The Portevin-Le Chatelier (PLC) effect is one of the best studied plastic instabilities in metallic materials. One of its main manifestations, in the appropriate strain rate and temperature range, is the appearance of “serrated flow” in the stress–strain curve. The PLC effect has been investigated extensively, from both experimental and theoretical points of view, in body and face centered cubic materials. However, the study of this phenomenon in hexagonal close-packed (hcp) materials, particularly in binary Mg-Al alloys, has been limited.

Chaturvedi et al. [1] reported serrated flow in a binary Mg-Ag alloy in the 53–124 °C temperature range, and attributed the locking of mobile dislocations to the diffusion of silver atoms during deformation. Serrated flow was also observed by Gao et al. [2] in binary Mg-Y alloys during tensile deformation between 150 and 250 °C. In contrast, Caceres et al. [3] investigated the serrated behavior of as-cast binary Mg-Al alloys and did not observe serrations. Interestingly, Standford et al. [4] reported flow serrations during deformation of a Mg-Gd alloy at 200 °C. However, they could not observe serrations in a Mg-Al alloy processed under the same conditions to obtain a similar extrusion texture and grain size.

The PLC effect has also been found in ternary Mg-alloys, such as AZ91 by Corby et al. [5], and in alloys containing rare earth additions such as Mg-Y-Nd by Zhu et al. [6], Mg-Gd-Zn alloys by Wu et al. [7] and Mg-Mn-Nd by Dudamell et al. [8]. In all these systems, serrated flow was attributed to the dynamic interaction between solute atoms and dislocations, a phenomenon known as dynamic strain aging (DSA), first proposed by Cottrell in 1953 [9]. In the case of AZ91 at room temperature, Corby et al. [5] reported that both Al and Zn atoms were involved in the DSA. They suggested that Al might be the dominant solute, while Zn acted as a catalyst enabling the formation of a forest of dislocations through which Al atoms might pipe diffuse.

This study reveals the appearance of serrations at room temperature in a nanostructured binary Mg-10Al (wt.%) alloy processed via cryomilling and spark plasma sintering (SPS). Our previous work has demonstrated that cryomilling (mechanical milling in liquid nitrogen atmosphere) is an optimum severe plastic deformation technique to obtain nanocrystalline-sized grains in binary Mg-Al alloys [10,11]. As a consolidation technique, the SPS method allows cryomilled Mg-powders to be synthesized in a very short time into a bulk nanostructured material with a characteristic bimodal grain distribution [10–12]. This feature results in a simultaneous enhancement of the strength and ductility [13–15]. Coarse grains facilitate dislocation mobility and contribute to the ductility, while nano-sized grains contribute to the strengthening by the Hall-Petch mechanism. Additionally, we have demonstrated that small amounts of impurity elements may be introduced during cryomilling in liquid nitrogen and form nanoscale dispersoids such as oxides and nitrides. Indeed, thorough TEM analyses of the cryomilled powders have revealed some traces of nitrogen and oxygen atoms [10]. The presence of second phases, however, is known to play a significant role in stabilizing the microstructure after heat treatment [16].

Although to the best of our knowledge serrations have not been reported for binary Mg-Al alloys [3,4], if serrations are indeed associated with a solute atom-dislocation interactions, they should logically also occur in our nanostructured Mg-10Al alloy, as all the ingredients for the PLC effect to occur are also present by design. The idea that impurity atoms introduced during cryomilling might promote DSA during mechanical testing was also hypothesized for a nanostructured Al alloy processed by cryomilling [17]. This work is motivated by this hypothesis, and is focused on in-situ nanomechanical testing to understand the
underlying mechanisms leading to serrated flow in binary nanostructured Mg-Al alloys.

Pure Mg and Al powders were blended to formulate the final composition Mg-10Al (wt.%) alloy. Cryomilling was performed for 8 h in liquid nitrogen atmosphere. Spark-plasma-sintering (SPS) process was carried out in vacuum for 5 min at 400 °C and under uniaxial pressure of 100 MPa to consolidate the cryomilled powders. More details about the material processing can be found in refs. [10,11].

In order to evaluate the mechanical behavior of nanostructured Mg-10Al alloys under different strain rate conditions, in-situ SEM microcompression experiments were conducted at room temperature under displacement control mode and at a strain rates between $5 \times 10^{-4}$ and $10^{-1} \text{s}^{-1}$. A PI 85 SEM PicoIndenter (Hysitron Inc.) with a 5-μm flat punch diamond probe inside a FEI Nova 600 Nanolab Dual-Beam Focused Ion Beam-Scanning Electron Microscope (FIB-SEM) was used for these experiments. Micropillars of 4-μm in diameter and an aspect ratio of 1:3 were machined from the sintered samples by FIB operated at 30 kV. A series of concentric annular milling patterns with different currents were applied. In order to tailor the pillars into the desired shape and minimize the tapering [18,19], a low beam current (0.3 nA) was used as final milling step.

Fig. 1 shows the results from in-situ SEM microcompression tests at room temperature of a nanostructured binary Mg-10Al alloy. The plot in Fig. 1a shows representative engineering stress-strain curves at different strain rates for Mg-10Al micropillars of 4 μm in diameter. A minimum of three tests per strain rate were performed but only one curve is shown here for clarity. We find that serrations start to appear at a strain rate of $2 \times 10^{-3} \text{s}^{-1}$ with a remarkable serrated effect occurring at $2 \times 10^{-2} \text{s}^{-1}$ (red curve). These results can be better seen in a schematic representation of strain rate as a function of the inverse of temperature (Fig. 1b), where the two straight lines mark the boundaries of the region where serrations occur. All strain rates shown in Fig. 1a have been highlighted with the same color code at room temperature in this figure. Serrations are classified as of type A, B, or C, depending on applied load, strain rate, and temperature [20,21]. Type B serrations that are characterized by regular oscillations around the general level of the stress-strain curve are observed at all strain rates. A transition from type B to type C, which is characterized by abrupt load drops below the general level of the stress-strain curve can be also observed at $10^{-1} \text{s}^{-1}$ of strain rate. (See Supplementary Fig. 1 for a better observation of the serrations).

Fig. 1c displays the average yield stress as a function of strain rate with the standard deviation of the average as error bars. It is worth noting the abnormal strain rate sensitivity (SRS) at room temperature, which is positive for lower strain rates and negative for strain rates higher than $2 \times 10^{-2} \text{s}^{-1}$. This has also been reported for a ternary Mg-Li-Al alloy [22] and rationalized in terms of the competition between two mechanisms: shearing of precipitates by dislocations (leading to a positive SRS) at lower strain rates [23,24]; and DSA (resulting in a negative SRS) at higher strain rates [25].

The negative SRS at room temperature caused by DSA is considered as a sufficient condition for serrated flow in different material systems [5,9,22,25–27]. The other mechanism of shearing of precipitates by dislocations that was firstly reported by Zambo et al. [28], Kumar et al. [29], Tian et al. [23], Wu et al. [24], and recently by Choudhuri et al. [30] tends to be suitable for alloys with a large number of fine precipitates in the matrix. As the deformation progresses, substantial pileups of dislocations may shear the fine precipitates, and then the stress concentration can be relaxed temporarily. Thus, repeated shearing can lead to successive stress drops resulting in stress–strain curves with a jerky appearance. From our previous work [11], we have evidence that γ-Al$_{3}$Mg$_{17}$ precipitates are primarily obstacles for dislocation movement contributing to the enhancement of strength in a Mg-10Al alloy. Therefore, the presumption that these precipitates act as shearable objects contributing to the PLC effect by repeated shearing [9,23–24,28–30] is not substantiated by strength measurements.

Additionally, it is interesting to point out a possible relationship between the sign of the SRS slope and the grain size. A high SRS has been reported to be related to the nanocrystalline grain size [31,32]. For instance, Schwaiger et al. [31] found that nanocrystalline pure Ni exhibited a positive SRS in flow stress, an effect that was not found in ultrafine and microcrystalline Ni. Fan et al. [33] and Joshi et al. [34] observed the PLC effect in ultrafine-grained Al-alloys only over a small strain range immediately following yield, while the coarse-grained alloy exhibited serrated flow over nearly the entire plastic strain range for the same applied displacement rate. Del Valle and Ruano [35] reported that SRS strongly increases by decreasing grain size below 15 μm in a Mg–Al–Zn alloy at moderate temperatures.

In view of the bimodal grain size distribution displayed by our specimens after SPS [11], we speculate that serrated flow is governed by the competition between positive SRS contributed by nanocrystalline grains (~35 nm) at low strain rates (caused by GB strengthening mechanisms) and negative SRS originating in coarse grains (~400 nm) from DSA. Although this explanation is somewhat speculative, our argument is that true DSA can only fully develop in coarse grained systems. In very fine grained materials, dislocation/grain boundary interactions and other grain boundary effects are likely to govern plastic flow, not leaving enough room for solute diffusion and/or dislocation glide to develop to the extent required by DSA.

Fig. 2 shows as an example two characteristic nanostructured Mg-10Al micropillars before and after microcompression at different strain rates. Left images in Fig. 2a and b show 52° tilted-view SEM images of those micropillars before microcompression. All micropillars are slightly tapered as a result of the annular cutting method by FIB. However, we...
Deformation bands are apparent on the micropillars’ surface, especially and show 52° tilted-view SEM images of the compressed micropillars. The images on the right hand side of Fig. 2a and b show 52° tilted-view SEM images of the compressed micropillars. Deformation bands are apparent on the micropillars’ surface, especially after increasing the strain rate up to $2 \times 10^{-2} \text{s}^{-1}$. This observation suggests a direct correlation between the DSA and the macroscopic response of the micropillar.

A detailed analysis of the serrations at $2 \times 10^{-2} \text{s}^{-1}$ of strain rate is shown in the Supplementary Video 1 and Fig. 3. From the video, we can correlate SEM images of the micropillar under compression with its loading-displacement curve. In addition to the deformation bands on the micropillar surface corresponding to the serrated flow in the loading-displacement curve, a thickening of the micropillar diameter is observed during compressive loading. Specifically, an increase of $\sim 18\%$ is measured for the diameter at half-height of the micropillar after the compression test is completed.

Fig. 3a shows the engineering stress-strain curve (red) along with the true stress-strain data by interpolation (grey curve). As expected from the increase of pillar diameter during microcompression, the apparent work-hardening effect from the engineering stress-strain curve is no longer visible in the true stress-strain curve. Typical type B serrations are shown in Fig. 3a. Our results suggest that this type of serrations appear when a critical stress is applied and begins with a pinning process. That means that dislocations accumulate some degree of slip prior to interacting with solute atoms, which then results in the first stress drop that gives rise to a serration. From the stress-strain curve, the critical stress and strain to induce serrated flow are around 600 MPa and 3.6%, respectively. This type B serration can be better seen in Fig. 3b, where the two main processes, i.e., pinning and unpinning of dislocations by solute atoms are clearly identified. We assume that at this particular strain rate, the diffusivity of solute atoms might become similar to the dislocation mobility, leading first to dislocation pinning and stress buildup, which is in turn followed by dislocation unpinning and rapid stress drop. The stress drop, $\Delta \sigma$, and strain increment, $\Delta \varepsilon$, are represented as a function of the number of drops in Fig. 4. Interestingly, $\Delta \sigma$ follows a linear relationship with the number of drops (Fig. 4a) while $\Delta \varepsilon$ is nearly constant at approximately $0.9\%$ (Fig. 4b). A linear relationship is also found between $\Delta \sigma / \Delta \varepsilon_{\text{max}}$ and the strain $\varepsilon$ (Fig. 4c), which indicates that the linearity between $\Delta \sigma$ and the number of drops is not simply an artifact of the engineering stress-strain curve. Since the stress drop can be related to the size of plastic events [20], the linear relationship of $\Delta \sigma$ with the number of drops suggests that the shear deformation of the dislocation bands linearly increases with the number of serrations.

Next we provide some micromechanical analysis of our results. DSA is brought about by dislocation solute-interactions, the nature of which depends critically on the type of solute (and dislocation). To begin our analysis, we hypothesize that the solutes present in our system in addition to substitutional Al atoms are interstitial O and N atoms introduced during cryomilling. To test this hypothesis and ascertain which of these atomic species – substitutional solutes or interstitial impurities – is responsible for DSA, we calculate the characteristic lengths of diffusion and slip, respectively, of atoms and dislocations in our deformation tests. Unfortunately, there are no reliable data for diffusion of oxygen 

![Fig. 2. SEM images of Mg-10Al micropillars before (left) and after (right) microcompression under strain rates of $5 \times 10^{-4} \text{s}^{-1}$ (a) and $2 \times 10^{-2} \text{s}^{-1}$ (b). Scale bar is 4 $\mu$m in all images.](image1)

![Fig. 3. (a) Engineering and true stress-strain curves at $2 \times 10^{-2} \text{s}^{-1}$ of strain rate. (b) Zoom in the periodic region enclosed by a rectangle in (a) showing type B serrations. The stress drop, $\Delta \sigma$, and the strain increment, $\Delta \varepsilon$, along with the pinning and unpinning regions are highlighted in this plot.](image2)
and/or nitrogen in Mg or Mg-Al alloys in the literature. Thus, we carry out the analysis for Al and test whether it can be eliminated as the solute responsible for DSA.

The characteristic slip distance travelled by a dislocation during one of the teeth of the stress–strain curve serrations can be obtained from Orowan’s equation [38]:

\[ l_s = \frac{k}{b \rho v} \]

where \( \dot{\varepsilon} \) is the strain rate, \( b \) is the Burgers vector and \( v \) is the dislocation velocity. From Eq. (1):

\[ l_d = \frac{\dot{\varepsilon} \delta t}{b \rho} = \frac{\Delta \varepsilon}{\dot{\varepsilon}} \]

with \( \delta t \) being the time duration of one of the serrations, estimated as \( \delta t = \Delta \varepsilon/\dot{\varepsilon} \). For a strain rate of \( 2 \times 10^{-2} - 1 \) and with \( \Delta \varepsilon = 9 \times 10^{-3} \) from Fig. 4, we have \( \delta t = 0.45 \) s. Next, with a value of \( b = 3.21 \times 10^{-10} \) m for basal dislocations and an upper bound for the dislocation density of \( \rho \approx 10^{13} \) m\(^{-2} \), we obtain \( l_d \approx 3 \times 10^{-5} \) m.

For its part, the characteristic diffusion length of Al solute atoms diffusing in Mg, \( l_c \), can be calculated as:

\[ l_c \approx \sqrt{D_o \exp \left( \frac{E_a}{R T} \right) \delta t} \]

where \( D \) is the diffusion coefficient, \( D_o \) is the diffusivity pre-factor, \( R \) is Boltzmann’s constant, \( T \) is the absolute temperature, and \( E_a \) is the activation energy for diffusion. Using values reported in the literature of \( D_o = 3.9 \times 10^{-10} \) m\(^2\) s\(^{-1} \) and \( E_a = 155 \) kJ/mol [39], we get \( l_c \approx 10^{-15} \) m at room temperature and under the same strain rate condition.

Given the six order of magnitude discrepancy found between the estimated values of \( l_d \) and \( l_c \) for Al atoms, we conclude that the diffusion of Al atoms in Mg at a strain rate of \( \dot{\varepsilon} = 2 \times 10^{-2} \) s\(^{-1} \) and room temperature cannot explain the observed PLC effect in our experiments. Therefore, by elimination, we attribute the DSA mechanism in nanostructured binary Mg-Al alloys to interstitial O and/or N impurities (introduced during cryomilling) and their interaction with dislocations.

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Fig. 4. Plots showing the values of the stress drop, \( \Delta \sigma \), and the strain increment, \( \Delta \varepsilon \), as a function of the number of drops: (a) and (b), and \( \Delta \sigma/\Delta \varepsilon \) as a function of strain, \( \varepsilon \), (c) from the data shown in Fig. 3.