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Authors

Murray, Megan M Krone, Mary Griffin Bernstein, Summer L <u>et al.</u>

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Amyloid β -Protein: Experiment and Theory on the 21–30 Fragment

Megan M. Murray,[†] Mary Griffin Krone,[†] Summer L. Bernstein,[†] Andrij Baumketner,[‡] Margaret M. Condron,[§] Noel D. Lazo,[§] David B. Teplow,[§] Thomas Wyttenbach,[†] Joan-Emma Shea,^{†,|} and Michael T. Bowers^{*,†}

Department of Chemistry and Biochemistry, and Department of Physics, University of California, Santa Barbara, California 93106, University of North Carolina Charlotte, 9201 University City Boulevard, Charlotte, North Carolina 28262, Department of Neurology, David Geffen School of Medicine, and Molecular Biology Institute and Brain Research Institute, University of California, Los Angeles, California 90095

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The structure of the 21–30 fragment of the amyloid β -protein (A β) was investigated by ion mobility mass spectrometry and replica exchange dynamics simulations. Mutations associated with familial Alzheimer's disease (E22G, E22Q, E22K, and D23N) of A β (21–30) were also studied, in order to understand any structural changes that might occur with these substitutions. The structure of the WT peptide shows a bend and a perpendicular turn in the backbone which is maintained by a network of D23 hydrogen bonding. Results for the mutants show that substitutions at E22 do little to alter the overall structure of the fragment. A substitution at D23 resulted in a change of structure for A β (21–30). A comparison of these gas-phase studies to previous solution-phase studies reveals that the peptide can fold in the absence of solvent to a structure also seen in solution, highlighting the important role of the D23 hydrogen bonding network in stabilizing the fragment's folded structure.

Introduction

A growing body of evidence closely links the amyloid β -protein (A β) to Alzheimer's disease (AD). A β is the primary component of amyloid plaques found in the brain tissue of Alzheimer's patients. Originally the amyloid plaques themselves were believed to be the causative agents of AD, as outlined by the Amyloid Cascade Hypothesis.^{1,2} However, recent clinical and experimental studies have shown that soluble oligomers of A β are neurotoxic.³ The A β peptide typically exists as either a 40 or 42 amino acid residue peptide (A β 40 and A β 42, respectively). While A β 40 normally exists in the plasma at a concentration ~10-fold higher than that of A β 42, the latter peptide is the more neurotoxic⁴ and is the primary component of amyloid plaques. A small number of familial forms of AD exist that result in early onset of the disease. These forms typically arise from single amino acid substitutions of the primary structure of the peptide, all of which are known to occur at either the 22nd or 23rd residue of the A β protein.⁵⁻¹⁰

Limited proteolysis/mass spectrometry experiments¹¹ showed that both $A\beta(1-40)$ and $A\beta(1-42)$ contain a 10-residue trypsinresistant segment, Ala21-Ala30. The homologous decapeptide exhibited the same resistance to proteolysis, suggesting that the segment is particularly stable. Solution NMR studies with molecular dynamics simulations showed that the segment forms a turnlike structure at the center of the peptide, resides V24–K28. The stability of this region suggests that it may be involved in the folding and subsequent aggregation of the fulllength monomer peptide.

Proteolysis and solution-state NMR have also been employed to probe the stability and structure of the $A\beta(21-30)$ mutants

associated with familial Alzheimer's disease (FAD).¹² The results showed that the FAD-associated substitutions at Glu22 and Asp 23 have an increased susceptibility to proteolysis, compared to wild-type (WT) $A\beta(21-30)$. NMR experiments revealed that the increased susceptibility to proteolysis is caused by destabilization of the peptide's structure. This destabilization of the folding nucleus may explain the acceleration of $A\beta$ aggregation that is associated with FAD.

Here, we use nanoelectrospray (ESI) mass spectrometry (MS) and ion mobility spectrometry (IMS) to provide a structural characterization of WT $A\beta(21-30)$ as well as the mutant forms of the peptide associated with each type of hereditary AD. Ion mobility mass spectrometry (IM-MS) is a technique that has been successfully employed to study the full-length $A\beta$ peptide.^{13,14} These experiments are coupled with modeling in order to provide an accurate understanding of the molecular interactions that help to maintain the peptide's integrity as well as investigate possible conformational changes that may occur as a result of various FAD mutations.

Methods

Sample Preparation. The $A\beta(21-30)$ fragment, $A^{21}E^{22}D^{23}$ -VGSNKGA,³⁰ as well as the mutant forms associated with FAD: Italian (E22K),¹⁰ Arctic (E22G),^{7,8} Dutch (E22Q),⁹ and Iowa (D23N),⁶ were synthesized by FMOC (*N*-(9-fluorenyl)methoxycarbonyl) chemistry. All peptides were synthesized in their N-terminal N_{α} -acetyl and C-terminal carboxyamide (blocked) forms so as to mimic the structure of the segment within the full-length peptide. The samples were purified by reversed-phase HPLC, characterized by mass spectrometry and amino analysis,¹⁵ and lyophilized. The lyophilized peptide was dissolved in 25 mM ammonium acetate pH 7.4 for a final concentration of 400 μ M.

IM-MS Experiments. Mass spectra were obtained using an instrument built in-house with a nanoelectrospray ionization

 $^{^{\}dagger}$ Department of Chemistry and Biochemistry, University of California, Santa Barbara.

[‡] University of North Carolina Charlotte.

[§] University of California, Los Angeles.

Department of Physics, University of California, Santa Barbara.

(nano-ESI) source, ion funnel, temperature-controlled drift cell, and quadrupole mass filter.¹⁶ To record mass spectra, 5 μ L of sample solution is introduced into the instrument from goldcoated borosilicate capillaries (0.1 mm o.d./0.78 mm i.d.). The capillaries are made with a Sutter Instrument Co. P-97 Flaming/ brown micropipet puller and coated with gold by an Emitech K550X sputter coater. The ions are generated continuously at atmospheric pressure in the nano-ESI source and guided through the ion funnel to the analysis region of the instrument. The ion beam is passed through the drift cell, analyzed by the quadrupole mass filter, and detected.

Ion mobility measurements were recorded using the instrument described above.¹⁶ In ion mobility experiments, the ions are stored at the end of the ion funnel and pulsed into the drift cell filled with 5 torr of helium gas. The injection energy of the ions pulsed into the drift cell can be varied from near 0 to 150 eV. The ions in the cell are quickly thermalized by collisions with the helium buffer gas and are drawn through the cell under the influence of a weak electric field.

The force of the weak electric field and the frictional drag of the collisions with the He gas produce a constant drift velocity $v_{\rm D}$. The drift velocity is proportional to the applied field *E* where the proportionality constant *K* is termed the ion mobility:

$$v_{\rm D} = K \cdot E \tag{1}$$

As the ions exit the drift cell, they are detected as a function of time producing an arrival time distribution (ATD). Kinetic theory¹⁷ relates the ion mobility to the ion—He collision cross section (σ), which can in turn be related to the arrival time at the detector (t_A)

$$\sigma = 1.3 \left(\frac{q^2 E^2 T}{\mu k_{\rm B} p^2 N^2 \ell^2} \right)^{1/2} (t_{\rm A} - t_0) \tag{2}$$

where q is the ion charge, T is the temperature, μ is the reduced mass of the ion-He collision, N is the He number density at STP, ℓ is the drift cell length (4.503 cm), $k_{\rm B}$ is the Boltzmann constant, and t_0 is the time the ion spends outside of the cell before reaching the detector. All of the quantities on the right-hand side of eq 2 are either measured or known for each experiment, so that a very precise value of σ can be determined.

Simulations. In all our simulations the AMBER94 force field¹⁸ was used to describe the peptide. The temperature was maintained by the Nosé–Hoover algorithm¹⁹ with a coupling constant of 0.05 ps. The LINCS²⁰ algorithm was used to constrain covalent bonds involving hydrogens. The Lennard-Jones potential decays smoothly to 0 between 1.5 and 2.5 nm. Neighbor lists for the nonbonded interactions were updated every 1000 simulation steps. Short-range neighbors were defined within a cutoff of 4.0 nm. Generous twin-range cutoffs of 4.0 nm/4.0 nm were used to evaluate electrostatic interactions. No periodic boundary conditions are used. Simulations were performed using GROMACS software.^{21–23}

The initial structure for each peptide is taken from the central structure of the most populated cluster in the solution-phase simulations.²⁴ All-atom models constructed for the various 10-residue peptides are capped by a neutral acetyl and amide group at the N and C terminus, respectively. The species modeled here corresponds to the peptide at neutral pH. Here, $A\beta(21-30)$ has an overall -1 charge and contains two negatively charged sites (E22, D23) and one positively charged site (K28). The Arctic

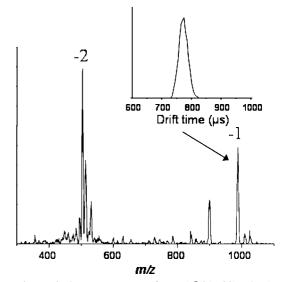


Figure 1. Typical mass spectrum of WT A β (21-30). The ATD for the z/n = -1 charge state is included as an inset (z is the charge and *n* is the oligomer number: n = 1 is a monomer, n = 2 is a dimer, etc). 82 × 80 mm (150 × 150 dpi).

mutant (E22G) and the Dutch mutant (E22Q) have no overall charge, and both contain one negatively charge site (D23) and one positively charged site (K28). The Italian mutant (E22K) has an overall +1 charge and has one negatively charge site (D23) and two positively charged (K22, K28). The Iowa mutant (D23N) has no overall charge and one negatively charged site (E22) and one positively charged site (K28). The replica-exchange algorithm²⁵ (REMD) was used to enhance equilibrium sampling. Forty replicas of the original system were exponentially spaced between 300 and 2200 K. At every 250 simulation steps (0.125 ps), exchanges were attempted among neighboring replicas and atomic coordinates were saved. Acceptance ratios for exchange ranged from 16% to 24%. Simulations were run for 28–30 ns, and all but the first 1 ns of the trajectories were analyzed.

Snapshots of the 300 K replica taken every 5 ps (~5800 structures) were clustered into families of structures. The GROMOS clustering method²⁶ is employed in which the root-mean-square deviations (rmsd) among conformations provide a measure of structural similarity. For each structure, the number of other structural neighbors, i.e., those with rmsd < 1 Å over C_{α} of resides 22–28, is calculated. The structure with the highest number of neighbors is taken as the central structure of a cluster (C1), which comprises this structure and all of its neighbors. All of the C1 structures are eliminated from the pool of available structures, and the procedure is repeated until all structures are eliminated.

Hydrogen bonds are considered present when the donor and acceptor are within 3.5 Å and the donor–acceptor-hydrogen angle is <60°. Salt bridges are defined as an interaction between oppositely charged groups in which the charged atoms are separated by \leq 4.5 Å. Cross sections of the model structures are calculated using a modified projection method.^{27,28}

Results

IM-MS. Figure 1 shows a typical mass spectrum for WT $A\beta(21-30)$. The spectrum has two groups of peaks corresponding to the -1 (m/z = 988) and -2 (m/z = 495) charge states with the addition of up to two sodium atoms (m/z = 1011 and 1034) for the -1 charge state and one sodium for the -2 (m/z

TABLE 1: Experimental and Theoretical Cross Sections for WT A β (21-30) and Its Mutant Forms^{*a*}

| | | cross sections (Å ²) | | | |
|--------|---------------|----------------------------------|------------------|-------------|--|
| mutant | charge (pH 7) | experimental ^b | theoretical c | % deviation | |
| WT | -1 | 227 | 231 | 1.73% | |
| D23N | 0 | 222.5 | 220 | 1.12% | |
| E22G | 0 | 210 | 208 | 0.95% | |
| E22K | +1 | 230 | 224 | 2.61% | |
| E22Q | 0 | 226 | 225 | 0.44% | |
| S26I | -1 | 229 | 233 | 1.75% | |
| D23G | 0 | 216 | 217 | 0.46% | |

^{*a*} Theoretical cross sections are given for the central structure of the most populated cluster (see Table 2 for spread of values). ^{*b*} Experimental cross sections correspond to the protein at pH 7. For proteins with a neutral charge, cross sections were taken for the +1 and -1 charge states and averaged. ^{*c*} Theoretical cross sections correspond to the protein at pH 7.

= 506) charge state. Although the -1 charge state is the solution charge state at pH 7.4, the -2 charge state is the dominant peak in the mass spectrum. Mass spectra were recorded for each mutant of $A\beta(21-30)$ and shared similar characteristics as those for the WT peptide. The mass spectra were recorded in positive or negative ion mode, according to the charge state of each species in solution. In the case of peptides with a neutral charge at pH 7, data were recorded for both positive and negative ions.

Ion mobility data was recorded for each charge state observed in the mass spectrum. A typical ATD of WT A β (21-30) is given as an inset of Figure 1. The ATD consists of one narrow and symmetric peak, suggesting that only one conformation of the peptide is present (or if multiple conformers are present, they either interconvert on the time scale of the experiment or have very similar cross sections). Therefore, the ATD of the z = -1 peak represents the peak for the monomer with a -1 charge. The ATDs for each A β mutant show a similar single, symmetric peak for the monomer. The experimental cross sections for each species are listed in Table 1.

Molecular Modeling. As shown in Table 2, most of the mutants of A β (21-30) spend a majority of time in a preferred family of structures (C1). The actual structures shown here are the central structures from this most populated cluster, C1. Figure 2a shows the central structure from the most populated cluster for the WT A β (21–30) at 300 K. The structure contains both a bend at residues G25-K28 and a turn at D23-S26. (A bend is defined as a region of curvature such that the bond angle formed by the three C_{α} atoms of residues i - 2, i, and i + 2 is at least 70°.29A turn is defined as a region where a hydrogen bond exists between CO of residue *i* and NH of residue i + n²⁹) The plane in which the turn lies is perpendicular to the bend at G25-K28. The bend at G25-K28 is stabilized by D23 O_{δ} side chain hydrogen bonds to G25, N27, and K28 amide hydrogen atoms (present in 96.6%, 99.0%, and 83.0% of the C1 structures, respectively) The turn in the peptide is stabilized by an O–N backbone hydrogen bond between residues D23(O) and S26(N) which is present in 82.8% of the structures in C1. There is a salt bridge at E22-K28 present in 100% of the C1 structures of the WT and a salt bridge between D23 and K28 in 99.2% of the structures.

Parts a and b of Figure 3 show the structures of the Arctic (E22G) and Dutch (E22Q) mutants, respectively, with the structure of the WT faded in the background. Both mutants show similar structural elements and intermolecular interactions to those of the WT. For all mutants, the positions of the termini do not match those of the wild type. The termini of the peptides were not included in the statistical analysis of the simulations

because they are not present within full-length A β . The turn and bend that are present in WT A $\beta(21-30)$ are maintained in the Arctic (E22G) and Dutch (D22Q) mutants. The rmsd from WT of E22Q is 0.943 Å and for E22G is 1.10 Å (Table 2). Table 3 shows that in E22G the amount of bonding between residues D23 and G25 and between D23 and N27 is noticeably lower (48.6% and 55.4%, respectively), and the percentage of hydrogen bonds formed with K28 is the same as in the WT (84.3%). Bonding between D23 O_{δ} side chain atoms and amide hydrogen of G25 is not present in E22Q, but the amount of D23 side chain bonding remains similar to that of the WT for D23 bonds with N27 and K28. The hydrogen bond between backbone atoms at D23(O) and S26(NH) are still present at high percentages for both E22G (71.9%) and E22Q (77.6%) (Table 4). A salt bridge formed between D23 and K28 in 98.7% and 99.9% of the C1 structures for E22G and E22Q, respectively, (Table 1).

The Italian (E22K) mutant has a higher rmsd from wild type of 2.67 Å although it still has similar structural elements to the WT (Figure 3c). This structure shows both the turn (K22–S26) and bend (G25-K28) that are present in the WT; however, the bonding motif is different. The turn at K22–S26 is stabilized by a K22(O) backbone hydrogen bond to S26(N) and G25(N) atoms (present in 74.2% and 98.4% of the structures, respectively). These bonds are not present in the WT, the Arctic mutant, or the Dutch mutant. The bend is stabilized by D23(O) side chain hydrogen bonds to N28 amide hydrogen atoms (present in 90.0% of the structures). A salt bridge at D23-K28 was present in 100% of the all of the C1 structures.

The structure of the Iowa form of the peptide (D23N) deviates the most from the WT, with a rmsd of 3.13 Å. The structure of $A\beta(21-30)$ D23N shows two turns at N23–N27 and at E22-S26 (Figure 3d). The first turn is stabilized by a backbone hydrogen bond between N23(O) and N27(N) atoms. The second turn (E22–S26) is maintained by another backbone hydrogen bond between E22(O) and S26(N) atoms, found in 90.7% of the structures. Backbone hydrogen bonds were calculated for the N_{δ} and O_{δ} atoms of the Asn side chain for this mutant, but these populated less than 25% of the C1 structures. For D23N, a salt bridge between E22 and K28 was present in 99.9% of the structures.

Solution-Phase vs Gas-Phase Structure. In solution-phase molecular dynamics studies,²⁴ the WT structures in cluster C1 contain a single backbone bend, while the WT structures in the second most highly populated structural cluster C2 have a single turn. In solution, these two structures are independent and in equilibrium with each other. This is in contrast to the results in the gas phase, where the bend and turn exist simultaneously, perpendicular to each other (Figure 2).

In solution, only a single salt bridge exists, located between residues E22 and K28. Substitutions at E22 remove this salt bridge although the backbone structure remains intact, suggesting that D23 hydrogen bonds are the primary forces stabilizing the structure of the fragment's backbone. In the gas phase however, two salt bridges occur: one between E22 and K28 and another between D23 and K28. Further details of the solution structures of the $A\beta(21-30)$ fragment and the related FAD mutants can be found elsewhere.²⁴

Discussion and Conclusions

A comparison between experimental cross sections and theoretical cross sections (Table 1) shows excellent agreement between experiment and theory. Deviations fall below 3% in all cases and below 2% in almost all cases. Substitutions at E22

TABLE 2: Most Populated Cluster of Structures Obtained from Replica Exchange Simulations at 300 K^a

| | % of structures in C1 | σ (Å ²) | rmsd from WT (Å) | E22-K28 salt bridge population (%) | D23-K28 salt bridge population (%) |
|------|-----------------------|----------------------------|------------------|---------------------------------------|---------------------------------------|
| WT | 99 | 231 ± 6.1 | _ | 100.0 | 99.2 |
| D23N | 61 | 222 ± 4.2 | 2.67 | 99.9 | _ |
| E22G | 79 | 210 ± 6.0 | 1.10 | _ | 98.7 |
| E22K | 99 | 225 ± 2.2 | 3.13 | _ | 100.0^{b} |
| E22Q | 89 | 225 ± 3.9 | 0.94 | _ | 99.9 |

^{*a*} The percentage of all structures belonging to the most populated cluster is given. The average cross section and standard deviation are calculated over 10 structures within C1. The root-mean-square deviation of C_{α} distances from the central structure of the WT C1 cluster is calculated for the C_{α} atoms of residues 22–28 only. The percentage of structures within cluster C1 that contain salt bridges among residues E22, D23, and K28 is given. ^{*b*} In the E22K mutant, a salt bridge exists between K22 and D23 in 100% of C1 structures.

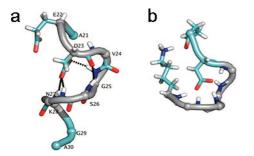


Figure 2. Central structures of the C1 cluster for WT A β (21–30) from gas-phase simulations (a) and solution-phase simulations (b). 83 × 50 mm (150 × 150 dpi).

TABLE 3: Percentages of C1 Structures Containing a Hydrogen Bond between One or Both D23 Side Chain O_{δ} Atoms and the Given Amide Hydrogen Atom^{*a*}

| | V24(NH) | G25(NH) | S26(NH) | N27(NH) | K28(NH) |
|------|---------|---------|---------|---------|---------|
| WT | 3.9 | 96.6 | 16.7 | 99.0 | 83.9 |
| E22G | 1.5 | 48.6 | 18.3 | 55.4 | 84.3 |
| E22K | 0.0 | 0.0 | 4.1 | 12.4 | 90.0 |
| E22Q | 0.2 | 1.3 | 36.1 | 95.9 | 71.9 |

 a Analogous hydrogen bonds were also calculated for the N_{δ} and O_{δ} atoms of the Asn side chain in the D23N mutant, and no hydrogen bonds are populated in more than 25% of the C1 structures for these cases.

do little to alter the turn-bend structure of the decapeptide, with subtle changes in the factors that stabilize this structure, as discussed here. Both E22Q and E22G show the same D23 O_{δ} backbone NH hydrogen bonds, which act to preserve the bend in residues 23-28. Still, these bonds are not as highly populated as in the wild type. E22Q has the lowest rmsd from WT, consistent with the very similar resistance to proteolysis displayed by the two peptides.¹² Hydrogen bonding between D23 and N27 is present in E22Q and WT but not present in E22G. Conversely E22G maintains a hydrogen bond between D23 and G25, which is not present in E22Q. Since both mutations have similar rmsd to the WT, we reiterate that these bonding differences have little impact on the core structure. The structure of E22K contains a dissimilar network of backbone hydrogen bonds to the WT but, perhaps surprisingly, has a similar structure.

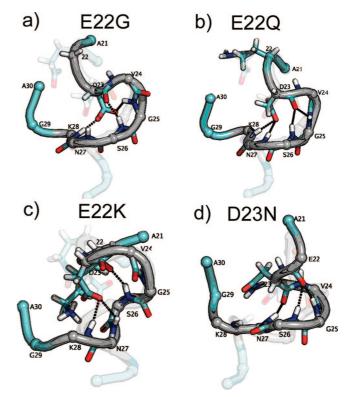


Figure 3. Gas-phase structures for $A\beta(21-30)$ mutants: Arctic E22G (a), Dutch E22Q (b), Italian E22K (c), and Iowa D23N (d). All structures are the central structure of cluster C1. The structure of the WT peptide is shown as a transparent image behind each mutant. For all the structures, the bend region at residues 23-28 is shown in gray and dotted lines denote key hydrogen bonds. 83×89 mm (150×150 dpi).

The Iowa substitution, D23N, on the other hand, does result in a change of backbone structure. Here, a negatively charged aspartic acid residue is replaced with a neutral asparagine residue. With this substitution, the hydrogen bonding between residue 23 and the surrounding residues is disrupted, suggesting that the D23 network of hydrogen bonds is primarily responsible for maintaining the backbone structure in the WT peptides, as well as in the E22X mutants. This is further emphasized by the decreased population of the C1 cluster of the D23N mutant, as compared to the E22X mutants (Table 2). With a mutation at

| TABLE 4: | Percentage of C1 | Structures Co | ontaining a I | Hvdrogen I | Bond between | Two Backbone Atoms |
|----------|------------------|---------------|---------------|------------|--------------|---------------------------|
| | | | | | | |

| | 23(O)-G25(NH) | 23(O)-S26(NH) | 23(O)-N27(NH) | 22(O)-G25(NH) | 22(O)-S26(NH) |
|------|---------------|---------------|---------------|---------------|---------------|
| WT | 3.0 | 82.8 | 0.0 | 0.0 | 0.0 |
| D23N | 0.0 | 0.2 | 0.0 | 17.9 | 90.7 |
| E22G | 78.3 | 71.9 | 0.0 | 0.0 | 0.0 |
| E22K | 0.4 | 2.0 | 45.5 | 98.4 | 74.2 |
| E22Q | 34.9 | 77.6 | 13.1 | 0.0 | 0.0 |

21–30 Fragment of Amyloid β -Protein

residue 23, the peptide is able to access slightly more conformational space than the WT and other E22X mutants.

These results confirm the presence of a turn in the 24-28region of the peptide, as previously reported in NMR experiments.^{11,12} Proteolysis experiments on the mutant forms of $A\beta(21-30)$ by Grant et al.¹² showed that substitutions at E22 and D23 increase the peptide's susceptibility to digestion, suggesting that these changes act to destabilize the core structure, perhaps encouraging higher order A β assembly. Our results for the Iowa form support this hypothesis, showing a change from a bend at G25-K28 with a turn at D23-S26 for the wild-type $A\beta(21-30)$ to two turns between residues N23-N27 and E22-S26 in the mutant. On the other hand, our calculations on substitutions at E22 show that none of these FAD associated mutations significantly alter the backbone structure of WT $A\beta(21-30)$. Regardless, subtle changes in structure for the various E22X mutants may either alter the intramolecular interactions between the A β (21-30) segment and the rest of the full-length peptide or the intermolecular interactions between the A β monomers. Consequently, the kinetics of monomer folding and the ensuing peptide assembly processes may be altered by these mutations.

This hypothesis however, does not explain the differences between the results presented here and a previous proteolysis study on $A\beta(21-30)$.¹² In those studies, the E22G, E22Q, and E22K mutants all showed decreased resistance to digestion relative to the WT. There can be several reasons for the differences. First, the proteolysis occurs at K28, which is at the C-terminal end of the conserved core structure of the peptide. Hence, other factors may dominate the trypsininduced proteolysis. Second, the proteolysis is a relatively slow process and may occur from minor structural components that are not considered in our structural analysis of Figure 3. In fact, a central feature of the kinetic analysis of the proteolysis experiments was the equilibrium among protease-sensitive and protease-insensitive conformers. Finally, proteolysis occurs in solution, and solution structures²³ are more variable than our solvent free structures, where the vast majority of the structures fall in a single cluster family. For example, the population of cluster C1 for the gas- phase structure of WT A β (21-30) is 99%, while the population in solution is 44%.

Overall, however, the above conclusions are consistent with those drawn from our solution-phase molecular dynamics simulations,²⁴ even though some important differences result from the loss of solvent. The small size of the peptide ensures that it is completely desolvated during the IM-MS experiments. Without the presence of solvent, electrostatic interactions and salt bridges play a significant structural role. The presence of stable salt bridges in the gas phase is primarily responsible for the simultaneous existence of the bend and turn that is found in the gas phase for the WT and the E22X mutants. In solvent, the D23–K28 salt bridge is not present. As a consequence, the fraction of structures containing a backbone bend is substantially reduced, although the bend remains the dominant structural element for the WT and E22X mutants.

Conversely, the D23N structure consists of a single turn in both the gas and solution phase. This provides further evidence that the D23-K28 salt bridge is the cause of the structural difference upon transitioning from solution to gas phase for the WT and E22X mutants.

Simulations in explicit solvent²⁴ show that a network of D23 hydrogen bonds serves as the primary source of structural

stabilization for the bend in the peptide backbone of the WT and E22X mutants. This same network of D23 hydrogen bonds still exists in the gas phase even in the presence of the D23–K28 salt bridge. Even though the gas-phase structure is completely desolvated, the important interactions from the biologically relevant solution- phase structure remain intact. These results speak to the persistence of the fundamental intramolecular interactions in these peptides and why they are so resistant to proteolysis.

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