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## 1 The role of surface forces in environment-enhanced cracking of brittle solids

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### 3 ABSTRACT

4 Fracture initiation and propagation in brittle materials is promoted in surface-reactive (sorptive) 5 environments, a phenomenon known as subcritical crack growth (SCG). Laboratory measured crack-6 propagation velocity vs. stress intensity factor relationships typically exhibit highly nonlinear, multi-7 stage characteristics that are sensitive to environmental factors such as adsorbate concentration and 8 temperature. For practical purposes, empirical relationships (e.g., a power law) have been used to 9 describe this complex phenomenon. However, how the overall SCG behavior emerges from the 10 underlying fundamental processes near the crack tip, such as the interaction of the crack surfaces 11 separated by only a few nanometers and mass transport within the nano-confined space, is still not well 12 understood. This paper develops a mechanistic, surface-force-based fracture theory (SFFT) which 13 integrates surface force models, fluid transport models, and linear elastic fracture mechanics to 14 quantitatively explain the multi-stage characteristics of SCG in brittle solids. A numerical model is 15 developed based on SFFT and solved through an implicit partitioned scheme for efficiency and 16 modularity. The results are validated by Wiederhorn's data on crack propagation in soda-lime glasses at 17 a wide range of relative humidity levels. We show that, for the first time, the entire range of an SCG 18 curve can be captured by a single physics-based model. The predicted SCG curves reveal that the 19 development of repulsive disjoining pressure behind the crack tip can be responsible for the reduced 20 apparent fracture toughness in a sorptive environment. The shape of the SCG curve, and its changes with 21 respect to the environment, is found to critically depend on the assumed transport models.

22 Key words: fracture, kinetics, subcritical crack growth, surface force, sorption.

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#### 23 1. BACKGROUND

24 Subcritical crack growth plays a crucial role in the long-term mechanical performance of natural and 25 engineered materials. In sorptive environments, or as referred to by Rice (1978) the surface-reactive environments, stable crack growth with a finite velocity  $(v_c)$  can occur at stress intensity factors  $(K_1)$ 26 significantly lower than the critical one  $(K_{IC})$  measured under the apparent 'dry' or vacuum conditions. 27 28 This phenomenon is known as subcritical crack growth (SCG) (Atkinson, 1982) or static fatigue 29 (Charles, 1958). SCG controls the rate of important geological processes such as the diagenesis of 30 granular rocks (Chester et al., 2007) and weathering (Eppes and Keanini, 2017). Because it can be 31 viewed as a precursor to abrupt fracturing, understanding and modeling SCG can help improve the 32 prediction of volcanic eruptions (Kilburn and Voight, 1998) and fault failure (Lennartz-Sassinek et al., 33 2014). SCG may also contribute to the slow degradation of engineering materials (such as concrete, 34 glass, metal, sand, rock), which is important in civil and mechanical engineering practice (Freiman et al., 35 2009; Karimpour and Lade, 2010; Le et al., 2009; Simmons et al., 1978; Zhang and Buscarnera, 2018).

36 The key challenge of quantitative modeling of SCG is rooted in its nonlinear multistage behavior 37 resulted from complex physiochemical interactions at various length scales. SCG typically exhibits three 38 distinct stages in the  $v_c - K_I$  space (Fig. 1a). At relatively low  $K_I$  (stage I),  $v_c$  increases rapidly with 39  $K_1$  due to solid-environment interactions; at intermediate  $K_1$  (stage II),  $v_c$  is believed to be controlled by the transport of the active species; stage II continues until a sudden and rapid increase of  $v_c$  (stage III) 40 as  $K_1$  exceeds the intrinsic fracture threshold ( $K_{1C}$ ) of the material in vacuum (Freiman et al., 2009; 41 42 Wiederhorn, 1967). The stage I and II curves strongly depend on environmental conditions such as 43 temperature and reactant (adsorbate) concentration in the environment (Fig. 1b) (Crichton et al., 1999; 44 Wiederhorn et al., 1980). In this context, our SCG model aims to address two key questions: (Q1) how

45 do solid and environment interact to cause the reduced fracture toughness (from  $K_{1C}$  to  $K_{10}$  in Fig. 1a) 46 and the associated exponential kinetics in stage I? and (Q2) what transport mechanism (or combination 47 of mechanisms) is at work near the crack tip to effectively limit the crack velocity in stage II?







Fig. 1. (a) Typical SCG curve and the two fundamental questions. (b) SCG data of soda-lime glass at different relative humidity, RH (Wiederhorn, 1967; Wiederhorn et al., 1980).

52 Regarding the first question, many hypotheses have been proposed for different material-53 environment systems. Energy-based approaches explain the reduction of apparent fracture toughness 54 through the decreased surface energy caused by the adsorption of fluid species on the crack surfaces 55 (Rice, 1978; Wan et al., 1990). In contrast, atomistic approaches focus on crack-tip chemical reactions 56 such as hydrolysis of stretched crack-tip bonds, dissolution and transport of crack tip material, and ion 57 exchange with the crack tip (Atkinson, 1984; Michalske and Freiman, 1982). Concurrent with these 58 mechanisms are the crack kinetic models by making analogy between crack propagation and reaction kinetics (Bazant and Planas, 1997; Lawn, 1975). Alternatively, empirical  $v_c - K_I$  relationships such as 59 60 power law are used as a descriptive tool without considering the underlying mechanisms driving the 61 crack propagation (Bazant and Planas, 1997; Charles, 1958).

For the second question, some mechanisms such as Fickian diffusion (Wiederhorn, 1967), Knudsen
diffusion (Lawn, 1974), and surface diffusion (Crichton et al., 1999) have been proposed to govern the

64 fluid invasion along a crack. However, the temperature-dependence of SCG data observed in phosphate 65 laser glasses suggests that none of these mechanisms alone can fully describe the stage II behavior 66 (Crichton et al., 1999). Mass transport in nano-confined spaces is still an active area of research (Choi et 67 al., 2001; Cihan et al., 2019) and is poorly understood at this moment for the near-crack-tip region. For these reasons, to this day, the modeling of stage II and the transition from stages I to II remains highly 68 69 simplistic and phenomenological. Frequently, the stage II and III behaviors are neglected altogether, and only power-law relationships are fitted to the  $v_c - K_I$  data obtained in stage I, across a range of stresses 70 71 and environmental conditions (Brantut et al., 2013; Eppes et al., 2018; Nara et al., 2012; Olson, 1993).

72 Recently, atomic force microscope (AFM) and surface force apparatus (SFA) experiments revealed 73 that repulsive forces (e.g., electrical double layer, hydration, crystallization) can become dominant 74 between mineral surfaces at nanometer separations in the presence of sorptive species (Dziadkowiec et 75 al., 2018; Røyne et al., 2015). Specifically, SFA measurements and molecular dynamic simulations (MD) 76 suggest that the nano-confinement of solutions between calcite surfaces can exert repulsion over 77 hundreds of nanometers of separation (Diao and Espinosa-Marzal, 2016; Dziadkowiec et al., 2019). 78 AFM measurements confirm that forces of similar range exist between glass surfaces in aqueous 79 solutions (Acuña and Toledo, 2008), and their magnitudes depend on the chemical composition of the 80 environmental fluid (Adler et al., 2001). This inspires an alternative explanation of the complex SCG 81 behavior: the repulsive forces along the crack walls can possibly facilitate the initiation and propagation 82 of cracks in sorptive environments, resulting in the characteristic trend of an SCG curve in Fig. 1. 83 Although the potential of surface forces in affecting solid fracture was debated in the glass community a 84 few decades ago (Lawn, 1985; Wiederhorn and Fuller Jr, 1989), it has been largely neglected since then, 85 without much progress.

#### 87 2. A SURFACE-FORCE BASED FRACTURE THEORY

88 Provided the renewed evidence of repulsive surface forces induced by surface sorption, we propose 89 an SCG theory to assess a mechanistic, surface-force-based answer to both questions. The key strategy 90 is illustrated in Fig. 2 and explained in the following. In this model, surface forces along crack surfaces 91 are accounted for via a force-separation relation. For a homogenous material in perfect vacuum,  $\Pi(w)$ 92 is defined by the force required to overcome the solid-solid intermolecular forces in separating two 93 surfaces and is the same everywhere along the crack. The interacting force at different location x along 94 the crack surface can be read from this  $\Pi(w)$  curve given the local crack separation w(x). The 95 resistance against crack opening provided by the attractive surface force is macroscopically manifested as the fracture toughness  $K_{IC}$  in classical fracture mechanics (Meng and Thouless, 2019). This part is 96 97 similar to the well-established cohesive crack theory (CCT), where the cohesive forces in the fracture 98 process zone (FPZ) counterbalance the far-field tensile stress to give a net-zero stress intensity factor at 99 the crack tip (Barenblatt, 1962).

100 Now consider that the crack is invaded by sorptive species, resulting in a non-uniform distribution of 101 species concentration C = C(x) along the crack as shown at the bottom of Fig. 2. Intrusion of the 102 sorptive molecules can shift the local  $\Pi(w)$  towards the repulsive side (Eskandari-Ghadi and Zhang, 2022), the magnitude of which depends on the local chemistry C(x). The surface force at different 103 104 locations therefore starts to travel along different  $\Pi(w)$  curves as opposed to the vacuum case. The 105 C(x) profile is dictated by the species transportation along the crack and is thus coupled with the crack 106 opening profile w(x), the ambient concentration  $C_0$ , and the crack velocity  $v_c$ . The proposed theory can 107 potentially explain the first question (Q1) through the reduced attraction or the development of repulsion

- 108 due to the presence of sorptive species. The second question (Q2) can be also directly addressed through
- 109 the coupling between fracture propagation and species transport in cracks.





Fig. 2. The key elements of the proposed surface-force based fracture theory.

112 The proposed theory operates at an intermediate scale which is larger than the atomistic theories 113 without relying on the dynamics of the crack-tip bond but is smaller than the macroscopic energy 114 approaches by requiring the full resolution of crack opening profile and stress fields near the crack tip. It 115 also differs from the cohesive crack theory in that, the traction curve in CCT is a constitutive relation commonly fitted to macroscale fracture testing data, involving only attractive force, and does not vary 116 117 along the crack. In contrast, the proposed approach is rooted in surface physics, involves intermediate-118 scale force-separation relationships which are either directly measurable by SFA experiments or 119 theoretically derivable based on intermolecular potentials, and can be strongly repulsive and vary with 120 the local environment. We therefore refer to the proposed theoretical framework as the Surface-Force 121 based Fracture Theory (SFFT) to distinguish from the previous works.

122 The remainder of this paper shall tackle two specific goals: (1) to formalize the mathematical 123 structure of SFFT for mode-I fractures and (2) to numerically implement an SFFT model and test its 124 ability in capturing typical SCG behaviors. For demonstration, the model will be specialized for the 125 glass-in-vapor experiments of Wiederhorn (1967) as our first step towards advancing our general 126 understanding of environment-enhanced cracking of brittle materials. We start by laying out the stress 127 intensity, opening profile, and propagation kinetics of cracks with considering both far-field stress and 128 distributed stresses along the crack surface (Section 3). We then formulate simplified analytical models 129 for surface force under variable environment (Section 4) and species transport along a moving non-130 parallel slit (i.e., a propagating crack) (Section 5). We numerically solve the system via a partitioned 131 implicit scheme to ensure a robust and efficient coupling between the aforementioned processes (Section 132 6). Finally, the model performance is assessed using the SCG data of soda-lime glass at different 133 humidity levels as presented in Fig. 1b (Section 7). The main conclusions and possible future extensions of this work are discussed at the end (Section 8). 134

## 135 3. FRACTURE MODELING CONSIDERING SURFACE FORCES

#### 136 **3.1. Fracture mechanics analysis**

137 We adopt the approach of Lawn (1985) by directly acknowledging the disjoining pressure ( $\Pi$ ) in the LEFM analysis. The *disjoining pressure*  $\Pi$  is here defined as the net surface force between two 138 139 solid surfaces normalized by the surface area (Clarke et al., 1986). It lumps the contributions of all solid-140 fluid interactions together without discerning their physiochemical origins and takes a positive value 141 when the net force is repulsive (or disjoining) and negative when cohesive (or joining). Such distributed 142 pressure along the crack surface can alter the stress field, characterized by the *effective* stress intensity 143 factor  $(K_{le})$ , in the vicinity of the crack tip.  $K_{le}$  should be distinguished from the applied stress intensity 144  $(K_{Ia})$  which solely accounts for the external forces or far-field stresses on the cracked solid.

Let us consider a 2D (plane-strain) edge crack of finite length *a* (Fig. 3a) in an elastic half-space subject to far-field tensile stress and disjoining pressure  $\Pi(x)$  over the crack surface. The boundary conditions are selected such that the problem is equivalent to an infinite body with a line crack (Irwin, 148 1957) (Fig. 3b). This allows one to take advantage of the readily available fundamental solutions to calculate the mode-I crack opening and stress intensity under various loading combinations (Tada et al., 2000). The dependency of  $\Pi(x)$  on fluid chemistry and species transportation will be addressed in section 4.



Fig. 3. Schematic of the model geometry and loading scenarios. (a) A finite edge crack in semi-infinite elastic domain. (b) A
 finite center crack in elastic infinite domain. The boundary conditions in (a) guarantee equivalent stress-deformation solution as (b).

## 156 LEFM allows one to write the effective stress intensity factor in an additive fashion

157 
$$K_{Ie} = \underbrace{k_{I\sigma}\sigma}_{K_{Ia}} + \underbrace{\int_{s=0}^{s=a} k_{I\Pi}(s)\Pi(w(s);s)ds}_{K_{I\Pi}}$$
(1)

where  $k_{I\sigma}$  is the stress intensity factor caused by unit far-field tensile stress;  $\sigma$  is the magnitude of the tensile stress;  $k_{I\Pi}(s)$  is the stress intensity factor caused by unit repulsive force on the crack surface at location s; w(x) is the crack opening at location x;  $\Pi(w)$  describes a disjoining pressure vs. separation relation which will be given by the surface force model (see Section 4);  $K_{Ia}$  is the applied 162 stress intensity factor;  $K_{I\Pi}$  is the stress intensity contribution by the disjoining pressure and takes a 163 negative value when the net effect of  $\Pi$  is cohesive.

For a given  $\sigma$ , the distribution  $\Pi(x) = \Pi(w(x); x)$  is not known a priori.  $\Pi$  depends on the surface separation while its change in turn alters the crack opening profile. The expression of w(x) is given below based on LEFM (Bazant and Planas, 1997):

167 
$$w(x) = c_{\sigma}(x)\sigma + \int_{s=0}^{s=a} c_{\Pi}(x,s)\Pi(w(s);s)ds$$
(2)

168 where  $c_{\sigma}(x)$  is the crack opening at location x along the crack caused by unit far-field tensile stress; 169  $c_{\Pi}(x,s)$  is the crack opening at the same location caused by a unit repulsive force at location s.

170 For the geometry sketched in Fig. 3a and b,  $k_{l\sigma}$  and  $c_{\sigma}(x)$  are given by Tada et al. (2000)

171 
$$k_{1\sigma} = \sqrt{\pi a} \text{ and } c_{\sigma}(x) = \frac{4(1-\nu^2)}{E}\sqrt{a^2 - x^2}$$
 (3)

## 172 Similar fundamental solutions can be found for unit surface force applied at *s* :

173 
$$c_{\Pi}(x,s) = \frac{8(1-\nu^2)}{\pi E} \begin{cases} \tanh^{-1}\sqrt{\frac{a^2-s^2}{a^2-x^2}} & ; & x < s \\ \cosh^{-1}\sqrt{\frac{a^2-s^2}{a^2-x^2}} & ; & x > s \end{cases}$$
(4)

174 
$$k_{I\Pi}(s) = 2\sqrt{\frac{a}{\pi}} \frac{1}{\sqrt{a^2 - s^2}}$$
(5)

175 It should be noted that Eqns. (4) and (5) are singular at x = s. This introduces difficulties to the 176 numerical integrations of Eqns. (1) and (2). We have circumvented this difficulty by means of numerical 177 approximation which allows the usage of non-singular analytical solutions (see Appendix A for details). The non-linear coupling between w(x) and  $\Pi(w)$  can be assembled as a system of two algebraic equations, referred to as the  $\Pi - w$  system hereafter, to be solved simultaneously (see Appendix A). The solution to the  $\Pi - w$  system provides a crack opening profile w(x) which will be substituted into Eq. (1) in conjunction with the  $\Pi(w)$  model (see Section 4) to obtain  $K_{le}$ .

## 182 **3.2.** Crack propagation kinetics

The rate at which crack propagates can be studied in the framework of reaction rate theory (Charles, 184 1962). Griffith's criterion suggests that non-dissipative crack propagation occurs once  $(\mathbf{G}-2\gamma) > 0$ , 185 where **G** is known as the energy release rate and  $\gamma$  is the surface tension of the newly created crack 186 faces. By analyzing the statistics of bond breakage and healing under the energy difference  $(\mathbf{G}-2\gamma)$  in a 187 way similar to a chemical reaction driven by chemical potential differences (Lawn, 1975), one can arrive 188 at the following kinetic equation

189 
$$v_c = v_0 \exp\left(-\frac{Q_0}{RT}\right) \sinh\left(\frac{\tilde{\kappa}_0(\mathbf{G}-2\gamma)}{RT}\right)$$
(6)

190 where  $\tilde{\kappa}_0$  is a proportionality factor;  $Q_0$  is the activation energy for bond breaking;  $v_0$  contains 191 information of the solid lattice vibration frequency and lattice length-scale (Meng and Thouless, 2019). 192 In Eq. (6), the "driving force" for crack propagation is  $(\mathbf{G}-2\gamma)$ . Bazant and Planas (1997) have shown 193 that stress intensity factor can be also taken as the driving force to replace  $(\mathbf{G}-2\gamma)$  in Eq. (6). Inspired 194 by this, we propose a slightly different equation for crack growth in the presence of disjoining pressure

195 
$$v_c = v_0 \exp\left(-\frac{Q_0}{RT}\right) \sinh\left(\frac{\kappa_0 K_{Ie}}{RT}\right)$$
(7)

196 where  $\kappa_0$  is a proportionality factor related to  $\tilde{\kappa}_0$ . The decomposition of  $K_{le}$  to  $K_{la}$  and  $K_{I\Pi}$  (i.e., Eq. 197 (1)) reveals some connections between Eqns. (6) and (7). Firstly, the celebrated Irwin equation directly

links  $K_{la}$  and  $K_{l\Pi}$ , respectively, to **G** and  $2\gamma$  at equilibrium.  $K_{le} = 0$  implies that  $\mathbf{G} = 2\gamma$  and  $v_c = 0$ , 198 199 agreeing that a non-propagating crack attains zero stress intensity and thus finite tensile stress at the tip 200 (Barenblatt, 1962). Secondly, the generation of repulsive disjoining pressure in sorptive environment 201 will reduce  $K_{III}$  (Eq. (1)), thus promoting crack propagation under the same applied  $K_{la}$ . This is equivalent to the surface energy reduction mechanism discussed by Rice (1978). We further argue that 202 203 Eq. (7) has the unique advantage of directly incorporating the nonuniform reduction of surface energy along the crack (via  $K_{III}$ ) caused by transient processes such as fluid transport, the inclusion of which in 204 205 energy-based descriptions is not straightforward.

206 4. SURFACE FORCE MODEL

The magnitudes of surface forces depend on the crack aperture w (Bazant and Planas, 1997) as well as the local environment C (Israelachvili, 2011). The net effect of these forces over unit surface area can be collectively encapsulated in the disjoining pressure  $\Pi(w, C)$ , the modeling of which is essential for the proposed fracture theory. Before tackling this task, we should clarify the ambiguity related to the definition of "crack tip".

212 From an atomistic perspective, the crack tip may be defined at the location where the solid chemical 213 bonds are broken. Thermodynamically, one may define the crack tip as the point behind which the 214 separation of opposite surfaces is irreversible and involves energy dissipation, and in front of which the 215 separation occurs reversibly or elastically. Indeed, the location of the "crack tip" behind and ahead of 216 which different inelastic processes may be at work is only a matter of perspective and convenience (Sills 217 and Thouless, 2015). From a continuum point of view, undamaged material obeys elastic behavior, and 218 the damaged material in the fracture process zone (FPZ) obeys a softening stress-strain relation (Bažant 219 and Oh, 1983). The point of divergence between continuum and localized crack happens at the

maximum value of traction on the force-separation curve. This motivates us to define the crack tip as the point of maximum traction between two surfaces, and the corresponding crack separation at this reference point is taken as w=0 (See Fig. 4).

223 The consideration of purely cohesive interactions (traction) over several micrometers behind crack 224 tip leads to the well-known cohesive crack theory. It has been successfully applied, for example, to 225 remove the stress singularity predicted by LEFM at the crack tip (Barenblatt, 1962), model the kinetics 226 of crack growth (Meng and Thouless, 2019), and capture crack fatigue (Allegri, 2020; Nguyen et al., 227 2001). It also hosts various material-specific fracture models by treating the traction-separation curve as 228 a constitutive relation (Hillerborg et al., 1976; MAI and Lawn, 1987; Needleman, 1990; Tijssens et al., 229 2000; Tvergaard and Hutchinson, 1992). However, the true surface forces originated from the physical 230 separation (<100 nm) of solid surfaces (electrostatic, van der Waals, etc.) and environmental attack 231 (surface hydration, electrical double layer, etc.) has been mostly neglected in classical fracture analysis. 232 The goal here is to substantialize a  $\Pi(w, C)$  model to capture these nanometer-scale interactions which 233 can be attractive or repulsive.

Generally, the disjoining pressure between two solid surfaces in arbitrary environment can be decomposed into two contributions:

$$\Pi(w,C) = \Pi_0(w) + \Delta \Pi(w,C) \tag{8}$$

where *C* stands for the concentration of the sorptive species in between surfaces. *C* could be generalized to a vector  $\mathbf{C} = [C_1, C_2, ..., C_k]$  should multiple sorptive species present in the environment. The first component,  $\Pi_0$ , accounts for the intrinsic surface forces between two solid surfaces in vacuum; the second term,  $\Delta \Pi$ , collects the change of disjoining pressure due to solid-species interaction (Adapa and Malani, 2021; Dziadkowiec et al., 2018; Røyne et al., 2015).

## 242 **4.1. Surface force in vacuum** $\Pi_0(w)$

 $\Pi_0(w)$  may be constructed through smooth interpolations of experimental data provided by SFA/AFM measurements, or rigorously derived with the knowledge of the total intermolecular pair potentials of the solid (Israelachvili, 2011). Here we opt for a well-behaved and versatile analytical expression to represent the in-vacuum surface forces of generic materials. Simple analytical models also benefit the numerical implementation in terms of both convergence and efficiency. A two-parameter equation is adopted for this purpose:

249 
$$\Pi_0 = -\Pi_{0,\max} \frac{w + w_0}{w_0} \exp\left(1 - \frac{w + w_0}{w_0}\right)$$
(9)

where  $\Pi_0$  attains maximum attraction  $\Pi_{0,max}$  at w=0 and decays exponentially for w>0, the rate of which is controlled by  $w_0$ . Eq. (9) also contains an exponentially increasing repulsive branch for w<0, which physically corresponds to the steric repulsion when molecules are brought too close together (Israelachvili, 2011). This feature is also expected to be numerically beneficial by serving as a penalty contact to avoid the overlap of two surfaces when they come to contact, in cases where crack healing may<sub>le</sub>occur (i.e., K < 0). Eq. (9) is visualized in Fig. 4 for three sets of parameters.

The solution of Eqns. (2) and (9) together allows us to study crack opening at different levels of applied far-field stress in vacuum. Using the numerical method detailed later in Section 6 and parameters in Table 1, the *w* and  $\Pi_0$  profiles for a crack of length 36mm under increasing stresses are plotted in Fig. 5. When the stress is low, the in-vacuum surface force contribution of stress intensity  $K_{I\Pi}$ can effectively balance out the applied stress intensity  $K_{Ia}$ , thus gives a low  $K_{Ie}$  value and a nearequilibrium crack per Eq. (7). This results in a smoothly converging crack tip profile (the solid blue curve in Fig. 5a). This is consistent with Barenblatt (1962) who analytically showed that the crack opening profile in the immediate vicinity of the crack tip becomes proportional to the 1.5 power of the distance from the crack tip when  $K \rightarrow 0$  (the dashed cyan curve). As the applied force increases, the crack is perturbed further away from equilibrium, and its tip geometry becomes blunt and approaches an elliptical shape. Besides the near-tip region, the overall crack opening profile (Fig. 5b) does not vary significantly for the range of applied  $\sigma$ , except for a proportional widening of the crack with increasing  $\sigma$ . Fig. 5c shows that the resultant  $\Pi_0$  increasingly concentrates at the crack tip as the applied stress increases and the crack aperture widens up.



270 w (mm) 271 Fig. 4. The in-vacuum surface force described by Eq. (9).  $\Pi_{0,\max}$  controls the magnitude of  $\Pi_0$  while  $w_0$  controls its spread 272 along w.



274 275 in vacuum for progressively increasing far-field tensile stress. The model parameters used here are listed in Table 1. Figures 6a and 6b plot the computed  $K_{le}$  and  $K_{I\Pi}$  against a wide range of applied  $K_{la}$  using the 277 same parameters as Fig. 5. As expected, the presence of cohesive forces results in a reduced  $K_{le}$ 278 compared to  $K_{la}$ . Increasing  $K_{la}$  results in a decrease of  $K_{I\Pi}$  (Fig. 6b), and thus  $K_{le}$  asymptotes 279 towards  $K_{la}$  (Fig. 6a). Lawn (1985) analytically derived an expression of  $K_{I\Pi}$  as  $K_{I\Pi} = -E'\gamma/K_{le}$ , 280 where  $E' = E/(1-v^2)$  for plane-strain, by neglecting the dependence of crack opening profile on  $\Pi_0$ . 281 For the selected parameters, this approximation is surprisingly accurate for large  $K_{Ia}$  values but deviates 282 from our exact solution of the full  $\Pi - w$  system at lower  $K_{la}$  levels (Fig. 6b). This implies that the 283 284 consideration of the two-way coupling between  $\Pi$  and w is essential for studying the onset of SCG which always occurs at low  $K_{Ia}$  values. Fig. 6c shows that  $K_{III}^2 / E'$  asymptotes to  $2\gamma$  as equilibrium 285  $(K_{le}=0)$  is approached. This further supports our arguments that Eq. (7) can be viewed as the 286

287 mechanistic counterpart to Eq. (6), and the Irwin's equation respectively links  $K_{Ia}$  and  $K_{III}$  to G and

288  $2\gamma$  at equilibrium.



291  $-K_{I\Pi}$  versus  $K_{Ia}$  to compare with the approximate analysis by Lawn (1985); and  $K_{I\Pi}^2 / E'$  versus  $K_{Ie}$  to study the 292 application of Irwin's equation in linking  $K_{I\Pi}^2 / E'$  and  $2\gamma$  at equilibrium. E' is the elastic constant and equal to  $E/(1-v^2)$ 293 for plane strain condition. Parameters used are summarized in Table 1,

294

289 290

#### **4.2.** Induced disjoining pressure $\Delta \Pi$

The magnitude of  $\Delta \Pi$  depends on both the surface separation w as well as the local concentration 295 296 of sorptive species C (Adler et al., 2001; Eskandari-Ghadi and Zhang, 2021). To construct a  $\Delta\Pi$  model, 297 let us focus the discussion on Wiederhorn (1967)'s glass-in-vapor system from hereon. Particularly, we 298 consider a single mode-I crack propagation in soda-lime glass in gaseous nitrogen at 25° with variable 299 relative humidity. The relevant chemistry along the crack can be straightforwardly characterized by the 300 partial pressure of the water vapor p. In this setting, the environment-induced disjoining pressure can be 301 expressed by  $\Delta \Pi(w, p)$ . Note that we did not consider  $\Delta \Pi$  as a function of time in this first treatment. In other words, we have assumed that the adsorption kinetics is much faster than the rate of subcritical 302 crack propagation ( $< 10^{-4}$  m/s), and the only rate-limiting step in generating the repulsive disjoining 303 pressure  $\Delta \Pi$  is the transport of water vapor along the crack, as described by the transport models 304 detailed in the next section. 305

Physically, many molecular-scale mechanisms can lead to the generation of  $\Delta\Pi$  at various vapor pressures (Clarke et al., 1986). It is believed that these mechanisms mainly have origins in Van der Waals, electric double layer, and hydration forces, with the hydration force being the dominant contribution (Dziadkowiec, 2019). The exhaustive discussion of each of them is beyond the scope of this work. A smooth 4-parameter model is instead adopted for the environment-induced disjoining pressure:

311 
$$\Delta \Pi(w, p) = f_{\Lambda \Pi}(w) g_{\Lambda \Pi}(p) \tag{10}$$

312 where

313 
$$f_{\Delta\Pi}(w) = \begin{cases} \Delta\Pi_{\max} \frac{w + \tilde{w}_0}{\tilde{w}_0} \exp\left(1 - \frac{w + \tilde{w}_0}{\tilde{w}_0}\right) & ; \ w \ge 0\\ f_{\Delta\Pi}(0) & ; \ w < 0 \end{cases}$$
(11)

314 captures the dependency of  $\Delta \Pi$  on w by taking similar form as Eq. (9); and

315 
$$g_{\Delta\Pi}(p) = \left(\frac{p}{p_{sat}}\right)^{\alpha} \exp\left(\beta\left(1 - \frac{p_{sat}}{p}\right)\right)$$
(12)

characterizes the dependency of  $\Delta \Pi$  on the vapor pressure p.  $\Delta \Pi_{\text{max}}$ ,  $\tilde{w}_0$ ,  $\alpha$ , and  $\beta$  are model 316 parameters;  $p_{sat}$  is the saturation vapor pressure xpfEhsion exhibits an overall exponential 317 decay, which is consistent with expressions adopted for hydration forces (Dziadkowiec et al., 2018; 318 319 Israelachvili, 2011). The expression of g is constructed to 1) monotonically increase with increasing 320 water vapor pressure p, 2) smoothly increase from zero, and 3) contain parameters that control the pressure and the steepness of the steepest increase. The value of  $f_{\Delta\Pi}$  in the range of w < 0 is assigned to 321 322 ensure the overall  $\Pi$  model is smooth at w=0. Eqns. (10), (11), (12) and the total  $\Pi$  from Eq. (8) are 323 visualized in Fig. 7 for different levels of  $p/p_{sat}$  and values of model parameters. Fig. 7a shows that 324 parameters  $\Delta \Pi_{max}$  and respectively and independently, control the spread of  $\Delta \Pi$  over separation

and its magnitude. Fig. 7b shows that the overall curvature of  $g_{\Delta\Pi}$  decreases with increasing  $\alpha$  and  $\beta$ .  $\alpha$  also controls the curvature at very small p values, while the effect of  $\beta$  is less concerned with small p. The resultant  $\Delta\Pi$  in Fig. 7c resembles the disjoining pressure sketched in Ash et al. (1973) and the one derived by Eskandari-Ghadi and Zhang (2021) based on the BDDT isotherm (Brunauer et al., 1940). The total  $\Pi$  in Fig. 7d can be compared with the conceptual sketch by Røyne et al. (2015) for calcite surfaces.



331 332 333 334 335 336 337 338 339

Fig. 7. Environment-induced disjoining pressure as a function of (a) separation and (b) relative partial pressure; (c) the  $\Delta\Pi$  surface in the w - p space with  $\Delta\Pi_{\text{max}} = 300$ MPa,  $\tilde{w}_0 = 1.8 \times 10^{-5}$  mm,  $\alpha = 0.2$ , and  $\beta = 0.3$ ; and (d) the total disjoining pressure Eq. (8) with  $\Pi_{0,\text{max}} = 1200$ MPa and  $w_0 = 3.18 \times 10^{-6}$  mm.

### 340 5. MASS TRANSPORT ALONG THE CRACK

The effect of sorption on crack growth is contingent upon the accessibility of sorptive species near 341 342 the crack tip. This is controlled by the species transport along the crack path and is expected to trail 343 behind as the crack propagates faster. The spatiotemporal evolution of species concentration in turn impacts the resultant disjoining pressure (Fig. 2) and further the  $K_{le}$  and the crack velocity, thus 344 forming a two-way coupling. Resolving species transport is therefore essential to the proposed SFFT 345 346 model for capturing the full SCG behavior. Below we develop a generic transport model to capture the 347 most basic features of gas flow in cracks. A one-dimensional treatment is adopted, assuming that gas 348 migrates towards the crack tip along the crack path (x-direction in Fig. 3a). The steady-state solution 349 with respect to the crack tip of the transport model will be sought, considering crack propagation 350 velocities in stages I and II are typically slow (of order 10<sup>-4</sup> m/s or less). In other words, we examine the steady-state crack propagation scenarios at each  $K_{la}$  value during an SCG test without considering the 351 short transient stages between  $K_{la}$  increments. 352

353 Consider a small segment  $\delta x$  along the crack with respect to the x - y coordinate fixed in space 354 (Fig. 8), the number of moles of sorptive species must be conserved:

$$\delta n(x,t)Lw(x)\delta x = J(x,t)Lw(x)\delta t - J(x+\delta x,t)Lw(x+\delta x)\delta t$$
(13)





opening treated here as a known. By Taylor expansion of  $J(x+\delta x)$  and  $w(x+\delta x)$ , and taking the limit of  $\delta x \rightarrow 0$  and  $\delta t \rightarrow 0$ , Eq. (13) can be rewritten as

363 
$$\frac{\partial n(x,t)}{\partial t} = -\frac{1}{w(x,t)}\frac{\partial}{\partial x}\left(w(x,t)J(x,t)\right)$$
(14)

For an ever-expanding domain (i.e., a moving crack tip), "steady state" only makes sense for an observer that moves together with the crack tip, i.e., the near-tip species concentration profile stops evolving for that observer. Let us therefore define a new coordinate system x' - y' that moves according to  $x' = x - v_c t$  and y' = y (Fig. 8). It follows  $n_t(x,t) = n_t(x',t) - v_c n_{x'}(x',t)$  and  $(w(x)J(x,t))_x = (w(x)J(x,t))_{x'}$ , where  $(\cdot)_t = \partial(\cdot)/\partial t$ ,  $(\cdot)_x = \partial(\cdot)/\partial x$ , and  $(\cdot)_{x'} = \partial(\cdot)/\partial x'$ . Thus the molar balance in Eq. (14) can be written in the x' - y' coordinate as

370 
$$\frac{\partial n(x',t)}{\partial t} = v_c \frac{\partial n(x',t)}{\partial x'} - \frac{1}{w(x',t)} \frac{\partial}{\partial x'} (w(x',t)J(x',t))$$
(15)

371 Examining the steady-state condition  $\partial(\cdot)/\partial t = 0$  of Eq. (15) gives the following ordinary differential 372 equation:

373 
$$v_c \frac{\partial n}{\partial x'} = \frac{1}{w} \frac{\partial}{\partial x'} (wJ)$$
(16)

374 It relates the steady-state profiles n(x'), w(x'), and J(x') at given boundary conditions. Without losing 375 generality, the species transport may be depicted by

376 
$$J(x) = -D\frac{\partial n}{\partial x}$$
(17)

377 where *D* is a transport coefficient. At chemical equilibrium, the partial pressure (*p*), concentration (*n*), 378 and chemical potential ( $\mu$ ) of the sorptive species are one-to-one related through the equation of state. Thus, *D* can be re-written in terms of permeability for viscous and Knudsen flow or diffusivity for molecular and surface diffusion. This preserves the generality of the model and enables more detailed examinations of the various transport mechanisms in future studies. Substitution of Eq. (17) in Eq. (16) gives

383 
$$v_c \frac{\partial p}{\partial x'} = -\frac{1}{w} \frac{\partial}{\partial x'} (wD \frac{\partial p}{\partial x'})$$
(18)

where the ideal gas law (p = nRT) and isothermal process (dT = 0) are assumed. The steady-state 384 pressure distribution p(x') can be obtained by numerically solving Eq. (18) together with the  $\Pi - w$ 385 system introduced in the previous section. This is however numerically taxing considering solving the 386 387  $\Pi - w$  system already requires an iterative scheme. Alternatively, an explicit analytical expression of p(x') can be obtained by assuming  $\partial w/\partial x' \approx 0$  and constant D in Eq. (18). This compromise 388 significantly eases the global solution scheme and permits quantitative studies of the first-order 389 390 behaviors of the proposed SFFT model. With these simplifying assumptions, Eq. (18) can be immediately integrated under the boundary conditions of  $p = p_0$  at the crack mouth (x' = -a) and 391  $J = v_c n$  at crack tip (x' = 0) to give: 392

393 
$$p(x') = p_0 \exp\left(-\frac{v_c}{D}(x'+a)\right)$$
(19)

394 where  $p_0$  is the partial gas pressure in the environment outside the crack.

In reality, it is expected that gas transport from crack mouth to tip involves multiple mechanisms each of which control the transport rate at different crack width (Fig. 9a). For example, bulk gas flow may be dominant at large crack openings (from M to A in Fig. 9b) because of the  $w^2$  – dependency of its diffusivity or permeability *D* (Wu et al., 2015). As the crack tip is approached and the crack opening

399 narrows, viscous flow rapidly diminishes, and a secondary transport mechanism gains dominance to 400 permit molecular diffusion in nanometer spaces (from B to T in Fig. 9b). This specific transport 401 mechanism near crack tip is currently not well understood. Lawn (1974) argues that molecular collisions 402 with the crack walls, known as Knudsen diffusion, is the mechanism of near-tip transport. Wiederhorn (1967) postulated that there exists a near-tip region, the so called "boundary layer", that is always 403 inaccessible to bulk gas flow. Water molecules can only migrate through the gas mixture in this region 404 405 via Fickian diffusion at the same rate that they are chemically consumed at the crack tip. Contrarily, 406 experimental data from glass SCG at different temperatures by Crichton et al. (1999) does not support 407 the dominance of Fickian diffusion nor Knudsen diffusion in the boundary layer by comparing the 408 anticipated versus the observed temperature-dependence of the stage-II crack velocity. They suggested 409 surface diffusion is a plausible mechanism to explain the temperature-dependence, but at the same time 410 also suggested that more complex mechanisms or their combinations could be responsible. An ongoing 411 research effort at the Lawrence Berkeley National Laboratory is aimed at providing better understanding 412 of these complex transport mechanisms in glass cracks at nano-scale.

Given the lack of a comprehensive transport model that captures the various mechanisms along the crack, we opt for a simplified, segmented, analytical steady-state pressure profile to account for the multi-regime transport of species in complementary to the single-regime model in Eq. (19). Particularly, species transport is assumed to take form of rapid bulk flow for crack width  $w > w_{th}$  (from M to C in Fig. 9c) where the pressure drop is negligible and slow diffusion for  $w_{th} < w$  (from C to T in Fig. 9c) where the pressure profile is similarly described by Eq. (19):

419  

$$\begin{cases}
p(x') = p_0 & ; \quad \forall x' | w(x') > w_{th} \\
p(x') = p_0 \exp\left(-\frac{v_c}{D}(x'+d)\right) & ; \quad \forall x' | w(x') \le w_{th}
\end{cases}$$
(20)

420 where  $w_{th}$  is a threshold crack width to be discussed later; d is defined such that  $w(x') > w_{th}$  for all 421 x' < -d. This dual-regime model (Fig. 9c) is conceptually similar to Wiederhorn's boundary layer 422 model (Wiederhorn, 1967). In essence, the only difference between the single-regime flow Eq. (19) and 423 the dual-regime flow Eq. (20) is that d is fixed at d = a in the former, but becomes a variable related to 424  $w_{th}$  in the latter.



Fig. 9. Schematic of the near tip crack profile (a), the realistic pressure profile in nano-spaces (b) inspired by Cihan et al. (2021), and the approximated dual-regime pressure profile (c).

428	The threshold crack width $w_{th}$ may be selected based on the Knudsen number, $\text{Kn} = \lambda / w$ , where $\lambda$
429	is the mean free path of the gas (Lawn, 1974). Kn 🛛 1 suggests transport dominated by viscous flow,
430	while Kn 1 suggests transport by slower diffusion mechanisms (Choi et al., 2001). Therefore, it is
431	reasonable to define $w_{th} = \lambda / Kn_0$ where $Kn_0$ marks the transition point taken to be 1/3 here, well
432	within the range of $0.01 < Kn_0 < 1$ recommended by Lawn (1974). The experiments of glass SCG by
433	Wiederhorn (1967) is performed in a nitrogen-water vapor environment at room temperature and
434	atmospheric pressure, based on which the mean free path can be approximated as $\lambda \approx 7 \times 10^{-8}$ m (Bird et
435	al., 2006; Chapman and Cowling, 1990; Jennings, 1988).

Fig. 10a and b plot the steady-state vapor pressure profiles at different crack velocity for crack length a = 36mm and ambient humidity RH<sub>0</sub> =  $p_0 / p_{sat} = 0.3$ . The dual-regime flow based on the current estimation of  $w_{th}$  predicts that the slow molecular diffusion only gets activated extremely close (tens of µm) to the crack tip. Both models predict exponential decay of crack-tip vapor concentration with increasing velocity (Fig. 10c).



Fig. 10. Steady-state vapor partial pressure profile based on the (a) single-regime model Eq. (19), and (b) dual-regime model
 Eq. (20) at different crack velocities. (c) Crack-tip pressure exponentially decreases as crack velocity increases. The model
 parameters used are those of Table 3.

#### 445 6. SOLUTION SCHEME

Our SFFT model specialized for glass SCG is now complete with fracture mechanics Eqns. (1) and (2), crack kinetics Eq. (7), surface force model Eqns. (9) and (10), and transport model Eq. (19) or (20). This section seeks the solution of the system subject to stress,  $\sigma(t)$ , over domain  $x \in [0, a]$ , and at each point in time  $t_k$ . Over each time increment  $(t_k, t_k + \Delta t_k)$ , the crack length is updated based on

450 
$$a(t_k + \Delta t_k) = a(t_k) + v_c(t_k)\Delta t_k$$
(21)

451 where  $\Delta t_k$  is initially set to 0.05 sec. It is reduced if the relative error at  $t_k$  is small and increased if the 452 relative error is large, within the range of 0.02 sec-1 sec.

For spatial discretization, nodes are distributed along the crack at the mid-plane, permitted by the one-dimensional nature of the problem. The crack is divided by *N* nodal points located at  $(x_1, x_2, ..., x_N)$  that are densely distributed (~ every 6 nm) near the crack tip and coarsely distributed (~ every 70 µm) close to the crack mouth. This is because of the nature of surface forces: they take large magnitudes and vary dramatically at short separations, then diminish quickly as the separation increases, making the system extremely non-linear near the crack tip. With this discretization, the unknown vectors **p**, **w**, and **II** are defined as  $p_i = p(x_i)$ ,  $w_i = w(x_i)$ , and  $\Pi_i = \Pi(x_i)$ , respectively.

For the modularity of the numerical scheme, the equations are grouped in three subsystems: **S1**) the fluid transport subsystem, Eq. (19) or (20); **S2**) the fracture mechanics  $\Pi - w$  subsystem, Eqns. (1), (2), (9), and (10); and **S3**) the crack propagation subsystem, Eq. (7). They are then iteratively coupled at the global level to search for a solution that simultaneously satisfies all the underlying equations.

464 For S1, the transport models in Eqns. (19) and (20) provide explicit solution at all points. In discrete
465 form, the nodal pressure described by these models are, respectively,

466 
$$p_i = p_0 \exp\left(-\frac{v_c}{D}x_i\right)$$
(22)

467 and

468

469 
$$p_{i} = \begin{cases} p_{0} & ; \quad \forall x_{i} \mid w(x_{i}) > w_{th} \\ p_{0} \exp\left(-\frac{v_{c}}{D}(x_{i} - a + d)\right) & ; \quad \forall x_{i} \mid w(x_{i}) \le w_{th} \end{cases}$$
(23)

470  $x_i$  is the coordinate of node *i* in the coordinate system set at the crack mouth, consistent with that used 471 for the fracture mechanics solution (Fig. 3). For S2, an iterative method is required to solve the  $\Pi - w$ 472 equations at fixed *p* profiles. Consistent  $\Pi$  and *w* profiles together with the resultant  $K_{le}$  are obtained 473 by following the solution method of Planas and Elices (1991), while taking advantage of the analytical 474 solutions made available by Tada et al. (2000) (Appendix A). For S3, Eq. (7) simply inputs an effective 475 stress intensity and outputs the resultant crack velocity.

476 The global coupling of S1, S2, and S3 is achieved through a partitioned Newton-Raphson method. 477 This method follows the multi-level Newton method of Kim et al. (2003). At local level, S1, S2, and S3 478 are solved separately and independently. It is important to distinguish the global unknowns as the 479 guessed inputs and the response of each subsystem to its pertaining guessed inputs as the local outputs. 480 The global unknowns are defined here as the partial pressure of the adsorptive fluid, **p**; the crack opening, w; the effective stress intensity factor at the crack tip,  $K_{le}$ ; and the crack velocity,  $v_c$ . We 481 482 shall then denote the solution to the fluid profile in response to guess values of crack opening and crack 483 velocity by  $\hat{\mathbf{p}}(\mathbf{w}, v_c)$ , solved by S1; the crack opening and effective stress intensity in response to guess values of fluid state profile by  $\hat{\mathbf{w}}(\mathbf{p})$  and  $\hat{K}_{le}(\mathbf{p})$ , solved by S2; and the crack propagation velocity in 484

485 response to a guess value of stress intensity factor by  $\hat{v}_c(K_{Ie})$ , solved by S3. The coupling is achieved 486 by directing the global residual to zero in an iterative manner.

487 The vector of global unknowns is constructed as

488 
$$\mathbf{u} = \begin{bmatrix} \mathbf{p} \\ \mathbf{w} \\ K_{le} \\ v_c \end{bmatrix}$$
(24)

489 The global residual is defined as

490 
$$\mathbf{R} = \begin{bmatrix} \mathbf{p} - \hat{\mathbf{p}}(\mathbf{w}, v_c) \\ \mathbf{w} - \hat{\mathbf{w}}(\mathbf{p}) \\ K_{le} - \hat{K}_{le}(\mathbf{p}) \\ v_c - \hat{v}_c(K_{le}) \end{bmatrix}$$
(25)

491 where  $\mathbf{p}$ ,  $\mathbf{w}$ ,  $K_{le}$ , and  $v_c$  are the values of the unknowns at the current iteration. The global solution 492 that simultaneously satisfies S1, S2, and S3 can be found by iteratively marching towards  $\mathbf{R} = \mathbf{0}$  via the 493 Newton-Raphson method. In each iteration, the correction to  $\mathbf{u}$ ,  $\delta \mathbf{u}$ , can be obtained from 494  $\delta \mathbf{u} = -\mathbf{J}^{-1}(\mathbf{u}) \cdot \mathbf{R}(\mathbf{u})$  where  $\mathbf{J}$  is the global Jacobian matrix defined as:

495 
$$\mathbf{J} = \frac{\partial \mathbf{R}}{\partial \mathbf{u}} = \begin{bmatrix} \frac{\partial (\mathbf{p} - \hat{\mathbf{p}}(\mathbf{w}, v_c))}{\partial \mathbf{p}} & \frac{\partial (\mathbf{p} - \hat{\mathbf{p}}(\mathbf{w}, v_c))}{\partial \mathbf{w}} & \frac{\partial (\mathbf{p} - \hat{\mathbf{p}}(\mathbf{w}, v_c))}{\partial K_{le}} & \frac{\partial (\mathbf{p} - \hat{\mathbf{p}}(\mathbf{w}, v_c))}{\partial v_c} \\ \frac{\partial (\mathbf{w} - \hat{\mathbf{w}}(\mathbf{p}))}{\partial \mathbf{p}} & \frac{\partial (\mathbf{w} - \hat{\mathbf{w}}(\mathbf{p}))}{\partial \mathbf{w}} & \frac{\partial (\mathbf{w} - \hat{\mathbf{w}}(\mathbf{p}))}{\partial K_{le}} & \frac{\partial (\mathbf{w} - \hat{\mathbf{w}}(\mathbf{p}))}{\partial v_c} \\ \frac{\partial (K_{le} - \hat{K}_{le}(\mathbf{p}))}{\partial \mathbf{p}} & \frac{\partial (K_{le} - \hat{K}_{le}(\mathbf{p}))}{\partial \mathbf{w}} & \frac{\partial (K_{le} - \hat{K}_{le}(\mathbf{p}))}{\partial K_{le}} & \frac{\partial (K_{le} - \hat{K}_{le}(\mathbf{p}))}{\partial v_c} \\ \frac{\partial (v_c - \hat{v}_c(K_{le}))}{\partial \mathbf{p}} & \frac{\partial (v_c - \hat{v}_c(K_{le}))}{\partial \mathbf{w}} & \frac{\partial (v_c - \hat{v}_c(K_{le}))}{\partial K_{le}} & \frac{\partial (v_c - \hat{v}_c(K_{le}))}{\partial v_c} \end{bmatrix} \end{bmatrix}$$
(26)

496 Upon simplification, one has



498 A schematic of this partitioned Newton-Raphson scheme is presented in Fig. 11 for clarity.

The derivatives in Eq. (27) can be computed from Eq. (7), Eqns. (A.6) and (A.7) of Appendix A, and
Eq. (22) or (23) depending on the choice of the transport model:

501 
$$\frac{\partial \hat{v}_c}{\partial K_{le}} = \frac{v_0 \kappa_0}{RT} \exp\left(-\frac{Q_0}{RT}\right) \cosh\left(\frac{\kappa_0 K_{le}}{RT}\right)$$
(28)

502 
$$\frac{\partial \hat{w}_i}{\partial p_j} = c_{\Pi,ik} \frac{\partial \hat{\Pi}_k}{\partial p_j} \approx c_{\Pi,ik} \frac{\partial \Pi_k}{\partial p_j}$$
(29)

503 
$$\frac{\partial \hat{K}_{le}}{\partial p_{j}} = k_{I\Pi,k} \frac{\partial \hat{\Pi}_{k}}{\partial p_{j}} \approx k_{I\Pi,k} \frac{\partial \Pi_{k}}{\partial p_{j}}$$
(30)

504 
$$\frac{\partial \hat{p}_i}{\partial w_j} = \frac{\partial \hat{p}_i}{\partial d} \frac{\partial d}{\partial w_j} \approx 0$$
(31)

505 
$$\frac{\partial \hat{p}_i}{\partial v_c} = -\frac{x_i}{D} p_0 \exp\left(-\frac{v_c}{D} x_i\right)$$
(32)

506 for single-regime flow, and

507 
$$\frac{\partial \hat{p}_i}{\partial v_c} = \begin{cases} 0 & ; \quad \forall x \mid w(x) > w_{th} \\ -\frac{x_i - a + d}{D} p_0 \exp\left(-\frac{v_c}{D}(x_i - a + d)\right) & ; \quad \forall x \mid w(x) \le w_{th} \end{cases}$$
(33)

508 for dual-regime flow.

509 The errors are defined as

510 
$$\varepsilon_{p} = \left| \frac{\delta p_{i}}{p_{i}} \right|, \ \varepsilon_{w} = \left| \frac{\delta w_{i}}{w_{i}} \right|, \ \varepsilon_{K_{Ie}} = \left| \frac{\delta K_{Ie}}{K_{Ie}} \right|, \text{ and } \varepsilon_{v_{c}} = \left| \frac{\delta v_{c}}{v_{c}} \right|$$
(34)

511 The iterations continue until the maximum of the four errors falls within a tolerance of  $10^{-5}$ .





Fig. 11. A schematic of (a) the inputs and outputs of the S1, S2, and S3 solvers; and (b) the partitioned Newton-Raphson scheme to seek solutions satisfying all subsystems.

Note that the length of the slow transport regime d varies continuously as the crack opens or closes, while the numerical representation of the crack opening **w** is discrete and does not allow smooth variation of d. Therefore, a linear interpolation for the value of w(x) between two nodes is adopted to allow smooth variation of d. Consequently, the derivative  $\partial d/\partial \mathbf{w}$  cannot be effectively computed without explicit expression of d. This term is expected to be small anyway given the initial guess for **w**  is close to the solution. The derivative in Eq. (31) is thus approximated by zero. In addition, the  $\partial \hat{\mathbf{\Pi}} / \partial \mathbf{p}$ term in Eq. (29) must be approximated since the  $\Pi - w$  system is implicitly solved without being written in terms of  $\mathbf{p}$ . Two versions of the implementation were tested, one with  $\partial \hat{\mathbf{\Pi}} / \partial \mathbf{p} \approx (\partial \hat{\mathbf{\Pi}} / \partial \mathbf{p})_{\hat{\mathbf{w}}}$ and  $\partial \hat{\mathbf{w}} / \partial \mathbf{p}$  given by Eq. (29), and the other with  $\partial \hat{\mathbf{w}} / \partial \mathbf{p} \approx \mathbf{0}$ . The second approximation yielded better convergence and was adopted in the final implementation. Finally, a proper initial guess is required to start the search for the solution at each time  $t_{k+1}$  due to the highly nonlinear nature of the problem. A robust initial guess algorithm is described in Appendix B.

## 527 **7. PERFORMANCE OF THE FRAMEWORK**

528 This section compares the SFFT prediction against Wiederhorn's SCG data (Fig. 12) to validate the 529 underlying hypotheses depicted in Fig. 1.

## 530 **7.1. Model calibration**

We shall calibrate the least coupled parts of the model first. The intrinsic material properties including the elastic parameters, in-vacuum surface force, and the reaction kinetic parameters are therefore calibrated first here. These parameters are expected to control the in-vacuum stage-III SCG response.

The standard Young's modulus of soda-lime glass is 70 GPa (572-1, 2012) while the reported ones in literatures vary between 50 GPa and 85 GPa (Meyland et al., 2021). A Poisson ratio in the range of 0.2 ~ 0.3 has been reported in some studies (Li and Wei, 2020; Meyland et al., 2021; Xu et al., 2017). Within this ballpark, the selected elastic parameters are shown in Table 1.



 $\begin{array}{c} 539 \\ 540 \\ 541 \\ 542 \end{array}$ Fig. 12. Data for crack propagation in soda-lime glass at 25°C and different relative humidity (Wiederhorn, 1967; Wiederhorn et al., 1980), open circles; at different temperatures at 24°C, 214°C, and 400°C in vacuum (Wiederhorn, 1974), squares; and in liquid water (Wiederhorn and Bolz, 1970), filled circles.

543 Two types of data are adopted to calibrate the  $\Pi_0$  parameters. The first data is surface energy for 544 soda-lime glass measured by Wiederhorn (1969) at near room temperature in an environment with background nitrogen gas (inert) as  $\gamma = 3.82 \text{ J/m}^2$ . This measurement in inert gas can estimate the area 545 underneath the  $\Pi_0 - w$  curve, in absence of sorptive species (Gdoutos, 2020). For the selected  $\Pi_0$ 546 547 expression Eq. (9), this area is  $\gamma = \prod_{0, max} w_0$ . The second data is the SFA measurements between a glass sphere in air (Yaminsky and Stewart, 2003), scaled by a factor of 912.6 such that the area underneath 548 this data matches the surface energy measurement. Parameters  $\Pi_{0,max}$  and  $w_0$  are then selected (Table 1) 549 550 to match this data (Fig. 13).



Fig. 13. Calibrated in-vacuum surface force ( $\Pi_{0,max} = 1200$ MPa and  $w_0 = 3.1833 \times 10^{-6}$  mm) vs. SFA measurements. A constant multiplying factor is applied to the SFA data to roughly match the area under this curve with the surface energy measurement of  $\gamma = 3.82$ J/m<sup>2</sup>. The  $\Pi_0$  data is converted from *F*/*R* data multiplied by 912.6.

The reaction kinetic parameters are calibrated against the in-vacuum  $v_c - K_{Ia}$  data. At relatively large 555  $K_{la}$  values, the  $K_{le} - K_{la}$  relation becomes linear (see Fig. 6) and so does the  $\log(v_c) - K_{le}$  relation (Eq. 556 (7)). Combining the two gives a linear one-to-one correspondence between  $K_{la}$  and  $v_c$  at high  $K_{la}$ 557 558 levels. Note that there are infinite combinations of  $v_0$ ,  $Q_0$ , and  $\kappa_0$  to match a linear set of  $\log(v_c) - K_{la}$ 559 data. Such arbitrariness can be removed by considering the temperature dependence of the crack kinetics. Specifically, the same set of  $v_0$ ,  $Q_0$ , and  $\kappa_0$  should capture the in-vacuum  $v_c - K_{la}$  curves obtained at 560 561 different temperatures, assuming the Arrhenius term in Eq. (7) outweighs the temperature dependence of the elastic parameters and in-vacuum surface forces. Fig. 14 plots soda-lime glass  $v_c - K_{la}$  data obtained 562 at 24°C, 214°C, and 400°C (Wiederhorn, 1974) based on which the kinetic parameters  $v_0$ ,  $Q_0$ , and  $\kappa_0$ 563 564 are calibrated (Table 1). Their values are the same orders of magnitude as those presented in Wiederhorn 565 (1974).

566

551

Parameter	Unit	Values				
Elasticity						
E	[MPa]	50000				
V	[-]	0.2				
In-vacuum surface force						
$W_0$	[mm]	3.1833×10 <sup>-6</sup>				
$\Pi_{0,\max}$	[MPa]	1200				
Crack propagation reaction						
$v_0$	[mm/s]	3.303×10 <sup>5</sup>				
$Q_0$	[J/mol]	1.31×10 <sup>5</sup>				
$\kappa_0$	[m <sup>2.5</sup> /mol]	0.24				

Table 1. Model parameters that control stage III SCG behavior



Fig. 14. Calibration of reaction rate parameters  $v_0$ ,  $Q_0$ , and  $\kappa_0$  against SCG data at three temperatures, 25°C, 214°C, and 400°C in vacuum. The irrelevant data are faded for clarity.

575 The model is now fully calibrated for the in-vacuum condition. We now move on to determine 576 parameters that control the environmental dependency, specifically the environment-induced disjoining 577 pressure  $\Delta\Pi$  and the pressure profile p(x). The former is expected to affect the strength reduction in 578 stage I, and the latter limits the rate of crack propagation in stage-II SCG.

In Eqns. (10), (11), and (12), parameters  $\Delta \Pi_{\text{max}}$  and  $\tilde{w}_0$  control the dependence of  $\Delta \Pi$  on the crack 579 opening and parameters  $\alpha$  and  $\beta$  independently control the dependence of  $\Delta \Pi$  on the partial pressure 580 581 of the adsorbate. AFM measurements in glass-liquid water system (Acuña and Toledo, 2008) provides a benchmark for the shape of the  $\Delta \Pi - w$  relation at  $p = p_{sat}$  which allows the determination of  $\tilde{w}_0$  (Fig. 582 15). The area underneath  $\Delta\Pi$  reflects the change of surface energy in presence of water vapor, meaning 583 that  $\Delta \Pi_{max}$  can be calibrated based on the reduction of apparent fracture toughness from RH<sub>0</sub> = 0% to 584 100% (Fig. 16). Finally, parameters  $\alpha$  and  $\beta$  in Eq. (12) are tuned to capture the strength reductions at 585  $RH_0 = 0.017\%$ , 0.2%, 1%, 10%, and 30% (Fig. 16). These parameters are summarized in Table 2. 586

587

588

Table 2. Parameters	that control stag	ge I SCG in cor	nplement to	Table 1.

Parameter	Unit	Values			
Environment-induced disjoining pressure					
$ ilde{w}_0$	[mm]	1.8×10 <sup>-5</sup>			
$\Delta \Pi_{0,\max}$	[MPa]	167			
α	[-]	0.067			
$\beta$	[-]	10-5			





Fig. 15. Calibrated  $\Delta \Pi$  with parameters  $\tilde{w}_0 = 1.8 \times 10^{-5}$  mm and  $\Delta \Pi_{max} = 167$  MPa at  $p = p_{sat}$  based on the *F/R* 592 measurements from AFM (Acuña and Toledo, 2008) scaled by 120.5. The main purpose here is to match the shape via 593 adjusting  $\tilde{w}_0$ .  $\Delta \Pi_{\text{max}}$  is determined in Fig. 16.



Fig. 16. Measured vs. simulated strength reduction at  $RH_0 = 100\%$  with parameters  $\Delta \Pi_{max} = 167 MPa$  and  $\tilde{w}_0 = 1.8 \times 10^{-5}$  mm (light blue coarse-dashed line). The strength reductions for intermediate RH values are captured by 596 selecting  $\alpha = 0.067$  and  $\beta = 10^{-5}$  (green dashed lines). The cross-hatching is meant as a visual guide for the extension of the 597

598  $RH_0 = 100\%$  data based on the data in liquid water. The reason for this extension is to obtain a visual of the  $RH_0 = 100\%$ 599 curve at lower velocities for the purpose of calibration.

On the transport side, for both p(x) profiles depicted by Eq. (19) and Eq. (20), D is selected such that the adsorbate relative pressure at the crack tip drops to an arbitrary small value (here taken as  $2 \times 10^{-6}$ ) at the stage-II plateau velocity for the RH<sub>0</sub> = 100% case ( $v_c = 2.5 \times 10^{-4}$  m/s at  $K_{1a} = 7.44 \times 10^5$  N/m<sup>1.5</sup> from Fig. 16). The near-tip region length d for the dual-regime flow model is estimated from the in-vacuum crack opening profile at  $K_{1a} = 7.44 \times 10^5$  N/m<sup>1.5</sup>. This gives  $d \approx 2.13 \times 10^{-5}$  m; i.e.,  $w < w_{th} = \lambda / \text{Kn}_0$  for  $a - 2.13 \times 10^{-5} < x$ . The values of D for the two cases are given in Table 3.

607

Table 3. Parameters that control stage II SCG in complement to Table 1 and Table 2.

Parameter	Unit	Values			
Single-regime flow					
D	$[m^2/s]$	6.94×10 <sup>-7</sup>			
Dual-regime flow					
D	$[m^2/s]$	3.96×10 <sup>-10</sup>			

608 These calibrated D values can be quantitatively compared with some theoretical estimates. In a study of gas transport in shale nanopores, Wu et al. (2015) predicted an equivalent Darcian permeability k (as 609 in  $v_{fluid} = -(k/\mu)\partial p/\partial x$ ) of order  $10^{-17}$  m<sup>2</sup> for the case of surface diffusion. Conversion of k to D can 610 be made by rewriting  $J(x) = -D \partial n / \partial x$  as  $v_{\text{fluid}} = J / n = -(D / p) \partial p / \partial x$ , giving  $D = kp / \mu$ . A 611 viscosity of order 10<sup>-5</sup>Pa · s is assumed (Tsilingiris, 2008; Wu et al., 2015), and a pressure of order 612  $<10^{3}$  Pa can be assumed considering  $O(p_{sat}) = 10^{3}$  Pa for water vapor. This gives 613  $O(D_{\text{surf. diff.}}) < 10^{-9} \text{ m}^2/\text{s}$ . for surface diffusion, comparable with the calibrated value of D for dual-614 615 regime flow. Similar comparison can be made in the case of Knudsen diffusion (Wu et al., 2015). On the other hand, considering viscous flow in a slit with aperture *w*, one can derive  $D_{\text{bulk flow}} = RTw^2/12\mu V_m$ for ideal gas. At room temperature, the molar volume of ideal gas is  $V_m = 2.45 \times 10^{-2} \text{ m}^3/\text{mol}$ . Fig. 5 suggests that the crack opening varies from order  $10^{-5} \text{ m}$  to  $10^{-8} \text{ m}$  from the crack mouth to where surface forces are observed. Combining the above gives  $O(D_{\text{bulk flow}}) = 10^{-2} \sim 10^{-8} \text{ m}^2/\text{s}$ . Note that the estimated  $D_{\text{bulk flow}}$  is several orders larger than our calibrated D values (Table 3) and those estimated based on Wu et al. (2015). This justifies our assumption that the pressure drop is negligible in the bulk flow regime up to the transition point (Fig. 9c).

## 623 7.2. Model prediction

624 Using the calibrated parameters in Tables 1, 2, and 3, the SCG curves predicted by the SFFT model 625 for single-regime and dual-regime flow at  $RH_0 = 100\%$  are presented together with experimental data in Fig. 17a. It is exciting to see that, for the first time, the key features of all three stages of SCG and the 626 627 smooth transition between them are directly predicted by a single mechanistic model. Comparing the 628 two computed SCG curves suggests that the choice of transport model directly impacts the stages I and 629 II SCG behavior. This cannot be observed in existing phenomenological SCG models, since the shape of 630 the whole SCG curve is imposed rather than predicted in these approaches (Charles, 1962; Lawn, 1975). 631 Fig. 17b presents the predicted vapor pressure profiles at different stages of SCG for the dual-regime model. As  $K_{la}$  increases, the crack opening widens, and the inviscid regime moves closer to the crack 632 633 tip. However, the simultaneous increase in crack velocity outpaces the crack widening, which results in 634 an overall decrease of fluid concentration at the crack tip. Fig. 17c shows that the disjoining pressure is 635 extremely concentrated near the crack tip and is highly sensitive to the local vapor pressure and thus the 636 crack velocity. This suggests that the introduction of slight repulsive disjoining pressure in the near-tip 637 region can cause significant reduction of the apparent fracture toughness for brittle solids. The profiles

#### 638 plotted in Fig. 17b and c confirm the common speculation that fluid transport is indeed the rate-limiting





 $\begin{array}{c} 640 \\ 641 \\ 642 \\ 643 \end{array} \begin{array}{c} K_{Ia} (N/m^{1.3}) \\ Fig. 17. (a) Measured vs. predicted SCG curves at RH_0 =100\% based on single-regime flow and dual-regime flow. (b) \\ Pressure profiles at different stages along the SCG curve depicted by the dual-regime flow model. (c) Disjoining pressure \\ distribution at different stages along the SCG curve. \end{array}$ 

644 Recalling that the SFFT model was calibrated only for the apparent toughness (where the SCG curve intersects with the x - axis) and the plateau velocity (where stages II and III intersects) at RH<sub>0</sub> = 100%, 645 646 everything in between should be regarded as true predictions. It is observed that the current model 647 generally overestimates the crack velocity in stages I and II, with the dual regime flow model making slightly better prediction. This is because the experimental data suggests the stage-I slope is lower 648 649 compared to the in-vacuum portion, while the current model predicts the same slopes for stages I and III. 650 We may remediate this through the reaction rate parameters or the fluid transport parameters. Fig. 18a demonstrates that the decrease of  $\kappa_0$  or  $Q_0$  can reduce the slope of SCG curve in the  $\log(v_c) - K_{la}$  plane. 651 652 This implies that the lowered stage-I slope can be accounted for by introducing some environmental

dependency on the kinetic parameters. A rigorous proposal of  $\kappa_0(p)$  or  $Q_0(p)$  requires detailed study of how the energy barrier for bond rupture at the very crack tip lowers under environmental attack, which is beyond the scope of this paper. Alternatively, an apparent slope reduction can be readily observed by cropping out the very low velocity range of the SCG plots (Fig. 18b).



Fig. 18. (a) Variation of SCG slope about a fixed point ( $K_{la} = 7.18 \times 10^5 \text{ N/m}^{1.5}$  and  $v_c = 1.12 \times 10^{-6} \text{ m/s}$ ) controlled by parameters  $\kappa_0$  and  $Q_0$ :  $\kappa_0 = 0.60 \text{ m}^{2.5}$ /mol and  $Q_0 = 257720 \text{ J/mol}$  for Simulation A;  $\kappa_0 = 0.24 \text{ m}^{2.5}$ /mol and  $Q_0 = 131000 \text{ J/mol}$  for Simulation B; and  $\kappa_0 = 0.10 \text{ m}^{2.5}$ /mol and  $Q_0 = 88760 \text{ J/mol}$  for Simulation C. (b) Zoomed-in presentation of Fig. 17 giving an impression of reduced stage-I slope.

657

This implies that a rate-limiting mechanism that is activated over a wide range of crack velocities can cause the reduced slope in stage I. Perhaps a mechanism with weaker dependence on  $v_c$  compared to the current transport models (e.g., a different near-tip diffusion mechanism, or sorption kinetics) can potentially close this gap. In partial support of this hypothesis, the experimental SCG data for crack propagation in liquid water in Fig. 16 which extends to much lower crack propagation velocities shows a slope close to the in-vacuum one. This speculation challenges the commonly accepted picture that SCG in stage I is free from any rate-limiting influences. More investigations are required in this regard toclarify the origin of environment-dependent slope in stage I SCG.

670 Fig. 19 presents the predicted SCG curves over a range of relative humidity from 0% to 100%.



It is observed that both flow models significantly underestimate the variation of plateau velocities at different humidity levels. Based on our previous observation that the shape of SCG critically depends on the transport model, we suspect that this underprediction is related to the insensitivity of the vapor pressure profile p(x) to the RH<sub>0</sub>. Physically, the threshold crack width  $w_{th}$  characterizing the point of transition from bulk flow to molecular diffusion (point C in Fig. 9c) should not be a constant but rather vary with respect to the boundary conditions (i.e., the ambient humidity). This possibility is explored in the next subsection.

### 682 **7.3. Effect of variable** $W_{th}$

Intuitively, lower RH means less water molecules in the environment, and they can be more easily captured by the crack walls to form the first few adsorbed layers. This promotes the molecular diffusion of water in adsorbed states rather than free bulk gas flow, thus extending the length of the "boundary layer" d in Fig. 9a.  $w_{th}$  is therefore proposed to be inverse proportional to RH<sub>0</sub>. A simple proposition to account for this effect is:

688 
$$w_{th} = \frac{\lambda / \text{Kn}_0}{1 + h_0 (\text{RH}_0 - 1)}$$
(35)

where  $h_0 < 1$  is a parameter to control how rapid  $w_{th}$  varies with  $RH_0$  such that  $w_{th}(RH_0 = 1) = \lambda / Kn_0$ . Here, a value of  $h_0 = 0.919$  is found to match the plateau velocity for case of  $RH_0 = 0.017\%$ . The predicted SCG curves using the updated  $w_{th}$ , Eq. (35), for all other  $RH_0$  values are presented in Fig. 20. We can see that the SFFT model is now capable of quantitatively capturing the shifting of the plateau velocity and the apparent fracture toughness under humidity changes. Again, this observation highlights the importance of detailing the transport of sorptive species in the near-tip regime in order to correctly predict the SCG behavior of brittle solids under various ambient conditions.

696 A subcritical cracking "stage diagram" can be produced (Fig. 21) in the normalized 697  $\log(v_c / v_0) - (K_{Ia} / \sqrt{2E'\gamma})$  space based on the slope of the predicted SCG curves in Fig. 20. The slope in 698 the normalized space is defined as  $I_c = d \log(v_c / v_0) / d(K_{Ia} / \sqrt{2E'\gamma})$ , which is analogous to the 699 *corrosion index* in the Charles' law (Atkinson, 1982).





It is apparent in Fig. 21 that stages I and III are represented by regions with larger  $I_c$ , while a drop of  $I_c$ 703 704 to lower values indicates stage II. Transition from stage II to III occurs abruptly (cyan dash-dot line), while the transition from stage I to II happens gradually with a much smoother and wider variation of  $I_c$ 705 values in between. A value of  $I_c = 15$  is selected to roughly represent this transition (pink dashed line). 706 707 Fig. 21 predicts that increased ambient relative humidity expands the stage-II zone, indicating that 708 species transport becomes more dominating in shaping the SCG curve over a wider  $K_{la}$  range at high RH levels. If an SCG curve is plotted in a limited range of  $K_{la}$  and thus  $v_c$ , one would have concluded 709 710 that the apparent corrosion index (i.e., the slope) is reduced as a function of the ambient humidity. This 711 reiterates one of our earlier speculations that the reduced slopes in stage I SCG could be just a visual 712 impression of an extended stage II zone due to the slow transport of environmental agents.



Fig. 21. A dimensionless stage diagram for SCG. Color indicates the slopes of normalized  $v_c / v_0$  versus  $K_{Ia} / \sqrt{2E'\gamma}$  curves from Fig. 20, based on which the three SCG stages can be clearly identified.

## 716 8. CONCLUSION

713

This study develops a surface-force based fracture theory (SFFT) to rationalize and predict the 717 718 subcritical cracking of brittle solids in sorptive environments. The theory highlights the fundamental role 719 of surface forces which present at a few tens of nanometer surface-surface separations on the apparent 720 fracture toughness, its environmental dependency, and the crack growth kinetics of SCG. Specifically, analytical LEFM solutions are adopted to consider the repulsive disjoining force induced by sorption 721 722 behind the crack tip. The competition between the material's intrinsic cohesive forces and environment-723 induced disjoining forces is manifested as the effective stress intensity factor operating at the crack tip 724 and serves as the driving force for crack propagation. This quantitatively explains the environmentally 725 induced strength reduction of brittle solids. The characteristic shape of the SCG curve is further 726 impacted by the rate-limiting effect of species transport in the crack. Simplified analytical transport 727 models are proposed to qualitatively represent the single-regime and dual-regime flows. Finally, the model is validated against the experimental data on soda-lime glass in water vapor studied by 728 729 Wiederhorn (1967). The main findings of this study are highlighted below:

- 730 1. The development of repulsive disjoining pressure behind the crack tip in sorptive environment731 can cause the reduction of apparent fracture toughness.
- 732 2. Species transport along the crack is the rate-controlling mechanism governing the plateau
  733 velocity and the stage-II behavior of SCG.
- 734 3. The effect of transport can go beyond stage II and shape the characteristics of stage-I SCG curve,
  735 for example, by reducing the apparent slope in stage I.
- 4. The SCG stage diagram generated by SFFT simulations allows one to visualize the conditions
  for each stage of SCG and the characteristics of transition between stages.

738 The theory and its numerical solution scheme are set up in such a way that more complex and 739 advanced model components can be readily incorporated for extensions to other brittle materials. Further 740 improvements can be contemplated by (1) incorporating more realistic fluid transport models that 741 automatically capture the smooth transition between different transport mechanisms as the crack 742 aperture varies; (2) considering the underlying sources of surface forces to motivate a physics-based 743 disjoining pressure model; and (3) extending to different systems such as the SCG of quartz or calcite in 744 aqueous solutions for geoscience and geo-engineering applications. Some fundamental aspects require 745 further investigations to enrich the calibration and validation of SFFT. For example, experimental 746 studies and molecular dynamic simulations are needed to quantify disjoining pressure between freshly 747 exposed solid surfaces in sorptive environment and to better understand species transport in nano-748 confined spaces.

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## 754 **REFERENCES**

- 572-1, D.E., 2012. Glass in Building–Basic Soda Lime Silicate Glass Products–Part 1: Definitions and
   General Physical and Mechanical Properties. Beuth Verlag Berlin.
- Acuña, S.M., Toledo, P.G., 2008. Short-range forces between glass surfaces in aqueous solutions.
   Langmuir 24, 4881-4887.
- Adapa, S., Malani, A., 2021. Role of cation size on swelling pressure and free energy of mica pores.
   Journal of Colloid and Interface Science 599, 694-705.
- Adler, J.J., Rabinovich, Y.I., Moudgil, B.M., 2001. Origins of the non-DLVO force between glass
   surfaces in aqueous solution. Journal of colloid and interface science 237, 249-258.
- Allegri, G., 2020. A unified formulation for fatigue crack onset and growth via cohesive zone modelling.
   Journal of the Mechanics and Physics of Solids 138, 103900.
- Ash, S.G., Everett, D.H., Radke, C., 1973. Thermodynamics of the effects of adsorption on interparticle
   forces. Journal of the Chemical Society, Faraday Transactions 2 69, 1256.
- Atkinson, B.K., 1982. Subcritical crack propagation in rocks: theory, experimental results and
   applications. Journal of Structural Geology 4, 41-56.
- Atkinson, B.K., 1984. Subcritical crack growth in geological materials. Journal of Geophysical Research:
   Solid Earth 89, 4077-4114.
- Barenblatt, G.I., 1962. The mathematical theory of equilibrium cracks in brittle fracture. Advances in
   applied mechanics 7, 55-129.
- Bažant, Z.P., Oh, B.H., 1983. Crack band theory for fracture of concrete. Matériaux et construction 16,
   155-177.
- Bazant, Z.P., Planas, J., 1997. Fracture and size effect in concrete and other quasibrittle materials. CRC
   press.
- Bird, R.B., Stewart, W.E., Lightfoot, E.N., 2006. Transport phenomena. John Wiley & Sons.
- Brantut, N., Heap, M., Meredith, P., Baud, P., 2013. Time-dependent cracking and brittle creep in crustal rocks: A review. Journal of Structural Geology 52, 17-43.
- Brunauer, S., Deming, L.S., Deming, W.E., Teller, E., 1940. On a Theory of the van der Waals
  Adsorption of Gases. Journal of the American Chemical Society 62, 1723–1732.
- Chapman, S., Cowling, T.G., 1990. The mathematical theory of non-uniform gases: an account of the kinetic theory of viscosity, thermal conduction and diffusion in gases. Cambridge university press.
- 785 Charles, R., 1958. Static fatigue of glass. I. Journal of Applied Physics 29, 1549-1553.
- Charles, R., 1962. The kinetics of glass failure by stress corrosion. Simposium sur la Resistance
   Mechanique du Verre et les Moyens de l'Ameliorer 511.
- Chester, F., Chester, J., Kronenberg, A., Hajash, A., 2007. Subcritical creep compaction of quartz sand
  at diagenetic conditions: Effects of water and grain size. Journal of Geophysical Research: Solid
  Earth 112.
- Choi, J.-G., Do, D., Do, H., 2001. Surface diffusion of adsorbed molecules in porous media: Monolayer,
   multilayer, and capillary condensation regimes. Industrial & engineering chemistry research 40,
   4005-4031.

- Cihan, A., Tokunaga, T.K., Birkholzer, J.T., 2019. Adsorption and capillary condensation-induced
   imbibition in nanoporous media. Langmuir 35, 9611-9621.
- Cihan, A., Tokunaga, T.K., Birkholzer, J.T., 2021. Diffusion to Imbibition Transition in Water
   Sorption in Nanoporous Media: Theoretical studies. Water Resources Research,
   e2021WR029720.
- Clarke, D., Lawn, B., Roach, D., 1986. The role of surface forces in fracture, Fracture Mechanics of
   Ceramics. Springer, pp. 341-350.
- Crichton, S.N., Tomozawa, M., Hayden, J.S., Suratwala, T.I., Campbell, J.H., 1999. Subcritical crack
   growth in a phosphate laser glass. Journal of the American Ceramic Society 82, 3097-3104.
- Biao, Y., Espinosa-Marzal, R.M., 2016. Molecular insight into the nanoconfined calcite–solution
   interface. Proceedings of the National Academy of Sciences 113, 12047-12052.
- Dziadkowiec, J., 2019. Interactions between Confined Calcite Surfaces in Aqueous Solutions: A surface
   forces apparatus study.
- Dziadkowiec, J., Javadi, S., Bratvold, J.E., Nilsen, O., Røyne, A., 2018. Surface Forces Apparatus
   measurements of interactions between rough and reactive calcite surfaces. Langmuir 34, 7248 7263.
- Dziadkowiec, J., Zareeipolgardani, B., Dysthe, D.K., Røyne, A., 2019. Nucleation in confinement
   generates long-range repulsion between rough calcite surfaces. Scientific reports 9, 1-15.
- Eppes, M., Hancock, G., Chen, X., Arey, J., Dewers, T., Huettenmoser, J., Kiessling, S., Moser, F.,
  Tannu, N., Weiserbs, B., 2018. Rates of subcritical cracking and long-term rock erosion.
  Geology 46, 951-954.
- Eppes, M.C., Keanini, R., 2017. Mechanical weathering and rock erosion by climate dependent
   subcritical cracking. Reviews of Geophysics 55, 470-508.
- Eskandari-Ghadi, M., Zhang, Y., 2021. Mechanics of shrinkage-swelling transition of microporous
  materials at the initial stage of adsorption. International Journal of Solids and Structures 222,
  111041.
- Eskandari-Ghadi, M., Zhang, Y., 2022. Modeling sorption-induced deformation of porous media: effect
   of pore size distribution. International Journal of Solids and Structures (Under Review).
- Freiman, S.W., Wiederhorn, S.M., Mecholsky, J., John J, 2009. Environmentally enhanced fracture of
   glass: a historical perspective. Journal of the American Ceramic Society 92, 1371-1382.
- 824 Gdoutos, E.E., 2020. Fracture mechanics: an introduction. Springer Nature.
- Hillerborg, A., Modéer, M., Petersson, P.-E., 1976. Analysis of crack formation and crack growth in
  concrete by means of fracture mechanics and finite elements. Cement and concrete research 6,
  773-781.
- 828 Irwin, G.R., 1957. Analysis of stresses and strains near the end of a crack traversing a plate.
- 829 Israelachvili, J.N., 2011. Intermolecular and surface forces. Academic press.
- B30 Jennings, S., 1988. The mean free path in air. Journal of Aerosol Science 19, 159-166.
- Karimpour, H., Lade, P.V., 2010. Time effects relate to crushing in sand. Journal of Geotechnical and
   Geoenvironmental Engineering 136, 1209-1219.
- Kilburn, C.R., Voight, B., 1998. Slow rock fracture as eruption precursor at Soufriere Hills volcano,
   Montserrat. Geophysical Research Letters 25, 3665-3668.
- Kim, J., Aluru, N., Tortorelli, D., 2003. Improved multi level Newton solvers for fully coupled
   multi physics problems. International journal for numerical methods in engineering 58, 463 480.
- Lawn, B., 1974. Diffusion-controlled subcritical crack growth in the presence of a dilute gas
   environment. Materials Science and Engineering 13, 277-283.

- Lawn, B.R., 1975. An atomistic model of kinetic crack growth in brittle solids. Journal of Materials
   Science 10, 469-480.
- Lawn, B.R., 1985. Interfacial forces and the fundamental nature of brittle cracks. Applied physics letters
   47, 809-811.
- Le, J.-L., Bažant, Z.P., Bazant, M.Z., 2009. Subcritical crack growth law and its consequences for
  lifetime statistics and size effect of quasibrittle structures. Journal of Physics D: Applied Physics
  42, 214008.
- Lennartz-Sassinek, S., Main, I., Zaiser, M., Graham, C., 2014. Acceleration and localization of
  subcritical crack growth in a natural composite material. Physical Review E 90, 052401.
- Li, D., Wei, D., 2020. Rate-Dependent Cohesive Zone Model for Fracture Simulation of Soda-Lime
   Glass Plate. Materials 13, 749.
- MAI, Y.W., Lawn, B.R., 1987. Crack interface grain bridging as a fracture resistance mechanism in ceramics: II, Theoretical fracture mechanics model. Journal of the American Ceramic Society 70, 289-294.
- Meng, F., Thouless, M., 2019. Cohesive-zone analyses with stochastic effects, illustrated by an example
   of kinetic crack growth. Journal of the Mechanics and Physics of Solids 132, 103686.
- Meyland, M.J., Nielsen, J.H., Kocer, C., 2021. Tensile behaviour of soda-lime-silica glass and the
   significance of load duration–A literature review. Journal of Building Engineering, 102966.
- Michalske, T.A., Freiman, S.W., 1982. A molecular interpretation of stress corrosion in silica. Nature
   295, 511-512.
- Nara, Y., Morimoto, K., Hiroyoshi, N., Yoneda, T., Kaneko, K., Benson, P.M., 2012. Influence of relative humidity on fracture toughness of rock: implications for subcritical crack growth. International Journal of Solids and Structures 49, 2471-2481.
- Needleman, A., 1990. An analysis of tensile decohesion along an interface. Journal of the Mechanics
   and Physics of Solids 38, 289-324.
- Nguyen, O., Repetto, E., Ortiz, M., Radovitzky, R., 2001. A cohesive model of fatigue crack growth.
  International Journal of Fracture 110, 351-369.
- 867 Olson, J.E., 1993. Joint pattern development: Effects of subcritical crack growth and mechanical crack
   868 interaction. Journal of Geophysical Research: Solid Earth 98, 12251-12265.
- Planas, J., Elices, M., 1991. Nonlinear fracture of cohesive materials, Current trends in concrete fracture
   research. Springer, pp. 139-157.
- Rice, J.R., 1978. Thermodynamics of the quasi-static growth of Griffith cracks. Journal of the
   Mechanics and Physics of Solids 26, 61–78.
- Røyne, A., Dalby, K.N., Hassenkam, T., 2015. Repulsive hydration forces between calcite surfaces and
  their effect on the brittle strength of calcite bearing rocks. Geophysical Research Letters 42,
  4786-4794.
- Sills, R., Thouless, M., 2015. Cohesive-length scales for damage and toughening mechanisms.
   International Journal of Solids and Structures 55, 32-43.
- Simmons, G.W., Pao, P., Wei, R., 1978. Fracture mechanics and surface chemistry studies of subcritical
   crack growth in AISI 4340 steel. Metallurgical Transactions A 9, 1147-1158.
- Tada, H., 1972. Studies of the crack opening stretch concept in application to several fracture problems.
   Lehigh University.
- Tada, H., Paris, P., Irwin, G., 2000. The analysis of cracks handbook. New York: ASME Press 2, 1.
- Tijssens, M., Van der Giessen, E., Sluys, L., 2000. Modeling of crazing using a cohesive surface
   methodology. Mechanics of Materials 32, 19-35.

- Tsilingiris, P., 2008. Thermophysical and transport properties of humid air at temperature range between
   0 and 100 C. Energy Conversion and Management 49, 1098-1110.
- Tvergaard, V., Hutchinson, J.W., 1992. The relation between crack growth resistance and fracture
   process parameters in elastic-plastic solids. Journal of the Mechanics and Physics of Solids 40,
   1377-1397.
- Wan, K.-T., Aimard, N., Lathabai, S., Horn, R.G., Lawn, B.R., 1990. Interfacial energy states of
   moisture-exposed cracks in mica. Journal of Materials Research 5, 172-182.
- Wiederhorn, S., 1967. Influence of water vapor on crack propagation in soda lime glass. Journal of the
   American Ceramic Society 50, 407-414.
- Wiederhorn, S., 1969. Fracture surface energy of glass. Journal of the American Ceramic Society 52,
   99-105.
- Wiederhorn, S., 1974. Subcritical crack growth in ceramics, Fracture mechanics of ceramics. Springer,
   pp. 613-646.
- Wiederhorn, S., Bolz, L., 1970. Stress corrosion and static fatigue of glass. Journal of the American
   ceramic society 53, 543-548.
- Wiederhorn, S., Fuller Jr, E., Thomson, R., 1980. Micromechanisms of crack growth in ceramics and
   glasses in corrosive environments.
- Wiederhorn, S.M., Fuller Jr, E.R., 1989. Effect of surface forces on subcritical crack growth in glass.
   Journal of the American Ceramic Society 72, 248-251.
- Wu, K., Li, X., Wang, C., Yu, W., Chen, Z., 2015. Model for surface diffusion of adsorbed gas in nanopores of shale gas reservoirs. Industrial & Engineering Chemistry Research 54, 3225-3236.
- Xu, X., Xu, J., Chen, J., Li, P., Liu, B., Li, Y., 2017. Investigation of dynamic multi-cracking behavior
   in PVB laminated glass plates. International Journal of Impact Engineering 100, 62-74.
- Yaminsky, V., Stewart, A., 2003. Interaction of glass surfaces in air: Dispersion forces in the retarded
   regime. Langmuir 19, 4037-4039.
- Zhang, Y., Buscarnera, G., 2018. Breakage mechanics for granular materials in surface-reactive
   environments. Journal of the Mechanics and Physics of Solids 112, 89-108.

## 913 APPENDIX A. NUMERICAL SOLUTION TO THE II-W SYSTEM

914 The solution to the  $\Pi - w$  system involves finding a consistent set of  $\Pi(x)$  and w(x) that satisfy 915 two equations of the general forms  $w(x) = w(x, \Pi(x))$  and  $\Pi(x) = \Pi(x, w(x))$ . Eq. (2) is an example of 916 the former, and Eqns. (9) and (10) take the form of the latter.

To write Eq. (2) in discrete form, an approximation for the integral on the right-hand side is required. Such approximation is commonly performed by adopting Gauss quadrature methods. However, due to the singularity of the fundamental solution, Eq. (4), at x = s, Gauss quadrature approximation encountered numerical convergence issues. Here an alternative approximation is adopted. Refer to the spatial discretization in Section 6. First, the disjoining pressure over the range of  $x_i$  and  $x_{i+1}$ ,  $\Pi_i^{elem}(x)$ , is approximated with a constant value,  $\Pi_i^{elem} = (\Pi(x_i) + \Pi(x_{i+1}))/2 = (\Pi_i + \Pi_{i+1})/2$ . With this, Eqns. (1) and (2) can be written at each node as

924 
$$w(x_{i}) = c_{\sigma}(x_{i})\sigma + \sum_{j=1}^{N-1} \prod_{j=1}^{elem} \underbrace{\int_{s=s_{j}}^{s=s_{j+1}} c_{\Pi}(x_{i},s)ds}_{\mathbf{c}_{\Pi}^{elem}} \text{ and } K_{Ie} = k_{I\sigma}\sigma + \sum_{j=1}^{N-1} \prod_{j=1}^{elem} \underbrace{\int_{s=s_{j}}^{s=s_{j+1}} k_{I\Pi}(s)ds}_{\mathbf{k}_{\Pi}^{elem}}$$
(A.1)

925 The fundamental functions  $c_{\Pi}(x, s)$  and  $k_{I\Pi}(s)$  are now the crack opening profile and the stress 926 intensity factor for uniformly distributed stress over range  $(s_j, s_{j+1})$  in Fig. A.1a, which are non-singular. 927 For unit repulsive stress ( $\Pi = 1$ ) applied over length *s* in Fig. A.1b, Tada (1972) provides

928 
$$\int_{0}^{s} c_{\Pi}(x,s) ds = \frac{8(1-\nu^{2})}{\pi E} \begin{cases} \sqrt{a^{2}-x^{2}} \sin^{-1}\frac{s}{a} + \frac{1}{2} |x-s| \cosh^{-1}\frac{a^{2}-sx}{a |x-s|} \\ + \frac{1}{2} |x+s| \cosh^{-1}\frac{a^{2}+sx}{a |x+s|} \end{cases}$$
(A.2)

929 
$$\int_{0}^{s} k_{III}(s) ds = 2\sqrt{\frac{a}{\pi}} \sin^{-1} \frac{s}{a}$$
(A.3)



The fundamental solutions for uniform stress over an arbitrary segment  $(s_j, s_{j+1})$  in the integrals of 932 (A.2) 933 directly terms Eq. (A.1) can be expressed in of Eqns. and (A.3), as  $\int_{s=s_j}^{s=s_{j+1}} c_{\Pi}(x_i, s) ds = \int_0^{s_{j+1}} c_{\Pi}(x_i, s) ds - \int_0^{s_j} c_{\Pi}(x_i, s) ds \quad \text{and} \quad \int_{s=s_j}^{s=s_{j+1}} k_{I\Pi}(s) ds = \int_0^{s_{j+1}} k_{I\Pi}(s) ds - \int_0^{s_j} k_{I\Pi}(s) ds \quad .$ 934

935  $c_{\sigma}(x_i)$  and  $k_{I\sigma}$  are given by Eq. (3). Eq. (A.1) can be written in a fully discretized form

936 
$$w_i = c_{\sigma,i}\sigma + \sum_{j=1}^{N-1} c_{\Pi,ij}^{elem} \Pi_j^{elem} \text{ and } K_{Ie} = k_{I\sigma}\sigma + \sum_{j=1}^{N-1} k_{I\Pi,j}^{elem} \Pi_j^{elem}$$
(A.4)

937 where  $c_{\sigma,i} = c_{\sigma}(x_i)$ .  $\Pi_i^{elem} = (\Pi_i + \Pi_{i+1})/2$  can also be written as a tensor transformation,  $\Pi^{elem} = \mathbf{A}_{\Pi} \cdot \mathbf{\Pi}$ ,

938 where

930 931

939 
$$\mathbf{A}_{\Pi} = \begin{bmatrix} \frac{1}{2} & \frac{1}{2} & 0 & 0 & 0 & 0\\ 0 & \frac{1}{2} & \frac{1}{2} & 0 & \cdots & 0 & 0\\ 0 & 0 & \frac{1}{2} & \frac{1}{2} & 0 & 0\\ \vdots & & \ddots & 0 & 0\\ 0 & 0 & 0 & 0 & 0 & \frac{1}{2} & \frac{1}{2} \end{bmatrix}_{(N-1) \times N}$$
(A.5)

940 Finally, Eq. (A.4) can be written in matrix-vector notation as:

941 
$$\mathbf{w} = \mathbf{c}_{\sigma} \boldsymbol{\sigma} + \mathbf{c}_{\Pi} \cdot \boldsymbol{\Pi}$$
(A.6)

$$K_{\mu} = k_{I\sigma}\sigma + \mathbf{k}_{III} \cdot \mathbf{\Pi} \tag{A.7}$$

943 where  $\mathbf{c}_{\Pi} = \mathbf{c}_{\Pi}^{elem} \cdot \mathbf{A}_{\Pi}$  and  $\mathbf{k}_{\Pi} = (\mathbf{k}_{\Pi}^{elem})^T \cdot \mathbf{A}_{\Pi}$ . Substituting the relation  $\Pi = \Pi(w, p)$  into Eq. (A.6) 944 yields a nonlinear algebraic equation that can be solved iteratively. The solution is sought here by 945 rewriting Eq. (A.6) in the same way as Planas and Elices (1991):

946 
$$\mathbf{\Pi}(\mathbf{w},\mathbf{p}) - \mathbf{M} \cdot \mathbf{w} + \mathbf{L}\boldsymbol{\sigma} = \mathbf{0}$$
(A.8)

947 where **p** is the nodal vector of gas partial pressure assumed to be known and constant in this local solver, 948  $\mathbf{M} = \mathbf{c}_{\Pi}^{-1}$ , and  $\mathbf{L} = \mathbf{c}_{\Pi}^{-1} \cdot \mathbf{c}_{\sigma}$ . A Newton-Raphson iteration method is implemented to seek the solution  $w_i$ 949 at each node. Starting with an initial guess,  $\mathbf{w}_0$ , the correction  $\mathbf{w}_{k+1} = \mathbf{w}_k + \delta \mathbf{w}_k$  at iteration *k* is made 950 where  $\delta \mathbf{w}_k = -(\mathbf{J}^M)^{-1} \cdot \mathbf{R}^M$ .  $\mathbf{R}^M$  and  $\mathbf{J}^M$  are computed as

951 
$$R_i^M = \Pi(w_i, p_i) - M_{ij}w_j + L_i\sigma \text{ and } J_{ij}^M = \frac{\partial \Pi}{\partial w}\Big|_{w_i, p_i} \delta_{ij} - M_{ij}$$
(A.9)

952 with  $\mathbf{w} = \mathbf{w}_k$ . The search for solution continues until the error defined as

953 
$$\mathcal{E}_M = \sqrt{\sum_{i=1}^{N-1} (R_i^M)^2 \Delta x_i}$$
(A.10)

is reduced below a tolerance (here taken as  $5 \times 10^{-4}$ ). Once  $\Pi$  and w profiles are obtained, Eq. (A.7) computes  $K_{le}$  to conclude the  $\Pi - w$  system solver.

956 Our  $\Pi$  – *w* system solver is verified by comparing the computed crack opening profile with the 957 analytical crack opening solutions of Tada et al. (2000) subjected to different surface force distributions 958 (Fig. A.2).



959

960 Fig. A.2 Verification of the normalized crack opening profile,  $\tilde{w} = wE / (1 - v^2)a$ , with exact solutions (Tada et al., 961 2000) for surface force distributions (a) linearly increasing toward the crack tip; (b) linearly decreasing toward the crack tip; 962 and (c) nonlinearly increasing toward the crack tip. Figure insets are courtesy of Tada et al. (2000).

#### 964 APPENDIX B. INITIAL GUESS FOR T > 0 AND T = 0

965 The current implementation is stress-controlled (i.e.,  $\sigma$  and a vary between timesteps). The crack 966 velocity is small and the timesteps are close, thus the variation of a over timesteps is very small. 967 However, a small variation in  $\sigma$  can create a large change in  $K_{le}$  which exponentially updates  $v_c$  and 968 subsequently p(x). It is thus apparent that using the solution at timestep  $t_k$  as the initial guess for 969 timestep  $t_{k+1}$  can involve large errors and prevent convergence.

970 We can obtain an initial guess for timestep  $t_{k+1}$  by a linear extrapolation. The rate of change of the 971 solution vector, **u**, over time  $\Delta t$  is linearized based on a small variation of  $\sigma$  over a small time  $.\theta\Delta t$ . as

972 
$$\frac{\Delta \mathbf{u}}{\Delta t} \approx \frac{\mathbf{u} |_{\sigma(t_k + \theta \Delta t)} - \mathbf{u} |_{\sigma(t_k)}}{\theta \Delta t}$$
(B.1)

The value of  $\theta$  is set to  $10^{-2}$ . Since the variation of  $\sigma$  over the time interval  $\theta \Delta t$  is now small, the solution at time  $t_k$  works as a good initial guess for the solution at time  $t_k + \theta \Delta t$ . Thus,  $\mathbf{u}|_{\sigma(t_k+\theta\Delta t)}$  is found using the same partitioned Newton-Raphson solver with the initial guess of  $\mathbf{u}|_{\sigma(t_k)}$ . The sole purpose of  $\mathbf{u}|_{\sigma(t_k+\theta\Delta t)}$  is to assist with finding a reasonable initial guess for  $\mathbf{u}|_{\sigma(t_k+\Delta t)}$ , it can thus be calculated with a more relaxed tolerance to reduce simulation time. Then, the initial guess for time  $t_{k+1}$  is computed via linear extrapolation:

$$\mathbf{u}|_{\sigma(t_k+\Delta t)} \approx \mathbf{u}|_{\sigma(t_k)} + \frac{\Delta \mathbf{u}}{\Delta t} \Delta t$$
(B.2)

While the procedure described here involves an additional set of iterations in search of  $\mathbf{u}|_{\sigma(t_k+\theta\Delta t)}$ , the amount of computation required for those iterations is far less than that required for finding the solution to  $\mathbf{u}|_{\sigma(t_k+\Delta t)}$  using the initial guess  $\mathbf{u}|_{\sigma(t_k)}$  even if the latter converges.

979

The initial guess for t = 0 can be obtained by starting simulations with a small  $\sigma$  so that  $v_c$  can be predicted a few orders of magnitude smaller than d/D. This way, the fluid profile is known,  $p(x) = p_0$ . Thus, the  $\Pi - w$  system can be solved alone for  $K_{le}$ . One must then verify that this  $K_{le}$  results in  $v_c$  of similar magnitude as the predicted one.