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1  
2 **Unravelling Biogeochemical Drivers of Methylmercury Production**  
3 **in an Arctic Fen Soil and a Bog Soil**  
4

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25

26 **Abstract**

27 Arctic tundra soils store a globally significant amount of mercury (Hg), which could be  
28 transformed to the neurotoxic methylmercury (MeHg) upon warming and thus poses serious  
29 threats to the Arctic ecosystem. However, our knowledge of the biogeochemical drivers of MeHg  
30 production is limited in these soils. Using substrate addition (acetate and sulfate) and selective  
31 microbial inhibition approaches, we investigated the geochemical drivers and dominant microbial  
32 methylators in 60-day microcosm incubations with two Arctic tundra soils: a circumneutral fen  
33 soil and an acidic bog soil, collected near Nome, Alaska, United States. Results showed that  
34 increasing acetate concentration had negligible influences on MeHg production in both soils.  
35 However, inhibition of sulfate-reducing bacteria (SRB) completely stalled MeHg production in  
36 the fen soil in the first 15 days, whereas addition of sulfate in the low-sulfate bog soil increased  
37 MeHg production by 5-fold, suggesting prominent roles of SRB in Hg(II) methylation. Without  
38 the addition of sulfate in the bog soil or when sulfate was depleted in the fen soil (after 15 days),  
39 both SRB and methanogens contributed to MeHg production. Analysis of microbial community  
40 composition confirmed the presence of several phyla known to harbor microorganisms associated  
41 with Hg(II) methylation in the soils. These observations suggest that SRB and methanogens were  
42 mainly responsible for Hg(II) methylation in these tundra soils, although their relative  
43 contributions depended on the availability of sulfate and possibly syntrophic metabolisms between  
44 SRB and methanogens.

45

46 **Key words:** Mercury, methylation, acetate, sulfate, microbial community, syntrophy

## 47 **1. Introduction**

48 Arctic tundra soil stores a globally significant amount of mercury (Hg) due to atmospheric  
49 deposition and a high content of soil organic matter that has a high binding affinity for Hg (Obrist  
50 et al., 2017; Schuster et al., 2018). The greatest toxicological concern of Hg pollution is the  
51 transformation of inorganic mercuric Hg(II) to methylmercury (MeHg), which is a potent  
52 neurotoxin that bioaccumulates and biomagnifies through food webs in the Arctic ecosystem.  
53 Production of MeHg has been reported in Arctic wetland ponds, lake sediments, and tundra soils  
54 (Barkay and Poulain, 2007; Lehnherr et al., 2012; Schartup et al., 2015; Yang et al., 2016a), which  
55 are important sources of MeHg in Arctic freshwater and the Arctic Ocean (Kirk et al., 2012;  
56 Lehnherr, 2014; Schartup et al., 2015). However, our current understanding of the biogeochemical  
57 drivers of Hg(II) methylation is limited in tundra soils. Recent studies reported that warming  
58 increased MeHg production in these soils, possibly through enhanced degradation of soil organic  
59 carbon (SOC) that releases low molecular weight (LMW) labile organic carbon and nutrients (e.g.,  
60 acetate, phosphorus, and nitrogen) and fuels microbial production of MeHg (MacMillan et al.,  
61 2015; Yang et al., 2016a). LMW labile organic carbon compounds, such as acetate, are important  
62 substrates and can be consumed in anaerobic biogeochemical processes of methanogenesis, ferric  
63 iron [Fe(III)] reduction, and sulfate reduction. Net accumulation of acetate has been found in the  
64 top organic soil layer but net consumption was found in the lower mineral soil layer after  
65 incubation at 8 °C (Yang et al., 2016a). The LMW labile organic carbon could also be transported  
66 into a deeper mineral layer through vertical movement or mixing due to cryoturbation between the  
67 organic and mineral layer soils (Drake et al., 2015; Yang et al., 2016b). Although correlations  
68 between MeHg production and labile organic carbon mineralization have been suggested

69 (MacMillan et al., 2015; Yang et al., 2016a), the roles of LMW organic acids (e.g., acetate) in  
70 microbial Hg(II) methylation remain unclear in the Arctic tundra soils.

71         Moreover, it remains elusive about who are the dominant Hg(II) methylators in these Arctic  
72 soils. MeHg is produced by a small group of anaerobic microorganisms carrying a two-gene cluster  
73 *hgcAB* essential for Hg(II) methylation (Parks et al., 2013). These microbes may include, but are  
74 not limited to, sulfate-reducing bacteria (SRB), iron-reducing bacteria (FeRB), methanogens, and  
75 fermenters (Bravo et al., 2018; Gilmour et al., 2013; Jones et al., 2020; Liu et al., 2018; Yu et al.,  
76 2013), although their contributions to MeHg production vary with geochemical conditions in  
77 different environmental systems. Several studies reported that methanogens and FeRB were  
78 dominant members of Hg(II) methylating community in rice paddy soils and in boreal lakes (Bravo  
79 et al., 2018; Liu et al., 2018), while others found that SRB were the principal Hg(II) methylators  
80 in some freshwater sediments or boreal wetlands (Jones et al., 2020; Schaefer et al., 2020). In the  
81 high Arctic wetlands, one study suggested that SRB were not the major Hg(II) methylators due to  
82 a low prevalence of SRB (Loseto et al., 2004), while others proposed that SRB could be active in  
83 these soils (Barkay and Poulain, 2007). A laboratory incubation study of Arctic tundra soils found  
84 no significant correlations between Hg(II) methylation and sulfate reduction because of low sulfate  
85 concentrations (Yang et al., 2016a). However, sulfate concentrations can vary greatly among these  
86 soils through precipitation or seawater intrusion (Mitchell et al., 2008; Tully et al., 2019). High  
87 sulfate concentrations may stimulate the growth and activity of SRB and promote MeHg  
88 production (Gilmour et al., 1992; Jones et al., 2020; Mitchell et al., 2008; St. Pierre et al., 2014),  
89 but other studies reported that sulfate addition may not affect Hg(II) methylation as SRB and  
90 related syntrophs may instead metabolize alternative substrates via syntrophic fermentation of  
91 organic compounds with methanogens (Schaefer et al., 2020). While methanogens and FeRB play

92 important roles in coupled activities of SOC transformation, methanogenesis, and Fe(III) reduction  
93 (Woodcroft et al., 2018; Yang et al., 2017), their contributions to Hg(II) methylation in Arctic  
94 tundra soils are unclear. A better understanding is thus needed to determine who are the principal  
95 Hg(II) methylators in the Arctic tundra soils, and whether these methylators are site-specific and  
96 how they are related to geochemical characteristics of the soils.

97 In this study, we conducted laboratory microcosm incubation experiments to investigate  
98 microbial production of MeHg in two contrasting Arctic tundra soils: a circumneutral fen soil from  
99 the toe of a hillslope and an acidic bog soil, both collected near Nome, Alaska, United States. Our  
100 main objectives were to determine (1) how an increase in the availability of substrates, acetate and  
101 sulfate, may affect MeHg production, and (2) who are the dominant Hg(II) methylators. We  
102 hypothesize that suppression of SRB and methanogens by their respective inhibitors, molybdate  
103 and 2-bromoethanesulfonate, could slow down or cease MeHg production in the tundra soils. This  
104 hypothesis was tested by a series of incubation experiments in the presence or absence of various  
105 substrates and/or microbial inhibitors and by 16S rRNA gene sequence analyses to tease out major  
106 microbial compositions and changes during incubation.

107

## 108 **2. Materials and Methods**

### 109 *2.1. Study site and characterization of soil samples*

110 Active-layer soil cores were collected during an August 2018 field campaign from the  
111 Intensive Site 9 (fen) near Teller Road Mile 27 (64.729193°N, 165.944072°W) and from a  
112 thermokarst depression (bog) at the Council Road Mile 71 site (64.859618°N, 163.703477°W) in  
113 Alaska, United States, as part of the Next Generation Ecosystem Experiment (NGEE-Arctic)

114 (Philben et al., 2020). For the clarity of the presentation and comparisons, herein we refer the  
115 Teller soil as the “fen” soil, and the Council soil as the “bog” soil. The Teller site is characterized  
116 by a 130-m elevation gradient and a diverse vegetation cover, including tall shrubs, dwarf shrubs,  
117 mosses, and graminoids and a water table at or near the soil surface (Léger et al., 2019; Philben et  
118 al., 2020). The soil layer consists of an organic rich upper part and shows a gradual increase in  
119 bulk density with depth. The National Ocean and Atmospheric Administration’s (NOAA’s)  
120 meteorological station at Nome Airport indicates that over a 5-year average (2013 to 2017) the  
121 mean annual air temperature is  $-1.02$  °C, the yearly rain precipitation is 450.6 mm, and the yearly  
122 snowfall is 1704.8 mm (Léger et al., 2019). The Council site is composed of mostly moist acidic  
123 tussock tundra, and the dominant vegetation is cotton grass (*Eriophorum vaginatum*) or tussock,  
124 blueberry (*Vaccinium uliginosum*), and lichen and moss (*Sphagnum* spp.) beds. The mean annual  
125 air temperature is  $-2.1$  °C, and the annual precipitation is about 485 mm (Kim et al., 2016).  
126 Following the collection of soil cores (~5 cm in diameter and 20 cm in depth), they were sealed in  
127 Whirl-Pak bags, and stored at  $-22$  °C until use. Soil porewater was also collected from the  
128 subsurface of the sites by piezometers and MacroRhizons, and stored for analyses of pH and  
129 dissolved methane (CH<sub>4</sub>) (described in Section 2.3.).

130 Soil water contents (Table 1) were determined by oven-drying the pre-weighed wet soil at  
131 60 °C until a constant mass was obtained and then calculating the mass difference before and after  
132 drying normalized to the initial wet weight [i.e., (wet weight–dry weight)/(wet weight)×100%].  
133 Sequential extraction of the soils was conducted to determine extractable Hg pools using a  
134 modified method from previous studies (Bloom et al., 2003; Lu et al., 2019; Miller et al., 2013).  
135 The soils were extracted sequentially with (1) 0.01 M HCl and 0.1 M CH<sub>3</sub>COOH (F1), (2) 1 M  
136 KOH (F2), and (3) a mixture of concentrated HCl and HNO<sub>3</sub> (aqua regia) at a volumetric ratio of

137 10:3 (F3). These pools of Hg were operationally defined as acid-soluble Hg (F1), organo-chelated  
138 Hg (F2), and mercuric sulfide (HgS) Hg (F3) (Bloom et al., 2003; Lu et al., 2019; Miller et al.,  
139 2013). Briefly, a subsample of the homogenized wet soil (~0.5 g) was equilibrated with 20 mL of  
140 the extractant F1 on an end-to-end rotary shaker (120 rpm) for 18±4 h. The soil suspension was  
141 then centrifuged at 2000×g for 15 min, and the supernatant filtered through 0.2-µm filters to collect  
142 the filtrate. This step was repeated by mixing the soil pellet after centrifugation with another 20  
143 mL of the extractant F1, and the filtrate was again collected and combined with above filtrate. The  
144 remaining soil pellet was added with 20 mL of the extractant F2, and the same extraction procedure  
145 was used to extract organo-chelated Hg. Lastly, 13 mL aqua regia (F3) was added to the soil pellet  
146 after F2 extraction for overnight digestion, and then diluted with Milli-Q water to a final volume  
147 of 40 mL. Total Hg was extracted directly by digesting an aliquot of 0.5–1 g wet soil in 13 mL  
148 aqua regia overnight and then diluting it to 40 mL with Milli-Q water. The extractant solutions  
149 were finally oxidized with BrCl (0.5% v/v) and stored at 4 °C overnight or longer until analyses,  
150 as previously described (Miller et al., 2009; Zhang et al., 2021).

## 151 2.2. *Soil microcosm incubations*

152 Soil microcosm experiments were the same as those reported in a recent study focusing on  
153 greenhouse gas production and anaerobic respiration pathways of soil organic carbon (Philben et  
154 al., 2020), while the present study focused on Hg(II) transformation. Briefly, soil cores were  
155 thawed and homogenized inside an anoxic glove bag (Coy Lab Products, Grass Lake, MI) under  
156 ~98% N<sub>2</sub> and 2% H<sub>2</sub> atmosphere. The homogenized wet soil (3.0±0.1 g) from each site was  
157 transferred into a series of autoclaved serum bottles (26 mL) in the glove bag. Molybdate and 2-  
158 bromoethanesulfonate (BES) were used as selective inhibitors of SRB and methanogens,  
159 respectively, as previously described (Hamelin et al., 2011; Kronberg et al., 2018; Schaefer et al.,



160 2020). The soil was thoroughly mixed and then incubated following the addition of 0.2 mL each  
161 of the amendments: (1) degassed Milli-Q wssater (unamended control); (2) sodium acetate  
162 (acetate); (3) sodium acetate and BES (acetate+BES); (4) sodium acetate and sodium molybdate  
163 (acetate+Mo); and (5)  $^{202}\text{Hg}$ -enriched  $\text{Me}^{202}\text{Hg}$  (Olsen et al., 2016). The added concentrations of  
164 acetate, BES, molybdate, and  $\text{Me}^{202}\text{Hg}$  were 9.2, 7.1, 1.4  $\mu\text{mol}$ , and 0.3  $\text{ng g}^{-1}$  wet soil, respectively  
165 (Appendix A Table S1). Since the two soils differed in water contents (Table 1), the final  
166 amendment concentrations per g dry weight were also provided in Appendix A Table S2.  
167 Additionally, due to a low sulfate concentration in the bog soil, a sixth amendment with sulfate (at  
168  $\sim 13.3 \mu\text{mol g}^{-1}$  wet soil) was performed to investigate the effects of increasing sulfate availability  
169 on MeHg production. All the serum bottles were then capped with thick rubber stoppers, sealed  
170 with aluminum crimps, and flushed with ultra-pure  $\text{N}_2$  for 10 min. Three replicates were  
171 constructed for each treatment at each sampling timepoint. Note that inhibitor experiments with  
172 the fen soil (Amendments 3 and 4) were monitored up to 30 days due to limited availability of the  
173 soil, and all others were monitored up to 60 days. An incubation temperature of 8  $^\circ\text{C}$  was used to  
174 simulate relatively high temperatures observed in summer months for accelerated microbial  
175 methylation (Kim et al., 2016; Léger et al., 2019), although the average mean annual air  
176 temperatures are about  $-1.0$  to  $-2.1$   $^\circ\text{C}$  at these sites. Changes in ambient MeHg and  $\text{Me}^{202}\text{Hg}$   
177 concentrations over the course of incubation were measured separately to determine respective  
178 methylation and demethylation in the soil.

### 179 *2.3. Chemicals and chemical analyses*

180 All chemicals used in this study are certified analytical reagents from Sigma-Aldrich and  
181 Fisher Scientific, and no detectable amounts of Hg were found in the stock solution of these reagents, as  
182 previously reported (Wang et al., 2020). Enriched Hg stable isotopes,  $^{202}\text{Hg}$  (95.86%) and  $^{200}\text{Hg}$

183 (96.41%), were obtained from the Isotope Enrichment Program at the Oak Ridge National  
184 Laboratory, Oak Ridge, Tennessee, United States (Zhang et al., 2021), and used for the synthesis  
185 of isotopically labeled MeHg using the previously established method (Hintelmann and Ogrinc,  
186 2002; Parks et al., 2013).

187 Dissolved CH<sub>4</sub> concentrations in soil porewater (collected in sealed serum bottles in the  
188 field) were determined by equilibrating 1 mL porewater in crimp-sealed serum vials and then  
189 analyzing the headspace CH<sub>4</sub> using a gas chromatograph (GC) equipped with a methanizer and a  
190 flame ionization detector (SRI 8610C, SRI Instruments, Torrance, CA). The final CH<sub>4</sub>  
191 concentration was calculated based on Henry's law, as described previously (Philben et al., 2019).  
192 Similarly, headspace concentrations of CH<sub>4</sub> and CO<sub>4</sub> in the incubation bottles were analyzed at  
193 predetermined time intervals.

194 Following gas measurement, triplicate soil microcosms were destructively sampled. An  
195 aliquot of the soil (0.5–1 g wet weight) was digested in 10 mL of 1.5 M KBr in 5% H<sub>2</sub>SO<sub>4</sub> and 2  
196 mL of 1 M CuSO<sub>4</sub> for 1 h to release MeHg from the sample matrix as bromide derivatives by the  
197 combined reaction of KBr and Cu<sup>2+</sup> under acidic pH conditions (Alli et al., 1994; Furutani and  
198 Rudd, 1980). This was followed by the addition of 10 mL methylene chloride (99.99%, Fisher  
199 Scientific) for an additional 1 h to extract MeHg (Bloom et al., 1997; Yang et al., 2016a). Me<sup>200</sup>Hg  
200 was added to each sample as an internal standard to correct potential interferences or loss of MeHg  
201 during extraction and subsequent distillation processes. MeHg was then determined using a  
202 modified EPA Method 1630 via distillation, ethylation, trapping, and thermo-desorption with an  
203 automated MERX purge and trap system (Brooks Rand, Seattle, WA), followed by separation by  
204 gas chromatography, decomposition through a pyrolytic column, and detection on an inductively  
205 coupled plasma mass spectrometer (ICP-MS) (Elan DRC-e, PerkinElmer, Shelton, CT) (Zhang et

206 al., 2021). A sediment reference material (ERM-CC580 with a certified value of  $75 \pm 4$  ng MeHg  
207  $\text{g}^{-1}$ ) was extracted and run in parallel with samples for quality assurance and quality control  
208 (QA/QC). The measured values of the reference material ranged from 69.5 to 80.2 ng/g, and the  
209 recovery was within  $100 \pm 10\%$ . Calibration verification samples with known amounts (50 pg) of  
210 ambient MeHg (a reference standard from Brooks Rand) and  $\text{Me}^{202}\text{Hg}$  (25 pg) were also run about  
211 every 20 samples with a mean recovery of  $100 \pm 3.5\%$  (Zaporski et al., 2020; Zhang et al., 2021).  
212 The detection limit for MeHg on ICP-MS was about 6 pg Hg, equivalent to a detection limit of ~  
213  $0.02$  ng Hg  $\text{g}^{-1}$  dw soil. For inorganic Hg(II) analysis, the extractant solutions after sequential  
214 extraction (Section 2.1) were reduced with  $\text{SnCl}_2$  (100 mM), followed by purging with high-purity  
215  $\text{N}_2$  and detection of purgeable elemental Hg(0) on a Zeeman cold vapor atomic absorption  
216 spectrometer (CVAAS) (Lumex RA-915+, Ohio Lumex Co.). Hg calibration standards (0, 25, 50,  
217 75, 100 pg) were also run with the samples, and the detection limit of Hg by CVAAS was about  
218 10 pg (Lu et al., 2019).

219 Another two aliquots of the remaining soil (0.75 g wet wt) were extracted with 10 mL of  
220 degassed Milli-Q water or 0.1 M KCl. The soil slurry was shaken for 1.5 h and centrifuged at  
221  $3000 \times g$ , and the supernatant was collected by filtering through a  $0.2\text{-}\mu\text{m}$  PTFE syringe filter. The  
222 KCl extracts were immediately used to determine pH and ferrous Fe(II) concentrations with 1,10-  
223 phenanthroline (Hach method 8146) using a portable colorimeter (Hach DR900 Colorimeter)  
224 (Philben et al., 2019; Yang et al., 2016b). The detection limit was  $\sim 0.25$   $\mu\text{mol Fe g}^{-1}$  dw. All water  
225 extract samples were analyzed within three days of collection or frozen at  $-23$   $^\circ\text{C}$  until analysis.  
226 Major anions (e.g.,  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ ) and organic acids (e.g., formate, acetate, and propionate) were  
227 analyzed using previously established methods (Philben et al., 2019; Yang et al., 2016b). They  
228 were separated on a Dionex AS15 analytical column and an AG15 guard column and analyzed by

229 an ion chromatograph (Dionex Integrion, Thermo Scientific) equipped with a conductivity  
230 detector. The detection limit for each anion and organic acid was  $\sim 0.14 \mu\text{mol g}^{-1} \text{ dw}$ . WEOC was  
231 measured on a total organic carbon analyzer (TOC, Shimadzu) after the samples were acidified in  
232 0.1% HCl and purged to remove inorganic carbonates (Philben et al., 2019; Yang et al., 2016b),  
233 and the detection limit was  $\sim 0.31 \mu\text{mol C g}^{-1} \text{ dw}$ . For statistical analyses, all data were validated  
234 for normality using Shapiro-Wilk test (Schwartz et al., 2019). Differences in Hg species  
235 distributions between the two soils were tested by the two-sample t-test. Multi-comparisons of  
236 MeHg concentrations among samples with different amendments at day 15 or day 30 were tested  
237 by analysis of variance (ANOVA) and pairwise comparisons (Bonferroni test), using the OriginPro  
238 analysis software (OriginLab, Northampton, MA) with a significance threshold of  $P=0.05$ .

#### 239 *2.4. Soil DNA extraction and 16S rRNA amplification*

240 Soil DNA extractions were performed in triplicate from pre-incubation bulk samples ( $n=3$   
241 technical replicates) and the incubated samples ( $n=3$  per sampling point per treatment). We  
242 analyzed a total of 86 samples generated through incubations from the fen and bog soils (Philben  
243 et al., 2020). Total DNA was extracted by using 0.25 g of wet soil as input to the DNeasy PowerSoil  
244 Kit (Qiagen, Germantown, MD, USA), where the samples were incubated in bead-solution at 65  
245 °C for 5 min prior to bead-beating. Samples were disrupted by bead beating with a 1600 MiniG  
246 (SPEX Sample Prep, Metuchen, NJ, USA) at a setting of 1500 RPM for 60 s, and the DNA was  
247 further purified according to the kit protocol. 16S rRNA genes were amplified by PCR using  
248 primers (F515/R806) that target the V4 region of the 16S rRNA gene where the reverse PCR  
249 primer was barcoded with a 12-base Golay code (Caporaso et al., 2011). These primers contained  
250 degeneracy added to both the forward and reverse primers and were not biased in detecting  
251 methanogenic Archaea, hence used to perform concurrent analysis of bacteria and archaea in the

252 samples. The PCR reactions were performed as described previously (Philben et al., 2020). Each  
253 sample was amplified in triplicate, combined, purified with Agencourt AMPure XP PCR  
254 purification system (Beckman Coulter, Brea, CA), and quantified using the Qubit dsDNA HS  
255 assay (Invitrogen, Carlsbad, CA, USA). Amplicons were pooled (10 ng/sample) and sequenced on  
256 one lane of the Illumina Miseq platform (Illumina Inc, San Diego, CA). Quality filtered sequence  
257 data were deposited at European Nucleotide Archive (PRJEB37429).

### 258 2.5. 16S rRNA gene sequence analysis

259 Paired-end amplicon sequences were overlapped and merged using FLASH (Magoč and  
260 Salzberg, 2011). Quality filtering ( $>q30$ ) and demultiplexing (Bokulich et al., 2013) resulted in  
261 445,195 sequences for the fen soil and 327,007 sequences for the bog soil. Sequences were grouped  
262 into operational taxonomic units (OTUs) based on 97% sequence identity, and chimeric sequences  
263 were removed using UPARSE (Edgar, 2013). OTUs were given taxonomic assignments in QIIME  
264 (Caporaso et al., 2010) version 1.7.0 using the RDP classifier (Wang et al., 2007) and the SILVA  
265 database 132 (Quast et al., 2012). Phylogenetic trees were created using FastTree (Price et al.,  
266 2010) under QIIME's default parameters. All remaining analyses were performed in R version  
267 3.5.1 (Bates et al., 2014; Team, 2013) via use of phyloseq (McMurdie and Holmes, 2013), taxize  
268 (Chamberlain and Szöcs, 2013), vegan (Oksanen et al., 2013), and ggstatsplot (Patil, 2019)  
269 packages. Amplicon data were proportionally rarified (to the 90% of the minimum sample depth  
270 in the dataset without replacement resulting in 7000 sequences per sample) and normalized, and  
271  $\beta$ -diversity (Shannon's H) was assessed using Bray-Curtis distance (Bray and Curtis, 1957). For  
272 multiple comparisons,  $P$  values were adjusted via Bonferroni method. Strains and taxa associated  
273 with Hg(II) methylation based on the presence of the marker gene *hgcA* were retrieved from the  
274 Integrated Microbial Genomes & Microbiomes (IMG/M) database of the Joint Genome Institute

275 (JGI). The sequence of *hgcA* from the well characterized methylator *Desulfovibrio desulfuricans*  
276 ND132 (locus tag DND132\_1056, IMG Gene ID 2503785873) was used as the query gene for the  
277 Homolog Toolkit in IMG/ER, which uses the Basic Local Alignment Search Tool (BLAST) with  
278 an e-value cutoff of  $1 \times 10^{-5}$  to identify homologs present in IMG. Homologous sequences without  
279 the conserved sequence motif [G(I/V)NVWC(A/S/G)(A/S/G)GK] were removed (Cooper et al.,  
280 2020; Parks et al., 2013), which resulted in 604 unique genomes with confirmed *hgcA* genes  
281 (Appendix B). Full taxonomic lineages for the identified *hgcA* genomes were retrieved using the  
282 taxize package (Chamberlain and Szöcs, 2013). The results were compared to 16S rRNA  
283 sequencing results to select taxa that may harbor mercury methylating bacteria.

284

### 285 **3. Results and Discussion**

#### 286 *3.1. Geochemical properties of the fen and bog soils*

287 Geochemical properties of the fen and bog soils are shown in Table 1. Notably, the bog  
288 soil was more acidic, with a pH of  $5.2 \pm 0.2$ , than the fen soil, with a pH of  $6.7 \pm 0.2$  (Table 1). The  
289 dissolved  $\text{CH}_4$  concentration was lower in the porewater in the fen soil ( $0.04 \pm 0.002$  mM) than in  
290 the bog soil ( $0.36 \pm 0.04$  mM). The total C contents were 26% and 17% in the fen and bog soils,  
291 respectively. Consistently, WEOC in the fen soil extract [ $50.0 \pm 22.0$   $\mu\text{mol g}^{-1}$  dry weight (dw) soil]  
292 was higher than that in the bog soil ( $31.7 \pm 2.0$   $\mu\text{mol g}^{-1}$  dw). The total extractable LMW organic  
293 acids ( $2.38 \pm 0.07$   $\mu\text{mol g}^{-1}$  dw) in the fen soil were about twice of those in the bog soil ( $1.22 \pm 0.06$   
294  $\mu\text{mol g}^{-1}$  dw). Extractable Fe(II) and sulfate concentrations were  $5.4 \pm 0.2$   $\mu\text{mol g}^{-1}$  dw and  $< 0.14$   
295  $\mu\text{mol g}^{-1}$  dw (or below the detection limit), respectively, in the bog soil, and were  $2.7 \pm 0.2$   $\mu\text{mol g}^{-1}$   
296 dw and  $0.45 \pm 0.04$   $\mu\text{mol g}^{-1}$  dw in the fen soil (Table 1). These observations suggest that the bog

soil was likely more reducing than the fen soil, as the water table at the Teller fen site is more fluctuating due to runoff along the hillslope (Philben et al., 2020).

299

**Table 1.** Comparisons of field porewater and soil extract properties of the fen and bog soils

<b>Field properties</b>	<b>Fen</b>	<b>Bog<sup>c</sup></b>
pH of field porewater	6.7±0.2 <sup>a</sup>	5.2±0.2 <sup>*</sup>
Dissolved CH <sub>4</sub> in porewater (mM)	0.040±0.002	0.36±0.04 <sup>**</sup>
Water content (%)	71±0.1	62±0.2 <sup>*</sup>
Total C (%)	26±0.3	17±0.2 <sup>*</sup>
<b>Soil extract properties</b>		
pH	6.7±0.2	5.2±0.2 <sup>*</sup>
Sulfate (μmol g <sup>-1</sup> dw)	0.45±0.04	<0.14 <sup>**</sup>
Fe(II) (μmol g <sup>-1</sup> dw)	2.7±0.2	5.4±0.2 <sup>*</sup>
WEOC <sup>b</sup> (μmol g <sup>-1</sup> dw)	50.0±22.0	31.7±2.0
Sum of organic acids (μmol g <sup>-1</sup> dw)	2.40±0.07	1.20±0.06 <sup>*</sup>
THg (ng g <sup>-1</sup> dw)	63.6±4.5	34.6±2.0 <sup>*</sup>
MeHg (ng g <sup>-1</sup> dw)	0.11±0.01	0.86±0.05 <sup>**</sup>

301

Notes: a. Data averaged from triplicate samples with errors representing one standard deviation.

303

b. WEOC = water-extractable organic carbon; THg = total mercury.

304

c. Significant differences between the two soils are marked as (\*)  $P < 0.05$  and (\*\*)  $P < 0.01$ .

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Sequential extraction results showed similar patterns of extractable Hg pools in both the

fen and bog soils (Fig. 1). The acid-soluble Hg accounted only for about 0.24% of the total Hg in

both soils. The HgS-like pool accounted for ~22% and 17.8% in the fen and bog soils, respectively.

However, the majority of Hg in the tundra soils was associated with soil organic matter (Fig. 1),

resulting from its high binding affinities with Hg (Lehnerr, 2014; Zhang et al., 2019). The fen

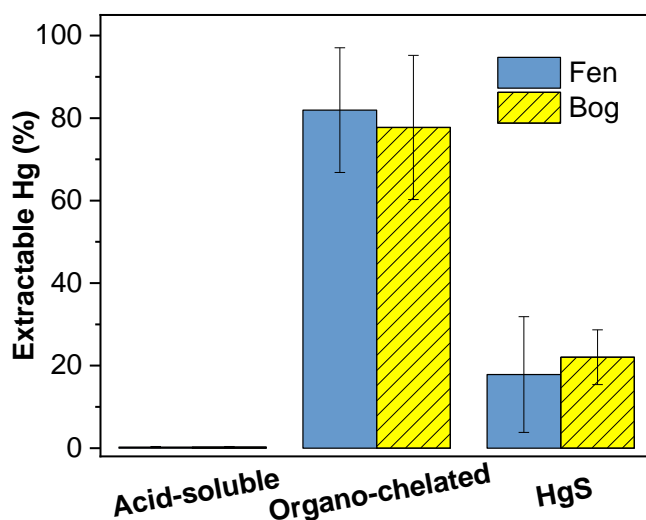
soil showed ~77.7% organo-chelated Hg and the bog soil ~82%, and no significant differences in

Hg species distributions were observed between the two soils. Interestingly, although the total Hg

(THg) in the bog soil (34.6±2.0 ng g<sup>-1</sup> dw) was only ~54% of that in the fen soil (63.6±4.5 ng g<sup>-1</sup>

dw), the MeHg concentration in the bog soil (0.86±0.05 ng g<sup>-1</sup> dw) was about 7-fold higher than

318 that in the fen soil ( $0.11 \pm 0.004 \text{ ng g}^{-1} \text{ dw}$ ). The higher total Hg but lower MeHg concentrations in  
 319 the fen soil than the bog soil imply that differences in soil hydrobiogeochemical conditions may  
 320 have influenced net Hg(II) methylation. More specifically, the more stagnant, reducing  
 321 environments in the bog soil likely favored MeHg production by anaerobic microorganisms (Parks  
 322 et al., 2013; Podar et al., 2015), as compared to the hillslope fen soil with frequent water table  
 323 fluctuations and runoffs (Philben et al., 2020).



324 **Fig. 1.** Sequential extraction of Hg in the fen and bog soils. Three Hg pools were extracted and  
 325 defined as: (1) acid-soluble Hg extracted by 0.01 M HCl and 0.1 M CH<sub>3</sub>COOH, (2) organo-  
 326 chelated Hg extracted by 1 M KOH, and (3) HgS extracted by aqua regia. Data were averaged  
 327 from triplicate samples with error bars representing one standard deviation. Each extractable Hg  
 328 pool was normalized to the measured total Hg in the fen soil ( $63.6 \pm 4.5 \text{ ng g}^{-1} \text{ dw}$ ) or in the bog  
 329 soil ( $34.6 \pm 2.0 \text{ ng g}^{-1} \text{ dw}$ ). No significant changes were observed in each Hg pools between the two  
 330 soils at  $P = 0.05$ .

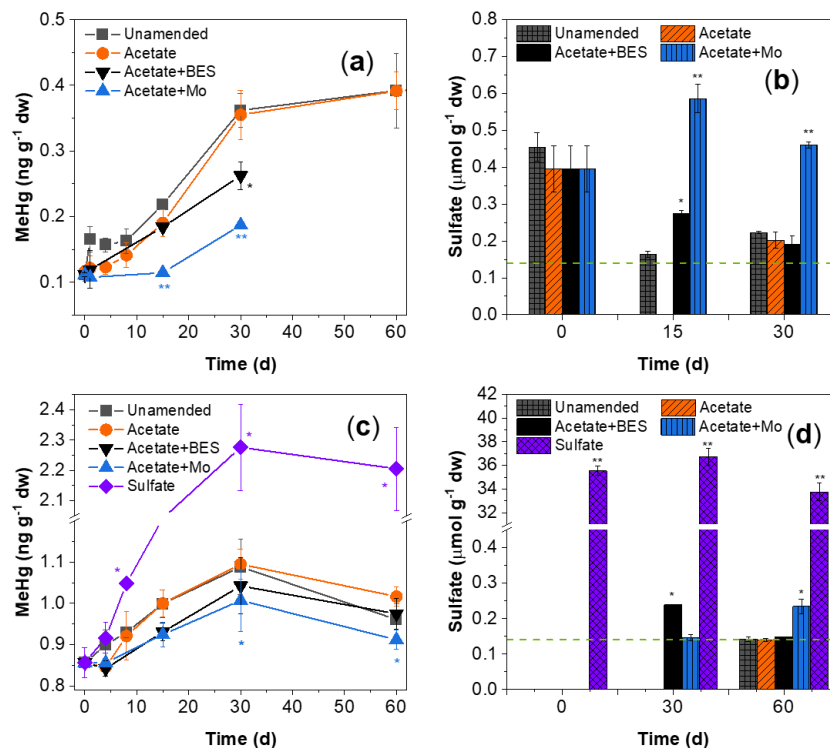
### 332 3.2. Methylmercury production in the fen soil

333 Microcosm incubations resulted in a rapid increase in MeHg production in the  
 334 circumneutral fen soil, particularly in the first 30 days either with or without (unamended) acetate



335 addition (Fig. 2a). The MeHg concentration in the unamended soil increased from  $0.11 \pm 0.004$  at  
 336 day 0 to  $0.36 \pm 0.02$  ng g<sup>-1</sup> dw at day 30 (Fig. 2a). However, after 30 days, MeHg production slowed  
 337 down, and its cumulative concentration increased to only  $\sim 0.39 \pm 0.06$  ng g<sup>-1</sup> dw from day 30 to 60.  
 338 The incubation with acetate addition produced similar amounts of MeHg as in the unamended soil  
 339 (Fig. 2a), indicating that increasing acetate concentration had little effect on MeHg production or  
 340 accumulation in the fen soil.

341



342

343 **Fig. 2.** Effects of acetate addition and microbial inhibitors on MeHg production and water-  
 344 extractable sulfate concentrations in the fen (a, b) and bog (c, d) soils at 8 °C. Data were averaged  
 345 from triplicate incubations with error bars representing one standard deviation. Sulfate  
 346 concentrations (Philben et al., 2020) below the detection limit (indicated by dotted lines) are not  
 347 shown. Significant changes among treatment groups at each timepoint are marked as (\*)  $P < 0.05$   
 348 and (\*\*)  $P < 0.01$ . Inhibitor-addition incubations of the fen soil were not conducted at day 60.

349 Experiments with selective microbial inhibitors showed that the addition of BES did not  
350 significantly affect MeHg production in the fen soil within the first 15 days (Fig. 2a), as a similar  
351 amount of MeHg ( $0.18 \pm 0.01 \text{ ng g}^{-1} \text{ dw}$ ) was observed as those in the control incubations  
352 (unamended and acetate-amended treatments) ( $P=1$ ). The result suggests that methanogens played  
353 a negligible role in MeHg production, as BES is a structural analog of coenzyme M (CoM) and a  
354 competitive inhibitor of methyl-CoM reductase enzyme required for methanogenesis (Hamelin et  
355 al., 2011; Kronberg et al., 2018; Schaefer et al., 2020). In contrast, molybdate addition nearly  
356 halted Hg(II) methylation in the first 15 days and no net increase in MeHg production was observed  
357 from day 0 to day 15 ( $P < 0.001$ ) (Fig. 2a). Consistent with the pattern of MeHg production, addition  
358 of molybdate inhibited sulfate reduction, as molybdate is known to inhibit SRB by mechanisms  
359 including inhibition of the key ATP sulfurylase enzyme and interference with sulfate transport  
360 (Biswas et al., 2009). Therefore, the sulfate concentration increased, rather than decreased, within  
361 the first 15 days (Fig. 2b). This increase in sulfate concentrations in the acetate+molybdate  
362 treatment at days 15 and 30 was attributed to both the lack of microbial reduction and the addition  
363 of a relatively high concentration of acetate ( $31.7 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) and molybdate ( $4.8 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ )  
364 resulting in desorption or anion exchange of sulfate initially adsorbed on the soil (Antelo et al.,  
365 2012; Philben et al., 2020). In contrast, water-extractable sulfate concentrations in all other  
366 treatments decreased from  $\sim 0.4 \text{ } \mu\text{mol g}^{-1} \text{ dw}$  at day 0 to near or below the detection limit ( $\sim 0.14$   
367  $\text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) after 15 days, indicating active sulfate reduction. Evidently, the halted Hg(II)  
368 methylation in the molybdate-amended soil was well correlated to suppressed sulfate reduction or  
369 microbial activities of SRB by molybdate. These results suggest that SRB likely played a dominant  
370 role in Hg(II) methylation whereas methanogens played a negligible role in the first 15 days.

371           The inhibition patterns of BES and molybdate on MeHg production, however, varied after  
372 15 days when sulfate became depleted (Fig. 2a). From day 15 to day 30, while the addition of  
373 molybdate still inhibited MeHg production, it did not completely halt Hg(II) methylation as  
374 observed within the first 15 days, and MeHg concentration increased from  $0.11 \pm 0.003 \text{ ng g}^{-1} \text{ dw}$   
375 at day 15 to  $0.19 \pm 0.003 \text{ ng g}^{-1} \text{ dw}$  at day 30 in the molybdate-amended incubation. The amount of  
376 MeHg accumulation ( $0.08 \pm 0.01 \text{ ng g}^{-1} \text{ dw}$ ) was only  $33.3 \pm 5.4\%$  of that observed in the control  
377 incubation ( $0.24 \pm 0.025 \text{ ng g}^{-1} \text{ dw}$ ), implying that SRB potentially contributed up to  $\sim 67\%$  to Hg(II)  
378 methylation. Coincidentally, MeHg production in the BES-amended incubation was  $\sim 38\%$  lower  
379 than that in the control experiment at day 30 (Fig. 2a), suggesting that methanogens contributed  
380 up to  $\sim 38\%$  of the methylation. Therefore, both SRB and methanogens appeared to be the major  
381 contributors to Hg(II) methylation after 15 days when sulfate was consumed in the incubations  
382 without molybdate or BES additions (Fig. 2b). SRB thus could be inhibited and unable to  
383 outcompete methanogens (Dang et al., 2019; Hausmann et al., 2016 ). Instead, SRB might form  
384 syntrophic partnerships with methanogens (Yu et al., 2018), and both clades could contribute to  
385 Hg(II) methylation. These observations are consistent with recent work reporting concomitant  
386 SRB and methanogen activities in methylating Hg(II) in boreal wetlands (Schaefer et al., 2020). It  
387 was found that, in the absence of sulfate and other electron acceptors, SRB syntrophs could act as  
388 fermenters and transfer electrons to methanogens, thereby increasing MeHg production (Yu et al.,  
389 2018). Inhibition of either one of the microbial groups may thus suppress MeHg production  
390 (Schaefer et al., 2020). Therefore, there appears a transition of responsible microbial communities  
391 from predominantly SRB in the first 15 days to both SRB and methanogens after 15 days in  
392 methylating Hg(II), and this transition was affected by geochemical properties of the soil or the  
393 availability of sulfate.

394 3.3. Methylmercury production in the bog soil and comparisons with the fen soil

395 In the acidic bog soil, rapid MeHg production was also observed either with or without the  
396 addition of acetate during incubation (Fig. 2c). The MeHg concentration in the unamended  
397 incubation increased from  $0.86 \pm 0.05 \text{ ng g}^{-1} \text{ dw}$  at day 0 to  $1.09 \pm 0.07 \text{ ng g}^{-1} \text{ dw}$  at day 30 (Fig. 2c),  
398 which was comparable to the amount of MeHg increased in the Teller fen soil. After 30 days,  
399 MeHg amount reached a plateau and even decreased in some incubations at day 60, especially in  
400 the unamended and acetate-only-amended incubations. The addition of acetate again showed  
401 negligible effects on MeHg production, as similar concentrations of MeHg were observed in the  
402 unamended and acetate-only-amended bog soil treatments ( $P=1$ ) (Fig. 2c). These results suggest  
403 that the observed slower MeHg accumulation in both the fen and bog soils after 30 days of  
404 incubation (Figs. 2a and 2c) could not be attributed to limited availability of labile organic C.  
405 Moreover, acetate concentration declined ( $\sim 10 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) in the acetate-only-treated fen soil  
406 (Appendix A Fig. S1b), but no consumption of acetate was observed in the bog soil (Appendix A  
407 Fig. S2b). Hence, MeHg accumulation did not correlate with acetate consumption. These  
408 observations are consistent with a previous study showing that acetate addition did not affect  
409 MeHg production in peatlands, which also contained high amounts of labile organic C (Mitchell  
410 et al., 2008). However, our results differ from a recent study reporting that acetate addition  
411 substantially increased MeHg production in sand dune sediments (Zaporski et al., 2020). These  
412 different findings may be attributed to a much lower availability of labile C substrates in the sand  
413 dune sediments than that in the tundra soils. The sand dune sediments contained low concentrations  
414 of acetate ( $<0.1 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) and WEOC ( $1.5 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) (Zaporski et al., 2020), while the  
415 tundra soils used in this study contained much higher amounts of acetate ( $0.8\sim 1.7 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ )  
416 and WEOC ( $>30 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ ) (Table 1). Therefore, the availability of abundant organic C

417 substrates did not limit microbial activities in methylating a small quantity of Hg(II) in both the  
418 fen and bog soils, so that increasing acetate availability showed insignificant effects on MeHg  
419 production.

420 With exception of the sulfate-amended treatment, there were no clear trends of sulfate  
421 reduction observed in all the incubations of the bog soil (Fig. 2d), in part due to its low sulfate  
422 concentrations (Fig. 2d, Table 1). However, the addition of sulfate to the bog soil substantially  
423 promoted MeHg production as compared to those without sulfate amendment (Fig. 2c), suggesting  
424 that sulfate was a limiting factor of Hg(II) methylation in this soil. At day 30, net MeHg production  
425 in the sulfate-amended incubation was  $1.42 \pm 0.14 \text{ ng g}^{-1} \text{ dw}$ , about 5-fold higher than those  
426 observed in the no-sulfate-amended incubations. The result suggests that Hg(II) methylation was  
427 strongly influenced by sulfate reduction in the soil. However, between day 30 and 60, MeHg  
428 production in the sulfate-amended incubation slowed down or decreased, similarly as observed in  
429 the fen soil, although the sulfate concentration remained high after 60 days ( $\sim 33.7 \text{ } \mu\text{mol g}^{-1} \text{ dw}$ )  
430 (Fig. 2d). The decrease in MeHg concentrations could be partially attributed to potential  
431 demethylation in the soil, as previously reported (Barkay and Gu, 2021; Bridou et al., 2011;  
432 Hamelin et al., 2015; Lu et al., 2017). This conclusion is supported by additional experiments  
433 incubating both the fen and bog soils with isotopically labeled  $\text{Me}^{202}\text{Hg}$ . The results showed an  
434 insignificant decrease in  $\text{Me}^{202}\text{Hg}$  concentrations at day 30 in both soils (Appendix A Fig. S3).  
435 However, at day 60, the  $\text{Me}^{202}\text{Hg}$  concentration decreased significantly ( $P=0.028$ ) in the bog soil.  
436 Therefore, the plateaued or decreased MeHg concentrations during incubation could be attributed  
437 to concurrent methylation and demethylation, although additional biogeochemical factors, such as  
438 the formation of HgS, bioavailable Hg(II) pools, and electron acceptors, may also limit MeHg  
439 production in these soils (Lin et al., 2014; Luo et al., 2017; Mitchell et al., 2008).

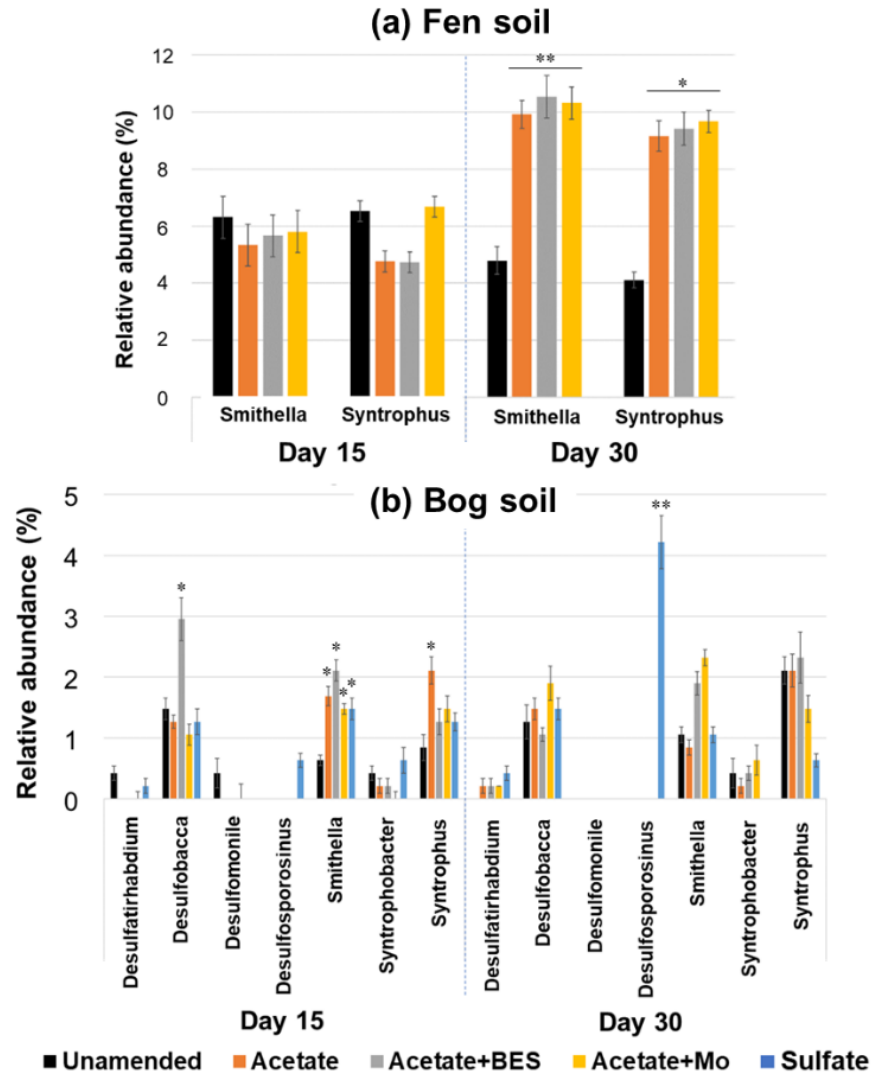
440 The effects of BES and molybdate inhibitors on Hg(II) methylation differed between the  
441 bog and fen soils. Unlike the fen soil, the addition of BES or molybdate only partially inhibited  
442 MeHg production (differences insignificant at  $P=0.05$ ) in the first 15 days in the bog soil (Fig. 2c).  
443 At day 15, the average amounts of MeHg produced in the BES-amended and molybdate-amended  
444 soils were  $0.076\pm 0.042$  and  $0.069\pm 0.047$  ng g<sup>-1</sup> dw, respectively, equivalent to about  $52.8\pm 33.6\%$   
445 and  $47.5\pm 37.9\%$  of MeHg produced in the controls (i.e.,  $0.15\pm 0.05$  ng g<sup>-1</sup> dw in the unamended  
446 and acetate-only-amended soils). This inhibition pattern did not change over time, suggesting that  
447 both SRB and methanogens continuously contributed to Hg(II) methylation in the bog soil. The  
448 effectiveness of molybdate and BES in inhibiting SRB and methanogens was evident in  
449 substantially decreased sulfate reduction (Appendix A Figs. S1 and S2) or stalled CH<sub>4</sub> production  
450 (Appendix A Fig. S4) over the incubation period.

451 Together, these results indicate that Hg(II) methylation was not limited by acetate, likely  
452 due to its high abundance in the fen and bog soils. However, sulfate availability is an important  
453 factor in affecting SRB activity for MeHg production in these tundra soils. Comparing the  
454 incubation results between the fen and bog soils, SRB appeared to play a dominant role in MeHg  
455 production via sulfate reduction when sulfate was available (e.g., in the first 15 days of incubation  
456 of the fen soil) (Fig. 2a). In the sulfate-deficient bog soil, sulfate amendment therefore substantially  
457 increased MeHg production (Fig. 2c). However, when sulfate availability was limited, similar  
458 inhibition patterns by BES and molybdate on MeHg production were observed in the fen soil after  
459 15 days (Fig. 2a) and in the bog soil (Fig. 2c), suggesting a potential role of SRB-methanogen  
460 syntrophic partnerships in Hg(II) methylation, as previously reported (Schaefer et al., 2020; Yu et  
461 al., 2018). While it is difficult to pinpoint whether both SRB and methanogens or only one of them  
462 were methylating Hg(II) as MeHg production could be suppressed by inhibition of either one of

463 the metabolic groups (Schaefer et al., 2020), both methanogens and SRB at least directly or  
464 indirectly contributed to MeHg production in these tundra soils (see additional details in Section  
465 3.4 and 3.5). For a long time, the role of SRB in Hg(II) methylation in tundra soils have been  
466 controversial, likely due to low sulfate concentrations or low SRB activities in these soils (Barkay  
467 and Poulain, 2007; Loseto et al., 2004; Yang et al., 2016a). Our results indicate that SRB  
468 abundance and activities could be stimulated by increasing sulfate concentrations and thus play an  
469 important role in biogeochemical transformation of Hg(II) in these soils (Åkerblom et al., 2020;  
470 Hu et al., 2020; Mitchell et al., 2008). Moreover, even when sulfate is limiting, SRB could  
471 potentially develop syntrophic partnership metabolizing alternative substrates with methanogens  
472 (Plugge et al., 2011), thereby contributing to MeHg production. Depending on sulfate availabilities  
473 in the fen and bog soils, SRB could contribute about 50%–100% directly or indirectly to MeHg  
474 production. However, we cannot rule out that other microbes may have contributed to Hg(II)  
475 methylation. Nitrate concentrations did not change appreciably over the incubation in both soils  
476 (Philben et al., 2020), suggesting that denitrifiers were not involved in Hg(II) methylation in these  
477 soils. FeRB may contribute to MeHg production (discussed in Section 3.5), as active Fe(III)  
478 reduction was observed in both the fen and bog soils during incubation (Philben et al., 2020).  
479 However, there are no specific inhibitors for FeRB (Bravo et al., 2018; Schaefer et al., 2020),  
480 making it difficult to assess its contribution to Hg(II) methylation.

481

482



483  
484

485 **Fig. 3.** Changes in relative abundances of microbial guilds containing SRB at the Genus level in  
486 the fen (a) and bog (b) soils either without or with amendments of acetate, acetate+BES,  
487 acetate+Mo, or sulfate after 15 and 30 days. Relative abundances were averaged from triplicate  
488 incubations with error bars representing one standard deviation. Significant changes with different  
489 amendments at each timepoint are marked as (\*)  $P < 0.05$  and (\*\*)  $P < 0.01$ .

490

### 491 3.4. Microbial community responses to incubation

492 Philben et al. (2020) previously showed that microbial community compositions before  
493 incubation were different between the fen and bog soils used in this study. Beyond the phylum



494 level differences among different bacteria, the composition of methanogenic archaea was  
495 significantly different between the fen and bog soils, where the hydrogenotrophic Rice Cluster II  
496 dominated the *Euryarchaeota* in the fen soil, while the *Euryarchaeota* in the bog soil were more  
497 diverse and contained both acetoclastic and hydrogenotrophic OTUs (Philben et al., 2020). In both  
498 soils, microbial guilds containing SRB constituted a small portion of the population (Appendix A  
499 Fig. S5) before incubation. *Smithella* and *Syntrophus* containing SRB members (Plugge et al.,  
500 2011) were the most abundant genera in both soils, consisting of  $12.8\pm 0.2\%$  and  $3.8\pm 0.3\%$  of total  
501 bacteria in the fen and bog soils, respectively. However, SRB diversity was higher in the bog soil,  
502 as OTUs belonging to *Desulfatirhabdium*, *Desulfobacca*, *Desulfomonile* and *Syntrophobacter*  
503 were also detected in addition to *Smithella* and *Syntrophus* (Appendix A Fig. S5). More diverse  
504 methanogenic archaea and SRB in the bog soil than in the fen soil may have contributed to  
505 comparable MeHg production in the two soils (Figs. 2a and 2c), overcoming potential limitation  
506 of lower concentrations of sulfate and pH in the bog soil (Table 1).

507         Analyses of species richness and  $\beta$ -diversity trends tied to the incubations showed that  
508 incubation time and amendments did not significantly change community diversity in both the fen  
509 and the bog soils over the course of 30 days (Appendix A Fig. S6). Focusing on microbial  
510 populations that are potentially significant to Hg(II) methylation, further analyses showed that  
511 microbial community structure response to acetate addition was limited in both soils. Changes in  
512 relative abundance of methanogenic archaea were marginal in both soils (Philben et al., 2020). For  
513 microbial guilds containing SRB, significant increases in *Smithella* ( $P=0.003$ ) and *Syntrophus*  
514 ( $P=0.028$ ) abundances were observed in the acetate-amended fen soil at day 30 (Fig. 3a). In the  
515 acetate-amended bog soil, similar increases were observed relative to the unamended incubation  
516 at day 15, but not at day 30 (Fig. 3b). These different microbial responses suggest that future

517 studies are needed to further elucidate the mechanisms and interactions of microbial communities,  
518 as the two soils differed in pH and sulfate availability (Liu et al., 1999; Schöcke and Schink, 1997).  
519 We also note that, in the sulfate-amended bog soils, the relative abundance of *Desulfosporosinus*,  
520 a genus of sulfate-reducing *Firmicutes*, significantly increased and became the most dominant  
521 SRB following the incubation (Control vs Sulfate:  $t_{(17)} = -4.22$ ,  $P=0.001$ ) (Fig. 3b), while  
522 *Desulfosporosinus* was not detected in the bog soil before incubation (Appendix A Fig. S5). The  
523 result is consistent with the observed promoted MeHg production after sulfate amendment in the  
524 bog soil (Fig. 2c), as strains of *Desulfosporosinus* have been reported to methylate Hg(II) in acidic  
525 habitats (Gilmour et al., 2013). These observations thus corroborate that increased sulfate  
526 availability stimulated microbial growth of SRB and the subsequent MeHg production. A recent  
527 study also reported that Hg(II) methylating communities at higher sulfate amendments in  
528 freshwater lake sediments were dominated by SRB populations (Jones et al., 2020).

529         Microbial community response to selective inhibitors (BES and molybdate), however, was  
530 minor after incubation of the two soils. In the fen soil, while relative abundances of microbial  
531 guilds containing SRB increased from day 15 to day 30, especially of *Smithella* ( $P=0.017$ ) and  
532 *Syntrophus* ( $P=0.038$ ) (Fig. 3a), there were no significant differences observed among the acetate-  
533 only, BES-, and molybdate-amended soils (Fig. 3a). Likewise, methanogens remained unchanged  
534 among different treatments (Appendix A Fig. S7). In the bog soil, addition of BES and molybdate  
535 also did not significantly affect the relative abundances of microbial guilds containing SRB (Fig.  
536 3b) or FeRB *Geobacter* spp. and methanogens (Philben et al., 2020). Compared to the acetate-  
537 amended soil, only *Desulfobacca* increased in the BES-amended bog soil after day 15, and  
538 *Smithella* increased in both the BES- and Mo-amended bog soil after 30 days. The relatively stable  
539 microbial communities with inhibitor amendments may be explained by potential syntrophy that

540 supports growth and survival of populations, as *Smithella* and *Syntrophus* are known to develop  
541 syntrophic interactions with methanogens when sulfate reduction is limited (Plugge et al., 2011).  
542 We note that 16S rRNA analysis could not distinguish whether the affected microorganisms are  
543 metabolically active or not (Li et al., 2017). However, despite the effects of microbial inhibitors  
544 on the overall community structures was insignificant or could not be identified, the inhibitors  
545 effectively inhibited sulfate reduction and methanogenesis (Figs. 2b and 2d, and Appendix A Figs.  
546 S1 and S4), suggesting important roles of each metabolic group in Hg methylation. Additionally,  
547 although acetate concentration was not a limiting factor in MeHg production in these tundra soils,  
548 acetate-induced changes in SRB abundances may still affect other biogeochemical processes, such  
549 as greenhouse gas production in these soils (Philben et al., 2020).

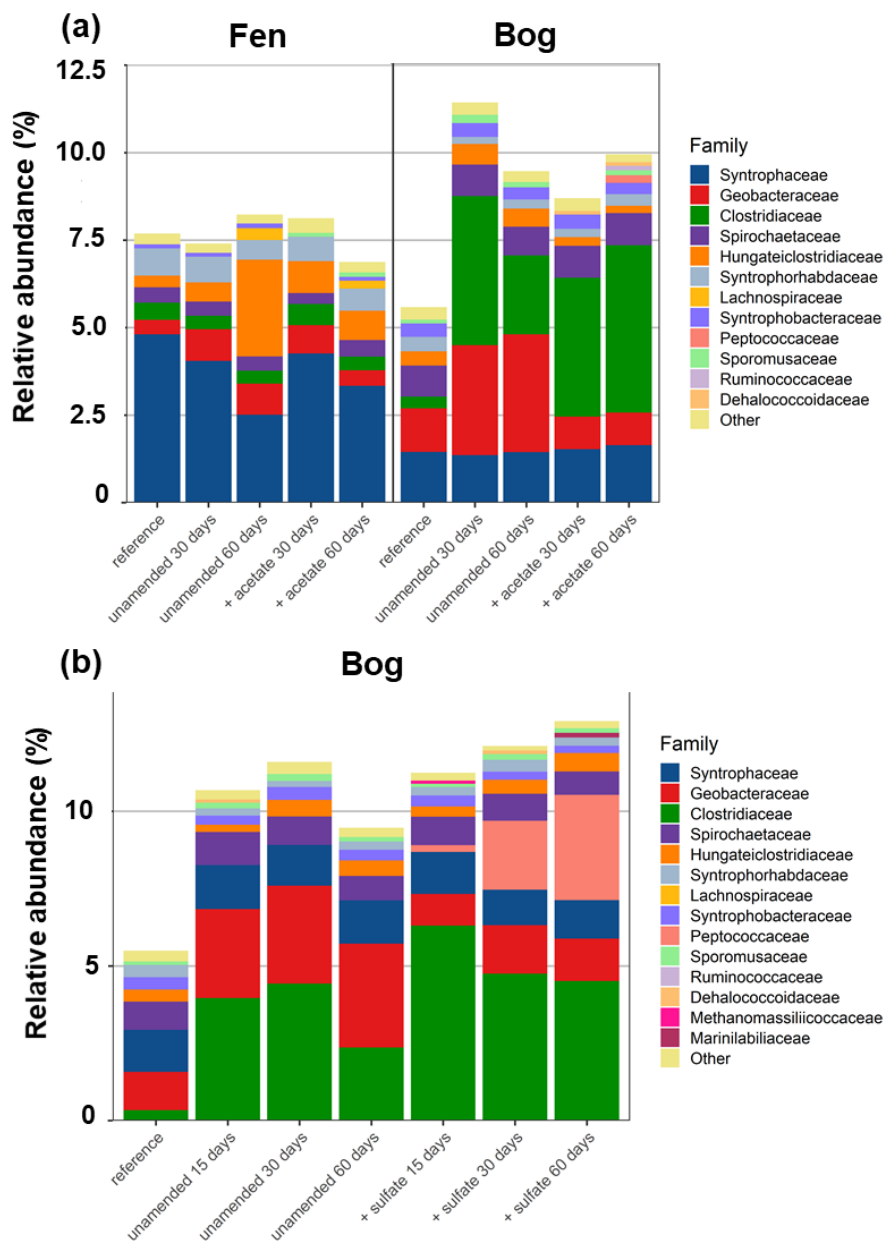
### 550 3.5. Potential microbial drivers of Hg(II) methylation

551 Arctic tundra is a significant sink of Hg due to its high soil organic matter content and  
552 could become an increasingly important source of MeHg to the Arctic ecosystems, as microbial  
553 production of MeHg in soil increases upon climate warming (Yang et al., 2016a). Microbial Hg(II)  
554 methylation has been associated with diverse groups of predominantly anaerobic bacteria and  
555 archaea (Gilmour et al., 2018; Gilmour et al., 2013). While Hg(II) methylation activity has been  
556 confirmed experimentally only for a small set of microorganisms, the presence of the genes, *hgcA*  
557 and *hgcB*, and their highly conserved sequence motifs correlate well with the ability to methylate  
558 Hg(II) (Podar et al., 2015; Schaefer et al., 2020). Therefore, these genes were used as proxies  
559 indicating the capacity to methylate Hg(II). Based on the distribution of *hgcA* among sequenced  
560 genomes and metagenomes publicly available in IMG/M (as of January 2021), individual Hg(II)  
561 methylating species appear to occur across 30 different phyla including several currently  
562 designated as candidate phyla (Appendix B). The majority of known methylators are associated

563 with the Deltaproteobacteria, the methanogenic Archaea, and the Firmicutes. Although not all  
564 species within a taxon may carry *hgcAB*, the genes are known to be more prevalent within certain  
565 clades and environments (Podar et al., 2015). While we identified a total of 12 family level taxa  
566 (distributed across 3 phyla) with relative abundances >0.1% that may harbor Hg(II) methylators  
567 in the fen and bog reference soils (Fig. 4a), family level abundances indicate that the vast majority  
568 of clades that may harbor Hg(II) methylators are within the Proteobacteria (Fig. 4a). On the family  
569 level, the Syntrophaceae accounted for 4.8% and 1.4%, respectively, in the reference fen and bog  
570 soils before incubation and was among the selected taxa showing the highest relative abundances  
571 in both soils (Fig. 4a). The genera *Smithella*, *Syntrophus*, and *Desulfobacca*, which are members  
572 of the Syntrophaceae, were observed in both the fen and bog soils (Fig. 3). Other groups in the  
573 order of relative abundance were the Syntrophorhabdaceae (0.77%), the Clostridiaceae (0.49%),  
574 and the Spirochaetaceae (0.45%) in the fen soil, and the Geobacteraceae (1.3%), the  
575 Spirochaetaceae (0.88%), and the Syntrophorhabdaceae (0.42%) in the bog soil.

576 As noted earlier, the addition of acetate led to some changes in the overall microbial  
577 community composition (Fig. 3 and Appendix A Fig. S8) between the unamended and acetate-  
578 amended fen soils, but minor differences were observed among family-level taxa associated with  
579 Hg(II) methylation, except for Hungateiclostridiaceae (Fig. 4a). However, in the bog soil the  
580 relative abundance of the Clostridiaceae increased from 2.4% in the unamended soil to 4.8% in  
581 the acetate-amended soil after 60 days (Fig. 4a). Although Hg(II) methylators are shown in seven  
582 genera among the Clostridiaceae (Appendix B), none of these specific genera appear to be present  
583 in the fen or bog soils (Appendix A Fig. S9), which suggests that the observed increase in the  
584 abundance of the Clostridiaceae should not impact MeHg production. The Geobacteraceae  
585 decreased from 2.9% in the unamended bog soil to 0.9% in the acetate-amended soil after 60 days

586 (Fig. 4a), although MeHg production was not impacted either with or without the addition of  
587 acetate (Fig. 2c). The presence of FeRB *Geobacter* was also detected (Appendix A Fig. S9), but a  
588 small decrease in their relative abundances was observed during incubation, suggesting that FeRB  
589 unlikely played an important role in Hg(II) methylation either. This result appears contradicting to  
590 a previous study, in which *Geobacteraceae* were thought to be important Hg(II) methylators in  
591 wetland soils (Schaefer et al., 2020). We therefore suggest that future studies are performed to  
592 further determining the role of FeRB in MeHg production, particularly when sulfate availability is  
593 limited in these soils. However, amendment with sulfate in the bog soil substantially increased the  
594 relative abundance of sulfate-reducing Clostridia, specifically the family Peptococcaceae (Fig. 4b),  
595 which harbors the genus *Desulfosporosinus* (Appendix A Fig. S10) containing 11 species carrying  
596 the *hgcA* gene essential for Hg(II) methylation (Appendix B). The relative abundances of other  
597 groups capable of sulfate reduction (e.g. the Syntrophaceae) did not change or decreased slightly  
598 (Fig. 4b). The results again suggest that metabolic interdependencies of syntrophic  
599 microorganisms might play an important role in driving the production of MeHg in these soils. We  
600 caution, however, that these estimates based on 16S rRNA gene results may have limitations with  
601 respect to DNA extraction method, primers, PCR cycles, and even the differences in 16S rRNA  
602 gene copy numbers per genome may impact identification of microbial communities and their  
603 abundances (Pollock et al., 2018). Future studies using more advanced and direct techniques (e.g.,  
604 transcriptomics) targeting syntrophic species carrying *hgcAB* are thus warranted to reveal  
605 additional factors controlling MeHg production in these arctic soils (Barkay and Poulain, 2007;  
606 Grégoire and Poulain, 2018; Yu et al., 2018).



607

608 **Fig. 4.** Relative abundances of taxa associated with Hg(II) methylation at the family level in the  
 609 fen and bog soils. Families at levels of less than 0.1% were combined into the group “Other”. (a)  
 610 Reference soils before incubation, unamended and acetate-amended soils after 30 and 60 days of  
 611 incubation. The relative abundance of individual taxa ranged from < 0.001 % to 4.8% in the fen  
 612 soil and from < 0.006 % to 4.8 % in the bog soil. (b) Effect of sulfate amendment on relative  
 613 abundances (> 0.1%) of taxa associated with Hg(II) methylation at the family level in bog soil  
 614 incubations.

### 615 **3. Conclusions**

616 Overall, our microcosm incubation experiments demonstrated that MeHg production in the  
617 tundra soil was not limited by acetate but sulfate availability, particularly in the bog soil with a  
618 low sulfate concentration. Increasing acetate concentration did not affect MeHg production, but  
619 sulfate addition significantly promoted MeHg production likely through stimulation of Hg(II)  
620 methylating SRB. Sulfate amendment in the bog soil incubations increased the relative abundance  
621 of Firmicutes, specifically the genus *Desulfosporosinus*. Our results indicate that the dominant  
622 microbial groups for Hg(II) methylation were dependent on soil geochemistry. In the fen soil,  
623 SRB appeared to be the predominant microbes in producing MeHg when sufficient amounts of  
624 sulfate were available. In contrast, both methanogens and SRB contributed to MeHg production  
625 directly and/or indirectly, possibly through syntrophic interactions when sulfate was limited.  
626 Together, these results reveal potential biogeochemical drivers of MeHg production in the Arctic  
627 tundra soils, an important step to advance our understanding of the transformation and  
628 bioaccumulation of Hg and MeHg in the rapidly changing Arctic ecosystem due to climate  
629 warming.

630

### 631 **Appendix A. Supplementary Information**

632 Changes in methane, acetate, sulfate, and methylmercury concentrations in the fen and bog  
633 soil incubations; Me<sup>202</sup>Hg concentrations over the course of demethylation experiments; Relative  
634 abundance of SRBs at *Genus* level in the fen and bog soils before incubation; Changes in species  
635 richness and diversity in the fen and bog soils after incubations for 30 day; Relative abundance of  
636 identified taxa at the phylum level in soil incubations after 60 days; Relative abundance of taxa (>  
637 0.1%) associated with Hg(II) methylation at the genus level in unamended and acetate-amended  
638 fen and bog soil incubations after 60 days; Relative abundance of taxa (> 0.1%) associated with

639 Hg(II) methylation at the family level in unamended and sulfate-amended bogl soil incubations;  
640 Incubation experimental setups and concentrations of amendments.

## 641 **Appendix B. Supplementary Data**

642 Genome IDs and description of microbial species with confirmed *hgcA* marker genes  
643 obtained from the Joint Genome Institute's IMG/M database and derived taxonomic lineages with  
644 Hg methylating microorganisms. Both Appendix A and B to this article can be found online at  
645 <https://doi XXXX>.

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656

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