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## Methyl iodide: Atmospheric budget and use as a tracer of marine convection in global models

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[1] We simulate the oceanic and atmospheric distribution of methyl iodide ( $\text{CH}_3\text{I}$ ) with a global 3-D model driven by assimilated meteorological observations from the Goddard Earth Observing System of the NASA Data Assimilation Office and coupled to an oceanic mixed layer model. A global compilation of atmospheric and oceanic observations is used to constrain and evaluate the simulation. Seawater  $\text{CH}_3\text{I}(\text{aq})$  in the model is produced photochemically from dissolved organic carbon, and is removed by reaction with  $\text{Cl}^-$  and emission to the atmosphere. The net oceanic emission to the atmosphere is  $214 \text{ Gg yr}^{-1}$ . Small terrestrial emissions from rice paddies, wetlands, and biomass burning are also included in the model. The model captures 40% of the variance in the observed seawater  $\text{CH}_3\text{I}(\text{aq})$  concentrations. Simulated concentrations at midlatitudes in summer are too high, perhaps because of a missing biological sink of  $\text{CH}_3\text{I}(\text{aq})$ . We define a marine convection index (MCI) as the ratio of upper tropospheric (8–12 km) to lower tropospheric (0–2.5 km)  $\text{CH}_3\text{I}$  concentrations averaged over coherent oceanic regions. The MCI in the observations ranges from 0.11 over strongly subsiding regions (southeastern subtropical Pacific) to 0.40 over strongly upwelling regions (western equatorial Pacific). The model reproduces the observed MCI with no significant global bias (offset of only +11%) but accounts for only 15% of its spatial and seasonal variance. The MCI can be used to test marine convection in global models, complementing the use of radon-222 as a test of continental convection. *INDEX TERMS:* 0312 Atmospheric Composition and Structure: Air/sea constituent fluxes (3339, 4504); 0322 Atmospheric Composition and Structure: Constituent sources and sinks; 0368 Atmospheric Composition and Structure: Troposphere—constituent transport and chemistry; *KEYWORDS:* methyl iodide, marine convection, atmospheric tracer, global budget of methyl iodide

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### 1. Introduction

[2] Methyl iodide ( $\text{CH}_3\text{I}$ ) is emitted to the atmosphere by the oceans and photolyzes with a lifetime of the order of a week. Aircraft measurements over the past decade have provided large data sets of  $\text{CH}_3\text{I}$  concentrations over the oceans up to 12 km altitude [Blake *et al.*, 1999, and

references therein]. These observations have been used to constrain vertical mixing rates in 1-D models of atmospheric transport [Davis *et al.*, 1996; Wang *et al.*, 2000, 2002] and to diagnose outflow from marine convection in the upper troposphere [Cohan *et al.*, 1999; Staudt *et al.*, 2002]. Beyond its value as a tracer of transport,  $\text{CH}_3\text{I}$  is also of interest in the upper troposphere and lower stratosphere as a source of iodine radicals for ozone destruction [Solomon *et al.*, 1994; Davis *et al.*, 1996]. In addition, methyl iodide is a leading candidate to replace methyl bromide as a soil fumigant [Waggoner *et al.*, 2000].

[3] We present here a global 3-D simulation of  $\text{CH}_3\text{I}$  in the ocean-atmosphere coupled system and use a large ensemble of worldwide atmospheric and oceanic  $\text{CH}_3\text{I}$  observations to evaluate model results. Our first objective is to develop  $\text{CH}_3\text{I}$  as a tracer for testing marine convection in global 3-D atmospheric models, complementing the standard use of  $^{222}\text{Rn}$  as a tracer for continental convection [Jacob *et al.*, 1997]. Our second objective is to quantify better the global atmospheric budget of  $\text{CH}_3\text{I}$ .

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**Table 1.** Global CH<sub>3</sub>I Emission Estimates in the Literature

Source	Global Emission, Gg/yr	Reference
Oceanic <sup>a</sup>	270	<i>Liss and Slater</i> [1974]
	1300	<i>Rasmussen et al.</i> [1982]
	300–500	<i>Singh et al.</i> [1983]
	140	<i>Nightingale</i> [1991]
	800	<i>Reifenhauser and Heumann</i> [1992]
	150	<i>Campos et al.</i> [1996]
	130–350	<i>Moore and Groszko</i> [1999]
Terrestrial		
Biomass burning	<10	<i>Andreae et al.</i> [1996]
	3.4	<i>Blake et al.</i> [1996b]
Plant-soil systems		
Rice paddies	20	<i>Muramatsu and Yoshida</i> [1995]
	71	<i>Redeker et al.</i> [2000]
Peatland ecosystems	1.4	<i>Dimmer et al.</i> [2000]
Wetlands	7.3	<i>Dimmer et al.</i> [2000]

<sup>a</sup> Extrapolated from field measurements over limited regions.

[4] A simple way to implement CH<sub>3</sub>I as a tracer of vertical transport in atmospheric models is to use observed marine boundary layer (MBL) concentrations as a lower boundary condition and simulate the vertical gradient in the tropospheric column. The 1-D models cited above used this approach to constrain their mean vertical transport rates. The approach is less attractive in a time-dependent 3-D model where the MBL concentration should vary in response to the MBL ventilation rate. A boundary condition in the ocean is preferable. We are led therefore to an examination of the oceanic CH<sub>3</sub>I budget.

[5] Data for oceanic concentrations of CH<sub>3</sub>I are available from a number of ship cruises. Extrapolations from individual regional data sets have led to global emission estimates in the literature ranging from 130 to 1300 Gg yr<sup>-1</sup> (Table 1). Consideration of the global data base of CH<sub>3</sub>I(aq) concentration measurements allows a more constrained estimate, as described in this paper. Biological sources from algae and phytoplankton have been examined in a number of laboratory studies and could contribute to elevated CH<sub>3</sub>I in coastal environments, but they appear to be negligible on a global scale when compared to the oceanic emission estimates in Table 1 [Manley and Dastoor, 1987, 1988; Nightingale et al., 1995; Manley and de la Cuesta, 1997; Giese et al., 1999]. A laboratory study by Moore and Zafiriou [1994] indicates that photochemical degradation of dissolved organic carbon (DOC) could possibly provide the main oceanic source of CH<sub>3</sub>I(aq). Field measurements by Happell and Wallace [1996] also suggest that the source of CH<sub>3</sub>I(aq) is photochemical. The simulation of CH<sub>3</sub>I(aq) in our global model is more consistent with a photochemical source than with a biological source, as described below.

[6] The main sinks of CH<sub>3</sub>I(aq) in the oceans are thought to be emission to the atmosphere and nucleophilic S<sub>N</sub>2 reaction with Cl<sup>-</sup>. These two sinks are of comparable magnitude [Zafiriou, 1975]; either may dominate depending on surface wind speed (which affects emission to the atmosphere) and seawater temperature (which affects the rate of reaction with Cl<sup>-</sup>). Hydrolysis is typically an order of magnitude slower than reaction with Cl<sup>-</sup> [Moelwyn-Hughes, 1938]. Photolysis is unimportant [Zika et al., 1984].

[7] Little is known regarding continental sources of CH<sub>3</sub>I (Table 1). There is a source from rice paddies [Muramatsu and Yoshida, 1995] which Redeker et al. [2000] have estimated to be as large as 71 Gg yr<sup>-1</sup> globally. There is a small CH<sub>3</sub>I source from biomass burning of 3–10 Gg yr<sup>-1</sup> [Andreae et al., 1996; Blake et al., 1996b]. Dimmer et al. [2000] have presented evidence for a small source from peatland ecosystems. The continental source of CH<sub>3</sub>I may be much less than the oceanic source on a global scale, but measurements of atmospheric concentrations of CH<sub>3</sub>I over the continents are too few at present to be illuminating.

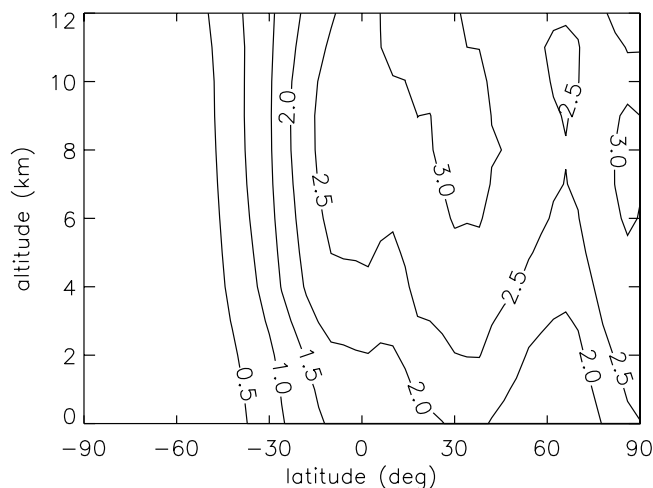
## 2. Model Description

### 2.1. General

[8] We use the GEOS-CHEM global 3-D model of tropospheric chemistry [Bey et al., 2001a; Liu et al., 2001] driven by assimilated meteorological observations from the Goddard Earth Observing System (GEOS) of the NASA Data Assimilation Office (DAO). Allen et al. [1996a] and Liu et al. [2001] previously used simulations of <sup>222</sup>Rn with the GEOS fields to test the representation of continental convection. The GEOS-CHEM model has been applied to a range of tropospheric chemistry problems [Bey et al., 2001a, 2001b; Fiore et al., 2002; Jacob et al., 2002; Li et al., 2000, 2001, 2002; Liu et al., 2001, 2002; Martin et al., 2002a, 2002b; Palmer et al., 2001].

[9] In the present application we use GEOS fields for 1994 available with 2° latitude × 2.5° longitude horizontal resolution and 20 sigma levels in the vertical extending up to 10 hPa. We degrade the horizontal resolution to 4° × 5° for computational expediency. The lowest 3 km of the atmosphere are resolved by 6 levels centered at 0.15, 0.35, 0.71, 1.2, 1.9, and 2.6 km above the surface for an air column based at sea level. Advection is computed every 30 min with a flux-form semi-Lagrangian method [Lin and Rood, 1996]. Moist convection is computed using the GEOS convective, entrainment, and detrainment mass fluxes as described by Allen et al. [1996a, 1996b]. We assume full vertical mixing within the GEOS-diagnosed atmospheric mixed layer generated by surface instability. The simulation is conducted for 18 months (July 1993 through December 1994) starting from low CH<sub>3</sub>I concentrations as initial conditions. The first 6 months are used to achieve initialization and we focus our attention on the 12-month simulation for 1994. Methyl iodide emitted from different sources (ocean, biomass burning, wood fuel, rice paddies, and wetlands), is carried as different tracers in the model.

[10] The atmospheric CH<sub>3</sub>I sink from photolysis is calculated in GEOS-CHEM using the FAST-J algorithm of Wild et al. [2000], implemented as described by Bey et al. [2001a]. Surface albedos and vertically resolved cloud optical depths are taken from the GEOS data with 6-hour resolution. CH<sub>3</sub>I absorbs over the wavelength range 300–340 nm and we use absorption cross sections from Roehl et al. [1997]. We assume unit quantum yield as has been observed for CH<sub>2</sub>ClI and CH<sub>2</sub>I<sub>2</sub> in that wavelength range [Roehl et al., 1997, and references therein]. The resulting lifetime of CH<sub>3</sub>I is about 4 days in the tropical troposphere (Figure 1). Oxidation of CH<sub>3</sub>I by OH is a negligibly small



**Figure 1.** Altitude-latitude plot of the mean 24-hour  $\text{CH}_3\text{I}$  photolysis frequency ( $10^{-6} \text{ s}^{-1}$ ) calculated in the GEOS-CHEM model for July 1994.

sink, amounting to only about 1% of photolysis [Brown *et al.*, 1990].

## 2.2. Oceanic Emission of Methyl Iodide

[11] Oceanic emission of  $\text{CH}_3\text{I}$  to the atmosphere is computed following Liss and Slater [1974],

$$\Phi = k([\text{CH}_3\text{I}(\text{aq})] - K_H[\text{CH}_3\text{I}(\text{g})]). \quad (1)$$

Here  $\Phi$  is the net flux of  $\text{CH}_3\text{I}$  from the ocean to the atmosphere representing a balance between emission from the ocean ( $\Phi_{em} = k[\text{CH}_3\text{I}(\text{aq})]$ ) and deposition from the atmosphere ( $\Phi_{dep} = -kK_H[\text{CH}_3\text{I}(\text{g})]$ ) and  $k$  is the sea-to-air transfer velocity.  $K_H$  is the dimensionless Henry's law constant for  $\text{CH}_3\text{I}$  defined as the ratio of aqueous to gas-phase concentrations at equilibrium ( $K_H = 3.4$  at 298 K and 1 atm,  $\Delta H_{298}^0/R = -4300 \text{ K}$  (available at <http://www.mpch-mainz.mpg.de/sander/res/henry.html>)). We parameterize  $k$  as a function of the surface wind speed following Nightingale *et al.* [2000], who found from tracer experiments an optimal wind speed dependence intermediate between the previous results of Liss and Merlivat [1986] and Wanninkhof [1992].

[12] Using a box model for the oceanic mixed layer of depth  $z$ , and assuming no horizontal flux divergence, we write a steady state expression for the local concentration of  $\text{CH}_3\text{I}(\text{aq})$ ,

$$[\text{CH}_3\text{I}(\text{aq})] = \frac{P + (k/z)K_H[\text{CH}_3\text{I}(\text{g})]}{(k/z) + k_{\text{Cl}}[\text{Cl}^-]}, \quad (2)$$

where  $P$  is the oceanic production rate per unit volume ( $\text{ng L}^{-1} \text{ h}^{-1}$ ), and  $k_{\text{Cl}}$  is the rate constant for reaction with  $\text{Cl}^-$ . We use  $k_{\text{Cl}} = 7.78 \times 10^{13} \exp[-13518/T] \text{ M}^{-1} \text{ s}^{-1}$  from Elliott and Rowland [1993], who provide the only temperature-dependent data for this reaction. The temperature dependence is very strong; an increase in ocean mixed layer temperature (MLT) of 20 K as observed between polar latitudes and the tropics corresponds to an approximately twenty-fold increase in  $k_{\text{Cl}}$ .

[13] In our simulation we use monthly mean MLT fields from the data set of Woodruff *et al.* [1987] and a uniform  $[\text{Cl}^-] = 0.54 \text{ M}$ . For typical values  $k = 10 \text{ cm h}^{-1}$ ,  $z = 50 \text{ m}$ , and  $k_{\text{Cl}} = 1.2 \times 10^{-3} \text{ h}^{-1}$  (where  $\text{MLT} = 292 \text{ K}$ ), the lifetimes of  $\text{CH}_3\text{I}(\text{aq})$  against transfer to the atmosphere and reaction with  $\text{Cl}^-$  are 21 and 35 days, respectively, for an overall lifetime of 13 days. Hence ventilation and chemical reaction with  $\text{Cl}^-$  make comparable contributions to the  $\text{CH}_3\text{I}(\text{aq})$  sink depending on local wind speed and MLT [Zafiriou, 1975].

[14] For our model, we specify the atmospheric concentration  $[\text{CH}_3\text{I}(\text{g})]$  in equations (1) and (2) with the local value from the previous time step of the GEOS-CHEM simulation, so that the oceanic and atmospheric simulations are fully coupled. The  $\text{CH}_3\text{I}(\text{aq})$  production rate  $P$  is the effective lower boundary condition of the model. Following the laboratory experiments of Moore and Zafiriou [1994] and the supporting observational evidence of Hoppel and Wallace [1996], we assume that the production of  $\text{CH}_3\text{I}(\text{aq})$  in the open oceans is driven by a photochemical mechanism dependent on the supply of dissolved organic carbon (DOC). As a crude parameterization, we scale the production rate as the product of the solar radiation flux at the surface (RAD) and the dissolved organic carbon concentration (DOC),

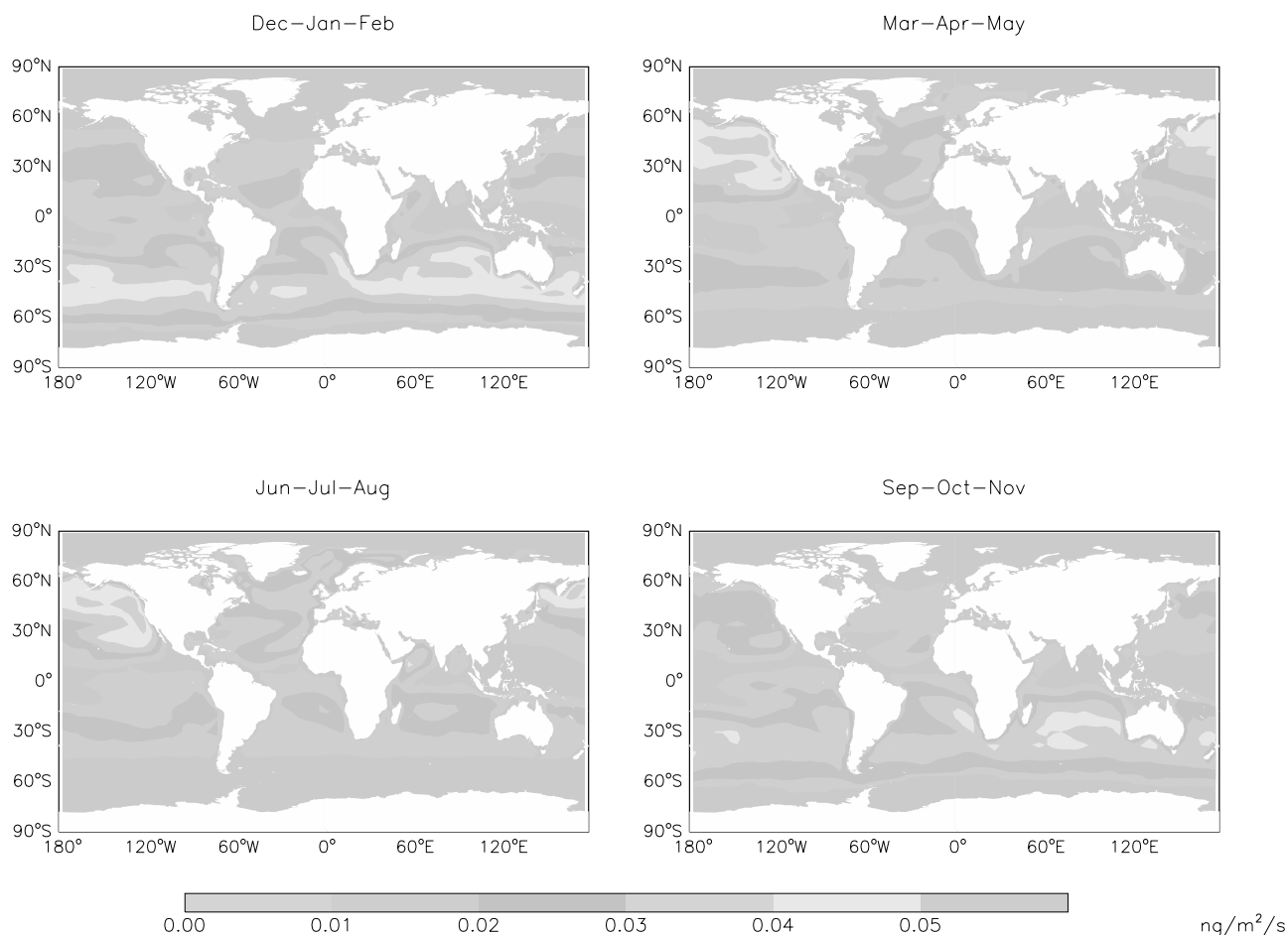
$$P = \alpha[\text{RAD} \times \text{DOC}], \quad (3)$$

where  $\alpha$  is a scaling parameter. Monthly average values of RAD are taken from the GEOS fields for 1994, and monthly average fields of DOC are taken from the oceanic GCM of Six and Maier-Reimer [1996]. A value  $\alpha = 0.1 \text{ m}^2 \text{ W}^{-1} \text{ h}^{-1}$  was obtained by least-squares fit of model results to observed seawater  $\text{CH}_3\text{I}(\text{aq})$  concentrations, as described in section 3. The resulting  $r^2$  is 0.4. A sensitivity simulation where  $P$  was scaled to net primary productivity (NPP) to simulate a biological source of  $\text{CH}_3\text{I}(\text{aq})$  was found to have no success in reproducing the observed distributions of  $\text{CH}_3\text{I}(\text{aq})$ .

[15] The oceanic emission of  $\text{CH}_3\text{I}$  as computed by the model is thus determined by MLT, RAD, DOC, and wind speed (Figure 2). The model predicts maximum emission at midlatitudes in the spring-summer hemisphere where solar radiation is high and MLT is relatively low. DOC is less variable than RAD in space and time but shows subtropical maxima that contribute to the high  $\text{CH}_3\text{I}$  fluxes there. The annual net emission flux to the atmosphere in the model is  $214 \text{ Gg yr}^{-1}$  (Table 2), comparable to the median ( $270 \text{ Gg yr}^{-1}$ ) of previous estimates (Table 1) but at the low end of the range ( $130\text{--}1300 \text{ Gg yr}^{-1}$ ).

## 2.3. Terrestrial Emission of Methyl Iodide

[16] We include emission of  $\text{CH}_3\text{I}$  from biomass burning and wood fuel by applying a  $\text{CH}_3\text{I}/\text{CO}$  emission ratio of  $4.0 \times 10^{-6} \text{ vol/vol}$  [Ferek *et al.*, 1998] to a CO emission inventory with  $1^\circ \times 1^\circ$  spatial resolution and monthly temporal resolution (J.A. Logan, personal communication, 2000). The recent review of Andreae and Merlet [2001] gives  $\text{CH}_3\text{I}/\text{CO}$  emission ratios of  $1.5 \times 10^{-6} \text{ vol/vol}$ ,  $1.3 \times 10^{-5} \text{ vol/vol}$ , and  $1.1 \times 10^{-6} \text{ vol/vol}$  for savanna, tropical forest, and extratropical forest, respectively.



**Figure 2.** Mean fluxes of  $\text{CH}_3\text{I}$  from the ocean to the atmosphere for each season computed with the coupled ocean-atmosphere GEOS-CHEM model.

[17] Global  $\text{CH}_3\text{I}$  emissions from rice paddies and natural wetlands are taken to be  $71 \text{ Gg yr}^{-1}$  and  $7.3 \text{ Gg yr}^{-1}$ , respectively, following the estimates of Redeker *et al.* [2000] and Dimmer *et al.* [2000]. These emissions are distributed spatially following the corresponding patterns of  $\text{CH}_4$  emission from the inventory of Fung *et al.* [1991]. Peatland ecosystems have been estimated to provide an additional  $1.4 \text{ Gg yr}^{-1}$  source of  $\text{CH}_3\text{I}$  [Dimmer *et al.*, 2000] but this is neglected in the model. Muramatsu and Yoshida [1995] estimated a global source from rice paddies of  $20 \text{ Gg yr}^{-1}$ , much less than the estimate of Redeker *et al.* [2000]. As we will see in section 4.1, it appears that the Redeker *et al.* [2000] estimate may be too high.

[18] Table 3 shows the annual emission flux of  $\text{CH}_3\text{I}$  to the atmosphere in the model, where each of the five sources

**Table 2.** Global Oceanic Mixed Layer Budget of  $\text{CH}_3\text{I}(\text{aq})$  in the Model

	Rate, $\text{Gg yr}^{-1}$
<i>Sources</i>	
Photochemical production	477
Atmospheric deposition	16
<i>Sinks</i>	
Emission to atmosphere	230
Reaction with $\text{Cl}^-$	263

is represented by a separate tracer. The ocean accounts for 70% of the global source. Emissions from rice paddies provide a sizeable fraction (24%) of the global budget of  $\text{CH}_3\text{I}$  and are highly localized in southeast Asia.

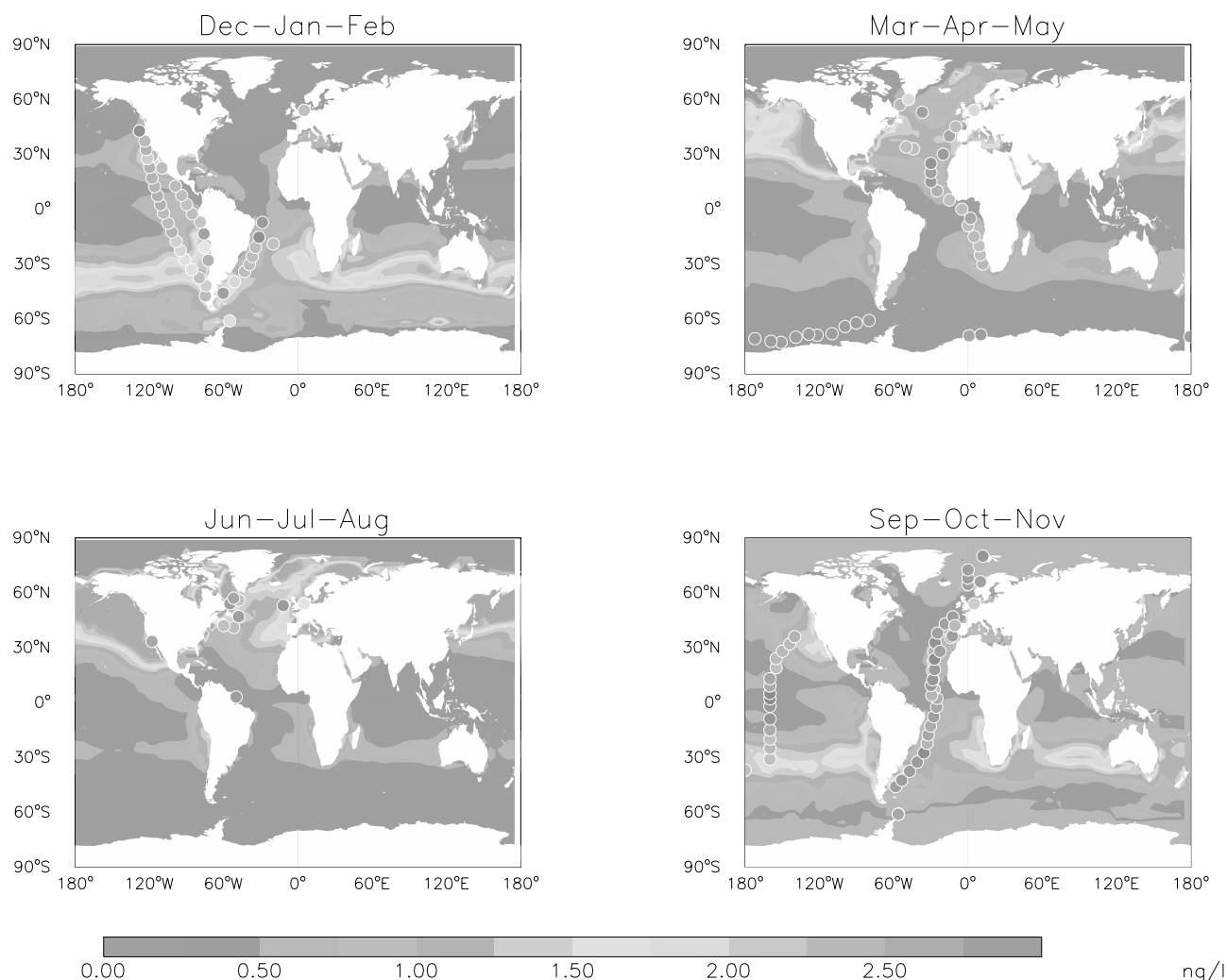
### 3. Methyl Iodide Distribution in the Ocean

[19] Concentrations of  $\text{CH}_3\text{I}(\text{aq})$  in seawater have been measured on a number of ship cruises. Figure 3 shows these

**Table 3.** Global Atmospheric Budget of  $\text{CH}_3\text{I}$  in the Model

	Values
<i>Sources, <math>\text{Gg yr}^{-1}</math></i>	
Ocean <sup>a</sup>	214
Rice paddies	71
Wetlands	7
Biomass burning	9
Wood fuel	3
Total source	304
<i>Sink, <math>\text{Gg yr}^{-1}</math></i>	
Photolysis	304
<i>Additional Parameters</i>	
Inventory, Gg	4.8
Lifetime, days	6

<sup>a</sup>Net source representing a balance between emission to the atmosphere and atmospheric deposition to the ocean (see Table 2).

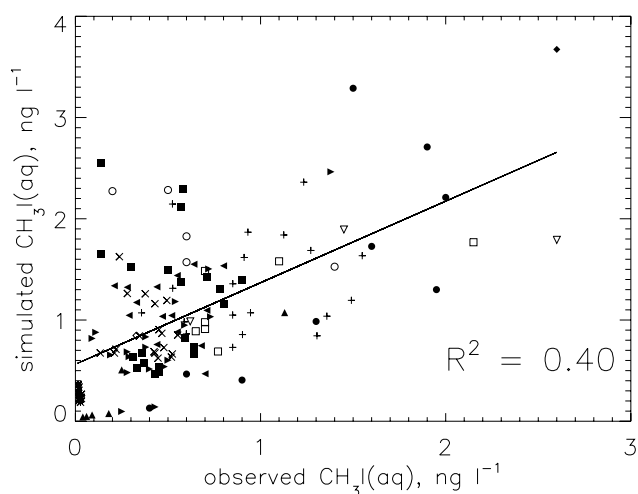


**Figure 3.** Global distribution of seawater concentrations of  $\text{CH}_3\text{I}(\text{aq})$  computed with the GEOS-CHEM model for each season. Observations from ship cruises are superimposed as circles. These observations include data from *Lobert et al.* [1995] (BLAST 1, January–February 1994, Eastern Pacific), *Campos et al.* [1996] (January–February 1989, North Sea), *Happell and Wallace* [1996] (February 1991, South Atlantic), *Reifenhauser and Heumann* [1992] (December 1987, Antarctica), *Schall et al.* [1997] (December 1991 to January 1992, Atlantic), and *Singh et al.* [1983] (December 1991, Eastern Pacific) in DJF; J. Butler (BLAST 3, NOAA/CMDL, unpublished data, 1999) (March–April 1992, Antarctic), *Campos et al.* [1996] (March–May, 1989, North Sea), *Moore and Tokarczyk* [1993] (April–May 1991, Greenland/Labrador/Atlantic), *Schall et al.* [1997] (March 1992, Atlantic), and *Tanzer and Heumann* [1992] (March–April 1989, Eastern Atlantic) in MAM; *Campos et al.* [1996] (June–August 1989, North Sea), *Happell and Wallace* [1996] (June 1993, South Atlantic), *Manley and Dastoor* [1987] (July 1985, Labrador Sea), and *Moore and Groszko* [1999] (July 1995, Greenland/Labrador Sea; June 1996, Ireland) in JJA; *Lobert et al.* [1995] (BLAST 2, October–November 1994, Atlantic), *Campos et al.* [1996] (September 1989, North Sea), *Happell and Wallace* [1996] (November 1996, Greenland/Norwegian Sea), *Moore and Groszko* [1999] (November 1995, Pacific), *Reifenhauser and Heumann* [1992] (October–November 1987, Antarctica), *Schall and Heumann* [1992] (September 1992, Spitzbergen, Fjord), and *Schall et al.* [1997] (November 1991, Atlantic) in SON.

observations superimposed as circles on the global seawater distribution of  $\text{CH}_3\text{I}(\text{aq})$  concentrations computed with GEOS-CHEM for each season. Observations from Blast 1 are believed to be artificially high because of a calibration problem and are therefore not included in this data set. There has been no intercalibration between measurements from the different experimental groups. Certain coastal regions appear to be hot spots, for example, the Antarctic

Peninsula [*Reifenhauser and Heumann*, 1992] and the Peru coast [*Singh et al.*, 1983].

[20] Simulated (monthly mean) and observed  $\text{CH}_3\text{I}(\text{aq})$  concentrations are correlated with  $r^2 = 0.4$  (Figure 4). The model predicts maximum oceanic concentrations at subtropical to middle latitudes in the summer season where solar radiation is high, but MLT is relatively low. Similar subtropical maxima are seen in some of the cruises [*Schall*



**Figure 4.** Scatterplot of simulated (monthly mean) versus observed oceanic  $\text{CH}_3\text{I}(\text{aq})$  concentrations, for the ensemble of observations listed in the legend and shown in Figure 3.

*et al.*, 1997; Moore and Groszko, 1999]. Most of the fine structure in the model distribution is driven by the variability in MLT. Observations are higher in the eastern than in the central Pacific, a feature reproduced by the model and reflecting the gradient in MLT. The relative seasonal amplitude in  $\text{CH}_3\text{I}(\text{aq})$  concentrations calculated for midlatitudes is about a factor of 6–10. Observations were carried out for 9 months in the North Sea region [Campos *et al.*, 1996], and the results indicate a similar seasonal amplitude and phase.

[21] The model is more successful in the Pacific than the Atlantic. It generally overestimates concentrations in the Atlantic by about a factor of 2. Most severe is the model overestimate in the Labrador Sea in spring and summer, where simulated concentrations are 1–2 orders of magnitude higher than observations from two separate cruises [Moore and Tokarczyk, 1993; Moore and Groszko, 1999]. Moore and Groszko [1999] noted no correlation with solar radiation in their Labrador Sea data.

[22] More generally, it appears that the model systematically overestimates observations at middle to high latitudes in spring-summer where solar radiation is high and MLT is low. Under these conditions the  $\text{CH}_3\text{I}(\text{aq})$  loss rate in the model is determined by ventilation to the atmosphere. In order to match observations, one would need an additional loss process with a rate constant of the order of  $1 \times 10^{-2} \text{ h}^{-1}$ . Biological degradation of  $\text{CH}_3\text{I}$  in seawater or high-latitude upwelling could possibly provide this missing sink but no data are available. A parallel can be drawn to methyl bromide ( $\text{CH}_3\text{Br}$ ), for which measurements in the open ocean surface waters of the North Atlantic indicate a rate constant for biological degradation that can reach  $1.2 \times 10^{-2} \text{ h}^{-1}$  [Tokarczyk and Saltzman, 2001]. In high-latitude ocean waters a loss rate of unknown origin of  $2.4 \times 10^{-3} \text{ h}^{-1}$  was required to explain the observed  $\text{CH}_3\text{Br}$  subsaturation [Moore and Webb, 1996; Lobert *et al.*, 1997].

[23] The NOAA/CMDL BLAST cruises in the eastern Pacific (BLAST 1), the Atlantic (BLAST 2), and the Antarctic (BLAST 3), provide  $\text{CH}_3\text{I}$  mixing ratio data for both surface air and seawater [Lobert *et al.*, 1995, 1996; J. Butler, unpublished data, 1999]. These data allow determi-

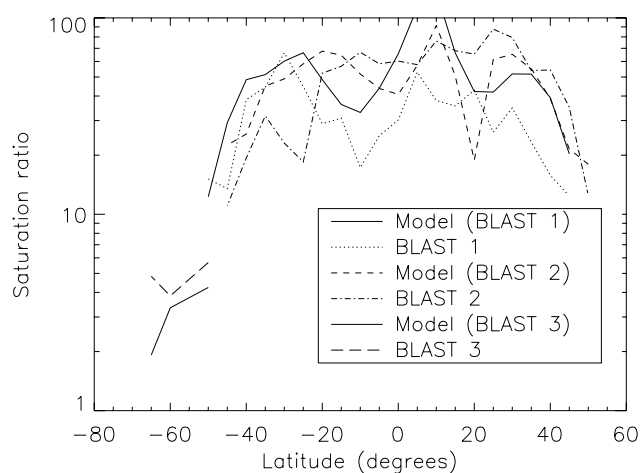
nation of the saturation ratio  $S = \Phi_{em}/\Phi_{dep} = [\text{CH}_3\text{I}(\text{aq})]/(K_H[\text{CH}_3\text{I}(\text{g})])$ . The resulting values of  $S$  are shown in Figure 5 and compared to the GEOS-CHEM model results. Since the calibration problems associated with the Blast 1  $\text{CH}_3\text{I}$  data affect both the air and seawater values, the offset cancels out in the calculation of  $S$ , and these values are included in Figure 5. Observed saturation ratios are of the order of 50–100 at  $30^\circ\text{S}$ – $30^\circ\text{N}$  and decrease at higher latitudes. The model captures the magnitude and latitudinal variability in the saturation ratio (Figure 5). In the model, the decrease in  $S$  at higher latitudes is driven by larger atmospheric  $\text{CH}_3\text{I}$  concentrations due to a longer  $\text{CH}_3\text{I}$  lifetime.

[24] Although  $S$  is low for the BLAST 3 cruise carried out at southern latitudes in austral fall, it is greater than unity, indicating that the Southern Ocean is a net source of  $\text{CH}_3\text{I}$ . Previously, Hoppel and Wallace [1996] measured subsaturation of  $\text{CH}_3\text{I}$  in the Greenland and Norwegian Seas in November. The model does not find subsaturation in this region, despite close agreement between the model and observed oceanic concentrations of  $\text{CH}_3\text{I}$  (0.05 ng/L). The observed atmospheric mixing ratio (2.4 pptv) is much higher than the corresponding model value (0.6 pptv).

[25] We conducted sensitivity simulations to investigate the possible role of biological production as a source of  $\text{CH}_3\text{I}(\text{aq})$  to the oceans but results were negative. Scaling the  $\text{CH}_3\text{I}(\text{aq})$  production rate  $P$  to net primary productivity (NPP) fields (taken from Six and Maier-Reimer [1996]) rather than to  $\text{DOC} \times \text{RAD}$  in the model resulted in no significant correlation between simulated  $\text{CH}_3\text{I}(\text{aq})$  fields and the observations of Figure 3. Addition of a biological source scaled to NPP to the photochemical source used in the standard simulation led to a systematic degradation of the comparison between simulated and observed  $\text{CH}_3\text{I}(\text{aq})$  concentration fields.

#### 4. Atmospheric Methyl Iodide Distribution

[26] Our compilation of atmospheric  $\text{CH}_3\text{I}$  observations comprises six ship cruises and eight aircraft missions (Table



**Figure 5.**  $\text{CH}_3\text{I}$  seawater saturation ratio  $S = [\text{CH}_3\text{I}(\text{aq})]/(K_H[\text{CH}_3\text{I}(\text{g})])$  calculated from the BLAST cruises (Figure 3), as a function of latitude. The corresponding GEOS-CHEM results are shown for comparison.

**Table 4.** Atmospheric Methyl Iodide Observations

Observation	Location	Date
<i>Aircraft Missions<sup>a</sup></i>		
ACE 1 (6, 10, 19, 20)	Australia	1995 (Nov.–Dec.)
ASTEX (7)	subtropical North Atlantic	1992 (June)
PEM West A (3, 4, 5, 8, 9)	northwest Pacific	1991 (Sept.–Oct.)
PEM West B (3, 4, 8)	northwest Pacific	1994 (Feb.–March)
PEM Tropics A (6, 10, 11, 12, 13, 20, 21)	south tropical Pacific	1996 (Sept.)
PEM Tropics B (8, 10, 12, 13, 21)	tropical Pacific	1999 (March–April)
SONEX (1, 2)	North Atlantic	1997 (Oct.–Nov.)
TRACE A (14, 15, 16, 17, 18, 22)	South Atlantic	1992 (Aug.)
<i>Ship Cruises</i>		
BLAST 1 [Lobert <i>et al.</i> , 1995, 1996]	eastern Pacific	1994 (Jan.–Feb.)
BLAST 2 [Lobert <i>et al.</i> , 1995, 1996]	Atlantic	1994 (Nov.)
BLAST 3 <sup>b</sup>	Antarctic	1996 (March)
Reifenhauer and Heumann [1992]	Antarctic	1987 (Dec.)
Singh <i>et al.</i> [1983]	eastern Pacific	1981 (Dec.)
Yokouchi <i>et al.</i> [1997]	South China Sea	1992 (Sept.–Oct.)
Yokouchi <i>et al.</i> [1997]	Bay of Bengal	1994 (Jan.–March)

<sup>a</sup>Data from D.R. Blake [e.g., Blake *et al.*, 1996a, 1996b]. Region numbers (in parentheses) correspond to Figure 6.

<sup>b</sup>NOAA/CMDL data (J. Butler, unpublished data).

4). Data from the eight aircraft missions were divided into 21 regions (Figure 6) chosen for their relatively homogeneous air mass characteristics and for conformity with the regions previously selected by Emmons *et al.* [2000] for averaging of aircraft atmospheric chemistry observations. The aircraft data are mostly over the Pacific and are clustered in the September–October and February–March periods. Data from Blast 1 (Pacific Ocean, January–February 1994) are not included because a calibration offset results in an overestimation of the atmospheric mixing ratios.

#### 4.1. Marine Boundary Layer

[27] Figure 6 shows the geographical distribution of boundary layer (0–1 km) concentrations for the two seasons. Comparison of ship with collocated aircraft (0.2–1 km) measurements show that the latter are on average 50% lower. Although the discrepancy might be due to a calibration difference, it could also reflect a vertical gradient of CH<sub>3</sub>I concentrations between the surface layer (sampled by the ships) and the MBL (sampled by aircraft). Such a gradient between the surface layer and the MBL has been previously noted in the case of DMS between concurrent ship and aircraft measurements during the ACE-1 mission [Suhre *et al.*, 1998; Mari *et al.*, 1998]. The discrepancy is inconsequential for the model calculation of the sea-to-air flux of CH<sub>3</sub>I since  $S \gg 1$  under almost all conditions.

[28] The right-hand panels of Figure 6 show the distributions of atmospheric CH<sub>3</sub>I in the GEOS-CHEM model. For comparison with the aircraft regions, the model data have been averaged between 0 and 1 km, and for comparison with the ship observations the model data are for the lowest model layer (0–0.15 km). The model overestimates observed mixing ratios by a factor of 3 over the Southern Ocean and Tasman Sea (regions 19 and 20) during the austral summer, suggesting an overestimated source or a missing seasonal sink, perhaps biological, for CH<sub>3</sub>I in the ocean; such a sink was proposed previously in section 3 from comparison of simulated and observed CH<sub>3</sub>I(aq) concentrations. The high concentrations off the coast of China in the model summer reproduce qualitatively the same feature in the observations

but are too high by a factor of 2. Rice paddies are the dominant source of CH<sub>3</sub>I for this region in the model, and the overestimate suggests that the rice paddy source of CH<sub>3</sub>I from Redeker *et al.* [2000] is too high.

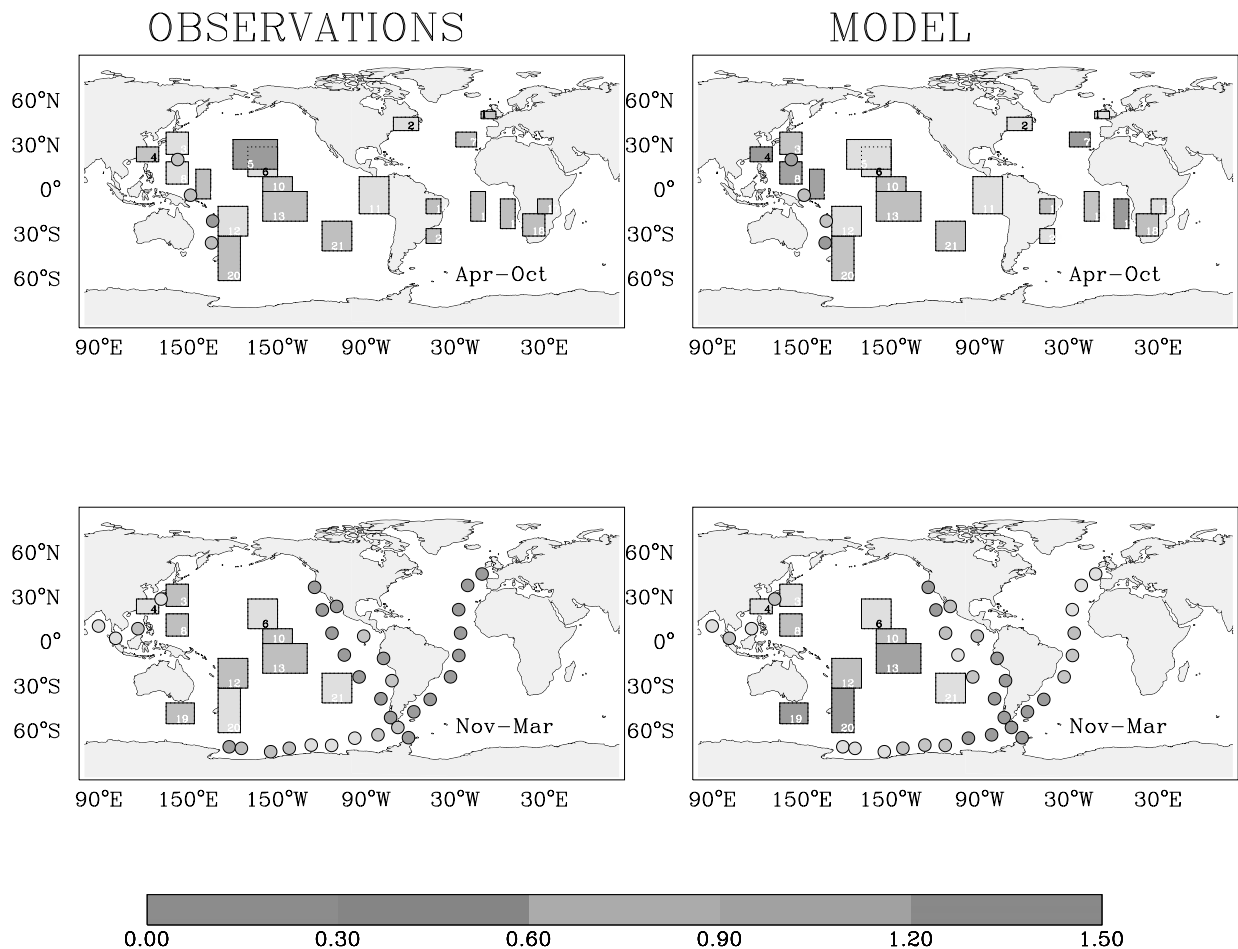
[29] It is evident from Figure 6 that the model has difficulty reproducing the patterns observed in the CH<sub>3</sub>I MBL concentration field. This problem does not necessarily compromise the use of the relative CH<sub>3</sub>I vertical profiles as a test of convection, as demonstrated in the following section.

#### 4.2. Vertical Profiles

[30] Observed vertical profiles of CH<sub>3</sub>I concentrations for selected regions of Figure 6 are compared to model results in Figure 7. In most cases the simulated concentrations are dominated by the oceanic source. When another source contributes more than 5% to the total sum, it is explicitly indicated. The rice paddy source makes a 30% contribution to the CH<sub>3</sub>I concentration in the upper troposphere over the North Pacific and Hawaii during PEM-West A and ACE 1. Biomass burning accounts for 50% of the boundary layer concentration of CH<sub>3</sub>I over Africa (regions 17 and 18) during the Trace-A mission.

[31] Concentrations of CH<sub>3</sub>I in the tropics decrease with altitude up to 2–3 km, representing the base of the trade wind inversion (TWI). Transport to higher altitudes generally involves wet convection. Many of the profiles in Figure 7 show enhanced concentrations in the middle and upper troposphere (UT) associated with deep convection. In most cases the overall vertical gradient in the tropical free troposphere is weak, indicating that convective outflow is distributed over all altitudes [Wang *et al.*, 2000, 2002]. Comparison of the simulated and observed profiles indicates that the model reproduces the gradual decrease with altitude in the lower troposphere (LT), the location of the TWI at 2–3 km altitude, and the lack of mean vertical gradient at higher altitudes. There is a tendency in the model, and to a somewhat lesser degree in the observations, for a “C-shaped” vertical profile with high concentrations in the UT reflecting preferential deep convection [Prather and Jacob, 1997].





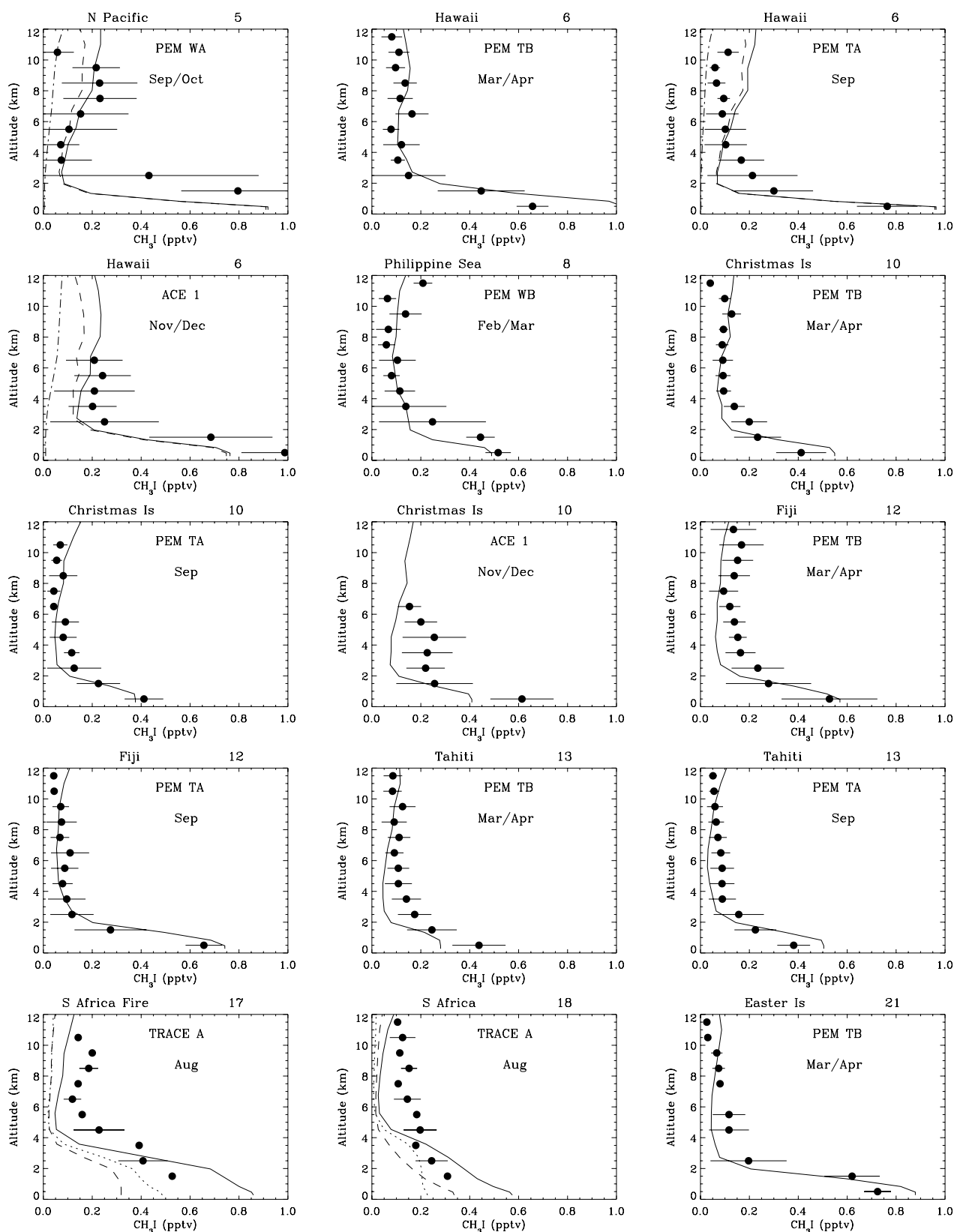
**Figure 6.** Atmospheric  $\text{CH}_3\text{I}$  concentrations in the marine boundary layer in (top) April–October and (bottom) November–March. Observations are shown in the left-hand panels, and the model results are shown in the right-hand panels. Circles are ship cruise data. Rectangles are aircraft data at 0- to 1-km altitude averaged over coherent regions [Emmons *et al.*, 2000]. References for the observations are given in Table 4. The model data are averaged between 0 and 1 km for the aircraft regions, while for the ship regions, the model data from the lowest atmospheric layer (0–0.15 km altitude) are shown. The scale has been chosen to highlight the variability between the different regions; the BLAST 1 ship cruise data in the eastern Pacific in January–February (Table 4) are anomalously high and exceed the 1.5 pptv maximum of the scale.

[32] Although errors in the simulation of MBL concentrations (section 4.1) complicate the interpretation of the vertical profiles in Figure 7 it is in fact the shapes of these profiles, rather than the absolute concentrations, that provide a diagnostic of vertical transport. We define therefore a marine convection index (MCI) as the ratio of the mean  $\text{CH}_3\text{I}$  concentration at 8–12 km (upper troposphere, UT) to the mean concentration at 0–2.5 km (lower troposphere, LT) for a given region. The observed and simulated values of the MCI are shown in Table 5 for the Pacific regions of Figure 7. Table 5 also shows the  $\text{CH}_3\text{I}$  lifetime for each region, which falls in a narrow range (4.0–4.8 days).

[33] The observed MCI values in Table 5 range from 0.11 (Easter Island) to 0.40 (Fiji in March–April), representing large differences in deep convective activity. During PEM-Tropics B (March–April) the observed MCI is highest over the tropical western Pacific (Fiji) and decreases from west to east, consistent with the large-scale upwelling and subsidence associated with the Walker circulation. During PEM-

Tropics A (September), the MCI is highest over the central South Pacific (Tahiti). The relatively low MCIs observed over Hawaii (0.20–0.26) reflect the general subsidence due to the North Pacific High. All regions in the equatorial and South Pacific indicate higher MCIs (greater convective mixing) during PEM-Tropics B than PEM-Tropics A, consistent with the wet season timing of PEM-Tropics B [Wang *et al.*, 2002]. The strongest seasonal difference is over Fiji where the MCI decreases from 0.4 in March–April (PEM-Tropics B) to 0.16 in September (PEM-Tropics A). Higher ratios are also found over Hawaii for PEM-Tropics B (0.26) than for PEM-Tropics A (0.20), for reasons that are less clear since the timing of PEM-Tropics B corresponds to the local dry season.

[34] The model reproduces the observed MCI with no significant bias (overall offset of only +11%) for the ensemble of data in Table 5, indicating that the GEOS data provide on average a good simulation of deep marine convective mass fluxes over the Pacific. However, the



**Figure 7.** Vertical profiles of  $\text{CH}_3\text{I}$  concentrations measured from aircraft and averaged over the regions of Figure 6 (region numbers indicated above each plot correspond to those of Figure 6). Horizontal bars are the standard deviations computed from the individual measurements. The corresponding model profiles for the appropriate months are shown as solid black lines. For regions where sources other than oceanic contribute more than 5% to the total  $\text{CH}_3\text{I}$  concentrations in the model, the contributions from individual sources are also shown (oceanic (dashed line), biomass burning (dotted line), and rice paddies (dash-dotted line)).

**Table 5.** Marine Convection Index (MCI) Over the Pacific: Ratio of Upper Tropospheric (UT; 8–12 km) to Lower Tropospheric (LT; 0–2.5 km) CH<sub>3</sub>I Concentrations

Region	Campaign	Lifetime, <sup>a</sup> days	MCI (Observed)	MCI (Simulated)
5. North Pacific	PEM-WA	4.8	0.22	0.41
6. Hawaii	PEM-TB	4.4	0.26	0.18
	PEM-TA	4.7	0.20	0.39
8. Philippine Sea	PEM-WB	4.4	0.27	0.31
10. Christmas Island	PEM-TB	4.0	0.32	0.30
	PEM-TA	4.1	0.24	0.37
12. Fiji	PEM-TB	4.3	0.40	0.22
	PEM-TA	4.7	0.16	0.14
13. Tahiti	PEM-TB	4.2	0.34	0.44
	PEM-TA	4.1	0.23	0.19
21. Easter Island	PEM-TB	4.4	0.11	0.11

<sup>a</sup>Mean model lifetime of CH<sub>3</sub>I in the 0- to 12-km column (24-hour average).

correlation between simulated and observed values of the MCI is weak ( $r^2 = 0.15$ ). The model reproduces qualitatively the spatial gradients and seasonal variations observed over the South Pacific. The model indicates greater deep convection over Hawaii during PEM-Tropics A than during PEM-Tropics B, consistent with the local wet season but contrary to the observations.

## 5. Summary

[35] We used an oceanic mixed layer model coupled to the GEOS-CHEM 3-D atmospheric transport model to provide a first global simulation of oceanic and atmospheric CH<sub>3</sub>I. The model was driven by GEOS assimilated meteorological observations from the NASA Data Assimilation Office. Model results were evaluated with an extensive data base of global observations for CH<sub>3</sub>I in the atmosphere and the oceans. We applied the results to understand the factors controlling the global budget of CH<sub>3</sub>I and to examine the usefulness of CH<sub>3</sub>I as a tracer of marine convection in global atmospheric models, complementing the use of <sup>222</sup>Rn as a tracer of continental convection.

[36] The concentration of CH<sub>3</sub>I(aq) in the oceanic mixed layer was modeled by assuming steady state between photochemical production (dependent on solar radiation and DOC), chemical reaction with Cl<sup>-</sup> ions (dependent on temperature), and exchange with the atmosphere (dependent on wind speed and on the local atmospheric concentration of CH<sub>3</sub>I). This formulation accounts for 40% of the variance in the ensemble of seawater observations. The model predicts high concentrations in the spring-summer hemisphere at midlatitudes where solar radiation is high and seawater temperature is relatively low, but corresponding observations in both the ocean and the air show much lower values than the model. The discrepancy may be possibly due to a missing biological sink for CH<sub>3</sub>I(aq). Measurements of the biological degradation of CH<sub>3</sub>I(aq) are needed.

[37] We calculate a gross production of CH<sub>3</sub>I(aq) in the ocean of 477 Gg yr<sup>-1</sup> and a net global CH<sub>3</sub>I source from the ocean to the atmosphere of 214 Gg yr<sup>-1</sup>. The total emission of CH<sub>3</sub>I from all sources in the model is 304 Gg yr<sup>-1</sup>, including additional contributions from rice paddies (71 Gg yr<sup>-1</sup>), wetlands (7 Gg yr<sup>-1</sup>), biomass burning (9 Gg yr<sup>-1</sup>), and wood fuel (3 Gg yr<sup>-1</sup>). Comparison of model results to

observations off the coast of China suggests that the estimate of the source from rice paddies [Redeker *et al.*, 2000] is too high.

[38] We used vertical profiles of CH<sub>3</sub>I observed from aircraft to test the model simulation of vertical transport over the oceans. Observations in the tropical marine atmosphere indicate a gradual decrease up to the trade wind inversion (TWI) at 2–3 km, a sharp transition across the TWI, and little vertical gradient through the rest of the troposphere reflecting convective outflow at all altitudes. We find that the GEOS meteorological fields reproduce qualitatively these features. We went on to use the ratio of observed CH<sub>3</sub>I in the upper troposphere (8–12 km) to the lower troposphere (0–2.5 km) as a marine convection index (MCI). The observed MCI over the Pacific ranges from 0.11 for strongly subsiding regions (southeastern subtropical Pacific) to 0.40 for strongly upwelling regions (western equatorial Pacific). The GEOS-CHEM model reproduces the observed MCI values over the Pacific with no significant bias (overall offset of only +11%), implying a good mean simulation of marine convection, but the correlation with the observed MCI for individual regions is weak ( $r^2 = 0.15$ ). The model reproduces qualitatively the spatial gradients and seasonal variations of the MCI observed over the South Pacific, but the seasonal variations are reversed over Christmas Island and Hawaii. We propose that the MCI can provide a useful diagnostic for testing the simulation of marine convection in global atmospheric models.

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