Photochemistry in biomass burning plumes and implications for tropospheric ozone over the tropical South Atlantic

Denise L. Mauzerall, Jennifer A. Logan, Daniel J. Jacob, Bruce E. Anderson, Donald R. Blake, John D. Bradshaw, Brian Heikes, Glenn W. Sachse, Hanwant Singh, and Bob Talbot

Abstract. Photochemistry occurring in biomass burning plumes over the tropical south Atlantic is analyzed using data collected during the Transport and Atmospheric Chemistry Near the Equator-Atlantic aircraft expedition conducted during the tropical dry season in September 1992 and a photochemical point model. Enhancement ratios (ΔY/ΔX, where Δ indicates the enhancement of a compound in the plume above the local background mixing ratio, Y are individual hydrocarbons, CO, O₃, N₂O, HNO₃, peroxyacetyl nitrate (PAN), CH₃O, acetone, H₂O₂, CH₃OOH, HCOOH, CH₃COOH or aerosols and X is CO or CO₂) are reported as a function of plume age inferred from the progression of CH₄-methane hydrocarbons/ΔCO enhancement ratios. Emission, formation, and loss of species in plumes can be diagnosed from progression of enhancement ratios from fresh to old plumes. O₃ is produced in plumes over at least a 1 week period with mean ΔO₃/ΔCO = 0.7 in old plumes. However, enhancement ratios in plumes can be influenced by changing background mixing ratios and by photochemical loss of CO. We estimate a downward correction of -20% in enhancement ratios in old plumes relative to ΔCO to correct for CO loss. In a case study of a large persistent biomass burning plume at 4-km we found elevated concentrations of PAN in the fresh plume. The degradation of PAN helped maintain NOₓ mixing ratios in the plume where, over the course of a week, PAN was converted to HNO₃. Ozone production in the plume was limited by the availability of NOₓ, and because of the short lifetime of O₃ at 4-km, net ozone production in the plume was negligible. Within the region, the majority of O₃ production takes place in air above median CO concentration, indicating that most O₃ production occurs in plumes. Scaling up from the mean observed ΔO₃/ΔCO in old plumes, we estimate a minimum regional O₃ production of 17 x 10¹⁰ molecules O₃ cm⁻² s⁻¹. This O₃ production rate is sufficient to fully explain the observed enhancement in tropospheric O₃ over the tropical South Atlantic during the dry season.

1. Introduction

Biomass burning in the tropical dry season is widespread and is thought to be increasing [Malingreau and Tucker, 1988; Caboon et al., 1992]. Human activities, including the burning of savannas to improve grazing, the clearing of forests and bushland for agricultural use, the use of biomass for fuel, and the burning of agricultural wastes, are the primary cause of tropical biomass burning. Emissions from biomass burning have a major influence on the photochemistry and radiation balance of the troposphere [Crutzen et al., 1979; Crutzen and Andreae, 1990]. Emitted compounds include CO, CO₂, CH₄, nonmethane hydrocarbons (NMHCs), and nitrogen oxides (NOₓ = NO + NO₂). The photochemical oxidation of CO and hydrocarbons in the presence of NOₓ produces O₃. Tropospheric O₃ has both detrimental and advantageous effects. It is an oxidant which is damaging to vegetation and human health. In the upper troposphere it is an important greenhouse gas. However, it is also a key precursor of the hydroxyl radical (OH). OH is the primary oxidant in the troposphere and is responsible for the removal of reactive pollutants released into the atmosphere by anthropogenic and natural processes. Most oxidation of reactive long-lived gases takes place in the tropical troposphere where high insolation and humidity foster the formation of OH from the photolysis of ozone. Therefore an understanding of the influence of tropical biomass burning on tropospheric ozone production is crucial to understanding the
influence of tropical fires on the global oxidizing capacity of the atmosphere and on the radiative forcing of ozone in the upper troposphere.

The Transport and Atmospheric Chemistry Near the Equator-Atlantic (TRACE A) expedition was conducted during the 1992 southern tropical dry season. An overview of results from this expedition are described by Fishman et al. [1996] with detailed papers included in the section "Southern Tropical Atlantic Region Experiment (STARE): TRACE A and SAFARI" in Journal of Geophysical Research, 101 (D19), 23,514-24,330, 1996. TRACE A was designed to evaluate the contribution of biomass burning to the tropospheric ozone maximum observed in satellite data off the west coast of southern Africa [Fishman et al., 1990] and in seasonal enhancements of ozone concentrations measured with ozone sondes over Natal, Brazil [Logan and Kirchhoff, 1986]. Satellite data indicated that the magnitude of the tropospheric ozone enhancement over the South Atlantic during the burning season (August-October) is comparable to the enhancement observed in northern mid-latitudes during summer (June-August) [Fishman et al., 1990]. TRACE A was coordinated with the Southern African Fire Atmospheric Research Initiative (SAFARI) which used prescribed fires and ground and low-altitude aircraft flights to characterize emissions from biomass burning fires in Africa in September-October 1992 [Lindesay et al., 1996].

The TRACE A expedition made chemical and meteorological measurements during aircraft flights in the NASA DC-8 over Brazil, the South Atlantic, and southern Africa. Regions over active fires in Brazil and southern Africa, as well as continental outflow, biomass burning plumes of various ages, and clean regional background air over the south Atlantic were surveyed. Measurements in the southern hemisphere between September 22, 1992 and October 24, 1992 extended from 0°S to 38°S latitude, from 52°W to 40°E longitude, and from 0 to 12 km altitude. Flight paths are shown in Figure 1. Species measured aboard the aircraft included $O_3$, NO, peroxyacetyl nitrate (PAN), HNO$_3$, CO, CO$_2$, CH$_4$, N$_2$O, 35 different NMHCs, peroxides (H$_2$O$_2$ and CH$_3$OOH), formaldehyde (CH$_2$O), acetone (CH$_3$COCH$_3$), formic and acetic acids (HCOOH, CH$_3$COOH), aerosol density and size distribution, and several chlorinated compounds of exclusively industrial origin including C$_2$Cl$_4$, CH$_3$CCl$_3$, CFC-11, CFC-12, and CFC-113. Meteorological parameters, UV radiation, temperature and dew point were also measured.

Extensive research has been conducted in the field and laboratory to characterize emissions from biomass burning in both temperate [Cofer et al., 1989, 1990; Hegg et al., 1990; Laursen et al., 1992; Lobert et al., 1991; Yokelson et al., 1996] and tropical regions [Greenberg et al., 1984; Hurst et al, 1994a, 1994b; Bonsang et al., 1991, 1995; Lacaux et al., 1995; Helas et al., 1995; Rudolph et al., 1995; Cofer et al., 1996; Hao et al., 1996]. Efforts to determine enhancement ratios of tracers of biomass combustion have also been made from measurements collected during other large integrated aircraft expeditions such as Amazon Boundary Layer Experiment (ABLE) 2A in the dry season over the Brazilian rainforest [Andreae et al., 1988], Arctic Boundary Layer Expedition (ABLE) 3A over the Alaskan arctic in summer [Wofsy et al., 1992, 1994], Arctic Boundary Layer Expedition (ABLE) 3B over eastern Canada in summer [Blake et al., 1994; Wofsy et al., 1994; Mauzerall et al., 1996], Chemical Instrumentation Test and Evaluation (CITE) 3 during the dry season over the equatorial and tropical south Atlantic [Andreae et al., 1994], Dynamique et Chimie Atmospherique en Foret Equatoriale (DECAFE) over the Ivory Coast in the dry season [Bonsang et al., 1995; Delmas et al., 1995; Helas et al., 1995; Rudolph et al., 1995], SAFARI over southern African savannas [Ward et al., 1996; Hao et al., 1996; Lacaux et al., 1996; Cofer et al., 1996; Andreae et al., 1996] and TRACE A [Blake et al., 1996].
have examined the evolution of an idealized African biomass burning plume using a two-dimensional (2-D) model. Although other work has examined aged biomass burning emissions and their impact on the TRACE A region [Browell et al., 1996; Gregory et al., 1996; Singh et al., 1996; Smyth et al., 1996; Talbot et al., 1996; Thompson et al., 1996] less work has focused on how biomass burning emissions evolve in plumes as they photochemically age, or on the contribution of O₃ production in plumes to the regional O₃ enhancement. In this paper we address these issues.

This paper is organized into five sections. In section 2 we describe a method of classifying plumes by photochemical age and we examine the temporal evolution of enhancement ratios (ΔY/ΔX) of photochemical tracers in the TRACE A plumes. We also discuss the influence of changing background concentrations and photochemical loss of CO on enhancement ratios. Section 3 is a case study of a large biomass burning plume sampled off the west coast of Africa. In section 4 we examine the influence that the dispersal of biomass burning plumes has on regional ozone production. Results of our analysis are summarized in section 5.

2. Data Analysis
2.1. Identification of Plumes and Determination of Enhancement Ratios

We identified biomass burning plumes sampled during the TRACE A expedition in time series plots of CO mixing ratio using the criteria that a plume exhibit an enhancement of at least 20 ppbv CO relative to local background. Concurrent enhancements in other emitted species including CO₂ and C₂H₂ were used for confirmation. Enhancements in halocarbon concentrations (e.g. C₂Cl₄ and CH₃CCl₃) identified urban pollution plumes (flight 5 along the east coast of Brazil and flight 8 returning to Rio de Janeiro), and these plumes were discarded. For each plume, least square regressions of every species were calculated relative to CO and CO₂, and the slope of each regression was obtained using the reduced-major-axis method [Hirsch and Gilroy, 1984]. The slope provides the enhancement ratio in the plume relative to the local background (ΔY/ΔX). Statistically significant regressions were diagnosed using a t-test with 95% confidence limits. In order to exclude regressions which appeared statistically significant due to a large number of observations but which actually included two separate populations r² > 0.3 was also required.

We based our analyses on merged time series of aircraft measurements where all the data were averaged over a common time interval. In TRACE A the frequency and time resolution of measurements varied considerably. Measurements for some species were reported with 5-second resolution (CO, CO₂, CH₄, and N₂O), while averaging times were longer for other species (10 seconds for O₃, 60 seconds for CH₂O, 90 seconds for NO, 130 seconds for acetone, and 150 seconds for PAN). For still other species the frequency of measurements and the sampling interval were variable (NMHCs and halocarbons, H₂O₂, and HNO₃). We chose to average the data over the measurement time interval of the critical species with coarsest time resolution. Depending on the question we are addressing, the critical species changes. We therefore use eight different merges matching the sampling time interval of particular species: 10-second merge (CO, CO₂, CH₄, N₂O, and O₃), 60-second merge (CH₂O), 90- or 180-second merge (NO), 130-second merge (acetone), 150-second merge (PAN), and three variable-time merges (NMHCs, HNO₃, H₂O₂).

2.2. Plume Classification and Locations

The temporal evolution of enhancement ratios is a function of both photochemistry and plume dilution [McKeen and Liu, 1993]. NMHCs provide a diagnostic of the relative age of plumes, with significant correlations between short-lived NMHCs and CO present only in relatively fresh plumes and correlations between longer-lived NMHCs and CO being maintained in older plumes. We classified biomass burning plumes into four age categories using the presence or absence of enhancement ratios for 10 NMHCs with a range of lifetimes (as shown in Table 1) to CO. Table 2 gives mean enhancement ratios relative to CO (ANMHC/ΔCO) for hydrocarbons used to assign relative plume ages.

### Table 1. Typical Lifetimes of O₃ and Nonmethane Hydrocarbons (NMHCs) in Transport and Atmospheric Chemistry Near the Equator-Atlantic (TRACE A)

<table>
<thead>
<tr>
<th>Species</th>
<th>Lifetime, a days</th>
</tr>
</thead>
<tbody>
<tr>
<td>O₃</td>
<td>6</td>
</tr>
<tr>
<td>Isoprene</td>
<td>0.04</td>
</tr>
<tr>
<td>1-Butene</td>
<td>0.09</td>
</tr>
<tr>
<td>Propene</td>
<td>0.15</td>
</tr>
<tr>
<td>Ethene</td>
<td>0.5</td>
</tr>
<tr>
<td>Toluene</td>
<td>0.7</td>
</tr>
<tr>
<td>n-Butane</td>
<td>1.8</td>
</tr>
<tr>
<td>Benzene</td>
<td>3.5</td>
</tr>
<tr>
<td>Propane</td>
<td>4.3</td>
</tr>
<tr>
<td>Acetylene</td>
<td>6.5</td>
</tr>
<tr>
<td>Ethane</td>
<td>20</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Temperature, K</th>
<th>0-4 km</th>
<th>4-8 km</th>
<th>8-12 km</th>
</tr>
</thead>
<tbody>
<tr>
<td>OH, a molecules cm⁻³</td>
<td>2.7 x 10⁶</td>
<td>1.6 x 10⁶</td>
<td>1.2 x 10⁶</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Species</th>
<th>Lifetime, a days</th>
</tr>
</thead>
<tbody>
<tr>
<td>O₃</td>
<td>25</td>
</tr>
<tr>
<td>Isoprene</td>
<td>0.05</td>
</tr>
<tr>
<td>1-Butene</td>
<td>0.13</td>
</tr>
<tr>
<td>Propene</td>
<td>0.25</td>
</tr>
<tr>
<td>Ethene</td>
<td>0.9</td>
</tr>
<tr>
<td>Toluene</td>
<td>1.0</td>
</tr>
<tr>
<td>n-Butane</td>
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</tr>
<tr>
<td>Benzene</td>
<td>6.3</td>
</tr>
<tr>
<td>Propane</td>
<td>8.9</td>
</tr>
<tr>
<td>Acetylene</td>
<td>13</td>
</tr>
<tr>
<td>Ethane</td>
<td>48</td>
</tr>
</tbody>
</table>

a Mean values calculated for TRACE A conditions as determined by Jacob et al. [1996].

b Lifetimes were calculated on the basis of oxidation by OH only.
Table 2. Mean Hydrocarbon Enhancement Ratios in Plumes (pptv/ppbv)

<table>
<thead>
<tr>
<th></th>
<th>Fresh Plumes (11)</th>
<th>Recent Plumes (10)</th>
<th>Aged Plumes (16)</th>
<th>Old Plumes (14)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean Error Plumes</td>
<td>Mean Error Plumes</td>
<td>Mean Error Plumes</td>
<td>Mean Error Plumes</td>
</tr>
<tr>
<td>ΔIsoprene / ΔCO</td>
<td>0.29 ± 0.031</td>
<td>NA</td>
<td>0</td>
<td>NA</td>
</tr>
<tr>
<td>Δ1-Butene / ΔCO</td>
<td>0.24 ± 0.12</td>
<td>NA</td>
<td>0</td>
<td>NA</td>
</tr>
<tr>
<td>ΔPropene / ΔCO</td>
<td>1.18 ± 0.47</td>
<td>NA</td>
<td>0</td>
<td>NA</td>
</tr>
<tr>
<td>ΔEthene / ΔCO</td>
<td>6.88 ± 1.34</td>
<td>1.32 ± 0.27</td>
<td>7</td>
<td>NA</td>
</tr>
<tr>
<td>ΔToluene / ΔCO</td>
<td>0.67 ± 0.10</td>
<td>0.24 ± 0.11</td>
<td>1</td>
<td>NA</td>
</tr>
<tr>
<td>Δn-Butane / ΔCO</td>
<td>0.33 ± 0.11</td>
<td>0.20 ± 0.09</td>
<td>3</td>
<td>0.23 ± 0.10</td>
</tr>
<tr>
<td>ΔBenzene / ΔCO</td>
<td>1.30 ± 0.06</td>
<td>0.81 ± 0.054</td>
<td>8</td>
<td>0.48 ± 0.048</td>
</tr>
<tr>
<td>ΔPropane / ΔCO</td>
<td>2.35 ± 0.82</td>
<td>0.97 ± 0.15</td>
<td>4</td>
<td>1.22 ± 0.43</td>
</tr>
<tr>
<td>ΔC2H2 / ΔCO</td>
<td>4.22 ± 0.37</td>
<td>3.40 ± 0.47</td>
<td>8</td>
<td>2.62 ± 0.15</td>
</tr>
<tr>
<td>ΔC2H6 / ΔCO</td>
<td>7.53 ± 0.63</td>
<td>5.87 ± 0.69</td>
<td>5</td>
<td>7.07 ± 1.07</td>
</tr>
</tbody>
</table>

The total number of plumes in a given age category is in parentheses. Enhancement ratios in each plume category include the mean and the standard error (σ/μ).

*a The number of plumes with statistically significant correlations. When that number is zero, NA is entered for the mean and standard error.

for each of the four age categories. For a particular ΔNMHC/ΔCO a systematic decrease in slope from one age category to the next is usually observed because of the more rapid disappearance of the NMHC than of the CO. As plumes age, enhancements of short-lived hydrocarbons progressively disappear. Significant enhancements of at least one NMHC with a lifetime at low altitudes of <0.2 days (isoprene, 1-butene, or propene) was required to categorize a plume as fresh; between 0.5 and 1-day (ethene or toluene) was required to categorize a plume as recent; between 2 and 5 days (n-butane, benzene, or propane) was required to categorize a plume as aged; and > six days (acetylene or ethane) was required to categorize a plume as old. The ages of individual plumes were also estimated independently from advanced very high resolution radiometer (AVHRR) satellite imagery of fire locations in combination with meteorological back trajectories. Fresh plumes were typically less than half a day old, recent plumes were < 1 day old; aged plumes were < five days old; and old plumes were < 1 week old.

Figure 2 presents examples of a plume in each age category ((a) fresh, (b) recent, (c) aged, and (d) old). For each age category a time series of selected data from a flight segment is shown. Vertical lines on the time series indicate the interval identified as containing a plume. Scatter plots of the data within the time interval of the plume are shown with superimposed lines indicating a significant correlation between the plotted species. Slopes of these lines provide enhancement ratios.

The locations of individual plumes are shown on the regional map in Figure 3a, while the altitude of the CO maximum of individual plumes is shown versus longitude in Figure 3b. Fresh plumes are directly over source regions and are typically within the boundary layer. Recent plumes are also usually directly over the source regions but are generally within the free troposphere. Aged plumes can be found over both the continents and the Atlantic at any altitude but are usually in the free troposphere. Old plumes are found remote from continental source regions in the free troposphere.

2.3. Analysis of Enhancement Ratios

Previous analyses of biomass burning have characterized biomass burning emissions by determining the ratio of ΔY/ΔX (where Y is a compound either emitted by the fire or produced in the fire plume, X is a reference tracer assumed to be conserved in the plume (usually CO or CO2), and Δ indicates the mixing ratio of a compound in the plume minus the mixing ratio in local background air). The chemical composition of the plume is characterized by the regression slope of Y versus X. Emission ratios of this type have been used in combination with estimates of the amount of biomass burned to quantify the contribution of biomass burning to trace gas budgets for CO, CH4, NOx, and other gases [Crutzen et al., 1979; Crutzen and Andreae, 1990; Laursen et al., 1992]. A similar approach has also been employed to estimate photochemical production of ozone by assuming that ΔO3/ΔCO indicates the number of ozone molecules produced per CO molecule emitted during combustion [Parrish et al., 1993, Andreae et al., 1994]. In this study we examine the validity of this approach.
Tables 3a and 3b present enhancement ratios relative to CO and CO$_2$ for a number of species: O$_3$, CH$_4$, N$_2$O, CH$_3$Cl, HNO$_3$, PAN, CH$_2$O, (CH$_3$)$_2$CO, H$_2$O$_2$, CH$_3$OOH, HCOOH, CH$_3$COOH, and aerosols. We examine enhancement ratios versus both CO and CO$_2$ because each compound has particular advantages and disadvantages. CO$_2$ is chemically inert in the atmosphere but exchanges with the biosphere; it also has a large background concentration which can make small enhancements difficult to detect above natural variability. CO, emitted primarily during smoldering combustion, is an attractive reference because it correlates well with other partially oxidized and reduced compounds emitted during similar combustion regimes. It is not taken up by vegetation, and its background mixing ratio is 3 orders of magnitude lower than CO$_2$, making detection of combustion plumes easier. However, CO is not chemically inert.

We observe that as plumes age $\Delta$CO/$\Delta$CO$_2$ decreases, with the most appreciable decrease between fresh and recent aged plumes (see Table 3b). There are two possible explanations for this decrease. First, background mixing ratios are not the same in all plume age categories (see Table 4). Second, there is photochemical loss of CO over the lifetime of the plume. These two factors have implications for all enhancement ratios and will be discussed, in turn, below.

2.3.1. Implications of changing background mixing ratios. A limitation on the use of enhancement ratios is imposed by variable background mixing ratios. In environments where background mixing ratios vary from one location to another, enhancement ratios of a particular pair of species relative to their local background mixing ratios vary as a plume dilutes [McKeen and Liu, 1993]. To illustrate this point, we show in Figure 4a the schematic of a fire plume containing two inert species with concentrations $(x',y')$ and diluting with background air of composition $(x_b',y_b')$ that is different from the background air at the point of emission $(x_b,y_b)$ in which the fresh plume $(x,y)$ formed. Let $R$ be the emission ratio of $y$ to $x$ in the fire. Linear regression of concentration measurements within the fresh plume
yields a slope of $R = \Delta y / \Delta x$ representing the emission ratio from the fire. However, as the plume dilutes with background air of changing composition, the slope will change (Figure 4b), biasing the interpretation of enhancement ratios in the plume relative to the emission ratio. For all cases of interest here, with $x_b - x_4' > 0$ and $y_b - y_4' > 0$, the bias is negative (i.e., $(y_b - y_4') / (x_b - x_4') < R$). This largely explains the dramatic decrease in $\Delta CO / \Delta CO_2$ from the fresh to the recent plume age categories.

### 2.3.2. Chemical loss of CO

Photochemical loss of CO is also responsible, in part, for the decrease in $\Delta CO / \Delta CO_2$ as plumes age. We calculated CO lifetimes (Figure 5) using mean model OH values reported for the TRACE A conditions at different altitudes by Jacob et al. [1996]. Results indicate a CO lifetime of about 20 days below 6-km altitude, rising to 60 days at 12 km. Below 6-km the lifetime of CO is short enough that over a 1 week transit time, one-third of the CO in a plume may be lost. Thus photochemical loss of
CO contributes to a decrease in ΔCO/ΔCO₂ as well as to an upward bias on enhancement ratios of ΔX/ΔCO (where X is any chemical species).

2.4. Photochemical Evolution of Plumes Determined From Enhancement Ratios

Through an examination of the mean enhancement ratios relative to CO and CO₂ shown in Table 3 we can gain insight into which species are emitted from the fire and which are produced through oxidation processes in the plume. Species which exhibit strong correlations with CO in fresh plumes and weaker correlations as plumes age and which have enhancement ratios which decrease as plumes age are presumably emitted from the fire or formed in the fresh plume. Species which show increasing correlation with CO and enhancement ratios which increase more rapidly than can be explained by the loss of CO as plumes age are presumably produced gradually in the plumes.

2.4.1. Ozone. For the TRACE A region, as shown in Table 3, ΔO₃/ΔCO increases from 0.15 to 0.32 to 0.71 to 0.74 ppbv/ppbv in fresh, recent, aged, and old plumes, respectively. These values can be compared to previously reported biomass burning ΔO₃/ΔCO values of 0.056 ppbv/ppbv in the Amazon basin, of 0.06 ppbv/ppbv in North America and of 0.46 ppbv/ppbv over the South Atlantic [Andreae et al., 1994]. However, this study is the first time that plumes have been segregated on the basis of age. The range observed here is similar to that reported by Andreae et al. [1994]. Photochemical loss of CO in the plumes causes ΔCO to be smaller and ΔO₃/ΔCO to be larger than they would be if CO were an inert tracer. Aged plumes at lower altitudes may exhibit a ΔO₃/ΔCO as much as 30% higher than would have been observed were CO inert. At higher altitudes, because of lower OH concentrations, CO loss is not as rapid, and the correction is less. Changes in the background O₃ and CO mixing ratios conversely cause a downward bias on ΔO₃/ΔCO as plumes age from fresh to recent categories, but because CO enhancements in fresh plumes are so large relative to the change in the background, the downward bias will be small, and the influence of changes in the background become neg-
ligible when considering further aging (see Table 4). We therefore estimate a downward correction of 20% in $\Delta O_3/\Delta CO$ in old plumes because of the photochemical loss of CO.

2.4.2. NOx. We calculate $\Delta NO_x$ in the plumes by combining measurements of NO with NO2 values calculated using a point model (see Jacob et al., [1996] for a description of the model). The $\Delta NO_x/\Delta CO$ ratio in fresh plumes of 5.2 pptv/ppbv is similar to the 3.4 pptv/ppbv ratio observed previously in boreal biomass fires over Alaska [Wofsy et al., 1992] but is low relative to the range 2 - 50 pptv/ppbv reported in the review by Crutzen and Andreae [1990]. The $\Delta NO_x/\Delta CO$ ratio in recent, aged, and old plumes remains relatively constant near 3 pptv/ppbv, possibly sustained by decomposition of PAN to NOx and subsequent conversion of NOx to HNO3. Because of relatively sparse measurements and the aggregation of plumes which have travelled at a variety of altitudes, the progression of $\Delta PAN/\Delta CO$ and $\Delta HNO_3/\Delta CO$ do not reflect the conversion of PAN to HNO3. However, the case study we discuss in section 3 exemplifies this conversion.

2.4.3. Methane. CH4 is emitted from fires and because of its long atmospheric lifetime is conserved over the lifetime of the plumes. Values of $\Delta CH_4/\Delta CO$ slowly increase as the plumes age because of loss of CO, while $\Delta CH_4/\Delta CO_2$ remains effectively constant over the lifetime of the plumes. However, as shown in Figure 6a, median mixing ratios of CH4 increase with altitude in the TRACE A region at a rate of 2.3 ppbv/km [Bartlett et al., 1996]. Krishnamurti et al. [1993] estimate a regional descent rate of upper tropospheric air to 900 mb of about 5 days. Thus northern hemisphere air carried to the upper troposphere in the Hadley cell circulation and subsiding in the southern hemisphere tropics likely contributes to the upper tropospheric enhancement in CH4. In addition, we calculate a median photochemical lifetime for CH4 which increases with altitude (Figure 6b). Despite
rapid downward mixing of upper tropospheric air in this region, loss rates of 2 ppbv/day in the lower troposphere likely also contribute to the observed vertical gradient of CH4 mixing ratio by providing a larger sink at lower altitudes.

2.4.4. Formaldehyde. Examining enhancement ratios for the four plume age categories, we find the largest fraction of significant correlations (55% of plumes) of ΔCH2O/ΔCO in fresh plumes and the smallest fraction (7%) in old plumes. In addition, there is a progressive decrease in the slope of the mean enhancement ratios as the plumes age from 0.0095 in fresh plumes to 0.0009 in old plumes. This indicates that CH2O is emitted from the fire or formed shortly thereafter from the oxidation of short-lived hydrocarbons and then progressively lost as the plumes age. Direct emission of CH2O in biomass burning is confirmed by Lee et al. [1997].

2.4.5. Peroxides. Examining enhancement ratios (Table 3) for the four plume age categories, we do not find as clear a signature for initial production of peroxides within the fire as we did for CH2O. Only 33% of the correlations in fresh plumes were significant for H2O2, and none were significant for CH3OOH. The fraction of plumes in the recent and aged categories which exhibited significant correlations for ΔH2O2/ΔCO and ΔCH3OOH/ΔCO never exceeded 30% and were < 10% in old plumes. The lifetimes of the peroxides against photolysis and reaction with OH are typically < 1 day (see section 3). The fact that significant correlations did not entirely disappear indicates that the net photochemical production of peroxides in the plumes was sufficient to maintain H2O2 mixing ratios enhanced over the background and to produce CH3OOH.

2.4.6. Acetone. A biomass burning source of acetone was identified in boreal biomass fires by Singh et al. [1994], and when all TRACE A acetone measurements are plotted versus CO, a strong positive correlation results, indicating a regional biomass burning source of acetone (see Figure 7). However, examining enhancement ratios (Table 3) for the four plume age categories, we find no significant enhancements in fresh plumes and significant enhancement ratios in
### Table 3a. Mean Photochemical Enhancement Ratios Relative to CO in Plumes

<table>
<thead>
<tr>
<th></th>
<th>Fresh Plumes (11)</th>
<th>Recent Plumes (10)</th>
<th>Aged Plumes (16)</th>
<th>Old Plumes (14)</th>
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</thead>
<tbody>
<tr>
<td></td>
<td>Mean</td>
<td>Standard Error</td>
<td>Significant Plumes&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Mean</td>
</tr>
<tr>
<td>ΔO&lt;sub&gt;3&lt;/sub&gt; / ΔCO</td>
<td>0.15 ± 0.37</td>
<td>4</td>
<td></td>
<td>0.32 ± 0.76</td>
</tr>
<tr>
<td>ΔCH&lt;sub&gt;4&lt;/sub&gt; / ΔCO</td>
<td>0.16 ± 0.05</td>
<td>10</td>
<td></td>
<td>0.37 ± 0.003</td>
</tr>
<tr>
<td>ΔN&lt;sub&gt;2&lt;/sub&gt;O / ΔCO</td>
<td>0.002</td>
<td>1</td>
<td></td>
<td>0.022</td>
</tr>
<tr>
<td>ΔCH&lt;sub&gt;3&lt;/sub&gt;Cl / ΔCO</td>
<td>0.77 ± 0.08</td>
<td>8</td>
<td></td>
<td>0.85 ± 0.04</td>
</tr>
<tr>
<td>ΔNO&lt;sub&gt;x&lt;/sub&gt;&lt;sup&gt;b&lt;/sup&gt; / ΔCO</td>
<td>5.2 ± 0.64</td>
<td>4</td>
<td></td>
<td>3.8 ± 2.8</td>
</tr>
<tr>
<td>ΔHNO&lt;sub&gt;3&lt;/sub&gt; / ΔCO</td>
<td>2.2 ± 0.49</td>
<td>2</td>
<td></td>
<td>5.8</td>
</tr>
<tr>
<td>ΔPAN / ΔCO</td>
<td>6.0 ± 1.7</td>
<td>3</td>
<td></td>
<td>7.5 ± 1.6</td>
</tr>
<tr>
<td>ΔCH&lt;sub&gt;2&lt;/sub&gt;O / ΔCO</td>
<td>9.5 ± 0.65</td>
<td>6</td>
<td></td>
<td>1.8 ± 0.81</td>
</tr>
<tr>
<td>ΔAcetone / ΔCO</td>
<td>NA</td>
<td>0</td>
<td></td>
<td>10.3 ± 1.9</td>
</tr>
<tr>
<td>ΔH&lt;sub&gt;2&lt;/sub&gt;O&lt;sub&gt;2&lt;/sub&gt; / ΔCO</td>
<td>11.8 ± 0.6</td>
<td>2</td>
<td></td>
<td>68.0 ± 44.9</td>
</tr>
<tr>
<td>ΔCH&lt;sub&gt;2&lt;/sub&gt;OOH / ΔCO</td>
<td>2.5</td>
<td>1</td>
<td></td>
<td>15.5 ± 6.8</td>
</tr>
<tr>
<td>ΔHCOOH / ΔCO</td>
<td>14.7 ± 4.6</td>
<td>3</td>
<td></td>
<td>43.1 ± 3.6</td>
</tr>
<tr>
<td>ΔCH&lt;sub&gt;3&lt;/sub&gt;COOH / ΔCO</td>
<td>16.5 ± 1.6</td>
<td>4</td>
<td></td>
<td>106.5</td>
</tr>
<tr>
<td>Δ (0.12-3.12um Aerosols) / ΔCO</td>
<td>18.2 ± 1.85</td>
<td>10</td>
<td></td>
<td>14.8 ± 3.25</td>
</tr>
</tbody>
</table>

Units are ppbv/ppbv for ΔO<sub>3</sub>/ΔCO, ΔCH<sub>4</sub>/ΔCO and ΔN<sub>2</sub>O/ΔCO and are pptv/ppbv for all other ratios. The total number of plumes in a given age category is in parentheses in the top row. Enhancement ratios in each plume category include the mean and the standard error (σ/√n).

<sup>a</sup>The number in parentheses gives the total number of plumes with statistically significant correlations; when that number is zero, NA is entered for the mean and standard error.

<sup>b</sup>NO<sub>x</sub> = (NO)<sub>measured</sub> + (NO<sub>2</sub>)<sub>calculated</sub>. NO<sub>2</sub> is calculated for every point in the plume using a point model and assuming diurnal steady state [Jacob et al., 1996].
Table 3b. Mean Photochemical Enhancement Ratios Relative to CO₂ in Plumes

<table>
<thead>
<tr>
<th></th>
<th>Fresh Plumes (11)</th>
<th>Recent Plumes (10)</th>
<th>Aged Plumes (16)</th>
<th>Old Plumes (14)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean ± Error</td>
<td>Significant Plumes</td>
<td>Mean ± Error</td>
<td>Significant Plumes</td>
</tr>
<tr>
<td>ΔCO / ΔCO₂</td>
<td>47.4 ± 6.4</td>
<td>10</td>
<td>23.2 ± 2.0</td>
<td>10</td>
</tr>
<tr>
<td>ΔCH₄ / ΔCO₂</td>
<td>4.6 ± 1.2</td>
<td>8</td>
<td>9.0 ± 1</td>
<td>1</td>
</tr>
<tr>
<td>ΔO₃ / ΔCO₂</td>
<td>5.1 ± 2.0</td>
<td>4</td>
<td>8.6 ± 1.1</td>
<td>3</td>
</tr>
<tr>
<td>ΔN₂O / ΔCO₂</td>
<td>0.14 ± 0.063</td>
<td>4</td>
<td>0.014 ± 0.02</td>
<td>1</td>
</tr>
<tr>
<td>ΔNO₂ / ΔCO₂</td>
<td>0.1 ± 0.063</td>
<td>1</td>
<td>0.17 ± 0.02</td>
<td>1</td>
</tr>
<tr>
<td>ΔPAN / ΔCO₂</td>
<td>0.25 ± 0.14</td>
<td>3</td>
<td>0.18 ± 0.014</td>
<td>3</td>
</tr>
<tr>
<td>ΔCH₂O / ΔCO₂</td>
<td>0.57 ± 0.056</td>
<td>4</td>
<td>0.044 ± 0.02</td>
<td>3</td>
</tr>
<tr>
<td>ΔAcetone / ΔCO₂</td>
<td>NA</td>
<td>0</td>
<td>0.18 ± 0.01</td>
<td>1</td>
</tr>
<tr>
<td>ΔH₂O₂ / ΔCO₂</td>
<td>0.55 ± 0.13</td>
<td>2</td>
<td>1.91 ± 1.14</td>
<td>2</td>
</tr>
<tr>
<td>ΔCH₃OOH / ΔCO₂</td>
<td>0.42 ± 0.29</td>
<td>3</td>
<td>0.40 ± 0.21</td>
<td>3</td>
</tr>
<tr>
<td>ΔHCOOH / ΔCO₂</td>
<td>0.64 ± 0.21</td>
<td>4</td>
<td>0.99 ± 0.37</td>
<td>2</td>
</tr>
<tr>
<td>ΔCH₃COOH / ΔCO₂</td>
<td>0.79 ± 0.23</td>
<td>4</td>
<td>NA ± 0</td>
<td>0</td>
</tr>
<tr>
<td>Δ(0.12- 3.12 ūm</td>
<td>0.86 ± 0.15</td>
<td>6</td>
<td>0.3 ± 0.11</td>
<td>4</td>
</tr>
</tbody>
</table>

* Units are ppbv/ppmv for ΔCO/ΔCO₂, ΔCH₄/ΔCO₂ and ΔN₂O/ΔCO₂ and are pptv/ppmv for all other ratios. See Table 3a for further explanations.
Table 4. Mean Background Mixing Ratios

<table>
<thead>
<tr>
<th></th>
<th>Fresh Plumes (11)</th>
<th>Recent Plumes (10)</th>
<th>Aged Plumes (16)</th>
<th>Old Plumes (14)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CO, ppbv</td>
<td>139 ± 11</td>
<td>87 ± 10</td>
<td>79 ± 3.5</td>
<td>83 ± 2.9</td>
</tr>
<tr>
<td>CO₂, ppmv</td>
<td>356 ± 0.4</td>
<td>354 ± 0.4</td>
<td>354 ± 0.3</td>
<td>352.8 ± 0.2</td>
</tr>
<tr>
<td>O₃, ppbv</td>
<td>49 ± 3.5</td>
<td>44 ± 4</td>
<td>41 ± 4.5</td>
<td>49 ± 4</td>
</tr>
</tbody>
</table>

The total number of plumes in a given age category is in parentheses. Mean background mixing ratios and the standard error (σ/N₁) are shown for every age category for each compound. Background mixing ratios were obtained by averaging the local minimum concentrations observed on either side of the plumes within each age category.

only about 25% of the older age categories, reflecting, we believe, the relatively sparse acetone measurements. The data suggest that acetone is formed as a secondary product in the plume, possibly from the oxidation of propane (see section 3).

3. African Plume Case Study

On October 14-18 the TRACE A aircraft sampled a large biomass burning plume transported from the west coast of Africa to the Atlantic Ocean. The plume extended from the top of the marine boundary layer (1.5 km altitude) to -5.5 km. Flights 13 and 14 were designed to sample this plume 1 day downwind of each other. Flight 15 sampled the same plume again 3 days after flight 14. AVHRR data for October 12, 1992 (Plate 1) and 5 day back trajectories for flights 13 (Figure 8a), 14 (Figure 8b) and 15 (Figure 8c) indicate that air parcels for these flights last passed over burning areas on October 12, 1992, 2, 3, and 5 days before they were intercepted by the aircraft. Although the origin of the emissions sampled in mission 10 (used to initialize the calculation) was east of the origin of the emissions sampled in missions 13, 14, and 15, the African ecosystem at this latitude is primarily grassy and woody savanna [Lavenu, 1987], and drought conditions prevailed in both areas. We therefore assume that the time since emission is the most critical factor determining plume composition.

3.1. Evolution of NOₓ Partitioning

In Plate 2 we show vertical profiles of CO, HNO₃, PAN, and NOₓ mixing ratios in the plume sampled during mis-

![Figure 4a](image)

Figure 4a. Schematic of plume formation and transport. The (x₀, y₀) and (xᵢ, yᵢ) indicate mixing ratios within the fresh plume at the time of sampling and in background air in proximity to the fire, respectively. The (xᵢ', yᵢ') and (xᵢ', yᵢ') indicate mixing ratios within the plume sampled by the aircraft and in background air in the vicinity of the sampled plume, respectively.

![Figure 4b](image)

Figure 4b. Schematic of plume sampling. Long-dashed line indicates a fresh plume with enhancements (xᵢ', yᵢ') relative to background mixing ratios near the fire (x₀, y₀). Dotted line indicates a plume remote from the fire which the aircraft measures relative to local background mixing ratios (xᵢ', yᵢ'). Enhancement ratios are obtained from the slopes of these lines.
sions 10, 13, 14, and 15 (M10, M13, M14, and M15). We use measured NO mixing ratios and calculate NO2 mixing ratios using a point model (see Jacob et al., [1996] for model details) to obtain NOx values. In the fresh emissions of M10 (Plate 2a), PAN is the dominant form of NOy with a maximum mixing ratio of 4000 pptv, ~25% larger than the 3000 pptv mixing ratio of NOx and 4 times greater than the 1200 pptv mixing ratio of HNO3. The dominance of PAN as an immediate product of NOx oxidation here contrasts with fossil fuel combustion where HNO3 usually accounts for a larger fraction of NOy [Parrish et al., 1993]. After 2 days of aging at 4 km, in M13 (Plate 2b) maximum PAN mixing ratios of 1500 pptv are 3 times the 500 pptv mixing ratio of HNO3. NOx mixing ratios have dropped to ~100 pptv which are maintained in the plume in M14 and M15 as the plume ages. After 3 days, in M14 (Plate 2c), HNO3 has become the dominant form of NOy at 1000 pptv and maximum mixing ratios of PAN have decreased to 700 pptv. After 5 days, in M15 (Plate 2d), HNO3 mixing ratios are 600 pptv, and PAN has decreased to 220 pptv. The data thus indicate rapid conversion of emitted NOx to PAN followed by relatively slow conversion of PAN to NO2 and on to HNO3 as the plume ages. The effect of rapid PAN formation followed by slow decomposition is to increase the effective lifetime of NOx, thus helping sustain photochemistry in the plume as it ages. The decrease in total NOy in the plume as it ages is a result of dilution.

3.2. Modeling

A lagrangian plume model is used to simulate the chemical evolution of this African plume. The model is initialized with mixing ratios determined from the fresh plume sampled during mission 10 at 4-km altitude and ~11°S latitude and 30°E longitude. The model uses the chemical mechanism described by Jacob et al. [1996] to simulate the chemical evolution of the plume over 1 week at 4-km altitude. Photolysis rates are enhanced by 50% over clear-sky conditions, as indicated by observations, because of persistent stratus cover over the ocean. The ozone column is 7.6 x 1018 molecules cm-2, the mean of values observed above the plume during missions 13, 14, and 15. Initial conditions and background air composition assumed for the calculation are given in Table 5.

Simulations both with and without horizontal plume dilution are conducted. Although the diluting case is more realistic, we use the nondiluting case to examine photochemical influences isolated from the influence of dilution. The width $Y(t)$ of the diluting plume at time $t$ is computed following Sillman et al. [1990]:

$$Y(t) = \sqrt{Y^2 + 8K_y t}$$

where $Y(0)$ is the width of the fresh fire plume assumed to be pumped to 4-km altitude at time $t = 0$, and $K_y$ is a constant cross-flow diffusion coefficient. We assume $Y(0) = 100$ km, based on AVHRR data for areal extent of fires. In the diluting plume scenario, we choose $K_y = 5 \times 10^4$ m2 s-1 to match the observed decrease of CO2. This value of $K_y$ is consistent...
with values recommended by Gifford [1982] for plume widening calculations. Concentrations in the diluting plume ($C_p$) are calculated as mixing ratios and are updated at hourly time steps $[t_n, t_{n+1}]$, by entrainment of background air ($C_b$) as shown in equations (2) and (3).

$$C_p(t_{n+1}) = C_p(t_n)(1 - \Delta Y) + C_b(\Delta Y)$$

$$\Delta Y = \frac{Y(t_{n+1}) - Y(t_n)}{Y(t_{n+1})}$$

Time-dependent results for the diluting and nondiluting plume simulations are shown in Figure 9 together with the mean plume measurements from missions 10, 13, 14, and 15 superimposed at the appropriate time intervals. The diluting simulation reproduces well the observed mixing ratios for CO and C$_2$H$_2$, but the nondiluting simulation shows a 35% decrease of CO over the course of the week due to chemical loss. We see that over the times of plume transport, CO is not a conserved tracer.

The nondiluting plume simulation indicates that 80% of net ozone production occurs within the first 24 hours of emission. In subsequent days, ozone production is only sufficient to maintain elevated O$_3$ mixing ratios in the nondiluting plume while the diluting plume experiences net loss. Measurements within the plume on flights 13 and 14 show even lower ozone mixing ratios than are obtained in the diluting plume simulation. The low O$_3$ production in this plume contrasts with the more generally observed increase of $\Delta$O$_3$/ACO from the fresh to the old plume categories (Table 3a). We attribute the discrepancy to the relatively low altitude of the simulated plume, resulting in short lifetimes for both NO$_2$ and O$_3$.

Photochemical production of ozone in the plume is NO$_x$-limited as shown in Figure 10. Doubling initial NO, NO$_2$, HNO$_3$, and PAN concentrations increased O$_3$ mixing ratios by ~7 ppbv, while doubling CO, CH$_4$, and NMHC concentrations increased O$_3$ mixing ratios by only 2 ppbv throughout the weeklong simulation.

Simulated and observed concentrations of CH$_2$O and peroxides offer an additional measure of chemical activity in the plume. The initially high mixing ratio of CH$_2$O indicates that it is primarily emitted during the combustion process or formed rapidly in the fresh plume. Oxidation of propene and ethene helps maintain CH$_2$O concentrations for the first day. In this simulation, CH$_2$O has a lifetime of ~4 hours, indicating that net photochemical production, primarily from the oxidation of CH$_4$, maintains the 400 pptv enhancement above background for the rest of the weeklong simulation.

The diluting and nondiluting model simulations bracket the H$_2$O$_2$ measurements, but the model greatly underestimates observed CH$_3$OOH. The reason for this is unclear but may be due to too low a rate constant in the model for HO$_2$+CH$_3$O. Model lifetimes of H$_2$O$_2$ and CH$_3$OOH in the simulation are ~0.8 and 0.3 days, respectively, indicating that net photochemical production is responsible for maintaining H$_2$O$_2$ mixing ratios above background. Mean $\Delta$H$_2$O$_2$/ACO and $\Delta$CH$_3$OOH/ACO enhancement ratios in the data indicate the formation of peroxides as the plumes age; however, in the model this is only observed for H$_2$O$_2$ in the nondiluting simulation.

Acetone has a lifetime in the model of ~7 days with losses
Plate 2. Evolution of NOx partitioning between NOy, HNO3, and PAN. (a) mission 10 measurements of fresh emissions after convection to 4-km altitude, (b) mission 13 measurements, 2 days, (c) mission 14, 3 days, and (d) mission 15, 5 days after last encountering fresh emissions.
due to photolysis and reaction with OH. The nondiluting simulation indicates that there is a net photochemical loss of acetone over the week. The model indicates that production which does occur is the result of oxidation of propane and larger NMHC.

The model simulation reproduces well the evolution of NO$_3$ partitioning from domination by PAN in the fresh plume to domination by HNO$_3$ in the aged plume (see Figure 9). Loss of PAN is primarily via thermal decomposition. Measurements within the plume indicate lower mixing ratios of HNO$_3$ than our diluting simulation suggests. One possible explanation would be conversion of HNO$_3$ into nitrate aerosols which is a process not included in the model and which would result in lower HNO$_3$ concentrations.

4. Regional Budget Analyses

4.1. Scaling Up From Measurements

To examine the relative importance of plumes on regional O$_3$ production, we plot a cumulative sum of both gross ozone production and loss calculated using the 0-D model versus the CO mixing ratio. Because chemistry varies with altitude, Figure 11 shows these plots in altitude intervals of 8-12, 4-8, and 0-4 km. The cumulative O$_3$ production at a particular CO concentration represents the total O$_3$ production which has been cumulatively produced by all points below that CO level. We consider O$_3$ production occurring in air masses with CO above median mixing ratios to represent O$_3$ production in plumes. Figure 11 shows that at median
Table 5. Initial Model Conditions for Biomass Burning Plume Simulation

<table>
<thead>
<tr>
<th>Species</th>
<th>Unit</th>
<th>Initial Concentrations (Fresh Plume)</th>
<th>Background Concentrations* (Diluting Plume)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CO₂</td>
<td>ppmv</td>
<td>361</td>
<td>356</td>
</tr>
<tr>
<td>CO</td>
<td>ppbv</td>
<td>390</td>
<td>102</td>
</tr>
<tr>
<td>O₃</td>
<td>ppbv</td>
<td>85</td>
<td>60</td>
</tr>
<tr>
<td>NO</td>
<td>pptv</td>
<td>179</td>
<td>19</td>
</tr>
<tr>
<td>NO₂ (calc)</td>
<td>pptv</td>
<td>434</td>
<td>46</td>
</tr>
<tr>
<td>HNO₃</td>
<td>pptv</td>
<td>931</td>
<td>277</td>
</tr>
<tr>
<td>PAN</td>
<td>pptv</td>
<td>2990</td>
<td>95</td>
</tr>
<tr>
<td>CH₄</td>
<td>ppbv</td>
<td>1720</td>
<td>1688</td>
</tr>
<tr>
<td>Ethane</td>
<td>pptv</td>
<td>1766</td>
<td>462</td>
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<tr>
<td>Ethene</td>
<td>pptv</td>
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<td>5</td>
</tr>
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<td>Acetylene</td>
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<td>80</td>
</tr>
<tr>
<td>Propane</td>
<td>pptv</td>
<td>246</td>
<td>17</td>
</tr>
<tr>
<td>Propene</td>
<td>pptv</td>
<td>139</td>
<td>2</td>
</tr>
<tr>
<td>Butene</td>
<td>pptv</td>
<td>10</td>
<td>0</td>
</tr>
<tr>
<td>Benzene</td>
<td>pptv</td>
<td>319</td>
<td>13</td>
</tr>
<tr>
<td>Toluene</td>
<td>pptv</td>
<td>98</td>
<td>0</td>
</tr>
<tr>
<td>Xylene</td>
<td>pptv</td>
<td>11</td>
<td>0</td>
</tr>
<tr>
<td>4C alkanes</td>
<td>pptv</td>
<td>61</td>
<td>0</td>
</tr>
<tr>
<td>7C alkanes</td>
<td>pptv</td>
<td>11</td>
<td>0</td>
</tr>
<tr>
<td>Isoprene</td>
<td>pptv</td>
<td>38</td>
<td>0</td>
</tr>
<tr>
<td>CH₂O</td>
<td>pptv</td>
<td>1580</td>
<td>113</td>
</tr>
<tr>
<td>Acetone</td>
<td>pptv</td>
<td>1820</td>
<td>515</td>
</tr>
<tr>
<td>Ethanol</td>
<td>pptv</td>
<td>254</td>
<td>0</td>
</tr>
<tr>
<td>H₂O₂</td>
<td>pptv</td>
<td>5060</td>
<td>3160</td>
</tr>
<tr>
<td>CH₃OOH</td>
<td>pptv</td>
<td>1320</td>
<td>1160</td>
</tr>
<tr>
<td>HCOOH</td>
<td>pptv</td>
<td>5130</td>
<td>720</td>
</tr>
</tbody>
</table>

* Mixing ratios of background air are taken to be the bottom decile of data sampled between 1 and 7 km during portions of flights 13-15 that are in proximity to the plume.

CO in all altitude intervals, not more than 30% of the cumulative gross O₃ production has taken place. The majority of O₃ production takes place in air above the median CO concentration, indicating the importance of the contribution of biomass burning plumes to the regional O₃ enhancement. In contrast, at northern high latitudes, Mauzerall et al. [1996] found that biomass burning plumes made a negligible contribution to the region’s O₃ budget. Since in the TRACE A region the majority of O₃ production takes place in plumes, we are justified in scaling up from the observed ΔO₃/ΔCO enhancement ratio in old plumes (section 2.4) to estimate a minimum regional O₃ production.

To assess the role of biomass burning plumes as regional sources of O₃ for the TRACE A conditions, we scale our enhancement ratio for old plumes (ΔO₃/ΔCO) as follows:

\[
S_{O_3} = \left( \frac{B \cdot E}{A} \right) \frac{\Delta O_3}{\Delta CO}_{old} \tag{4}
\]

Here \( S_{O_3} \) is the net O₃ source from production in biomass burning plumes. \( B = 294 \) Tg month⁻¹ (Tg = x10¹²g) is the amount of biomass burned south of the equator in South America and Africa in the month preceding the rainy season [Hao and Liu, 1994]. \( E = 58g \) CO per 1kg fuel is a CO emission factor for cerrado (grassland) in Brazil [Ward et al., 1992]. \( A = 4.8 \times 10^{17} \) cm² is the surface area of the South Atlantic air basin (between 65°W and 40°E longitude and 0°S and 40°S latitude) and is used to approximate the region influenced by biomass burning. \( \langle \Delta O_3/\Delta CO \rangle_{old} = 0.59 \) mol mol⁻¹ is the mean enhancement ratio observed in old plumes, corrected for an estimated 20% loss of CO. We thus calculate \( S_{O_3} = 17 \times 10^{12} \) molecules O₃ cm⁻² s⁻¹.

The average enhancement in the tropospheric O₃ residual (calculated using total ozone mapping spectrometer (TOMS) version 6) from the wet to the dry season in the south Atlantic air basin is ~10 DU (1 DU = 2.69 x 10¹⁶ molecules O₃ cm⁻²) [Olson et al., 1996]. Therefore, with an average residence time of 2-3 weeks in the air basin, the observed tropospheric O₃ enhancement can be generated from O₃ production in biomass burning plumes.

4.2. Model Characterization of Ozone Production in Plumes

In this section we use a point (0-D) photochemical model to evaluate the importance of biomass burning plumes of different ages to the regional O₃ budget. The model is used to calculate diel steady state production and loss rates for O₃ at each point for which measurements exist along the aircraft flight tracks and is described in detail by Jacob et al. [1996]. Diel steady state is defined by the reproducibility of concentrations over a 24-hour solar cycle. CO mixing ratios for each point are used as a proxy for plume age.

The ensemble of aircraft measurements at each point along the flight tracks is merged over the sampling interval of NO for analysis of O₃ production and loss. We divide the points into three altitude ranges (0-4, 4-8, and 8-12 km), and within each altitude range we divide the points based on their observed CO mixing ratio (<70, 70-100, 100-200, 200-300, and >300 ppbv). We use the mean CO mixing ratio to indicate the relative time since the parcel was last exposed to biomass burning emissions with CO > 300 ppbv indicating relatively fresh biomass burning plumes, CO between 200-300 and 100-200 ppbv indicating progressively aging...
Figure 9. Time-dependent photochemical model simulation at 4-km. Calculations were started at noon of day 1. Ticks on the x-axis mark midnight. In each panel the thick (thin) solid (dashed) line indicates the evolution of mixing ratios of species in the diluting (undiluting) plume simulation. The thick dashed line indicates the background mixing ratio used to dilute the plume. Points at noon of days 1, 3, 4, and 6 are mean mixing ratios measured within the plume during flights 10, 13, 14, and 15.

biomass burning influence, CO between 70-100 ppbv indicating background air, and CO < 70 ppbv indicating marine boundary layer air in the 0-4 km altitude plot and stratospheric air in the 8-12 km plot. For each altitude range in Figure 12 we plot the mean production and loss rates for O$_3$ versus the mean CO mixing ratio within each CO interval. Vertical bars on the plots indicate a standard error (σ/N) in the mean rates calculated from all points within a particular CO bin.

Figure 12 shows a maximum in gross ozone production and loss at low altitudes and within fresh plumes and shows a decrease in both production and loss as altitude increases and plumes age. O$_3$ production is dominated by HO$_2$ + NO with CH$_3$O$_2$ + NO being a relatively minor secondary source at all altitudes. O$_3$ production is largest in the freshest most polluted plumes because of elevated NO$_x$ mixing ratios. In background air at high altitudes, net ozone production remains positive, while in background air at low altitudes, net production is close to zero. In the stratospheric air (mean CO ~ 70 ppbv in the 8-12 km plot) net ozone production remains slightly positive, while in the marine boundary layer (mean CO ~ 70 ppbv in the 0-4 km plot), net ozone production is negative. Throughout the column O$_x$ + H$_2$O is the primary loss route for O$_3$, followed by O$_3$ + HO$_2$ and O$_3$ + OH.

We compare the magnitude of ozone production and loss in background air in the wet and dry seasons by conducting a wet season simulation based on a simplified atmosphere including only CO, CH$_4$, O$_3$, NO, and H$_2$O. In order to be a sensitivity analysis of the impact of biomass burning emissions on regional photochemistry, wet season temperatures, dew points, and zenith and nadir UV fluxes were taken from median TRACE A measurements for each altitude interval [Jacob et al., 1996]. Values used in the wet season calcula-
Figure 10. Sensitivity analysis to NO$_x$ versus hydrocarbon limitation on O$_3$ production in the diluting plume simulation.

5. Conclusions

Photochemistry occurring in biomass burning plumes over the tropical south Atlantic was analyzed using data collected during the TRACE A aircraft expedition conducted during the tropical dry season in September 1992. During TRACE A the chemical composition of biomass burning emissions over source regions and over the South Atlantic were measured. We report enhancement ratios ($\Delta Y/\Delta X$,
where Δ indicates the enhancement of a compound in the plume above the local background mixing ratio, \( Y \) is individual NMHC, CO, O₃, CH₄, N₂O, HNO₃, PAN, CH₂O, acetone, H₂O₂, CH₃OOH, HCOOH, CH₃COOH, or aerosols and \( X \) is CO or CO₂) as a function of plume age inferred from the progression of \( \Delta \text{NMHC}/\Delta \text{CO} \) enhancement ratios. We diagnose emission, formation, and loss of species in plumes from progression of enhancement ratios from fresh to old plumes. We find that O₃ is produced in plumes over at least a 1 week period with mean \( \Delta \text{O}_3/\Delta \text{CO} = 0.7 \) observed in old plumes; that CH₂O is primarily emitted during the fire or formed shortly thereafter, then subsequently lost as the plume ages; and that acetone is formed after emission as the plume ages. We found \( \Delta \text{CO}/\Delta \text{CO}_2 \) decreases from 0.05 to 0.02 from regions of fresh fires to remote ocean locations as a result of the photochemical loss of CO and decreasing background mixing ratios of CO and CO₂. We estimate that a downward correction of ~20% in enhancement ratios relative to \( \Delta \text{CO} \) is necessary in old plumes due to photochemical loss of CO.

In our case study of a large biomass burning plume sampled at 4-km altitude over the course of 5 days off the west coast of Africa we found elevated concentrations of PAN in the fresh plume. The degradation of PAN helped maintain NOₓ mixing ratios in the plume where, over the course of a week, PAN was converted to HNO₃. Ozone production in the plume was limited by the availability of NOₓ, and because of the short lifetime of O₃ at 4-km, net ozone production in the plume was negligible.

We examined the influence that biomass burning plumes have on regional ozone production. We found that the majority of O₃ production takes place in air above the median CO concentration, indicating that most O₃ production occurs in plumes. Maximum gross ozone production and loss rates occur at low altitudes and within fresh plumes. Rates decrease as altitude increases and as plumes age. A simple wet season model simulation indicates that gross O₃ production rates in background air during the dry season are ~3 times larger than during the wet season.

Scaling up from the mean observed \( \Delta \text{O}_y/\Delta \text{CO} \) in old
Table 6. Wet and Dry Season Mixing Ratios

<table>
<thead>
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<th>0-4 km</th>
<th>4-8 km</th>
<th>8-12 km</th>
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<tr>
<td>Wet Season, Values Used in Simulation</td>
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<tr>
<td>CO&lt;sub&gt;a&lt;/sub&gt;, ppbv</td>
<td>80</td>
<td>80</td>
<td>80</td>
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<td>O&lt;sub&gt;3&lt;/sub&gt;&lt;sup&gt;b&lt;/sup&gt;, ppbv</td>
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<tr>
<td>NO&lt;sub&gt;2&lt;/sub&gt;, pptv</td>
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<td>8</td>
<td>40</td>
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</table>

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<th>Dry Season, Median TRACE A Mixing Ratios</th>
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<tr>
<td>CO, ppbv</td>
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<td>103</td>
<td>93</td>
</tr>
<tr>
<td>O&lt;sub&gt;3&lt;/sub&gt;, ppbv</td>
<td>50</td>
<td>69</td>
<td>74</td>
</tr>
<tr>
<td>NO, pptv</td>
<td>25</td>
<td>29</td>
<td>120</td>
</tr>
</tbody>
</table>

<sup>a</sup> CO is assumed to be well mixed throughout the troposphere during the wet season. Mixing ratios are specified from measurements made during the wet season in the lower troposphere over the southern tropical Atlantic Ocean [Harriss et al., 1990].

<sup>b</sup> O<sub>3</sub> values are from wet season ozone sonde measurements made over Amazonia [Kirchhoff et al., 1990].

<sup>c</sup> The 0-8 km mixing ratios are specified from observations during the wet season from Singh et al., [1990]. The 8-12 km mixing ratios are specified from median mixing ratios observed during the wet season on the southern hemispheric portion of flights between Rio de Janeiro and Natal Brazil and between Natal and Dakar, Senegal [Drummond et al., 1988].

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