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Charge density wave and order-disorder in the antiferromagnetic metallic series $Eu(Ga_{1-x}Al_x)_4$

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The solid solution $\operatorname{Eu}(\operatorname{Ga}_{1-x}\operatorname{Al}_x)_4$ was grown in single crystal form to reveal a rich variety of crystallographic, magnetic, and electronic properties that differ from the isostructural end compounds EuGa_4 and EuAl_4 , despitethe similar covalent radii and electronic configurations of Ga and Al. Here we report the onset of magnetic spin reorientation and metamagnetic transitions for x=0-1 evidenced by magnetization and temperature-dependent specific heat measurements. $T_{\rm N}$ changes non-monotonously with x, and it reaches a maximum around 20 K for x=0.49, where the a lattice parameter also shows an extreme (minimum) value. Anomalies in the temperature-dependent resistivity consistent with charge density wave behavior exist for x=0.49 and 1 only. Density functional theory calculations show the increased ionic character between the Ga-Al bonds in the x=0.49 structure compared to the end compounds, such that lack of crystallographic disorder and chemical pressure are proposed as the causes of the charge density wave behavior.

I. INTRODUCTION

The interplay of structural, magnetic, and electronic properties of rare earth based intermetallics often results in emergent phenomena and competing ground states, such as unconventional superconductivity, heavy fermion behavior, intermediate valence, and quantum criticality. ¹ Particularly, pressure, magnetic field, or chemical doping in Ce and Yb compounds in their magnetic or nonmagnetic sublattices has been extensively used to tune the balance between their versatile ground states. $^{2-4}$ Comparatively less work has been done to explore the effects of pressure or doping in Eu-based intermetallics, even though Eu presents similar opportunities to tune the ground state through valence fluctuations between magnetic Eu²⁺ and nonmagnetic Eu³⁺ ions.⁵ In this study, we explored the effects of the isovalent doping in the $Eu(Ga_{1-x}Al_x)_4$ series, motivated by the wide range of apparently conflicting results observed when tuning the properties of the end compounds EuGa₄ and EuAl₄.

Previous studies on single crystals of the stoichiometric compounds EuGa₄ and EuAl₄ revealed similar magnetic behavior, with antiferromagnetic (AFM) ordering with very similar Neél temperatures $T_{\rm N}=15$ K and 15.4 K, respectively. The compounds are isostructural, forming in a tetragonal crystal structure consisting of two distinct transition metal sites, forming a covalently-bound anionic framework with divalent body-centered cations. The structural and magnetic similarities between these two compounds may be easily understood considering the chemical similarities of Ga and Al: they are isova-

lent, with very close covalent radii of 1.22 Å and 1.21 Å, respectively. However, drastic differences have also been noted with either doping or applied pressure, which cannot be readily explained. While no evidence for mass renormalization has been reported in EuAl₄, electrical resistivity measurements have suggested heavy fermion behavior in EuGa₄.^{7,8} At ambient pressure, a plausible charge density wave (CDW) was reported in the former compound below $T^* = 140$ K, and increasing pressure suppressed T^* to zero for p = 2.5 GPa. However, in the latter compound, a plausible CDW is observed only under applied pressure, with $T^* = 105 \text{ K}$ for p = 0.75 GPa, which subsequently increased to 160 K for p = 2.15GPa. Doping EuM_4 (M = Ga or Al) on either the magnetic (Eu) or nonmagnetic (M) sublattice has also shown notable changes in the magnetic, electronic, and crystallographic properties. When Eu is substituted by Yb in $(Eu_{0.5}Yb_{0.5})Ga_4$, T_N is suppressed to 13 K.¹⁰ By comparison, doping EuGa₄ in the nonmagnetic sublattice has shown that the AFM order is suppressed down to $T_{\rm N} = 9.6$ K and 6.3 K in polycrystalline Eu(Ga_{1-x}A_x)₄ (A,x) = (Mg,0.14) or (Li,0.18), respectively.¹¹ In contrast, EuAl₄ doped with Si resulted in ferromagnetic (FM) order below $T_{\rm C} = 17 \text{ K}$ in Eu(Al_{0.75}Si_{0.25})₄. ¹²

The versatile interplay between spin, charge, and orbital degrees of freedom in $\mathrm{Eu}M_4$ motivates the current systematic study of the solid solution between the Ga and Al end compounds in the series $\mathrm{Eu}(\mathrm{Ga}_{1-x}\mathrm{Al}_x)_4$ with x=0 to 1. Such a substitution should minimize the chemical effects brought about by doping, since replacing Ga with isoelectronic and similarly-sized Al does not change the

electron count or the volume of the unit cell (and hence the chemical pressure). Thermodynamic and transport measurements on $Eu(Ga_{1-x}Al_x)_4$ single crystals reveal strong correlations between the structural, magnetic, and electronic properties. The compounds remain tetragonal with space group I4/mmm at room temperature for the whole doping range, with Ga and Al preferentially occupying one or the other of the two transition metal element sites. Remarkably, for x = 0.49, the two transition metals fully separate into two sublattices and form an ordered structure EuGa₂Al₂ with a minimum unit cell volume in the series. This, in turn, favors the occurrence of a plausible CDW state at ambient pressure at $T^* = 51 \text{ K}$, while T_N is maximum in this composition at \sim 20 K. These results should be contrasted with those from isoelectronic doping (Ca²⁺ or Sr²⁺) or hole doping (La³⁺)¹³ in EuGa₄ on the magnetic sublattice, where in some cases structural distortions preclude the occurrence of a CDW transition down to 2 K.

II. EXPERIMENTAL METHODS

Single crystals of $Eu(Ga_{1-x}Al_x)_4$ were grown using a self-flux technique. Elemental metals were assembled in alumina crucibles with a 1:9 ratio of Eu:Ga/Al. In a typical growth, the metals were melted and homogenized at 900°C and cooled to 700°C at 3°C/hour in an inert argon atmosphere. Single crystals were separated from the flux using centrifugation through an alumina strainer placed between the crucibles. Powder x-ray diffraction was performed at ambient and low temperatures on a Bruker D8 Advance equipped with a Bruker MTC-LOWTEMP sample stage using Cu K α radiation. Rietveld refinements were done using Topas software. Single crystal x-ray diffraction was performed on a Bruker Apex II diffractometer or a Rigaku SCX Mini diffractometer using Mo K α radiation. Integration of raw frame data was done with Bruker Apex II software or CrystalClear 2.0. Refinement of the diffraction data was performed using XPREP and ShelXTL software packages.

Electron microprobe analysis (EMPA) was performed using a Cameca SX-100 electron probe microanalyzer with a wavelength-dispersive spectrometer. An accelerating potential of 15 kV and a beam current of 20 nA in a 1 μ m fixed beam were used to collect elemental intensities from 15 representative points on a polished surface of each crystal. The composition of each crystal was determined using the averages and standard deviations of the elemental intensities of Eu, Ga, and Al. The elemental intensities of Eu and Ga were determined from a standard sample of EuGa₄, and the elemental intensity of Al was similarly determined from a standard sample of Al_2O_3 . Chemical formulas for each crystal were calculated assuming 5 atoms per formula unit and full occupancy of the Ga/Al site. The compositions obtained from EMPA and XRD free variable refinement were comparable, so the doping fractions determined by EMPA were rounded

to two decimal places and used for all of the analyses throughout this paper.

Single energy images, elemental maps, and Eu $\rm M_{5,4}$ -edge x-ray absorption spectra (XAS) were acquired using the scanning transmission x-ray microscope instrument at the spectromicroscopy beamline 10ID-1 at the Canadian Light Source according to data acquisition methodology described previously. ^{14,15} Samples were prepared by grinding crystals of the analyte into a fine powder with a mortar and pestle and brushing the powder onto carbon support films (3-4 nm carbon, Electron Microscopy Sciences) with a fiber, which arranged a large number of micron-sized particles in a compact area suitable for Eu $\rm M_{5,4}\text{-}edge~XAS.$

DC magnetic susceptibility measurements were performed on a Quantum Design Magnetic Properties Measurement System. Heat capacity measurements were performed using adiabatic thermal relaxation technique on a Quantum Design Physical Properties Measurement System (PPMS). Temperature-dependent ac resistivity measurements were performed on a Quantum Design PPMS using the current i=2 mA and f=462.02 Hz for a duration of 7 seconds with i||ab.

III. RESULTS

A. Crystallography

Single crystals of $Eu(Ga_{1-x}Al_x)_4$ with dimensions of approximately $3 \times 2 \times 1 \text{ mm}^3$ were grown with x = 0, 0.18, 0.33, 0.49, 0.68, and 1. Powder x-ray diffraction indicates that all crystals in this series crystallize in the tetragonal I4/mmm space group at 300 K. A typical Rietveld analysis is shown for x = 0.49 in Fig. 1, indicating no significant flux inclusions or impurity phases. Temperaturedependent powder x-ray diffraction measurements (Supplementary Materials Fig. S1) on EuAl₄ at T = 300 Kand 93 K confirm that the tetragonal crystal structure is preserved down to low temperatures with no structural phase transition, as was reported in some isostructural BaAl₄-type structures. ¹⁶ Single crystal x-ray refinements confirm the I4/mmm space group in all compounds reported herein and indicate full occupancy of all lattice sites. In EuGa₄ and EuAl₄, the Ga and Al atoms occupy two inequivalent crystallographic sites corresponding to the 4d site, M(1), at $(0, \frac{1}{2}, \frac{1}{4})$ and the 4e site, M(2), at (0, 0, z). Upon substituting Ga for Al, a clear site preference is shown: Al fully occupies the 4d site before occupying the 4e site. Diffraction data for single crystal x-ray refinements can be found in the Supplementary Materials in Table S1.

B. Physical Properties

Eu $M_{5,4}$ -edge x-ray spectromicroscopy was used to probe electronic structure and bonding in selected sam-

ples of Eu(Ga_{1-x}Al_x)₄ with x=0, 0.18, 0.49, and 1. In general, each of the Eu M₅- and M₄-edges exhibits characteristic multiplet splitting patterns with fine structure that closely resembles expectations from earlier Eu M_{5,4}-edge studies of divalent Eu compounds.^{17,18} Preliminary calculations in the atomic limit for Eu²⁺ that described transitions from $3d^{10}4f^7$ to $3d^{10}4f^8$ states also reproduced the salient features of the experimental spectra, including the high energy shoulders observed at approximately 1132.5 eV. Hence, the Eu M_{5,4}-edge spectra support a ground state Eu²⁺ valence formulation for each Eu(Ga_{1-x}Al_x)₄ compound, and no evidence for mixed valence character was detected. A detailed study of these results in addition to the Ga L_{2,3}-edge are forthcoming.¹⁹

Previous reports showed AFM order in EuGa₄ and EuAl₄ at $T_{\rm N}=15$ K and 15.4 K, respectively, and the appearance of spin reorientation transitions in EuAl₄.^{6,8} However, in the doped series Eu(Ga_{1-x}Al_x)₄ it appears that, as Al replaces Ga(1) at the 4d site, multiple spin reorientation transitions occur, while $T_{\rm N}$ changes nonmonotonously with x. Magnetic susceptibility measurements with H||ab| and H||c| are shown in Fig. 2(a) and 2(b). As many as three magnetic transitions occur down to 1.8 K in x=0.49 and x=1. The magnetic transition temperatures were determined from magnetization derivatives d(MT)/dT and $C_p(T)$ data.²⁰ Even though the end compounds order at virtually identical $T_{\rm N}$ values, it appears that the ordering temperature is significantly enhanced at intermediate compositions, and is maximum

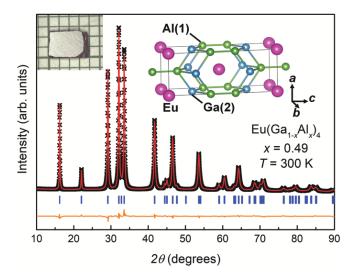


FIG. 1: Powder x-ray diffraction (black symbols) of a doped single crystal $\operatorname{Eu}(\operatorname{Ga}_{1-x}\operatorname{Al}_x)_4$ with x=0.49 indicates that this crystal (and all crystals in this doped series) crystallizes in the I4/mmm space group with no significant flux inclusion or impurity phases. The red line is the diffraction pattern calculated from Rietveld refinement and the blue ticks are the calculated peak positions. The orange line is the difference between the measured points and the calculated diffraction. The left inset is a picture of a crystal with each square = 1 mm x 1 mm and the right inset shows the tetragonal crystal structure.

at $T_{\rm N}=19.0$ K near the ordered structure at x=0.49 (purple, Fig. 2). A summary of the magnetic transition temperatures for these compounds is given in Table S2.

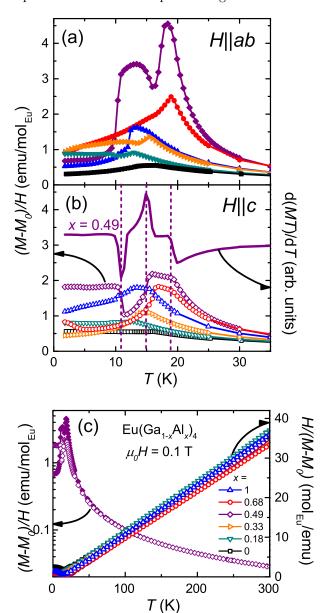


FIG. 2: Temperature-dependent magnetic susceptibility measurements with (a) H||ab and (b) left: H||c. Right: Peaks determined from d(MT)/dT were used to indicated $T_{\rm N}$ and spin reorientation transition temperatures. At high temperatures, (c) left: $(M-M_0)/H$ for x=0.49 with closed symbols representing H||ab and open symbols representing H||c. Right: the inverse magnetic susceptibility of the polycrystalline average indicates that these crystals show Curie-Weiss behavior and fully divalent Eu ions.

High-temperature inverse magnetic susceptibility $H/(M-M_0)$ indicates Curie-Weiss behavior across the series as $H/(M-M_0)$ are linear (Fig. 2c) above ~ 25 K. The linear fits are used to determine the effective magnetic moment p_{eff} and Weiss temperatures $\theta_{\rm W}$, and these

are listed in Table S2. The p_{eff} values are comparable to the theoretical $p_{eff}^{theory}=7.94$ for Eu²⁺, while the $\theta_{\rm W}$ values are positive and close to the $T_{\rm N}$ temperatures for the whole series. Positive $\theta_{\rm W}$ values are indicative of FM correlations, which were also observed in an isostructural compound EuRh₂Si₂.²²

No crystal electric field (CEF) effects are expected for Eu²⁺ ions, and this is indeed consistent with identical H||ab| and H||c| high temperature curves, with the x=0.49 data shown in Fig. 2c as an example. However, in the ordered state, slight differences in $(M - M_0)/H$ are registered in the moment orientation relative to the applied field below 50 K, as shown in Fig. 2a-b. This is even better evidenced by the anisotropic M(H) isotherms measured at T = 1.8 K (Fig. 3a-b). The magnetization saturation for all measured compounds, except x =0, is $7 \mu_{\rm B}/{\rm Eu}^{2+}$, as expected for the J = 7/2 Hund's rule ground state multiplet. EuGa₄ (black squares, Fig. 3a-b) appears to approach saturation slightly above the 7 T maximum field for these measurements. As Al replaces Ga across the $Eu(Ga_{1-x}Al_x)_4$ series, metamagnetic (MM) transitions are observed for x = 0.33, 0.49,0.68, and 1 with crystallographic anisotropy. Figure 3c shows an example of how the MM critical fields were determined from the peaks in dM/dH. As expected, the number of MM transitions at low T (Fig. 3, T = 1.8 K) coincides with the number of transitions in the low Hmagnetic susceptibility (Fig. 2).

Specific heat measurements (Fig. 4) confirmed the presence of multiple magnetic transitions in these compounds, with the transition temperatures consistent with those derived from temperature-dependent magnetization measurements. Nakamura et al. argued for heavy fermion behavior in EuGa₄ based on a Fermi liquid relation between the measured quadratic resistivity coefficient A and the calculated electronic specific heat coefficient γ with a modest mass renormalization from $\gamma=138~{\rm mJ/mol~K^2.^8}$ However, our low temperature ${\rm C}_P/T$ data show no evidence for strong mass renormalization in any of the Eu(Ga_{1-x}Al_x)₄ compounds (x = 0 - 1), as shown in the inset of Fig. 4.

No Kondo correlations are present in the H=0 electrical resistivity of $Eu(Ga_{1-x}Al_x)_4$ (Fig. 5). For all x values, the high temperature resistivity decreases with T, until loss of spin disorder scattering at T_N is marked by an abrupt drop. The residual resistivity ratios RRR $= \rho(300 \text{K})/\rho_0$ (listed in Table ??) with $\rho_0 = \rho(2 \text{K})$ are an order of magnitude larger for the end compounds (x = 0 and 1) compared to the doped samples. Remarkably, we observed a sharp resistivity increase occurring for x= 0.49 and 1 around 51 and 140 K, respectively. In the latter compound, Nakamura et al.⁸ associated the resistivity increase at 140 K with a CDW-like transition. Notably, such a transition appears in $Eu(Ga_{1-x}Al_x)_4$ only for x = 0.49, where (i) x-ray diffraction indicates an ordered structure, with Ga and Al fully occupying the two separate sublattices to form EuGa₂Al₂, and (ii) resistivity measurements reveal the lowest residual resistivity ρ_0

and an enhanced RRR value compared to all other doped (disordered) samples.

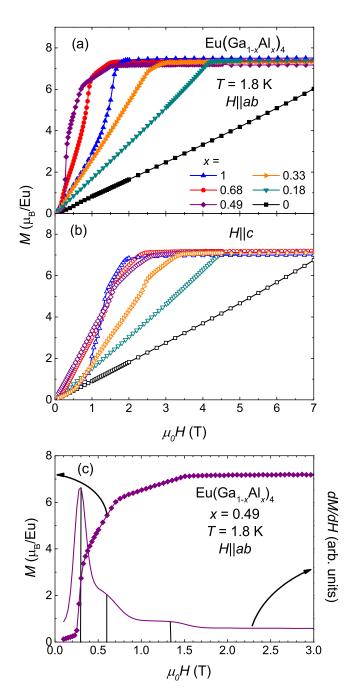


FIG. 3: Field-dependent magnetization measurements with (a) $H\|ab$ and (b) $H\|c$ show multiple metamagnetic transitions that are anisotropic. An example of a metamagnetic transition in this series is shown in (c) with an example of how critical fields were determined using peaks from dM/dH vs. H.

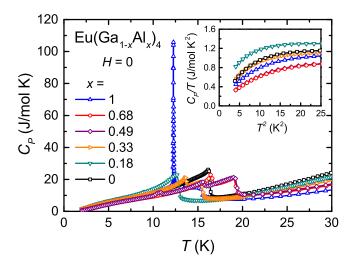


FIG. 4: Specific heat measurements confirm multiple magnetic transitions and a first-order phase transition in EuAl₄. The inset shows no evidence of mass renormalization in this system from C_P/T vs. T^2 .

IV. DISCUSSIONS AND CONCLUSIONS

Given the chemical similarities between Ga and Al (iso-electronic, similar covalent radii of 1.22 Å and 1.21 Å, respectively⁹), no substantive differences in crystallographic or physical properties are expected between the isostructural EuGa₄ and EuAl₄ compounds. However, as Al replaces Ga in Eu(Ga_{1-x}Al_x)₄, the magnetic, electronic, and structural properties change non-monotonously: (i) As shown in Fig. 6a, a maximum $T_{\rm N}$ occurs in x=0.49. This is the result of the minimum Eu-Eu ion spacing in this composition as evi-

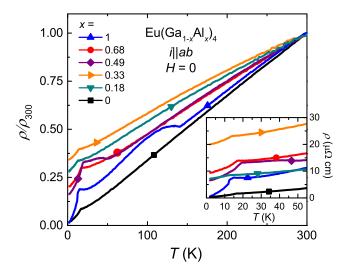
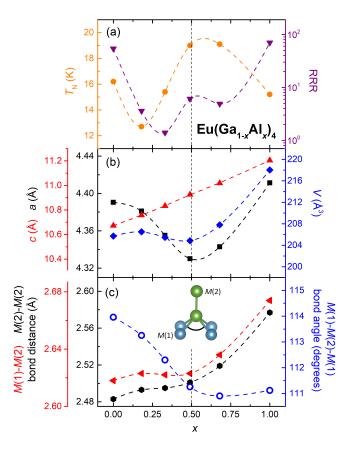


FIG. 5: Temperature-dependent resistivity scaled by ρ_{300} . Anomalies in x=0.49 and 1 are consistent with CDW-like behavior. Inset: Absolute resistivity values at low temperature.

denced by the non-linear change in the a lattice parameter and unit cell volume (squares and diamonds, Fig. 6b), which are minimum for x = 0.49 while c (triangles) increases linearly from x = 0 to 1. The ground state across the series is AFM (Fig. 2), even though the spin correlations appear FM ($\theta_W > 0$, $\theta_W \sim T_N$). In the absence of frustration or CEF effect, magnetic order is likely a result of strong next-nearest-neighbor interactions (with exchange coupling $J_2 > 0$), in addition to the nearest neighbor Rudermann-Kittel-Kasuya-Yosida coupling (exchange coupling $J_1 < 0$), such that $J_2 > |J_1|^{21}$. This is consistent with the proposed magnetic structure of EuGa₄, where intra-plane Eu magnetic moments are thought to couple ferromagnetically, while inter-plane Eu magnetic moments couple antiferromagnetically.²² (ii) The observation of a possible CDW transition in $\operatorname{Eu}(\operatorname{Ga}_{1-x}\operatorname{Al}_x)_4$ with x = 0.49 and 1 may stem directly from the ordered structure, considering the evidence for full site separation for Ga and Al in the x= 0.49 compound. This, however, does not explain the lack of a CDW in the x = 0 (also ordered) analogue, even though applied pressure appeared to induce such a transition. Additional qualitative differences exist even in the pressure-dependence of the plausible CDW transition in EuGa₄ and EuAl₄. According to the change in lattice parameters shown in Fig. 6b, it seems that Al substituting for Ga acts as positive pressure, resulting in the occurrence of a CDW at x = 0.49 in Eu(Ga_{1-x}Al_x)₄, similar to the behavior in EuGa₄ under applied pressure. (iii) Most notable of the non-monotonous trends in this series is the minimum in the in-plane lattice parameter a at x = 0.49 compared to the linear increase in c across the entire series (Fig. 6b). In order to explain this nonlinear structural trend, density functional theory (DFT) calculations with the local density approximation (LDA) were carried out in the linear muffin tin orbital tight binding atomic spheres approximation (LMTO-TB-ASA) to probe the bonding character between Al and Ga in the doped compounds.

DFT calculations were performed for x = 0, 0.49, and1. To avoid complications arising from the unpaired felectrons of Eu²⁺, Sr²⁺ was substituted as an analog in the calculations. In order to ensure that the non-linear changes in a were associated solely with the Ga-Al bonds and not the Eu atoms, single crystals of SrGa₄, SrGa₂Al₂, and SrAl₄ were grown from self-flux, and their lattice parameters were measured from powder x-ray diffraction (shown in the Supplementary Materials Fig. S2). Trends in lattice parameters similar to those in the Eu analogues were observed, with a minimized in $SrGa_2Al_2$ and c increasing linearly from $SrGa_4$ to $SrAl_4$. As expected given the isoelectronic nature of the series, all three band structures are qualitatively very similar (Supplementary Materials Fig. 3. However, analysis of the electron distribution extracted from the integrated density of states (DOS) up to $E_{\rm F}$ reveals substantive differences between the end compounds and the x = 0.49 composition: there is charge transfer from the M(1) to the

M(2) site as the composition approaches x=0.49 from both end compounds, such that the M(1) [M(2)] electron density is minimum [maximum] for x=0.49 (see Supplementary Materials Table S3). This maximum charge transfer manifests when the two M sites are preferentially occupied by $M(1)=\mathrm{Al}$ and $M(2)=\mathrm{Ga}$, implying an enhanced ionic character of the M(1)-M(2) bond at x=0.49 compared to both x=0 and 1. Despite the similar trends toward less ionic character in the Al-rich and Ga-rich compounds, the increased ionicity from x=0



(a) Left: Increasing x corresponds to a nonmonotonic change in T_N (orange circles) that could be associated with changes in lattice parameters a and c. Right: RRR values (purple down triangles) calculated from resistivity measurements show the low amount of disorder in the end compounds and the decreased disorder in x = 0.49 compared to other doped compounds in the series. (b) Left: Lattice parameters a (black squares) and c (red triangles) as a function of doping fraction x indicating a linear change in c and a non-linear change in a with increasing x resulting in a local minimum. Right: Unit cell volume V (blue diamonds) as function of x. (c) Left: Bond distances between atoms located at the M(1)-M(2) (red left triangles) and M(2)-M(2)(black hexagons) crystallographic sites remain constant up to x = 0.49 but increase from x = 0.49 to 1. Right: The tetrahedral bond angle between M(1) - M(2) - M(1) atoms (blue open circles) decreases up to x = 0.49 and remains constant from x = 0.49 to 1. All dashed lines are guides to the eye.

0 to x = 0.49 prevents bond length expansion (as M(1) is replaced by Al but M(2) remains occupied by Ga), but then ionicity is reduced again from x = 0.49 to x = 1 (as M(2) is also replaced by Al), resulting in a greater increase in bond lengths.

This unexpected deviation from Vegard's law²³ can be further explained by examining the trends in the M(1) - M(2) and M(2) - M(2) bond lengths and the M(1) - M(2) - M(1) bond angle, where M = Al or Ga. As shown in Fig. 6c, as Al occupies the M(1) site up to x = 0.49, the bond distance between M(1) and Ga(2)remains relatively unchanged. However, the bond angle M(1)-Ga(2) – M(1) in the Ga-centered tetrahedron decreases linearly up to x = 0.49. These crystallographic trends acting together expand the c lattice parameter while simultaneously contracting the a lattice parameter to a minimum. As Al substitutes Ga in the M(2) site up to x = 1, a different trend emerges. Here we observe that the tetrahedral bond angle remains constant while the bond lengths between Al(1)-M(2) and M(2)-M(2)increase, thus leading to both lattice parameters a and c increasing. These behaviors are likely caused by the greater electronegativity of Ga, which renders the Ga-Ga bonds more ionic.

In summary, we have observed that although Ga and Al are very similar in their valence and size, substituting Al for Ga in the doped system $Eu(Ga_{1-x}Al_x)_4$ produces striking and unexpected magnetic, electronic, and structural transitions. Substituting Ga with Al up to x =0.49 decreases a to a minimum and appears to increase the ferromagnetic interactions in the system, resulting in higher $T_{\rm N}$ and multiple magnetic transitions. Additionally, temperature-dependent $\rho(T)$ measurements show pronounced changes in electronic transport as manifested by CDW formation in $Eu(Ga_{1-x}Al_x)_4$ for x = 0.49 and 1. The CDW behavior is markedly different between EuAl₄ and EuGa₄, and chemical and hydrostatic pressure can be used as tools to elucidate the factors contributing to the CDW formation in this series. Future studies will aim to distinguish between the effects of doping in the magnetic versus the nonmagnetic sublattice in EuGa₄ and to explore the effects of hole-doping, positive chemical pressure, and disorder on the magnetic and electronic properties of EuGa₄.

V. ACKNOWLEDGEMENTS

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neering Research Council of Canada, the University of Saskatchewan, the Government of Saskatchewan, Western Economic Diversification Canada, the National Research Council Canada, and the Canadian Institutes of Health Research. Tyrel, add yours

¹ M. Brian Maple, J. Phys. Soc. Jpn., **74**, 222 (2005).

² M. Nicklas, M. E. Macovei, J. Ferstl, C. Krellner, C. Geibel, and F. Steglich, *Phys. Status Solidi B*, **247**, 727 (2010).

- ³ T. Takabatake, F. Iga, T. Yoshino, Y. Echizen, K. Katoh, K. Kobayashi, M. Higa, N. Shimizu, Y. Bando, G. Nakamoto, H. Fujii, K. Izawa, T. Suzuki, T. Fujita, M. Sera, M. Hiroi, K. Maezawa, S. Mock, H. v. Löhneysen, A. Brückl, K. Neumaier, and K. Andres, J. Magn. Magn. Mater., 177-181, 277 (1998).
- ⁴ T. N. Voloshok, N. V. Mushnikov, N. Tristan, R. Klingeler, B. Büchner, and A. N. Vasiliev, *Phys. Rev. B*, **76**, 172408 (2007).
- Y. Ōnuki, A. Nakamura, F. Honda, D. Aoki, T. Tekeuchi, M. Nakashima, Y. Amako, H. Harima, K. Matsubayashi, Y. Uwatoko, S. Kayama, T. Kagayama, K. Shimizu, S. Esakki Muthu, D. Braithwaite, B. Salce, H. Shiba, T. Yara, Y. Ashitomi, H. Akamine, K. Tomori, M. Hedo and T. Nakama, *Philos. Mag.*, (2016).

⁶ Svilen Bobev, Eric D. Bauer, J.D. Thompson, and John L. Sarrao, J. Magn. Magn. Mater., 277, 236 (2004).

- Ai Nakamura, Yuichi Hiranaka, Masato Hedo, Takao Nakama, Yasunao Miura, Hiroki Tsutsumi, Akinobu Mori, Kazuhiro Ishida, Katsuya Mitamura, Yusuke Hirose, Kiyohiro Sugiyama, Fuminori Honda, Rikio Settai, Tetsuya Takeuchi, Masayuki Hagiwara, Tatsuma D. Matsuda, Etsuji Yamamoto, Yoshinori Haga, Kazuyuki Matsubayashi, Yoshiya Uwatoko, Hisatomo Harima, and Yoshichika Ōnuki, J. Phys. Soc. Jpn., 82, 104703 (2013).
- ⁸ Ai Nakamura, Taro Uejo, Fuminori Honda, Tetsuya Takeuchi, Hisatomo Harima, Etsuji Yamamoto, Yoshinori Haga, Kazuyuki Matsubayashi, Yoshiya Uwatoko, Masato Hedo, Takao Nakama, and Yoshichika Ōnuki, *J. Phys. Soc. Jpn.*, **84**, 124711 (2015).
- ⁹ Beatriz Cordero, Verónica Gómez, Ana E. Platero-Prats, Marc Revés, Jorge Echeverría, Eduard Cremades, Flavia Barragán, and Santiago Alvarez, *Dalton Trans.*, 2832 (2008).
- ¹⁰ G. D. Loula, R. D. dos Reis, D. Haskel, F. Garcia, N. M. Souza-Neto, and F. C. G. Gandra, *Phys. Rev. B*, **85**, 245128 (2012).
- Abishek K. Iyer, Lahari Balisetty, Sumanta Sarkar, and Sebastian C. Peter, J. Alloys Compd., 582, 305 (2014).
- Paul H. Tobash and Svilen Bobev, J. Alloys Compd., 418, 58 (2006).
- 13 M. Stavinoha, et al., in preparation.
- ¹⁴ Stefan G. Minasian, Enrique R. Batista, Corwin H. Booth, David L. Clark, Jason M. Keith, Stosh A. Kozimor, Wayne W. Lukens, Richard L. Martin, David K. Shuh, S. Chantal E. Stieber, Tolek Tylisczcak, and Xiao-dong Wen, J. Am. Chem. Soc., 139, 18052 (2017).
- Alison B. Altman, C. D. Pemmaraju, Selim Alayoglu, John Arnold, Eric D. Bauer, Corwin H. Booth, Zachary Fisk,

- Joseph I. Pacold, David Prendergast, David K. Shuh, Tolek Tyliszczak, Jian Wang, and Stefan G. Minasian, *Phys. Rev. B*, **97**, 045110 (2018).
- Gordon J. Miller, Fan Li, and Hugo F. Franzenh, J. Am. Chem. Soc., 115, 3739 (1993).
- ¹⁷ G. Kaindl, G. Kalkowski, W. D. Brewer, B. Perscheid, and F. Holtzberg, J. Appl. Phys., 55, 1910 (1984).
- ¹⁸ B. T. Thole, G. van der Laan, J. C. Fuggle, G. A. Sawatzky, R. C. Karnatak, and J.-M. Esteva, *Phys. Rev. B*, **32**, 5107 (1985).
- ¹⁹ S. G. Minasian, et al., in preparation.
- 20 Michael E. Fisher, *Philos. Mag.*, **7**, 1731 (1962).
- Neeraj Kumar, S. K. Dhar, A. Thamizhavel, P. Bonville, and P. Manfrinetti, Phys. Rev. B, 81, 144414 (2010).
- ²² Z. Hossain, O. Trovarelli, C. Geibel, and F. Steglich, *J. Alloys Compd.*, **323-324**, 396 (2001).
- ²³ A.R. Denton and N.W. Ashcroft, *Phys. Rev. A*, **43**, 3161 (1991).

Supplementary Materials

Additional crystallographic data for $Eu(Ga_{1-x}Al_x)_4$ is available in CIF format at ... (ICSD references here)

A. Summary of magnetic, transport, and crystallographic data

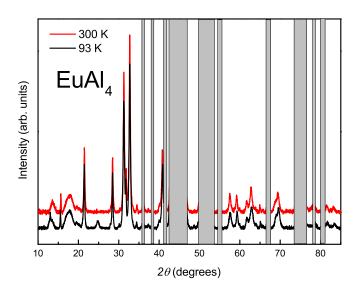


FIG. S1: Powder x-ray diffraction of EuAl₄ performed at 300 K (red line) and 93 K (black line). This indicates that the tetragonal space group is preserved above and below the CDW-like transition, and the anomaly in resistivity is not caused by a structural phase transition. Gray bars are due to large background peaks from the metal sample holder.

B. Lattice parameters and band structure calculations for $SrGa_4$, $SrAl_2Ga_2$, and $SrAl_4$

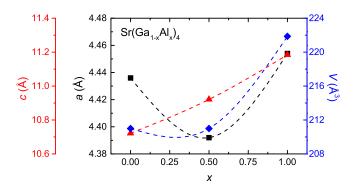


FIG. S2: Lattice parameters from powder x-ray diffraction of SrGa₄, SrAl₂Ga₂, and SrAl₄ single crystals. Trends seen here are consistent with trends observed in the Eu analogues, indicating that the non-linear change in a is associated with the Ga—Al sublattice.

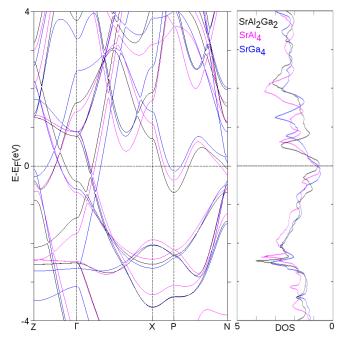


FIG. S3: Band structure calculations for $SrGa_4$, $SrAl_2Ga_2$, and $SrAl_4$. Sr^{2+} is used as a substitute for $Eu^2 +$ to avoid complications arising from unpaired 4f electrons.

TABLE S1: Crystallographic data for single crystals of $Eu(Ga_{1-x}Al_x)_4$ (space group I4/mmm). Values for x determined from EMPA.

x = 0	x = 0.18	x = 0.33	x = 0.49	0.00	1
		x = 0.55	x = 0.49	x = 0.68	x = 1
0	0.15	0.31	0.47	0.68	1
4.3904(7)	4.381(3)	4.3551(9)	4.3301(7)	4.3429(13)	4.4113(9)
10.6720(18)	10.757(7)	10.833(2)	10.9253(17)	11.018(3)	11.204(3)
205.71(7)	206.5(3)	205.47(9)	204.85(7)	207.80(14)	218.02(11)
40.640	36.87	32.93	29.14	23.57	14.968
1656	969	1734	1725	1769	1722
137	92	138	139	139	140
0.036	0.031	0.022	0.017	0.047	0.048
1.23	1.20	1.28	1.20	1.12	1.529
0.014	0.024	0.012	0.009	0.015	0.018
0.037	0.057	0.029	0.021	0.025	0.038
0.0127(11)	0.0022(13)	0.0103(9)	0.0019(5)	0.0019(8)	0.0057(15)
90	90	90	90	90	188
	4.3904(7) 10.6720(18) 205.71(7) 40.640 1656 137 0.036 1.23 0.014 0.037 0.0127(11) 90	4.3904(7) 4.381(3) 10.6720(18) 10.757(7) 205.71(7) 206.5(3) 40.640 36.87 1656 969 137 92 0.036 0.031 1.23 1.20 0.014 0.024 0.037 0.057 0.0127(11) 0.0022(13) 90 90	4.3904(7) 4.381(3) 4.3551(9) 10.6720(18) 10.757(7) 10.833(2) 205.71(7) 206.5(3) 205.47(9) 40.640 36.87 32.93 1656 969 1734 137 92 138 0.036 0.031 0.022 1.23 1.20 1.28 0.014 0.024 0.012 0.037 0.057 0.029 0.0127(11) 0.0022(13) 0.0103(9) 90 90 90	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

 ${}^{a}R_{1} = \sum ||F_{o}| - |F_{c}|| / \sum |F_{o}|$ ${}^{b}wR_{2} = [\sum [w(F_{o}{}^{2} - F_{c}{}^{2})^{2}] / \sum [w(F_{o}{}^{2})^{2}]]^{1/2}$

TABLE S2: Summary of magnetic and transport properties in $Eu(Ga_{1-x}Al_x)_4$

					cere and cramspore	FF	- (1	2 2/4		
\overline{x}	$T_{\rm N}^{a}$ (K)	$T_{\rm N} ({\rm K})^b$	$T_{\rm N} ({\rm K})^c$	p_{eff}	χ_0	θ_W	H_{c1} (T)	H_{c1} (T)	RRR	T^*
	T_2^a (K)	$T_2 (K)^b$	$T_2 (K)^c$		(emu/mol_{Eu})	(K)	H_{c2} (T)	H_{c2} (T)		(K)
	T_3^a (K)	T_3 (K) ^b	T_3 (K) ^c				H ab	$H\ c$		
0	15.9	15.9	16.2	8.13	0.0015	6.64	>7	0.6	54	
	13.3									
0.18	12.4	12.4	12.7	7.91	0	11.16	4.0		3.6	
	8.4	8.9								
0.33	14.9	14.9	15.4	8.15	0	12.26	2.5		1.4	
	12.9	13.4	13.6							
0.49	17.4	18.4	19.0	7.96	0	22.59	1.5	0.6	6.1	51
	15.4	14.9	15.6							
	10.4	10.9	10.9							
0.68	18.4	18.4	19.1	8.23	0	17.82	1.4	0.9	4.9	
	15.9	15.9	16.4							
1	14.9	14.4	15.2	7.98	0	15.02	1.6	1.4	70	141
	12.4	11.9	13.3							
	10.4	10.4	12.3							

^afrom d(MT)/dT with H||ab ^bfrom d(MT)/dT with H||c ^cfrom $C_p(T)$

TABLE S3: Analysis of the electron distribution extracted from the integrated density of states up to \mathbf{E}_F provides insight into the ionicity of the Ga-Al bonds. In contrast to both end members, in $\mathrm{SrAl_2Ga_2}$ there is increased charge transfer to the M(2) site. This charge transfer only manifests when $M(1)=\mathrm{Al}$ and $M(2)=\mathrm{Ga}$, implying an enhanced ionic character to the M(1)-M(2) bonds in $\mathrm{SrAl_2Ga_2}$.

() ()		
compound	$e^{-}/M(1)$	$e^{-}/M(2)$
$\mathrm{SrGa_4}$	5.70	4.40
SrAl_4	5.63	4.40
$\mathrm{SrAl_2Ga_2}$	5.50	4.70
$\mathrm{SrGa_4}$	5.56	4.40
(with $SrAl_2Ga_2$		
structure parameters)		
SrAl_4	5.75	4.30
(with $SrAl_2Ga_2$		
structure parameters)		