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Title

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Journal

Journal of the Less Common Metals, 111(1-2)

ISSN

0022-5088

Authors

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Publication Date

1985-09-01

DOI

10.1016/0022-5088(85)90193-6

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BULK AND SURFACE VALENCE OF YbBe₁₃*

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Summary

We report angle-integrated photoemission results on clean and oxidized YbBe₁₃ samples, at and below 300 K, which show YbBe₁₃ to have a trivalent bulk ($\bar{v} = 2.9^{+0.1}_{-0.02}$) and a divalent surface layer. The bulk valence, independently derived from temperature-dependent L_{III}-edge X-ray absorption measurements, is found to be three.

1. Introduction

The isostructural XBe₁₃ alloys (X = rare earth or actinide element) form an interesting class of materials. For superconducting UBe₁₃ [1] and magnetic NpBe₁₃ [2], an anomalously high electronic specific-heat value has been reported ($\gamma \sim 1 \text{ J mol}^{-1} \text{ K}^{-2}$) which classifies these materials as heavyfermion systems. On the other hand, CeBe₁₃ provides evidence of a configurational instability of the Ce 4f shell [3, 4]. The subject of this paper, YbBe₁₃, has been found to order antiferromagnetically below $T_N \simeq 1.3 \text{ K}$ [5], but above T_N it shows an anomalously large linewidth in inelastic magnetic neutron scattering, which may indicate that the Yb ions are also near a valence instability [6]. The latter could also be inferred from Mössbauer measurements for the paramagnetic state [7]. The lack of volume dependence of resistivity features suggests, however, that YbBe₁₃ is not mixed-valent in the ground state, but above T_N resistive anomalies exist, possibly owing to "Kondo-sideband" scattering [8]. Here, we apply bulksensitive L_{III}-edge X-ray absorption and photoemission spectroscopy at

^{*}Paper presented at the International Rare Earth Conference, ETH Zurich, Switzerland, March 4 - 8, 1985.

 $h\nu = 40.8 \text{ eV}$ (*i.e.*, bulk and surface sensitive) to YbBe₁₃. We find an essentially trivalent bulk with a divalent surface layer, as was previously found, for example, for Sm [9], EuPd₃ [10] and TmS [11]. Divalent surface layers are generally present on all promotional-type mixed-valent systems, *e.g.*, TmSe [11], Sm_xY_{1-x}S [12], and Yb_xY_{1-x}Al₂ [13].

2. Experimental details

Flux-grown single crystals of YbBe₁₃ [8] were fractured under a vacuum of $p < 1 \times 10^{-10}$ Torr and immediately irradiated with He(II) ($h\nu = 40.8 \text{ eV}$) radiation from a helium discharge lamp (Leybold-Heraeus) in a system originally designed for inverse photoemission [14]. Photoemission energy-distribution curves (EDCs) were taken within 2 min using a 180° hemispherical energy analyzer (Leybold-Heraeus). The overall resolution (electrons and photons) was 0.2 eV. The sample could be cleaved at room temperature as well as at 100 K.

For the X-ray absorption measurement, a sample previously used for photoemission was finely powdered under a dry argon atmosphere and pressed into paraffin wax. The intensities of the incident and transmitted Xrays were monitored as a function of energy using tunable synchrotron radiation from the EXAFS-II spectrometer at the HASYLAB, DESY, Hamburg [15]. Near the Yb-L_{III} edge, around 8.9 keV, the energy resolution was 2 eV. Data were taken at 300 K and 5 K.

3. Results and Discussion

In Fig. 1(a) we present a photoemission EDC of clean $YbBe_{13}$ fractured at room temperature. We note the $Yb^{2+}(4f^{13})$ doublet near the Fermi level $E_{\rm F} = 0$. The Yb³⁺ (4f¹²) multiplet can be recognized in the binding-energy range 5-11 eV similar to that in mixed-valent YbAl₂ [13]. For clarity, in Fig. 1(a) we have included the bar diagram representing the $4f^{13}$ final-state multiplet taken from Tm^{2+} [11] with an increase of the splittings by 10% owing to the additional nuclear charge in Yb. As becomes clear from the EDC in Fig. 1(a), the Yb^{3+} multiplet peaks rest on a broad emission feature with a maximum around 7 eV. By comparison with a He(I) ($h\nu = 21.2 \text{ eV}$) EDC (not shown), we attribute this broad emission to s,p states of Be similar to those in UBe₁₃ [16]. There have been discussions as to whether oxygen 2pemission could be the origin of this feature [17, 18]. To check this, we have exposed clean YbBe₁₃ to 0.75 L O₂ (1 L = 1 Langmuir = 10^{-6} Torr × 1 s), Fig. 1(b). The changes are dramatic. The Yb³⁺ emission is completely suppressed by the huge increase in the O 2p emission (note the factor, 2, in Fig. 1(a) relative to Fig. 1(b), indicating the high sensitivity of the surface to O_2 . This leads us to believe that the freshly-fractured surface is oxygenfree. In fact, further exposure of O_2 increased the O 2p emission as can be



Fig. 1. Angle-integrated photoemission EDCs at $h\nu = 40.8$ eV for (a) clean YbBe₁₃; (b) plus 0.75 L O₂. Difference curves; (c) "0.75 L O₂" minus "clean"; (d) "1.5 L O₂" minus "0.75 L O₂".

seen from the difference spectra in Figs. 1(c) and 1(d). The saturation level was only reached after $10 L O_2$.

As an additional check, we fractured YbBe₁₃ at 100 K to avoid oxygen diffusion from the bulk [18]. Again, we found an EDC similar to the one in Fig. 1(a), in particular the trivalent Yb multiplet resting on top of the broad Be s,p emission. The difference between cold and warm-fractured YbBe₁₃ is an ~ 20% reduced intensity of the Yb²⁺ doublet near E_F . This may indicate some Yb atoms segregating to the surface at room temperature. Upon oxidation, Yb forms trivalent Yb₂O₃, but the multiplet structure is completely smeared out by the O 2p emission [19]. The fact that we can clearly discern the f¹² peaks in fig. 1(a) is another indication that the fractured YbBe₁₃ surfaces are clean and that the underlying broad s,p emission is Be-derived.

We now turn to the Yb²⁺ (4f¹³) doublet near the Fermi level. It is interesting to find the ${}^{2}F_{7/2}$ component at 0.9 eV and a shoulder cut by E_{F} . For a homogeneously mixed-valent system, the bulk divalent component has to be pinned at $E_{\rm F}$, if screening effects are neglected [13]. On the other hand, it has now been established that all mixed-valent systems (with the exception of Ce systems) have a divalent surface layer, e.g., EuPd₂Si₂ [10], TmSe [11], Sm_xY_{1-x}S [12], and Yb_xY_{1-x}Al₂ [13]. Even some trivalent systems such as Sm [9], EuPd₃ [10] and TmS [11] exhibit a divalent surface layer. From the non-zero binding energy of the intense f¹³ doublet in Fig. 1(a), we can immediately conclude that YbBe₁₃ has a divalent surface layer. The 0.9 eV position of the surface ${}^{2}F_{7/2}$ component is close to the value found for Yb_xY_{1-x}Al₂ [13], which is around 1.1 eV. By contrast, elemental Yb metal exhibits the ${}^{2}F_{7/2}$ surface component at 1.7 eV binding energy [20].

The question now arises, whether a bulk $4f^{13}$ doublet with low spectral intensity could cause the shoulder near E_F in Fig. 1(a), as would be expected for a nearly trivalent ($\bar{v} < 3$) system. We have therefore tried to fit the experimental spectrum in the range 0 - 4.5 eV with two sets of Lorentzian broadened f^{13} doublets (ratio of the 5/2 - 7/2 component 0.75; spin-orbit splitting 1.27 eV [13]) for the bulk and the surface. Although, in principle, the EDC could be fitted by a single surface doublet, owing to the ambiguity in subtracting the underlying Be emission near $E_{\rm F}$, 10% of the divalent spectral intensity could stem from a Yb²⁺ bulk doublet. Hence, the lower limit for the bulk mean valence amounts to $\bar{v} = 2.9^{+0.1}_{-0.02}$. An independent experimental check could involve quenching of the surface emission by oxygen exposure, which would leave a possibly existent divalent bulk f¹³ emission unaffected. While it is known that this works quite well with rareearth NaCl compounds [11, 12], intermetallic alloys such as YbAl₂ [13] do not clearly show this effect, probably because of the simultaneous oxygen reaction of the non-rare-earth constituents in the alloy.

For YbBe₁₃, we have studied the behavior of the surface $4f^{13}$ emission with O₂ exposure in Figs. 1(b) - (d). The first 3/4 L O₂ clearly reduces the divalent surface emission, as is clear from the difference spectrum of Fig. 1(c). The remaining emission, however, is still not pinned at E_F in Fig. 1(b), pointing to the non-existence of a bulk Yb²⁺ signal. Further O₂ exposure results in an increase of the spectral intensity in the 0 - 4 eV range (see the difference curve of Fig. 1(d)), which saturates after 10 L O₂. This complex behavior cannot be used to clarify the question of a small Yb²⁺ signal hidden in the intense Yb²⁺ surface emission of Fig. 1(a).

To obtain a definitive answer, we measured the X-ray absorption at the Yb L_{III} edge at 8.934 keV. This technique is bulk sensitive only. Our results are displayed in Fig. 2. The experimental curve measured at 5 K is identical to a spectrum of YbBe₁₃ at 300 K. This is in contrast to YbPd₂Si₂ with a temperature-dependent mean-valence which varies from $\bar{v} = 2.9$ at 300 K to 2.8 at 16 K [15]. A fit to the L_{III} -edge profile performed in the same way as for YbPd₂Si₂ finally revealed that no divalent Yb is present in bulk YbBe₁₃, as the photoemission results could only suggest (see the discussion above).



Fig. 2. Yb L_{III} -edge X-ray absorption of YbBe₁₃. The dotted line represents the experiment at a sample temperature of 5 K and is identical to a spectrum taken at 300 K. The solid line is a fit using a single Yb³⁺ L_{III} edge. Deviations at higher energies are due to the EXAFS oscillations.

4. Conclusion

From temperature-dependent L_{III} -edge measurements, we derive that $YbBe_{13}$ is trivalent in the bulk. With photoemission spectroscopy, we find a trivalent bulk and a divalent surface layer. The Be s,p emission produces a broad emission feature centered around 7 eV.

Acknowledgments

We thank H. Statz, IBM Germany, for his interest and support, and W. Krone, G. Kaindl, R. Schlittler, G. Wortmann, and the HASYLAB staff at DESY, Hamburg for their help.

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