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ABSTRACT

The spin of neodymium-141 was measured by the method of atomic beams and found to be 3/2. A lower limit placed on the hyperfine separation of the states with total angular momentum $\mathbf{F} = 11/2$ and $\mathbf{F} = 9/2$, Δv (11/2, 9/2) by second-order perturbation theory shows $\Delta v (11/2, 9/2) \ge 1630$ Mc/sec.

THE NUCLEAR SPIN OF NEODYMIUM-141*

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INTRODUCTION

Neodymium-141 is an even-odd nucleus with 60 protons and 81 neutrons. Since the neutron number is one less than the magic number 82, the properties of the nucleus should be well described by the single-particle shell model, which predicts that the state of the eighty-second neutron is $d_{3/2}$. Hence the spin of Nd¹⁴¹ should be 3/2 and the ground state should have even parity.

Evidence concerning the spin comes from the measurements by Polak et al., 1 who have investigated the positron decay of Nd¹⁴¹ and the associated gamma rays. The resulting spectrum is consistent with the assignment of 3/2 for the spin of Nd¹⁴¹.

A prerequisite for the determination of the nuclear spin from hyperfine-structure measurements is the knowledge of the electronic structure. The ground-state configuration of neodymium is known to be $(4f)^4$, from the optical spectroscopic investigations by Schuurmans.² The atomic beam work of Smith and Spalding³ has further established the ground state to be characterized by ${}^{5}I_{4}$ with $g_{J} = -0.6031$. This value is used throughout this work.

METHOD

The method used is the "flop-in" atomic-beam resonance method first proposed by Zacharias. $^{4, 5}$ The apparatus is of conventional design,

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utilizing an oven arrangement particularly convenient for handling materials with high radiation levels. Both the apparatus and oven have been discussed elsewhere. ^{6,7} Beam was collected on freshly flamed platinum for neodymium is at least 25%, and very probably 100%. This is comparable to what has been observed with other elements in the lanthanide and actinide series. After collection, the deposited neodymium is counted in methane counters.

For an odd-A isotope such as Nd^{141} in an electronic state with J = 4, at least two transitions are observable in a "flop-in" apparatus. These transitions will be between levels characterized by the quantum number

$$(\mathbf{F} = \mathbf{I} + 4, \mathbf{m}_{\mathbf{F}} = -\mathbf{I} + 1 \leftrightarrow \mathbf{F} = \mathbf{I} + 4, \mathbf{m}_{\mathbf{F}} = -\mathbf{I} - 1)$$

and

$$(\mathbf{F} = \mathbf{I} + 3, \mathbf{m}_{\mathbf{F}} = -\mathbf{I} + 2 \iff \mathbf{F} = \mathbf{I} + 3, \mathbf{m}_{\mathbf{F}} = -\mathbf{I}),$$

where F is the total angular momentum of the atom, m_F is the projection along the axis of quantization, and I is the nuclear spin. Resonances arising from transitions in the state F = I + 4 will be denoted as a resonances, and those in the state F = I + 3 will be denoted as β resonances. Both of the above transitions are of a double-quantum nature, i.e., $\Delta m_F = \pm 2$. This is a consequence of an integral J value and can best be seen from the schematic energy level diagram, Fig. 1. Such transitions were first observed in Pu^{239} , 6,7 and have been observed since in many isotopes.⁸

In the Zeeman region, i.e., in the weak-magnetic-field region, the frequency ν of an observable transition is given by

$$v = g_F \mu_0 H/h$$

where g_F is the splitting factor for the particular F level in which the

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transition is observed, μ_0 is the Bohr magneton, H is the magnetic field, and h is Planck's constant. The splitting factor g_F is related to the electronic splitting factor g_I by the relation

$$g_{F} \approx g_{J} \frac{F(F+1) + J(J+1) - I(I+1)}{2F(F+1)}$$
, (2)

where a term in the nuclear moment has been neglected. Using $g_J = -0.6031$ for the J = 4 state of Nd, one determines the transition frequencies for a and β type resonances to be

$$\nu \approx \frac{2.4124}{1+4} \frac{\mu_0 H}{h} \qquad (a type) \qquad (3)$$

and

$$\mu \approx \frac{0.30155 (3I + 16)}{(I + 4)(I + 3)} \frac{\mu_0 H}{h} \qquad (\beta \text{ type})$$

Equations (3) determine the nuclear spin from the resonant frequencies observed in weak magnetic fields.

The experimental procedure consisted of searching for a signal from the isotope of unknown spin at the discrete frequencies given by Eq. (3). This procedure was repeated at several magnetic fields.

The magnetic field was determined from the resonant frequency of K^{39} (F = 2, $m_F = -1 \leftrightarrow F = 2$, $m_F = -2$).

SAMPLE PREPARATION

The radioactive sample was produced in the Berkeley 60-inch cyclotron by the reaction $Pr^{141}(p,n)Nd^{141}$. The target material, praseodymium, was machined into discs 0.025 in. thick and bombarded with 12-Mev protons for from 5 to 8 hours. The total integrated flux was generally about 175 μ a-hr. At the completion of a scheduled bombardment, the target material was immediately unloaded and placed in a sharp-lipped tantalum crucible contained in a tantalum oven and introduced into the atomic beam machine. Optical line-up was effected and oven outgassing was carried out. Usually less than 2 hours elapsed between the completion of the cyclotron bombardment and the production of a stable beam. The duration of a run was ordinarily about 4 hours.

RESULTS

An initial search made at a low magnetic field at frequencies determined from Eq. (3) for all likely half-integral spin values yielded the results shown in Fig. 2. It is seen that I = 3/2 is strongly indicated. The transitions observed were seen at several higher fields, and the linear relation of transition frequency and field established. Two resolved resonances are shown in Figs. 3 and 4, and the observations are summarized in Table I.

In the last two columns of this table, the experimentally observed frequency and its compounded uncertainty are given. The compounded uncertainty, Δf_i , is defined by

$$(\Delta f_i) = \left[(\Delta v_i)^2 + (\frac{\partial v_i}{\partial H_i})^2 (\Delta H_i)^2 \right]^{1/2},$$

where Δv_i is the uncertainty associated with the finite line width of the neodymium resonance, $\frac{\partial v_i}{\partial H_i}$ is the rate of change of the resonant frequency with field, and ΔH_i is the uncertainty in the magnetic field.

Identification of the observed material as Nd^{141} is assured in several ways. The target material is known to be greater than 99% Pr^{141} by weight, from spectroscopic analysis. The method of production, the

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observed g_J value, the half-integral spin, and a half-life of 2.5 hours observed in the decay of several resonance foils serve to establish unambiguously that the observed material is Nd¹⁴¹.

CONCLUSIONS

The measured spin of 3/2 is in agreement with the shell-model prediction and the work of Polak et al.¹ as noted in the introduction.

The observed data make possible the setting of a lower limit to the zero-field hyperfine-structure separation between the F = 11/2 and F = 9/2 states, $\Delta v (11/2, 9/2)$. From second-order perturbation theory the deviation of an a resonance from the Zeeman frequency. δ_a , should be related to $\Delta v (11/2, 9/2)$ according to

$$\delta_{a} = 0.1190 \left(g_{J} \frac{\mu_{0}}{h} H \right)^{2} / \Delta \nu (11/2, 9/2).$$
(4)

Assuming the value of δ_a to be equal to or less than the compounded uncertainty at the highest observed field, we obtain

$$\Delta v(11/2, 9/2) > 1630 \text{ Mc/sec}$$

as a lower limit to the hyperfine separation.

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FOOTNOTES AND REFERENCES

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No.	Resonance type	Field (gauss)	Calculated Zeeman frequency (Mc/sec)	Observed frequency (Mc/sec)	Compound uncertainty
1	a	12.264	7.529	7.545	0.045
2	a	19.566	12.012	12.025	0.052
3	a	36.198	22.223	22.215	0.068
4	β	6.915	4.835	4.838	0057
5	β	19.566	13.680	13.665	0.062
6	β	36.198	25.310	25.325	0.113

Table I. Resolved resonances in Nd¹⁴¹. The Zeeman frequency is calculated on the assumption I = 3/2, J = 4, and $g_J = -0.6031$.

FIGURE LEGENDS

Fig. 1. Schematic energy-level diagram; I = 3/2.

Fig. 2. Low field spin search. Arrows along abscissa give predicted values of resonant frequencies for appropriate spin assignments.

Fig. 3. A resolved a-resonance.

Fig. 4. A resolved β -resonance.



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Fig. 2











Fig. 4

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