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# **Title**

Accelerated microbial turnover but constant growth efficiency with warming in soil

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105 increased MBC gross production rates (0.97  $\mu$ g MBC g<sup>-1</sup> dry soil d<sup>-1</sup> °C<sup>-1</sup>, r<sup>2</sup> = 0.99 in

106 mineral soil and 3.63  $\mu$ g MBC g<sup>+</sup> dry soil d<sup>+</sup> °C<sup>+</sup>, r<sup>2</sup> = 0.98 in organic soil). However,





 In order to assess the implications of microbial turnover to soil C predictions, we 153 used the Allison-Wallenstein-Bradford (AWB) model<sup>55</sup>. The AWB model uses rates of microbial processes that are based on the best estimate of steady state conditions, which allowed us to extrapolate the significance of our short-term results to long-term steady-156 state C stocks. We simulated three different scenarios. In the first scenario, neither MGE nor turnover was altered by temperature and soil C decomposition was modeled with a first-order decay function and Michaelis-Menten enzyme kinetics, the current assumption 159 in most biogeochemical models<sup>729</sup>. In this scenario there was no change in MBC with warming and SOC declined as a result of accelerated enzymatic decomposition (Figure  $\pm$  4). In the second scenario, MGE decreased by 0.016 °C<sup> $\pm$ </sup>, as in prior theoretical studies  $\pm$ . Here, the reduction in MGE limited microbial growth at higher temperatures, resulting in 163 a 5 % decline of MBC <sup>o</sup>C<sup>1</sup> averaged from 5 to 20 °C. As a result, SOC increased with temperature as decomposition became limited by MBC. The third scenario corresponded to our experimental observations of a constant MGE and accelerated microbial turnover with warming. Accelerated microbial turnover at higher temperature caused decreases in MBC and increase in SOC, which were larger than for the scenario of constant turnover and declines in MGE. We conclude that, although MGE did not decline, accelerated microbial turnover is an alternative mechanism that can moderate the effects of temperature on soil C stocks. These model simulations suggest that temperature-sensitive microbial turnover produces an effect on MBC and SOC that is not accounted for in current biogeochemical or microbial models. Our results show that accelerated enzyme kinetics and increased microbial

turnover are the main mechanisms associated with an increased respiration at higher

 elevated temperatures. This effect on SOC is similar to those that have been predicted in models assuming a decline in MGE, but differs in direction from the predictions traditional biogeochemical models. Consequently, soil microbial models should include a temperature-sensitive microbial turnover rate. The lack of temperature sensitivity in MGE, which is controlled at the cellular level, suggests that microbial biochemical efficiency is a weak control on soil C dynamics. **Methods** Soil samples were collected in October 2012 from the Marcell Experimental 185 Forest in Grand Rapids, MN (MAT =  $3^{\circ}$ C, MAP = 750 mm). Mineral soil samples were collected from the A horizon in a hardwood forest and organic soil samples were collected from an ombrotrophic peatland (top 40 cm after removing the living layer of moss). Soil samples were stored at 4 °C until the experiment began in April 2013. 189 Replicates ( $n = 6$ ) from both soils were randomly assigned to one of four incubators and 190 incubated for seven days at 5, 10, 15, or 20 °C (See Supplementary Methods Section I). After a seven-day incubation period, MGE was determined using two position-

temperatures and, in model simulations, lead to a small increase in SOC content under

192 specific <sup>13</sup>C-labeled isotopologues of glucose (U-<sup>13</sup>C and 1-<sup>13</sup>C) and two of pyruvate (1-<sup>13</sup>C)

193 and  $2,3$ -<sup>13</sup>C) as metabolic tracers<sup>1430</sup>. We measured <sup>13</sup>CO<sub>2</sub> accumulation in each jar three

194 times over the course of 60, 90, 135, or 180 min at 20, 15, 10, and 5 °C respectively. The

195 ratios between  $°CO<sub>2</sub>$  production rates from glucose and pyruvate isotopologues were

196 calculated and used to model metabolic pathway activities and  $MGE^{\omega}$  (Table S2). One

complete replicate (i.e. 4 temperatures x 2 soils x 4 isotopologues) was incubated and

198 analyzed each week. For more details and background information on metabolic probing 199 and modeling, see Supplementary Methods Section II and Figure S2.

 Two weeks after the MGE measurements, another incubation was set up under 201 identical conditions to measure respiration and MBC. Each of the four incubators was systematically assigned to one of the four treatment temperatures and both soils were 203 incubated for seven days. After the seven-day incubation period,  $CO<sub>2</sub>$  concentrations were measured at 0 and 24 h. After the respiration measurement, MBC concentration was measured using chloroform fumigation-extraction (See Supplementary Methods Section III, Table S1).

 We calculated microbial turnover using the experimentally measured respiration (*R*), MGE, and MBC (Supplementary Methods Section IV, Figure S3). We applied the assumptions that MBC was at steady state and that all turned over MBC was released as 210 CO<sub>2</sub>. Our findings of temperature-sensitive turnover were not affected much by non-

211 steady state of MBC and whether C from turnover was released as  $CO<sub>2</sub>$  or added to the

212 SOC pool (Supplementary Methods Section V, Figure S4 and S5).

213 The gross microbial production was calculated as

$$
214 \qquad \qquad \Delta MBC_g = MGE * R,
$$

215 and microbial turnover  $(\tau)$  assuming steady state MBC pools and all C from turnover

216 going to  $CO<sub>2</sub>$  as follows

$$
z = \frac{MGE * R}{MBC}
$$

218

219 To calculate the effect of microbial turnover and incubation duration on  $MGE$ <sub>a</sub>, we used 220 the following equation

221  $MGE_{A} = (1-\tau)^{n} \times MGE_{i}$ 



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- the microbial turnover rates for the mineral soil in our study (Figure 2). b) The modeled
- 371 relationship of  $MGE$ <sub>a</sub> and temperature in mineral soil after two days.
- 
- **Figure 4. The relative change in soil organic C (SOC) and microbial biomass C**
- **(MBC) from 5 to 20 °C under three scenarios using the Allison-Wallenstein-**
- **Bradford model.** In the constant turnover, constant MGE scenario there is no change in
- MBC with temperature.

























24 replicates of all soil x temperature x isotopologue combinations. For the organic soil 25 incubated at  $5^{\circ}$ C, one replicate was lost. Each replicate consisted of four jars, as is 26 required for the metabolic tracer probing<sup>1</sup>. Mineral soil (25 g dry weight, 66.5 % soil 27 moisture content) and organic soil (4 g dry weight, 400 % soil moisture content) was 28 incubated in specimen cups and placed in mason jars (473 mL). Mason jars were initially 29 covered with Saran<sup>™</sup> plastic wrap, to limit moisture loss but allow oxygen into the jars 30 during the seven-day incubation period. We used iButton data loggers (Maxim Integrated, 31 San Jose, CA, USA) to monitor soil temperature during the incubation and the metabolic 32 tracer probing experiment.

33

#### *II. Position-Specific* 34 *13C-Labeled Metabolic Tracer Experiment*

35 After incubating the soil for one week, headspace was refreshed before the jar was 36 closed using an airtight lid with a septum. Next,  $10$  ml of pure  $CO$  was added to each jar  $37$  in order to meet the  $CO<sub>2</sub>$  concentration and amount required by the Picarro G2201-*i* CO<sub>2</sub> 38 cavity ring-down isotope spectrometer (Picarro Inc., Sunnyvale, CA, USA). Thirty 39 minutes after addition of  $CO<sub>2</sub>$ , we took a sample of the headspace (time=0). We then 40 added one of four metabolic tracer isotopomers to each of the four parallel incubations 41 per replicate, following the procedure in Dijkstra et al.<sup>12</sup>. We used two glucose 42 isotopologues (U-<sup>13</sup>C and 1-<sup>13</sup>C) and two pyruvate isotopologues (1-<sup>13</sup>C and 2,3-<sup>13</sup>C). All of 43 the metabolic tracers were dissolved in deionized water at a concentration of 10.7  $\mu$ mol C 44 per mL. Two mL of tracer was added to each incubation, equivalent to 10.3  $\mu$ g C g<sup>-1</sup> dry 45 mineral soil and 64.3  $\mu$ g C g<sup>-1</sup> dry organic soil. After tracer addition, 10 ml of headspace

46 was sampled three times at 20 min intervals in the 20 °C incubation, at 30 min intervals 47 at 15 °C, at 45 min intervals at 10 °C, and at 60 min intervals at 5 °C. 48 All gas samples were analyzed on the Picarro G2201-*i* CO<sub>2</sub> cavity ringdown 49 isotope spectrometer. The ratios of  ${}^{13}$ C production for each isotopologue pair were 50 calculated as:

51 
$$
\frac{c_U}{c_1} \text{ratio} = \frac{^{13}CO_2 \text{ production from } U - ^{13}C \text{ glucose}}{^{13}CO_2 \text{ production from } 1 - ^{13}C \text{ glucose}} \tag{1}
$$

52 and

53 
$$
\frac{C_1}{C_{2,3}} \text{ratio} = \frac{^{13}C_2 \text{ production from } 1^{-13}C \text{ pyruvate}}{^{13}C_2 \text{ production from } 2,3^{-13}C \text{ pyruvate}}
$$
(2)

54 The  $C_0/C_1$  ratio for glucose and the  $C_1/C_{23}$  ratio for pyruvate are determined by the 55 characteristics of the central C metabolic network (i.e. glycolysis, citric acid cycle, and 56 pentose phosphate pathway, Figure S2) which cause some C-atoms to be preferentially 57 used for biosynthesis (for example in lipids and amino acids), while others are 58 preferentially lost in decarboxylation reactions. If cells use substrate mainly for the 59 production of ATP and very little for biosynthesis (substrate-limited microbial activity), 60 the  $C_U/C_1$  ratio for glucose will be close to 6:1 (all C-atoms are being released as  $CO_2$ ), 61 and the  $C_1/C_{2,3}$  ratio of pyruvate be close to 1:2. Observed ratios (Table S2) for glucose 62 and pyruvate were significantly different from the expected values for microbes without 63 biosynthesis ( $p < 0.05$ ). 64 The calculated ratios (Table S2) were used to model the metabolic flux patterns 65 through the central C metabolic network as described in Dijkstra et al.<sup>1</sup> (Figure S2). It is

- 66 assumed in this model that glucose is the only C substrate utilized by microbes. All
- 67 model rates are expressed relative to glucose uptake (*v1*), which is set at 100 moles. This

 model has nine unknowns, seven of which are estimated using a known bacterial and 69 fungal metabolite precursor demand<sup>2</sup>. In this paper, we assumed a fungi : bacteria ratio of 50:50 for the modeled microbial community. Previous studies have shown that assumptions about microbial community composition do not much alter model MGE estimates<sup>2,3</sup>, because the average precursor requirements are not that different between 73 fungi and bacteria<sup>2,4</sup>. The remaining two unknown model variables are then estimated using the observed isotopologue ratios of glucose and pyruvate. The Excel linear programming tool Solver was used to change the rates of *v14* and *v10* until modeled isotopomer ratios matched observed values. The final output of the model is relative rates for all 21 reactions of the central C metabolic network, which are used to calculate 78 MGE. The MGE is calculated from the uptake rate and  $CO<sub>2</sub>$  producing reactions ( $v10$ ,  $v5$ , *v7* and *v8*) as:

$$
80 \t\t MGE = \frac{6*v1 - \Sigma CO_2}{6*v1} \t\t(3)
$$

## *III. Respiration and Microbial Biomass Measurements*

 In a separate incubation two weeks after the final MGE measurement, we assessed the effects of temperature on respiration and microbial biomass C (MBC) concentration, and calculated the specific respiration rate (μg CO<sub>2</sub>-C mg<sup>-1</sup> MBC h<sup>-1</sup>)(n=4). In each incubator, we set up mason jars for each soil type following the same procedure that we used for the metabolic tracer incubation. After a one-week incubation, respiration rate was determined over 24 h using LICOR 6262 (LI-COR Biosciences, Lincoln, NE). Afterwards, we determined MBC using the chloroform-fumigation extraction method. Half of each sample was fumigated with chloroform for 7 days



114 Therefore respiration from turnover  $(R<sub>\tau</sub>)$  is equal to:

$$
115 \t R_{\tau} = \frac{MGE}{1 - MGE} * R_g \t\t(7)
$$

116 Combining equation 4 and equation 1, total respiration  $(R)$  is equal to:

117 
$$
R = R_g + \frac{MGE}{1 - MGE} * R_g
$$
 (8).

118 So respiration from creating new microbial biomass  $(R_g)$  can be calculated as:

119 
$$
R_g = \frac{R}{1 + \frac{MGE}{1 - MGE}} = R(1 - MGE)
$$
 (9).

120 And respiration from turnover  $(R<sub>\tau</sub>)$  is:

121 
$$
R_{\tau} = R - R_g = R(1 - (1 - MGE)) = R * MGE
$$
 (10).

122 And turnover  $(d^{-1})$  is calculated as flux of C out of MBC divided by MBC:

$$
123 \quad \tau = \frac{R_{\tau}}{MBC} = \frac{R * MGE}{MBC} \tag{11}
$$

124 A conceptual diagram of these equations is available in Figure S3.

125

## 126 *V. Sensitivity of Turnover to Calculation Assumptions*

- 127 1) Non-Steady State MBC Pool
- 128 When the MBC pool is not at steady state,  $\Delta MBC_g$  is divided over turnover
- 129 and net microbial growth. So, ΔMBC<sub>a</sub> is calculated as before (eq. 2). A portion of this
- 130 C is added to the MBC pool (ΔMBC<sub>n</sub>),

$$
131 \quad \Delta MBC_n = \alpha * \Delta MBC_g \tag{12}
$$

132 while the remainder is lost as  $CO<sub>2</sub>$  due to turnover:

133 
$$
\Delta MBC_g - \Delta MBC_n = (1 - \alpha)\Delta MBC_g = \tau * MBC
$$
 (13).

134 When  $\alpha$ =1, all MBC produced is added to the existing MBC, and no C is

135 available for turnover. When  $\alpha$ =0, then all MBC that is produced is turned over

- 136 (steady state assumption described above). When  $\alpha$ <0, a net decline in MBC occurs,
- 137 and more C is available for turnover than is produced.
- 138 Under these conditions,  $R_{\tau}$  is calculated as

$$
139 \qquad \tau = \tau * MBC = (1 - \alpha) \Delta MBC_g \tag{14}
$$

$$
140 \qquad R_{\tau} = (1 - \alpha) \left( \frac{MGE}{1 - MGE} \right) * R_g \tag{15}
$$

141 
$$
R = R_g + (1 - \alpha) \left( \frac{MGE}{1 - MGE} \right) * R_g
$$
 (16).

142 
$$
R_g = \frac{R}{1 + (1 - \alpha)(\frac{MGE}{1 - MGE})} = \frac{R(1 - MGE)}{1 - \alpha * MGE}
$$
 (17).

143 
$$
R_{\tau} = R - R_g = R(1 - \frac{(1 - MGE)}{1 - \alpha * MGE})
$$
 (18).

$$
144 \quad \tau = \frac{R_{\tau}}{MBC + 0.5 * DMBC_n} = \frac{R}{MBC + 0.5 * DMBC_n} (1 - \frac{1 - MGE}{1 - \alpha * MGE}) \tag{19}
$$

We assessed the sensitivity of our results to the assumption that MBC was at steady  
state, by calculating turnover rate assuming that there had been a 20 % increase in  
MBC over our incubation period. This is the same as testing 
$$
\alpha
$$
=0.2. There was no  
significant difference between microbial turnover rate calculated with the  
assumption of steady state MBC and microbial turnover rate calculated with an  
assumed 20 % increase in MBC (Figure S4).

## 152 2) Fate of C from Turnover

153 In the following equations, MBC that is turned over is respired  $(R_{\tau})$  or added 154 to the SOC pool  $(ASOC)$ 

$$
155 \t\tau * MBC = \Delta MBC_g = R_{\tau} + \Delta SOC \tag{20}
$$

156 Introducing f as the fraction of microbial C that is being turned over into  $CO<sub>2</sub>$ , and (1-157 *f*) as the fraction of microbial C that is turned over to dead SOC yields:

$$
158 \t R\tau = f * \tau * MBC = f * \Delta MBC_g \t (21).
$$

$$
159 \t R_{\tau} = f\left(\frac{MGE}{1 - MGE}\right) * R_g \t\t(22).
$$

$$
160 \t R = R_g + f(\frac{MGE}{1 - MGE}) * R_g \t (23).
$$

161 
$$
R_g = \frac{R}{1 + f\left(\frac{MGE}{1 - MGE}\right)} = \frac{R(1 - MGE)}{1 - (1 - f)MGE}
$$
 (24).

162 
$$
R_{\tau} = R - R_g = R(1 - \frac{(1 - MGE)}{1 - (1 - f)MGE})
$$
 (25).

163 and

164 
$$
\tau = \frac{R_{\tau}}{f * MBC} = \frac{R}{f * MBC} \left( 1 - \frac{(1 - MGE)}{(1 - (1 - f)MGE)} \right)
$$
 (26).

165 We assessed the sensitivity of our results to the assumption that all turned over 166 C is released as  $CO<sub>2</sub>$ , by comparing calculated turnover rates under the "all C to  $CO<sub>2</sub>$ " 167 condition  $(f=1)$ , with "all C to SOC pool"  $(f=0)$  and "C going for 50% to CO2 and 50% 168 to SOC" ( $f=0.5$ ). When all or half the microbial turnover is directed to SOC, the 169 calculated turnover rates are higher and the relationship with temperature is stronger 170 (Figure S5). The assumption made in this experiment, that all turned over MBC goes 171 to CO<sub>2</sub>, represents the most conservative estimate of microbial turnover.

172

## 173 *VI. Estimating Effects of Experiment Duration on MGEA*

 $174$  In most studies<sup>8-10</sup>, MGE is determined by adding a stable or radioactive isotope 175 labeled substrate, followed by measuring the incorporation of the label into microbial 176 biomass. Here MGE is calculated as:

$$
177 \quad MGE = \frac{MEC}{s} \tag{27}
$$

or

$$
179 \quad MGE = \frac{MBC}{MBC + R} \tag{28}
$$

 where MBC is the labeled microbial C produced from the substrate-C added (S), and R as 181 the labeled C respired as  $CO<sub>2</sub>$ . As pointed out by Frey et al.<sup>11</sup>, the two definitions of MGE are similar, unless a portion of S remains in the soil solution, or if some of the initial

- labeled MBC ends up as dead organic matter, but is not released as CO2. We make the
- 184 assumption that all S is taken up and turned into MBC at t=0 with an instantaneous MGE
- (*MGEI*) = 0.72. At time 0, MBC equals:

$$
186 \quad MBC_0 = MGE_I * S \tag{29}
$$

- However, as soon as new microbial biomass is produced, it becomes susceptible to
- 188 turnover (either viruses, grazing or natural senescence). So  $MBC_1$  at t=1 becomes

189 
$$
MBC_1 = MBC_0 - \tau * MBC_0 = (1 - \tau) * MBC_0
$$
 (30)

190 with  $\tau$  as the turnover rate (fraction of biomass that dies and is returned as  $CO_2$  to the

191 atmosphere and /or remains in the soil as dead organic matter). At  $t=2$ , MBC<sub>2</sub> becomes

192 
$$
MBC_2 = (1 - \tau) * MBC_1 = (1 - \tau)^2 * MBC_0
$$
 (31)

193 So, at  $t=n$ 

194 
$$
MBC_n = (1 - \tau)^n * MBC_0 = (1 - \tau)^n * MGE_l * S
$$
 (32)

and

196 
$$
MGE_A = \frac{((1-\tau)^n * MGE_I * S)}{S} = (1-\tau)^n * MGE_I
$$
 (33)

#### *VII. Statistical Analyses*

We performed a multifactor ANOVA on all experimental data using soil type and

 incubation temperature as main factors. In two cases, the metabolic model could not find matches with the observed isotopomer ratios. This was the case with one of the replicates 202 of mineral soil at 10  $\degree$ C and mineral soil incubated at 15  $\degree$ C, reducing the number of replicates to 5 for modeled metabolic rates and MGE for these treatments. Microbial biomass C and N data were log-transformed to meet the assumptions for ANOVA; the microbial biomass N data had one outlier that was excluded from statistics. The calculated microbial turnover and microbial production data were analyzed using a regression analyses on the means from the bootstrap resampling against temperature. Sensitivity analyses of assumptions used to calculate turnover were done using 95 % CI of calculated turnover rates within each soil x temperature combination and using an ANOVA on the regression of the mean turnover rates.

## *VIII. Microbial Enzyme Model*

 We made two modifications to the Allison-Wallenstein-Bradford (AWB) microbial 214 model version in Li et al.<sup>12</sup>. Instead of being constant, we made the microbial turnover 215 rate  $(\tau_B)$  an Arrhenius function of temperature:

$$
\tau_B(T) = \tau_{B,ref} * \exp\left[\frac{-E a_\tau}{R} * \left(\frac{1}{T} - \frac{1}{T_{ref}}\right)\right]
$$
\n(34)

216 where *T* is temperature (K),  $\tau_{B,ref}$  is the microbial turnover rate at the reference

temperature  $T_{ref}$  (20°C or 293 K), R is the ideal gas constant (8.314 J mol<sup>-1</sup> K<sup>-1</sup>) and *Ea*<sub>t</sub> is the activation energy for microbial turnover (Table S3). The Arrhenius equation was used as we expect sensitivity of microbial turnover to temperature to be driven by biological or biochemical processes that do not usually respond linearly. The second modification was

221 to introduce a coefficient  $\theta$  that determines the fraction of microbial turnover that enters 222 soil carbon pools versus being respired to  $CO<sub>2</sub>$ . When  $\theta = 1$ , all dead microbial biomass 223 enters soil carbon pools, and when  $\theta = 0$ , all dead biomass is respired to CO<sub>2</sub>. Although  $\theta$ 224 can vary, it was set to zero for all analyses reported here. The model equations are given 225 below.

226 Microbial biomass  $(B)$  increases with DOC  $(D)$  uptake  $(F_U)$  times microbial 227 growth efficiency (*MGE*) and declines with death  $(F_B)$  and enzyme production  $(F_E)$ :

$$
\frac{dB}{dt} = F_U * MGE - F_B - F_E \tag{35}
$$

228 where assimilation is a Michaelis-Menten function scaled to the microbial biomass pool 229 size:

$$
F_U = \frac{V_U * B * D}{K_U + D} \tag{36}
$$

230 and where *EC* is a linear function of temperature with intercept *MGE,ref* and slope *m*:

$$
E_C(T) = MGE_{ref} + m * (T - T_{ref})
$$
\n
$$
(37)
$$

231 Microbial biomass turnover is modeled as a first-order process with the temperature-

232 sensitive rate constant  $\tau_B$ :

$$
F_B = \tau_B * B \tag{38}
$$

233 Enzyme production is modeled as a constant fraction  $(\tau_E)$  of microbial biomass:

$$
F_E = \tau_E * B \tag{39}
$$

234 Temperature sensitivities for *V*,  $V_U$ , *K*, and  $K_U$  follow the Arrhenius relationship as in Eq.

235 34. CO<sub>2</sub> respiration is the fraction of DOC that is not assimilated into MBC plus the

236 respired fraction of microbial biomass turnover:

$$
C_R = F_U * (1 - MGE) + F_B * (1 - \theta)
$$
\n(40)

237 The enzyme pool (*E*) increases with enzyme production and decreases with enzyme 238 turnover:

$$
\frac{dE}{dt} = F_E - F_L \tag{41}
$$

239 where enzyme turnover is modeled as a first-order process with a rate constant  $\tau_L$ :

$$
F_L = \tau_L * E \tag{42}
$$

240 The SOC pool (*S*) increases with external inputs and a fraction of dead microbial biomass 241 ( $a_{BS} * \theta$ ) and decreases due to decomposition losses:

$$
\frac{dS}{dt} = I_S + F_B * a_{BS} * \theta - F_S \tag{43}
$$

242 where decomposition of SOC is catalyzed according to Michaelis-Menten kinetics by the 243 enzyme pool:

$$
F_S = \frac{V * E * S}{K + S} \tag{44}
$$

244 The DOC pool receives external inputs, a fraction of dead microbial biomass, the

245 decomposition flux, and dead enzymes, while assimilation of DOC by microbial biomass

246 is subtracted:

$$
\frac{dD}{dt} = I_D + F_B * (1 - a_{BS}) * \theta + F_S + F_L - F_U \tag{45}
$$

247 The steady-state analytical solutions for SOC, DOC, MBC, and ENZC are given here:

$$
S = \frac{-K * \tau_L * (x_1 * \tau_E + x_2 * MGE * \tau_B * \theta - I_S * \tau_B)}{\tau_E * ((I_S + I_D) * MGE * V + x_1 * \tau_L) + \tau_L * \tau_B * (x_2 * MGE * \theta - I_S)}
$$
(46)

248 where

$$
x_1 = I_S * (MGE - 1) \text{ and } x_2 = I_S - a_{BS} * (I_D + I_S)
$$
\n(47)

$$
D = \frac{-K_U * (\tau_B + \tau_E)}{\tau_B + \tau_E - MGE * V_U}
$$
\n
$$
(48)
$$

$$
B = \frac{MGE * (I_D + I_S)}{(1 - MGE) * \tau_E + \tau_B * (1 - MGE * \theta)}
$$
(49)

$$
E = \frac{B * \tau_E}{\tau_L} \tag{50}
$$

249 To generate Figure 4, we calculated the relative change in steady-state solutions for SOC 250 and MBC at from 0 to 20<sup>o</sup>C under three scenarios with  $\theta = 0$ , meaning that all microbial 251 turnover is respired as  $CO_2$ . For the "constant turnover, constant MGE" scenario,  $Ea_0 = 0$ 252 kJ mol<sup>-1</sup> and  $m = 0$ . For the "constant microbial turnover, declining MGE" scenario,  $Ea_1 =$ 

- 253 0 kJ mol<sup>-1</sup> and  $m = -0.016$  °C<sup>-1</sup>. For the "increasing microbial turnover, constant MGE"
- 254  $Ea = 47 \text{ kJ} \text{ mol}^{-1} \text{ and } m = 0.$

### **SUPPLEMENTARY NOTE**

## *Thermodynamic Limits to MGE.*

 The value of MGE in soil microbial communities is important for our understanding of soil C cycling processes. This value is explicitly or implicitly part of 259 soil C cycling models, usually a constant value is used<sup>8,13,14</sup> ranging from 0.15 to 0.60 (ref. 13).

 Efforts to predict MGE from thermodynamic and chemical principles have 262 been ongoing for several decades<sup>15,16</sup>. Experimental data are mostly limited to pure culture studies where substrate availability is usually high relative to substrate 264 availabilities in natural environments. MGE is limited by thermodynamic constraints<sup>14</sup>. The theoretical maximum value of MGE was calculated in several studies. The thermodynamic maximum yield can be predicted from the ratio between the degree of 267 reduction of the substrate (e.g., glucose  $\gamma_s = 4$  or formate  $\gamma_s = 2$ ) and product, ( $\gamma_p$  biomass ~  $\pm$  4.2)<sup>17</sup>. The MGE<sub>max</sub> is about 0.95 (ref. 13, 16). A second theoretical maximum is defined 269 by the cost of making new biomass. This yields an MGE<sub>max</sub> from glucose of about 0.88 (ref. 4). The observed MGE values in this study (MGE ranged from 0.67 to 0.75) were lower than the MGEmax values identified above, but higher than the average 272 thermodynamic efficiency in pure culture studies<sup>13,14,16</sup>. Variability of yield (MGE) in culture studies spans almost two orders of magnitude from 0.01 to 0.8 (ref. 15-19) associated with species differences, substrate type and concentration, and environmental factors. Experimental values for MGE in pure culture studies are always lower than the theoretical maximal yield values described 277 above. The ratio between experimentally observed MGE<sub>max</sub> and the theoretical MGE<sub>max</sub>

- 278 is called the thermodynamic efficiency<sup>13-16</sup>. This value is used as a "first approximation"
- 279 according to Roels  $1980^{16}$ , and should not be mistaken for a theoretical thermodynamic
- 280 maximal yield, as higher values have been observed in pure culture studies<sup>15,16,18,19</sup> and
- 281 soil and aquatic ecosystems (ranging from close to zero to  $>0.8$  for both environments<sup>13</sup>).

# **SUPPLEMENTARY TABLES**

- **Table S1**. Microbial biomass C (MBC) and microbial biomass N (MBN) means and
- standard error for four incubation temperatures and two soil types (n=4). Microbial

286 biomass was calculated using  $k_{EC}$  = 0.45 and  $k_{EN}$  = 0.54.



	$Cu/C1$ Ratio of Glucose		$C_1/C_{23}$ Ratio of Pyruvate	
Temperature (°C)	<b>Mineral Soil</b>	Organic Soil	<b>Mineral Soil</b>	Organic Soil
5	$3.21 \pm 0.31$	$2.92 \pm 0.24$	$4.41 \pm 0.40$	$3.56 \pm 0.20$
10	$3.26 \pm 0.15$	$2.91 \pm 0.14$	$3.38 \pm 0.20$	$3.73 \pm 0.35$
15	$3.60 \pm 0.39$	$2.94 \pm 0.16$	$3.75 \pm 0.30$	$3.51 \pm 0.19$
20 291	$2.70 \pm 0.12$	$3.15 \pm 0.26$	$3.94 \pm 0.46$	$3.61 \pm 0.31$

 **Table S2.** Glucose and pyruvate isotopomer ratios for mineral and organic soil at 5, 10, 290 15, and 20 °C (means  $\pm$  standard error, n=6 except organic soil at 5 °C; n=5).

Parameter	Description	Value	<b>Units</b>
$T_{ref}$	Reference temperature	20	$\rm ^{o}C$
$MGE_{ref}$	MGE at reference temperature	0.31	
$\,m$	MGE change with temperature	$[0, -0.016]$	$^{\circ}$ C <sup>-1</sup>
$I_{\mathcal{S}}$	SOC input rate	0.00015	mg C g <sup>-1</sup> soil h <sup>-1</sup>
$I_D$	DOC input rate	0.00001	mg C $g^{-1}$ soil h <sup>-1</sup>
$V_{ref}$	SOC reference $V_{max}$	$\mathbf 1$	mg C mg <sup>-1</sup> C h <sup>-1</sup>
$V_{U,ref}$	DOC uptake reference $V_{max}$	0.01	mg C mg <sup>-1</sup> MBC $h^{-1}$
$K_{ref}$	SOC reference $K_m$	250	mg $C g^{-1}$ soil
$K_{U,ref}$	DOC uptake reference $K_m$	0.26	mg $C g^{-1}$ soil
$\tau_{B,ref}$	Reference MBC turnover rate	0.00028	mg C mg <sup>-1</sup> C h <sup>-1</sup>
$Ea_V$	SOC $V_{max}$ activation energy	47	kJ mol <sup>-1</sup> $K$ <sup>-1</sup>
$Ea_{VU}$	Uptake $V_{max}$ activation energy	47	kJ mol <sup>-1</sup> $K$ <sup>-1</sup>
$Ea_K$	SOC $K_m$ activation energy	30	kJ mol <sup>-1</sup> $K$ <sup>-1</sup>
$Ea_{KU}$	Uptake $K_m$ activation energy	30	kJ mol <sup>-1</sup> $K$ <sup>-1</sup>
$Ea_{\tau}$	MBC turnover activation energy	47	kJ mol <sup>-1</sup> $K$ <sup>-1</sup>
$\tau_E$	Enzyme production rate	$5.6 \times 10^{-6}$	mg C mg <sup>-1</sup> MBC $h^{-1}$
$\tau_L$	Enzyme loss rate	0.001	mg C mg <sup>-1</sup> C $h^{-1}$
$a_{BS}$	Fraction of dead MBC partitioned to SOC	0.5	
$\theta$	Fraction of dead MBC transferred to soil pools	$\mathbf 0$	

292 **Table S3**. Microbial- enzyme model parameter descriptions, values, and units.



297 *Figure S1.* Response of a) soil respiration rate and b) specific respiration rate of mineral 298 and organic soil to temperature (means and se; some standard errors are smaller than data 299 points).



 *Figure S2.* Model for metabolic processes in soil microbial communities. Flux rates (v2- v21) are normalized relative to glucose uptake (v1, set at 100 moles). Insert depicts details of the pentose phosphate pathway. Abbreviations: G6P, glucose-6-phosphate; F1,6P, fructose-1,6-phosphate; GAP, glyceraldehyde-phosphate; PYR, pyruvate; ACCO, acetyl-CoA; ICIT, isocitrate; αKG, α-ketoglutarate; OAA, oxaloacetate; RU5P, ribulose- 5-phosphate; S7P, sedoheptulose-7-phosphate; E4P, erythrose-4-phsophate. Reprinted from ref. 1.



311 *Figure S3.* Schematic showing relationships between measured (total respiration, MGE,

312 and MBC pool) and calculated values (production rate and turnover rate). Total

313 respiration rate is the sum of respiration from turnover and growth  $(R = R_g + R_t)$ .





*Figure S4.* Effect of change in MBC on calculated microbial turnover rates at each

temperature for a) mineral soil and b) organic soil (means and 95% CI).





*Figure S5.* Effect of changing the fate of turned over MBC on calculated microbial



# **SUPPLEMENTARY REFERENCES**



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