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Photon echo studies of photosynthetic light harvesting

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Abstract The broad linewidths in absorption spectra of photosynthetic complexes obscure information related to their structure and function. Photon echo techniques represent a powerful class of time-resolved electronic spectroscopy that allow researchers to probe the interactions normally hidden under broad linewidths with sufficient time resolution to follow the fastest energy transfer events in light harvesting. Here, we outline the technical approach and applications of two types of photon echo experiments: the photon echo peak shift and two-dimensional (2D) Fourier transform photon echo spectroscopy. We review several extensions of these techniques to photosynthetic complexes. Photon echo peak shift spectroscopy can be used to determine the strength of coupling between a pigment and its surrounding environment including neighboring pigments and to quantify timescales of energy transfer. Two-dimensional spectroscopy yields a frequency-resolved map of absorption and emission processes, allowing coupling interactions and energy transfer pathways to be viewed directly. Furthermore, 2D spectroscopy reveals structural information such as the relative orientations of coupled transitions. Both classes of experiments can be used to probe the quantum mechanical nature of photosynthetic lightharvesting: peak shift experiments allow quantification of correlated energetic fluctuations between pigments, while 2D techniques measure quantum beating directly, both of which indicate the extent of quantum coherence over multiple pigment sites in the protein complex. The mechanistic and structural information obtained by these techniques reveals valuable insights into the design principles of photosynthetic light-harvesting complexes, and a multitude of variations on the methods outlined here.

Keywords Photon echo · Photon echo peak shift · 2D Fourier transform photon echo spectroscopy · Spectral band broadening · Reaction center · FMO · LH3 · Electronic quantum coherence

Abbreviations

2D Two-dimensional FMO Fenna–Matthews–Olson PMT Photomultiplier tube

3PEPS Three-pulse photon echo peak shift 1C3PEPS One-color three-pulse photon echo peak

shift

2C3PEPS Two-color three-pulse photon echo

peak shift

BChl Bacteriochlorophyll **B**pheo Bacteriopheophytin Rb. sphaeroides Rhodobacter sphaeroides Rs. rubrum Rhodospirillum rubrum **CCD** Charge-coupled device LH₁ Light-harvesting complex 1 LH2 Light-harvesting complex 2 LH3 Light-harvesting complex 3

LO Local oscillator

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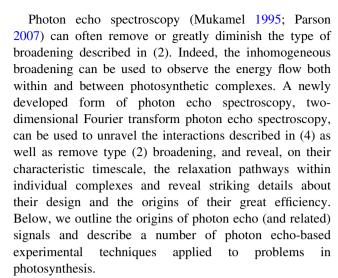
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Introduction

The electronic absorption and emission spectra of photosynthetic pigment–protein complexes (Blankenship 2002) are generally broad and lack the kind of details needed to provide insight into their structure, function, and design principles. While the broad absorption bands are advantageous for solar energy absorption, progress in understanding their exquisite effectiveness in light harvesting and trapping, and in charge separation to initiate the chemistry of photosynthesis, requires that we find ways to remove at least some of the broadening that obscures the information content.

What are the origins of the broadening of spectra (Fleming and Cho 1996; Parson 2007) of photosynthetic complexes? In general, there are five factors at work. (1) The bare electronic transitions are broadened by the vibrational transitions (of both chromophore and protein) that accompany them. These transitions arise because the atomic nuclei have different equilibrium positions when the chromophore (e.g., chlorophyll) is in the excited state than in the electronic ground state. This is generally called homogeneous broadening. (2) Individual chromophores in different members of the entire protein ensemble may have slightly different electronic transition energies, even though they occupy structurally identical locations. This can arise because of different ionization states of protein side chains close to their pK_a, different orientations of side chains, slight distortions of the overall protein structure, and a host of similar small influences. The overall effect is to smear out the transition to produce inhomogeneous broadening. (3) Chlorophylls and other pigments are generally bound in a variety of non-equivalent sites in an individual protein complex. For example, the Fenna-Matthews-Olson (FMO) complex of green sulfur bacteria binds seven bacteriochlorophyll molecules each in a unique site. This type of inhomogeneous broadening may produce a set of more discrete transition energies than the broadening arising by mechanism (2). Both (2) and (3) give transition energies that vary slowly or not at all on the time scale of the optical functions of photosynthetic complexes. (4) In many photosynthetic complexes, the chromophores are held very close to one or more neighbors leading to electronic mixing and associated spectral shifts from the individual molecule's unperturbed transition. This can lead to a set of chemically identical chromophores having a significantly broader spectrum than a similar, but non-interacting, set of molecules. (5) Finally, several processes can, and often do, happen very fast in photosynthetic complexes, leading to lifetime broadening. An excellent summary of the spectroscopy of photosynthetic complexes can be found in Van Amerongen et al. (2000).



The basis of photon echo spectroscopy, as with other "ultrafast" techniques, is the interrogation of a system with laser pulses short enough to track dynamical processes of interest. In this work, ultrafast means tens of femtoseconds (where a femtosecond is 10^{-15} s), a timescale on which the fastest energy transfer processes occur between neighboring pigments in light-harvesting complexes. The method requires a sequence of laser pulses to interrogate the sample and, as with pump-probe and related experiments, allows observation of excited state dynamics. However, in photon echo experiments, the laser pulse sequence is specifically designed to take advantage of the *reversibility of dephasing*, illustrated in a general way as follows.

Upon irradiation by a laser pulse, the system begins to oscillate between quantum energy levels. A full quantum mechanical description is beyond the scope of this article, but an analogy can be drawn to a collection of springs, set into motion by the external perturbation (the pulse). Imagine that each of the springs oscillates with a slightly different frequency, analogous to inhomogeneous broadening wherein the electronic transition frequencies of a collection of chromophores vary, described by (2) above for photosynthetic light-harvesting complexes. The result of this distribution of frequencies is that the "springs," oscillating in phase immediately after interaction with the pulse, become gradually less synchronized over time. This is known as dephasing. Imagine then that at some later instant, the motion of the springs is simultaneously reversed by another perturbing pulse. As long as each of the springs maintains its original oscillation frequency and changes only its direction, the overall dephasing is reversed also. When this reverse dephasing or rephasing process occurs not with springs but with a collection of chromophores interacting with laser pulses, the effect is for the sample to emit a light pulse "echoing" the input pulse at the instant when the oscillators are once more in phase.



The key to the unique information contained in photon echo signals is that the appearance of a photon echo signal depends on each of the springs remembering its initial oscillation frequency and phase. If, on the other hand, the frequencies are individually modified or the phases shifted (as can occur through coupling to vibrational motions of the pigments or proteins), the collective motion of the springs devolves into random noise; the constructive interference—rephasing—is never realized, and a photon echo signal is not emitted. Thus, the signal is uniquely sensitive to the coupling between the electronic transitions on the pigments and the nuclear motions of the "bath" (motions of the pigments themselves and of the surrounding protein). Recent work, including some of the experiments summarized here, has shown that, in fact, the detailed pigment-protein interactions in photosynthesis play an important role in controlling energy flow through the complexes. Furthermore, photon echo signals track energy transfer between the electronic states of neighboring chromophores. Therefore, photon echo experiments are well suited to the study of photosynthetic light harvesting.

The experimental pulse sequence for three-pulse photon echo experiments is shown in Fig. 1. The first input pulse instigates the initial dephasing process described above. The effect of the second pulse is to freeze the "springs," halting dephasing; the third pulse triggers rephasing by setting the motion in reverse. The initial dephasing period, the interval between pulses 1 and 2, is labeled τ or the "coherence" time (as the oscillation occurs because the system is in a coherent quantum mechanical superposition state). The second interval is known as T or the "population" time (also called the "waiting" time), and the third, t, as the "echo" time. The system's ability to rephase and generate a photon echo diminishes with increasing population time, as each chromophore gradually loses its "memory" of its initial frequency due to interactions with the solvent or energy transfer between pigments. Therefore, experimentally varying T allows experimentalists to observe dynamical processes.

How does the experimentalist enforce the above sequence of events? One way is to experimentally "select" these processes using a noncollinear beam geometry. The geometry varies depending on the experimental scheme, as

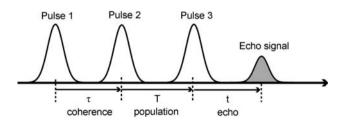


Fig. 1 Pulse sequence for three-pulse photon echo experiment

described below. Due to conservation of momentum, the constructive interference of polarization (the photon echo) is scattered in a predictable, so-called phase-matched, direction. Phase matching selects desired signals and allows background-free detection. The use of phase matching is a distinct advantage of optical photon echo techniques over their nuclear magnetic resonance analogs, and is made possible by the large sample size relative to the wavelength of the incident radiation. Detection schemes depend on the particular type of photon echo experiment performed. The experiments illustrated below demonstrate how three-pulse photon echo experiments can be designed to conveniently probe various aspects of the pigment-protein interactions and energy transfer processes in photosynthesis.

Photon echo peak shift spectroscopy

Experimental considerations

Preparing samples of photosynthetic proteins for the study by echo experiments generally requires solubilizing the isolated proteins of interest in a buffer solution with a small amount of detergent. The use of a narrow ($\sim 100-200 \mu m$) quartz sample cell minimizes effects due to reabsorption of emitted signals. Furthermore, the optical density (OD) of the sample must be chosen to minimize signal distortions due to propagation of signal and pulses through the sample (exacerbated at high ODs) or interference with the solvent response (exacerbated at low ODs) (Christensson et al. 2008). In the experiments presented here, sample ODs were in the range of 0.1-0.3. In high temperature studies, the sample is often flowed through the cell to continuously regenerate fresh sample in the focal spot of the laser beams. However, photon echo experiments are often performed on low temperature glasses to reduce broadening due to the nuclear motions as described above. These samples are prepared by mixing the protein solution with a cryoprotectant such as glycerol (to prevent formation of cracks, which would scatter laser light into the detector) and freezing the sample in a windowed cryostat to the desired temperature. The cryotstat is mounted on a movable stage in the laser beam path, such that the sample may be aligned to the focal point of the laser beams. Localized sample damage is avoided by periodically shifting the cell laterally or vertically to an unused spot and by minimizing the input power of the laser beams as much as possible. Also, at very high excitation energies, it is possible to create multiple excitations (excitons) in the sample and produce spurious signals in the same phase-matched directions as the third order signal. This possibility is discussed by Bruggemann et al. (2007).



Routine generation of tunable, femtosecond laser pulses using Ti:Sapphire sources has been achieved over the last two decades (Jimenez and Fleming 1996; Demtroder 2003; Rulliere 2003; Parson 2007). In the photon echo experiments described below, three ultrashort pulses are aligned to pass the vertices of an equilateral triangle on a plane perpendicular to pulse propagation and tightly focused on a sample (Fig. 2). Echo signals are generated in phase-matched directions (e.g., $-\mathbf{k}_1+\mathbf{k}_2+\mathbf{k}_3$, $+\mathbf{k}_1-\mathbf{k}_2+\mathbf{k}_3$, or $+\mathbf{k}_1+\mathbf{k}_2-\mathbf{k}_3$, where the ks are the momentum vectors of the laser beams). The photon echo signals in selected phase-matched directions are spatially filtered into the detection system by placing a mask after the sample, thereby blocking other signals and scattered light. A photomultiplier tube (PMT) or a photodiode collects the signals. Since the detectors respond more slowly than the experimental time scale, one obtains time t-integrated photon echo signals as a function of τ and T.

One-color three-pulse photon echo peak shift (1C3PEPS)

In disordered systems like photosynthetic complexes where electronic dephasing is extremely rapid, it is well established that the photon echo peak shift provides useful information about solvation dynamics, i.e., the rearrangement of the "solvent" (the protein environment) nuclei to accommodate electronic excitations on the chromophores. The peak shift (τ^*) is defined simply as the coherence time (τ) at which the photon echo signal reaches maximum intensity for a given T. For precise determination of τ^* , the average peak shift of echo signals from two different phase matching directions $(-\mathbf{k}_1+\mathbf{k}_2+\mathbf{k}_3)$ and $+\mathbf{k}_1-\mathbf{k}_2+\mathbf{k}_3$ is often obtained (Fig. 2). The usefulness of 1C3PEPS lies in the fact that it closely follows the time correlation function of a transition frequency of a pigment, which contains solvation dynamics information (Cho et al. 1996). In general, the initial peak shift reflects the system-bath coupling strength because it is proportional to the reciprocal value of

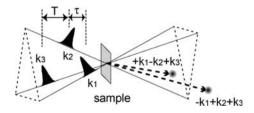


Fig. 2 Three-pulse photon echo peak shift experiment configuration. Three pulses are focused on a sample and the photon echo signals are emitted in the phase-matched direction, $-\mathbf{k}_1+\mathbf{k}_2+\mathbf{k}_3$ and $+\mathbf{k}_1-\mathbf{k}_2+\mathbf{k}_3$. $\lambda_1=\lambda_2=\lambda_3$ for 1C3PEPS, $\lambda_1=\lambda_2<\lambda_3$ for downhill 2C3PEPS, $\lambda_1=\lambda_2>\lambda_3$ for uphill 2C3PEPS, and $\lambda_1=\lambda_3\neq\lambda_2$ for 2CECPE. **ks** and λ s are the momentum vectors and the wavelengths of the pulses, respectively

the mean square fluctuation amplitude of the transition frequency. Peak shifts at large T indicate the extent of static disorder, and the decay captures population dynamics. For example, Jimenez et al. (1997) revealed that initial peak shifts for light-harvesting complexes (LH1 and LH2) of purple photosynthetic bacteria, Rhodobacter (Rb.) sphaeroides are large (~ 25 fs) compared to the peak shifts of typical dyes in polar solvents (10-15 fs), which indicates weak coupling of the pigments in these complexes to the surrounding protein matrix. This relatively weak coupling may be essential to minimize heat dissipation to the surroundings and, therefore, maximize the energy transfer efficiency from LH2 to LH1 to the reaction center. Another 1C3PEPS experiment on the isolated B820 subunit (a subunit of the LH1 complex, so-called because it absorbs near 820 nm) of LH1 in *Rhodospirillum rubrum*, in comparison with 1C3PEPS on the whole LH1 complex, clearly demonstrated the contribution of energy transfer to the 1C3PEPS signal decay (Fig. 3) (Yu et al. 1997). The signal from the LH1 complex showed a rapid decay component in early T corresponding to energy transfer around the ring and resulting in a small peak shift value at long T (circles). Note that (excitation) energy transfer from one (excited) molecule to another leads to loss of correlation. To the contrary, the energy transfer out of the subunit is blocked in the B820 subunit, which consists only of one α and one β transmembrane polypeptide and two BChla molecules. Therefore, the B820 subunit exhibits a generally large peak shift (squares, Fig. 3). The solid line indicates the simulated 1C3PEPS profile with the same parameters for the LH1 complex but without an energy transfer factor. The experiments also demonstrate that the photon echo peak shift is

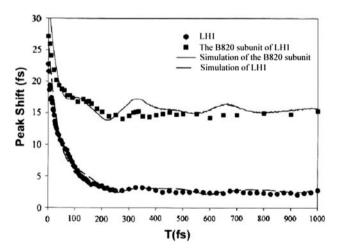


Fig. 3 1C3PEPS measurements of LH1 of *Rhodobacter* (*Rb.*) *sphaeroides* (*circles*) and the B820 subunit from LH1 of *Rhodospirillum* (*Rs.*) *rubrum* (*squares*). The solid lines represent two simulations with identical input parameters except that the energy transfer rate is set to zero for the B820 sample (Yu et al. 1997). Figure reprinted by permission from Elsevier (Yu et al. 1997)



sensitive to energy transfer within the laser pulse window as well as energy transfer out of the detection window because the peak shift measures the rephasing capability. Moreover, unlike conventional transient absorption or time-resolved fluorescence studies, it is insensitive to reverse energy transfer between transitions of similar energies. These features are useful in studying the diagonal elements of a Hamiltonian of photosynthetic systems in which multiple replicas of pigments are common. In this sense, the evolution of photon echo peak shift reflects excited state dynamics of a photosynthetic system in detail.

Two-color three-pulse photon echo peak shift (2C3PEPS)

Pigments in photosynthetic pigment–protein complexes are often closely packed and form an excitonically coupled system, described as broadening source (4) above and shown schematically in Fig. 4. For instance, the reaction center of the purple bacterium Rb. sphaeroides has three distinct absorption bands assigned to the Q_v bands of the primary bacteriochlorophylls (BChls), accessory BChls, and bacteriopheophytins (Bpheos), named the P, B, and H bands. However, the excitonically coupled states contain the properties of electronic states from different molecules, and they are correlated to some extent. The notations P, B and H denote only the major contributing molecules to each state. The recently developed 2C3PEPS method is suitable for investigating such correlated electronic states when the states are of different energies. The more correlation between the states, the larger the peak shift signal

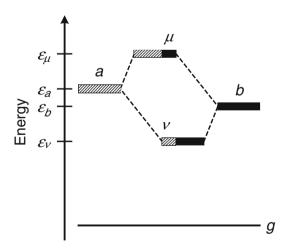


Fig. 4 Energy diagram for an excitonically coupled system. a and b are the localized states of two different chromophores (if the coupling interaction energy were zero), and μ and v are the upper and lower excitonically coupled states formed from the mixing of a and b. Hatched and solid bars depict conceptually the portion of the contribution from a and b individually to the mixed states. Because μ and v contain character of both a and b, they are correlated

generated, as an extended concept of 1C3PEPS. In practice, the 2C3PEPS experiment is performed using one wavelength for the first two laser pulses and a different wavelength for the last pulse in a setup similar to that in Fig. 2. If the first two pulses are at a higher energy than the last one, the experiment is called "downhill" and if lower, it is called "uphill." In particular, 2C3PEPS enables us to directly determine the coupling constant, J, between the two coupled states, i.e., the off-diagonal elements of the Hamiltonian, without prior knowledge about the site energies of the pigments in the protein matrix. That is, it allows researchers to differentiate between broadening sources (3) and (4) described in the Introduction. J is related to the mixing angle (θ) by:

$$\tan(2\theta) = \frac{J}{\varepsilon_a - \varepsilon_b},$$

where ε_a and ε_b are the monomer energies (or site energies in protein matrix) of molecules a and b, respectively. The mixing angle can be obtained from the experimental mixing coefficient, C_{uv} :

$$C_{\mu
u} = 2 \sin^2 heta \cos^2 heta pprox rac{ au_{\mu
u}^*(T)}{ au_{\mu
u}^*(T) + rac{1}{2\kappa}(au_{\mu}(T) + au_{
u}(T)\kappa^3)},$$

where $\kappa = \tau_{\mu\nu}^*(T)/\tau_{\nu\mu}^*(T)$ (Mancal and Fleming 2004). As can be seen in the notation of Fig. 4, $\tau_{\mu}^*(T)$ and $\tau_{\nu}^*(T)$ represent the 1C3PEPS values for upper and lower excitonic states, respectively, and $\tau_{\mu\nu}^*(T)$ and $\tau_{\nu\mu}^*(T)$ represent uphill and downhill 2C3PEPS, respectively. The value of J can be determined using the following equation based on the difference in energy between the two observed exciton states:

$$E_{\mu} - E_{\nu} = 2J\sqrt{1 + \left(\frac{1}{\tan(2\theta)}\right)^2}.$$

The method was applied to Bpheos (H band) and accessory BChls (B band) of the reaction center of *Rb. sphaeroides* using pulses centered at 750 and 800 nm, yielding an estimate of the coupling constant between them of $\sim 170 \pm 30 \text{ cm}^{-1}$ (Parkinson et al. 2007).

Another version of the two-color three-pulse photon echo experiment alternates colors in the pulse sequence to generate specific electronic coherences (two-color electronic coherence photon echo, 2CECPE) (Lee et al. 2007). A coherence between two excitonic states is prepared during T, whereas the 3PEPS experiments described above generate population states during T. Therefore, the photon echo signal along the T axis contains dynamics of quantum mechanical coherence between μ and ν , and between g and μ (or g and ν) along τ (see Fig. 4). Surprisingly long-lived (440 fs at 77 K) electronic coherence between B and H was observed in the Rb. sphaeroides reaction centers with



the primary electron donor chemically oxidized, suggesting that quantum coherence plays an important role in enhancing the efficiency of energy transfer in pigment-protein complexes. Lee et al. (2007) argued that correlated protein motion surrounding the pigments is essential to protect the quantum coherence, a mechanism that may be ubiquitous in photosynthetic machinery. The application of the 2CECPE technique to the bacterial reaction center shows how mechanistic details of photosynthetic systems can be obtained using photon echo methods.

Two-dimensional Fourier transform photon echo spectroscopy

Principles of two-dimensional Fourier transform photon echo spectroscopy

As demonstrated above, photon echo experiments afford control over multiple frequency and time "handles" (i.e., pulse color and duration of time periods T and τ). Furthermore, the Fourier relationship between time- and frequency-domain signals is of central importance, and can be exploited by researchers in the design of experiments. Whereas the frequency domain enables observation of excitation energy levels and transition strengths, the time domain allows direct measurement of dynamics. Adopting a time or frequency view accesses different types of information, and the various possibilities underlie the flexibility of photon echo-based techniques. Here we briefly outline the technique of two-dimensional (2D) Fourier transform photon echo spectroscopy, in which a mixed time/frequency view gives a wealth of information about the system of interest.

A 2D spectrum graphs transitions along two frequency axes and contains information about correlation between transitions at different frequencies (Jonas 2003). The direct visualization of correlation, or coupling, in the off-diagonal region of a 2D spectrum is the method's most immediate advantage over linear spectroscopy. The technique is the most versatile photon echo technique, taking advantage of rephasing to remove inhomogeneous broadening, while presenting both a frequency and time view of the system, allowing simultaneous characterization of the system's energetics and dynamics. This is accomplished by viewing the photon echo signal as a function of the two Fourier frequency axes corresponding to the coherence evolution periods τ and t for a series of population times, T.

Experimental considerations

Measuring a 2D spectrum requires spectral resolution (measurement of all frequency components) of the photon

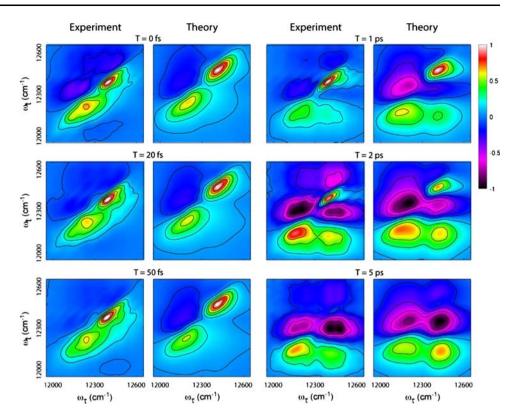
echo signal (for a detailed treatment of the experimental 2D apparatus, see Brixner et al. 2005). The signal is measured in only one phase-matched direction, and a beam alignment is adopted in which the three excitation beams pass through three corners of a square and the signal propagates in the direction of the fourth corner. The photon echo signal, measured while scanning the coherence time, τ , for a given population time, T, is directed into a spectrometer and imaged on a CCD (charge-coupled device) camera. Thus, signal evolution over the echo time t is indirectly measured through its Fourier analog, ω_t . Heterodyne detection, performed by interfering the signal with a "local oscillator" pulse, identical to the excitation pulses except attenuated by a neutral density filter, allows measurement of both the amplitude and phase of the signal electric field. The signal field is thus measured as a function of τ , T, and ω_t , and Fourier transformation along τ yields the signal as a function of ω_{τ} , T, and ω_{t} . The spectrum is displayed (in our convention) with the ω_{τ} axis as the abscissa and the ω_t axis as the ordinate, and the evolution of the spectrum with increasing T allows observation of dynamics. In analogy to transient absorption experiments, the ω_{τ} axis corresponds to the "pump" frequency, while the ω_t axis corresponds to the "probe" frequency.

Applications

The experimental and simulated 2D spectra of light-harvesting complex 3 (LH3) from purple bacteria Rhodopseudomonas acidophila, shown in Fig. 5 (Zigmantas et al. 2006), illustrate the general features of a 2D spectrum. The overall appearance results from the interference of signals from different processes: positive signals arise from stimulated emission or ground-state bleaching (depletion of population in the ground state as a result of excitation), both of which result in more light being emitted. Excited state absorption to yet-higher levels results in less light emitted and thus in negative signals (Brixner 2005). For example, in Fig. 5, positive signals dominate at early population times (T < 1 ps), while negative signals dominate at later times. Peaks along the diagonal in early-population-time 2D spectra match the peaks observed in a linear absorption spectrum. In the LH3 spectra at T = 0, two strong diagonal bands are observed corresponding to the dominant Q_v bands in the near-IR spectrum. A cross peak in a 2D spectrum connecting two diagonal peaks indicates coupling between the two states. When electronic coupling is sufficiently strong relative to the coupling to the bath, quantum coherence may be preserved long enough for observation, giving rise to coherent cross peaks. However, cross peaks may also arise as a result of irreversible, dissipative energy transfer, namely from higher to



Fig. 5 The experimental and theoretical 2D spectra of the LH3 complex, corresponding to the real part of electric field at 77 K at population times T = 0 fs, 20 fs, 50 fs, 1 ps, 2 ps, and 5 ps. The B800 and B820 peaks appear at $(\omega_{\tau} \sim 12450 \text{ cm}^{-})$ $\omega_t \sim 12450 \; {\rm cm}^{-1}) \; {\rm and}$ $(\omega_{\tau} \sim 12150 \text{ cm}^{-1})$. $\omega_t \sim 12150 \text{ cm}^{-1}$). respectively, in the T=0spectrum. All spectra are normalized to the absolute maximum; positive features correspond to "more light" and negative to "less light" (Zigmantas et al. 2006)



lower energy states. This type of cross peak can be observed in the energy funnelling processes of light harvesting. In reality, cross peaks in 2D spectra arise from multiple sources, and it may be difficult to distinguish between the limits of coherent and incoherent signals. In the case of LH3, the early-time spectra show faint off-diagonal signals (both positive and negative), and strong cross peaks are first observed at ~ 2 ps.

LH3 is a low-light adapted variant of the more common LH2 peripheral antenna complex, containing 27 BChla pigments arranged in two parallel rings, known as B820 and B800, due to their absorption wavelengths. Note that the B820 ring of LH3 discussed here is different from the solubilized dimer subunit of LH1, also called B820, discussed earlier. The 18 BChls of the B820 ring are closely packed, resulting in nearest-neighbor coupling interactions of about 300 cm⁻¹. In contrast, the nine BChls of the B800 ring are more widely spaced, coupled by only 30 cm⁻¹. These interactions and resulting dynamics are apparent in the 2D experimental spectral features. While no strong cross peaks are apparent at T = 0, the weak off-diagonal features, and in particular the above-diagonal negative signal, indicate coherent coupling in the LH3 complex. The effect of coherent coupling is more apparent in the lower energy (B820) peak, in that it is shifted further down off the diagonal relative to the B800 peak (as a result of interference with the above-diagonal negative feature) and it exhibits coherent dynamics within the first 50 fs, while the B800 peak remains unchanged. Still, the off-diagonal signal above the B800 peak shows that coherence effects are present even for the weakly coupled BChla ring: if coherent coupling were not present, the B800 peak would be perfectly centered on the diagonal. Thus, 2D spectra are exquisitely sensitive even to weak interactions between chromophores.

The theoretical spectra match the experimental results well, particularly the long-time dynamics in which strong cross peaks appear as energy transfers irreversibly from the B800 to B820 ring. In the T = 1 ps spectrum, a positive cross peak begins to appear, and by 5 ps no population is visible in the B800 band. For a detailed treatment of the theoretical methods used, see Brixner et al. (2004) and Zigmantas et al. (2006). The excellent match between the experimental and simulated spectra demonstrates that the model captures the energy level structure and general dynamical behavior of the LH3 complex. Furthermore, while the experimental spectra provide a wealth of information alone, the theoretical calculations give deeper insight to the energy transfer mechanisms at work. For example, the theoretical calculations showed energy transfer from the B800 band to dark, high-lying energetic states of the B820 band, a mechanism which increases the rate of energy transfer over that predicted by the traditional Förster theory of energy (Scholes and Fleming 2000).

The LH3 results hint at the importance of quantum coherence effects in photosynthetic light harvesting. A 2D experiment can also be devised to probe quantum coherence effects directly, in a manner related to the 2CECPE

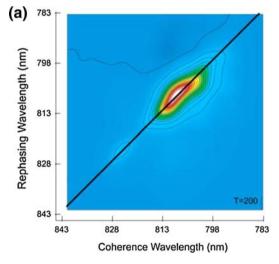


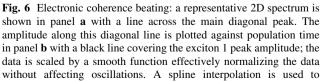
experiment, as first demonstrated by a study of the FMO complex at 77 K (Engel et al. 2007). In this experiment, 2D spectra are measured with smaller intervals between T time-points, such that rapid oscillations in signal amplitude are sampled. These oscillations result from the reversible, wavelike motion of quantum superposition states. The persistence of the oscillations (longer than 660 fs) indicates that the coherent nature of the electronic-excited states spanning multiple pigments is maintained for a surprisingly long time after laser excitation, whereas it was assumed previously that vibrational motions would destroy the electronic coherence within ~100 fs. Figure 6 demonstrates how taking slices through the 2D spectra of FMO from *Chlorobium tepidum* and lining them up in *T* reveals the oscillatory motion. The Fourier transform of the oscillations gives a beat spectrum, revealing the energy differences between coupled excitonic states giving rise to the quantum interference. As discussed above in the 2CECPE study of the bacterial reaction center, the quantum mechanical nature of energy transfer may be advantageous for more efficient sampling of a complex energy landscape in photosynthetic systems, as well as for robustness against trapping in local energetic minima.

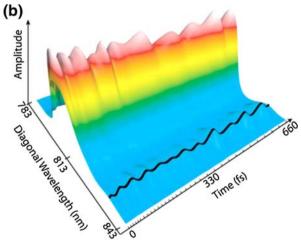
The above 2D experiments illustrate that key mechanistic information can be extracted from spectra measured using identical input pulses. Experimental strategies involving different polarizations of the laser pulses may be used to probe an increased, or more specific, set of interactions. Such an approach is analogous to linear and circular dichroism methods in one-dimensional spectroscopy (see Garab and Van Amerongen, this issue), except with

increased versatility as here four pulses can be independently controlled (Hochstrasser 2001; Zanni et al. 2001; Dreyer et al. 2003). The usefulness of rotating the pulse polarizations lies in the fact that in 2D spectra measured with parallel-polarized input pulses, diagonal peaks dominate the spectra (as evident in the T = 0 LH3 spectrum of Fig. 5), obscuring off-diagonal peaks that report on chromophore interactions. The first example of 2D electronic spectroscopy using polarization techniques to uncover weak signals was also in an application to the FMO complex (Read et al. 2007). Extending techniques applied in the infrared regime (Hochstrasser 2001; Zanni et al. 2001), the polarization sequence $(60^{\circ}, -60^{\circ}, 0^{\circ}, 0^{\circ})$ for pulses (1, 0)2, 3, LO) was used to completely eliminate the diagonal peaks from the spectrum of FMO from Pelodictyon phaeum (Fig. 7).

In addition to highlighting otherwise weak features, polarization techniques can be used to report on structures of photosynthetic complexes. The amplitude of a cross peak in a 2D spectrum, and the way that the amplitude changes with input pulse polarization, depends in part on the relative orientation between the coupled transition dipoles. In a measurement on the FMO complex from *Prosthecochloris aestuarii* using one set of pulse polarizations, a cross peak is only weakly visible in the spectrum at 400 fs (Fig. 8 (45°, -45°, 0°, 0°) spectrum), while under another polarization scheme (Fig. 8 (75°, -75°, 0°, 0°) spectrum) the cross peak appears strongly (Read et al. 2008). Based on these data in combination with theoretical calculations, we concluded that the relative angle between the two dominant transition dipoles in FMO is roughly 40°.







connect the spectra; the times at which spectra were taken are denoted by tick marks along the time axis. The oscillatory black curve shows coherent quantum beating of exciton 1. Figure reprinted with permission from Macmillan Publishers Ltd: Engel et al. (2007); Copyright 2007



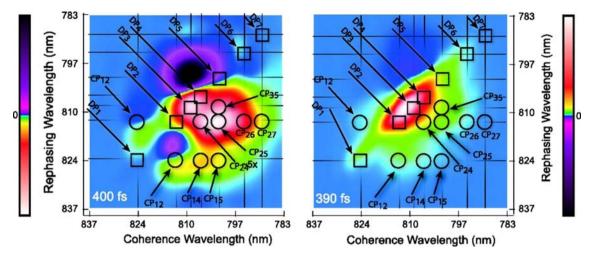


Fig. 7 The cross peak specific 2D and conventional 2D electronic spectra for FMO from *Pelodictyon phaeum*. The cross-peak specific spectrum reveals off-diagonal features obscured by the diagonal peaks in the conventional 2D spectrum. Both spectra are colored using a nonlinear ArcSinh coloration to emphasize smaller features, and the cross-peak specific coloration is inverted to facilitate direct visual comparison of the cross peaks to those in the conventional 2D

spectrum. Diagonal peaks (DP_i) are shown with squares while cross peaks (CP_{ij}) are denoted with circles. The shape of the edge of the cross peak regions agrees between the spectra, but significant additional structure is visible in the cross peak specific spectrum. Figure from Read et al. (2007); Copyright 2007, National Academy of Sciences, USA

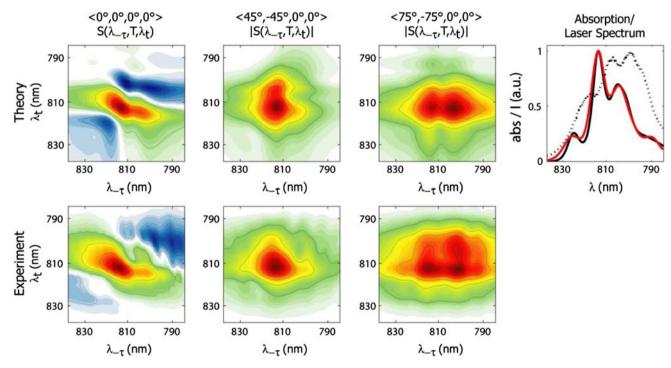


Fig. 8 Theoretical and experimental spectra of FMO from *Prosthecochloris aestuarii* at T=400 fs and 77 K (nonrephasing part, for details, see Read et al. 2008). Top row, left to right: theoretical <0°, 0°, 0°, 0°>, <45°, -45°, 0°, 0°>, and <75°, -75°, 0°, 0°> 2D spectra. The top right panel shows experimental and theoretical linear absorption spectra in black and red, respectively, and the dotted line is the laser spectrum of the pulses used to measure 2D spectra. Bottom

row, left to right: experimental $<0^{\circ}$, 0° , 0° , 0° , $<45^{\circ}$, -45° , 0° , 0° , and $<75^{\circ}$, -75° , 0° , 0° > 2D spectra. The differently polarized spectra show different cross peak amplitudes. In particular, a strong cross peak visible in the $<75^{\circ}$, -75° , 0° , 0° > spectrum is absent from the $<45^{\circ}$, -45° , 0° , 0° > spectrum. Figure reprinted with permission from *Biophysical Society*, Read et al. (2008); Copyright 2008

The experiment and corresponding theoretical calculations point to the way that 2D experiments can be designed to probe particular interactions in a multi-chromophore

system and thus yield detailed quantitative insight on the coupling strengths and relative orientations between transition dipole moments.



Conclusions

In summary, photon echo-based experiments may be designed to probe a number of aspects of photosynthetic light-harvesting complexes in detail, including coupling among pigments, coupling between pigments and the surrounding protein environment, contributions to spectral broadening, dynamical time scales, and mechanisms of energy transfer in light harvesting. Perhaps most exciting at this juncture is the recently realized capability of photon echo techniques to directly probe the quantum mechanical underpinnings of ultrafast energy transfer in photosynthesis, first discussed over 50 years ago (see review by Knox 1996), but elusive of direct experimental observation until now. The experiments described above demonstrate some of the experimental techniques that can be utilized to probe various aspects of light harvesting in detail. However, the flexibility of photon echo techniques means that a myriad of different experiments could be devised in addition to those outlined here in this review. From an experimental standpoint, the technological implementation of photon echo experiments is still in an early phase. While routine generation of sub-100 fs pulses has now been achieved, phase detection and control still present a problem for programmable pulse sequences, which would significantly aid in widespread applicability of these techniques. Thus, coming years will likely see rapid expansion of experimental methods related to those described here, and much is to be gained in our understanding of photosynthetic light harvesting from such developments.

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