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Electron scattering from 1-Methyl-5-Nitroimidazole: Cross sections for modelling electron transport through potential radiosensitizers

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Abstract: In this study we present a complete set of electron scattering cross sections from 1-Methyl-32 5-Nitroimidazole (1M5NI) molecules for impact energies ranging from 0.1 to 1000 eV. This infor-33 mation is relevant to evaluate the potential role of 1M5NI as a molecular radiosensitizers. The total 34 electron scattering cross sections (TCS) we previously measured with a magnetically confined elec-35 tron transmission apparatus have been considered as reference values for the present analysis. Elas-36 tic scattering cross sections have been calculated by means of two different schemes: The Schwinger 37 Multichannel (SMC) method for the lower energies (below 15 eV) and the Independent Atom Model 38 based Screening Corrected Additivity Rule with Interferences (IAM-SCARI) for higher energies 39 (above 15 eV). The latter has also been applied to calculate the total ionization cross sections which 40 have been complemented with experimental values of the induced cationic fragmentation by elec-41 tron impact. Double differential ionization cross sections have been measured with a reaction mi-42 croscope multi-particle coincidence spectrometer. Using a momentum imaging spectrometer, direct 43 measurements of anion fragment yields and kinetic energies by dissociative electron attachment are 44 also presented. Cross sections for other inelastic channels have been derived with a self-consistent 45 procedure by sampling their values at a given energy to ensure that the sum of the cross sections of 46 all the scattering processes available at that energy coincides with the corresponding TCS. 47

Keywords:electron scattering cross sections; electron impact molecular fragmentation; molecular48radiosensitizers; radiation damage; ionization; dissociation49

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1. Introduction

Using high-atomic number (Z) elements to enhance the energy deposition (absorbed 52 dose) during X-ray irradiation of living tissues was introduced long time ago [1]. Metal 53 atoms (Au, Pt, Gd) and metallic nanoparticles [2-4] have been extensively studied as po-54 tential radiosensitizers in conventional (photon beam based) radiotherapy treatments [5]. 55 The absorption coefficient of photons in matter increases with Z, together with the prob-56 ability of generating Auger electrons [6]. These metallic structures have also proven to 57 enhance the energy deposition of charged particle beams (electrons, protons and heavy 58 ions) as those used in new advanced radiotherapy techniques such as electron-flash in-59 traoperative radiotherapy [7], proton-therapy [8] and heavy ion (He, C, O) beam radio-60 therapy [9, 10, 11]. In these cases, the radiosensitizing mechanisms are not well under-61 stood and could be related to induced secondary processes on the surface of the nanopar-62 ticles and their molecular coating [12]. Recently, the concept of molecular radiosensitizer 63 [13] has been introduced as a more appropriate targeting procedure for charged particle 64 beams. These molecules enhance radiation effects in tumoral areas with less toxicity than 65 heavy atom nanoparticles. Basically, low energy secondary electrons, generated by the 66 primary beam, dissociate the molecular radiosensitizer [14] and create abundant reactive 67 radical species that are able to efficiently attach to the DNA molecular components or 68 reduce the tumor hypoxia, which is a barrier to effective radiation therapy [15]. Note that 69 the main goal of radiotherapy is damaging the DNA of tumoral cells but preserving, as 70 much as possible, the DNA in healthy tissue. In this context, electron induced dissociation 71 to potential molecular radiosensitizers are critical processes and it is important to under-72 stand their radiosensitizing mechanisms. In order to model the radiation effects when mo-73 lecular radiosensitizers are deposited in the tumor, the corresponding electron induced 74 chemistry needs to be characterized in terms of interaction probabilities (cross sections) of 75 all the available reaction channels at a given electron energy. 76

The main goal of this study is to provide a self-consistent electron scattering data set 77 for a potential molecular radiosensitizer, 1-methyl-5-nitroimidazole (1M5NI), within the 78 impact energy range (0.1-1000 eV). This consistent data set has been obtained by combin-79 ing the theoretical and experimental data available in the literature with our own meas-80 urements and calculations of the total electron scattering, integral elastic and inelastic 81 cross sections, as well as the induced cationic and anionic fragmentation. 82

The remainder of this article is organized as follows: Section 2 provides the present 83 results and their interpretation. These results are discussed in Section 3 together with a 84 critical comparison with previous data. Theoretical and experimental methods used in 85 this study are described in Section 4. 86

2. Results

In order to obtain a comprehensive self-consistent data set, for modelling purposes, 88 we followed the procedure described in previous studies (see ref. [16] and references 89 therein). Essentially, accurate measurements of the total electron scattering cross sections 90 (TCS) are used as reference values to ensure the self-consistency of the adopted database. 91 The sum of the cross sections of all the scattering processes, which are available at a given 92 energy (open channels), should give the reference TCS value at that impact energy. In this 93 case the reference values have been derived by combining the present TCS measurements 94 in the range 0-300 eV with our calculated values for impact energies above 300 eV, as 95 described in the next subsection. 96

2.1. Total electron scattering cross sections

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As already mentioned, our recently published TCS measurements [17] have been 98 taken as reference values for electron impact energies from 1 to 300 eV. These results were 99 obtained with a magnetically confined electron beam apparatus [18], and their assigned 100uncertainties are within 5%. TCS measurements based on the attenuation of a linear elec-101 tron beam passing through a low-pressure gas cell containing the target of interest are 102 probably the most accurate results that we can use for this purpose. They are affected by 103 the "missing angles" systematic error (see ref [18] for details), but they have been corrected 104according to the procedure described below. 105

Table 1. Experimental results of the total electron scattering by 1M5NI cross sections and the corre-106sponding corrected values, accounting for the elastic scattering in the "missing angles (MA)" (see107text for details).108

$\mathbf{F}(\mathbf{a}\mathbf{V})$	TCS_exp	TCS+MA	$\mathbf{F}(\mathbf{aV})$	TCS_exp	TCS+MA
E(ev)	(10 ⁻²⁰ m ²)	(10 ⁻²⁰ m ²)	E(ev)	(10 ⁻²⁰ m ²)	(10 ⁻²⁰ m ²)
1	84.4	84.4 90.13		67.8	69.47
1.2	81.8	87.77	8	69.7	71.47
1.4	78	84.10	8.2	67.6	69.46
1.5	80.4	86.52	8.4	66.4	68.34
1.6	86.3	92.40	8.6	62.9	64.93
1.8	90.2	96.13	8.8	64.8	66.92
2	92.6	98.16	9	65.7	67.93
2.1	89.2	94.49	9.2	67.3	69.66
2.2	86.8	91.77	9.5	64.7	67.26
2.3	82.3	86.93	9.8	67.2	69.93
2.4	78.8	83.06	10	73.2	75.99
2.6	74.7	78.23	10.5	71.4	74.13
2.8	74	76.85	11	74.2	76.86
2.9	75.7	78.27	12	71.8	74.77
3	76.7	79.06	13	72	74.90
3.2	76.1	78.20	14	71.6	74.84
3.4	75.5	77.57	16	71.5	74.12
3.6	72.8	74.97	18	70.5	73.05
3.8	68.5	70.88	20	70.7	73.18
4	68.1	70.72	22	69.1	71.52
4.1	66	68.72	25	69	71.34
4.2	63.2	66.02	30	67.9	70.13
4.4	65	67.98	35	66.4	68.55
4.6	65.9	69.00	40	64.3	66.37
4.8	65.9	69.06	45	62.7	64.71
5	65.3	68.46	50	59.7	61.65
5.2	64.6	67.68	55	57.9	59.80
5.4	63.5	66.46	60	56.3	58.16
5.6	62.4	65.17	65	53.9	55.72
5.8	64.5	67.07	70	52.1	53.88
6	66.9	69.24	80	49.6	51.32

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6.2	70.4	72.51	90	47.9	49.56
6.4	67.6	69.51	100	46.2	47.81
6.6	64.4	66.12	120	42.6	44.13
6.8	68.6	70.18	150	39.3	40.74
7	68.3	69.80	200	34.8	36.12
7.2	65.1	66.57	250	31.3	32.54
7.4	66.8	68.31	300	27.6	28.78

69.98

For energies above 20 eV we have used our screening corrected additivity rule, 110 within the framework of the independent atom model, (IAM-SCAR) procedure [19] in-111 cluding interference effects (IAM-SCARI) [20] to calculate differential elastic as well as 112 integral elastic, inelastic, and therefore total, scattering cross sections. This method has 113 proven to be reliable, within 10%, for a large number of molecular targets [16] and impact 114 energies above 20 eV. In the case of 1M5NI, the agreement of the calculated TCS with the 115 experimental results is excellent [17] for impact energies higher than 20 eV. Accordingly, 116 we have used the calculated values to extrapolate the present experimental TCS values up 117 to 1000 eV. 118

As described in previous publications [18, 21], calculated differential elastic cross sections (DCS) can be used to quantify the magnitude of the "missing angle" effect by inte-120 grating these DCS values over the acceptance angle of the detector for each electron incident energy. The corresponding results are shown in Table 1. Considering that the exper-122 imental values have been corrected for this systematic error, we have estimated an overall 123 uncertainty limit of about 10% for the present TCS reference values. 124

2.2. Differential and integral elastic cross sections

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For energies below 15 eV, we used the Schwinger multichannel (SMC) method [22, 23] to calculate the differential and integral elastic scattering cross sections [17].

For intermediate and high energies (15-1000 eV) we used our IAM-SCARI method 128 [19-20]. It is based on an independent atom representation but it includes the screening of 129 the atomic cross sections within the molecule and considers interference effects due to the 130 multicenter scattering process. Details on the calculation are given in Section 4 (Materials 131 and Methods). 132

Representative differential elastic cross section (DCS) for impact energies ranging 133 from 1 to 1000 eV are shown in Figure 1. Numerical values of these DCS and the corre-134 sponding integral elastic scattering cross section (ICS), derived by integrating the differ-135 ential values over the whole scattering angle range (0-180 deg), are shown in Table S1 (see 136 Supplementary Information). These data have been calculated with the above procedures, 137 from 0.1 to 15 eV the SMC data have been chosen while the IAM-SCARI results have been 138 taken for energies above 15 eV. For the overlapping impact energy (15 eV) there is a good 139 agreement between both methods. At that energy, the SMC results tend to be lower than 140 the IAM-SCARI in the forward direction (close to 0 degrees) which is due to the limited 141 number of partial waves included in this calculation (no dipole Born corrections [24] are 142 included). There is an excellent agreement from 4 to 27 deg but for higher scattering an-143 gles, although they have a similar shape, the minimum cross section around 120 deg is 144much more pronounced in the IAM-SCAR calculation (see Figure 1). Nonetheless, by in-145 tegrating over the whole scattering angle range (0-180 degrees) the agreement between 146 both ICS values (58.64 and 60.48, respectively) was found to be better than 3%. 147

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Figure 1. Representative differential elastic cross section calculated with the SMC methodh for the (0.1-15 eV) impact energy range and the IAM-SCARI from 15 to 1000 eV. The agreement between both methods at 15 eV is discussed in the text (numerical results of the full calculation are given in Table S1 of the Supplementary Information).

2.3. Differential and integral inelastic cross sections

For energies above the threshold energy to excite the different inelastic channels (rotational excitation, electron attachment, vibrational excitation, electronic excitation and 155 ionization) differential and integral cross section related to the channel in question need 156 to be known in order to obtain a realistic representation of the scattering problem. In the following subsubsections, these inelastic channels will be analysed. 158

2.3.1. Electron attachment cross sections

Incident electrons may be temporarily trapped in the potential well formed by the 160 potential of the target molecules in combination with the centrifugal potential of the inci-161 dent electron leading to the formation of a resonance (unstable anion with short lifetime). 162 This can happen from the ground state of the target (shape resonance) or involve one of 163 its electronic excited states (Feshbach resonance). Resonances can be experimentally ob-164 served as local maxima in the total cross section energy dependence [17] or through elec-165 tron transmission measurements [25], which can be combined with a time of flight (TOF) 166 analysis of the formed anion fragments (see Ref. 25 for details). Although they are essen-167 tially inelastic processes, normally appear as sharp peaks connected with the calculation 168 of the integral elastic cross sections [17, 26]. The excess energy associated with the for-169 mation of the parent anion can be released along different pathways: autodetachment fol-170 lowed by energy relaxation, autodetachment followed by neutral dissociation or anionic 171 dissociation. The latter channel is commonly known as dissociative electron attachment 172

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(DEA) and can be experimentally studied by means of a mass/charge analysis of the produced fragments after the electron attachment process. In most cases this is carried out
with a time of flight (TOF) spectrometer, which means that only anionic fragments, together with the parent anion, can be analyzed. More complete experiments may incorporate a quadrupole mass spectrometer to analyze neutral fragments but, as far as we know,
this information is not available for 1M5NI.

In order to obtain a full description of the electron attachment (EA) process, in terms 179 of EA cross sections, we need to compile and critically discuss the results of integral elastic 180 scattering cross sections (accounting for inelastic channels in the calculation procedure), 181 total electron scattering cross section measurements and charge/mass experimental anal-182 ysis of produced fragments. In addition, high-resolution electron transmission experi-183 ments are highly valuable to verify the positions and widths of the theoretically predicted 184 resonances. We started our analysis with the energy dependence of our integral elastic 185 cross section that was calculated with the SMC method (see Section 4 for details). The 186 results of this calculation are shown in Figure 2. 187



Figure 2. Integral elastic cross section calculated with the SMC method (see text for details).

We considered that the local maxima in the IECS values shown in Figure 2 are due 190 to resonant electron attachment while the pure IECS follows a smooth energy dependence 191 as it corresponds to elastic processes. This assumption is supported by our TCS measure-192 ments which confirm the positions of these resonances (see Figure 2). The electron attach-193 ment cross sections are then derived by extracting the resonances from the elastic plus 194 electron attachment cross section curve. Some small resonances appearing at energies 195 above 10 eV, which are not confirmed by any experimental evidence, have been consid-196 ered as pseudo-resonances, i.e., artifacts originated by not including inelastic channels in 197 the elastic scattering cross section calculation. Numerical EA cross section data are shown 198 in Table 3. 199

2.3.2. Anion yield analysis

We investigated the anion fragmentation produced via dissociative electron attachment to 1M5NI by using a momentum imaging spectrometer, which has been described in detail previously [27]. The most relevant details of the experimental systems are described in Section 4.

Figure 3 shows the mass-resolved relative yields of anions produced in dissociative 205 electron attachment to 1M5NI in the resonant 3.1 eV – 4.7 eV energy range where the anion 206 product yields were observed to be highest. The most abundant anions are NO2⁻ and CN-207 , followed by heavier anions having lost neutral O, OH, CH₃, NO, and/or NO₂ radicals. 208 These results agree with those previously obtained by Tanzer et al. [28]. Note that, the 209 mass resolution (m/ Δ m \approx 20) of the present momentum imaging spectrometer does not 210 allow us to distinguish the number of hydrogen atoms remaining in most cases. At these 211 attachment energies, there is no significant yield of H- produced from the sample. For the 212 3.1 eV and 4.7 eV measurements, a small peak is clearly visible at m/q = 35 u, correspond-213 ing to either C₃⁻ or H₂O₂⁻. Also clearly visible at all energies is a peak around m/q = 54 u, 214 which is possibly due to $C_2N_2H_x$, for x=1,2 or 3. 215

The branching ratios (BRs), representing the contribution of a specific anion with re-216 spect to the sum of all detected fragmentation channels, have been measured for 1M5NI 217 and are shown in Figure 4. At 3.1 eV electron energy, the contribution of NO2⁻ accounts 218 for ~58% of the total anion yield which reduces to ~35% at 4.7 eV. The relative cross-section 219 for NO_2^{-1} production exhibits a peak around 3 eV, and decreases as the electron energy 220 increases [28], which is consistent with our observed reduction of the BR for NO2. In con-221 trast, the BRs for CN⁻ and CNO⁻ increase as the electron energy increases, reaching a max-222 imum of 21% and 10%, respectively. The BRs of the other anions in Figure 4 (right panel) 223 exhibit little variation in the energy range of 3.1-4.2 eV, they increase slightly between 4.2 224 and 4.7 eV. While none of these fragments contribute with more than 10%, their combined 225 yield accounts for 23-30% of the total anion yield. 226

The kinetic energy distributions for the five most abundant anions produced in DEA 227 to 1M5NI are shown in Figure 5. The inset in each panel shows a central slice of each 3-228 dimensional momentum image. In all cases the anion momentum is highly isotropic. The 229 anion fragments kinetic energy distributions all peak near 0 eV, with little variation across 230 the present range of incident electron energies. Perhaps most remarkable is the CN⁻ kinetic 231 energy distribution, which is significantly broader than that of NO²⁻ and the heavier frag-232 ments. This suggests a three-body breakup or a stochastic mechanism in the dissociation 233 of CN-, whereby the available energy is broadly distributed into any of the available nu-234 clear degrees of freedom, in addition to the CN⁻ kinetic energy. Clearly, significant motion 235 or rearrangement of the C_3N_3 ring, the NO₂, or the CH₃ moieties are required for the re-236 lease of CN⁻. This contrasts with the dominant NO² dissociation, which may only require 237 stretching of the nitro C-N bond. 238

Figure 3. Time of flight mass spectra showing the relative yields of anion fragments produced in240dissociative electron attachment to 1M5NI at 4 incident energies. The vertical scale is not normalized241between the four electron energies; it shows the number of ions on a logarithmic scale for each242measurement at a fixed electron beam energy. Vertical lines indicate the most prominent anion frag-243ments. The region around 16 a.m.u is subject to contamination from DEA to H2O, O2 and CO2, all of244which produce O' following attachment of electrons on the high energy side of the electron beam245energy distribution.246

Figure 4. The experimental 1M5NI branching ratios of the negative ions formed as a function of248electron energy. Left panel: for ions CN^- (blue), CNO^- (orange) and NO_2^- (yellow). Right panel: for249ions $C_2N_2H_x^-$ (blue), $C_3N_2H_x^-$ (orange), $C_4N_2H_x^-$ (yellow), NO loss (violet), and the sum of OH and250 CH_3 loss (green).251

Figure 5. Measured mass-resolved kinetic energy distributions of the anion fragments, and momen-
tum-sliced images (insets) in atomic units (a.u.) for anion fragments produced in dissociative attach-
ment of 3.1 eV (left column) and 4.2 eV (right column) electrons to 1M5NI. The incident electron is
in the +Py direction in the momentum images.253254255255256

2.3.3. Total ionization cross sections and cation formation by electron impact

Although electron impact ionization is probably the most relevant process in radia-258 tion damage and plasma processing applications, in general, related cross section data are 259 not abundant in the literature, especially for complex polyatomic molecules like 1M5NI, 260 which we are studying here. For the higher energies, the first-Born approximation is com-261 monly used to calculate total ionization cross sections by means of the Born-Bethe formula 262 [29], while assuming an independent atom representation, which is a good approach for 263 energies above 200-300 eV. Calculation procedures considering distorted waves within 264 the second-Born approximation (DWSBA) allow to extend this method to lower energies, 265 which provides information on the single and double differential ionization cross sections. 266 In order to reproduce the maximum cross sections, around 60-90 eV, more elaborated cal-267 culations such as the Binary-Encounter-Bethe (BEB) formulation [30] have demonstrated 268 to give accurate data, within 10%, for a large number of molecular targets. A comparative 269 study including results for different approximations can be found in Ref. [31]. Our screen-270 ing corrected additivity rule (IAM-SCAR), which considers an independent atom repre-271 sentation but corrects for the atomic screening within the molecule and includes relativ-272 istic and velocity dependent effects [32], is also competing with the BEB method. Both 273 methods generally agree within their respective uncertainty limits for energies above 20-274 30 eV. For the lower energies, these approximations are not accurate enough and more 275 reliable data would require employing more sophisticated ab-initio methods such as the 276 R-matrix [33] and the Convergent Close Coupling (CCC) approaches [34]. The limitation 277 of these methods is mainly the size of the target to be treated, being restricted mainly to 278 atoms and small molecules. We have not found any experimental result on the total or 279 partial ionization cross sections in the literature. Itälä et al. [35] measured the photo frag-280 mentation patterns of 1M4NI by soft X-ray synchrotron radiation. Photon decomposition 281 of nitroimidazole compounds has been investigated by Yu and Bernstein [36] and the dis-282 sociation of nitroimidazole ions was studied by Feketová et al. [37]. Since the amount of 283 available data is not enough to build a reliable ionization data set, we followed a proce-284 dure based on our own calculation. We used our IAM-SCAR method to calculate total 285 electron impact ionization cross sections from threshold up to 1000 eV incident energy 286 according to the procedure described in [32]. The corresponding results are given at the 287 end of Section 2 (see Figure 9 and Table 3). 288

Figure 6. Mass analysis of the positive ions formed after the collision of 67 eV electrons with 1M5NI290molecules. (see also the discussion in the text to identify cationic species which are shown in Table2912).292

Cation formation by electron impact can be experimentally determined by analyzing 293 the mass spectrum and intensity of the different positive ions formed. Figure 6 shows the 294 mass spectrum of the cationic species formed by the collision of 67 eV electrons with 295 1M5NI molecules as recorded with the time-of-flight spectrometer described in section 4. 296 From the mass analysis cations can be assigned, within the mass resolution limitation 297 (~2u). As this figure shows, the most intense positive ion formed corresponds to the parent 298 ion ($C_4H_5N_3O_2^+$) with 127 u. Other relevant features are found at 56 u, assigned to the 299 $C_{2}H_{4}N_{2^{+}}$ (Diazoethane) ion; 45 u, to the CH₃NO⁺ (Formaldoxine) ion; 29 u, to the CHO⁺ 300 (Formyl radical) ion and 15 u, assigned to the 301

NH⁺ (Imidogen radical) ion. The reactions producing these molecular fragments need 302 to be investigated in order to evaluate the radiosensitizing properties of 1M5NI and its 303 potential toxicity. Concerning the most representative fragments, diazoethane is a metab-304 olite and its role has not been extensively studied. It has been characterized as highly re-305 active with specific sites of the O and N atoms of relevant biomolecules such as uridine 306 and thymidine [42]. Formaldoxime is highly reactive, since its carbonyl group (-C=O) un-307 dergoes different chemical reactions such as reduction, oxidation and hydrolysis, thus act-308 ing as an efficient antibiotic although is limited by its toxicity. The formyl radical is very 309 active producing H⁻ anions by low energy electron attachment [43] which in turn easily 310 initiates reactions with sensitive biomolecules and imidogen and in combination with H2O 311 initiates the prototypical amidation reaction of O-H bonds [44] 312

Table 2. Identification of the main cationic species formed after the collision of 67 eV electrons with3131M5NI molecules (see also Figure 6).314

Mass (u)	Chemical composition	Relative intensity
127	C4H5N3O2 ⁺ (1M5NI ion)	1

124	C4H2N3O2 ⁺ (1M5NI-3H)	0.415
112	$C_{4}H_{4}N_{2}O_{2^{+}}$ (1M5NI-NH)	0.0564
84	C4HNO (Acrylonitrile-ketone)	0.152
75	C ₂ H ₅ NO ₂ (Glycine)	0.0978
69	C ₂ H ₃ N ₃ (Triazole)	0.137
62	CH4NO (Hydroxymethyl-amino-oxy radical)	0.0900
56	C ₂ H ₄ N ₂ (Diazoethane)	0.571
51	C ₃ HN (Cyanoacetylene)	0.139
48	H2NO2 ⁺ (Nitronium)	0.124
45	CH ₃ NO ⁺ (Formaldoxime)	0.633
42	C ₂ OH ₂ + (Ketene)	0.230
38	C ₂ N (Carbon Cyanide radical)	0.0720
33	H ₃ NO (Hydroxylamine)	0.201
29	CHO (Formyl radical)	0.768
18	H ₂ O ⁺ (Water)	0.0512
15	NH (Imidogen)	0.349
12	C (Carbon)	0.0435

2.3.4. Differential inelastic cross sections

The relevant information to define the track structure of electrons passing through 316 molecular media are the differential inelastic cross sections. This information is provided 317 by the double differential cross sections (DDCS) which give the scattering angle distribu-318 tions as a function of the energy transferred to the target molecule. Direct measurements 319 of DDCS for the whole impact energy range covered in this study, considering all possible 320 transferred energies, would be tedious and are generally not available for any molecular 321 target. Calculations are more accessible but generally require drastic approximations to 322 cover the wide impact and transferred energy ranges required for modelling purposes. 323 One of the most commonly used approximations is based on the Distorted Wave Second 324 Born Approximation (DWSBA) [38], which has been extensively used for representative 325 molecules as water both in the gas [39] and condensed [40] phases. However, to our 326 knowledge no approximate calculations of the DDCS of complex molecules such as 327 1M5NI have been published. We thus followed the semiempirical procedure proposed for 328 other molecular targets, such as pyridine [41]. From measurements of the angular distri-329 butions of scattered electrons performed for different incident energies (E) we proposed 330 the following formula which depends on the energy transferred to the medium (ΔE) and 331 the corresponding differential elastic cross sections: 332

$$\frac{d^2\sigma(E)}{d\Omega\Delta E} \propto \left(\frac{d\sigma(E)}{d\Omega}\right)_{el}^{1-\frac{KEE}{E}},$$
(1) 333

where $\frac{d\sigma(E)}{d\Omega}$ represents the differential elastic cross section and k is a parameter to mod-334 ify the weight of the dependence on $\Delta E/E$. The value of k should be derived for each 335 considered target molecule from direct measurements of the DDCS at representative elec-336 tron impact and transferred energies. For 1M5NI we have measured the angular distribu-337 tion of scattered electrons corresponding to the generation of the parent ion and the three 338 more intense peaks of the fragmentation mass spectrum (see Figure 6). In all cases the 339 incident and the transferred energies have been found to be 67 and 30 eV, respectively. 340 The results are shown in Figure 7. As can be seen in this figure, the observed angular 341 distribution of the scattered electrons does not depend on the produced cation and pre-342 sents a good agreement with that derived from Eq. (1) for *k*=1. 343

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Figure 7. Angular distribution of scattered electrons after the collision of 67 eV electrons with 1M5NI345transferring 30 eV to the target to produce cationic species (see also figure legend).346

2.3.5. Rotational, vibrational and electronic excitation cross sections

1M5NI is a polar molecule, with a strong permanent dipole moment of 4.4 D and 348 therefore dipole rotational excitations by electron impact are relevant processes. However, 349 the average rotational excitation transferred to the molecule at 300K temperature is 0.617 350 meV which is too low to be resolved by any of the experiments considered in this study. 351 From the theoretical point of view, the calculation methods used in this study assume that 352 the nuclei are fixed during the collision. Hence neither rotational nor vibrational excitation 353 processes could be calculated. For the lower energies, where we here considered the SMC 354 calculation for the elastic scattering to be accurate, an indirect way to account for dipole 355 interactions is to include the dipole moment in the scattering potential and apply the so-356 called Born correction [45] to the higher order partial waves. This implies to introduce a 357 minimum transferred energy to avoid the 0-angle singularity of the Born approximation. 358 Cross section values obtained with this procedure are commonly termed "rotationally 359 summed elastic cross sections" to distinguish from the "pure" elastic cross sections that 360 are calculated without the Born correction (see [46] and references therein for details). This 361 is very important when comparing theoretical and experimental elastic cross sections 362 (note that, as aforementioned, most of the experimental elastic cross sections do not 363 properly resolve rotational excitations). 364

Rotational excitation cross sections can also be calculated independently by assuming 365 the molecule as a rigid rotor and applying the first-Born approximation (FBA). Although 366 the incident energy could be very low, down to 0.1 eV in this case, the average rotational 367 excitation mentioned above is at least three orders of magnitude lower and therefore may 368 validate the FBA approach. Details on this calculation method are given in the case of 369 pyridine by Sieradzka et al. [47]. Differential and integral cross sections calculated with 370 this procedure are shown in Figure 8 (a) and (b), respectively. Since they have been de-371 rived from a crude approximation, we recommend to consider them just as a qualitative 372 indication of the expected energy dependence, yet using their absolute values would re-373 quire additional verifications (numerical results of the full calculation are given in Table 374 S2 of the Supplementary Information). 375

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Figure 8. Rotational excitation cross sections of 1M5NI calculated with the first-Born approximation.379(a), differential cross sections; (b), integral cross sections.380

According to the self-consistent procedure we are following here to obtain the rec-381 ommended integral cross sections, the total inelastic cross section is derived by subtract-382 ing the pure integral elastic shown in Figure 9 from the reference TCS+MA values of Table 383 1. The electron attachment cross sections are derived from the resonance analysis de-384 scribed above. After subtracting the total ionization cross sections from the remaining in-385 elastic channels, we separated the electronic and vibrational excitation cross section by 386 sampling their magnitudes from their respective thresholds to get the sum of their contri-387 butions at any energy, in order to be consistent with the reference TCS values at that en-388 ergy. Finally, as already mentioned, the rotational excitation cross sections are estimated 389 through an independent FBA calculation and are not included in the reference TCS values, 390 i.e. they are not considered in this self-consistency procedure. The recommended cross 391 section data obtained with this method are shown in Table 3 and plotted in Figure 9 (the 392 rotational excitations shown in Fig. 8 are not considered in this figure). As shown in Table 393 3 the sum of the recommended integral cross sections is in good agreement with the ref-394 erence TCS data discussed in Subsection 2.1. 395

Figure 9. Recommended integral cross sections (CS). —, elastic CS; —, ionization CS; —, electronic 397 excitation CS; ----, sum of all the considered 398 scattering channels of 1M5NI (excluding the rotational excitation, see text for details); • , TCS+MA 399 reference data (experimental values corrected for elastic missing angles, see text). 400

Table 3. Recommended integral cross sections for electron scattering cross sections from 1M5NI, in units of 10⁻²⁰ m², derived from the proposed self-consistent procedure (see text for details). .

Energy (aV)	Elastic	Electron	Vibrational	Electronic	Ioniza-	SUM
$\frac{(ev)}{0.1}$	390	attachillent	excitation	excitation	tion	390
0.2	214					214
0.3	162					162
0.4	138					138
0.5	122					122
0.7	104		0.1			104
1.0	86.8		3.28			90.1
1.5	72.2	2.39	11.9			86.5
2.0	63.3	14.5	20.3			98.2
3.0	55.0	4.12	20.0			79.1
4.0	52.9	0.425	17.3	0.1		70.7
5.0	52.5	1.74	13.9	0.257		68.5
7.0	54.4	0.136	14.2	1.07		69.8
10	65.8	0.1		10.1		76.0
15	60.5	0.1		13.7	0.328	74.5
20	54.9			15.2	3.11	73.2
30	46.8			13.8	9.52	70.1
40	42.3			11.5	12.6	66.4
50	38.6			9.29	13.7	61.6
70	33.6			11.5	12.5	66.4
100	29.1			4.78	13.9	47.8

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450	24.3	3.83	12.6	40.7
200	21.2	3.38	11.3	35.9
300	17.1	2.62	9.41	29.2
400	14.6	2.27	8.04	24.89
500	12.7	2.04	7.06	21.8
700	10.2	1.74	5.68	17.7
1000	7.95	1.43	4.42	13.8

3. Discussion

The results presented in Section 2 constitute a complete dataset of self-consistent scat-406 tering cross sections ready to be used for modelling electron transport in biologically rel-407 evant media, such as water containing traces of 1M5NI. This is a common situation in 408 many radiobiological experiments [48] that are devoted to quantifying the living cell (as-409 sumed to be water) survival fractions with and without additional radiosensitizers. This 410 data set is self-consistent in the sense that the sum of the recommended integral cross 411 section of all the considered scattering channels (elastic, ionization, electronic excitation, 412 vibrational excitation and electron at-attachment) at a given energy coincides with the 413 reference value at that energy. These reference data are derived from our accurate (within 414 5%), previously measured, total electron scattering cross sections together with the pre-415 sent calculation to correct the elastic "missing angle" effect (see above) and complemented 416 with our IAM-SCARI calculations up to 1000 eV impact energy. As 1M5NI is a polar mol-417 ecule, additional electron scattering rotational excitation cross sections have been calcu-418 lated by means of the Born approximation. Although this is a simple approximation it 419 gives an idea of the energy dependence of rotational excitation processes and can be used 420 in applications in which heating processes may be relevant. For any of the scattering pro-421 cesses considered here, there are no previous theoretical or experimental cross section 422 data available in the literature. However, considering previous studies in which we ap-423 plied a similar method to that proposed here to other molecules such as water [49], pyri-424 dine [41] or benzene [50], we can estimate a 10 % uncertainty limit to the most relevant 425 scattering channels (elastic, ionization and electronic excitation) and up to 20-25% to the 426 remaining channels (vibrational excitation and electron attachment). For electron impact 427 rotational excitations, we only provide qualitative information. 428

Since damaging effects of molecular radiosensitizers are connected to the production 429 radicals by electron induced molecular dissociation, we have studied the generation of 430 anionic fragments by low energy electrons and the formation of cationic fragments by 431 relatively high energy electrons. The observed anionic and cationic fragmentation pat-432 terns agree with the prediction of previous studies [13, 14, 17, 28, 35, 37] and confirm the 433 potential activity of 1M5NI as a radiosensitizer. Experiments like that in Ref. [48] should 434 be repeated by adding 1M5NI to the living cell target in order to confirm and quantify its 435 radiosensitizing effect. 436

4. Materials and Methods

4.1. Calculation methods

For the lower electron impact energies (<15 eV), the elastic differential and integral 439 cross sections were computed with the SMC method [22]. The calculations relied on the 440 fixed-nuclei approximation, where the geometry was obtained with density functional 441 theory, using the B3LYP functional and the aug-cc-pVDZ basis set. The restricted Hartree-442 Fock approximation was employed to describe the electronic ground state, with the Car-443 tesian Gaussian functions given in Ref. [23]. As mentioned before, electronic excitation 444 channels were not considered in the scattering calculations. Besides the usual spin-445 adapted configurations associated with the static-exchange approximation, we further ac-446 counted for polarization effects by introducing a set of spin-adapted configurations built 447

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from single excitations. For that, we used the orbital energy criterion introduced in Ref. 448 [51], with an energy cutoff of ε_{cut} = 1.43 Hartree. Also, the canonical virtual orbitals were 449 replaced by modified virtual orbitals generated in the field of the cation of charge +8. To 450 avoid possible numerical problems, the singular value decomposition technique was 451 made use of. We have removed the combination of configurations associated with the 452 three lowest singular values for the A' symmetry and with the lowest singular value for 453 the A" symmetry. The long-range electron-molecule dipolar interaction was accounted 454 for in the calculations via the Born-closure procedure [53]. Additional details about the 455 calculations can be found in а previous publication 456 [https://pubs.acs.org/doi/10.1021/acs.jpclett.2c01704]. 457

For the higher collision energies (>15 eV), we employed the IAM-SCAR+I method to 458 obtain differential and integral elastic cross sections as well as integral inelastic electronic 459 excitation and ionization cross sections. This method has been described in detail [19] and 460 its reliability has been thoroughly verified [58-60]. Briefly, the molecular target is consid-461 ered as an aggregate of its individual atoms. Each atom is represented by an "ab-initio" 462 optical potential, where the real part accounts for elastic scattering, while the imaginary 463 part represents the inelastic processes, considered as the "absorption part". The differen-464 tial scattering cross sections (DCSs) are obtained from the atomic data by the screening 465 corrected additivity rule (SCAR) procedure, incorporating interference (I) corrections by 466 summing all the atomic amplitudes, where the phase coefficients are included. Then, by 467 integrating over all the scattered angular range, the integral scattering cross sections 468 (ICSs) are obtained. Finally, the rotational excitation cross sections stem from the first-469 Born approximation. 470

4.2. Experimental methods

The total electron scattering cross sections, which have been used as reference values 472 to evaluate the self-consistency of the present data set, were taken from our recent meas-473 urements [17] that were performed with a magnetically confined electron transmission 474 apparatus [18]. Details on this experimental setup can be found in Ref. [18]. 475

An anion fragment momentum imaging spectrometer was employed at Lawrence Berkeley National Laboratory to analyze the relative yields and kinetic energies of anion 477 fragments produced by dissociative electron attachment (DEA) to 1M5NI. The experi-478 mental arrangement to perform these measurements has been previously described in de-479 tail [27], thus we include here only the information most relevant for the present experi-480 ments. Briefly, a stainless-steel capillary was employed to produce an effusive jet of mol-481 ecules, which crosses orthogonally with a pulsed electron beam in a coaxial magnetic field 482 inside the spectrometer. At one end of a gas manifold system, glass sample holder con-483 taining approximately 10 g of 1M5NI was heated to a temperature range of 40-50° C. The 484 gas manifold feeding the gas jet capillary was also gradually heated by increasing the 485 temperature (< 80°C). This process caused the sublimated 1M5NI vapor to accumulate in 486 the gas manifold and build up a pressure of 10-100 mTorr, before the gas was introduced 487 into the sample gas inlet and capillary. The electron beam energy spread (0.5 eV full width 488 at half maximum) and absolute electron beam energy was determined and checked before 489 and after the present experiments by measuring the anion yields across the thermody-490 namic threshold for O⁻ production from CO₂. The anion fragment momentum was cali-491 brated against the well-known O⁻ momentum distribution from DEA to O₂. The time-of-492 flight and positions of each ion hit were recorded by a time- and position-sensitive multi-493 channel plate detector with delay-line readout in an event list-mode format. The raw po-494 sition and time data were sorted and converted to momenta after each experiment was 495 completed. 496

The induced cationic fragmentation and the double differential ionization cross sec-497 tions have been measured with a reaction microscope (ReMi) multi-particle coincidence 498 spectrometer [62, 63]. The schematic diagram of the apparatus is shown in Figure 10. 499

Figure 10. Schematic view of the reaction microscope (ReMi) multi-particle coincidence spectrome-
ter used to measure the induced cationic fragmentation and the double differential ionization cross
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A pulsed electron beam is used as projectile beam. It is produced with a photoemis-504 sion source (electron gun) containing a tantalum photocathode which was illuminated 505 with UV light (266 nm) of 0.5 ns pulse duration from a diode laser having 40 kHz repeti-506 tion rate. The electron gun is mounted within the drift tube of the spectrometer. The pro-507 duced electron beam propagates along the spectrometer axis (Z axis) and is dumped into 508 a hole of 8 mm radius on the electron detector. Electrostatic lenses in the electron gun 509 assembly and an axial magnetic guiding field focus the beam to a diameter of ~ 0.5 mm at 510 the interaction region. The 1-Methyl-5-Nitroimidazole (1Me5NI) target is prepared using 511 a syringe setup as shown in Figure 11. This newly built gas target is required since the 512 standard supersonic gas target cannot be employed due to the sample's particularly low 513 vapor pressure of 0.003 mbar at room temperature. 1Me5NI powder is filled inside a sy-514 ringe which is fixed on a xyz-manipulator. The target sample undergoes sublimation and 515 the vapor is introduced to the interaction region within the spectrometer by moving the 516 setup along the X-direction in the lab frame, such that the tip of the needle of the syringe 517 is close to but not hitting the electron beam. To increase the target density, the sample was 518 slightly heated up to 56°C. For guiding of the outgoing electrons to the electron detector 519 homogenous electric and magnetic fields of 4 V/cm and 7.4 G, respectively, were used. 520 For this purpose, the syringe needle is kept on the proper spectrometer potential. After 521 the electrons have reached the detector the electric spectrometer field is ramped up to 22.6 522 V/cm for extraction of the ions. In order to prevent the low momentum ions (e.g. parent 523 ions) from colliding with the electron gun assembly a pulsed voltage higher than the volt-524 age of the spectrometer ring at the interaction region was applied to the syringe needle. 525 In this way the low momentum ionic fragments are pushed transversally out of the inter-526 action region and subsequently by-pass the electron gun and reach the ion detector. Both 527 the electron and the ion detectors are time and position sensitive microchannel plate de-528 tectors with hexagonal (multi-hit) delay line read out. 529

Figure 11. a) The target vapor source consists of a standard medical syringe with a 1 mm inner531diameter, 10 cm long stainless-steel needle. b) View into the electrode array of the reaction micro-532scope with the syringe needle introduced horizontally through a hole in a spectrometer ring elec-533trode. In the back the ion detector is visible while the spectrometer drift tube and the electron de-534tector on the front side are not shown. The vertically aligned helium supersonic gas jet was used in535the present measurements for calibration of the spectrometer fields.536

From the time and position information, the momenta and, therefore, the kinetic en-537 ergies and the emission angles of the detected electrons are reconstructed. The acceptance 538 angle is close to 4π for electrons with energies between 0.3 eV and 18 eV. Electrons moving 539 in small forward or backward angles and those with energies below 0.3 eV are not de-540 tected due to the presence of the central hole in the electron detector. The solid angle cov-541 erage for the intact parent ions is almost 100 deg. However, the initial momentum infor-542 mation of the ionic fragments is lost due the inhomogeneous electric field used to push 543 the fragments away from the center. Therefore, only the time-of-flight information of the 544 ions is used to differentiate different fragmentation channels by the ionic fragment mass. 545

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