same  $T_{opt}$  (Alster *et al.*, 2018). This result may be explained by enzymes experiencing different temperature regimes, but similar mean temperatures. Temperature regime is thought to play a role in determining temperature sensitivity, but few studies have examined this relationship (Zhu & Cheng, 2011; Bai *et al.*, 2017).

317

#### 318 **5. New questions and opportunities**

319 Several questions arise from these hypotheses, providing new 320 opportunities for inquiry and development of thermal adaptation theory for 321 soil biological systems. These questions include: Which of the proposed 322 hypotheses, if any, are supported experimentally? Which enzyme thermal response traits change with long-term environmental warming (i.e.,  $\Delta C_P^{\dagger}$  or 323  $T_{opt}$ )? How much divergence, if any, does warming cause in the temperature 324 response curves? How can we incorporate responses of  $K_m$  into this new 325 thermal adaptation framework? How will adaptation of  $\Delta C_{P}^{t}$  and  $T_{opt}$  vary with 326 327 substrate availability or temperature variability?

Another set of key questions centers on scaling enzyme MMRT. Prior studies have already demonstrated MMRT's validity for higher level biological processes (Alster *et al.*, 2016a; Robinson *et al.*, 2017; Duan *et al.*, 2018; Liang *et al.*, 2018); however, what this theory means at scales beyond the enzyme level is still unknown. What is the relationship between single enzymatic reactions versus reactions involving a suite of enzymes (e.g.,

microbial respiration) (Alster et al., 2018)? Can we expect changes in 334 335 thermal adaptation to vary by ecosystem, microbe, or enzyme? How will 336 production of different isoenzymes and changes in microbial community 337 composition be reflected in the thermal adaptation of temperature sensitivity 338 of a microbe or community? We previously hypothesized that reactions 339 involving multiple enzymes would reflect the summation of the temperature response curves and thus have a less negative  $\Delta C_{P}^{\dagger}$  (Alster *et al.*, 2018). 340 341 However, we did not find that the data were entirely consistent with this 342 hypothesis. Additionally, soils are composed of both organic and inorganic 343 elements. These inorganic elements lack enzyme catalysts and follow 344 Arrhenius-type kinetics for chemical reactions (e.g., sorption, desorption, or 345 diffusion processes). Schipper et al. (2019) argue that soil processes should 346 thus be determined by combining the MMRT and Arrhenius models based on 347 substrate supply in the system. However, how to determine the relative 348 contribution of each model and how to vary each of the model parameters 349 for this mixed-model requires attention.

350

#### **351 6.** Conclusion

We have several recommendations to facilitate testing and application of MMRT. We advise researchers to design experiments with as many independent temperatures as possible, up to 20, and across a range that is biologically relevant and includes  $T_{opt}$ . This approach is important to secure adequate data for model fitting. We also encourage researchers to collect

357 data that can be used to fit and compare alternative models. Macromolecular 358 Rate Theory offers universal metrics for comparing temperature sensitivity 359 across microbes and systems. Measurements that are only analyzed using a 360  $Q_{10}$  temperature coefficient or the Arrhenius equation represent a missed 361 opportunity to test MMRT. Using MMRT, we can examine more biochemically 362 relevant parameters, which could provide insights into how enzymes and 363 organisms adapt to temperature. Therefore, we suggest MMRT as a powerful 364 tool for representing the biochemical mechanisms operating in soil systems. 365 In doing so, we may improve predictions of microbial temperature responses 366 to climate change.

367

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376

377 Figure 2. Hypotheses for thermal adaptation of temperature response 378 curves for cold-adapted (blue lines) and warm-adapted (red, dashed lines) 379 biological reactions. Panel A corresponds to the thermal adaptation hypothesis generated from the Arrhenius equation and panels B-D 380 381 correspond to hypotheses for thermal adaptation developed for the MMRT 382 framework. With all hypotheses, the magnitude and direction of the change 383 in rate will depend on how much the temperature response curve shifts, and at what temperature the reaction occurs. See table S1 for the example 384 385 parameter values corresponding to each plot.

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