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Source Provenance of Obsidian Artifacts from the Spring Creek Site (CA-SHA-69), Shasta County, California

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**Author** Shackley, M. Steven

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GEOARCHAEOLOGICAL X-RAY FLUORESCENCE SPECTROMETRY LABORATORY 8100 Wyoming Blvd., Ste M4-158 Albuquerque, NM 87113 USA

# **SOURCE PROVENANCE OF OBSIDIAN ARTIFACTS FROM THE SPRING CREEK SITE (CA-SHA-69), SHASTA COUNTY, CALIFORNIA**



The ca. 950 year old Glass Mountain (East Glass Mountain chemical group here) obsidian flow on the east rim of the Medicine Lake caldera in northeastern California (entire ridge in center of image). The entire ridge in the background is composed of a glassy breccia with single blocks as much as 3 meters in height. Abundant core reduction and biface preform production is evident everywhere (Dillian 2002).

by

M. Steven Shackley, Ph.D., Director Geoarchaeological XRF Laboratory Albuquerque, New Mexico

and

Richard E. Hughes, Ph.D., Director Geochemical Research Laboratory Portola Valley, California

Report Prepared for

Heidi Shaw ENPLAN Redding, California

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## **INTRODUCTION**

The analysis here of 63 obsidian artifacts from the Spring Creek Site (CA-SHA-69) in Shasta County, California indicates a nominally diverse assemblage dominated by the Grasshopper Flat/Lost Iron Well (GF/LIW) chemical group in the Medicine Lake Highlands.

Given the expertise of Dr. Richard Hughes in this region both in archaeology and the understanding of the geochemical sources of archaeological obsidian, Hughes interpreted the data analyzed by Shackley in Albuquerque, and assigned the samples to source.

# **LABORATORY SAMPLING, ANALYSIS AND INSTRUMENTATION**

 All archaeological samples are analyzed whole. The results presented here are quantitative in that they are derived from "filtered" intensity values ratioed to the appropriate xray continuum regions through a least squares fitting formula rather than plotting the proportions of the net intensities in a ternary system (McCarthy and Schamber 1981; Schamber 1977). Or more essentially, these data through the analysis of international rock standards, allow for interinstrument comparison with a predictable degree of certainty (Hampel 1984; Shackley 2011).

 All analyses for this study were conducted on a ThermoScientific *Quant'X* EDXRF spectrometer, located at the University of California, Berkeley. It is equipped with a thermoelectrically Peltier cooled solid-state  $Si(Li)$  X-ray detector, with a 50 kV, 50 W, ultrahigh-flux end window bremsstrahlung, Rh target X-ray tube and a 76  $\mu$ m (3 mil) beryllium (Be) window (air cooled), that runs on a power supply operating 4-50 kV/0.02-1.0 mA at 0.02 increments. The spectrometer is equipped with a 200 l min<sup>-1</sup> Edwards vacuum pump, allowing for the analysis of lower-atomic-weight elements between sodium (Na) and titanium (Ti). Data acquisition is accomplished with a pulse processor and an analogue-to-digital converter. Elemental composition is identified with digital filter background removal, least squares empirical peak deconvolution, gross peak intensities and net peak intensities above background.

 For the analysis of mid Zb condition elements Ti-Nb, Pb, Th, the x-ray tube is operated at 30 kV, using a 0.05 mm (medium) Pd primary beam filter in an air path at 200 seconds livetime to generate x-ray intensity Ka-line data for elements titanium (Ti), manganese (Mn), iron (as  $Fe<sub>2</sub>O<sub>3</sub><sup>T</sup>$ ), cobalt (Co), nickel (Ni), copper, (Cu), zinc, (Zn), gallium (Ga), rubidium (Rb), strontium (Sr), yttrium (Y), zirconium (Zr), niobium (Nb), lead (Pb), and thorium (Th). Not all these elements are reported since their values in many volcanic rocks are very low. Trace element intensities were converted to concentration estimates by employing a quadratic calibration line ratioed to the Compton scatter established for each element from the analysis of international rock standards certified by the National Institute of Standards and Technology (NIST), the US. Geological Survey (USGS), Canadian Centre for Mineral and Energy Technology, and the Centre de Recherches Pétrographiques et Géochimiques in France (Govindaraju 1994). Line fitting is linear (XML) for all elements but Fe where a derivative fitting is used to improve the fit for iron and thus for all the other elements. When barium (Ba) is analyzed in the High Zb condition, the Rh tube is operated at 50 kV and up to 1.0 mA, ratioed to the bremsstrahlung region (see Davis 2011; Shackley 2011). Further details concerning the petrological choice of these elements in Southwest obsidians is available in Shackley (1988, 1995, 2005; also Mahood and Stimac 1991; and Hughes and Smith 1993). Nineteen specific pressed powder standards are used for the best fit regression calibration for elements Ti-Nb, Pb, Th, and Ba, include G-2 (basalt), AGV-2 (andesite), GSP-2 (granodiorite), SY-2 (syenite), BHVO-2 (hawaiite), STM-1 (syenite), QLO-1 (quartz latite), RGM-1 (obsidian), W-2 (diabase), BIR-1 (basalt), SDC-1 (mica schist), TLM-1 (tonalite), SCO-1 (shale), NOD-A-1 and NOD-P-1 (manganese) all US Geological Survey standards, NIST-278 (obsidian), U.S. National Institute of Standards and Technology, BE-N (basalt) from the Centre de Recherches Pétrographiques et Géochimiques in France, and JR-1 and JR-2 (obsidian) from the Geological Survey of Japan (Govindaraju 1994). RGM-1 a USGS obsidian standard from Glass Mountain, Medicine Lake Highlands is analyzed during each sample run of 20 to check stability of machine calibration (Table 1).

Source assignments were made by plotting the quantitative composition estimates in Table 1 against the Zr/Sr parameters of archaeologically significant obsidians identified in this part of northern California. The correspondences indicate that most of the artifacts were manufactured from obsidian of the Grasshopper Flat/Lost Iron Well chemical type (GF/LIW in Table 1) of the Medicine Lake Highland (Hughes 1986: Table 8) with a smaller number matching the Zr/Sr parameters of East Medicine Lake (EML in Table 1) volcanic glass from the Highland. Four artifacts correspond to the chemical profile of Buck Mountain obsidian (BM in Table 1) from the Warner Mountains (Hughes 1986: Table 7), and one other specimen (no. 43- 003-1) has a trace element composition unlike any of the geological reference standards in Hughes' current regional reference collection. We note that several of the artifact-to-chemical type attributions (*sensu* Hughes 1998) in Table 1 are accompanied by a ?; this uncertainty may be the result of Sr enrichment due to surface incrustation of a calcium carbonate-like substance on some artifacts which Shackley was unable to remove completely. While this prevented us from making more precise attributions for these artifacts, we are nonetheless confident that they derive from the Medicine Lake Highland Volcanic Field.

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<b>SAMPLE</b>	Mn	Fe	Zn	<b>Rb</b>	Sr	Υ	Zr	<b>Nb</b>	Ba	Pb	Th	<b>SOURC</b>
11-001	302	1076 3	44	140	76	28	184	11		28	20	Е GF/LIW
11-002	278	9418	51	131	77	29	179	11		25	16	<b>GF/LIW</b>
11-003	268	9606	33	109	62	26	185	13		21	11	GF/LIW
11-004-1	291	1151 $\overline{2}$	69	141	68	27	183	10		27	21	GF/LIW
11-004-2	332	1080 5	66	136	74	28	184	10		28	14	GF/LIW
11-005	330	1182 2	69	140	74	29	186	13		26	12	GF/LIW
13-000	361	1018 0	58	149	70	29	184	12		25	16	GF/LIW
14-000	308	1278 3	38	150	82	30	213	10		29	24	EML?
$15 - 001 - 1$	274	1024 3	39	135	71	26	201	12		25	13	EML?
15-001-2	317	9539	60	134	74	26	172	12		25	19	<b>GF/LIW</b>
15-002-1	305	1148 4	37	146	73	27	197	14		27	19	GF/LIW?
15-002-2	351	1332 7	49	157	77	34	207	12		31	20	<b>EML</b>
15-003-1	414	7623	77	116	71	16	94	13	834	23	10	BM
15-003-2	352	1451 1	69	151	84	28	191	10		28	18	GF/LIW
19-001	318	1150 0	48	157	76	27	188	12		28	24	GF/LIW
19-002-1	309	1115 5	40	140	77	29	187	14		26	15	GF/LIW
19-002-2	320	1304 2	59	128	71	30	181	14		26	20	GF/LIW
19-003	289	1099 9	55	143	74	30	185	12		28	18	GF/LIW
19-007	319	1145 3	58	150	73	31	195	13		31	14	GF/LIW?
19-013	316	1179 0	45	138	75	27	189	12		26	16	GF/LIW
19-014	310	1108 0	44	139	72	33	184	13		27	10	GF/LIW
22-002	333	1218 6	45	150	75	29	201	12		29	18	EML?
22-003	323	1123 3	80	131	67	28	184	10		27	17	GF/LIW
22-004-1	277	9359	40	125	67	27	176	8		24	8	GF/LIW
22-004-2	338	1175	50	141	75	30	188	14		26	15	GF/LIW
24-001	354	9 1230 9	55	146	90	30	198	15		26	15	EML?
24-002	333	1191	73	142	75	26	191	10		28		15 GF/LIW

Table 1. Elemental concentrations and source assignments for the archaeological specimens and USGS RGM-1 obsidian standard. All measurements in parts per million (ppm). Ba acquired for some samples to enhance discrimination.





Figure 1. Bivariate plot of Zr/Sr concentrations for archaeological specimens from CA-Sha-69. Source composition ranges of archaeologically significant obsidians in northern California from Hughes (2014).



Dashed lines represent the range of variation measured in geological obsidian source samples. Filled triangles plot the artifacts listed in Table 1. Source composition ranges from Hughes (2014).