

# 1 High-resolution exposure assessment for volatile organic compounds in 2 two California residences

3 *David M. Lunderberg,<sup>1,2\*</sup> Pawel K. Misztal,<sup>2†</sup> Yingjun Liu,<sup>2‡</sup> Caleb Arata,<sup>1,2</sup> Yilin Tian,<sup>2,3</sup> Kasper*  
4 *Kristensen,<sup>2§</sup> Robert J. Weber,<sup>2</sup> William W Nazaroff,<sup>3</sup> and Allen H. Goldstein<sup>2,3</sup>*

5 <sup>1</sup> *Department of Chemistry, University of California, Berkeley, CA, USA.*

6 <sup>2</sup> *Department of Environmental Science, Policy, and Management, University of California, Berkeley, CA, USA.*

7 <sup>3</sup> *Department of Civil and Environmental Engineering, University of California, Berkeley, CA, USA.*

8 <sup>†</sup> *Now at Department of Civil, Architectural and Environmental Engineering, The University of Texas at Austin, TX,*  
9 *USA*

10 <sup>‡</sup> *Now at BIC-ESAT and SKL-ESPC, College of Environmental Sciences and Engineering, Peking University, China*

11 <sup>§</sup> *Now at Department of Engineering, Aarhus University, Denmark*

12 *\*Corresponding email: [david\\_lunderberg@berkeley.edu](mailto:david_lunderberg@berkeley.edu)*

## 13 **0 ABSTRACT**

14 Time spent in residences substantially contributes to human exposure to volatile organic  
15 compounds (VOC). Such exposures have been difficult to study deeply, in part because VOC  
16 concentrations and indoor occupancy vary rapidly. Using a fast-response online mass  
17 spectrometer, we report time-resolved exposures from multi-season sampling of more than 200  
18 VOC species in two California residences. Chemical-specific source apportionment revealed that  
19 time-averaged exposures for most VOCs were mainly attributable to continuous indoor  
20 emissions from buildings and their static contents. Also contributing to exposures are occupant-  
21 related activities, such as cooking, and outdoor-to-indoor transport. Health-risk assessments are  
22 possible for a subset of observed VOCs. Acrolein, acetaldehyde, and acrylic acid concentrations  
23 were above chronic advisory health guidelines, whereas exposures for other assessable species  
24 were typically well below guideline levels. Studied residences were built in the mid-20th  
25 century, indicating that VOC emissions even from older buildings and their contents can  
26 substantially contribute to occupant exposures.

27

## 28 **KEYWORDS:**

29 source apportionment, time-activity patterns, air pollution, indoor, health risk assessment,  
30 cooking, building-material emissions

## 31 1 INTRODUCTION

32 Exposures to volatile organic compounds (VOCs) can be associated with adverse health  
33 outcomes (1–5). Considering that modern populations spend most of their time in residences and  
34 that most VOC concentrations and associated exposures are substantially higher indoors than  
35 outdoors (6–8), evaluating the residential component is critical for understanding total VOC  
36 exposures. Certain VOCs and very volatile organic compounds (henceforth bundled as VOCs for  
37 brevity), such as formaldehyde, some halogenated solvents, and BTEX (benzene, toluene, ethyl  
38 benzene, xylene), among others, have been the subject of substantial attention (9–11). However,  
39 for many VOCs, the levels of exposure, the controlling physical processes, and the consequent  
40 health risks remain understudied (12, 13).

41 Residential VOC abundances are attributable to multiple sources, which vary among compounds.  
42 Potential sources include building materials (14), outdoor-to-indoor transport (8), indoor  
43 chemistry (15), consumer products (16), activities such as cooking (17, 18), and the human  
44 occupants themselves (19, 20). For most VOCs and in most buildings, the relative contributions  
45 from each source to indoor concentrations are not known. In addition, indoor VOC abundances  
46 are modulated by physicochemical parameters, such as the ventilation rate (21), temperature  
47 (through sorptive interactions with interior surfaces (22) and temperature-dependent material  
48 emissions (18)), humidity (23) and both seasonal and diel variations in outdoor VOC  
49 concentrations (18). Some studies have suggested that VOC emissions from building materials  
50 are most important immediately after construction or renovation and can decline by orders of  
51 magnitude within a few years (24–26).

52 Time-resolved measurements of VOCs in residences and associated exposures are scarce. Time-  
53 averaged VOC measurements, either actively collected on sorbent tubes or passively collected on  
54 diffusive samplers over fixed intervals, have been used to characterize VOC exposures in cross-  
55 sectional studies (7, 8, 27, 28). Samples are sometimes collected in multiple locations specific to  
56 each research participant, including direct monitoring of air immediately proximate to study  
57 subjects using personal samplers. In many cases, studies implement a targeted screening  
58 approach where analysis is restricted to a subset of chemicals of relevant health interest.

59 Typically, source apportionment has not been conducted via intensive study of individual  
60 residences. Studies focusing on source apportionment of indoor VOCs may collect time-  
61 averaged samples and conduct source apportionment via multivariate receptor models. These  
62 analyses typically use data reduction techniques such as principal component analysis (PCA) or  
63 positive matrix factorization (PMF) on population-level data acquired in survey studies (8, 29–  
64 31). In one such study of Canadian residences, Bari et al. reported that a majority of indoor  
65 VOCs were attributable to indoor sources with household products being the largest contributor  
66 and off-gassing of building materials the smallest (31).

67 Personal exposures measured by portable samplers are often greater than those predicted by  
68 stationary monitoring of outdoor, home, and occupational environments (7, 32). Predictions  
69 based on stationary monitoring often assume a well-mixed single-zone space, an assumption  
70 which may not be fully justified for strongly localized source events that create spatial

71 heterogeneities (33, 34). Proximity of occupants to source events has been used to explain the  
72 discrepancy between personal monitoring and stationary monitoring for particles (35–37). This  
73 effect may partially explain the discrepancy between microenvironmental monitoring and  
74 personal monitoring when assessing exposures to VOCs released by occupant activities, such as  
75 cooking, cleaning or application of personal care products. Another important point is that  
76 stationary exposure estimates using time-integrated sampling rest on the assumption that average  
77 measured VOC concentrations are representative of average VOC concentrations during  
78 occupancy. For VOCs with significant contributions from occupant-related episodic sources,  
79 such as cooking or use of personal care products, concentrations may be elevated during  
80 occupancy and so occupant exposures would be underestimated if based on time-average  
81 concentration measurements that include periods of vacancy. Sporadic, high-concentration  
82 events might produce significant temporal variability in VOC abundances and, if so, would  
83 necessitate the acquisition of time-resolved VOC measurements and occupant activity data for  
84 estimating VOC exposures and apportioning exposure sources. Owing to advances in analytical  
85 capabilities, such measurements have begun to be acquired in recent years in several field  
86 monitoring campaigns (17–20, 22).

87 In this work, we discuss the first broad-scale application of time-resolved VOC measurements  
88 coupled with a) occupant time budgets to assess residential exposures to VOCs and b) occupant  
89 time-activity budgets to attribute exposures to source categories. The 250 monitored analytes in  
90 this work span a broad range of volatilities, chemical functionalities, and toxicities (Tables S1–  
91 S3). Among the characterized analytes are aromatics (including benzene, toluene, xylene),  
92 carbonyls (including acetaldehyde and acrolein), furanoids, halogenates, organosulfur  
93 compounds, carboxylic acids, alcohols and alkenes, siloxanes, and other compounds of potential  
94 health interest (acetonitrile, acrylic acid, acrylonitrile).

95 In total, we assess residential VOC exposures for more than 200 VOCs sampled with high time  
96 resolution over three multiweek monitoring campaigns conducted in two California households.  
97 We implement a source apportionment model for VOCs to attribute exposures to four broad  
98 source categories: continuous indoor sources, outdoor origin, cooking, and other. We  
99 contextualize our observations by utilizing information that relates exposure to health risk, as  
100 reported by two U.S. governmental agencies and one California state agency. Lastly, we assess  
101 the performance of stationary time-averaged exposure estimates relative to time-resolved best  
102 estimates of exposure.

## 103 2 EXPERIMENTAL METHODS

104 **Site information.** This work utilizes measurements from three field campaigns conducted at two  
105 normally occupied California residences in the East Bay region of the San Francisco Bay Area in  
106 summer and winter seasons (H1 summer, H1 winter, H2 winter). The monitoring campaigns in  
107 these residences have been described elsewhere in detail (18, 38, 39, 40). This paper represents a  
108 continuation of prior analyses describing indoor ventilation patterns (38) and VOC emissions  
109 (18) at the H1 residence. It also represents the first reporting of VOC data acquired at the H2  
110 residence. Briefly, the H1 site is a single-family, split-level wood-frame residence, constructed in  
111 the 1930s, with two adult occupants designated here as “H1M1” and “H1F1.” Measurements  
112 were made for 8 weeks in 2016 from mid-August to early October (H1 summer) and for 5 weeks  
113 in 2017 from late January to early March (H1 winter). The house did not have air conditioning.  
114 During (mild) winter, the residence was sometimes heated by a gas furnace. The H2 site  
115 consisted of a single-family, single-story wood-frame residence, constructed in the 1950s, with  
116 two adult occupants designated “H2M1” and “H2F1”, a teenager, and a dog. A third adult  
117 occupant designated “H2M2” was present for approximately half of the measurement period.  
118 Measurements were made for an 8-week campaign in 2017-18, from early December to early  
119 February. The H2 site was heated by a forced-air gas-fired furnace operating twice a day, in  
120 morning and evening.

121 At each residence, occupants maintained daily logs describing their temporal presence and  
122 activity. Occupant presence data was recorded with approximately 5-min resolution and coded as  
123 “indoors,” “indoors sleeping,” and “away.” Time-resolved activities, including cooking,  
124 cleaning, candle combustion, and some use of personal care products, were also recorded by the  
125 occupants. A network of >50 wireless sensors collected metadata during each measurement  
126 period, including temperature, humidity, motion, appliance use, and door/window status.

127 **Chemical measurements.** A proton-transfer-reaction time-of-flight mass spectrometer (PTR-  
128 ToF-MS, Ionicon Analytik GmbH, Austria, PTRTOF 8000) was used to measure VOCs during  
129 each campaign. The PTR-ToF-MS was located in an external garage (H1 summer, H1 winter) or  
130 an external shed (H2 winter). The PTR-ToF-MS uses soft ionization with  $\text{H}_3\text{O}^+$  as the primary  
131 reagent ion, yielding parent masses (1.000 amu – 500.00 amu) of gaseous species whose proton  
132 affinities are greater than the proton affinity of water. Most unsaturated hydrocarbons and  
133 compounds with heteroatoms are quantifiable with detection limits on the order of 0.005 ppb and  
134 time-resolution on the order of minutes. PTR-ToF-MS operating parameters and quantification  
135 during the H1 summer and winter campaigns have been described elsewhere (18, 38) and are  
136 summarized in the supporting information (SI, Section 1).

137 **Exposure Analysis.** Time-integrated exposures (hereafter “exposures”) were calculated as the  
138 integral of time-instantaneous exposures over the time period of exposure. Daily occupant  
139 exposures in units of ppb-hours per day ( $\text{ppb}\cdot\text{h d}^{-1}$ ) were calculated by integrating ion formula  
140 concentrations for each day during periods when each occupant was present. Measurement  
141 periods with major interferences from researcher activities were excluded from all analyses. The  
142 kitchen concentration was used at H1 as the best-estimate for calculating occupant exposures and  
143 the living room concentration was used at H2. VOC concentrations throughout the living spaces

144 and kitchen in both residences were well-coupled. During sleeping hours at H2, the  
145 concentration in the kitchen may not be fully representative of occupant exposures as each closed  
146 bedroom is partially decoupled from the main living space. In this condition, occupant exposures  
147 to bioeffluents emitted during sleeping periods may be underestimated. At H1, the bedroom  
148 doors were left open throughout the sampling campaign and so the sleeping environment is  
149 expected to be well-coupled to the kitchen, although biogenic emissions may still be higher  
150 within the bedroom. Simulated distributions of time-averaged exposures were constructed by  
151 random sampling. These distributions may be biased by 1-2 data gaps (<2 days) in concentration  
152 time series related to instrumental down time or intentional manipulation experiments conducted  
153 by researchers. Data gaps encountered during random selection were treated by using the next  
154 available measurement in the “time-averaged” sample.

155 **Source Apportionment.** VOC time series were typically characterized by a persistent  
156 background modulated by regular changes in temperature in association with the diurnal  
157 temperature cycle (H1 summer) or house heating cycle (H2 winter) and punctuated by rapid  
158 concentration spikes related to specific events. The temperature-dependent background was  
159 clearly identified during periods of vacancy when occupant-related events did not occur.  
160 Concentration spikes and associated peak boundaries were manually identified for each ion  
161 formula and assigned to specific source categories based on coincident events during the growth  
162 phase of the peak. The peak baseline was determined by a linear regression between the  
163 concentrations observed at peak boundaries. Net peak areas were defined as the total integrated  
164 area above the peak baseline. Event peaks were excluded if at least 50% of the event by peak-  
165 height was attributable to outdoor infiltration. More detailed information about the peak selection  
166 process is available in the SI (Section 1b). Outdoor VOC concentrations were assumed to enter  
167 the indoors without attenuation, allowing for outdoor VOC concentrations to be used as a proxy  
168 for the indoor VOC concentrations of outdoor origin. On the rare circumstance that outdoor  
169 concentrations were greater than indoor concentrations, the indoor concentration was assumed to  
170 be derived wholly from outdoor-to-indoor transport. The integrated areal remainder after  
171 subtracting contributions from events and from outdoor transport was defined as “continuous  
172 indoor sources.” In total, more than 12,000 peaks were identified from among the 594 chemical  
173 concentration time series acquired during the three campaigns.

174 **Risk-Based Prioritization.** Advisory health guidelines for acute and chronic exposures to VOCs  
175 were acquired from three organizations: the U.S. Environmental Protection Agency (USEPA)  
176 Integrated Risk Information System (IRIS, chronic only),<sup>41</sup> the California Environmental  
177 Protection Agency Office of Environmental Health Hazard Assessment (OEHHA),<sup>42</sup> and the  
178 Agency for Toxic Substances and Disease Registry (ATSDR).<sup>43</sup> Each organization has  
179 independently developed concentration guidelines for chronic VOC exposures. The IRIS  
180 database reports reference concentrations (RfCs), “an estimate (with uncertainty spanning  
181 perhaps an order of magnitude) of a continuous inhalation exposure to the human population  
182 (including sensitive subgroups) that is likely to be without an appreciable risk of deleterious  
183 effects during a lifetime.” OEHHA reports reference exposure levels (RELs), an estimate of  
184 concentrations for inhalation exposure “at or below which no adverse health effects are

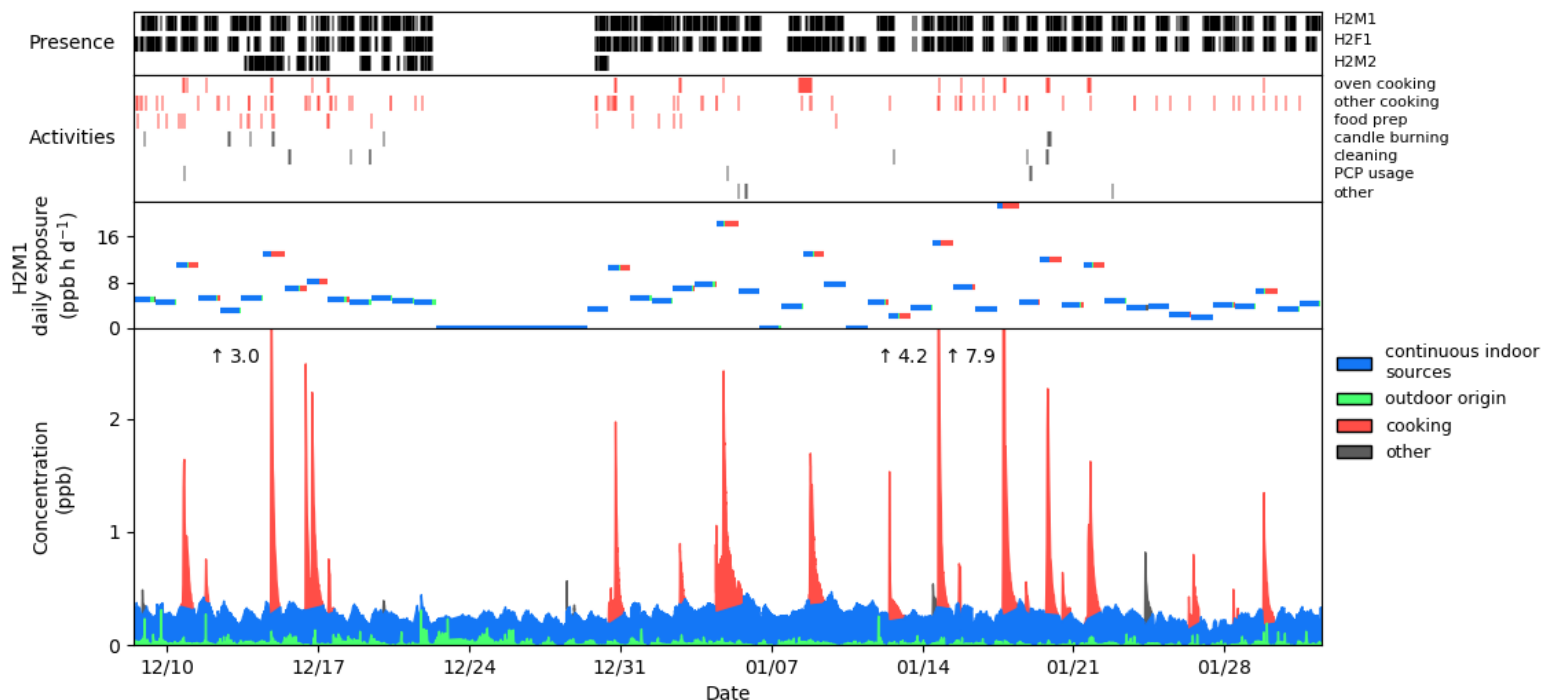
185 anticipated following long-term exposure.” ATSDR reports minimal risk levels (MRLs), “an  
186 estimate of the amount of a chemical a person can eat, drink, or breathe each day without a  
187 detectable risk to health.” These health guidelines were selected as three major guidelines from  
188 state and federal agencies in the United States that are relevant to the location of the studied  
189 residences. Other international guidelines have also been developed.<sup>44-48</sup>

### 190 **3 RESULTS AND DISCUSSION**

191 **Time-resolved exposures.** Indoor VOC concentrations and occupant time-activity budgets were  
192 acquired at two normally occupied California residences in the summer and winter seasons over  
193 three monitoring periods: the H1 summer campaign (8 weeks), the H1 winter campaign (5  
194 weeks), and the H2 winter campaign (8 weeks). In total, 250 ion masses corresponding to  
195 molecular formulas of VOCs were identified (H1 summer: 217 ions; H1 winter: 170 ions; H2  
196 winter: 205 ions), with substantial overlap among campaigns. Of these ions, 115 were assigned  
197 to specific chemical structures based on site-specific knowledge, fragmentation patterns, and  
198 supporting measurements using thermal desorption two-dimensional gas chromatography time-  
199 of-flight mass spectrometry (TD-GC×GC-ToF-MS). Paired indoor/outdoor VOC concentrations  
200 were acquired every 30 minutes. Figure 1 displays occupant time-budgets and activities for the  
201 H2 winter campaign. During each campaign, VOC concentrations generally increased with  
202 increasing temperature or specific occupant activities and decreased with increasing ventilation.  
203 Concentrations were higher indoors when compared to outdoors (> 2×) for most VOCs (~95%)  
204 at the H1 (18) and H2 sites, consistent with past field studies. Moreover, for about half of the  
205 observed VOCs, concentrations were substantially higher (> 10×) indoors than outdoors.

206 Source apportionment in this study attributed the residential VOC exposures to four source  
207 categories: “continuous indoor sources,” “outdoor origin,” “cooking,” and “other.” Time-  
208 resolved exposures and associated source apportionments are displayed in Figure 1 for site H2  
209 and subject H2M1 for the case of acrolein, an illustrative compound of health interest with prior  
210 evidence of both continuous indoor sources (lumber) and episodic sources (combustion, cooking)  
211 (48–52). We note that acrolein has been previously measured in outdoor ambient air and retail  
212 stores near the H1 and H2 residences (53, 54). In the present study, daily occupant exposures to  
213 acrolein were mainly attributable to continuous indoor sources, with a median exposure of ~ 4.6  
214 ppb-hours per day (ppb-h d<sup>-1</sup>). Substantial variation was observed in daily exposures, principally  
215 due to variations in the fraction of time that occupants spent indoors combined with the effects of  
216 event-specific sources. Certain days with cooking events doubled the daily occupant acrolein  
217 exposures. Due to the infrequency of cooking events that emitted acrolein at H2, exposures  
218 attributable to continuous indoor sources accounted for ~65% of cumulative occupant acrolein  
219 exposures, with cooking contributing 25-30% and outdoor-to-indoor transport the small  
220 remainder. Most episodic enhancements in acrolein concentrations were specifically attributed to  
221 cooking. Only two episodic enhancements remained unassociated with any reported events.  
222 (These enhancements did not significantly influence cumulative occupant acrolein exposures,  
223 indicating that episodic contributors to acrolein exposures are well understood at this site and are  
224 principally derived from cooking.) At H2, oven cooking produced the largest enhancements in  
225 acrolein concentrations. At H1, stovetop cooking produced the largest acrolein enhancements.  
226 Both the H2 oven and the H1 stovetop were electric, suggesting that heated food (rather than  
227 natural gas combustion) was the largest episodic contributor to acrolein emissions in agreement

228 with past studies observing no differences between types of cooking appliances (48). While  
 229 enhancements in acrolein concentrations at H2 were also observed in association with use of a  
 230 natural gas stove-top, these enhancements were much smaller in abundance than those from the  
 231 electric oven. Most acrolein enhancements at H2 were observed in association with the use of  
 232 cooking oil. This finding aligns with previous reports that heated cooking oil is a primary source  
 233 of acrolein during cooking (50–52).

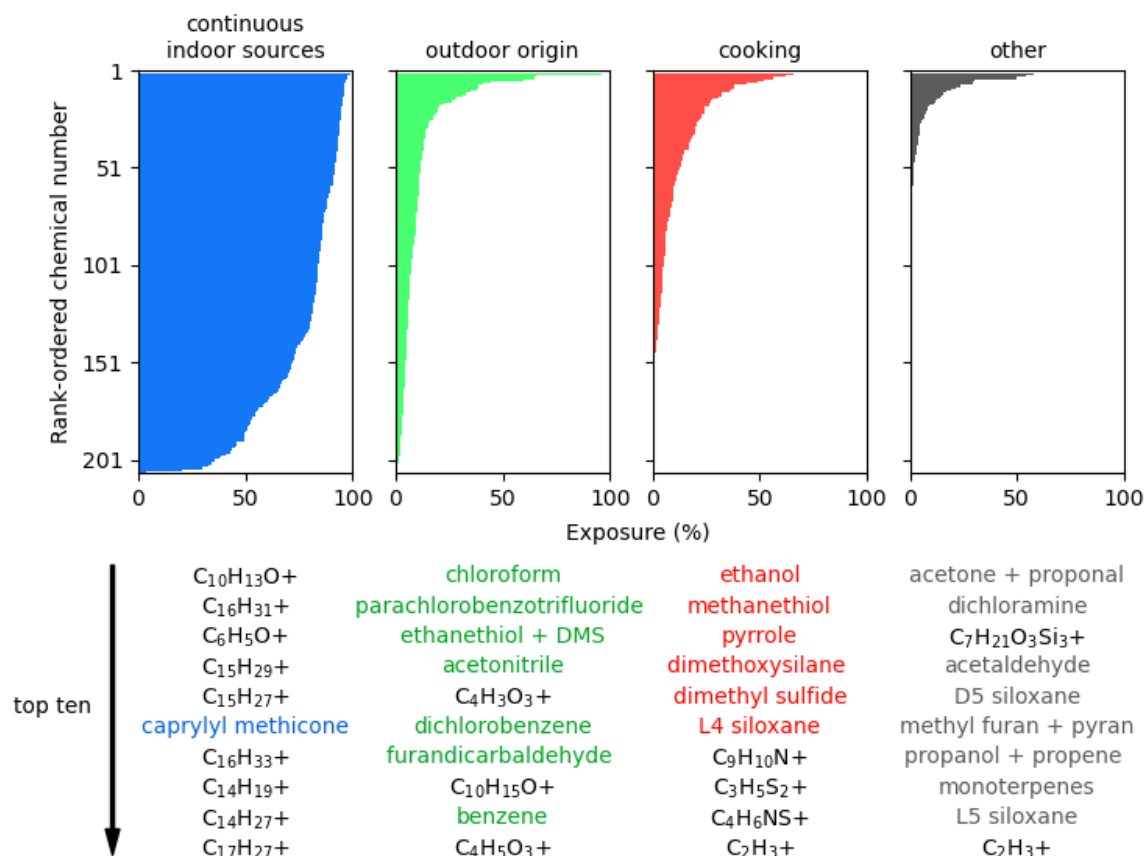


235 **Fig. 1. Acrolein exposure data at H2.** In the two uppermost panels, occupant presence  
 236 and activity data are displayed during the H2 campaign. In the bottommost panel, the  
 237 time series and associated source apportionment of indoor acrolein ( $C_3H_5O^+$ ) is  
 238 displayed, with colors corresponding to the relative contribution from each source  
 239 category. Integrated daily exposures are shown in the second panel from the bottom for  
 240 occupant H2M1. The shading of the horizontal bar is colored according to the proportion  
 241 of exposure attributable to each source category.

242 **Source apportionment.** Source apportionment of occupant VOC exposures was performed for  
 243 all VOCs observed in the H1 winter, H1 summer, and H2 winter campaigns following the  
 244 method shown in Figure 1. At each site, exposures are assessed separately for each of the  
 245 primary adult occupants. The relative contributions to the total exposure from each source  
 246 category were identified for each VOC and are reported on a percentage basis. Source  
 247 apportionment of residential exposures to 205 VOCs for occupant H2M1 is shown in Figure 2.  
 248 Approximately 90% of observed VOCs produced time-averaged exposures that were primarily  
 249 attributable to continuous indoor sources during all three monitoring campaigns at these two  
 250 residences. Compounds with predominantly indoor continuous sources spanned diverse ranges of  
 251 volatilities and chemical functionalities, including carboxylic acids (octanoic acid – 95% of  
 252 exposure from indoor continuous emissions), siloxanes (caprylyl methicone – 95%), aromatics

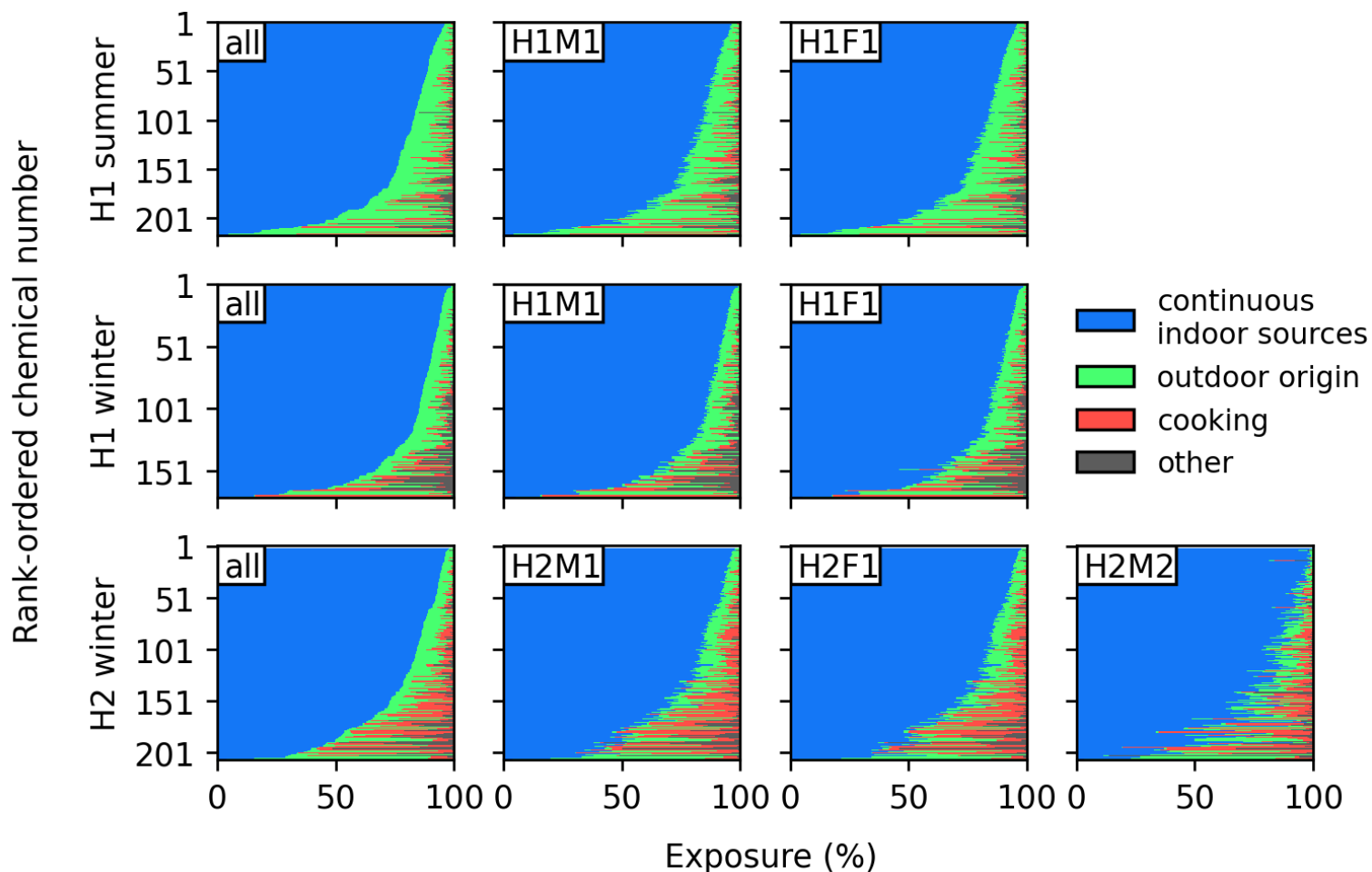
253 (phenol – 93%), carbonyls (C9 carbonyl – 92%), among others, and were not easily  
254 generalizable to specific categories. Caprylyl methicone is a trisiloxane commonly used in  
255 cosmetics; surprisingly, its presence was dominated by continuous emissions with the exception  
256 of two short-term concentration spikes in the evening. Cooking was the most important episodic  
257 source of indoor VOC exposures (18); however, only eight specific VOCs had exposures that  
258 were principally attributable to cooking during any of the campaigns for any occupant: ethanol,  
259 methanethiol, pyrrole, dimethyl sulfide,  $C_6H_9O_4^+$ , D4 siloxane, L5 siloxane, and  
260 dimethoxysilane. The latter three compounds were emitted synchronously with cooking events,  
261 but it is not certain that they originated directly from the food, cookware or cooking appliances.  
262 Apart from cooking, exposures of only three volatile compounds—acetone, an unidentified  
263 siloxane ( $C_7H_{21}O_3Si_3^+$ ), and the inorganic gas dichloramine—were otherwise largely attributable  
264 to episodic indoor emission events for any occupant. Acetone was identified as originating from  
265 the indoor use of nail polish remover during the H2 winter campaign. Dichloramine was  
266 identified as originating from episodic cleaning events. Cyclic siloxanes are known additives to  
267 consumer personal care products. Noting that three of five concentration spikes of the  
268 unidentified siloxane over the H2 winter campaign occurred immediately following sleeping  
269 periods when a personal care product might have been applied as part of a morning routine, it is  
270 suspected that the unidentified siloxane is a component of a personal care product. Only ten  
271 specific VOCs had exposures that were primarily attributable to outdoor-to-indoor transport for  
272 any occupant. Four of these compounds were observed in all campaigns (chloroform,  
273 acetonitrile, parachlorobenzotrifluoride,  $C_4H_3O_3^+$ ), four compounds were unique to the H1  
274 summer campaign ( $C_{10}H_9O_3^+$ ,  $C_2H_5O_3^+$ ,  $C_5H_5O_3^+$ ,  $CCl_2F^+$ ), and two compounds were unique to  
275 the H2 winter campaign (ethane thiol + DMS and  $C_{10}H_{15}O^+$ ). We also highlight  $C_8H_9O_2^+$   
276 (anisaldehyde, a floral fragrance) and  $C_{10}H_{17}^+$  (monoterpenes, emitted during citrus  
277 consumption) as compounds of interest. In addition to cooking, we observed large contributions  
278 from episodic events unrelated to cooking (31% and 36%, respectively during the H1 winter  
279 campaign), expected to partially originate from personal care product use and citrus  
280 consumption.





281  
 282 **Fig. 2. Rank-ordered VOC exposures by source category for occupant H2M1 during**  
 283 **the winter campaign at H2.** Residential VOC exposures to 205 VOCs were calculated  
 284 for occupant H2M1. The contributions from each source category to the total exposure of  
 285 each VOC were identified and are reported on a percentage basis. The relative  
 286 contributions from each VOC were then rank-ordered from greatest-to-least for each  
 287 source category. Source categories include continuous indoor sources (blue), outdoor-  
 288 origin (green), cooking (red), and other (gray). The unique rank-orderings for each source  
 289 category are displayed in the upper panel and the top ten VOCs in each rank-ordered  
 290 source category are identified in the lower section of the figure.

291 Relative contributions for each source category were calculated for all observed VOCs and for  
 292 each occupant during each campaign (Figure 3). Absolute daily exposures and relative source  
 293 apportionments can be found in the SI (Figure S1, Tables S1-S3). Time-activity budgets exhibit  
 294 some differences (Figure S2); however, source apportionments for each occupant were  
 295 qualitatively and quantitatively similar. For the H1 campaigns, relative contributions from source  
 296 categories for occupant exposures were within 15 absolute percentage points of relative source  
 297 apportionments derived from concentration time series for all but C<sub>6</sub>H<sub>9</sub>O<sub>4</sub><sup>+</sup>. (Source  
 298 apportionments from concentration time series are equivalent to relative source apportionments  
 299 of exposures for an occupant who is always present.)



301 **Fig. 3. Relative source apportionment of VOC exposures.** Residential exposures to  
 302 (217, 170, 205) detected VOC ions were calculated for each occupant during the (H1  
 303 summer, H1 winter, H2 winter) monitoring campaigns and were apportioned into specific  
 304 source categories. Subpanels display relative VOC exposures for individual occupants;  
 305 the heading ‘all’ refers to a hypothetical occupant who is present for all of the  
 306 measurement period. In subpanels, each horizontal bar corresponds to a single VOC and  
 307 is colored according to the relative contribution from each source category (continuous  
 308 indoor sources [blue], outdoor-origin [green], cooking [red], and other [gray]). Each row  
 309 of panels uses the same rank-ordering defined by the first ‘all’ subpanel, which has VOCs  
 310 rank-ordered from greatest-to-least by the relative contribution from “continuous indoor  
 311 sources.” The top, middle, and bottom rows correspond to the H1 summer, H1 winter,  
 312 and H2 winter campaigns, respectively.

313 In contrast, deviations were observed between source apportionments for a substantial subset  
 314 of VOCs that were dominantly event-driven during the H2 winter campaign (19 of 205 ions).  
 315 These mainly occurred for the H2M2 occupant, who was present for only part of the  
 316 measurement campaign. If the H2M2 occupant is excluded, only two exceptions are  
 317 observed, the inorganic gas dichloramine (related to cleaning) and the  $C_3H_6^+$  ion. Generally,

318 if an occupant was absent for a large emission event, as in the case of H2M2 for much of the  
319 measurement campaign, discrepancies were observed.

320 Absolute exposures were comparable between the H1 summer and winter seasons for most  
321 VOCs. Of 166 ions concurrent to both seasons, absolute mean concentrations for 66% (97%) of  
322 VOCs differed by no more than 50% (factor of 2) between seasons (Figure S3). Large exceptions  
323 ( $> 2\times$  difference) were observed for D5 siloxane, monoterpenes,  $C_2H_3^+$ , and methane sulfonic  
324 acid (greater during the H1 winter season) and for cinnamaldehyde (greater during the H1  
325 summer season). During the H1 summer season, 47 more compounds were observed above the  
326 detection threshold than during the H1 winter season. These compounds are chemically diverse,  
327 including halogens, aromatics, oxygenates, and organosulfurs. We are unable to assign most of  
328 these ions to specific chemical structures. While only four VOCs were principally attributable to  
329 outdoor-to-indoor transport during the H1 winter season, the same four plus four additional  
330 VOCs were attributable to outdoor-to-indoor transport during the H1 summer season, of which  
331 three were highly oxygenated species ( $C_2H_5O_3^+$ ,  $C_5H_5O_3^+$ ,  $C_{10}H_9O_3^+$ ). The highly oxygenated  
332 species had cyclical outdoor fluctuations with afternoon peaks, suggesting that these compounds  
333 may be arising as secondary VOCs produced through summertime enhanced outdoor  
334 photochemistry or as biogenic emissions. In general, outdoor-to-indoor transport was more  
335 important during the summer season as opposed to the winter season. Indoor concentrations  
336 attributable to outdoor-to-indoor transport were greater on an absolute basis for 147 compounds  
337 concurrent to both the H1 summer and H1 winter campaigns (88%). However, contributions for  
338 these compounds remained low, accounting, on average, for 16% and 11% of total indoor  
339 concentrations in summer and winter, respectively.

340 A recent review of longitudinal studies measuring total volatile organic compound (TVOC)  
341 levels asserts that emissions from buildings and their contents are a “new building” phenomenon  
342 that is most important only soon after construction, renovation, or refurbishing as new materials  
343 off-gas. After an initial 1-2 year off-gassing period, TVOC emissions from building materials  
344 decline such that they are assumed to be unimportant relative to VOC emissions related to  
345 occupants and their activities (26). However, measurements of TVOC levels are not  
346 generalizable to specific VOCs that may be of interest to public health. One longitudinal study of  
347 25 speciated VOCs including alkanes, aromatics, and aldehydes in 251 Japanese residences  
348 similarly found that VOC concentrations in wood-framed housing will generally decrease over a  
349 1-2 year off-gassing period before stabilizing at levels consistent with much older residences  
350 (24). In contrast, homes with ferroconcrete or light-steel frames, regardless of age, produced  
351 VOC profiles consistent with aged-wood frames. The present study focused on measurements  
352 acquired at two houses of wood-frame construction, as is common in the US housing stock. We  
353 caution that certain features of wood-framed houses, such as temperature-dependent emissions  
354 originating from wood and substantial buffering of indoor humidity (55), may not be fully  
355 generalizable to other construction types.

356 While it may be true for some chemicals that VOC emissions from buildings are less important  
357 in older residences, observations at the H1 and H2 sites indicate that continuous indoor sources  
358 were the largest contributors to VOC exposures for more than 90% of observed species, even  
359 though the residences were many decades old and had not been recently remodeled or  
360 refurnished. Plausible sources might include continued material off-gassing as deeply bound

361 VOCs migrate to surfaces; oxidative, thermal or microbial decomposition of building materials  
362 yielding decomposition products; or sorptive interactions as VOCs related to occupant events  
363 deposit on surfaces and then slowly off-gas. The last mechanism has been traditionally  
364 understood to apply to semivolatile organic compounds (SVOCs), but recent studies suggest that  
365 sorption in surface reservoirs is also important for VOCs (22). That phenomenon may cause  
366 primary emissions from episodic sources to be interpreted as originating from building materials.

367 Source control efforts prioritize building materials and furnishings with the specific goal of  
368 encouraging use of materials with low short-term VOC emissions. Unified regulatory limits are  
369 generally not available, but some countries have implemented product certification and voluntary  
370 emissions testing programs. The H1 and H2 sites suggest that ongoing VOC emissions can be  
371 expected to occur from building materials long after any short-term VOC reservoirs are depleted.  
372 While we expect that these sites are ordinary for their age and location, further study is  
373 warranted to see if these findings generalize to other sites, especially those with higher  
374 concentrations of airborne VOCs. If exposures associated with long-term VOC emissions from  
375 building materials prove to be significant for occupant health and well-being, current control  
376 strategies may need augmentation. Some compounds associated with wood degradation, such as  
377 carboxylic acids, furanoids and acetaldehyde, may prove particularly challenging to address via  
378 source-control strategies of the type currently employed.

379 Past source apportionment studies using multivariate receptor models of population-level data  
380 have diverged on the relative importance of off-gassing from building materials (29–31). For  
381 instance, Bari et al. noted that off-gassing from building materials contributed only 6% of  
382 measured total VOCs ( $n = 119$  analytes; alkanes, alkenes, aromatics, halogens, carbonyls,  
383 alcohols, terpenes, others) in Canadian residences (31). Conversely, Guo reported that 77% of  
384 measured VOCs in Hong Kong residences were attributable to emissions from building materials  
385 ( $n = 16$  representative analytes; alkanes, aromatics, carbonyls, terpenes, acids, others) (30). Such  
386 differences between studies may reflect the choice of studied analytes rather than differences in  
387 building age and composition. Furthermore, dimensionality reduction techniques, such as  
388 positive matrix factorization (PMF) and principal component analysis (PCA), reduce complex  
389 datasets to specific factors or components unique to each dataset that must be interpreted and  
390 assigned to specific source categories. While such source apportionment studies yield  
391 qualitatively similar results reflecting commonalities in residential VOC sources, it is difficult,  
392 and often unjustifiable, to generalize quantitatively between different source apportionment  
393 analyses. Contrasting findings between the present study, indicating substantial contributions  
394 from continuous indoor sources, and past studies, indicating that off-gassing from aged building  
395 materials is less important, may be related to subjective choices for source assignment in  
396 multivariate receptor models. This observation emphasizes that detailed characterizations of  
397 single-residences are needed to complement and justify inferences from population-level  
398 surveys.

399 **Risk-based prioritization.** Chronic advisory health guidelines from at least one agency were  
400 acquired for 31 VOCs consistent with chemical formulas identified at the H1 and H2 field sites.  
401 Of these 31 VOCs, 18 were specifically identified at the H1 and H2 field sites with high degrees  
402 of confidence (Table 1). Considering residential exposures at these two study sites, three  
403 compounds exceeded a health-risk guideline given by IRIS, OEHHA, or ATSDR. Acrolein

404 concentrations were 1-2 orders of magnitude higher than the health-risk guidelines of IRIS and  
405 ATSDR, and ~3–4 times higher than the OEHHA guideline, for each occupant during all three  
406 campaigns. Acetaldehyde concentrations were above the USEPA IRIS health guideline and  
407 within an order of magnitude of the OEHHA health guideline for all occupants during all three  
408 campaigns. Acrylic acid concentrations were comparable to the USEPA IRIS health guideline  
409 for all occupants during all three campaigns. Health guidelines for acrylic acid were not available  
410 from OEHHA and ATSDR. Acrolein and acetaldehyde have already been identified as two of  
411 nine “priority hazards based on the robustness of measured concentration data and the fraction of  
412 residences that appear to be impacted.” (3)

413 We also consider acute advisory health guidelines from OEHHA and ATSDR for 15 VOCs  
414 consistent with chemical formulas identified at the H1 and H2 fields sites, all of which were  
415 identified with high confidence (Table S4). Acrolein concentrations exceeded acute  
416 concentration guidelines 1–2 times each week during the H1 summer and H2 winter campaigns,  
417 often in association with cooking. Exceedances were less common during the H1 winter  
418 campaign, occurring only twice over the five-week monitoring period. Three events during the  
419 H2 winter campaign produced acetaldehyde concentrations that exceeded OEHHA acute  
420 concentration guidelines. Benzene concentration spikes reached concentrations roughly one third  
421 that of the acute guideline value. A short-term spike in acetone concentration associated with the  
422 use of nail polish remover reached 20% of the acute guideline value during the H2 winter  
423 campaign. All other compounds with available acute concentration guidelines were typically 2–3  
424 orders lower than the guideline value.

425 Public health guidelines often incorporate large uncertainty factors to account for limitations  
426 in the available toxicological data. In some cases, this may lead to uncertainties “spanning  
427 perhaps an order of magnitude” as explicitly stated by USEPA IRIS. In the event that  
428 USEPA guidelines are an order-of-magnitude overestimate, several of the flagged VOCs  
429 would no longer be expected to be of health interest at observed levels. If guidelines are an-  
430 order-of-magnitude underestimate, additional compounds may be of health interest at  
431 observed levels.

432 **Table 1.** Chronic hazard assessment of select VOCs. Indoor concentrations are presented as  
433 time averages of measured living space concentrations for the periods that individual  
434 occupants are indoors at home.<sup>a</sup>

435 [see table on following page]

436 <sup>a</sup> Reference concentrations (USEPA IRIS), reference exposure levels (OEHHA), and  
437 minimal risk levels (ATSDR) are presented for chronic and intermediate (int.) VOC  
438 exposures. Ion formulas with uncertain confidence in compound assignment are  
439 designated with an asterisk. All other compounds are possible assignments. OEHHA and  
440 IRIS values were converted from mass per volume concentrations to mixing ratios by  
441 applying the ideal gas law and assuming standard conditions (298 K, 1 atm). The  
442 “Concentration-to-Guideline Ratio” is defined as the ratio between the greatest mean  
443 occupant exposure concentration among the campaigns and the smallest health guideline  
444 concentration. All concentrations are reported in ppb. Values marked with “#” are  
445 uncertain ion formula assignments that, if assigned correctly, may be exceeding health  
446 advisory guidelines or warrant additional interest.

VOC	Mean Abundance (ppb)									Compound Assignment			Health Guideline (ppb)				
	Ion Formula	H1 summer			H1 winter			H2 winter			Name	Confidence	Concentration Guideline Ratio	OEHHA REL	USEPA IRIS RfC	ATSDR MRL	
<b>C<sub>3</sub>H<sub>5</sub>O<sup>+</sup></b>	0.57	0.56	0.57	0.44	0.44	0.45	0.36	0.39	0.39	0.38	acrolein		66	0.15	0.0087	0.04 (int.)	
<b>C<sub>2</sub>H<sub>5</sub>O<sup>+</sup></b>	9.3	9.7	9.5	8.3	8.2	9.1	24	28	26	14	acetaldehyde		5.6	78	5		
<b>C<sub>3</sub>H<sub>5</sub>O<sub>2</sub><sup>+</sup></b>	0.37	0.37	0.37	0.29	0.29	0.31	0.27	0.29	0.29	0.27	acrylic acid		1.1		0.34		
<b>C<sub>6</sub>H<sub>7</sub><sup>+</sup></b>	0.24	0.24	0.24	0.46	0.46	0.46	0.31	0.33	0.33	0.27	benzene		0.15	94	9.4	3	
<b>C<sub>8</sub>H<sub>11</sub><sup>+</sup></b>	0.25	0.24	0.25	0.25	0.24	0.24	0.33	0.34	0.34	0.31	xylene		0.15#	160	2.3	50 (int)	
											ethyl benzene		0.0057	460		60	
<b>CH<sub>5</sub>O<sup>+</sup></b>	28	27	27	28	28	28	45	50	50	42	methanol		0.033	3100	1500		
<b>C<sub>3</sub>H<sub>7</sub>O<sup>+</sup></b>	11	11	11	9.9	10	10	30	39	39	19	acetone		0.03			1300	
											propionaldehyde	*	11		3.4		
<b>C<sub>7</sub>H<sub>9</sub><sup>+</sup></b>	0.36	0.34	0.35	0.75	0.76	0.75	1.3	1.3	1.3	1	toluene		0.016	80	1300	1000	
<b>C<sub>6</sub>H<sub>7</sub>O<sup>+</sup></b>	0.52	0.52	0.52	0.48	0.48	0.48	0.34	0.36	0.36	0.31	phenol		0.001	52			
<b>CCl<sub>3</sub><sup>+</sup></b>	0.17	0.17	0.17	0.13	0.13	0.13	0.14	0.13	0.13	0.14	chloroform		0.0085			20	
<b>C<sub>2</sub>H<sub>4</sub>N<sup>+</sup></b>	0.16	0.16	0.16	0.11	0.11	0.11	0.14	0.14	0.14	0.14	acetonitrile		0.0044		36		
<b>C<sub>6</sub>H<sub>15</sub>O<sub>2</sub><sup>+</sup></b>	<LOD	<LOD	<LOD	<LOD	<LOD	<LOD	0.05	0.053	0.053	0.043	2-butoxyethanol		0.0031	17			
											butoxyethanol		0.00027			200	
<b>C<sub>6</sub>H<sub>5</sub>Cl<sub>2</sub><sup>+</sup></b>	0.018	0.018	0.018	0.019	0.019	0.019	0.01	0.011	0.01	0.01	dichlorobenzene		0.0019	130	130	10	
<b>C<sub>7</sub>H<sub>9</sub>O<sup>+</sup></b>	0.18	0.17	0.17	0.15	0.14	0.14	0.1	0.11	0.11	0.1	cresols		0.0011	140			
<b>C<sub>4</sub>H<sub>9</sub>O<sup>+</sup></b>	0.76	0.76	0.76	0.75	0.74	0.78	1.3	1.5	1.5	1.2	ethyl methyl ketone		0.00088			1700	
											tetrahydrofuran	*	0.0022		680		
<b>C<sub>8</sub>H<sub>9</sub><sup>+</sup></b>	0.14	0.14	0.14	0.15	0.14	0.15	0.16	0.17	0.17	0.15	styrene		0.00085	210	240	200	
<b>C<sub>4</sub>H<sub>5</sub>O<sub>3</sub><sup>+</sup></b>	0.014	0.015	0.014	0.019	0.019	0.019	0.015	0.016	0.016	0.014	maleic anhydride	*	0.11#	0.17			
<b>C<sub>4</sub>H<sub>7</sub>O<sub>2</sub><sup>+</sup></b>	0.69	0.68	0.69	0.59	0.58	0.6	0.59	0.62	0.62	0.56	vinyl acetate	*	0.012	57	57	10 (int)	
<b>C<sub>3</sub>H<sub>4</sub>N<sup>+</sup></b>	0.022	0.021	0.021	0.022	0.022	0.022	0.034	0.035	0.035	0.033	acrylonitrile	*	0.038	2.3	0.92		
<b>C<sub>6</sub>H<sub>13</sub>O<sup>+</sup></b>	0.23	0.22	0.23	0.16	0.16	0.16	0.21	0.21	0.21	0.19	2-hexanone	*	0.032			7.3	
											methyl isobutyl ketone	*	0.00032			730	
											ethylmethylbenzene	*	NA	NA	NA	NA	
<b>C<sub>9</sub>H<sub>13</sub><sup>+</sup></b>	0.13	0.12	0.12	0.14	0.14	0.13	0.16	0.18	0.18	0.14	cumene	*	0.0023			80	
											trimethylbenzene	*	0.015			12	
<b>C<sub>5</sub>H<sub>9</sub>O<sub>2</sub><sup>+</sup></b>	0.54	0.53	0.54	0.31	0.31	0.31	0.31	0.32	0.32	0.28	acetylpropionyl		NA	NA	NA	NA	
											glutaraldehyde	*	27#	0.02			0.03 (int)
											methyl methacrylate	*	0.0032			170	

449 Source apportionment reveals that acrolein, acetaldehyde and acrylic acid at these study sites are  
450 principally attributable to continuous indoor sources with important contributions from cooking  
451 and smaller contributions from outdoor-to-indoor transport. During the H1 summer and H1  
452 winter campaigns, acetaldehyde exposures were principally attributed to continuous indoor  
453 sources (~60-70%) with contributions from cooking (~20-30%) and outdoor-to-indoor transport  
454 (~5-10%). In contrast, during the H2 winter campaign, continuous indoor sources accounted for  
455 half of occupant exposures, with significant contributions from both cooking (~20%) and  
456 unidentified indoor source events (~30%). Acrylic acid exposures were principally attributed to  
457 continuous indoor sources (~60-80%) with contributions from cooking (~5-20%) and outdoor-to-  
458 indoor transport (~15-25%) across the three campaigns. Acrolein exposures were principally  
459 attributed to continuous indoor sources (~70-90%) with contributions from cooking (~5-10%)  
460 and outdoor-to-indoor transport (~5-20%) across the three campaigns.

461 We highlight other VOCs with concentrations within an order magnitude of health guidelines as  
462 meriting further consideration. Concentrations of benzene were well below the OEHHA health  
463 advisory limit but are within an order of magnitude of the USEPA IRIS and ATSDR guidelines.  
464 Residential benzene exposures at these study sites were principally attributed to continuous  
465 indoor emissions (~50-60%) and outdoor-to-indoor transport (~40-50%), highlighting the  
466 importance of both indoor sources and outdoor emissions (which are likely traffic-related).  
467 Similar results are observed for other BTEX (benzene, toluene, ethylbenzene, xylene)  
468 compounds that have been the subject of significant interest and have been attributed to both  
469 indoor sources and outdoor fossil-fuel related sources (11).

470 There were no health guidelines available for ~90% of the 250 VOC ions observed. Furthermore,  
471 both VOC concentrations and associated toxicities can vary by orders of magnitude, indicating  
472 that it may not be sufficient to assume that there are minimal impacts on human health when  
473 compounds are present at relatively low concentrations. The possibility of additive or synergistic  
474 interactions arising from exposure to complex VOC mixtures may similarly influence health  
475 risks from air pollutant exposures in residences.

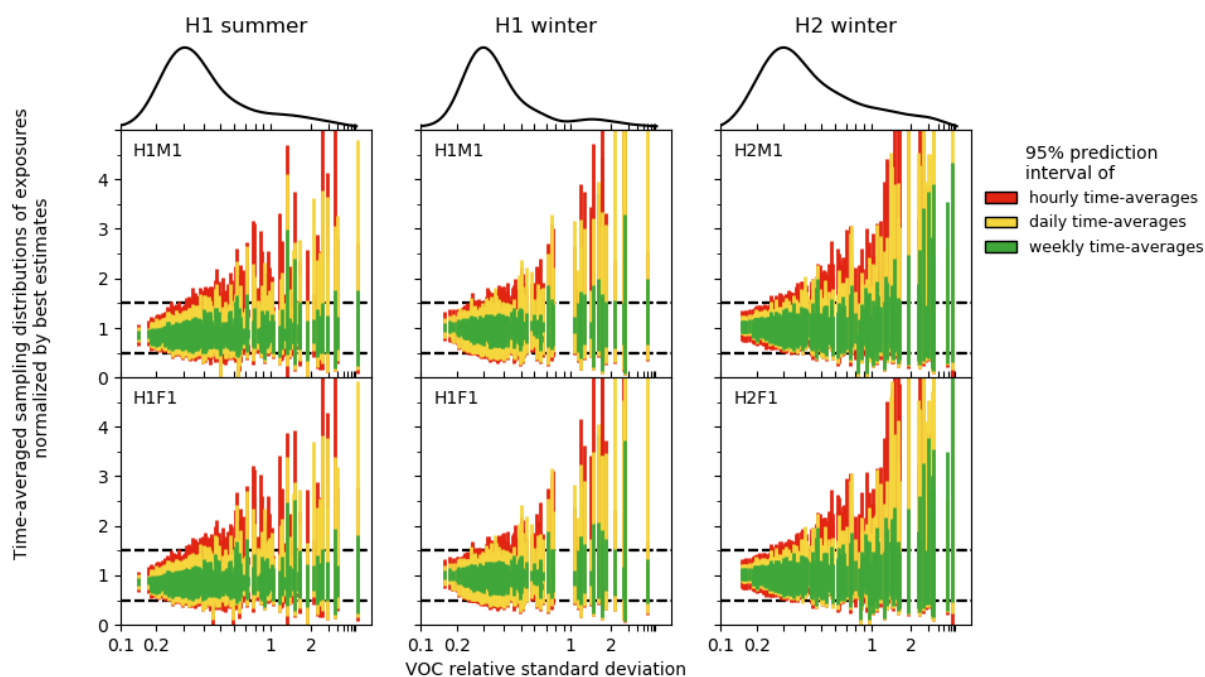
476 We were unable to conclusively assign some molecular ions to specific VOCs; these may  
477 represent an isomeric mixture of compounds with the same molecular formula but (potentially)  
478 different toxicities. We note that concentrations of the  $C_8H_{11}^+$ ,  $C_4H_3O_3^+$ , and the  $C_5H_9O_2^+$  ions, if  
479 attributed solely to xylene, maleic anhydride, and glutaraldehyde, respectively, would be of  
480 potential health-risk concern. Static measurements acquired by offline TD-GC×GC-ToF-MS at  
481 during the H2 winter campaign suggest that xylene isomers accounted for 80% of observed mass  
482 of  $C_8H_{11}^+$  with ethylbenzene accounting for the remainder. There was no evidence of any analyte  
483 with structural formulas consistent with  $C_4H_3O_3^+$  or  $C_5H_9O_2^+$  that might correspond to maleic  
484 anhydride or glutaraldehyde in the TD-GC×GC-ToF-MS measurements. However, given  
485 instrumental parameters, the presence or absence of maleic anhydride and glutaraldehyde was  
486 not assessable.

487 **Methodological considerations for future exposure studies.** Occupant exposures in their  
488 homes were determined using time-resolved data and compared against simulated time-averaged  
489 methods using the H1 summer, H1 winter and H2 winter datasets. Time-averaged samples



490 collected on sorbent tubes for later offline analysis commonly use sampling durations in the  
491 range of hours to weeks. We constructed simulated distributions of time-averaged samples using  
492 three hypothetical sampling durations (1-hour, 1-day, 1-week). Then, we normalized each value  
493 by the time-resolved best estimate of exposure. A value of “1” indicates that the time-averaged  
494 exposure estimate equals the time-resolved exposure estimate. Summary statistics of these  
495 distributions are compared against the relative standard deviation (RSD) of observed VOC  
496 concentrations for each compound during the three measurement campaigns (Figure 4). As a  
497 measure of variability, the RSD will increase with contributions from emission patterns that are  
498 both periodic (typically related to the temperature-based diurnal cycle) and episodic (typically  
499 related to emission events). Because the mean-to-median ratio (18), a measure of episodic  
500 variability, strongly correlates with the RSD ( $R^2 = 0.86$ ), it is qualitatively assumed that episodic  
501 variability is more important than periodic variability at this site. For VOCs with small RSDs  
502 ( $RSD < 0.4$ ) indicating low periodic fluctuations and low episodic emissions, all three sampling  
503 durations tend to produce exposures that are consistent (within  $\pm 50\%$ ) with best-estimate  
504 exposures. However, as the RSD of the VOC concentration time series increases, so too would  
505 the variability in occupant exposures derived from time-averaged samples. These effects can be  
506 particularly important when a sampling period directly coincides with an infrequently occurring  
507 episodic emission, which would produce exposures estimates potentially much higher than the  
508 best estimate based on time-resolved information. In the illustrative case of pyrrole during the  
509 H2 winter campaign, estimated exposures were up to 100 times higher for occupant H2M1 than  
510 the best-estimate when an hourly time-averaged sample coincided with an episodic cooking  
511 event that released pyrrole. The effects of this bias decreased with increasing sampling duration.  
512 Pyrrole exposures estimated from daily time-averaged samples were up to 13 times higher than  
513 the time-resolved best estimate of exposure and pyrrole exposures from weekly time-averaged  
514 samples were up to 3.3 times higher than the time-resolved best estimate of exposure.

515 Hourly time-averaged samples produce unsatisfactory estimates of exposures ( $> \pm 50\%$  of best-  
516 estimate) for most VOCs with moderate-to-high RSDs ( $RSD > 0.4$ ). The performance of the  
517 time-averaged method slightly improved when sampling durations increased to daily but was  
518 similarly unsatisfactory for compounds with moderate-to-high RSDs ( $RSD > 0.4$ ). Only when  
519 the time-averaged method increased to weekly sampling intervals did performance appear  
520 satisfactory for most VOCs ( $\sim 90\%$  of VOCs during H1 summer and H1 winter,  $\sim 75\%$  during H2  
521 winter), yielding exposure estimates that were within 50% of the best-estimate. However, even  
522 with this longer sampling duration, the time-averaged and best-estimate time-resolved methods  
523 were less comparable for VOCs with high RSDs, mostly due to the large influence of episodic  
524 events.



525  
 526 **Fig. 4. Comparison of traditional and time-resolved exposure estimates against**  
 527 **compound time-series variability.** Occupant exposures during the H1 summer, H1  
 528 winter, and H2 winter campaigns were calculated from a) simulated time-averaged  
 529 concentrations (1-h, 1-day, 1-week time-averages) coupled with occupant time budgets  
 530 and b) time-resolved methods ('best-estimate'). Distributions of randomly sampled time-  
 531 averaged exposure estimates ( $n = 1000$ ) normalized by the time-resolved best-estimate  
 532 are compared against a VOC's relative standard deviation. Vertical bars correspond to the  
 533 95% observation window for individual VOC distributions (with one outlier excluded in  
 534 each campaign). Red, orange, and green colors indicate distributions of hourly, daily, and  
 535 weekly time-averages, respectively. Columns correspond to the H1 summer, H1 winter,  
 536 and H2 winter campaigns. Subpanels in each column correspond to exposures for  
 537 individual occupants during a campaign. The kernel density estimate of the relative  
 538 standard deviation is shown above each scatterplot. Horizontal dashed lines designate the  
 539  $\pm 50\%$  boundary of the best-estimate exposure. Note that the horizontal axis has non-  
 540 linear scaling.

541 Population-level environmental risk assessments require the ability to estimate exposures with  
 542 both adequate precision and ease of implementation. Often, samples are collected on sorbent  
 543 tubes for offline analysis giving time-averaged VOC concentrations and then coupled with  
 544 occupant time budgets to estimate pollutant exposures. A comparison of time-averaged methods  
 545 with time-resolved methods has not been previously reported to substantiate this approach. In  
 546 this study, time-averaged sampling methods appeared to yield adequate measures of occupant  
 547 exposure, provided that the studied compound of interest has proportionately small contributions  
 548 from episodic sources. As expected, longer sampling times produce more precise estimates of  
 549 occupant exposures. Hourly and daily time-averaged samples would be ineffective at estimating  
 550 exposures to VOCs with substantial contributions from episodic sources. Weekly time-averaged

551 samples can yield more precise (+/- 50%) estimates of exposures for VOCs with moderate  
552 contributions from episodic sources and order of magnitude precision for compounds with  
553 substantial contributions from episodic sources. Conversely, sampling duration matters little for  
554 compounds with small contributions from episodic sources, indicating that many VOCs can be  
555 measured rapidly for exposure estimates, provided that source mechanisms are well understood.

556 VOCs were sampled in two or three locations in the occupied portion of each studied residence  
557 to provide some evidence about spatial variability. On the whole, concentrations were well-  
558 coupled between open living spaces. It has been noted that the assumption of a well-mixed room  
559 may not be sufficient when estimating exposures to strongly localized sources due to spatial  
560 heterogeneities. (33, 34) For this reason, contributions from episodic sources to occupant  
561 exposures within these residences may have been higher if an occupant was in close proximity to  
562 an episodic source, regardless of whether time-averaged or time-resolved methods were used to  
563 estimate exposures. Personal sampling overcomes this spatial limitation; however, no methods  
564 are presently available for conducting personal sampling for VOC exposures with both high  
565 chemical specificity and time-resolved concentrations.

566 This study incorporated health-risk data from three governmental organizations yielding advisory  
567 guidelines for chronic exposures to 31 separate VOCs (22 VOC ions). We compared the  
568 guideline values against VOC concentrations for summer (one residence) and winter seasons  
569 (two residences). Residential VOC concentrations were above guideline values for two  
570 compounds already well-known to the public health community (acrolein, acetaldehyde) and for  
571 one compound for which there is less evident awareness (acrylic acid). We also note that  
572 available advisory health guidelines address fewer than 10% of the 250 VOC ions observed.  
573 Larger VOC ions likely correspond to many distinct chemical isomers, further reducing the  
574 estimate of observed VOCs with associated health data. Because it is not feasible to acquire  
575 detailed toxicology or exposure data for the thousands of chemicals to which populations may be  
576 exposed, recent efforts have developed tools for high-throughput screening of chemical toxicities  
577 (ToxCast) and exposures (ExpoCast).<sup>56</sup> High-throughput models like ExpoCast may benefit from  
578 improved understanding that can be gained in high-resolution studies. This study was limited to  
579 two residences and five occupants. While we expect that these residences are ordinary for their  
580 age and location, it remains to be seen how well our findings generalize to other residences.  
581 Further research is warranted to study the health effects of the substantial chemical diversity  
582 found at levels much higher than outdoors in residential air.

## 583 4 SUPPORTING INFORMATION

584 Supporting Information: PTR-ToF-MS operating conditions, data analysis, absolute daily  
585 exposures, occupant time-activity budgets, seasonal concentration differences, H2  
586 residence floor plan, representative time series of peak assignments, summary data tables,  
587 acute health guidelines.

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599

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