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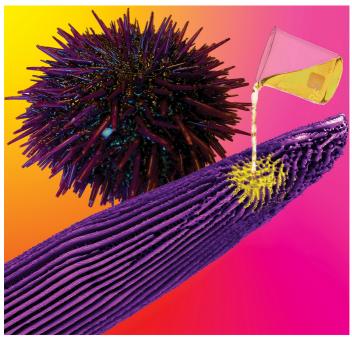
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INTEGRATING THE FIELDS OF CRYSTAL ENGINEERING AND CRYSTAL GROWTH FOR THE SYNTHESIS AND APPLICATIONS OF NEW MATERIALS







RESEARCH | IN OTHER JOURNALS



BIOMINERALOGY Precursor double down

ominerals that make up shells, coral skeletons, and other structures often grow differently than other types of crystals. Instead of atom-by-atom growth, solid amorphous precursor particles attach and crystallize. However, Stifler et al. found evidence of liquid precursor particles from observations of sea urchin spines regenerating from cuts. Liquid-like pools were evident in these regenerating spines, along with other observations consistent with a liquid morphology. The authors tentatively conclude that these liquid droplets are precursors to the well-known solid amorphous calcium carbonate precursor particles that help to form the spines. -BG Cryst. Growth Des. 10.1021/acs.cgd.1c00865 (2021).

The solid amorphous calcium carbonate precursor particles from which the spines of the purple sea urchin (pictured) grow are themselves derived from liquid precursor particles.

Ir-Pd-Pt-Rh-Ru. The discovered optimal binary alloys have been successfully validated experimentally. The proposed methodology substantially reduces the total number of experiments required for optimizing the composition space, making catalyst design for HEAs practically feasible, -YS

Angew, Chem, Int. Ed. 10.1002/ anie.202108116 (2021).

CELL BIOLOGY

Teaming up to finish a kill Cytotoxic T lymphocytes (CTLs) are thought to kill target cells in a rapid, all-or-nothing process. This may be the case when target cell immunogenicity is high; however, in solid tumors,

single CTL encounters may fail to kill target cells. Weigelin et al. found that migrating CTLs move between cancer cells and initiate apoptosis by additive sublethal interactions or "additive cytotoxicity." Sublethal damage occurs by perforin-dependent membrane pore formation, nuclear lamina rupture, and DNA disruption. In mouse melanoma tumors in vivo, additive cytotoxicity in interstitial tissue results in preferential elimination of invading tumor cells. –SMH

Nat. Commun. 12, 5217 (2021).

SKIN INFLAMMATION

A new culprit in psoriasis

Chronic inflammatory skin diseases such as psoriasis and atopic dermatitis are caused by overactive immune responses. Nakamizo et al. discovered a previously unrecognized type of immune cell in the skin of individuals with psoriasis. By performing single-cell flow cytometry and RNA-sequencing analysis on specific immune cells called macrophages and dendritic cells (DCs), the researchers created a molecular signature for skin inflammation and report that CD14+ type 3 DCs were more frequent in psoriasis lesions compared with healthy skin. This newly discovered DC subset expressed the cytokines IL23A and IL1B, which are known to drive inflammatory events in psoriasis. - PNK

J. Exp. Med. 218, e20202345 (2021).

ATOM INTERFEROMETRY Entanglement-enhanced atom interferometer

Atom interferometers work on the same principle as light interferometers but using the wave properties of atoms. Typically, a cloud of cold atoms is split, dropped into separate paths, and then brought back together to reveal interference fringes. Anders et al. demonstrate that a cloud of cold rubidium atoms cooled into a Bose-Einstein condensate, a quantum coherent state in which all of the atoms occupy the same quantum mechanical ground state, can enhance the sensitivity of the interferometer. Splitting the condensate into two momentum-entangled modes results in a sensitivity beyond the standard quantum limit. Using the setup in existing atom interferometers is expected to provide enhanced detection capability to test questions in fundamental physics. quantum gradiometers, and gravitational wave detectors. -ISO

Phys. Rev. Lett. 127, 140402 (2021).

DEVELOPMENTAL BIOLOGY Transitional metabolism

Various metabolites and metabolic pathways affect stem cell and embryo development. Zhao et. al. used a mass spectrometry-based method to measure metabolite abundance as mouse embryos transited between developmental stages. Combining metabolomics profiling with transcriptomics analysis revealed unique metabolic signatures as embryos developed from the totipotent to the pluripotent stage. Thus, as two-cell embryos become blastocysts, enzymes involved in the tricarboxylic acid cycle and oxidative phosphorylation increase activity and metabolite balance changes. For example, L-2-hydroxyglutarate in the twocell embryos declines, resulting in global histone methylation remodeling during preimplantation development. -BAP Nat. Metab. 10.1038/ s42255-021-00464-x (2021).

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576 29 OCTOBER 2021 • VOL 374 ISSUE 6567



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Evidence for a Liquid Precursor to Biomineral Formation

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ABSTRACT: The crystals in animal biominerals, such as sea urchin spines, mollusk shells, and coral skeletons, form by attachment of amorphous particles that subsequently crystallize. Do these solid amorphous precursor particles have liquid precursors? Polymer-induced liquid precursors (PILP) or prenucleation clusters coalescing into a liquid precursor to calcium carbonate crystallization have been observed extensively in synthetic systems. Molecular dynamics simulations also predict liquid–liquid phase separation. However, evidence for liquid precursors in natural biominerals remains elusive. Here, we present Scanning or PhotoEmission Electron Microscopy (SEM or PEEM) evidence consistent with a dense liquid-like precursor in regenerating sea urchin



spines. The observed precursor originates in tissue and ultimately transforms into a single crystal of calcite (CaCO₃) with complex stereom morphology.

A nimal biomineralization has played a central role during the evolutionary history of animals^{1,2} by providing a selective advantage over soft-bodied species. Biominerals have advantageous material properties,³ different from geologic minerals.⁴ The study of how animals construct mineralized structures is a compelling chemistry, materials science, structural, cell, and evolutionary biology problem. Many previous studies of biomineralization in animals found that amorphous precursors^{5,6} and crystal formation by attachment of amorphous precursor particles⁷ are vital components in forming calcium carbonate biominerals.

Liquid precursors, instead, have never been observed in natural biominerals. They have been assumed to exist, because they remarkably replicate in vitro a variety of naturally occurring structures, but direct observations are lacking. Liquid precursors were found in a variety of synthetic systems,⁸⁻¹³ but not naturally occurring biominerals despite extensive searches.¹⁴ Liquid precursors remain an unobserved assumption in 2021.¹² Here, we searched for evidence of a liquid precursor phase in regenerating sea urchin spines, using Scanning and PhotoEmission Electron Microscopies (SEM and PEEM). Sea urchin spines have a stereom structure, that is, an intricate arrangement of intersecting $10-\mu$ m-wide cylindrical calcite (CaCO₃) struts and 10- μ m-diameter holes common to echinoderm skeletal parts: the spines and tests of sea urchins, the armor plates and vertebrae of seastars and brittlestars, and the tests of sand dollars.¹⁵⁻¹⁷ The stereom diffracts like a single crystal of calcite,18 but unlike abiotic calcite crystals, it does not have a thermodynamically favorable euhedral morphology, with flat faces and sharp edges. Instead, the stereom has smooth, curved surfaces. Topologically, the stereom is a gyroid, that is, an infinitely connected triply periodic structure that minimizes surface area and does not contain straight lines.¹⁹ Here, we examined the nature of the

calcium carbonate precursors and searched for liquid phases, if they exist, in regenerating sea urchin spines.

When calcitic sea urchin spines are broken in nature or cut in the laboratory, they regenerate.²⁰ The regenerating part is distinct, as it starts growing from the center of the cylindrical spine and thus has a smaller diameter than the original spine. The color is also distinct: purple in the original spine and pale brown in the regenerating spine. Figure 1a shows spines cut using scissors and photographed approximately five days later in a living California purple sea urchin, Strongylocentrotus purpuratus. Seven to fourteen days after cutting them, the spines regrow to the original size near the cutting position and taper in diameter toward the regenerating tip (Figure 1b). The cut spine is an ideal system to observe calcite formation mechanisms. Adult sea urchin spines, along with sea urchin larval spicules and mollusk larval shells, were among the first biominerals in which amorphous calcium carbonate (ACC) precursors were observed.^{5,6}

A simple sample preparation and SEM imaging revealed a liquid-like precursor, as shown in Figure 1c-f. The preparation is entirely done on ice and includes rinsing in Tris base at pH 9, bleaching, rinsing in Tris base, and dehydrating in ethanol. The crucial difference between this and previous protocols is that the spines were kept vertical, with the regenerating tip pointing up during all preparation steps. Bleach oxidizes and removes the tissue around the spine stereom. With the spines upright and their regenerating tips up, precursor phases falling

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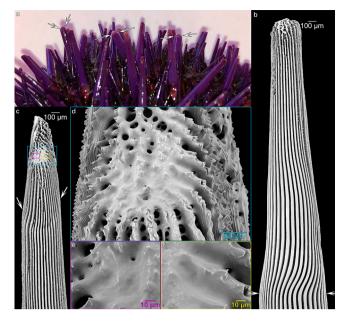


Figure 1. Regenerating sea urchin spines show a liquid-like amorphous precursor. a. Photograph of a living *Strongylocentrotus purpuratus* sea urchin, taken ~5 days after the spines were cut between arrowheads, obliquely between the arrowhead pair for the left spine and perpendicular to the spine axis between arrowheads pairs for the center and left spines. b,c. SEM images showing the original (below the arrows) and regenerating spine (above arrows) after ~10 days of regrowth. The spines were cut horizontally in b, and obliquely in c. In c, the original and regenerating spines are separated by a curved line between arrowheads, distinguishable because the spine diameter tapers down in the regenerating spine. Notice in c the liquid-like precursor filling stereom holes. d, e, f. Magnified images of the liquid-like precursor showing that it is smooth at the micron and submicron scale. Boxes in c indicate the regions magnified in d, e, f. All SEM images were acquired in backscattered electron (BSE) mode.

and accumulating on top of the stereom are less likely to wash away during rinsing and washing. The unusually dense mineral observed in Figure 1c-f covering the stereom, therefore, was not washed away but remained on the stereom and filled the stereom holes, behaving like a dense liquid. At the time of observation with SEM in vacuum, the material was likely no longer liquid. Still, it retained the morphology of liquid materials, filling all holes in a region that should otherwise show plentiful stereom holes. A comparison of regenerating spines in Figure 1c and b shows a regenerating stereom with and without liquid precursor on top. Figure S1 shows fractured spines, which, at their centers, only have stereom.²⁰ Thus, the location of the liquid-like material near the center of a spine's stereom provides an ideal substrate to distinguish solid from liquid materials. The SEM images in Figure 1 were acquired in backscattered electron (BSE) mode; thus, their brightness is proportional to the atomic number of the elements observed. The liquid-like material is as bright as the calcite stereom; therefore, it must have similar composition and density. Energy-dispersive X-ray (EDX) fluorescence emission data in Figure S2 show that the two spectra are similar but not identical, indicating that the liquid-like pool has the same relative amounts of Ca and O as calcite in the mature spine, but substantially more C. Greater C concentration in liquidlike pools than in mature stereom is consistent with organic molecules stabilizing the liquid, as they do with synthetic PILP.8

In Figure 2, PEEM component maps show a regenerating spine with many pixels in the stereom still amorphous, as well as in-tissue particles. These particles have average size ~ 200 nm and are located in the tissue, $\sim 5-10 \ \mu m$ away from the stereom. This distance is not surprising, as the skeletonforming cells are known to transport calcium to the forming biomineral through filopodia that are 5–10 μ m long.²² Three areas on the same spine were analyzed in total, containing 13 in-tissue particles. To quantify how disordered the in-tissue phases were, we used the empirical method recently used by Kahil et al.²³ for a similar biomineral-forming system: intracellular Ca-rich particles in the cells that deposit calcite spicules in sea urchin embryos, where the ratio of $L_{2'}/L_2$ peak intensities in Ca L-edge spectra provides an empirical estimate of order at the atomic scale in liquid and solid, amorphous and crystalline CaCO₃ phases.^{23–25} We note that the primary mesenchyme cells that form calcite spicules in sea urchin embryos and the sclerocyte cells that form calcite spines in adult sea urchins are similar. Thus, the results by Kahil et al. and those observed here are directly comparable. In the regenerating spine analyzed here, using $L_{2'}/L_2$ peak intensity ratios, we found three phases, calcite, ACC, and ACC-H₂O (Figure S3). The three phases were localized in the in-tissue particles of Figure 2, which were located in the spine where indicated by Figure S4. In both the regenerating spine here and the embryo cell in Kahil et al.²³ particles were observed intissue. In-tissue particles are much more amorphous, thus, they contain more red and yellow pixels compared to the stereom, which contains mostly blue or cyan pixels (Figure 2). Peakratio analysis shows that in-tissue particles can be more disordered than ACC-H₂O, in fact, a few of the in-tissue particles have a peak-ratio lower than that of ACC-H₂O (Figure S3). This observation is consistent with such particles being more disordered than the most disordered solid phase, ACC-H₂O, and thus were not completely solid during PEEM analysis.

Previous work showed that the occlusion of an Asprich peptide in calcite disorders the Ca spectrum slightly,²⁶ but the effect on the peak ratio is so small (0.47-0.46 on the vertical axis of Figure S3) compared to the effect of phase changes that it cannot explain the observed decrease.

We hypothesize that liquid droplets formed within the spine sclerocyte cells, explaining why they can be more disordered than ACC-H₂O in the in-tissue particles at the time of PEEM analysis. When the tissue was bleached away for SEM analysis, the droplets from many cells coalesced and formed the larger pool observed in Figure 1c-f. Such intracellular Ca-rich, membrane-bound particles or droplets were directly observed by Kahil et al. in single cells.²³

We prepared a total of 28 regenerating spines from 9 different sea urchins, all *Strongylocentrotus purpuratus*. Pools of liquid-like material were observed consistently in 20 spines, with the one in Figure 1c-f being the largest pool, and many smaller pools observed to occlude single holes in the stereom. In addition, at the time of observation in the SEM, these were probably mostly solid, but they retained the morphology consistent with a liquid surface in a pseudomorphic transformation. Figure 3 shows two liquid-like pools, demonstrating that they are smooth at the nanoscale and that they are as bright as the surrounding calcite stereom in both secondary electrons (SE-) and BSE-SEM mode.

After vertical-spine-bleaching preparations, in addition to the liquid-like pools, we also observe calcite rhombohedra, which

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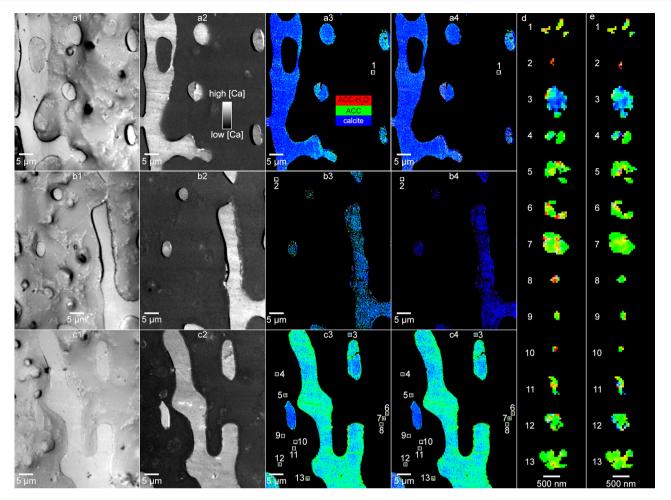


Figure 2. a1-c1, Average PEEM images; a2-c2, Ca distribution maps; a3-c3, PEEM-component maps; and a4-c4, repeat component maps of three different areas of a regenerating sea urchin spine. The numbered boxed regions in component maps a3-c3 and a4-c4 are magnified in d and e, respectively. In total, we found 13 different in-tissue particles in these 3 areas.

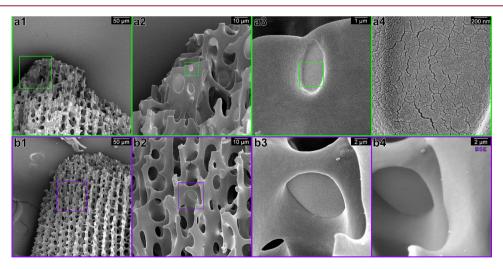


Figure 3. Liquid-like pools in regenerating spines. Increasing magnifications of the boxed regions are shown in subsequent panels and correspondingly colored. a3 and b3 show two liquid-like pools in two spines from different animals. a4 shows that at high magnification the liquid-like pool is smooth and featureless, except for cracking in the Pt coating. a1-a2, b1-b2 show many more pools, as does Figure S5. Notice that the pools in a3 and b3 are as bright as the rest of the stereom. b4 was acquired in backscattered electron (BSE) mode, all other images in secondary electron (SE) mode. b3 and b4, showing the same pool in SE and BSE, confirm that the liquid-like pool has similar composition to the stereom.

are never observed in sea urchin spines bleached horizontally. One possible interpretation is that these rhombohedra formed at the expense of either the liquid precursor or the ACC solid precursor particles known to exist in regenerating sea urchin spines.²⁰ The rhombohedra are presented in Figure S5 and Figure 3b3,b4. The rhombohedra are as bright as the stereom

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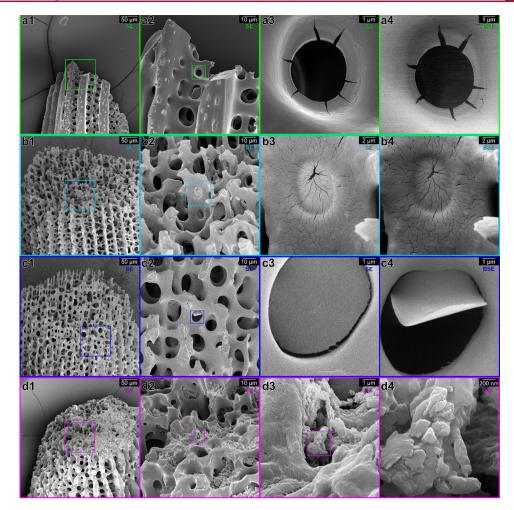


Figure 4. More clues about the liquid-like precursor from 4 spines at increasing magnifications, imaged in SE or BSE. a. This liquid-like pool had a circular opening when the precursor was liquid, then it cracked when the material either solidified, or crystallized, or got dehydrated in vacuum. b. This liquid-like pool and its surrounding stereom are both covered and cracked. One interpretation is that the liquid precursor coated both, then it shrank in volume when it solidified, or crystallized, or was dehydrated. c. This liquid-like pool shows damage with exposure to the e-beam and was seen moving during the experiment. c3 was acquired in SE before electron-induced motion; c4 was acquired in BSE after the motion. Notice how thin the liquid-like pool is. d. This regenerating spine's stereom holes are not filled with liquid but with solid particles. These particles, therefore, must have solidified before falling into the stereom holes. They were, presumably, within the tissue when it was beached.

in BSE-SEM (Figure 3b4, Figure S6a4), whereas cubic crystals of salt are bright in SE and dark in BSE (Figures S6d3,d4). We thus conclude that the bright euhedral crystals in Figure S5 are calcite rhombohedra. The possibility that the rhombohedra formed from the liquid precursor is supported by the morphology of the liquid-like material observed immediately adjacent to some rhombohedra, e.g., those in Figure S5b4,c4 or Figure S5b1,c1. These patches with dense-liquid morphology are unusual on the surface of the stereom, were never observed before, and only appear when using the present vertical-spinebleaching preparation. Thus, the unusual rhombohedra are more likely to have formed from the unusual dense liquid-like material observed here than from dissolution and reprecipitation of the solid particles previously reported.²⁰ Similar thin layers of synthetic PILP and nonpseudomorphic transformations were observed by Gower et al. on glass slides or microporous molds shaped after a natural sea urchin spine.^{27,28}

Other clues to the existence of a liquid precursor are presented in Figure 4. These include the observation that the thin liquid-like pools are \sim 100 nm thick (Figure 4c4), at least at the time of observation in SEM, consistent with a liquid with low surface tension forming a layer in a hole, like soapy water

in a soap bubble loop. Another observation, unique in this set of experiments, was a circular hole in a liquid-like pool, reminiscent of an inverted bubble, suggesting liquid-like behavior and cracking that is solid-like behavior (Figure 4a3,a4). A possible interpretation is that the material was liquid first, then solidified and cracked during water removal and volume change. A gel or colloidal suspension of particles could also explain this solid-like behavior, but not thin soap bubblelike layers.

We did not observe the liquid-like material in all spines. In 8 out of 28 spines, only the usual, smooth, solid stereom was found. In one spine only, we observed solid particles filling the stereom holes (Figure 4d2). A possible interpretation is that the liquid precursor solidified before reaching the stereom when the tissue was bleached, and the solid particles fell into the stereom holes.

We tentatively conclude that liquid droplets are precursors to the ACC precursor solid particles previously observed in sea urchin spine stereoms.²⁰ The whole set of phases and phase transitions involved in sea urchin spine biomineralization is presented in Table 1. Table 1. Phases and Phase Transitions in Sea Urchin Biominerals a

Phase	Chemical Formula	
Aqueous solution	H ₂ O, Ca ²⁺ , CO ₃ ²⁻ , HCO ₃ ,	
	Sr ²⁺ , Mg ²⁺ , …	
1. Liquid-liquid phase separation		
Liquid precursor denser than water	CaCO₃·(H₂O)n>1	
2. Precipitation of ACC-H ₂ O solid particles		
Hydrated amorphous calcium	CaCO₃-(H₂O)n≈1	Transitions
carbonate (ACC-H2O) solid particles		
3. Dehydration		
Anhydrous amorphous calcium	CaCO ₃	
carbonate (ACC)		
4. Crystallization		
Calcite, aragonite, or vaterite	CaCO ₃	

"These include 5 phases (pink) and 4 phase transitions (yellow) leading to the formation of crystalline calcite. Ions can attach to any of the phases and during all phase transitions.

The observation of liquid-like morphologies does not contradict previously observed ACC phases^{5,6,29} or biomineral formation^{30–32} by particle attachment.^{7,33,34} Several previous studies showed calcein-labeled intracellular vesicles in embryonic cells depositing sea urchin spicules.^{22,35–37} Kahil et al. recently showed hundreds of intracellular vacuoles and vesicles that initially contain seawater, form ACC-H₂O, ACC, and calcite,²³ the same phases previously observed with PEEM in sea urchin spine stereom²⁰ and in tissue here. We observe fewer in-tissue particles here compared to Kahil et al.²³ because the probing depth of PEEM is only 3 nm at the Ca L-edge;³⁸ thus, the probability of finding Ca-rich particles in a 3-nm-thick cell slice is much smaller than 6% (Figure S7).

In all in-tissue particles, we found more amorphous phases than in the forming stereom, suggesting that the precursor phases are formed in the tissue and are then deposited on the growing stereom. In a few cases, the spectra showed even greater disorder than the most disordered solid ACC-H₂O, which could be interpreted as not completely solidified (Figure S3). Such a disordered phase is consistent with a dense liquid phase, precursor to the solid ACC-H₂O and ACC phases, as listed in Table 1.

The liquid-like pools observed in SEM images cannot be solid particles originally suspended in a liquid, because solid particles in sea urchin spines were observed to be 100-400 nm in size. Particles with such sizes would appear nanoparticulate at 50,000×, precisely the magnification at which solid particles are observed here in Figure 4d4. Such particles were shown in sea urchin spines and many other biominerals by Gilbert et al.³³ At this scale, all liquid-like pools observed here are smooth, not nanoparticulate.

The liquid precursor was denser than water. Otherwise, it would not have fallen onto the underlying stereom in the bleach–water solution. Even if the liquid droplets were formed in membrane-bound intracellular vesicles in the tissues and were not in contact with water, when the membranes dissolved in bleach, the liquid droplets did not mix with and disperse in water. Thus, a liquid–liquid phase separation (LLPS) must have occurred previously in intracellular vesicles. Alternatively, LLPS may occur in membrane-less intracellular assemblies, as described for proteins³⁹ and synthetic systems that replicate membranes.¹¹ Since Kahil et al. observed membrane-bound

vesicles containing Ca-rich particles in sea urchin embryonic cells,²³ these are most likely membrane-bound in sea urchin spine cells as well.

LLPS leading to calcite formation was first observed in synthetic $CaCO_3$ systems by the Gower group,^{8,11,40} and then extensively reproduced by other groups in a variety of synthetic systems.^{9,10,41-44} LLPS was also predicted in molecular dynamics simulations.⁴⁵ The phase diagram of calcium carbonate solutions shows that at 15 °C, the temperature at which *S. purpuratus* sea urchins live, the spinodal line is at 2 mM; thus at this or greater concentrations, liquid–liquid separation can occur.⁴⁶ Even at concentrations lower than the spinodal line, dense liquid precursor droplets could nucleate within the parent liquid by overcoming a free energy barrier. Thus, dense liquid droplets could form whenever the concentration exceeds the solubility limit of the dense liquid phase, be it above or below the spinodal line. Seawater [Ca] is 10–11 mM, and in sea urchin intracellular vesicles [Ca] is much greater, 1–15 M;²³ thus, LLPS can occur.

The liquid precursor is likely common to other CaCO₃ biomineralizers. The same ACC-H2O and ACC solid precursors were observed in many fresh, forming biominerals from diverse animals. These biominerals include sea urchin spicules,²⁹ teeth,⁴⁷ and spines,²⁰ in aragonite (CaCO₃) mollusk shell nacre⁴⁸ and coral skeletons.^{34,49,50} Thus, we expect other CaCO₃ biomineralizers to share the dense liquid precursor observed in sea urchin spines and the sequence of phases and phase transitions in Table 1. Further experiments on diverse biominerals will demonstrate or refute the robustness of these preliminary results and their generality to other systems. If confirmed, a liquid precursor would solve long-standing mysteries in nacre formation.^{51,52} Twenty years after this idea was first proposed,⁸ and extensive searches,¹² we present the best evidence to date for a liquid precursor to biomineral formation. This evidence, however, is still indirect, because at the time of analysis by SEM and PEEM, the liquid-like precursor had solidified, via a pseudomorphic transformation, that is, retaining the liquid-like morphology. Isolating the liquid precursor and measuring its viscosity and composition remains a challenge.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.cgd.1c00865.

C1_Cni5_ACCH2O.txt (TXT) C2_Cni5_ACC.txt (TXT) C3_Cni5_Calcite.txt (TXT) Materials and Methods, imaging and micrographs, EDX

spectra, short-range order of Ca-rich particles, spherical model of a calcifying cell (PDF)

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Author Contributions

PG conceived the initial experiments. CEK and CAS prepared all the samples, PG collected all the SEM data, CAS and PG collected PEEM data. PG organized all the data into figures and wrote the manuscript first draft, all coauthors edited the manuscript.

Author Contributions

[#]Previously publishing as Gelsomina De Stasio.

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Notes

The authors declare no competing financial interest.

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Supplementary Information for

Evidence for a liquid precursor to biomineral formation

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This PDF file includes:

Figures S1 to S7, Materials and Methods

Other supplementary materials for this manuscript include the following:

C1_Cni5_ACCH2O.txt C2_Cni5_ACC.txt C3_Cni5_Calcite.txt

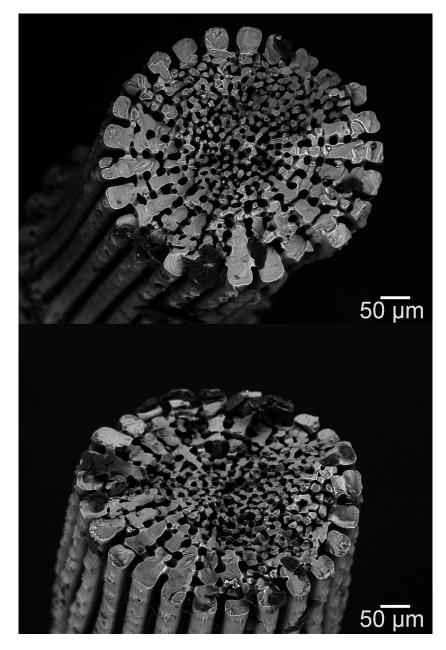


Fig. S1. Bleached and fractured sea urchin spines. The two spines from *Strongylocentrotus purpuratus* show that only holey stereom is present at the center. The radial structures visible at the outer diameters are termed septa and confer the striations visible on the lateral surface of the spine cylindrical structure.

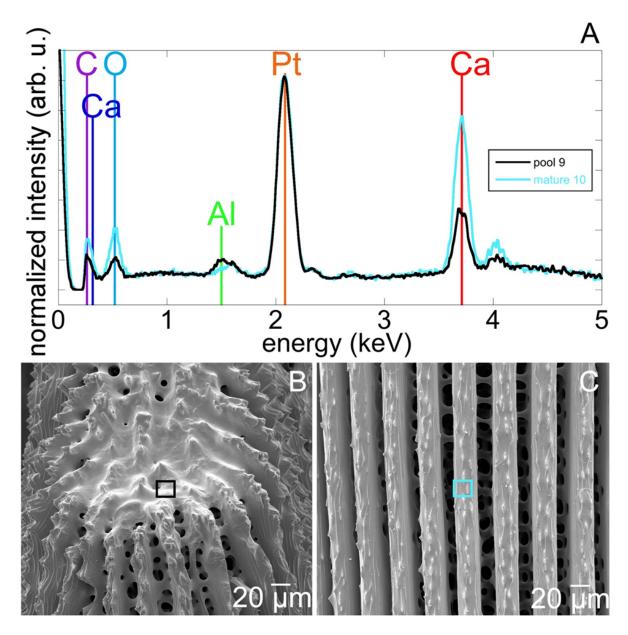


Fig. S2. Energy-dispersive x-ray (EDX) fluorescence emission data of the liquid-like pool and the mature spine. A. The two EDX spectra are similar but not identical. The Ca, O, C peak heights and therefore concentrations are greater in the mature-spine calcite (cyan) than in the liquid-like pool (black). Their relative decreases, however, differ. The approximate ratios after background subtraction, are as follows: [Ca pool]: [Ca mature]=0.4, [O pool]: [O mature]=0.4, and [C pool]: [C mature]=0.7. The Pt coating is identical in the two regions, despite topographic differences, because the 20 nanometers of Pt were deposited while tilting and spinning the sample. The ratio confirms that the Pt concentrations are identical, after background subtraction: [Pt pool]: [Pt mature]=1. B, C. SEM image acquired at 500x magnification, indicating the pool or mature regions from which the EDX spectra in A were acquired (black box in B, cyan box in C). Both spectra were acquired in the boxed regions at 10,000x magnification.

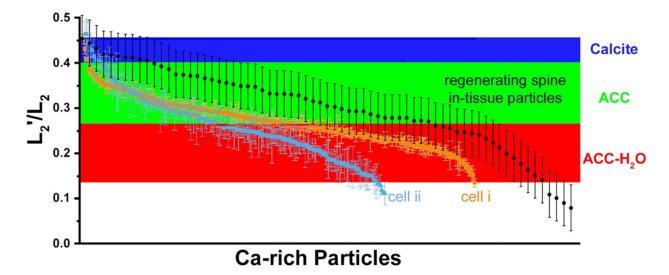


Fig. S3. Short-range order of Ca-rich particles in sea urchin embryo single-cells from Kahil et. al.¹ (cells i and ii) and in-tissue particles from regenerating adult sea urchin spine (this work). The shortrange order is defined as the first or second nearest neighboring atoms to the Ca atoms in the observed phase, measured by the peak intensity ratio L₂'/L₂ as described by Kahil et. al.¹. In their work or ours, pixel sizes were 13 nm or 55 nm, Ca L-edge spectra were acquired with 18 or 121 energy data points. We both minimized radiation damage by limiting the exposure time but made slightly different choices: Kahil et. al.¹ reduced the number of energy steps, we limited the spatial resolution and therefore the exposure time. The two approaches are completely equivalent because particle sizes are well above pixel sizes, thus it does not matter if they were observed in transmission or photoemission (their work and ours, respectively). The blue, green, and red bands in the background represent the peak ratio ranges for the three phases: calcite, ACC, and ACC-H₂O, respectively. The upper or lower bounds of each band are the peak ratio in their data or our data, respectively. The differences in peak ratios are due to differences in beamline resolution, energy sampling, and over-absorption in transmission. Peak ratios from the Ca L-edge spectra collected by PEEM in each pixel of in-tissue particles in a regenerating sea urchin spine in this work are represented by black data points. Our in-tissue particle data-points are superimposed on their single-cell data-points, demonstrating that the in-tissue particles in regenerating spines had a wider range of order compared to the ones observed in spicules. Only a subset of our data is shown, selected among the lowest noise spectra, which yield the most reliable peak ratios. In almost all intissue particles the L_2'/L_2 ratios were within the calcite, ACC, or ACC-H₂O bands, indicating that at the time of PEEM analysis these particles were solid. A few in-tissue particle pixels, however, were below the ACC-H₂O band, suggesting greater disorder.

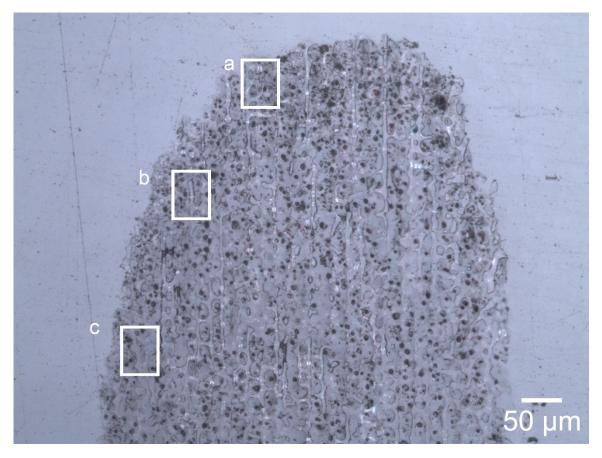


Fig. S4. Visible light micrograph of fixed, regenerating spine. The three boxed regions a, b, and c correspond to the identically labeled areas in Figure 2. The dark circles are air bubbles in the embedding epoxy.

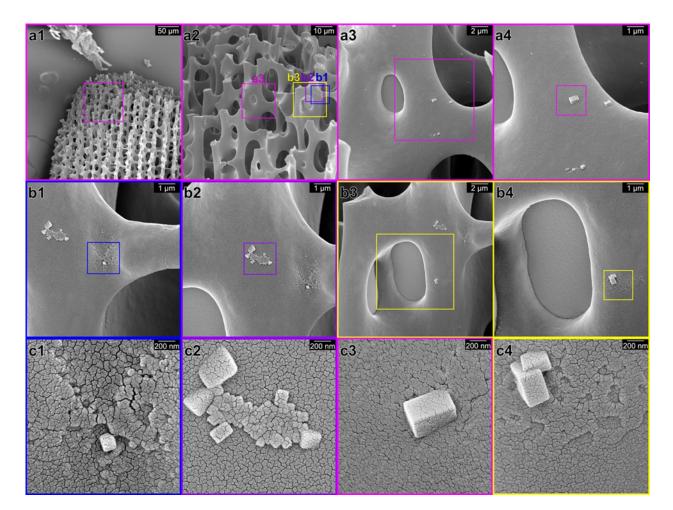


Fig. S5. Rhombohedra on a sea urchin spine stereom. All images were taken in SEM-SE mode. The rhombohedra show that the liquid-like precursor may crystallize as calcite. In Fig. S6 rhombohedra in BSE are as bright as or brighter than the stereom, whereas salts are darker. The rhombohedra, therefore, are likely calcite.

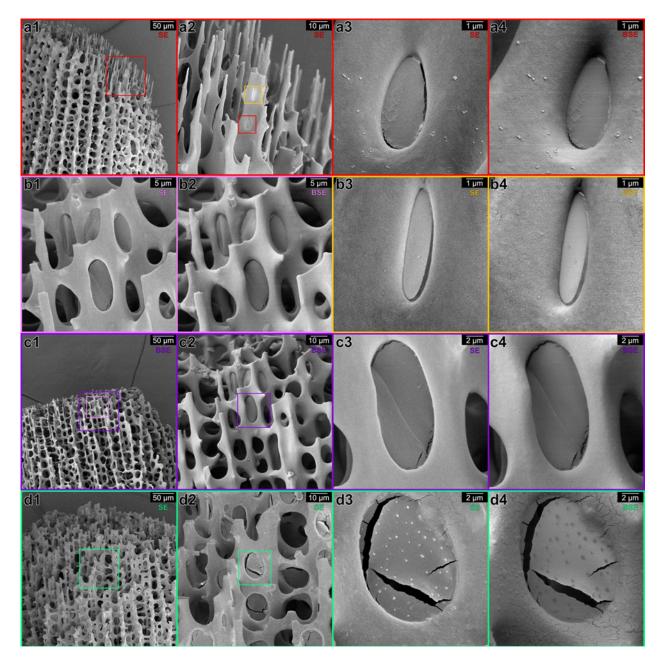


Fig. S6. Liquid-like pools in three regenerating spines from different animals, imaged in secondary electron (SE) or back-scattered electron (BSE) modes of SEM. SE and BSE labels appear in each panel. Boxes in a1,a2 or c1,c2 indicate regions magnified in a2,a3,a4,b3,b4, or in c2,c3,b1,b2 and correspondingly colored. b1,b2 show three liquid-like pools in the same frame. c1 shows more liquid-like pools at lower magnification. Notice that the pools in a4,b4,c4 are as bright as the rest of the stereom in BSE, indicating that the Ca density is similar. The pool in a3,a4 is surrounded by submicron rhombohedra bright in both SE and BSE. The pool in d3,d4 is covered and surrounded by salt crystals, which are bright in SE but dark in BSE. The rhombohedra in a3,a4, therefore, must be calcite. This indicates that the liquid-like material crystallizes as calcite.

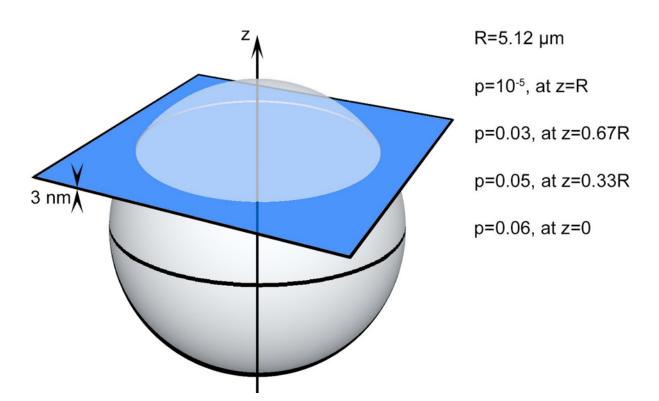


Fig. S7. A spherical model of a calcifying cell showing, on the right, the probability p that a 3nmthick slice (the probing depth of PEEM at the Ca L-edge 31) of the cell at different heights on the sphere's vertical axis contains at least one Ca-rich vesicle. We assumed that there are 140 particles, each with 250nm-diameter, within each cell, as observed by Kahil et al.¹. The probability of finding a Ca-rich particle in the 3nm-thick slice analyzed in PEEM is even lower, because the 55µm-field of view of PEEM images may or may not contain a 10.24µm-diameter stereom-forming cell, or a portion of a cell.

Materials and Methods

Sea urchins, spine regeneration

Twenty-five *Strongylocentrotus purpuratus* sea urchins were collected near Bodega Bay, CA, USA. They were then kept in outdoors tanks with seawater circulating, coming directly from the Pacific Ocean at the University of California - Bodega Marine Laboratory, Bodega, CA, USA. They were then transferred to UC-Berkeley, indoors, kept at 15 °C constant temperature in seawater, and fed kelp every day for ~ 1 week before the experiments to ensure that the urchins were healthy: they moved around and had their tube feet out. Several spines in each sea urchin were cut off at approximately ½ of their length using scissors. The spines were allowed to regenerate for 7-14 days and harvested. Individual spines were selected if they appeared to be regenerating. These regenerating spines were harvested by cutting them near the base.

A total of 28 spines were prepared vertically and 2 horizontally for SEM experiments and 1 vertically for PEEM experiments. Horizontally prepared spines showed liquid-like pools in 0 out of 2 spines; thus, the horizontal-spine preparation was abandoned. Vertically prepared spines showed liquid-like pools in 20 out of 28 spines. Therefore, only this procedure was used subsequently.

Sample preparations for SEM experiments

Each spine was placed vertically into a 1.5 ml Eppendorf tube, with the regenerating tip up.

The spines were treated, on ice, as follows:

- Two rinses with 50 mM Tris base at pH 9.
- Two incubations for 10 min each in 6% NaClO.
- Two rinses with 50 mM Tris base at pH 9.
- Two rinses with 100% ethanol.
- Air dried at room temperature.

For SEM experiments the spines were mounted on carbon tape (Ted Pella Inc., Redding, CA), and coated with 20-nm Pt, while rotating and tilting the samples, using a high-resolution Cressington 208hr sputter coater with rotary, planetary, and tilting movements (Cressington, UK), which is optimized for coating distribution and coverage of highly topographic samples such as sea urchin spines.

Sample preparations for PEEM experiments

One of the regenerating spines, harvested 10 days after cutting, was placed vertically in a tube of seawater with 5% MgCl₂ and subsequently fixed with formaldehyde in 0.05 M sodium cacodylate buffer and gradually dehydrated in five steps with ethanol concentrations increasing from 50% to 90% with each 10-minute step. Finally, the spine was dehydrated in 100% ethanol twice, for 10 minutes each time. The spine was then embedded in Solarez resin and cured according to the procedure in Sun et. al. 2020^2 . Once cured, the sample was polished and coated with 1 nm Pt in the area of interest and 40 nm elsewhere while spinning and tilting.

SEM experiments

Most of the SEM experiments were all done at the UC-Berkeley Electron Microscopy Laboratory, using either a Hitachi TM1000 or a Hitachi S5000 microscope.

The TM1000 has a fixed accelerating voltage of 15 kV, fixed back-scattered electron (BSE) mode, and 10,000x maximum magnification. This was used for Fig. 1b-f, Fig. 2b4, and Fig. S1. The S5000 was used with 10 kV accelerating voltage, in both secondary electrons (SE) and BSE modes, and has a 1,000,000x maximum magnification. For the present experiments the maximum magnification used was 50,000x. Each region of interest was first images at 50,000x, then at 10,000x, 5,000x, 1,000x, and 250x. The S5000 was use for Figs. 2, 4, S5, S6.

In all figures, the images were presented in reverse order, with low magnification first. All SEM images were adjusted for brightness and contrast in Adobe Photoshop[®] and assembled with colored grids and boxes also in Photoshop[®].

Energy-Dispersive X-ray (EDX) fluorescence emission analysis was done at Lawrence Berkeley National Laboratory, Chemical Sciences Division, building 30, using an FEI Quanta FEG 250 SEM, equipped with a Bruker EDX analyzer, and used with a 5 kV e-beam and 10 mm working distance.

PEEM experiments

PEEM experiments were done at the Advanced Light Source on beamline 11.0.1.1 at the Ca Ledge for component mapping with a lateral resolution of 60 nm and the probing depth 3 nm. All data from 3 areas of the same regenerating spine were acquired 30, 30.5, and 31 hours after extraction, to ensure that the amorphous phases stayed amorphous, as much as possible, considering the unavoidably long sample preparation time (30 hours). The spine was not

bleached, but fixed as described in ² to retain spine tissues and in-tissue particles. All data were acquired on the regenerating part of the spine. We limited the magnification and resolution (55 nm instead of the possible 20 nm or even 10 nm), to shorten the acquisition time and thus reduce the radiation dose on the spine to minimize radiation damage and maintain the phases amorphous. For component maps, stacks of PEEM images were acquired while scanning the illuminating x-ray photon energy across the Ca L-edge, from 340 to 360 eV, with 0.1-eV energy steps between 345 and 355 eV and 0.5 eV from 340 to 345 eV and 355 to 360 eV (121 energy steps). All Ca data were taken with circular polarization to avoid any linear polarization effects from crystals. The 121 images were converted to 8-bit tif stacks using Igor Pro® version 6.37, and the GG Macros developed for component analysis by our group, and available to any interested users free of charge (https://home.physics.wisc.edu/gilbert/software/). The component spectra used were obtained by averaging single-pixel spectra containing at least 90% of the relevant phase, in sea urchin spine data from Alberic et al. 2019³, but improved the component spectra compared with that publication. The new component spectra, called "Cni5", improved the peakfitting of the spectra, with identical backgrounds for all 3 phases, thus background variations potentially leading to misinterpretation were eliminated. The Cni5 spectra are included as text files. The results were reproduced on 3 diverse areas (Fig. 3) of the same regenerating spine. In component maps, the skeleton and the calcium containing particles in the tissue were preserved, and the rest of the pixels were masked (black) with a series of overlapping masks. One mask removed all pixels with a bad fit to the 3 component spectra, defined as one with too high a chisquare value: c² > 0.01, or 0.015, or 0.025 for areas 1, 2, or 3, respectively (Igor and GG Macros). A second mask called "Therm mask" removed pixels where the phase went thermodynamically uphill in the repeat movie (e.g. a green pixel becoming red, which is a nonphysical result) with an allowed tolerance of 100 color levels out of 255 for areas 1 and 3 and allowed tolerance of 25 color levels out of 255 for the much noisier area 2 (MATLAB[®], MathWorks). A third mask called "magic eraser" removed particles smaller than 2 pixels by 2 pixels, interpreted as noise that escaped the first 2 masks. The magic eraser places each pixel in the center of a 3x3 grid comprised of 8 surrounding pixels and masks it if the pixel does not border at least 3 other unmasked pixels (MATLAB®, MathWorks). All remaining pixels clustered into reliable particles in the tissue. A fourth mask was done by hand: single-pixel spectra from intissue particles were carefully examined, and any pixel whose spectrum was mostly noise was masked off by hand.

The peak intensity ratios in Fig. S3 were obtained by first subtracting a linear fit to the pre-edge of each spectrum, and then measuring the intensity of peaks L₂' and L₂, previously termed peaks 2 and 1, respectively ³⁻⁶. The background subtraction and peak intensity measurements were done in Igor Pro 6.37, using the GG Macros. The ratios were calculated in Microsoft Excel[®] for Mac, plotted in Kaleidagraph[®] 4.5.2, and superimposed on the Kahil et al. figure using Adobe Photoshop[®] version 22.2.0 for Mac. The upper bounds for peak ratios of calcite, ACC, and ACC-H₂O are the L₂'/L₂ ratio of known spectra from each phase as presented in Kahil et. al¹. and the lower bound for ACC-H₂O is the L₂'/L₂ ratios for the ACC-H₂O spectrum published in Albéric et. al.³ with the background subtracted as described above. The peak ratios for calcite and ACC from Albéric et. al.³ were slightly lower than the ACC and ACC-H₂O ratios from Kahil et. al¹, so to simplify Fig. S3, we chose to display the lower bound of calcite and ACC as the upper bound of ACC and ACC-H₂O. The peak ratios for each phase of Cni5 spectra used for component mapping fall in the range displayed in Fig. S3. These differences in peak ratios are due to differences in beamline resolution and energy sampling.

Probability calculations

The calcifying cell was modeled as a sphere (Fig. S7) with a radius of $5.12 \mu m$ based on the dimensions of the cell shown in Kahil et. al. ¹. The Ca-rich particles were modeled as smaller spheres with a diameter of 250 nm. To find the total number of Ca-rich vesicles that could fit inside the cell, the volume of the cell was divided by the volume of the particle and that ratio was multiplied by 0.63, the packing efficiency of randomly packed spheres⁷. Then, to find the probability that a single "spot" is filled with a vesicle, the average number of vesicles per cells from Kahil et. al. ¹, 140, was divided by the total number of spots, 43,293. The sphere was then

divided into 3nm-thick sections to represent the probing depth of PEEM⁸. The number of "spots" in a given section was calculated from the division of the volume of the spherical segment ⁹ by the volume of a 250nm vesicle. The probability that no particles are found at a given spot is then one minus the probability of finding a particle in that spot. From there, the probability that the spots in the 3nm slice contains no vesicles was calculated by the probability of a single spot having no particles raised to the number of vesicles in that slice. The probability that there is at least one particle in a given slice is one minus the probability of having no particles in that slice.

Cni5 component spectra (separate files). C1_Cni5_ACCH2O.txt, C2_Cni5_ACC.txt, and C3_Cni5_Calcite.txt

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