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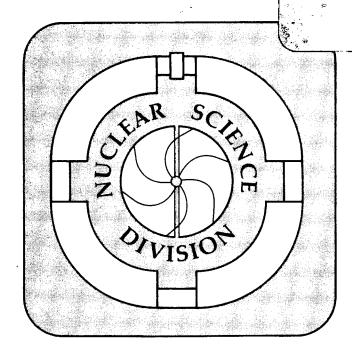
SEARCH FOR THE BETA-DECAY OF 180 Lu TO 180 Hf $^{\rm m}$

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August 1986

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SEARCH FOR THE BETA-DECAY OF 180 Lu TO 180 Hfm

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The $t_{1/2}=5.7$ minute ¹⁸⁰Lu isotope was produced in the ¹⁸⁰Hf(n,p) reaction and its subsequent β decay back to ¹⁸⁰Hf was studied with a Ge(Li) spectrometer. A radiochemical technique was used to measure the fractional population of the I^{π} , K=(8⁻,8) isomer in ¹⁸⁰Hf to be $f_m=0.005\pm0.018$ %. The limit on f_m is shown to be too small to account for the nucleosynthesis of ¹⁸⁰Ta^m in the r-process. The possible existence of a high-spin isomer in ¹⁸⁰Lu and its astrophysical consequences are discussed.

INTRODUCTION

Attempts have been made in recent years to explain the abundance of the naturally-occurring high-spin isomer $^{180}\mathrm{Ta}^\mathrm{m}$ in terms of the slow (s) and/or rapid (r) neutron capture processes that account for the bulk of heavy element nucleosynthesis in stars. 1,2 Beer and Ward have suggested that $^{180}\mathrm{Ta}^\mathrm{m}$ may be produced through a weak β -decay branch of $^{180}\mathrm{Hf}^\mathrm{m}$, which is known to be populated in the s-process. Fig. 1 summarizes the salient features of this theory. An r-process contribution to the abundance of $^{180}\mathrm{Ta}^\mathrm{m}$ requires that (1) a fraction f_m of the β decays of the $I^\mathrm{m} = (3^-, 5^+)$ $^{180}\mathrm{Lu}$ leads to the population of the $I^\mathrm{m} = 8^-$ isomer in $^{180}\mathrm{Hf}$, and (2) a fraction f_β of the resulting $^{180}\mathrm{Hf}^\mathrm{m}$ population β decays to the $I^\mathrm{m} = 9^-$ long-lived isomer in $^{180}\mathrm{Ta}$. If atomic ionization effects are neglected, the solar abundance data compiled by Cameron and the neutron capture cross-section work of Beer and Macklin can be used to extract the r-process component of the observed solar abundance of $^{180}\mathrm{Ta}^\mathrm{m}$:

$$\left| \frac{\frac{N_r}{N_o}}{N_o} (^{180}\text{Ta}^{m}) \right| = 1.2 \pm 0.2 \times 10^4 \text{ f}_m \text{ f}_\beta.$$
 [1]

Based on log-ft values for similar decays in neighboring nuclei, f_{β} is expected to fall between 0.1% and the recently established experimental upper limit of 0.35%.^{6,7} Numerically, therefore, a 100% r-process origin for ¹⁸⁰Ta^m requires 2% < $f_{\rm m}$ < 8%.

To date ¹⁸⁰Lu is the most neutron rich member observed on the A=180 isobar. Relatively little is known of its structure. The $t_{1/2} = 5.7$

minute ground state of this prolately-deformed nucleus is most commonly given an I^{π} , $K=(3^-,3)$ assignment, 8 , 9 though Ward and D'Auria argue for an I^{π} , $K=(5^+,5)$ assignment. 10 The subsequent β transitions of 180 Lu to states between 1.2 and 2.2 MeV of excitation in 180 Hf have been the subject of a number of investigations. 8 , 9 , 10 , 11 , 12 These half dozen states in 180 Hf are believed to have K=2, 3, or 4. All are seen to γ cascade $\approx 100\%$ of the time to the 2^+ and 4^+ components of the K=0 ground state rotational band. To an intensity of perhaps a percent, no strength has been observed to the well-studied I^{π} , $K=(8^-,8)$ isomer at 1142 keV in 180 Hf. In fact, no γ transition has been observed which feeds 180 Hf m , though it is readily produced in (n,γ) and (n,n') reactions. 13 , 14

The present study uses a straightforward radiochemical separation technique following the production of ¹⁸⁰Lu and detects the subsequent ¹⁸⁰Hf^m decay. Earlier we presented a limit of $f_m < 0.06 \ \%$. ¹⁴ Eschner et al. have subsequently reported a measurement of $f_m = 0.46 \pm 0.15 \ \%$ using a multinucleon transfer technique. ¹⁵ We present here our final results along with a possible explanation for the discrepancy between our measurement and that of Eschner et al.

EXPERIMENT

Bulk samples of natural Hf metal with masses between 1 and 2 grams were first activated with thermal neutrons at the University of Washington reactor to introduce the long-lived ($t_{1/2} = 42 \text{ day}$) isotope ¹⁸¹Hf as a radiochemical tracer. After a suitable cooldown period of 1 to 4 weeks, the "spiked" Hf samples were wrapped in Cd foils and inserted behind the water-cooled Be target/beam-stop at the University of Washington cyclotron. The Hf samples were bombarded with fast neutrons generated by the Be(d,n) reaction using 22 MeV deuterons. Following an irradiation period of 5 to 10 minutes, the samples were expeditiously transported to a radiochemistry laboratory. The desired 180 Lu, along with other rare earth activities, was extracted by dissolving the Hf sample in HF acid and precipitating out LuP, with the addition of Y as a carrier. precipitation was accelerated by the use of a centrifuge and a radiochemical "hafnium reduction factor" of 104 was achieved in less than a ¹⁸⁰Lu halflife (5 minutes). Geometrically similar samples containing about 2 ml of solution and precipitate were prepared. The precipitate was first counted with a 135 cm Ge(Li) detector in a reproducible geometry to obtain the abundance of 180 Lu. Several hours later, the solution was counted with the same detector to obtain the abundance of 160 Hf (produced directly in the 180 Hf(n,n') reaction) and of the 181 Hf tracer. precipitate was recounted in 1 hour time bins to determine the residual abundances of "" and "" Hf. The Ge(Li) detector was shielded for lowlevel counting throughout the experiment and its resolution was 1.5 keV

(FWHM) at 500 keV. Efficiency measurements were performed with standard sources in the experimental geometry.

Many activations were performed using Hf metal (containing 3% Zr) and "spectrographic-grade" Hf (156ppm Zr). Typical γ spectra between 275 and 525 keV at each stage of counting are shown in Fig. 2 for a Hf metal run. Table I lists the activities that were identified in the precipitate throughout the course of the experiment. Most of the activity between 5 and 10 hours after the fast neutron bombardment proved to be from the Y and Sr products of Zr(n,p) and $Zr(n,\alpha)$. While the spectrographic-grade runs were free of contaminating lines, the Hf reduction factor could not be pushed above \approx 1000.

DERIVATION OF f AND RESULTS

The final abundance of $^{180}\mathrm{Hf}^{\mathrm{m}}$ in the target has two sources. The first comes from prompt production reactions like inelastic scattering ($^{180}\mathrm{Hf}(\mathrm{n,n'})$) and thermal neutron capture ($^{179}\mathrm{Hf}(\mathrm{n,\gamma})$). The smaller second source comes from the fraction $\mathrm{f_m}$ of the decay of $^{180}\mathrm{Lu}$.

$$\left[N \left(^{180}Hf^{m} \right) \right] = \left[N \left(^{180}Hf(n,n') \right) \right] + f_{m} \left[N \left(^{180}Lu \right) \right]$$
 [2]

As described in the previous section, the radiochemistry employed separates activities which are precipitated out of HP acid from those left in solution. The Hf reduction factor, ξ , is defined as the ratio of solution (sn) to precipitate (pt) abundance. A quantity η is similarly defined to describe the effectiveness with which Lu is recovered in the precipitate:

$$\xi = \frac{\left[N_{sn}^{(181}Hf)\right]}{\left[N_{pt}^{(181}Hf)\right]}, \qquad \eta = \frac{\left[N_{pt}^{(180}Lu)\right]}{\left[N_{sn}^{(180}Lu)\right]}$$
[3]

where continuity [N] = [N_{pt}] + [N_{sn}] is assumed. A general figure of merit for the success of the radiochemistry in this experiment is $\eta\xi/(\eta+1)$. Typically, $\xi \approx 1000$ to 10000 and $\eta/(\eta+1) \approx 50$ % to 90% were achievable. The quantities of ¹⁸⁰Hf isomer which end up in the precipitate and solution can then be written as

Combining these two expressions to eliminate $[N(^{180}Hf(n,n'))]$ and then

substituting the ratio of tracer activities in [3] for ξ , f_m is given by:

$$f_{m} = (1+\delta) \left\{ \frac{\left[N_{pt}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]}{\left[N_{pt}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]} + \frac{\left[N_{pt}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]}{\left[N_{pt}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]} - \frac{\left[N_{sn}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]}{\left[N_{sn}^{(1+\delta)} + F_{pt}^{(1+\delta)} \right]} \right\}$$
 [5]

where $(1+8) = (1-1/\eta \xi)^{-1}$. 8 is of the order 10^{-4} and is subsequently neglected. The five bracketed abundances were measured by integrating photopeak areas in their most favorable time-windows $(t_{sta}, t_{sta} + t_{dur})$ and extrapolating back to an initial time t_o . The initial time was taken to be the last step in the radiochemical separation. Because f_m is expressed in terms of ratios, only relative γ line intensities and Ge(Li) detector efficiencies were needed. Furthermore, these only affect the scaling of the final result. The 408- and 482-keV γ lines, in 180 Lu and 181 Hf were used as the primary lutetium and tracer reference lines. The 332- and 443-keV γ lines in 180 Hf were used separately in the analysis. For the 443-keV line then

$$f_{m} = \frac{i_{\gamma} e_{\beta} e_{\beta}(408)}{i_{\gamma} e_{\beta}(443)} \left\{ \frac{[FN_{pt}(482)]}{[FN_{pt}(408)]} \right\} \left\{ \frac{[FN_{pt}(443)]}{[FN_{pt}(482)]} - \frac{[FN_{sn}(443)]}{[FN_{sn}(482)]} \right\}$$
[6]

where

$$\mathbf{F} = \left\{ \begin{bmatrix} -\lambda t \\ e \end{bmatrix} \begin{bmatrix} -\lambda t \\ 1 - e \end{bmatrix}^{-1} dur \right\}^{-1}$$

is the inverse of the fraction of total decays which occur in the time window (t_{sta} , t_{sta} + t_{dur}), i_{γ} is the absolute gamma line intensity, ϵ_{p} is the measured photopeak efficiency of the Ge(Li) detector, and ϵ_{s} is the correction for γ -cascade summing out of the photopeak.

Table II lists the photopeak areas for three independent runs, each analyzed for both the 332- and 443-keV γ lines. The limitation on the Hf metal runs (I and II) is purely statistical while the limitation on the spectrographic-grade run (III) is probably systematic since it is difficult to extract photopeak areas with an accuracy greater than 1%. Table III gives the six determinations of f_m . We combine only the 4 values of f_m from the Hf metal runs to obtain $f_m = 0.005 \pm 0.018\%$. To this is added a scaling uncertainty of 10%, mostly due to the uncertainty in $i_{\gamma}(408)$ to obtain a firm upper limit (67% C.L.) on f_m of 0.026%.

DISCUSSION

The measurement by Eschner et al. of $f_m=0.46\pm0.15\%^{15}$ is clearly incompatible with the null result reported here. There are two likely explanations for this discrepancy. (1) An error in experimental technique was made by one of the groups, or (2) a high-spin, short-lived isomer of Lu exists to which the experiment of Eschner et al. was more sensitive. It will be helpful to compare the technique of Eschner et al. with our own. It will also prove worthwhile to explore the second explanation for its nuclear and astrophysical consequences. In the following text, superscripts "g" and "m" are used to differentiate between quantities which refer to the ground state ($t_{1/2}$ =5.7 minutes) and the proposed short-lived isomeric state in 180 Lu.

Both experiments are similar in their use of a Ge(Li) detector to measure the activities of ¹⁸⁰Lu, ¹⁸⁰Hf^m and a "tracer" Hf isotope, but differ in the production reaction used and in the method employed to subtract the directly produced ¹⁸⁰Hf^m. In the experiment of Eschner et al., a beam of 8.6 MeV/u ¹³⁶Xe was directed onto a W target/ion source. ¹⁸⁰Lu (and other rare—earth reaction products) diffused out of their target on a time scale of 10 seconds and was then run through a mass spectrometer to a catcher foil for online and later offline counting. Discrimination against the promptly produced ¹⁸⁰Hf^m relied on (1) different Lu and Hf diffusion time constant, (2) different Lu and Hf ionization potentials, and (3) relative production cross—sections for ¹⁸⁰Hf^m and ¹⁷⁷Hf^m which played the role of the "tracer". ¹⁷ Eschner et al. reported a fast release time

constant for lutetium $\alpha_{Lu}^{=5}^{+5}_{-2}$ x10⁻² sec⁻¹ which corresponds to a half-life against release of between 7 and 23 seconds. $\alpha_{Hf}^{=7.4\pm1.0}$ x10⁻⁴ sec⁻¹ was much slower.

Both groups have proffered the possible existence of a high-spin, short-lived isomer in 180 Lu to explain the discrepancy in the measurement of f_m . 14,15 With this assumption, the 5.7 minute ground state need contribute nothing to the population of 180 Hf m (f_m^g =0), consistent with our results, while the proposed isomer must have a high-spin to allow $f_m^m > 0.5\%$, consistent with the measurement of Eschner et al. The isomeric production fraction

$$P_{m} = \frac{\sigma_{m}}{\sigma_{q} + \sigma_{m}} \tag{7}$$

is undoubtedly different in the two experiments, and is probably greater for the multinucleon transfer reaction. In fact, however, the (n,p) reaction readily populates high spin states. In the $^{178}\mathrm{Hf}(n,p)$ reaction, for example, the isomeric production fraction for the 9 isomeric states is $P_{m}=0.22$ as shown in Table 1.

The proposed isomer of ¹⁸⁰Lu should have a halflife in the range from 1 to 100 seconds, long enough to be recovered by the diffusion method of Eschner et al. but short enough so that it has transmuted into ¹⁸⁰Lu^g and ¹⁸⁰Hf^m during the 5 minute radiochemical separation in our experiment. This is possible if f_m^m competes not against β transitions that lead to the ground state of ¹⁸⁰Hf, but rather with γ transitions that feed the 5.7

minute 180 Lu 9 . It is reasonable then to express f_{m}^{m} as a ratio of partial β and γ transition rates.

$$f_{m}^{m} = \frac{\lambda_{\beta}^{m}}{\lambda_{\gamma}^{m} + \lambda_{\beta}^{m}}$$
 [8]

If the proposed ¹⁸⁰Lu isomer decays most of the time by low energy, heavily converted EM transitions, it may have escaped direct detection and gone unnoticed in previous experiments. Indeed the shape of the on-line build-up curve for ¹⁸⁰Lu displayed in Fig. 1 of Ref. 15 is identical for the 10 second release halflife described by Eschner et al. as it would be for the infeeding of ¹⁸⁰Lu^g from a 10 second halflife isomer.

Fig. 3 pictorially summarizes the results from the two experiments. It displays the limits on the product of P_m and f_m^m as a function of the halflife of the proposed isomer of ¹⁸⁰Lu. Our experiment excludes halflives greater than about 100 seconds, while that of Eschner et al. implies that the halflife must be greater than about 1 second. The shaded region is allowed by both experiments. The dashed line in Fig. 3 represents a solution for f_m^m assuming that the production fraction $P_m = 0.5$ and the partial β halflife $t_{1/2}^m = 42$ minutes. Such a halflife corresponds to a log-ft of 6.4 for a direct β transition to ¹⁸⁰Hf $_m^m$, which is a reasonable estimate for the strength of an allowed hindered transition. ¹⁸ The dashed line passes through the shaded region for isomeric halflives between 20 and 40 seconds. Finally, the data points at 5.7 minutes represent the measurements of Eschner et al. and ourselves for f_m^g with the assumption that the proposed isomer of ¹⁸⁰Lu does not exist (or $P_m = 0$).

There are theoretical reasons to believe that a low-lying, high-spin isomer could exist in 180 Lu. By coupling the known odd proton and neutron quasiparticles states near their respective Fermi surface it is possible to construct both low-spin (I=1-5) and high-spin (I=7-10) states. The region of isotopes around 180 Lu is filled with systematic examples of such lowand high-spin K-isomers. In terms of the unified model, the $t_{1/2} = 5.7$ minute ground state of 180 Lu is most probably the $I^{\pi}, K=(3^{-},3)$ coupling of the odd proton 7/2 [404] and the odd neutron 1/2 [510], quasiparticles. 8,9 Ward and D'Auria prefer an I^{π} , K=(5⁺,5) assignment, coupling 9/2 [514] with 1/2 [510]... These states are seen within 20 keV of each other in 182 Ta. The $7/2^{+}$ [404], and $9/2^{+}$ [624], states can couple with their particle spins aligned to form the low-spin 1 state observed as the ground states of 176 Lu, 178 Lu, and 180 Ta. With their particle spins antialigned, these same orbitals couple to form an 8 state seen at 174 keV of excitation in ¹⁸⁰Ta. ¹⁷ Finally, the $9/2^{-}[514]_{\pi}$ and $9/2^{+}[624]_{\nu}$ orbitals can couple to form the high-spin 9 isomers seen at 73 keV of excitation in 180 Ta and at about 300 keV in 176 Lu. Hoff has recently calculated that the 8 and 9 states should lie within 200 keV of the 3 ground state of 180 Lu. 19 Hoff also has placed the I" = 5 + state approximately 50 keV below these prospective 180 Lu isomer candidates.

If the 9 state is the lower of the two high-spin states in 180 Lu, and is thus the isomer, the M4 transition to the 5 state will be much slower than the allowed β transition to the 8 isomer in 180 Hf. Using the observed log-ft of 6.4 for the analogous decay of 182 Hf^m, a partial halflife of 40 minutes is calculated. Such an isomer should have been

produced easily and observed previously. Suppose, on the other hand, that the 8⁺ state plays the role of the ¹⁸⁰Lu isomer. The single particle estimate for the M3 transition to the 5⁺ level yields a partial halflife of about 10 seconds. The β -branching ratio for the high-spin isomer is then on the order of 1%, consistent with the measurement of Eschner et al. and with our null result. Other M3 transitions in the rare-earths occur a factor of 10 to 100 times slower than the single-particle rate, ²⁰ but this factor can be recovered by increasing the energy difference between the 5⁺ and 8⁺ states. We note that the prospective 8⁺ isomer cannot lie any lower than about 150 keV in energy since it is the spin antialigned member of a Gallagher-Moszkowski doublet with the 1⁺ state. ²¹ If the t_{1/2} = 5.7 minute activity of ¹⁸⁰Lu is the K^T = 3⁻ state, then the 1⁺ state must lie above it.

Beyond the circumstantial evidence discussed here, there is no direct evidence supporting the existence of a high-spin isomer in 180 Lu. Takahashi et al. reported a 2.5±0.5 minute β activity with a 3.3±0.1 MeV endpoint. This was interpreted as the β -decay from a low-spin isomer (1[†]?) directly to the ground-state of 180 Hf, but it has not been confirmed by subsequent β and γ experiments. Zychor et al. specifically searched for evidence of high-spin Lu isomers in a γ -ray singles experiment following the bombardment of nat W with 136 Xe and reported null results for halflives greater than 2 minutes. 17

ASTROPHYSICAL IMPLICATIONS

In the absence of a high-spin isomer of Lu, our measured limit on $\mathbf{f}_{\mathbf{m}}$ clearly excludes an r-process origin for $^{180}\mathrm{Ta}^{\mathbf{m}}$. The existence of an isomer would change this conclusion. Little is known about the site of the r-process which accounts for the nucleosynthesis of half of the elements heavier than iron and all of the actinides. 22 To achieve the high density of neutrons necessary to create the very neutron-rich progenitor matter probably involves a thermal transient to peak temperatures of 1-3×10 ° K. The post-r-process is characterized by a rapidly falling temperature during which time nuclei β -decay back to the valley of β stability. ¹⁸⁰Lu is probably reached within the first minute when the temperature remains high. Thermal equilibration of the high-spin isomeric and low-spin ground states will be mediated by higher lying states of intermediate spin via processes which include photoabsorption, Coulomb excitation, and positron annihilation. As the temperature continues to fall, this communication link eventually will be broken, freezing out the isomeric population. Lu would still be hindered from making direct isomeric transitions to the ground state because K- and L-shell ionization would block the internal conversion channel. (For example, $\alpha_r = 10^5$ for a 50-keV M3 transition.) Thus the β -decay branch of 180 Lum would be significantly enhanced. (f_m \approx 100%). Finally, the continuing drop in temperature increases the survivability of the high-spin isomers of 180 Hf and 180 Ta against the same processes responsible for the thermalization of Lu. 23

The β-decay yields of the isomeric and ground states of ¹⁸⁰Hf, and consequently the r-process production of ¹⁸⁰Ta^m, are sensitive to the time-integrated thermodynamics of the post-r-process. Therefore it may be possible to combine future detailed nuclear structure data for ¹⁸⁰Lu with the observed solar abundance of ¹⁸⁰Ta^m to place constraints on the time-temperature history of the post-r-process. Ultimately, such constraints may aid in the determination of the r-process site itself.

CONCLUSION

We have measured the fractional population of $^{180}\mathrm{Hf}^{\mathrm{m}}$ following the β decay of 180 Lu to be $f_{\rm m}$ = 0.005±0.018% in disagreement with the positive value reported by Eschner et al. We believe that our measurement, which employs a simpler technique and yields a null result with a much smaller statistical uncertainty, is the correct value for the 5.7 minute activity of Lu. Astrophysically, f is too small to account for the observed abundance of Tam by straightforward r-process production. We have shown that the experimental discrepancy can be reconciled if the multinucleon transfer reaction of Eschner et al. populated a high-spin, short-lived isomer in 180 Lu. Further, the existence of such an isomer may substantially enhance the resultant nucleosynthesis of ""Ta". With more complete nuclear structure information, it may be possible to establish constraints on the post-r-process astrophysical environment. Clearly more work is called for in the understanding of the doubly-odd A=180 nuclei and in the pinpointing of the r-process site before the origin and fate of Ta is known.

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TABLE I. Activities observed in the precipitate after fast neutron bombardment of natural Hf metal in Run I. A Zr content of 3% is assumed. The ratio of 180 Hf $^{m}/^{180}$ Lu was determined in a separate activation to be ≈ 30 . Excitation energies for the isomers have been subtracted from the Q-values of Ref. 16.

| OBSERVED | PRIMARY | T _{1/2} | Q-value | $\Delta I^{m{\pi}}$ | CROSS-SECTION |
|--|---------------------------|------------------|---------|-----------------------------------|--------------------------|
| ACTIVITY | REACTION | | (MeV) | | RELATIVE TO |
| | | | | | 180Hf ^m =1000 |
| a ₇ sr ^m | ⁹⁰ Zr(n,α) | 2.8 h | +1.362 | 1/2 | 22 |
| 91Sr | ⁹⁴ Zr(n,α) | 9.5 h | +2.067 | 5/2 ⁺ | 33 |
| $\mathbf{p}_{\mathbf{Y}^{\mathbf{g}}}$ | ⁹⁶ Zr(n,α) | 10.2 h | +0.170 | 1/2 | 750 |
| ao ^A w | 90Zr(n,p) | 3.19 h | -2.188 | 7 ⁺ | 280 |
| a T X w | ⁹¹ Zr(n,p) | 49.7 m | -1.318 | 2+ | 610 |
| 92 _Y | 92Zr(n,p) | 3.54 h | -2.841 | 2- | 430 |
| 94 <u>Y</u> | 94Zr(n,p) | 18.7 m | -4.22 | 2 | 160 |
| 175Yb | ¹⁷⁸ Hf(n,α) | 4.19 d | +7.905 | 7/2 | 51 |
| 177Yb ^g | 180 Hf(n, α) | 1.9 h | +6.856 | 9/2+,1/2- | 16 |
| 176Lu ^m | ¹⁷⁶ Hf(n,p) | 3.68 h | -0.533 | 1- | 63 |
| 177 _{Lu} g | ¹⁷⁷ Hf(n,p) | 6.71 đ | +0.285 | o ⁻ | 20 |
| 177Lu ^m | ¹⁷⁷ Hf(n,p) | 160.9 d | -0.685 | 8+ | . 3 |
| 178 _{Lu} g | ¹⁷⁸ Hf(n,p) | 28.4 m | -1.470 | 1+ | 56 |
| 178Lu ^m | 178Hf(n,p) | 23.0 m | -1.77 | 9- | 16 |
| ¹⁷⁹ Lu | ¹⁷⁹ Hf(n,p) | 4.9 h | -0.570 | 1+ | 420 |
| 180Lu | 180Hf(n,p) | 5.7 m | -2.52 | (3 ⁻ ,5 ⁺) | 33 |

TABLE II. Efficiencies and run by run photopeak areas used in three independent activations to determine f_m . The 332- and 443-keV ¹⁸⁰Hf^m y lines are treated separately. Only \approx 2 out of 50 ml of solution (sn) was counted to match the geometry used in the precipitate (pt) runs.

| | Activit | - Y | 190 ^{Lu} g | 180Hfm | 180Hfm | ¹⁸¹ Hf |
|--|----------------------------|---------------------|---------------------|---------------|------------|-------------------|
| | t _{1/2} | | 5.7 m | 5.518 h | 5.518 h | 42.4 d |
| | E, (kev | 7) | 408.1 | 332.3 | 443.2 | 482.0 |
| | iy | | 0.50 | 0.944 | 0.821 | 0.810 |
| | $\epsilon_{p}(E_{\gamma})$ | | 0.031 | 0.038 | 0.029 | |
| $\epsilon_{\mathbf{S}}(\mathbf{E}_{\gamma})$ | | 0.91 | 0.77 | 0.77 | | |
| RUN (pt/ | sn) t _{st} | a' ^t dur | (m) Phot | opeak Areas (| counts) | |
| I. H | f metal | | | | | |
| I.1 pt | 1.6 | 4.7 | 110270±330 | | | |
| I.2 sn | 226 | 18 | | 76260±280 | 50270±230 | 22620±150 |
| I.3 pt | 364 | 420 | | 973± 67 | 635± 48 | |
| I.4 pt | 79 4 0 | 850 | | | | 1015± 37 |
| II. H | f metal | | | | | |
| II.1 pt | 1.6 | 13.6 | 112300±350 | | | |
| II.2 sn | 101 | 63 | | 186860±450 | 121770±360 | 99230±320 |
| II.3 pt | 302 | 661 | | 280± 32 | 251± 35 | |
| II.4 pt | 5902 | 392 | | | | 2 49 ± 20 |
| III. s | pectrogra | aphic-g | rade Hf | | | |
| III.1 pt | 1.8 | 6.1 | 78940±290 | | | |
| | | | | | | |
| III.2 sn | 129 | 63 | | 103204±330 | 67554±310 | 36090±270 |

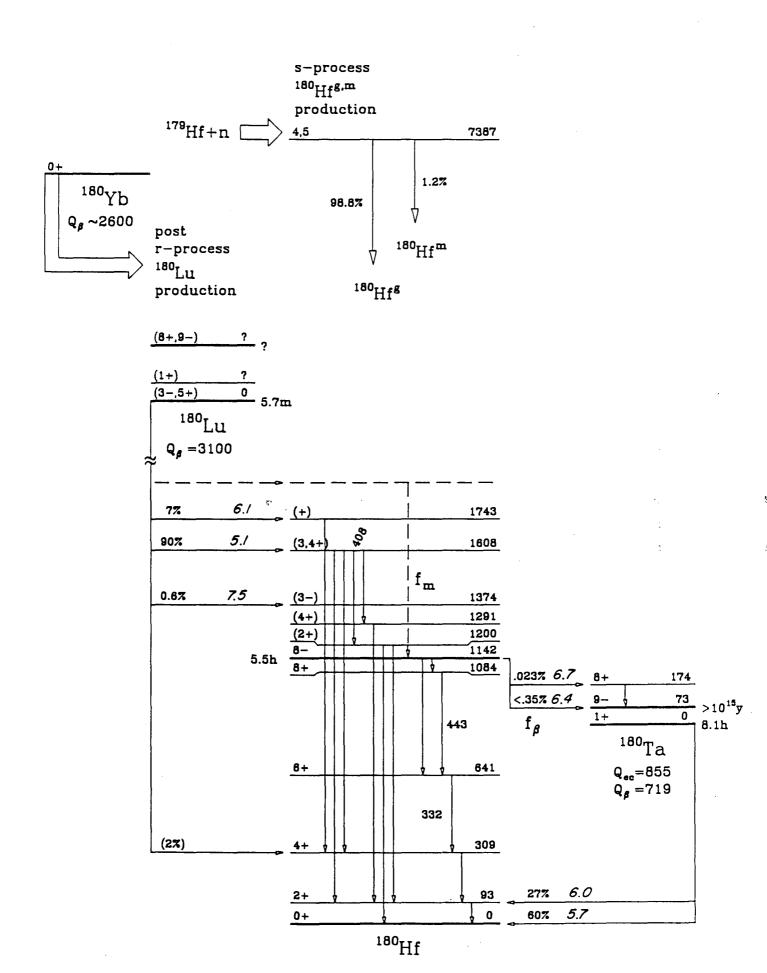
TABLE III. The fractional population, f_m , of ¹⁸⁰Hf^m following the β -decay of ¹⁸⁰Lu. The final experimental results are derived from formula [6] and the data in Table II for the 332 and 443 keV lines in each of three runs. While the values from Run III (which used spectrographic-grade Hf) are not included in the statistically weighted average, they are nevertheless consistent with the null result obtained from Runs I and II.

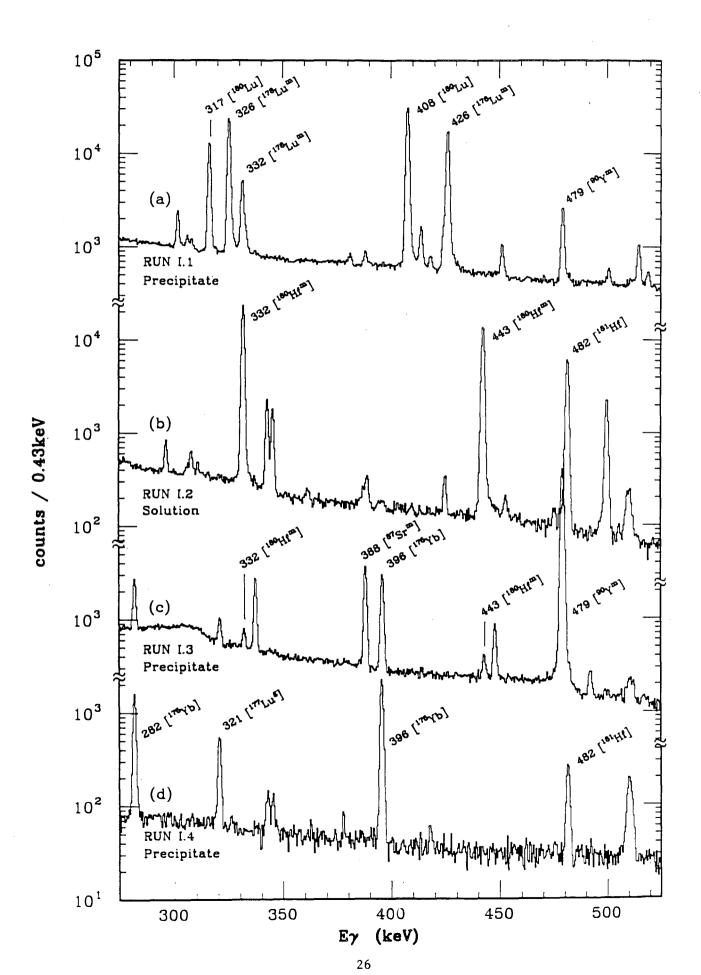
| RUN | f _m (332) | f _m (443) | |
|------------------------|-------------------------------|------------------------|--|
| ı. | +0.019±0.040% | +0.014±0.042% | |
| II. | -0.025±0.027% | +0.042±0.038% | |
| III. | +0.20 ±0.14 % | -0.06 ±0.15 % | |
| Statistically weighted | f _m = 0.005±0.018% | | |

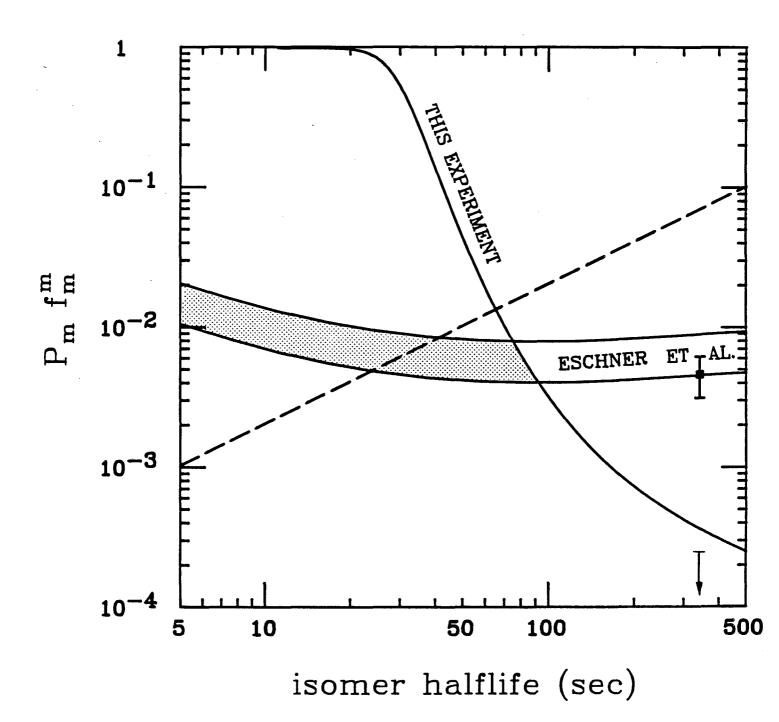
FIGURE 1. A partial energy level scheme for ¹⁸⁰Hf (Ref. 3) and the neutron-capture production paths to ¹⁸⁰Ta^M (Ref. 1). In the Beer and Ward model, ¹⁸⁰Ta^M can be produced in both the s-process and r-process by the population of ¹⁸⁰Hf^M followed by a weak β -decay branch, f_{β} . This experiment establishes a limit on f_{m} , the r-process fractional production of ¹⁸⁰Hf^M following the β -decay of ¹⁸⁰Lu (dashed beta-gamma path). Shown are gamma transitions in the decay of ¹⁸⁰Lu and ¹⁸⁰Hf^M with intensities greater than 2%. The three gamma-rays specifically used in the determination of f_{m} in this experiment are labeled. The levels in ¹⁸⁰Lu are discussed in the text. All energies are in keV and the italicized numbers are log-ft values.

PIGURE 2. Ge(Li) gamma-ray spectra following the radiochemical separation of lutetium (in precipitate form) from hafnium (in solution). The ¹⁸⁰Lu, ¹⁸⁰Hf^m and ¹⁸¹Hf gamma-rays used in this experiment are labeled as well as strong lines from other Lu, Yb, Y and Sr activities. The chronologically ordered spectra (a) through (d) emphasize activities with vastly different halflives. Irradiation and run times are listed in Table 2. The counting geometries were identical for the solution (b) and the precipitate (a), (c) and (d).

PIGURE 3. Interpretation of f_m in terms of a proposed high-spin isomer of ¹⁸⁰Lu. The ordinate is the product of the ¹⁸⁰Lu isomer's production factor, P_m , and its β -decay fraction that populates ¹⁸⁰Hf^m, f_m^m . The result of this experiment (shown as the lower data point at 5.7 minutes) also establishes an upper limit of about 100 seconds on the halflife of such an isomer. We have solved the equations of (Ref. 15) including a source term for a ¹⁸⁰Lu isomer to generate curves for the upper and lower limits implied by the positive value reported by Eschner et al. (upper data point). The shaded region is in agreement with both experimental measurements. The dashed line is described in the text.







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