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Special Issue: Biomass as a path to sustainability

## Opinion

# Prospects for carbon-negative biomanufacturing

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Biomanufacturing has the potential to reduce demand for petrochemicals and mitigate climate change. Recent studies have also suggested that some of these products can be net carbon negative, effectively removing  $CO_2$  from the atmosphere and locking it up in products. This review explores the magnitude of carbon removal achievable through biomanufacturing and discusses the likely fate of carbon in a range of target molecules. Solvents, cleaning agents, or food and pharmaceutical additives will likely re-release their carbon as  $CO_2$  at the end of their functional lives, while carbon incorporated into non-compostable polymers can result in long-term sequestration. Future research can maximize its impact by focusing on reducing emissions, achieving performance advantages, and enabling a more circular carbon economy.

#### Importance of biomanufacturing to climate change mitigation

Rising oil prices and the climate crisis are motivating many countries to once again evaluate their reliance on petroleum-derived liquid fuels and petrochemicals [1,2]. While electrification is likely to reduce demand for gasoline and potentially diesel fuel, petrochemical production has fewer lowor zero-emission alternatives [3]. **Biomanufacturing** (see Glossary) offers the promise of replacing these fossil-derived chemicals and materials with renewable drop-in or performance-advantaged alternatives, while also locking atmospheric carbon up in stable forms such as building materials and plastics (many of which are ultimately landfilled after use). A number of recent papers have presented new biomanufacturing processes accompanied by a claim that they are (or can be) net carbon negative [4–7]. This raises a few questions. First, are these processes truly carbon negative? Second, is the potential magnitude of carbon removal in biomanufacturing meaningful relative to what is required to stabilize the climate? Third, how can synthetic biology and metabolic engineering research be most effectively leveraged as part of broader efforts to make manufacturing more sustainable? To answer these questions, it is useful to discuss some target molecules that can be made biologically, what applications they are used for, and how carbon flows through feedstocks, microbial production systems, and products in typical biomanufacturing systems.

#### State of biomanufacturing and potential to reduce petrochemical consumption

Biomanufacturing is defined as manufacturing that uses biological systems, including microbes, plant cells, and enzymes, to produce commercially relevant molecules [8]. Biomanufacturing already plays an important role in the production of pharmaceuticals, flavors and fragrances, and cosmetic additives [9]. However, these are small-volume applications. Achieving meaningful net-negative emissions requires an entirely different scale of production than what is achievable in specialty chemical and pharmaceutical markets. As such, this review will focus primarily on commodity chemicals/products.

There are so far only a small number of microbially produced commodity chemicals, including 1,3-propanediol, 1,4-butanediol, isobutanol, farnesene, lactic acid, and succinic acid [10].



#### Highlights

Future biomanufacturing studies must recognize the important distinction between greenhouse gas mitigation relative to the status quo and processes that result in true carbon dioxide removal.

The potential for biomanufacturing to serve as a carbon dioxide removal strategy is limited because many target products are re-oxidized to  $CO_2$  at the end of their useful life, yet this is not accounted for in recent studies.

Many precursors to commodity polymers can be made biologically and polymers offer the largest opportunity to achieve carbon dioxide removal through biomanufacturing.

Bio-based materials do reduce reliance on petroleum and may offer performance advantages relative to conventional petrochemical alternatives.

Instead of focusing on net carbon negativity, future research should focus on using biotechnology to enable a more circular and sustainable carbon economy.

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However, there has been progress in the development and optimization of new biomanufacturing routes; significant strides have been made in gas fermentation processes that utilize gases that are either low-value or waste products, such as untreated biogas, gasified crop or forest residues, and steel mill waste gas [6,11,12]. Some commodity chemicals/polymers, such as polyhydroxybutyrate (PHB), muconic acid, catechol, limonene, and latex, can also be accumulated in plants and extracted as part of a biomanufacturing system [13,14].

The question of whether most bio-based commodity chemicals and materials can reduce reliance on oil and gas has a clear answer: yes. The majority of commonly used polymers and industrial solvents are produced from petroleum and/or natural gas and, while aerobic microbial production processes can be electricity intensive, their life-cycle consumption of oil and gas is generally lower than the conventional petrochemical alternative. A study from the Argonne National Laboratory reported that reductions in total fossil energy consumption between fossil and bio-based propylene glycol, 1,3-propanediol, acrylic acid, polyethylene, succinic acid, isobutanol, and 1,4-butanediol ranged from 24% to 76% [15]. The question of whether biomanufacturing can and should be claimed as **carbon negative** is more complex and requires knowledge of how carbon flows through these systems into the use phase and end of life of the final products.

#### Emissions mitigation versus carbon-negative products

There is consensus among energy systems modelers and climate modelers that, in order to slow and ultimately stop anthropogenic climate change, most sectors must reach net-zero greenhouse gas (GHG) emissions, through a combination of renewable electricity generation, energy storage, electric transmission infrastructure investments, hydrogen production and use, and production of renewable liquid fuels [3]. This outcome is sometimes referred to as decarbonization, although it does not literally mean that all carbon is eliminated from the systems. Carbon-based fuels from renewable feedstocks can be part of a decarbonization strategy. These measures are all a form of **emissions mitigation**. In other words, they can replace high(er) emitting technologies that we rely on today through technological advancements and infrastructure investments, thus reducing or in some cases eliminating emissions while maintaining similar levels of service (e.g., mobility, lighting, and thermal comfort in buildings). In the context of biomanufacturing, studies have repeatedly indicated that bio-based chemicals can reduce life-cycle GHG emissions when compared against conventional petrochemicals [7,15–18].

In contrast to mitigation, a carbon-negative technology will continue to achieve net **CO<sub>2</sub> removal** from the atmosphere regardless of how much emissions mitigation has occurred in the broader economy [19]. In the near term, there is no inherent benefit to removing CO<sub>2</sub> relative to avoiding a comparable mass of emissions. The impact on atmospheric CO<sub>2</sub> concentrations and climate is identical. Policy makers and industry decision-makers are best served by devoting available resources to the most cost-effective options for avoiding or removing CO<sub>2</sub>. However, net carbon removal is deemed necessary to reach long-term climate stabilization by compensating for sectors whose emissions, and halt or reverse anthropogenic climate change, humans must make long-term investments in processes that remove CO<sub>2</sub> or other GHGs from the atmosphere and sequester them in a stable form. This is why carbon-negative technologies are of specific interest to the research community.

#### Carbon uptake routes in biomanufacturing

There are two primary mechanisms for removing  $CO_2$  from the atmosphere: **direct air capture (DAC)** and uptake of  $CO_2$  by plants. Other options do exist, such as enhanced weathering [21],

#### Glossary

**Biomanufacturing:** manufacturing that uses biological systems, including microbes, plant cells, and enzymes, to produce commercially relevant molecules.

**Carbon negative:** this descriptor refers to a process that achieves net  $CO_2$ removal per unit of output. Usage of this term in the literature has been somewhat inconsistent, with some using it to indicate net removal, while others use the term more loosely to describe any process that achieves net removal or mitigation. For example, processing manure in anaerobic digesters can achieve net mitigation because it avoids substantial methane emissions from lagoon storage. However, this process does not achieve  $CO_2$  removal.

**CO<sub>2</sub> removal:** strategies that remove CO<sub>2</sub> from the atmosphere and either sequester gaseous CO<sub>2</sub> underground or convert the carbon to some other stable, storable form.

Cradle-to-gate: a term commonly used to denote system boundaries in a life-cycle assessment. This determines what is or is not included in the analysis. 'Cradle' refers to raw material extraction and 'gate' refers to the production facility gate. Thus, a cradle-to-gate life-cycle GHG assessment includes all emissions and sequestration occurring during raw material production (e.g., crop production), transportation of materials to the production facility, and the production process itself. Final product transportation beyond the facility gate, product use, and end of life is excluded from the analysis

Cradle-to-grave: a term commonly used to denote system boundaries in a life-cvcle assessment. This determines what is or is not included in the analysis. 'Cradle' refers to raw material extraction and 'grave' refers to the end of life (e.g., disposal or recycling) of the final product. Thus, a cradle-to-grave life-cycle GHG assessment includes all emissions and sequestration occurring during raw material production (e.g., crop production), transportation of materials to the production facility, the production process, transportation to the point of use, the use phase, and end of life. Direct air capture (DAC): systems

that pass air through either a chemical solution, solid sorbent, or other material to remove  $CO_2$ , which is later released for capture and storage (or use) in a concentrated form.



although these are earlier in their development. DAC offers a straightforward value proposition: removal of  $CO_2$  for a price (and an energy footprint). It also requires geologic  $CO_2$  storage or some path to  $CO_2$  utilization that does not re-emit carbon to the atmosphere, such as biological conversion by algae or gas fermentation. As long as the cost of DAC exceeds the cost of  $CO_2$ emissions mitigation strategies on a per-tonne basis, emissions mitigation should be prioritized over DAC. However, making early investments in research, development, and demonstration is important to ensure that future scale-up of  $CO_2$  removal can occur. In contrast to DAC,  $CO_2$ uptake by plants and the use of these feedstocks in biomanufacturing offer greater economic co-benefits in the form of saleable products, but less certainty as to the scalability and longterm stability of carbon storage. Plants will fix carbon in the form of cellulose, hemicellulose, lignin, starch, sugars, proteins, pectins, and other compounds. These forms of carbon are not stable and, in the absence of intervention, will decompose to  $CO_2$ . Deep-rooted plants may also sequester carbon to soils for some finite period of time [22], although soil carbon sequestration is outside the primary focus of this review.

Plant biomass, once harvested, can serve as the input to biomanufacturing processes. In many biomanufacturing processes, sugars serve as the feedstock, and these sugars are consumed by microbes capable of producing the products of interest or intermediates that can be chemically upgraded to final products. Lignin-derived compounds or an intermediate produced from plant material, such as biogas from anaerobic digestion or syngas from **gasification**, may also be used as a feedstock [7,23]. Biogenic (meaning nonfossil) carbon entering the facility in the form of sugars or other bioavailable compounds will be incorporated into the product, some will be emitted as  $CO_2$  during bioconversion, additional carbon will be incorporated into cell mass, and some residual carbon will remain in wastewater for further treatment. The two primary opportunities for sequestration lie in the  $CO_2$  waste stream from bioconversion (and potentially wastewater treatment), if captured, and the final product itself.

When exploring the potential for capturing the waste CO<sub>2</sub> stream from bioconversion, there is a crucial distinction between processes that operate anaerobically, those that operate aerobically by sparging with pure oxygen, and those that operate aerobically by sparging with air. Anaerobic processes and those sparged with pure oxygen produce a nearly pure CO<sub>2</sub> stream that can be captured and sequestered without the use of amine scrubbers, membrane separation, or other processes required to separate  $CO_2$  from  $N_2$  and other gases in combustion flue gases [24]. Cell-based meat production, for example, may sparge bioreactors with O<sub>2</sub>-enriched air and, as a result, can offer an opportunity for (comparatively) low-cost carbon capture and sequestration [25]. Aerobic processes, which are more common for the production of many advanced bioproducts, produce a more dilute CO<sub>2</sub> waste stream that will require scrubbers or some other separation as described previously. In either case, once this  $CO_2$  is captured, it must be pumped to underground storage or converted into some other stable form. Recent research has explored biological conversion of CO<sub>2</sub> to products provided an energy source is available (e.g., H<sub>2</sub> or CH<sub>4</sub>) [23]. While this does not guarantee net carbon negativity, it does offer the possibility of GHG emissions reductions and another useful life for carbon that would likely otherwise be emitted.

Once the feedstock, whether it is plant material or a gaseous input, is converted to a product, the ultimate use and disposal (or recycling) of that product is the key to whether it will sequester carbon or simply release  $CO_2$  back to the atmosphere. This is the distinction between achieving a more circular carbon economy and achieving net carbon negativity. Bio-based materials whose carbon is re-oxidized to  $CO_2$  at their end of life still mitigate climate change; that carbon will be taken up through re-growth of feedstocks. However, the only way to permanently (or semi-

Emissions mitigation: the avoidance of emissions to the atmosphere that would otherwise occur, through either a reduction in activity (e.g., reduction in energy use) or capture and sequestration.

Gasification: thermal conversion of carbonaceous material into a gaseous product called syngas through partial oxidation.

Incineration: combustion, particularly for waste streams, aimed at destroying the material. Incineration can be used with or without energy recovery. Municipal, industrial, and hazardous waste can all be incinerated. System boundary: the scope of activities that are incorporated in an analysis. System boundary is an important concept in life-cycle assessments.



permanently) remove carbon from the atmosphere is to sequester it underground or in some other stable form. Table 1 summarizes some key chemicals that can be produced biologically for use in materials, with a focus on products that currently rely on petrochemicals. It also lists the likely fate of different products once they reach the end of their useful life. Some products and precursors cannot yet be directly biologically produced. If its production requires one or more chemical upgrading steps, it is not included in Table 2. For example, terephthalic acid can only be produced through chemical or hybrid biological–chemical approaches [26], although some have suggested that there may be novel biological routes developed in the future [27].

Table 1. Biomanufactured chemicals, global demand, carbon content, likely fate of carbon by application type, and recent studies on biomanufacturing processes

Biomanufactured chemical	Approximate annual global demand (million tonnes/year)	Carbon content (mass %)	Market applications	Typical fate(s) of carbon at end of life	Refs <sup>a</sup>
Ethylene	165 (in 2017) <sup>i</sup>	86%	Polymers, precursor to many industrial chemicals, hormone for plant ripening/flowering	Polymers: landfilled, incinerated, leaked to environment Compostable polymers: landfilled,	[28–31]
Ethanol	98 (in 2021) <sup>ii</sup>	52%	Fuel, solvent, food, pharmaceuticals	incinerated, leaked to environment, oxidized to CO <sub>2</sub> during composting	[32,33]
Styrene	28 (in 2019) <sup>iii</sup>	92%	Polymers	Solvents: primarily oxidized to CO <sub>2</sub>	[34–37]
Monoethylene glycol (MEG)	25 (in 2020) <sup>iv</sup>	39%	Polymers, engine coolants, antifreeze	upon disposal through <b>incineration</b> or other destructive treatment	[38–40]
Propylene glycol	2.7 (in 2020) <sup>∨</sup>	47%	Polymers, food, pharmaceuticals, e-cigarettes	Lubricants: primarily oxidized to CO <sub>2</sub> (after recycling), some automotive lubricants incorporated into asphalt after disposal	[41,42]
Acetone	6.85 (in 2021) <sup>vi,vii</sup>	62%	Industrial solvent, precursor to acrylic glass and bisphenol A	Resins/coatings: landfilled, incinerated Pharmaceuticals, food, personal care, cleaning agents: oxidized to $CO_2$ (a fraction of landfilled food waste will be emitted as $CH_4$ from landfills)	[6,43]
Isobutanol	6.7 (in 2021) <sup>viii</sup>	65%	Resins and coatings, solvent, plasticizers, fuel		[44–48]
Acrylic acid	6.3 (in 2020) <sup>ix</sup>	50%	Polymers		[49,50]
Adipic acid	3.9 (in 2018) <sup>×</sup>	49%	Polymers, food and pharmaceutical additive	Fuels: oxidized to $\rm CO_2$	[51–54]
1,4-butanediol	2.79 (in 2020) <sup>xi</sup>	53%	Solvent, polymers		[55–57]
IPA	2.15 (in 2020) <sup>vii</sup>	60%	Pharmaceuticals, cosmetics and personal care products, solvent and cleaning agent		[6,58]
Lactic acid	1.39 (in 2021) <sup>xiii</sup>	40%	Polymers, pharmaceuticals, cleaning products		[59–61]
Sebacic acid	0.2 (production capacity, 2020) <sup>xiv</sup>	59%	Polymers, lubricants		[62,63]
1,3-propanediol	0.146 (in 2014) <sup>×v</sup>	47%	Polymers, solvent		[57,64–68]
Succinic acid	0.016–0.03 (in 2021) <sup>xvi,xvii</sup>	41%	Polymers, food, pharmaceuticals, solvent production		[69–71]
Farnesene	0.00815 (in 2015) <sup>xviii</sup>	88%	Lubricants, cosmetics, fragrances, fuel		[72,73]
Diamines	Varies by compound	~50% (varies)	Polymers, pharmaceuticals, many others		[74]

<sup>a</sup>Recent studies on biomanufacturing processes.



Polymer resin types	Global demand 2022 (million tonnes/year) <sup>a</sup>	Projected 2030 demand (million tonnes/year) <sup>a</sup>	Approximate carbon content (mass %) <sup>b</sup>	2030 annual carbon sink potential: 90% landfilled (million tonnes CO <sub>2e</sub> /year)°	2030 annual carbon sink potential: 50% landfilled (million tonnes CO <sub>2e</sub> /year)°	Key chemical precursors
PET	27	28	63%	58	32	Monoethylene glycol (MEG); terephthalic acid
HDPE	61	63	86%	180	99	Ethylene
PVC	56	58	38%	73	40	Chlorine, ethylene
LDPE/LLDPE	59	61	86%	170	96	Ethylene
PP	80	82	86%	230	130	Propylene
PS	23	24	92%	73	40	Styrene
PUR	20	20	41%	27	15	Diisocyanates, polyols (including 1,4-butanediol, MEG)
Fibers	66	68	70% <sup>d</sup>	160	87	MEG, terephthalic acid, acrylic acid, sebacic acid, adipic acid, caprolactam succinic acid, cellulose, 1,3-propanediol, diamines, many others
Other	110	110	66%	240	133	Many others
Total	502	514	N/A	1212	673	N/A

#### Table 2. Global demand and approximate carbon storage potential of commodity polymers

Abbreviations: HDPE, high-density polyethylene; LDPE, low-density polyethylene; LLDPE, linear low-density polyethylene; PET, polyethylene terephthalate; PP, polypropylene; PS, polystyrene; PUR, polyurethane; PVC, polyvinyl chloride.

<sup>a</sup>Extrapolated from 2019 global demand<sup>xx</sup> based on an assumed 3% annual growth rate.

<sup>b</sup>Carbon contents based on US Environmental Protection Agency data [81], except where noted.

<sup>c</sup>Calculated by multiplying 2030 projected demand by resin-specific fractional carbon content, fraction of resin type landfilled, and (44/12) to convert carbon mass to CO<sub>2</sub> mass.

<sup>d</sup>Calculated based on the average (mean) carbon contents of PET, nylon 6, and acrylic.

#### Bio-based products as carbon sinks

For a biomanufacturing process to be carbon negative, it must sequester more carbon than it emits over the entire lifetime of the process. Usually, this means biogenic carbon is incorporated into the product that remains sequestered in some stable form through its end of life (Figure 1, Key figure). This narrows the range of products that can reasonably be deemed carbon negative. Solvents, for example, (Table 1) are commonly combusted at their end of life or discharged to a wastewater treatment plant where their carbon will ultimately be oxidized to CO<sub>2</sub> and released into the atmosphere. Two recent studies are notable for having claimed net carbon-negative production. Liew and colleagues [6] published net-negative carbon footprints for acetone and isopropanol (IPA), which are primarily used as solvents but can be used in other applications. Wang and colleagues [4] also claimed a negative carbon footprint for production of a benzene, toluene, ethylbenzene, and xylene (BTEX) product mixture. In both cases, the studies used a system boundary (scope) known as cradle-to-gate (as opposed to cradle-to-grave) (see the description by Scown and Keasling [10]). In other words, they did not include the final fate of carbon in those products once they leave the production facility. If the results are used to compare a bio-based route to a fossil-based production route, the choice of system boundary is immaterial, as both products will be treated identically in their use and disposal. However, in an absolute sense, the choice of system boundaries affects the conclusions. These negative emissions values should not necessarily be interpreted as an indication that such processes will achieve net carbon removal from the atmosphere.



### Key figure

Carbon flows from feedstock through biomanufacturing and product end of life



Figure 1. Most products that have been targeted for biomanufacturing, with the exception of polymers and some building materials, are oxidized either over the course of their use or during their end of life. This means that, while biomanufacturing processes are able to utilize atmospheric carbon (in the form of plants or other carbon capture mechanisms), many of the target molecules will be used in products that will be re-oxidized to  $CO_2$  during their disposal. Only selected polymers and other materials that are stable in the long term will serve as a true carbon sink. Incorporating  $CO_2$  capture into biomanufacturing processes can tip the balance toward net carbon removal, which may be particularly advantageous for anaerobic processes or aerobic processes that sparge bioreactors with  $O_2$ -enriched gas. Abbreviations: CCUS, carbon capture, utilization, and storage; WWT, wastewater treatment.

Some bioproducts do have greater potential to lock carbon away in a stable form. Automotive lubricants may be recycled, with some portion diverted for use as a low-value fuel, and an additional fraction incorporated into asphalt. In the case of lubricants, it makes sense to partially credit this carbon going to asphalt as being sequestered, with practices likely varying by region [75]. The fate of bio-based precursors to polymers, such as polyethylene terephthalate (PET), high-density polyethylene (HDPE), low and linear low-density polyethylene (LDPE and LLDPE, respectively), polypropylene (PP), and nylon will depend on where the materials are ultimately used. In Germany, for example, 61% of plastics were incinerated as of 2016, 39% were recycled, and <1% were landfilled [76]. By contrast, >90% of plastics in the USA are landfilled<sup>xix</sup>. This can make a generalized carbon accounting for bio-based plastics challenging. The same material may be carbon negative in one country and not in another. Compostable polymers, if successfully



broken down in commercial composting facilities, will also release their carbon back into the atmosphere. However, evidence suggests that many do not; polylactic acid (PLA) does not break down sufficiently in commercial composting operations and must be screened out and landfilled [77]. If landfilled, the majority of carbon in PLA will remain sequestered.

Polymers likely represent one of the largest opportunities for biomanufacturing to sequester carbon in stable forms, assuming the majority of material is not incinerated or fully broken down through composting. Table 2 shows the most common commodity polymers, the carbon contained in total global production for each type, and the primary chemical precursors. Future demand is estimated based on the simplifying assumption of 3% annual demand growth each year across all polymer types. Only a subset of these materials can currently be replaced with bio-based alternatives. However, estimating the total carbon storage potential is useful for gauging the relative significance of bio-based polymers in broader climate change mitigation efforts. For perspective, 2021 global energy-related CO<sub>2</sub> emissions were estimated to be 34.9 Gt [78]. The simple calculations in Table 2 suggest that, even if all polymer products (plastics, rubbers, fibers, and so forth) were produced from bio-based materials and then 90% of those products were landfilled, this strategy would sequester approximately 1.2 Gt CO<sub>2</sub> annually, or 3.4% of current annual emissions. If recycle rates for these polymers increase, there will be a proportional decrease in the potential for carbon sequestration, effectively creating a circular carbon economy rather than net carbon removal. Achieving a 50% recycle rate for polymers would reduce the sequestration potential to around 0.67 Gt CO<sub>2</sub>/year or 1.9% of global energy-related CO<sub>2</sub> emissions (Table 2). It is worth noting that treating plastics as a carbon sink rather than focusing on improving recycle rates has other serious implications. This point is echoed in a recent article by Meys and colleagues [79], which focuses on the goal of achieving net carbon-neutral plastics through increased circularity rather than attempting to treat plastics as a carbon sink. Accumulation of waste in landfills limits the future use of that land for other beneficial purposes, while improper disposal of plastic waste causes the accumulation of microplastics in the environment, the health implications of which are still not fully understood [80].

Building materials are another potentially interesting carbon sink, although the role for biomanufacturing is still evolving. The largest opportunities for carbon sequestration are in the bulk materials that can include bio-based materials: timber and concrete. Bio-aggregates such as hemp or wood can be incorporated into concrete. Much like polymers, there are potential tradeoffs between treating building materials as carbon sinks versus designing buildings for increased recyclability and waste reduction. The dual (and sometimes competing) priorities of increased recyclability, carbon sequestration, and improved use-phase performance are reflected in recent selections for the US Advanced Research Projects Agency-Energy (ARPA-E) Harnessing Emissions into Structures Taking Inputs from the Atmosphere (HESTIA) program<sup>XXI</sup>, which includes projects that aim to develop everything from cellulosemycelium composites to bio-based adhesives for oriented strand board and bio-based concrete additives/aggregates. Increasing sequestration of biogenic carbon in buildings by using timber in large commercial or multi-family residential buildings, in contrast to some of the materials discussed previously, is occurring and is likely to continue. A recent analysis explored the magnitude of carbon that might be sequestered annually if timber use were increased in construction in the form of glue-laminated (glulam) beams or cross-laminated timber (CLT) panels [82]. In a business-as-usual scenario, only 0.037 Gt CO<sub>2</sub>/year would be sequestered and if timber use were dramatically increased in urban structures, as much as 2.5 Gt CO<sub>2</sub>/ vear could be sequestered [82]. These values do not account for any wood that is combusted or otherwise oxidized in construction and demolition waste each year. There



are additional efforts to develop materials with higher carbon contents; for example, biochar can be incorporated into building materials to increase the total carbon content to around 90% [82]. The field of biomanufacturing carbon-negative building materials is rapidly developing and, given recent research investments, some leading strategies may emerge in the next 5 years.

#### Concluding remarks and future perspectives

Based on the market sizes and target molecules, biomanufacturing seems poised to play a nonnegligible, but limited, role in capturing and sequestering carbon from the atmosphere. This role may expand if CO<sub>2</sub> capture is integrated with new biomanufacturing facilities on a large scale (see Outstanding questions) or if biomanufactured materials gain substantial uptake in markets for building materials. Readers should be wary of claims of net carbon-negative production if the final product is not a stable material that will be sequestered in a long-lived application or stored underground, as these values are often the result of incomplete system boundaries in life-cycle assessments rather than true net carbon removal. Solvents, for example, will not offer longterm, stable carbon storage. In terms of final product options, polymers present the largest near-term carbon sequestration opportunity, although this will require continued landfilling of large quantities of waste in exchange for modest climate benefits.

Rather than focusing on the narrow goal of achieving net carbon negativity, scientists, industry leaders, and policy makers should think more broadly about how biotechnology can enable a more circular and sustainable carbon economy. In the near term, GHG mitigation strategies are equally impactful when compared with net-negative options. Additionally, the benefits of reducing reliance on petrochemicals extend well beyond the climate to ecological, human health, and geopolitical impacts. Refocusing biomanufacturing on achieving emissions reductions, creating a circular carbon economy, and providing a wide array of performance-advantaged products will ensure that biotechnology has a leading role to play in a more sustainable future.

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#### **Declaration of interests**

None are declared.

#### Resources

- www.marketresearchfuture.com/reports/ethylene-market-931
- <sup>ii</sup>https://ethanolrfa.org/markets-and-statistics/annual-ethanol-production
- www.nexanteca.com/blog/202009/global-styrene-market-snapshot
- www.chemanalyst.com/industry-report/mono-ethylene-glycol-meg-market-646
- <sup>v</sup>www.expertmarketresearch.com/reports/propylene-glycol-market
- <sup>vi</sup>www.prnewswire.com/news-releases/global-acetone-market-to-reach-7-3-million-tons-by-2026–301355631.html
- viiwww.chemintel360.com/report-details.php?id=26
- viiiwww.chemintel360.com/reportdetails/Global-N-Butanol-Market/597
- <sup>ix</sup>www.expertmarketresearch.com/reports/acrylic-acid-market

#### Outstanding questions

What is the total CO<sub>2</sub> removal potential if all or most new biomanufacturing is coupled with carbon capture and sequestration (or utilization)?

How significant is the potential for biological production of major polymer precursors, including ethylene?

What are the climate implications of performance-advantaged biomanufactured materials based on differences in properties and usephase impacts?

Can products that are currently oxidized at their end-of-life, such as solvents, be converted to a safe, stable form that enables carbon sequestration?

What role will biomanufacturing play in carbon-sequestering building materials and how large is the potential carbon sink?



<sup>x</sup>www.bccresearch.com/market-research/chemicals/adipic-acid.html

xiwww.chemanalyst.com/industry-report/butanediol-market-657

xiiwww.expertmarketresearch.com/pressrelease/global-isopropyl-alcohol-market

xiiiwww.chemintel360.com/reportdetails/Lactic-Acid/227

- xivwww.ieabioenergy.com/wp-content/uploads/2020/02/Bio-based-chemicals-a-2020-update-final-200213.pdf
- xvwww.grandviewresearch.com/press-release/global-1-3-propanediol-market
- <sup>xvi</sup>www.grandviewresearch.com/industry-analysis/succinic-acid-market

<sup>xvii</sup>www.mordorintelligence.com/industry-reports/succinic-acid-market

xviiiwww.prnewswire.com/news-releases/increasing-penetration-from-cosmetics-flavors-and-fragrances-to-drivefarnesene-market-global-market-insights-inc-578765151.html

xixwww.epa.gov/facts-and-figures-about-materials-waste-and-recycling/advancing-sustainable-materials-management

<sup>xx</sup>www.oecd-ilibrary.org/environment/data/global-plastic-outlook\_c0821f81-en

xxiwww.arpa-e.energy.gov/technologies/programs/hestia

#### References

- Galán-Martín, Á. et al. (2021) Sustainability footprints of a renewable carbon transition for the petrochemical sector within planetary boundaries. One Earth 4, 565–583
- Liadze, I. et al. (2022) The Economic Costs of the Russia-Ukraine Conflict, National Institute of Economic and Social Research
- Larson, E. et al. (2021) Net-Zero America: Potential Pathways, Infrastructure, and Impacts, Princeton University
- Wang, Y. et al. (2022) A carbon-negative route for sustainable production of aromatics from biomass-derived aqueous oxygenates. *Appl. Catal. B* 307, 121139
- Vögeli, B. *et al.* (2022) Cell-free prototyping enables implementation of optimized reverse β-oxidation pathways in heterotrophic and autotrophic bacteria. *Nat. Commun.* 13, 3058
- Liew, F.E. *et al.* (2022) Carbon-negative production of acetone and isopropanol by gas fermentation at industrial pilot scale. *Nat. Biotechnol.* 40, 335–344
- Corona, A. *et al.* (2018) Life cycle assessment of adipic acid production from lignin. *Green Chem.* 20, 3857–3866
- Zhang, Y.-H.P. et al. (2017) Biomanufacturing: history and perspective. J. Ind. Microbiol. Biotechnol. 44, 773–784
- Jullesson, D. et al. (2015) Impact of synthetic biology and metabolic engineering on industrial production of fine chemicals. *Biotechnol. Adv.* 33, 1395–1402
- Scown, C.D. and Keasling, J.D. (2022) Sustainable manufacturing with synthetic biology. *Nat. Biotechnol.* 40, 304–307
- Rodríguez, Y. et al. (2020) Biogas valorization via continuous polyhydroxybutyrate production by Methylocystis hirsuta in a bubble column bioreactor. Waste Manag. 113, 395–403
- Köpke, M. and Simpson, S.D. (2020) Pollution to products: recycling of "above ground" carbon by gas fermentation. *Curr. Opin. Biotechnol.* 65, 180–189
- Yang, M. *et al.* (2020) Accumulation of high-value bioproducts in planta can improve the economics of advanced biofuels. *Proc. Natl. Acad. Sci. U. S. A.* 117, 8639–8648
- Yang, M. et al. (2022) Comparing in planta accumulation with microbial routes to set targets for a cost-competitive bioeconomy. Proc. Natl. Acad. Sci. U. S. A. 119, e2122309119
- Adom, F. et al. (2014) Life-cycle fossil energy consumption and greenhouse gas emissions of bioderived chemicals and their conventional counterparts. *Environ. Sci. Technol.* 48, 14624–14631
- Montazeri, M. *et al.* (2016) Meta-analysis of life cycle energy and greenhouse gas emissions for priority biobased chemicals. ACS Sustain. Chem. Eng. 4, 6443–6454
- Shaji, A. et al. (2021) Economic and environmental assessment of succinic acid production from sugarcane bagasse. ACS Sustain. Chem. Eng. 9, 12738–12746
- Semba, T. et al. (2018) Greenhouse gas emissions of 100% bioderived polyethylene terephthalate on its life cycle compared with petroleum-derived polyethylene terephthalate. J. Clean. Prod. 195, 932–938
- Tanzer, S.E. and Ramírez, A. (2019) When are negative emissions negative emissions? *Energy Environ. Sci.* 12, 1210–1218

- Fuss, S. et al. (2018) Negative emissions—Part 2: costs, potentials and side effects. Environ. Res. Lett. 13, 063002
- Goll, D.S. *et al.* (2021) Potential CO<sub>2</sub> removal from enhanced weathering by ecosystem responses to powdered rock. *Nat. Geosci.* 14, 545–549
- Yang, Y. et al. (2019) Soil carbon sequestration accelerated by restoration of grassland biodiversity. Nat. Commun. 10, 718
- Chou, A. et al. (2021) An orthogonal metabolic framework for one-carbon utilization. Nat. Metab. 3, 1385–1399
- Yang, M. et al. (2020) Cost and life-cycle greenhouse gas implications of integrating biogas upgrading and carbon capture technologies in cellulosic biorefineries. *Environ. Sci. Technol.* 54, 12810–12819
- 25. Humbird, D. (2021) Scale-up economics for cultured meat. Biotechnol. Bioeng. 118, 3239–3250
- Smith, P.B. (2015) Bio-based sources for terephthalic acid. In Green Polymer Chemistry: Biobased Materials and Biocatalysis (Cheng, H.N. et al., eds), pp. 453–469, American Chemical Society
- Lee, S.Y. et al. (2019) A comprehensive metabolic map for production of bio-based chemicals. Nat. Catal. 2, 18–33
- He, Y. et al. (2021) Metabolic engineering of Zymomonas mobilis for ethylene production from straw hydrolysate. Appl. Microbiol. Biotechnol. 105, 1709–1720
- Copeland, R.A. *et al.* (2021) Hybrid radical-polar pathway for excision of ethylene from 2-oxoglutarate by an iron oxygenase. *Science* 373, 1489–1493
- Wang, B. et al. (2021) A guanidine-degrading enzyme controls genomic stability of ethylene-producing cyanobacteria. Nat. Commun. 12, 5150
- Ding, Y. *et al.* (2019) Nanorg microbial factories: light-driven renewable biochemical synthesis using quantum dot-bacteria nanobiohybrids. *J. Am. Chem. Soc.* 141, 10272–10282
- Hoang Nguyen Tran, P. et al. (2020) Improved simultaneous cofermentation of glucose and xylose by Saccharomyces cerevisiae for efficient lignocellulosic biorefinery. Biotechnol. Biofuels 13, 12
- Huang, S. et al. (2019) Enhanced ethanol production from industrial lignocellulose hydrolysates by a hydrolysate-cofermenting Saccharomyces cerevisiae strain. Bioprocess Biosyst. Eng. 42, 883–896
- Lee, K. et al. (2019) Enhanced production of styrene by engineered Escherichia coli and in situ product recovery (ISPR) with an organic solvent. Microb. Cell Factories 18, 79
- Liu, C. et al. (2018) A systematic optimization of styrene biosynthesis in Escherichia coli BL21(DE3). Biotechnol. Biofuels 11, 14
- Grubbe, W.S. et al. (2020) Cell-free styrene biosynthesis at high titers. Metab. Eng. 61, 89–95
- Liang, L. et al. (2020) Genome engineering of E. coli for improved styrene production. Metab. Eng. 57, 74–84
- Islam, M.A. *et al.* (2017) Exploring biochemical pathways for mono-ethylene glycol (MEG) synthesis from synthesis gas. *Metab. Eng.* 41, 173–181

 Uranukul, B. et al. (2019) Biosynthesis of monoethylene glycol in Saccharomyces cerevisiae utilizing native glycolytic enzymes. Metab. Eng. 51, 20–31

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- Chae, T.U. et al. (2018) Production of ethylene glycol from xylose by metabolically engineered Escherichia coli. AICHE J. 64, 4193–4200
- Veeravalli, S.S. and Mathews, A.P. (2018) Exploitation of acidtolerant microbial species for the utilization of low-cost whey in the production of acetic acid and propylene glycol. *Appl. Microbiol. Biotechnol.* 102, 8023–8033
- Veeravalli, S.S. and Mathews, A.P. (2019) A novel low pH fermentation process for the production of acetate and propylene glycol from carbohydrate wastes. *Enzym. Microb. Technol.* 120, 8–15
- Capilla, M. et al. (2021) The combined effect on initial glucose concentration and pH control strategies for acetone-butanolethanol (ABE) fermentation by *Clostridium acetobutylicum* DSM 792. Biochem. Eng. J. 167, 107910
- Sherkhanov, S. et al. (2020) Isobutanol production freed from biological limits using synthetic biochemistry. Nat. Commun. 11, 4292
- Noda, S. *et al.* (2019) Reconstruction of metabolic pathway for isobutanol production in *Escherichia coli. Microb. Cell Factories* 18, 124
- Promdonkoy, P. et al. (2020) Improvement in D-xylose utilization and isobutanol production in S. cerevisiae by adaptive laboratory evolution and rational engineering. J. Ind. Microbiol. Biotechnol. 47, 497–510
- Song, H.-S. et al. (2018) Enhanced isobutanol production from acetate by combinatorial overexpression of acetyl-CoA synthetase and anaplerotic enzymes in engineered Escherichia coli. Biotechnol. Bioeng. 115, 1971–1978
- Ankenbauer, A. et al. (2021) Micro-aerobic production of isobutanol with engineered Pseudomonas putida. Eng. Life Sci. 21, 475–488
- Ko, Y.-S. *et al.* (2020) A novel biosynthetic pathway for the production of acrylic acid through β-alanine route in *Escherichia coli. ACS Synth. Biol.* 9, 1150–1159
- Oliveira, A. et al. (2021) A kinetic model of the central carbon metabolism for acrylic acid production in *Escherichia coli*. PLoS Comput. Biol. 17, e1008704
- Shin, J.H. et al. (2021) Exploring functionality of the reverse βoxidation pathway in Corynebacterium glutamicum for production of adipic acid. Microb. Cell Factories 20, 155
- Raj, K. et al. (2018) Biocatalytic production of adipic acid from glucose using engineered Saccharomyces cerevisiae. Metab. Eng. Commun. 6, 28–32
- Hao, T. et al. (2021) Engineering the reductive TCA pathway to dynamically regulate the biosynthesis of adipic acid in Escherichia coli. ACS Synth. Biol. 10, 632–639
- Kruyer, N.S. et al. (2020) Fully biological production of adipic acid analogs from branched catechols. Sci. Rep. 10, 13367
- Silva, R.G.C. *et al.* (2020) Identification of potential technologies for 1,4-butanediol production using prospecting methodology. *J. Chem. Technol. Biotechnol.* 95, 3057–3070
- Pooth, V. *et al.* (2020) Comprehensive analysis of metabolic sensitivity of 1,4-butanediol producing Escherichia coli toward substrate and oxygen availability. *Biotechnol. Prog.* 36, e2917
- Wang, J. *et al.* (2020) Bacterial synthesis of C3-C5 diols via extending amino acid catabolism. *Proc. Natl. Acad. Sci. U. S. A.* 117, 19159–19167
- Wang, C. *et al.* (2018) Enhanced isopropanol-butanol-ethanol mixture production through manipulation of intracellular NAD(P)H level in the recombinant *Clostridium acetobutylicum* XY16. *Biotechnol. Biofuels* 11, 12
- Mitsui, R. et al. (2020) Construction of lactic acid-tolerant Saccharomyces cerevisiae by using CRISPR-Cas-mediated genome evolution for efficient D-lactic acid production. Appl. Microbiol. Biotechnol. 104, 9147–9158

- Grewal, J. and Khare, S.K. (2018) One-pot bioprocess for lactic acid production from lignocellulosic agro-wastes by using ionic liquid stable *Lactobacillus brevis*. *Bioresour. Technol.* 251, 268–273
- 61. Yamada, R. et al. (2019) Toward the construction of a technology platform for chemicals production from methanol: D-lactic acid production from methanol by an engineered yeast *Pichia* pastoris. World J. Microbiol. Biotechnol. 35, 37
- Li, G. et al. (2020) Advances in microbial production of mediumchain dicarboxylic acids for nylon materials. *React. Chem. Eng.* 5, 221–238
- Jeon, W.-Y. et al. (2019) Microbial production of sebacic acid from a renewable source: production, purification, and polymerization. Green Chem. 21, 6491–6501
- 64. Yun, J. et al. (2021) Co-fermentation of glycerol and glucose by a co-culture system of engineered *Escherichia coli* strains for 1,3propanediol production without vitamin B12 supplementation. *Bioresour. Technol.* 319, 124218
- Lee, J.H. et al. (2018) Production of 1,3-propanediol from glucose by recombinant *Escherichia coli* BL21(DE3). *Biotechnol. Bioprocess Eng.* 23, 250–258
- 66. de Santana, J.S. et al. (2021) Production of 1,3-propanediol by Lactobacillus diolivorans from agro-industrial residues and cactus cladode acid hydrolyzate. *Appl. Biochem. Biotechnol.* 193, 1585–1601
- Fokum, E. et al. (2021) Co-fermentation of glycerol and sugars by *Clostridium beijerinckii*: enhancing the biosynthesis of 1,3propanediol. *Food Biosci.* 41, 101028
- Bao, W. et al. (2020) Regulation of pyruvate formate lyasedeficient Klebsiella pneumoniae for efficient 1,3-propanediol bioproduction. Curr. Microbiol. 77, 55–61
- Cui, Z. et al. (2017) Engineering of unconventional yeast Yarrowia lipolytica for efficient succinic acid production from glycerol at low pH. Metab. Eng. 42, 126–133
- Ahn, J.H. et al. (2020) Enhanced succinic acid production by Mannheimia employing optimal malate dehydrogenase. Nat. Commun. 11, 1970
- Ferone, M. et al. (2019) Continuous succinic acid fermentation by actinobacillus succinogenes: assessment of growth and succinic acid production kinetics. *Appl. Biochem. Biotechnol.* 187, 782–799
- Liu, Y. et al. (2019) Engineering the oleaginous yeast Yarrowia lipolytica for production of α-farnesene. Biotechnol. Biofuels 12, 296
- Liu, H. et al. (2021) Dual regulation of cytoplasm and peroxisomes for improved A-farnesene production in recombinant Pichia pastoris. ACS Synth. Biol. 10, 1563–1573
- Chae, T.U. et al. (2020) Metabolic engineering for the production of dicarboxylic acids and diamines. *Metab. Eng.* 58, 2–16
- Wu, L. et al. (2018) A hybrid biological-chemical approach offers flexibility and reduces the carbon footprint of bio-based plastics, rubbers, and fuels. ACS Sustain. Chem. Eng. 6, 14523–14532
- Wurm, F.R. *et al.* (2020) Plastics and the environment-current status and challenges in Germany and Australia. *Macromol. Rapid Commun.* 41, e2000351
- Bandini, F. et al. (2020) Fate of biodegradable polymers under industrial conditions for anaerobic digestion and aerobic composting of food waste. J. Polym. Environ. 28, 2539–2550
- Liu, Z. et al. (2022) Monitoring global carbon emissions in 2021. Nat. Rev. Earth Environ. 3, 217–219
- Meys, R. et al. (2021) Achieving net-zero greenhouse gas emission plastics by a circular carbon economy. Science 374, 71–76
- Vethaak, A.D. and Legler, J. (2021) Microplastics and human health. Science 371, 672–674
- 81. U.S. EPA (2021) Inventory of U.S. Greenhouse Gas Emissions and Sinks: 1990–2019, U.S. Environmental Protection Agency
- 82. Churkina, G. et al. (2020) Buildings as a global carbon sink. Nat. Sustain. 3, 269–276