UCLA UCLA Electronic Theses and Dissertations

Title

Molecular Amalgamations that Generate Constrained Peptidomimetic Macrocycles

Permalink <https://escholarship.org/uc/item/6jv2z2g2>

Author

Bernardino, Salvador Jurado

Publication Date 2022

Peer reviewed|Thesis/dissertation

UNIVERSITY OF CALIFORNIA

Los Angeles

Molecular Amalgamations that Generate Constrained

Peptidomimetic Macrocycles

A dissertation submitted in partial satisfaction of the

requirements for the degree Doctor of Philosophy in Chemistry

by

Salvador Jurado Bernardino

2022

© Copyright by

Salvador Jurado Bernardino

2022

ABSTRACT OF THE DISSERTATION

Molecular Amalgamations that Generate Constrained

Peptidomimetic Macrocycles

by

Salvador Jurado Bernardino Doctor of Philosophy in Chemistry University of California, Los Angeles, 2022 Professor Patrick G. Harran, Chair

Peptidomimetic macrocycles are increasingly growing as attractive scaffolds for the development of therapeutics targeting protein-protein interactions. Compounds of this type have been traditionally underrepresented in drug discovery platforms due to their structural characteristics not conforming to typical guidelines for 'drug-likeness'. We have developed methods to simultaneously investigate new bioactive chemotypes and their pharmacological properties in a systematic manner. Several libraries of peptidyl macrocyclic and polycyclic compounds are examined here based on previously developed template methodology and novel discoveries using a highly reactive fluorinated reagent. Both methods allow us to rapidly build

complexity from machine made, protecting-group-free bioactive starting materials. In this way, we can efficiently and broadly survey the chemical space of macrocycles through a diversity-oriented strategy.

In Chapter 2, we show that unprotected peptides react with commercial octafluorocyclopentene (OFCP) to afford hexafluorinated macrocycles via successive ipso substitutions of vinyl fluoride. Sequence variants having combinations of cysteine, tyrosine, histidine, and serine residues are shown to react rapidly at room temperature to give stable, isolable products. We also introduce our initial discoveries involving polysubstitution cascades with OFCP.

Chapter 3 details a broad scope of polycyclic compounds provided directly through onepot methods. Macrocycle and spirocycle variations are also examined. We found the vinyl fluoride position to be amenable to linker systems that provide access to polycycles incorporating previously unused amino acid residues. Non-cell-based permeability assays are performed to evaluate the properties of these structures.

In Chapter 4, using cell-based permeability assays we study the pharmacological properties of macrocycles incorporating various template generations derived from a core cinnamyl alcohol motif. Structure-permeability relationships are explored with these compounds systematically via a small library.

iii

The thesis of Salvador Jurado Bernardino is approved.

Jose A. Rodriguez

Hosea M. Nelson

Craig A. Merlic

Patrick G. Harran, Committee Chair

University of California, Los Angeles

2022

This dissertation is dedicated to my parents, Pablo and Maria Bernardino.

TABLE OF CONTENTS

LIST OF FIGURES

LIST OF SCHEMES

LIST OF TABLES

ACKNOWLEDGEMENTS

I am profoundly grateful to the many individuals who have helped me throughout my academic journey. My success is owed to the collective efforts of every mentor, Professor, friend, and family member I have encountered along the way. First, I would like to thank my advisor, Professor Patrick Harran for his unwavering guidance and support over the last five years. Thank you for believing in me and constantly challenging me to improve as a scientist. Your optimism and persistence towards difficult scientific problems has shaped my growth and will continue to inspire me throughout my career. Also, I'm grateful for the insightful advice and support of my committee, Professor Craig A. Merlic, Professor Jose. A. Rodriguez, and Professor Hosea M. Nelson.

The work described in the ensuing chapters was possible thanks to the contributions of many great scientists. Dr. Tomoyuki Tsunemi was instrumental to our discoveries with OFCP (Ch. 2) and following experiments. Thank you Tomo for your friendship and invaluable guidance as a mentor during your time here. Angel Mendoza provided significant contributions to the work described in Chapter 3, especially with the linker methodology. Previous lab members Dr. Tristin Rose, Dr. Brice Curtin, and Dr. Luke Sisto all contributed to the library of macrocycles described in Chapter 4. Thank you to Dr. Alan Mathiowetz and the Pfizer team for their work conducting cellular permeability assays (Ch. 4). Special thanks to Dr. Hui Ding for always being available for guidance and indispensable help with HPLC technical issues.

I also would like to thank past and present Harran lab members for their friendship and support including Hui Ding, Francesco Manoni, Tyler Allred, Emily Murzinski, Luke Sisto, Rupert Li, Evan Hurlow, Ishika Saha, Anton El Khoury, Morris Dweck, Angel Mendoza, Ani Mustafa, Gilbert Walker, Gabriella Cooper, Emma Greene, Alek Lotuzas, Jackie Bustamante, Kyle

Nagasawa and many others. I couldn't have asked for a better group of friends to experience graduate school with. You have all been amazing to work with and I wish you all the best.

I am indebted to many mentors I have encountered throughout my journey. To Mr. and Mrs. Precissi, thank you for serving as early sources of inspiration to pursue this field. To the faculty and staff at Delta College – Trini Araya, Dr. Kim, Dr. Ricardez, Ms. Demmons, Dr. Hastings Dr. Reedy, Dr. Gamarnik, and many others – I am tremendously grateful for your continued support and friendship. To Dr. Jesus Luna, thank you for taking me on and providing my first experiences in research. Your mentorship has been crucial to my success.

Finally, I would like to express my gratitude to my family. To my parents, thank you for instilling in me the value of hard work, responsibility, and humility. I am inspired everyday by your life's journey and compassionate values. You are my heroes. To my siblings, thank you for always being there and your support through tough love. Thank you to my brother and cousins, Juan and Justin, for making me feel like I'm back home with our entertaining late-night chats. I am also grateful for the love and support my entire extended family have given me throughout my life. And to my love Adriana, you have been there with me through it all. You have always been a source encouragement and motivation through the most difficult times. Thank you for everything.

VITA

FORMAL EDUCATION

University of California, Los Angeles, Los Angeles, CA Doctor of Philosophy in Chemistry **Expected: December 2022** Master of Science in Chemistry May 2019

University of California, Davis, Davis, CA Bachelor of Science in Pharmaceutical Chemistry Theorem Chemistry June 2017

San Joaquin Delta Community College, Stockton, CA May 2015 May 2015 Associate of Science in Mathematics and Science Associate of Art in Interdisciplinary Studies - Honors

PUBLICATIONS

S.J. Bernardino, A. Mendoza, M.J. Dweck, Declan, P.G. Harran, Cascade synthesis of fluorinated spiroheterocyclic scaffolding for peptidyl turns, loops, and macrobicyclic structures, *Manuscript in Preparation*.

T. Tsunemi, **S.J. Bernardino**, A. Mendoza, C. G. Jones, P. G. Harran, Syntheses of atypically fluorinated peptidyl macrocycles via sequential vinylic substitutions**,** *Angew. Chem.* **2020**, *132*, 684-688.

PRESENTATIONS

Bernardino, S.J., "Molecular Amalgamations that Generate Constrained Peptidomimetic Macrocycles" 2022 Organic Graduate Symposium, UCLA, Los Angeles, CA, June 10, 2022.

Bernardino, S.J., "Rapid Assembly of Tunable Fluorospiroheterocyclic Scaffolding for Peptidyl Turns, Loops and Macrobicyclic Structures" Merck UCC Symposium, October 11, 2021.

Bernardino, S.J.*, "Rapid Assembly of Tunable Fluorospiroheterocyclic Scaffolding for Peptidyl Turns, Loops and Macrobicyclic Structures" 6th UC Chemical Symposium, March 25-27, 2021. (**poster presentation**)

Bernardino, S.J.*; Zhu, X., Meijide, J., Zhang, P., Tugny, C., Derat, E., Zhang, Y., Mansuy, V.M., Menand, M., Fensterbank, L., Roland, S., Bistri, O., Sollogoub, M. "Synthesis of Functionalized NHC-Capped α-Cyclodextrin Metal Complexes" 253rd ACS National Meeting & Exposition, San Francisco, CA, April 2-6, 2017. (**poster presentation**)

HONORS - AWARDS

2022 Majeti-Alepati Dissertation Award 2021 Merck Research Award for Underrepresented Chemists of Color (UCC) 2017 – 2021 Eugene V. Cota-Robles Fellow 2017 American Chemical Society Scholar 2016 NSF-UF US/France REU Fellow 2015 Cal Aggie Alumni Association Leadership Scholar

Chapter 1 – Introduction

1.1 Background and Rationale

The use of small molecules at conventional protein targets such as enzymes, receptors and ion channels has historically been highly effective in treating many types of diseases.^{1,2} However, these compounds are generally limited to targeting well-defined binding pockets. A majority of the human proteome include proteins involved in intracellular signaling events through so-called protein-protein interactions (PPIs).³⁻⁵ In that regard, they provide the largest class of potential drug targets. However, the pockets and grooves present at the surfaces of proteins that bind to small molecules differ substantially from the shallow extended solvent-exposed surfaces of proteins involved in PPIs.⁶ For this reason the classic small molecule drug discovery approach is not amenable to targeting PPIs and have been conventionally considered as "undruggable". Most contact surfaces in protein-protein interfaces involve noncontiguous amino acid residues within the polymer chain. Although these interfaces are large, only a small subset of residues contribute to most of the binding-free energy known as 'hotspots'.⁷⁻¹² Traditional high-throughput screening libraries, tend to represent a relatively small number of chemical scaffolds geared towards typical enzyme targets, which are not suitable for the particularity of these PPI interfaces. 13-14 Historically, strict adherence to chemotypes following 'drug-likeness' has hampered progress in this area.¹⁵ Thus, creative construction of new chemical libraries through diversity-oriented methods is needed to more appropriately sample the chemical space required to target PPIs.

Peptides serve as a logical starting point for developing inhibitors of PPIs as they mimic natural protein-protein contacts. Macrocyclic peptides, however, retain the active structural components of linear peptides, while also improving key pharmacological properties. Their cyclic nature imparts on them several notable advantages over their linear counterparts including: (1) improved transport properties across biological membranes, achieved via the masking of polar

Figure 1.1. Notable examples of modified/naturally occurring cyclic peptides and peptidomimetic macrocycles.

functionality through internal hydrogen bonding¹⁶⁻¹⁷, (2) increased structural rigidity, in turn improving target binding affinity¹⁸⁻²⁰ and (3) their increased resistance to proteolytic degradation.²¹⁻ 23 Various conventional macrocyclization techniques currently exist along with numerous contemporary methods recently published with increased interest in the field.²⁴⁻²⁸ Over the years, peptidomimetic macrocycle design strategies have focused on mimicking secondary structures such as β- and γ-turns, β-strands, β-sheets and α-helices.²⁹⁻³⁴ Simple chemical modifications to further improve properties have also been developed including the use of D-amino acids,³⁵ retroinverse peptides, 36 N-methylation, 37 and side chain lipidation. 38 Cyclosporine A (CSA), a cyclic nonribosomal peptide natural product, is a well-known example of a permeable and orally bioavailable drug that violates conventional rules-of-thumb for drug-likeness.³⁹ CSA's capacity to passively diffuse across cell membranes has inspired many mechanistic studies with the intention of gaining insights to guide cell-permeable cyclic peptide design.⁴⁰⁻⁴¹ For example, N-methylations and D-amino acid incorporation of the Veber-Hirschman peptide (Figure 1), a somatostatin derivative with reported sst2 and sst5 somatostatin receptor selectivity, demonstrated improved bioavailability in Caco-2 cell monolayers, degradation resistance in plasma, and greater oral bioavailability in rats (F = 10%).⁴² Regardless of these modifications, the conformation of the bioactive non-N-methylated parent peptide was preserved with the presence of dual β-turn structures. Recently, Merck has developed a potent tricyclic peptidomimetic PCSK9 inhibitor (termed "Compound 44") currently at the preclinical stage for the treatment of high LDL cholesterol levels.⁴³ The structure has a molecular weight of 1612 g/mol and contains 9 H-bond donors yet demonstrates sufficient oral bioavailability to maintain therapeutic levels in animals. Despite these advances, efficient methods to incorporate unique biologically active lead structures with favorable properties are in demand.

Figure 1.2. Outline of research detailed in this dissertation. **A.** Previously developed template and new commercially available linker systems that are incorporated in peptidyl macrocycles. **B.** Peptidomimetic macrocycles and polycycles synthesized using linkers.

For more than a decade, our lab has been interested in developing efficient, protectinggroup-free methods for synthesizing macrocyclic peptides utilizing synthetic templates for facile macrocyclization.⁴⁴⁻⁴⁹ By incorporating novel functionality into subsequent generations of our templates, we've been able to develop increasingly complex macrocycles while fine-tuning their physical properties and subsequent transformations. Discovery- and target-oriented methods to survey the therapeutic potential of these hybrid macrocycles are needed, and the efficient assembly-line like protocols we have developed directly address this issue while expanding a previously narrow chemical space. This dissertation details the synthesis of peptidomimetic macrocycles/polycycles and subsequent studies of their pharmacological properties.

1.2 References

- 1. Makley, L. N. and Gestwicki, J. E. *Chem. Biol. Drug Des.* **2013**, *81*(1), 22-32.
- 2. Imming, P., Sinning, C., and Meyer, A. *Nat. Rev. Drug Discov.* **2006**, *5*(10), 821-834.
- 3. Bonetta, L. *Nature* **2010**, *468*(7325), 851-852.
- 4. Barabási, A. L., Gulbahce, N., and Loscalzo, *J. Nat. Rev. Genet.* **2011**, *12*(1), 56-68.
- 5. Lu, H.-C.; Fornili, A.; Fraternali, F. *Expert Rev. Proteomics* **2013**,*10*(6), 511.
- 6. Hopkins, A. L. and Groom, C. R. *Nat. Rev. Drug Discov.* **2002**, *1*(9), 727-730.
- 7. Clackson, T.; Wells, J. A. *Science* **1995**, *267*(5196), 383-386.
- 8. Clackson, T.; Ultsch, M. H.; Wells, J. A.; de Vos, A. M. *J. Med. Biol.* **1998**, *277*(5), 1111- 1128.
- 9. Muller, Y. A.; Li, B.; Christinger, H. W.; Wells, J. A.; Cunningham, B. C.; De Vos, A. M. *Proc. Natl. Acad. Sci.* **1997**, 94(14), 7192-7197.
- 10. Thanos, C. D.; DeLano, W. L.; Wells, J. A. *Proc. Natl. Acad. Sci.* **2006**, *103*(42), 15422- 15427.
- 11. Moreira, I. S.; Fernandes, P. A.; Ramos, M. J. *Proteins Struct. Funct. Bioinf.* **2007**, *68*(4), 803-812.
- 12. Thorn, K. S.; Bogan, A. A. *Bioinformatics* **2001**, *17*(3), 284-285.
- 13. Wells, J. A.; McClendon, C. L. *Nature* **2007**, *450*(7172), 1001-1009.
- 14. Lu, H.; Zhou, Q.; He, J.; Jiang, Z.; Peng, C.; Tong, R.; Shi, J. *Signal. Transduct. Target. Ther.* **2020**, 5(1), 1-23.
- 15. Terrett, N. *MedChemComm* **2013**, *4*(3), 474-475.
- 16. Dougherty, P. G.; Sahni, A.; Pei, D. *Chem. Rev.* **2019**, *119*(17), 10241-10287.
- 17. Bockus, A. T.; McEwen, C. M.; Lokey, R. S. *Curr. ToMed. Chem.* **2013**, 13(7), 821-836.
- 18. Clackson, T.; Wells, J. A. *Trends Biotechnol.* **1994**, 12(5), 173-184.
- 19. Angelini, A.; Cendron, L.; Chen, S.; Touati, J.; Winter, G.; Zanotti, G.; Heinis, C. *ACS Chem. Biol.* **2012**, *7*(5), 817-821.
- 20. Kessler, H. *Angew. Chem. Int. Ed. Engl.* **1982**, *21*(7), 512–523.
- 21. Craik, D. J. *Science* **2006**, *311*(5767), 1563–1564.
- 22. Samanen, J.; Ali, F.; Romoff, T.; Calvo, R.; Sorenson, E.; Vasko, J.; Storer, B.; Berry, D.; Bennett, D.; Strohsacker, M.; Powers, D., *J. Med. Chem.* **1991**, *34*(10), 3114-3125.
- 23. Werle, M.; Bernkop-Schnürch, A. *Amino Acids* **2006**, *30*(4), 351-367.
- 24. White, C. J.; Yudin, A. K. *Nat. Chem.* **2011**, *3*(7), 509-524.
- 25. Gongora-Benitez, M.; Tulla-Puche, J.; Albericio, F. *Chem. Rev.* **2014**, *114*(2), 901-926.
- 26. Huang, Y.; Wiedmann, M. M.; Suga, H. *Chem. Rev.* **2018** , *119*(17), 10360-10391.
- 27. Vinogradov, A. A.; Yin, Y.; Suga, H. *J. Am. Chem. Soc.* **2019** , *141*(10), 4167-4181.
- 28. Bechtler, C.; Lamers, C. *RSC Med. Chem.* **2021**, *12*(8), 1325-1351.
- 29. Fairlie, D. P.; Abbenante, G.; March, D. R. *Curr. Med. Chem.* **1995**, *2*(2), 654-686.
- 30. Fairlie, D. P.; West, M. L.; Wong, A. K. *Curr. Med. Chem.* **1998**, *5*(1), 29-62.
- 31. Tyndall, J. D.; Pfeiffer, B.; Abbenante, G.; Fairlie, D. P. *Chem. Rev.* **2005**, *105*(3), 793- 826.
- 32. Loughlin, W. A.; Tyndall, J. D.; Glenn, M. P.; Fairlie, D. P. *Chem. Rev.* **2004**, *104*(12), 6085-6118.
- 33. Cheng, P. N.; Pham, J. D.; Nowick, J. S. *J. Am. Chem. Soc.* **2013**, *135*(15), 5477-5492.
- 34. Fasan, R.; Dias, R. L.; Moehle, K.; Zerbe, O.; Vrijbloed, J. W.; Obrecht, D.; Robinson, J. A. *Angew. Chem. Int. Ed.* **2004**, *43*(16), 2109-2112.
- 35. Bauer, W.; Briner, U.; Doepfner, W.; Haller, R.; Huguenin, R.; Marbach, P.; Petcher, T.J.; Pless, J. *Life Sci.* **1982**, *31*(11), 1133-1140.
- 36. Shemyakin, M. M.; Ovchinnikov, Y. A.; Ivanov, V. T. *Angew. Chem. Int. Ed. Engl.* **1969**, *8*(7), 492-499.
- 37. Haviv, F.; Fitzpatrick, T.D.; Swenson, R.E.; Nichols, C.J.; Mort, N.A.; Bush, E.N.; Diaz, G.; Bammert, G.; Nguyen, A.; *J. Med. Chem.* **1993**, *36*(3), 363-369.
- 38. Rand, A.C.; Leung, S.S.; Eng, H.; Rotter, C.J.; Sharma, R.; Kalgutkar, A.S.; Zhang, Y.; Varma, M.V.; Farley, K.A.; Khunte, B.; Limberakis, C. 2012. *MedChemComm* **2012**, *3*(10), 1282-1289.
- 39. Faulds, D.; Goa, K. L.; Benfield, P. *Drugs* **1993**, *45*(6), 953-1040.
- 40. Ahlbach, C. L.; Lexa, K. W.; Bockus, A. T.; Chen, V.; Crews, P.; Jacobson, M. P.; Lokey, R. S. *Future Med. Chem.* **2015**, *7*(16), 2121-2130.
- 41. Dougherty, P. G.; Sahni, A.; Pei, D. *Chem. Rev.* **2019**, *119*(17), 10241-10287.
- 42. Biron, E.; Chatterjee, J.; Ovadia, O.; Langenegger, D.; Brueggen, J.; Hoyer, D.; Schmid, H.A.; Jelinek, R.; Gilon, C.; Hoffman, A.; Kessler, H. *Angew. Chem. Int. Ed.* **2008**, *47*(14), 2595-2599.
- 43. Tucker, T.J.; Embrey, M.W.; Alleyne, C.; Amin, R.P.; Bass, A.; Bhatt, B.; Bianchi, E.; Branca, D.; Bueters, T.; Buist, N.; Ha, S.N. *J. Med. Chem.* **2021**, *64*(22),16770-16800.
- 44. Lawson, K. V.; Rose, T. E. Harran, P.G.; *Tetrahedron* **2013**, *69*(36), 7683-7691.
- 45. Lawson, K. V.; Rose, T. E.; Harran, P. G. *Proc. Natl. Acad. Sci.* **2013**, *110*(40), E3753- E3760.
- 46. Rose, T. E.; Lawson, K. V.; Harran, P. G. *Chem. Sci.* **2015**, *6*(4), 2219-2223.
- 47. Rose, T. E.; Curtin, B. H.; Lawson, K. V.; Simon, A.; Houk, K. N.; Harran, P. G., *Chem. Sci.* **2016,** *7* (7), 4158-4166.
- 48. Curtin, B. H., Manoni, F., Park, J., Sisto, L. J., Lam, Y. H., Gravel, M., Roulston, A., Harran, P. G., *J. Org. Chem.*, **2018**, *83*, (6), 3090-3108.
- 49. Sisto, L. J., Harran, P. G., *Tet. Lett.*, **2020**, *61* (24), 151986.
- 50. Sisto, L. J., Harran, P. G., *Tet. Lett.*, **2020**, *61* (24), 152985.

Chapter 2 – Synthesis of atypically fluorinated peptidyl macrocycles via sequential vinylic substitutions

2.1 Introduction

Ipso substitution of halogen to form large rings was first demonstrated during synthetic studies on vancomycin class antibiotics. In those experiments, biarylether motifs were formed by unimolecular S_N Ar reactions involving phenoxide addition to fluoro nitro arenes.¹ More recently, exploiting reactivity of polyfluorinated aromatics, Spokoyny and Pentelute showed perfluorobenzene **2.1** could engage peptides harboring proximal thiols in successive ipso substitutions to form p-dithiotetrafluorophenyl ether bridged macrocycles.² The same group showed decafluorodiphenyl sulfone **2.2** would react with diamine bearing peptides to afford 4,4' diamino octafluorodiphenyl sulfone bridged macrocycles.³⁻⁵

Figure 2.1. Polyfluorinated arylation and vinylation reagents. Octafluorocyclopentene (OFCP) (**2.3**) has the potential to react multiple times as an electrophile, at both primary (blue) and secondary (red) positions.

In our own studies on constrained peptidomimetics, we were considering ways to build bridged bicyclic structures via transannulating existing macrocycles.⁶ The ease of fluoride ipso substitutions was appealing. However, relative to reagents **2.1** and **2.2**, we sought broader nucleophile participation and product geometries more compatible with smaller polyheterocyclic structures. Rather than aromatic substitutions, we choose to examine vinylic substitutions of fluoride from octafluorocyclopentene (OFCP, 2.3, Figure 2.1).⁷⁻¹⁶ Simple phenols⁷, amines¹⁰, mercaptans^{14,15} and alcohols¹⁴⁻¹⁶ were known to displace fluoride from this commercial reagent, and the extent of substitution could be varied. Herein we report that successive, dissymmetric substitutions¹⁷ of fluoride on OFCP by functionalized peptides occurs readily, and provides flexible means to generate constrained fluorinated macrocycles.

2.2 Results and Discussion

To model ring closure steps in projected macrocyclizations involving amino acid side chains, we first converted OFCP to mono phenoxy derivative 2.5 (0.9 eq PhOH, Et₃N, 74%). This molecule was found to react $(Cs_2CO_3, DMF, 0\degree C)$ individually with tyrosine, cysteine and histidine monomers to afford hexafluorocyclopentenylated adducts **2.8**, **2.9** and **2.10**, respectively, in high yields (Scheme 2.1). The reaction to give **2.9** was near instantaneous (< 15 sec) and those affording 2.8 and 2.10 were complete within minutes (see Appendix). L-Cbz-Ser-NH₂ reacted more slowly, and further. The carboxamide in initially formed substitution product **2.11c** internally displaced an allylic fluoride via 6-exo trig cyclization to afford spirooxazinones **2.11a**/**b** (d.r = 1:1). Products **2.8**-**2.10** and **2.11a**/**b** were stable to handling and chromatography and showed diagnostic ¹⁹F NMR spectra consistent with their assigned structures (see Appendix).

Scheme 2.1. Bimolecular amino acid derivatizations using OFCP. (a) OFCP, PhOH, Et₃N, DMF, 0 °C, 1 h, 74%. (b) OFCP, 3-butyn-1-ol, DMF, 0 °C to rt, 3 h, 57%. (c) i) OFCP, $HO(CH_2)_3CO_2Bn$, Et₃N, DMF, 0 °C to rt, 1 h, 61%. ii) H2, 10% Pd/C, EtOH. iii) HOSu, EDC·HCl, DCM, rt, 3 h, 83% (2 steps).

To examine if single vinyl fluoride substitutions on OFCP could be used to form peptidyl macrocycles, we prepared butynyloxy (**2.6**) and carboxypropyloxy derivatives (**2.7**) of OFCP as shown in Scheme 2.1. Both structures were stable oils that could be purified by chromatography. Starting with the tyramide derivative of L-AF (**2.12**, Scheme 2.2), N-acylation with 4-pentynoic acid and exposure of the product to 2.6 in the presence of Cs₂CO₃ (DMF, rt) gave diyne 2.13 as a white solid. Subjecting this material to Verniest's conditions for Glaser-Hay coupling^{18,19} afforded aryloxy hexafluorocyclopentene containing macrocyclic diyne **2.14**. We then tested if vinylic substitution could be made the ring closure event. Acylation of **2.12** with **2.7** and treating the product with Cs₂CO₃ (1.5 eq, DMF, 0 °C -> rt) gave fluorinated macrocycle **2.16** in good overall yield. To our knowledge, the ring systems in **2.14** and **2.16** have not been described previously.

Scheme 2.2. Stepwise synthesis of macrocycles using vinylic substitutions. (a) 4-pentenoic acid, EDC·HCl, HOBt, iPr2EtN, DMF, rt, 3 h. (b) **2.6**, Cs2CO3, DMF, rt, 3 h, 53%. (c) Cu(OAc)2·H2O, NiCl2, Et3N, pyridine, EtOH, 50 °C, 3 h, 52%. (d) **2.7**, iPr2EtN, DMF, 0 °C, 1 h. (e) Cs2CO3, DMF, 10 mM, 0 °C to rt, 2 h, 47% from **2.12**.

We next tested if linear peptide derivatives would react with OFCP itself en route to macrocycles. When a DMF solution of dipeptidyl alcohol **2.17** was treated with OFCP in the presence of Et3N, ether conjugate **2.18** was formed. An N-terminal histidine residue was then

appended to the molecule, during which the alkoxyheptafluorocyclopentene motif was stable (Scheme 2.3). Treatment of product 2.19 with Cs₂CO₃ in DMF selectively formed imidazole

Table 2.1. Direct macrocyclizations mediated by OFCP. Reagents and conditions: (a) OFCP, TEA, DMF, 0 °C, 2 h. (b) OFCP (5 eq), TEA, DMF, 0 °C, 30 min, then Cs_2CO_3 , 0 °C, 30 min - 1 h. (c) OFCP, Cs $_2CO_3$, rt, 1h. **―** HMBC correlations.

containing macrocycle **2.20**. The primary amine was unaffected by the reaction. N1-C1 connectivity in **2.20** was assigned based on C1 showing an HMBC correlation to C35H, but not C34H (compare to **2.26**, Table 2.1). Taken together, data from biomolecular models and incremental ring syntheses suggested linear peptides harboring combinations of cysteine, tyrosine, histidine and serine residues could participate in direct macrocyclizations mediated by OFCP. This proved facile for sequences containing two cysteine residues. Stirring Ac-CFAC-NH² (**2.21**) with 1.5 eq OFCP in the presence of Et3N (DMF, 0 ˚C) provided macrocycle **2.22** in 51% isolated yield. In mixed nucleophile systems, a two-stage, one-pot procedure proved most effective. Tetrapeptide **2.23** contained cysteine and tyrosine residues. Exposure of **2.23** to 5.0 eq OFCP in the presence of Et_3N (DMF, 0 °C) selectively derivatized the thiol within minutes. Subsequent removal of excess OFCP (b.p. 27 ˚C, 1 atm) via rotary evaporation and adding solid

Cs2CO³ to the solution gave unique cyclophane **2.24**. The same protocol executed on Ac-YFAH-NH² engaged the tyrosine and histidine side chains to afford macrocycle **2.26**. The histidine reacted with OFCP first, vinylating its distal nitrogen, which after tyrosine capture produced a regioisomeric variant of the imidazole linkage observed in macrocycle **2.20** (Scheme 2.3). HMBC correlations were observed between C5 and both imidazole protons in **2.26**. Table 1 lists additional fluorinated macrocycles prepared in the above manner, each representing a new ring system. Side chain protecting groups were not required. Reactions proceeded rapidly at 0 °C and no provisions were made to exclude oxygen or atmospheric moisture. In the three to five residue sequences examined, unprotected Trp, Asn, Gln and Thr side chains did not participate or interfere with fluoride substitution reactions. Macrocyclic products were stable to handling, isolation and extended storage²⁰ except for serine linked macrocycle 2.28. This molecule gradually ringed opened via net hydrolysis upon standing in solution at room temperature.²¹

Scheme 2.3 Free amine embedded OFCP macrocyclic products. (a) OFCP, Et₃N, DMF, rt, 3 h, 40%. (b) i) TFA, DCM, rt, 1 h. ii) Boc-His-OH, EDC·HCl, K-Oxyma, iPr2EtN, DMF, rt, 3 h. iii) TFA, DCM, rt, 2 h, 45% from **2.18**. (c) Cs2CO3, DMF, 0 °C to rt, 2h, 39%. **―** HMBC correlation.

In free base form, the amino side chains of lysine and ornithine would react with OFCP, but the resultant substitution products were generally not stable. In contrast, OFCP reacted with free N-terminal cysteine residues to form isolable hexafluoro cyclopentathiazine derivatives (Scheme 2.4). The reactions examined were complete within minutes. The ability to rapidly hexafluorocylopentenylate the N-terminus of peptides and peptidomimetics may have utility, for example, in probe syntheses for positron emission tomography.²²

Scheme 2.4. Hexafluoro cyclopentathiazinone derivatives formed from N-terminal capture of OFCP. Reagents and conditions: (a) OFCP, TEA, DMF (10 mM), 0 °C, 30 min. **―** HMBC correlations.

Substituting the two vinyl positions on OFCP with heteroatoms gave products potentially susceptible to further internal reactions, as was observed during conversion of **2.11c** to **2.11a**/**b** (Scheme 2.1). Experimenting initially with **2.22**, we found that treatment with PhOH/Cs₂CO₃ in DMF gave spirocyclic thiazinone-containing phenyl ether **2.41**, wherein two additional fluorine atoms had been displaced from the cyclopentene core. This result implied that phenoxide could be captured intramolecularly. In fact, we discovered that hexapeptides of consensus sequence YXCXXC would undergo remarkable polysubstitution reactions with OFCP. Stirring Ac-YACFAC- $NH₂$ with 1.5 eq OFCP in DMF at 0 °C rapidly formed a disubstitution adduct (monitored by LC/MS). Excess $Cs₂CO₃$ was then added and the solution was warmed to rt and stirred for 4 h.

Scheme 2.5. Initial discovery of cascade polysubstitutions of OFCP. Reagents and conditions: (a) OFCP, TEA, DMF (5 mM), 0 °C, 30 min, then Cs₂CO₃, rt, 6 h, 21% (unoptimized). (b) OFCP, TEA, DMF (5 mM), 0 °C, 30 min, then Cs₂CO₃, rt, 4 h, 28%. (c) OFCP, TEA, DMF (5 mM), 0 °C, 10 min, then Cs₂CO₃, rt, 6 h, 41%. ²³ **―** HMBC correlations.

This gave macrobicyclic structure **2.43**, wherein the peptide had displaced four fluorine atoms from OFCP, forming three new rings (Scheme 2.5) in the process. The outcome was rationalized in terms of the C-terminal carboxamide cyclizing at C4 in a first formed cyclic bis-thioether, generating a new vinyl fluoride that subsequently captured the pendant phenoxide. Stereochemistry at the newly formed spiro center (C4, dr = 10:1) was assigned by analogy to **2.47** (vide infra). Similar results were obtained starting with Ac-YNCTFC-NH₂ (2.44) wherein complex polycycle **2.47** was isolated in good yield. In that case, quenching the reaction with AcOH after 2 h provided a mixture of **2.47** and two intermediate structures (Figure 2.2A). Data obtained on purified samples of those compounds (Figure 2.2B) suggested they were cyclic bis-thioether **2.45** and spirocycle **2.46**, fully consistent with the progression of events proposed above. As expected, re-exposing 2.45 and 2.46 to Cs₂CO₃ in DMF produced 2.47. Scrambled variants of 2.44, namely Ac-CNYTFC-NH2, Ac-NYCAFC-NH2, and Ac-YCNTFC-NH2, each reacted smoothly with OFCP, but none substituted more than three of its fluorine atoms (see SI, Figure S7). The YXCXXC consensus was required to form macrobicyclic structures via four substitutions.

Scheme 2.6. NMR investigations supporting polycyclizations. a) Analytical HPLC trace (C18, 50 \rightarrow 100% MeCN + 0.1% TFA) of mixture obtained upon quenching (AcOH) the reaction of 2.44 with Cs2CO₃ after 2 h. B) Diagnostic ¹H-NMR signals indicative of structures of **2.45**, **2.46** and **2.47**, having m/z = 963.3, 943.3, and 923.2 respectively. See SI for details.

After much effort, the structure of **2.47** was firmly established by X-ray analysis of a single crystal grown using the vapor diffusion method (DCM into DMSO/PhH at 25 ˚C). In the solid state **2.47** exists in a single conformation. Large cavities in the unit cell are occupied by DMSO with two water molecules bound along the polar face of the structure (Figure 2.3).

2.3 Conclusions

The above reaction cascades convert linear, unprotected peptides into conformationally defined polyheterocycles having two turn surfaces scaffolded by a tetrafluorocyclopentene core. Constrained peptidomimetics of this kind have not been studied previously and may prove useful in medicinal chemistry programs. Experiments along these lines are ongoing, as are evaluations of the pharmacological properties of polyfluorocyclopentenylated macrocycles in general. The ease and simplicity of OFCP based polycyclizations should greatly facilitate access to complex fluorinated macrocycles.

2.4 References

- 51. D. A. Evans, M. R. Wood, B. W. Trotter, T. I. Richardson, J. C. Barrow, J. L. Katz, *Angew. Chem., Int. Ed.* **1998**, *37*, 2700-2704; *Angew. Chem*.**1998**,*110*, 2864–2864.
- 52. A. M. Spokoyny, Y. Zou, J. J. Ling, H. Yu, Y.-S. Lin, B. L. Pentelute, *J. Am. Chem. Soc.* **2013**, *135*, 5946-5949.
- 53. Y. Zou, A. M. Spokoyny, C. Zhang, M. D Simon, H. Yu, Y.-S. Lin, B. L. Pentelute, *Org. Biomol. Chem.* **2014**, *12*, 566-573. See also; S. Kalhor-Monfared, M. R. Jafari, J. T. Patterson, P. I. Kitov, J. J. Dwyer, J. M. Nuss, R. Derda, *Chem. Sci*., **2016**, *7*, 3785- 3790.
- 54. G. Lautrette, F. Touti, H. G. Lee, P. Dai, B. L. Pentelute, *J. Am. Chem. Soc.* **2016**, *138*, 8340-8343.
- 55. Cobb and co-workers have tagged the peptides with a perfluoroheteroaromatic. D. Gimenez, C. A. Mooney, A. Dose, G. Sandford, C. R. Coxon, S. L. Cobb, *Org. Biomol. Chem*. **2017**, 15, 4086-4095.
- 56. B. H. Curtin, F. Manoni, J. Park, L. J. Sisto, Y. Lam, M. Gravel, A. Roulston, P. G. Harran, *J. Org. Chem*., **2018**, *83*, 3090-3108.
- 57. P. J. Card, B. E. Smart, *J. Org. Chem.* **1980**, *45*, 4429-4432.
- 58. A. E. Bayliff, M. R. Bryce, R. D. Chambers, *J. Chem. Soc., Perkin Trans. 1* **1987**, 763- 767.
- 59. M. Negele, A. Marhold, D. Beilefeldt, T. Himmler, U.S. Patent 4,990,633, Feb 5, 1991.
- 60. S. Garg, B. Twamley, Z. Zeng, J. M. Shreeve, *Chem. - Eur. J.* **2009**, 15, 10554-10562.
- 61. J.-M. Cracowski, B. Sharma, D. K. Brown, K. Christensen, B. R. Lund, D. W. Smith, Jr, *Macromolecules* **2012**, 45, 766-771.
- 62. J. Wu, Y. Xi, G. T. McCandless, Y. Xie, R. Menon, Y. Patel, D. J. Yang, S. T. Iacono, B. M. Novak, *Macromolecules* **2015**, *48*, 6087- 6095.
- 63. J. Wu, Y. Xi, G. T. McCandless, O. V. Kulikov, R. Menon, B. M. Novak, *RSC Adv*. **2015**, 75547-75554.
- 64. S. Iacono, A. R. Jennings, U.S. Patent 9,695,203 Jul 4, 2017.
- 65. S. Iacono, A. R. Jennings, U.S. Pat. Appl. 20160280725, Sep 29, 2016.
- 66. Recently, Kubota and co-workers reported macrocyclization using OFCP and diethylene glycol for preparing phase transfer catalyst: H. Fukumoto, S. Nakashima, T. Agou, T Kubota, Jpn. Kokai Tokkyo Koho JP 2018150247 Sep 27 2018.
- 67. Shreeve and co-workers reported cyclization from symmetric disubstituted perfluorocyclopentene: Q.-C. Mir, J. M. Shreeve, *J. Fluorine Chem.* **1994**, 68, 269-275.
- 68. S. Verlinden, N. Geudens, J. C. Martins, D. Tourwe, S. Ballet, G. Verniest, *Org. Biomol. Chem.* **2015**, *13*, 9398- 9404.
- 69. A. P. Silvestri, P. A. Cistrone, P. E. Dawson, *Angew. Chem., Int. Ed.* **2017**, *56*, 10438- 10442; *Angew. Chem.,* **2017**, *129*, 10574-10578.
- 70. For example, macrocycle **2.22** was stable in 1M aq. HCl / MeOH (1:9) solution for 24 hours; whereas it decomposed within 1 hr in basic media (1M aq. NaOH in MeOH (1:9)). See Figures S5 and S6. In general, reactions promoted with $Cs₂CO₃$ were quenched with AcOH to avoid degradations during aqueous workup.
- 71. Macrocycle **2.28** was stable as a solid, but upon standing in DMSO-*d*6, gradually (several weeks) ring opened via hydrolysis (+ 2H2O –HF net). Related structure **2.48**, having a smaller macrocycle size, hydrolyzed more rapidly. See Appendix for details.

- 72. F. Qing, X. Jia, X. Zhou, H. Quan, Pat. Appl. CN105439806A, Nov 24, 2015.
- 73. Adding up to 20 vol % water to the reaction of 2.44 with OFCP in the presence of Et₃N had no discernible impact on the formation of **2.45** (see SI, Figure S4).
- 74. CCDC 1946197 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.

Chapter 3 – Studies on shape and property altering polysubstitution cascades

3.1 Introduction

Reaction processes that form multiple covalent bonds while creating new rings and stereocenters are inherently valuable in organic synthesis. They advance an overarching goal of the field, which is to enable syntheses of value-added products from abundant raw materials in the most direct manner possible. Reaction cascades have been used effectively for the synthesis of diverse carbocyclic and heterocyclic ring systems using both catalytic and stoichiometric methods.^{1,2} The typical aim is to build molecular complexity quickly. However, multi bond-forming processes can also manipulate the form and properties of pre-assembled structures. For example, we have developed reagents that can engage small peptides and related oligomers in successive ring forming reactions to afford polycyclic derivatives.³ In the example shown in Fig. 3.1B,³ three successive operations imbed the core of synthetic reagent **3.1** into unprotected WWY. Four new bonds (red) are formed between the oligomer and the scaffolding reagent, resulting in four new rings and three new stereocenters. Other methods to cyclize unprotected peptides using native functional groups (e.g. Fig. 3.1A)⁴⁻⁶ achieve less bonding with the scaffold and install fewer conformational constraints. Using **3.1** and related reagents, it is possible to convert machine made oligomers systematically into stable composite macrocycles having diverse shapes and improved pharmacological properties.

The scaffolding imparted by residual **3.1** (colored blue in **3.2**) is hydrophobic. It was of interest to develop methods that could install scaffolding having polar elements and the potential for transannular hydrogen bonding. We recently discovered polysubstitution cascades that achieve this outcome (Figure 3.1C).⁷ The reactions modify linear, unprotected peptides in a single flask at room temperature. They require no catalysts or heavy metals and generate a wealth of previously unknown heterocyclic ring systems with tunable properties.

A. Cyclizations Using Native Peptide Functionality

C. This Work: Relative Rate Controlled Polysubstitution Cascades

Figure 3.1. Background and rationale. **A**) Halo acetamides, perfluoroarenes and 1,3,5-tris-halomethyl benzenes engage nucleophilic peptide side chains in ring forming reactions. **B**) Reagent **3.1** sheds polar functionality while reacting with peptides to form composites with hydrophobic scaffolding (blue) bonded multiple times (red) to the oligomer. **C**) We have discovered that commercial octafluorocyclopentene can react with unprotected peptides in polysubstitution cascades that form multiple rings in a single flask.

Electron deficient aryl fluorides participate in nucleophilic ipso substitution (SNAr) reactions. Sanger's method for N-terminal peptide sequencing using 2,4-dinitrofluoro benzene is a seminal example.⁸ More heavily fluorinated aromatics such as hexafluorobenzene can engage two nucleophiles successively at ring positions 1 and 4 (Figure 3.1A). The reaction is efficient with sulfur nucleophiles and has been used to generate macrocyclic peptides via cysteine 'stapling'. Poly substitutions become possible when using perfluorinated cycloalkenes. Commercial octafluorocyclopentene (OFCP) reacts with nucleophiles rapidly at both its vinyl positions, and

subsequently at its allylic positions to afford adducts having up to six fluorine atoms replaced. Previously we showed peptide side chain nucleophile to form fluorocyclopentenylated macrocycles via cross-linked cysteine, tyrosine, histidine, and serine residues in various combinations. Here we describe OFCP initiated reaction cascades that convert initially formed peptidyl macrocycles into conformationally rigidified polycyclic derivatives via additional internal fluoride displacements. We show that spirothiazinone intermediates can be intercepted to afford diverse functionalization and conjugation products. We examine the molecular properties of products and computationally evaluate if fluorospirocyclic scaffolding can create new structural mimics of major loop types observed in the Protein Data Bank.

3.2 Results and Discussion

Linear peptides having three proximal nucleophilic residues react with OFCP (25 °C, Et₃N, DMF) according to the progression shown in Figure 3.2. Two successive vinylic substitutions occur rapidly (krel for Ser:Tyr:His:Cys ~ 1:30:45:1000) to afford macrocycles **3.6**. Excess OFCP is removed in vacuo and the residue is redissolved in a DMF/THF solution containing 3 equiv. of KOSiMe₃. This initiates spirocyclization via Sn2' displacement of a third fluorine atom (when Nu3 is a C-terminal carboxamide) to afford a new vinyl fluoride that is captured by a fourth competent nucleophile to give polycycles **3.8**. Compounds **3.8**, are stable, soluble and purified using standard chromatographic techniques. AcYNCTCF-NH $_2$ reacts with OFCP at its two cysteine residues within minutes at 25 °C. Subsequent KOSiMe₃ treatment installs the spirothiazinone motif while generating a new vinyl fluoride that then captures the N-terminal tyrosine residue to afford **9**. By initiating reaction cascades at a C-terminal cysteine amide, it is possible to synthesize a range of macrobyclic structures that are bridged by sulfide and imidazole linkages. There is flexibility in ring size on either side of the bridging residue and, in the case of histidine derivative **3.13**, bridge position epimer **3.14** can be prepared readily. Diastereoselectivity at the newly

Cys, His, and Tyr Mediated Macrobicyclizations

(where $K_{rel} \sim$ mercaptan >> imidazole \sim phenol)

Figure 3.2. Direct one-pot polycyclizations of peptides with OFCP

formed spiro center is high (>10:1) across the series. Products have well resolved ¹H NMR spectra

at ambient temperature. By repositioning the cysteinyl amide off of a glutamate or aspartate side

chain, it is possible to initiate alternate macrobicyclization cascades. For example, branched peptides AcD(C)VHSAC-NH² AcYGAE(C)H-NH² affords the striking double looped polycycles **3.16** and **3.17**, respectively when reacted successively with OFCP and KOSiMe₃. To our knowledge, each of the structural types shown in Figure 3.2 are without precedent in the literature.

Figure 3.3. Examining spirocyclic variations and coupling partners at vinyl fluoride position. Reagents and conditions: a) OFCP (1.5 equiv), Et₃N (2.5 equiv), DMF (5 mM), 0 °C, 30 min; conc. then KOSiMe₃ (2 equiv), 1:4 DMF/THF (5 mM), 0 °C -> RT, 1 h, (>10:1 d.r.). b) beta-D-Thioglucose sodium salt (1 equiv), DMF (50 mM), 0 °C, 1 h, >95%. c) Coniferyl Carbonate **3.27** (2 equiv), KOSiMe₃ (2 equiv), DMF (50 mM), 0 °C -> RT, 12 h, 53%. d) NaN³ (1 equiv), DMF (50 mM) 0 °C, 1 h, >95%. e) H2, Pd/C (10 wt %), EtOH, 35 °C, 12 h, 87%. f) Propargyl Alcohol (1 equiv), Phenylenediamine (15 mol %), Sodium ascorbate (10 mol%), CuSO4•5H2O (5 mol%), 2:3 H2O:tBuOH (0.2 M), 12 h, 75%.

It is also possible to isolate and characterize vinyl fluorides. Treatments of AcCWSC-NH₂ with OFCP followed by KOSiMe₃ affords spiro tricyclic compound **3.18**. Under identical conditions, the same sequence containing D-cysteine affords epimeric macrocycle **3.20**, while its homo cysteine variant affords spiro thiazepinone **3.19**. Smaller analogs are also accessible. Omitting the tryptophan residue results in the formation of spiro tricyclic substance **3.21**.

The vinyl fluoride in **3.18** can be intercepted in bimolecular reactions to give a variety of substituted and homologated derivatives. It reacts with commercial β-D-thioglucose sodium salt within minutes at 25 °C to afford unprotected glycoconjugate **3.22** in near quantitative yield. Variations in thioglycoside and peptide sequence could be used to generate a new class of glycosylated macrocycles. Compound **3.18** also reacts with a coniferyl alcohol derivative to afford **3.23**. The cinnamyl carbonate in **3.23** provides means to form an additional large ring via electrophilic capture of amines, carboxylates, imidazoles and pi basic aromatic yield. Azidation of **3.18** with NaN₃ in DMF provides vinyl azide **3.24** in >95% yield This molecule participates in Sharpless/Huisgen 'click' cycloadditions with terminal alkynes. Stirring **3.24** with 1 equiv. of propargyl alcohol in the presence of a phenylenediamine ligated Cu^{II} catalyst affords triazole 3.26 in 75% yield. Azide **3.24** can also be hydrogenated to afford β-thio enamine **3.25**. We note also the possibility that electrophiles of type **3.18** could be used to form novel antibody / macrocycle conjugates via reactions at solvent exposed cysteine or tyrosine residues.

Capture of spirocyclic vinyl fluorides with insert molecules allows syntheses of macrobicyclic structures from peptide sequences that would otherwise give mono macrocyclic products (Figure 3.4). These insert molecules can be coupled to the vinyl fluoride position of spirocyclic intermediates in an efficient one-pot manner (**3.28**). Various linker systems can be used to further expand the scope of polycycles by interacting with side chain residues that wouldn't normally react with OFCP directly either based on its nucleophilicity or strained configuration (**3.29**). For example, thiazole **3.30** rapidly reacts at the vinyl fluoride position the provide **3.31**. Ionization of the alcohol under acidic conditions gives Friedel-Crafts products (see **3.32** and **3.33)**, albeit after long reaction times while heated (12 h, 80 °C). Direct polycyclizations of these peptides (AcYACFAC-NH₂ and ACYCAC-NH₂) with OFCP would not be possible due to constrained nature required for the final ring closure. Reminiscent of our template work discussed earlier, we were inspired to develop a similar system arriving at the phenolic insert **3.27**. This

molecule couples slowly but effectively to provide stable and isolable carbonate linkers. Thus far, we have demonstrated their utility with tyrosine (**3.35**) and tryptophan (**3.36**) residues, most

Figure 3.4. Polycyclizations via various linker insert methods. Reagents and conditions: a) OFCP (1.5 equiv), Et₃N (3 equiv), DMF (50 mM), 0 °C, 30 min; conc. then KOSiMe₃ (3.5 equiv), thiazole 3.30 (1.1 equiv), 1:4 DMF/THF (50 mM), 0 °C -> RT b) 5% TFA, MeNO₂ (5 mM), 80 °C, 12 h. c) OFCP (1.5 equiv), Et3N (2.5 equiv), DMF (5 mM), 0 °C, 30 min; conc. then KOSiMe³ (4 equiv), **3.27** (2 equiv), 1:4 DMF/THF (50 mM), 0 °C -> RT, 12 h. d) ScOTf³ (1 equiv), MeNO² (5 mM), RT, 2 h. e) OFCP (1.5 equiv), Et3N (3 equiv), DMF (10 mM), 0 °C, 30 min; conc. then Cs_2CO_3 (4 equiv), Cysteamine (1.1 equiv), 0 °C, 1:4 DMF/THF (5 mM) f) NCS (1.5 equiv), 2:8 MeCN:NaOAc/AcOH (aq.) (1 mM), RT 45 min g) T3P (1.5 equiv), $iPr₂NEt$ (3.5 equiv), DMF (5mM), 0 °C.

importantly. Under lewis acidic conditions, tryptophan reacts rapidly to give three major regioisomers with **3.35** being the major product isolated. We've also recently obtained macrocycle products with carboxylic acid functionality. This was made possible with the help of masking Asn
or Gln residues as cyanosulfurylides (**3.37**), chemistry originally developed by the Bode group as a means to prevent aspartimide formation during peptide synthesis.⁹ This has provided the potential for macrolactone and macrolactam polycyclic products. For example, AcD*GSFAC-NH² can be reacted with OFCP followed by capture of the vinyl fluoride position using cysteamine and subsequently revealing the carboxylic acid using NCS as seen in **3.38**. Macrolactamization can then be achieved by subjecting **3.38** with T3P coupling conditions to give **3.39** in 30% yield from the linear peptide. We are currently working to expand the scope of polycycles with these methods including varying both the participating and spectating functionality of the peptide backbone and alternative insert molecules.

B. Ketene Capture

C. Benzyne Capture

Figure 3.5. Studies on several linker systems with alternative modes of capture.

Considering other linker system types, we continued to focus our strategies on methods which were both general and amenable to conditions compatible with the polar functionality present within our peptide substrates. We have since begun to examine three other systems that we believe have the potential to tolerate these guidelines (Figure 3.5). After coupling of hydroquinone to the vinyl fluoride position, electrochemical oxidation to the quinone intermediate may promote nucleophilic capture of neighboring amino acid residues (Figure 3.5A). However, one-pot methods to couple hydroquinone to form the acyclic starting material proved difficult. For example, when reacting AcWCASC-NH² with standard macrocyclization conditions (1.5 equiv. OFCP, 3 equiv. Et₃N) followed by hydroquinone and $Cs₂CO₃$, we observed slow reaction times (overnight) and incomplete conversion (Figure 3.6A). Reverse-phase HPLC purification of these quenched reaction mixtures was required due to increased polarity from the free phenol but did not lead to product isolation possibly due to instability. 4-methoxyphenol was then used to probe the reactivity and stability compared to hydroquinone. At small scales, we found its reactivity comparable to hydroxyquinone with incomplete conversion over long reaction times. Its increased lipophilic nature may allow normal flash column chromatography isolation techniques, which we will be examining in future experiments.

Figure 3.6. Attempts to synthesize hydroquinone incorporated structures and benzodioxinone substrates. De Brabander, and coworkers have developed a method to synthesize *ortho*-substituted salicylic esters and amides through the photolysis of *ortho*-substituted benzodioxinones (Figure 3.6B). ¹⁰ Photochemical fragmentation of benzodioxinones to quinoketenes allowed the capture

alcohols, phenols, amines, and anilines using mild reaction conditions (neutral, RT). This general method could prove incredibly useful if successfully incorporated as a linker molecule (Figure 3.5B). The reported method includes the use of 5-hydroxybenzodioxinone, however we chose to synthesize the unreported 6-hydroxy regiosiomer, from 2,5-dihydroxy benzoic acid and benzophenone, to avoid steric complications during macrocyclizations. Unfortunately, numerous attempts to access this ketene precursor were unsuccessful due to hydrolytic instability. Future attempts will focus on synthesizing regioisomeric analogs of this substrate and their potential as a method to form macrolactam/lactone products.

Figure 3.7. Our original and optimized routes to benzyne precursor substrate **3.42**.

We believe benzyne-type intermediates have the potential to engage a broad range of amino acid residues (Figure 3.5C). It is well known that these intermediates can capture a variety of nucleophiles including alcohols, amines, carboxylates, carboxamides, and others.¹¹ With this substrate our goal is to develop methods towards direct synthesis of ansa -bridged macrobicyclic structures from unprotected peptide precursors. Silyl aryl triflates are one of the most common benzyne intermediate precursors known, so we developed several routes to phenoxy silyl triflate **3.43** (Figure 3.7). Our initial route begins with the monobenzylation of resorcinol followed by monobromination to give **3.41** (Figure 3.7A). Initial silylation then lithiation promotes a Brooke-like rearrangement to ultimately capture Tf_2O from the resulting phenoxide. Finally, Raney Ni hydrogenolysis of **3.42** provided the desired substrate **3.43**. We later adopted a more efficient strategy through a one-pot method from 4-bromoresorcinol.¹²We found both **3.42** and **3.43** to be highly unstable, especially when neat for short periods of time, making both isolation and subsequent coupling reactions with this substrate very challenging. Loss of the silyl group was found to be the major form of decomposition as observed by NMR.

A. Benzyne precursor couplings and attempts at activation

Figure 3.8. Initial challenges when studying benzyne precursor substrate reactivity and stable alternatives. Couplings of **3.43** to the vinyl fluoride position of **3.45** was found to be sluggish, but most importantly unstable under these conditions (Figure 3.8). We found the presence of DMF under basic conditions promoted decomposition. DMF is necessary for the solubility of these substrates, leading to solvent mixtures of 10 – 20% DMF/THF to be ideal albeit with modest yield. Regardless, after numerous attempts to activate the silyl aryl triflate of **3.46** under various conditions we arrived at the same outcome with the major product being desilylation. In our efforts to find a more stable

alternative, we found that the Daugulis group has used silyl aryl bromides towards aryne reactivity, however with most of their examples using dimethyl silanes rather than the traditional TMS group.¹³ We synthesized the silyl aryl bromide **3.48** as an alternative substrate in two steps from 3-bromophenol (Figure 3.8B). We found this substrate to be considerably more stable compared to the previous silyl aryl triflate in basic conditions using either DMF or THF for long reaction times (Figure 3.8C). This silyl aryl bromide coupled to compound **3.45** with modest yields (41 – 56%)

Table A

	Entry Fluoride Source Temperature Time			Solvent	Result
	CsF	RT	6 h	THF	N.R.
2	CsF	50 °C	12h	THF	N.R.
3	TBAF	RT	6 h	THF	Partial desilylation
4	TMAF	RT	6 h	THF	N.R.
5	TMAF	40 °C	4 h	THF	Partial desilylation
6	TMAF	40 °C			4 h THF/MeCN Complete desilylation
7	CsF, AgF	RT		7 h THF/MeCN	Partial desilylation

Table B

Table 3.1. Silyl aryl bromide coupling and activation conditions attempted.

with no observed decomposition (Table 3.1A). Various conditions were used to examine benzyne activation which is outlined in Table 3.1B. CsF led to no reaction even at elevated temperatures over 12 h. TBAF and TMAF resulted in desilylation in varying amounts depending on the reaction

temperature. Once again, this substrate demonstrated instability in a similar manner to other cases, however mostly under forcing conditions in this case. The Harrity group reported a similar issue of protodesilylation when attempting activation of silyl aryl iodides for benzyne reactivity.¹⁴

A. OFCP Model system

Figure 3.9. Examining model systems for reactivity and considering future directions to promote activation over protodesilylation.

They found that by using AgF as an additive in the presence of CsF, ratios of protodesilylation to elimination were reduced. In our case, the use of AgF as an additive did not reduce undesired decomposition. We further tested this substrate in a simpler model system and continued to observe decomposition in the presence of various known coupling partners (Figure 3.9A). Finally as a control we successfully replicated known reaction of **3.51** to give regioisomeric products **3.52** and **3.53** using CsF and pyrrolidine (Figure 3.9B). Considering the various instances of protodesilylation we experienced, there are several approaches we could still examine (Figure 3.9C). The Daugulis and Harrity groups both reported successful benzyne reactivity with the use of either dimethylsilane or iodide functionality, respectively. Also, incorporating a methylene unit to the substrate as the benzyl alcohol/thiol analog may alter its electronic properties in a way that reduces its propensity to decompose readily. Future experiments should focus on these substrates to further probe reactivity.

As a compound class, cyclic peptides and peptidomimetics are often developed to target extracellular proteins because their passive membrane permeability is limited. Developing strategies to improve membrane permeability is critical for the targeting of intracellular PPIs. There has been a massive effort to understand factors dictating passive permeability in hopes that cell permeant analogs could be designed.¹⁵ The fluorospiroheterocycle-scaffolded structures described here have not previously been evaluated for passive permeability in any format. A 42 compound library including OFCP macrocycles and polycycles with varying characteristics was evaluated for permeability through a parallel artificial membrane permeability assay (PAMPA) kit. Select examples from this library are shown in Figure 3.6 (See Appendix for full data set). More than half of this set exhibited permeability greater than the low permeability standard theophylline. Interestingly, three compounds performed better than the high permeability standard, which included two polycycles with structural characteristics not typically associated with permeable compounds. Compared to compound **3.13**, its epimeric analog (**3.14**) displayed a notably greater permeability rate. This may suggest significant conformational differences wherein possible intramolecular hydrogen bonding (IMHB) present in **3.14** leads to the observed differences in outcomes between the epimers. A computational model of **3.14** was generated demonstrating this possibility with IMHB interactions from the bridging histidine and the peptide backbone. Overall, these results are promising for our OFCP-embedded macrocycles and polycycles and we are working to further evaluate additional analogs of similar structure types.

Figure 3.10. PAMPA permeability of select structures. Average permeability rate from assay performed in duplicate; error bars: standard deviation (see Appendix for more information). Gray = low (theophylline), medium (diclofenac), and high (chloramphenicol) permeability standards. (a) Structure was determined via geometry minimization of **3.14** carried out on Schrödinger's Maestro Macromodel® using AMBER* force field in water as the solvent. Hydrogens bonds were determined via identification of polar contact surface area using literature reported donor-acceptor distances and bond angle ranges.

3.3 Conclusions

Relative rate controlled polysubstitution cascades using perfluorocycloalkenes is a new, simple means to rapidly generate fluorinated composite macrocycles. Cyclic peptides and peptidomimetics serve as leads in drug discovery, as probe compounds in chemical biology and as building blocks in materials science. This application describes synthetic tools to generate a previously unknown class of fluorinated, conformationally constrained composite peptidomimetics also enabling the study of their novel stability, reactivity, and properties. In the future, this methods platform will enable us to test whether fluorospirocyclic scaffolding can create membrane permeant structural mimics of the major loop types observed in the Protein Data Bank.¹⁶

3.4 References

- 1. Nicolaou, K. C.; Chen, J. S., *Chem. Soc. Rev.* **2009,** *38*(11), 2993-3009.
- 2. Tietze, L. F., *Chem. Rev.* **1996,** *96*(1), 115-136.
- 3. Rose, T. E.; Curtin, B. H.; Lawson, K. V.; Simon, A.; Houk, K. N.; Harran, P. G., *Chem. Sci.* **2016,** *7*(7), 4158-4166.
- 4. Spokoyny, A. M.; Zou, Y.; Ling, J. J.; Yu, H.; Lin, Y.-S.; Pentelute, B. L., *J. Am. Chem. Soc.* **2013,** *135*(16), 5946-5949.
- 5. Zou, Y.; Spokoyny, A. M.; Zhang, C.; Simon, M. D.; Yu, H.; Lin, Y.-S.; Pentelute, B. L., *Org. Biomol. Chem.* **2014,** *12*(4), 566-573.
- 6. Heinis, C.; Rutherford, T.; Freund, S.; Winter, G., *Nat. Chem. Bio.* **2009,** *5*(7), 502-507.
- 7. Tsunemi, T.; Bernardino, S. J.; Mendoza, A.; Jones, C. G.; Harran, P. G., *Angew. Chem. Int. Ed.* **2020,** *59*(2), 674-678.
- 8. Sanger, F. *Biochem. J.* **1945**, 39, 507-515.
- 9. Neumann, K.; Farnung, J.; Baldauf, S.; Bode, J. W. *Nat. Commun.* **2020**, *11*(1), 1-10.
- 10. Soltani, O.; De Brabander, J. K. *Angew. Chem. Int. Ed.* **2005**, *44*(11), 1696-1699.
- 11. Liu, Z.; Larock, R. C. *J. Org. Chem.* **2006**, *71*(8), 3198-3209.
- 12. Xu, H.; He, J.; Shi, J.; Tan, L.; Qiu, D.; Luo, X.; Li, Y. *J. Am. Chem. Soc.* **2018**, *140*(10), 3555-3559.
- 13. Mesgar, M.; Daugulis, O. *Org. Lett.* **2016**, *18*(15), 3910-3913.
- 14. Crossley, J. A.; Kirkham, J. D.; Browne, D. L.; Harrity, J. P. *Tet. Lett.* **2010**, *51*(50), 6608- 6610.
- 15. Dougherty, P. G.; Sahni, A.; Pei, D. *Chem. Rev.* **2019**, *119*(17), 10241-10287.

16. Gavenonis, J.; Sheneman, B. A.; Siegert, T. R.; Eshelman, M. R.; Kritzer, J. A. *Nat. Chem. Biol.* **2014**, *10*(9), 716-722.

Chapter 4 – Cell-permeable peptidomimetic macrocycles from template precursors

4.1 Introduction

Macrocyclic peptide-based therapeutics have gained increasing attention for their potential to target clinically important protein-protein interactions (PPIs).¹⁻⁴ Their structural characteristics, however, often place them outside most classical definitions of 'drug-likeness' leading to underrepresentation in HTS drug discovery programs. These restrictions define property boundaries for hydrogen bond donor/acceptor groups, rotatable bonds, molecular weight, and polar surface area as predictors of oral bioavailability and cellular permeability.^{5,6} Widespread adoption of these guidelines by industry leaders and scientists has persisted for decades since its introduction, leading to a uniform chemical space represented in therapeutics.⁷⁻ ⁹ Recently, many have begun to reconsider how we assess 'drug-like' properties and the overall hypothesis itself. For example, Schultz has reported that since 1997 the threshold for both molecular weight and hydrogen bond acceptors has increased substantially based on hundreds of small molecules approved by the FDA over the last two decades.¹⁰ This evolving understanding is now being considered especially with the emergence of new chemical modalities in drug development.

Macrocyclic peptides often exhibit structural elements outside of desired ranges defined by these guidelines, yet there are many notable exceptions. Today, there exist many macrocyclic drugs in clinical use such as cyclosporine A (CSA), rapamycin, and vancomycin, however, they still only represent a fraction of the market. Considering their structural features, their surprising bioactivity and bioavailability has inspired extensive mechanistic studies that may help facilitate design of cell-permeable compounds. CSA's conformational flexibility for example, enables it to exhibit "chameleonic" behavior with its ability shift from displaying a more polar surface in aqueous environments to presenting a less polar surface when crossing lipophilic cell membranes through its ability to transiently form intramolecular hydrogen bonds (IMHBs).¹¹⁻¹² Recently, the Baker group successfully used computational-aided design of membrane permeable and orally bioavailable peptide macrocycles incorporating this conformational switching characteristic in their design models.¹³ Backbone N-methylation is also widely known to be correlated with increased membrane permeability.^{14,15} Interestingly, N-methylations have been found to do so not just by simply masking H-bond donors but by decreasing flexibility in the peptide backbone, thereby stabilizing hydrogen-bonded conformations.¹⁶ Computational estimations of topological polar surface area (TPSA) of a given molecule is commonly used for determining overall polarity when assessing bioavailability.¹⁷ Calculating TPSA is both rapid and convenient, however it is conformationally independent and does not account for potential intramolecular hydrogen bonding. Goetz and coworkers have developed a method using supercritical fluid chromatography which correlates chromatographic retention with the exposed polar surface area (EPSA) of a molecule.^{18,19} This technique relies on the idea that molecules that can form IMHBs are better able to hide their polarity than those that cannot, leading to lower retention times. EPSA has been used to determine unknown IMHB patterns in cyclic peptides via systematic N-methylations. For example, N-methylations resulting in a decrease or increase in EPSA would indicate a previously exposed N-H group or a significant conformational change now exposing polar functionality, respectively. Methylations leading to unchanged an EPSA value may indicate a hidden N-H group, likely due to intramolecular hydrogen bonding. Most importantly, correlations between EPSA and cellular permeability have been described using Ralph Russ Canine Kidney (RRCK) cells. With its ability to accurately identify IMHBs, independent of molecular weight and lipophilicity, this method acts as a valuable tool in drug design programs as a predictor of efficient membrane permeability. Combining these advancements with synthetic methods to develop unique structures with favorable properties are needed.

Our group has long-sought facile methods to convert linear peptide oligomeric sequences to diverse macrocyclic structures with favorable pharmacokinetic properties. Designed synthetic

35

templates **G1**-**G3** have provided access to increasingly complex peptidomimetic macrocycles (Figure 4.1).20-25 Each template generation contains a succinimidyl ester for efficient acylation and a cinnamyl carbonate core motif amenable to reaction conditions that selectively provide C-O, C-N, and C-C bonded products. Templates **G2** and **G3** contain additional functionality in the form of hidden aldehyde moieties allowing subsequent Pictet-Spengler transformations giving polycyclic structures. The alkyne in **G3** acts as a handle for further late-stage functionalization. This versatile system allows us to study the pharmacological properties of these structures in a systematic manner. Through collaborative efforts with Pfizer Inc., we have begun to explore structurepermeability relationships of peptidomimetic macrocycles in detail described here.

Figure 4.1. Previously developed synthetic templates **G1** – **G3** and representative outcomes for diverse macrocyclization.

4.2 Results and Discussion

Over several rounds, we synthesized a small library of 39 peptidomimetic macrocycles using our template methodology to evaluate cellular permeability (RRCK assays). This compound set consisted of compounds that surveyed a diverse array of functionality including systematic Nmethylations, nonproteinogenic amino acids, and incorporation of the **G1**-**G3** template systems. Overall, most compounds were found to exhibit at least moderate permeability, with nine falling in the "not permeable" range. Seven of these macrocycles displayed good permeability with RRCK apparent permeability (P_{apo}) values greater than 5×10^{-6} cm/s. Notably, when plotting cellular permeability as a function of molecular weight (Figure 4.2A), compounds throughout the 500 – 800 MW range exhibited moderate permeability. EPSA values were also determined for the entire set providing values ranging from 69 – 119 (Figure 4.2B). According to Goetz and coworkers, compounds with EPSA values below 100 are more likely to exhibit passive permeability compared to those above that cutoff. In our case, this criterion seems to hold true with most of our significantly permeable macrocycles occupying the space below this EPSA threshold (gray line, Figure 4.2B).

Figure 4.2. Correlation between RRCK apparent permeability and molecular weight or exposed polar surface area (EPSA) of macrocycle library. (Note: EPSA values are determined based on retention times directly compared with a set of calibration standards. See appendix for more information).

As mentioned previously, our template methodology enables facile synthesis of macrocycles with switchable selectivity. Following initial acylation of a tyrosine embedded peptide to G1, acyclic structures can undergo Friedel-Crafts reactivity (5-10% TFA in MeNO₂) or Tsuji-Trost type reaction conditions $(Pd(PPh₃)₄$ in DMF) arriving at C-C or C-O macrocycles, respectively. Using these methods, five pairs of compounds were synthesized using template **G1** to examine direct differences in permeability between these two macrocycle types (Figure 4.3). Most C-O linked macrocycles (**4.11** – **4.13**) exhibited greater passive permeability compared to their C-C linked counterparts (**4.6** – **4.8**) as predicted by EPSA. In two examples, we observed either a decrease in permeability (**4.9** & **4.14**) or no significant change (**4.10** & **4.15**). Based on these results, major conformational differences leading to favorable IMHB interactions, rather than simply masking H-bond donors, may explain the improved permeabilities.

Figure 4.3. Direct comparison of C-C and C-O linked macrocycles with their experimentally determined permeabilities.

We then explored various systematic modifications from parent structures **4.16** (Figure 4.4A) and **4.20** (Figure 4.4B) and their impact on permeability. Analogs were synthesized with changes at each amino acid position of the peptide backbone (P1 – P3). N-methyl amino acid incorporation at P1 for **4.16** significantly improved permeability with compounds **4.13** and **4.17**, whereas substituting valine with a more polar threonine (**4.18**) at P3 had a negative effect, unsurprisingly. In the case of compounds **4.13** and **4.17**, EPSA was not a reliable predictor of the observed permeability. Direct methyl capping of the tyrosine phenol in compound **4.8** (Figure 4.3) provided compound **4.19** with a substantial increase in permeability efficiency. Both macrocycles **4.13** and **4.19** positively affect RRCK P_{app} by masking the tyrosine phenol, however the significant

Figure 4.4. Evaluating pharmacological properties of macrocycles through various systematic changes.

disparity between the two suggests the presence of favorable conformational arrangements not easily predicted by such a minor modification. N-methylation of **4.20** at P1 lead to no noteworthy change in both permeability rate and EPSA retention (compound **4.13**). This likely indicates that the N-H of serine is involved in an IMHB in macrocycle **4.20**. Other P1 substitutions with Ser(OMe) and sarcosine greatly improved their permeability compared to the parent structure. Interestingly, cyclodehydration of **4.20** to give oxazoline **4.24** provided our most permeable macrocycle in this library (RRCK P_{app} 21.37 \times 10⁻⁶ cm/s). Unfortunately, attempts at assaying similar compounds with this functionality proved challenging due to its inherent instability. Use of nonproteinogenic lipophilic amino acids Ala(Cyclohexyl), Tle, and Gly(Cyclohexyl) at P3 had the expected desired effect. Compared to **4.21**, substituting the amino acid at P3 with either phenylalanine (**4.28**) or threonine (**4.29**) gave a notable increase and minor decrease in permeability, respectively.

Figure 4.5. Sub library of twelve macrocycles incorporating **G1**, **G2**, and **G3** templates and their corresponding permeability rates.

Subsequent studies focused on examining permeability differences among macrocycles bearing either the **G1**, **G2**, or **G3** template. A set of 12 compounds were synthesized using four peptides derived from the parent sequence H_2N -Trp-Tle-Ser-Tyramine (Figure 4.5). The peptides contain tryptophan at P1 for additional reactivity with the **G2**/**G3** templates and tyrosine at P4 for efficient Tsuji-Trost reactivity giving C-O linked macrocycles. The remainder of the sequence contains amphiphilic residues and individual N-methylations to study the potential for hybrid compounds of this type to carry polar side chains across a cell membrane. All 12 macrocycles exhibited apparent permeability ranging from $1.15 - 5.51 \times 10^{-6}$ cm/s. No change was observed between parent (**4.30** – **4.32**) and transposed (**4.33** – **4.35**) sequences, whereas N-methylations (**4.36** – **4.41**) led to the greatest increase in observed permeability. When comparing macrocycles across all three template systems, cellular permeability did not significantly vary. It's possible that structures bearing templates **G2**/**G3** benefit from increased lipophilicity, while those containing template **G1** are more flexible and can more effectively participate in IMHBs.

4.3 Conclusions

The work described here details various systematic modifications to examine structurepermeability relationships of a library of macrocycles. With peptidyl-based macrocycle inhibitors of PPIs increasingly growing in interest, studies on their pharmacological properties are greatly needed. The methodologies our group has developed to efficiently synthesize peptidomimetic macrocycles are uniquely amenable to this demand. Overall, we found most compounds exhibited at lease moderate permeability in RRCK cell lines with observed improvements based on siteselective modifications (e.g. N-methylations). In all cases the linear peptide starting materials were not permeable compared to their cyclic counterparts. This is a strong early starting point for us to understand important properties of these types of structures. Similar studies on analogous peptidomimetic macrocyclic systems are ongoing.

4.4 References

- 1. Vinogradov, A. A.; Yin, Y.; Suga, H. *J. Am. Chem. Soc.* **2019**, 141(10), 4167-4181.
- 2. Zhang, H.; Chen, S. *RSC Chem. Biol.* **2022**, 3(1), 18-31.
- 3. Zorzi, A., Deyle, K., & Heinis, C. *Curr. Opin. Chem. Biol.* **2017**, *38*, 24-29.
- 4. Doak, B. C.; Zheng, J.; Dobritzsch, D.; Kihlberg, J. *J. Med. Chem.* **2016**, *59*(6), 2312- 2327.

41

- 5. Lipinski, C. A.; Lombardo, F.; Dominy, B. W.; Feeney, P. J., *Adv. Drug Deliv. Rev.* **1997**, *23*, 3-25.
- 6. Veber, D. F.; Johnson, S. R.; Cheng, H. Y.; Smith, B. R.; Ward, K.W.; Kopple, K. D. *J. Med. Chem.* **2002**, *45*(12), 2615−2623.
- 7. Rask-Andersen, M.; Masuram, S.; Schioth, H. B. *Annu. Rev. Pharmacol. Toxicol.* **2014**, 54, 9−26.
- 8. Hopkins, A. L. and Groom, C. R. *Nat. Rev. Drug Discov.* **2002**, *1*(9), 727-730.
- 9. Rask-Andersen, M.; Masuram, S.; Schioth, H. B. *Annu. Rev. Pharmacol. Toxicol.* **2014**, 54, 9−26.
- 10. Shultz, M. D. *J. Med. Chem.* **2018**, *62*(4), 1701-1714.
- 11. Wang, C. K.; Swedberg, J. E.; Harvey, P. J.; Kaas, Q.; Craik, D. J. *J. Phys. Chem. B*, 122(8), 2261-2276.
- 12. Witek, J., Keller, B. G., Blatter, M., Meissner, A., Wagner, T., & Riniker, S.*J. Chem. Inf. Model*. **2016**, 56(8), 1547-1562.
- 13. Bhardwaj, G.; O'Connor, J.; Rettie, S.; Huang, Y.H.; Ramelot, T.A.; Mulligan, V.K.; Alpkilic, G.G.; Palmer, J.; Bera, A.K.; Bick, M.J.; Di Piazza, M.; Li, X.; Hosseinzadeh, P.; Craven, T.W.; Tejero, R.; Lauko, A.; Choi, R.; Glynn, C.; Dong, L.; Griffin, R.; Van Voorhis, W.; Rodriguez, J.; Stewart, L.; Montelione, G.T.; Craik, D.; Baker, D. *Cell* **2022**, *185*(19), 3520-3532.
- 14. Biron, E.; Chatterjee, J.; Ovadia, O.; Langenegger, D.; Brueggen, J.; Hoyer, D.; Schmid, H.A.; Jelinek, R.; Gilon, C.; Hoffman, A.; Kessler, H. *Angew. Chem. Int. Edn Engl.* **2008**, 47(14), pp.2595-2599.
- 15. Chatterjee, J.; Gilon, C.; Hoffman, A.; Kessler, H. *Acc. Chem. Res.* **2008**, *41*(10), 1331- 1342.
- 16. White, T.R.; Renzelman, C.M.; Rand, A.C.; Rezai, T.; McEwen, C.M.; Gelev, V.M.; Turner, R.A.; Linington, R.G.; Leung, S.S.; Kalgutkar, A.S.; Bauman, J.N.; Zhang, Y.; Liras, S.; Price, D.A.; Mathiowetz, A.M.; Jacobsen, M.P.; Lokey, S.R. *Nat. Chem. Biol.* **2011**, *7*(11), 810-817.
- 17. Ertl, P.; Rohde, B.; Selzer, P. *J. Med. Chem.* **2000**, *43*(20), 3714-3717.
- 18. Goetz, G. H.; Philippe, L.; Shapiro, M. J. *ACS Med. Chem. Lett.*, *5*(10), 1167-1172.
- 19. Goetz, G.H.; Farrell, W.; Shalaeva, M.; Sciabola, S.; Anderson, D.; Yan, J.; Philippe, L.; Shapiro, M.J. *J. Med. Chem.* **2014**, *57*(7), 2920-2929.
- 20. Lawson, K. V.; Rose, T. E. Harran, P.G.; *Tetrahedron* **2013**, *69*(36), 7683-7691.
- 21. Lawson, K. V.; Rose, T. E.; Harran, P. G. *Proc. Natl. Acad. Sci.* **2013**, *110*(40), E3753- E3760.
- 22. Rose, T. E.; Lawson, K. V.; Harran, P. G. *Chem. Sci.* **2015**, *6*(4), 2219-2223.
- 23. Rose, T. E.; Curtin, B. H.; Lawson, K. V.; Simon, A.; Houk, K. N.; Harran, P. G., *Chem. Sci.* **2016,** *7* (7), 4158-4166.
- 24. Curtin, B. H., Manoni, F., Park, J., Sisto, L. J., Lam, Y. H., Gravel, M., Roulston, A., Harran, P. G., *J. Org. Chem.*, **2018**, *83*, (6), 3090-3108.
- 25. Sisto, L. J., Harran, P. G., *Tet. Lett.*, **2020**, *61* (24), 151986.
- 26. Sisto, L. J., Harran, P. G., *Tet. Lett.*, **2020**, *61* (24), 152985.

Experimental Appendices

Chapter 2 Appendix Material

General Methods

Reagents were purchased from commercial vendors and used as received unless otherwise stated. Tetrahydrofuran (THF), diethyl ether (Et₂O), acetonitrile (MeCN) and toluene (PhMe) were passed through a Glass Contour solvent drying system.

Yields refer to chromatographically and spectroscopically (¹H NMR) homogeneous materials, unless otherwise stated. Thin-layer chromatography (TLC) was conducted on precoated plates (Sorbent Technologies, silica gel 60 PF254, 0.25 mm) visualized with UV 254 nm. Column chromatography was performed on silica gel 60 (SiliCycle, 240−400 mesh). Purification of peptides was performed using an Agilent 1200 HPLC system equipped with G1361A preparative pumps, a G1314A auto sampler, a G1314A VWD, a G1364B automated fraction collector, and a Waters Sunfire C18 column (5 μm, 19 mm 250 mm), unless otherwise noted. Analytical HPLC was performed using the same system, but with a G1312A binary pump. Mass spectra were recorded using an Agilent 6130 LC/MS system equipped with an ESI source. High-resolution mass spectra were recorded on Thermo Scientific Exactive® Mass Spectrometer with DART ID-CUBE Waters GST Premier, Waters LCT Premier, and Agilent 6545 LC-QTOF. NMR spectra were recorded on Bruker Avance (300, 400 or 500 MHz) spectrometers. HSQC, COSY, HMBC NMR experiments were used to aid assignment of NMR peaks when required.

General Procedure A – Peptide synthesis:

C-terminal carboxamide peptides were synthesized manually using standard Fmoc solid phase synthesis protocols on Rink Amide MBHA resin (200-400 mesh, 0.73 mmol/g, 1% DVB) on 0.37- 0.50 mmolscale using a fritted glass reaction vessel. Fmoc-deprotection was achieved with 20% piperidine in DMF (2 x 30 min). The reaction vessel was washed with DMF (3x) and CH_2Cl_2 (2x). The vessel was then charged with the appropriate Fmoc-amino acid (4 eq.) and HBTU (4 eq.) followed by DMF (10-20 ml) and $iPr₂NEt$ (10 eq). The resin was agitated for 2 hours, drained, and washed with DMF (3X). After all coupling were completed the resin was cleaved with TFA/ water/TIPS (90:5:5) for 1.5 hrs. The cleaved resin was removed by filtration and the filtrate was concentrated under vacuum. The peptide was precipitated with $Et₂O$ and isolated by centrifugation. The peptide pellet was repeated triturated with EtOAc and isolated by centrifugation to ensure complete removal of cleavage reagents. X was used to prepare PFCP containing peptides.

General Procedure B – Macrocyclization

A flask was charged with linear precursor (1 eq) and DMF (10 mM in substrate). The mixture was added Cs_2CO_3 (2 eq) at 0 °C, stirred for 2 hrs at rt. The mixture was diluted with EtOAc, washed with brine (5 times). The organic layer was dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. Purification condition — see details per example, below.

General Procedure C – Direct macrocyclization

A flask was charged with linear precursor (1 eq) and DMF (10 mM in substrate). The mixture was added octafluorocyclopentene (1M in MeCN, 5 eq) and Et_3N (1.5 eq) at 0 °C. The mixture was stirred for 30 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added Cs_2CO_3 (1 eq) at 0 °C and stirred for 1 hr at rt. The mixture was diluted with EtOAc, washed with brine (5 times). The organic layer was dried over Na2SO4, filtered and the solvent was removed by rotary evaporation. Purification condition — see details per example, below.

Experimental Procedures

((Perfluorocyclopent-1-en-1-yl)oxy)benzene (2.5)

A solution of phenol (1.09 g, 11.6 mmol) in DMF (60 ml) was cooled to 0 °C and added octafluorocyclopentene (1.70 ml, 12.7 mmol) and Et_3N (2.66 ml, 19.1 mmol). The resulting solution was allowed to stir for 1 hr at 0 °C. The reaction mixture was added H_2O , extracted with EtOAc, washed with brine (x5), dried over $N_{2}SO_{4}$, filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography $(SiO₂, hexane only)$ afforded the title compound (2.45 g, 8.56 mmol, 74%) as a colorless oil. ¹H NMR (DMSO-d₆, 400 MHz) δ 7.53-7.45 (m, 4H), 7.40-7.35 (m, 1H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 153.1, 136.5- 133.6 (m, 1C), 134.3-133.6 (m, 1C), 130.2, 127.1, 119.1, 114.2-106.2 (m, 3C). ¹⁹F NMR (DMSO*d*6, 282 MHz) δ -113.5 (d, *J* = 14 Hz, 2F), -114.0-(-114.1) (m, 2F), -128.1 (s, 2F), -150.4-(-150.6) (m, 1F). HRMS m/z [M]⁺ calc'd for $C_{11}H_{5}F_{7}O$ 286.0, found 286.0.

1-(But-3-yn-1-yloxy)-2,3,3,4,4,5,5-heptafluorocyclopent-1-ene (2.6)

A solution of 3-butyn-1-ol (1 ml, 13.2 mmol) in DMF (100 ml) was cooled to 0 °C and added octafluorocyclopentene (2.11 ml, 15.9 mmol) and Et_3N (2.22 ml, 15.9 mmol). The resulting solution was allowed to stir for 3 hr at rt. The reaction mixture was diluted with $Et₂O$. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. Purification by column chromatography ($SiO₂$, hexane only) afforded the title compound (1.96 g, 7.48 mmol, 57%) as a pale yellow oil.

¹H NMR (DMSO-*d*6, 500 MHz) δ 4.59 (dt, *J* = 6.3, 2.9 Hz, 2H), 2.95-2.93 (m, 1H), 2.69 (dt, *J* = 6.3, 2.6 Hz, 2H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 135.2 (t, *J* = 23.6 Hz, 1C), 134.7-131.8 (m, 1C), 114.1-106.4 (m, 3C), 79.7, 73.0, 71.5 (d, *J* = 4.6 Hz, 1C), 19.0 (d, *J* = 1.5 Hz, 1C). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -112.8-(-112.9) (m, 2F), -114.6-(-114.8) (m, 2F), -128.2-(-128.4) (m, 2F), -159.6-(-159.8) (m, 1F). HRMS m/z [M] ⁺ calc'd for C9H5F7O 262.0223, found 262.0223.

Benzyl 4-((perfluorocyclopent-1-en-1-yl)oxy)butanoate (S2.1)

A solution of benzyl 4-hydroxybutanoate^[S1] (2.95 g, 15.2 mmol) in DMF (60 ml) was cooled to 0 $°C$ and added octafluorocyclopentene (3.04 ml, 22.8 mmol) and Et₃N (4.24 ml, 30.4 mmol). The resulting solution was allowed to stir for 1 hr at rt. The reaction mixture was diluted with Et_2O . The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography (SiO2, gradient 10-20% EtOAc/hexane) afforded the title compound (3.59 g, 9.29 mmol, 61%) as a pale yellow oil. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.39-7.32 (m, 5H), 5.14 (s, 2H), 4.43 (td, *J* = 6.2, 2.9 Hz, 2H), 2.53 (t, *J* = 7.2 Hz, 2H), 2.13 (qui, *J* = 6.6 Hz, 2H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.3, 135.9 135.9-135.6 (m, 1C), 135.6-133.6 (m, 1C), 128.8, 128.6, 128.5, 114.0-106.6 (m, 3C), 72.5 (d, *J* = 4.4 Hz, 1C), 66.8, 29.9, 24.6 (d, *J* = 1.4 Hz, 1C). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ - 115.3 (brd, *J* = 12.7 Hz, 2F), -116.5-(-116.6) (m, 2F), -130.0 (s, 2F), -162.0-(-162.2) (m, 1F). HRMS m/z [M]⁺ calc'd for C₁₆H₁₃F₇O₃ 386.0747, found 386.0748.

2,5-Dioxopyrrolidin-1-yl 4-((perfluorocyclopent-1-en-1-yl)oxy)butanoate (2.7)

A solution of **S2.1** (2.75 g, 9.30 mmol) in DCM (100 ml) was added *N*-hydroxysuccinimide (2.14 ml, 18.6 mmol) and EDC·HCl (3.57 ml, 18.6 mmol). The resulting solution was allowed to stir for 3 hrs at rt. The reaction mixture was added 1N HClaq and extracted with CHCl₃. The organic layer was dried over Na₂SO₄, filtered and the solvent was removed by rotary evaporation. Purification by column chromatography (SiO₂, gradient 33-50% EtOAc/hexane) afforded the title compound (3.03 g, 7.72 mmol, 83%) as a pale yellow oil. ¹H NMR (DMSO-*d*6, 500 MHz) δ 4.61 (td, *J* = 6.2, 3.1 Hz, 2H), 2.83 (t, *J* = 7.4, 2H), 2.82 (brs, 4H), 2.09 (qui, *J* = 6.8 Hz, 2H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 170.2, 168.4, 135.5-135.2 (m, 1C), 134.3-132.1 (m, 1C), 113.7-106.7 (m, 3C), 72.4 (d, *J* = 4.4 Hz, 1C), 26.4, 25.4, 23.9 (d, *J* = 1.3 Hz). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -112.7-(- 112.9) (m, 2F), -114.7-(-114.8) (m, 2F), -128.2-(-128.3) (m, 2F), -160.7-(-160.8) (m, 1F). HRMS [M-OSu]⁺ calc'd for C9H6F7O² 279.0251, found 279.0247.

Benzyl (*S***)-(1-amino-3-(4-((3,3,4,4,5,5-hexafluoro-2-phenoxycyclopent-1-en-1 yl)oxy)phenyl)-1-oxopropan-2-yl)carbamate (2.8)**

A solution of **2.5** (96 mg, 0.336 mmol) in DMF (2.5 ml) was cooled to 0 °C and added Cbz-L-Tyr- $NH₂$ (81 mg, 0.258 mmol) and $Cs₂CO₃$ (110 mg, 0.336 mmol). The resulting solution was allowed to stir for 1 hr at 0 °C. The reaction mixture was added water, and then extracted with EtOAc. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography ($SiO₂$, gradient 0-5% MeOH/CHCl3) afforded the title compound (108 mg, 0.186 mmol, 72%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.46 (brs, 1H), 7.39 (d, *J* = 8.8 Hz, 1H), 7.32-7.24 (m, 5H), 7.20 (t, *J* = 8.5, 2H), 7.11 (d, *J* = 8.6 Hz), 7.06-7.03 (m, 2H), 6.75 (d, *J* = 8.0 Hz, 2H), 6.72 (d, *J* = 8.7 Hz), 4.95 and 4.93 (AB quartet, *J* = 12.8 Hz, 2H), 4.11-4.06 (m, 1H), 2.89 (dd, *J* = 13.7, 3.9 Hz), 2.64 (dd, *J* = 13.7, 10.8 Hz, 1H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.3, 155.9, 153.6, 152.0, 137.0, 135.5, 133.8-133.4 (m, 2C), 130.4, 129.8, 128.3, 127.7, 127.5, 125.2, 116.9, 116.6, 114.8- 107.0 (m, 3C), 65.2, 56.1, 36.6. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -112.7 (app. t, 4F), -128.6 (app. s, 2F);. HRMS m/z $[M + H]^{+}$ calc'd for $C_{28}H_{22}F_6N_2O_6H$ 581.1511, found 581.1532

Benzyl (*R***)-(1-amino-3-((3,3,4,4,5,5-hexafluoro-2-phenoxycyclopent-1-en-1-yl)thio)-1 oxopropan-2-yl)carbamate (2.9)**

A solution of **2.5** (105 mg, 0.367 mmol) in DMF (2.8 ml) was cooled to 0 °C and added Cbz-L-Cys-NH₂ (72 mg, 0.282 mmol) and $Cs₂CO₃$ (110 mg, 0.367 mmol). The resulting solution was allowed to stir for 1 hr at 0 °C. The reaction mixture was added water, and then extracted with EtOAc. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography (SiO₂, gradient 0-5% MeOH/CHCl₃) afforded the title compound (126 mg, 0.242 mmol, 86%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.56 (d, *J* = 8.7 Hz, 1H) , 7.52 (brs, 1H) , 7.44 (t, *J* = 8.3 Hz) , 7.26 - 7.36 (m, 7H) , 7.22 (d, *J* = 8.1 Hz, 2H)3.10 (dd, *J* = 9.4, 13 Hz, 1H), 3.37 (dd, *J* = 4.9, 13 Hz, 1H), 4.10 - 4.14 (m, 1H), 5.04 (dd, 1.6, 13 Hz, 2H). ¹³C NMR (DMSO*d*6, 126 MHz) δ 170.9, 155.9, 153.3, 148.0 (m, 1C), 136.8, 130.4, 128.2, 127.5, 127.2, 126.2, 122.1 (m, 1C), 118.5, 117.5 - 107.6 (m, 3C), 65.7, 54.0, 33.4. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ - 105.3 and -105.4 (AB quartet, *J* = 249 Hz, 2F), -112.2 and -112.4 (AB quartet, *J* = 255 Hz, 2F), - 128.6-(-128.7) (m, 2F). HRMS m/z $[M + H]$ ⁺ calc'd for $C_{22}H_{18}F_6N_2O_4SH$ 521.0970, found 521.0952.

Benzyl (*S***)-(1-amino-3-(1-(3,3,4,4,5,5-hexafluoro-2-phenoxycyclopent-1-en-1-yl)-1Himidazol-4-yl)-1-oxopropan-2-yl)carbamate (2.10)**

A solution of **2.5** (100 mg, 0.350 mmol) in DMF (2.7 ml) was cooled to 0 °C and added Cbz-L-His-NH₂ (77 mg, 0.269 mmol) and $Cs₂CO₃$ (110 mg, 0.350 mmol). The resulting solution was allowed to stir for 1 hr at 0 °C. The reaction mixture was added water, and then extracted with EtOAc. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography ($SiO₂$, gradient 0-5% MeOH/CHCl₃) afforded the title compound (120 mg, 0.216 mmol, 81%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.85 (s, 1H), 7.38–7.18 (m, 13H), 7.05 (brs, 1H), 4.98 (s, 2H), 4.16–4.14 (m, 1H), 2.81 (dd, *J* = 15, 4.2 Hz), 2.68 (dd, *J* = 15, 9.8 Hz). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.2, 155.8, 152.4, 140.2 (m, 1C), 140.0, 137.0, 136.9, 130.0, 128.3, 127.8, 127.8, 127.6, 126.6, 118.7, 116.5 (m, 1C), 115.3, 114.4-107.2 (m, 3C), 65.4, 54.0, 30.2. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ; -108.3, -111.4, -128.1. HRMS m/z [M + H]⁺ calc'd for C25H20F6N4O4H 555.1467, found 555.1492.

Benzyl ((8*R***)-2,3,3,4,4-pentafluoro-9-oxo-1-phenoxy-6-oxa-10-azaspiro[4.5]dec-1-en-8 yl)carbamate (2.11 a/b)**

A solution of **2.5** (78 mg, 0.273 mmol) in DMF (2.1 ml) was cooled to 0 °C and added Cbz-L-Ser- $NH₂$ (50 mg, 0.21 mmol) and $Cs₂CO₃$ (89 mg, .273 mmol). The resulting solution was allowed to stir for 3 hr at 0 °C. The reaction mixture was quenched with AcOH (31 µl, 2.6 eq) and concentrated under reduced pressure to give a crude oil. Purification by column chromatography $(SIO₂, gradient 10-20% MeCN/CH₂Cl₂)$ afforded the title compound as a white solid and as a mixture of diastereomers (1:1) and rotamers (50.2 mg, 0.104 mmol, 49%). The diastereomeric mixture was separated by preparative reverse-phase HPLC and identified by NMR. Cbz deprotection of **2.11a** provided a single compound, confirming the presence the initial rotamer mixture. **2.11a**: ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.64 (brs, 1H), 7.50 – 7.22 (m, 9H), 7.18 (d, *J* = 7.5 Hz, 1.5H), 6.90 (d, *J* = 6.9 Hz, 0.5H), 5.30 – 5.10 (m, 2H), 4.64 – 4.50 (m, 2H), 4.1 – 4.02 (m, 1H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 168.9 (major), 168.8 (minor), 153.5 (major), 153.2 (minor), 151.8 (major), 151.3 (minor), 138.5 - 137.9 (m, 1C), 135.9 (major), 135 (minor), 133.7 – 130.5 (m, 1C), 130, 128.5, 128.3, 127.9, 127.2, 126.4, 118.9, 118.5, 115.2 – 109.8 (m, 2C), 93.5 – 92.8 (m, 1C), 70.9 (major), 70.2 (minor), 68.2 (minor), 67.3 (major), 59.5 (minor), 58.9 (major). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -106.4 – (-108.5) (m, 1F), -116.4 – (-120.2) (m, 2F), -127 – (-128.8) (m, 1F), -158.7 – (-159.9) (m, 1F). HRMS m/z [M + Na]⁺ calc'd for C₂₂H₁₇F₅N₂O₃Na 507.0955, found 507.0967. **2.11b**: ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.44 (t, *J* = 7.23 Hz, 1.5H), 7.40 – 7.20 (m, 10H) 6.92 (d, *J* = 7.5, 0.5H), 5.31 – 5.03 (m, 2H), 4.65 – 4.56 (m, 1H), 4.45 – 4.31 (m, 2H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 169.5 (major), 169.3 (minor), 154.1 (major), 153.5, 151.9 (major), 151.2 (minor), 138.4 – 137.8 (m, 1C), 135.8 (major), 134.8 (minor), 134.7 – 131.3 (m, 1C), 129.9, 129.8, 128.7, 128.6, 128.4, 128.3, 127.9, 127.3, 126.1, 126.1, 118.8, 118.6, 115.7 – 109.8 (m, 2C), 93.1 – 91.7 (m, 1C), 71.3 (major), 70.5 (minor), 68.2 (minor), 67.2 (major), 58.5 (minor), 58 (major). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -109 (dd, *J* = 243.6, *J* = 17.7 Hz, 1F), -116.5 – (-118.6) (m, 2F), -126.7 (dd, *J* = 230.3, 22.9 Hz, 1F), -157.3 (dd, *J* = 68.7, 13.3 Hz, 1F). HRMS m/z [M + Na]⁺ calc'd for C₂₂H₁₇F₅N₂O₃Na 507.0955, found 507.0956.

Rate determination for individual reactions of L-Cbz-His-NH2, L-Cbz-Tyr-NH2, L-Cbz-Ser-NH2, and L-Cbz-Cys-NH² with mono phenoxy derivative 2.5

Amino acids (0.05 mmol) L-Cbz-Ser-NH₂, L-Cbz-Tyr-NH₂, L-Cbz-His-NH₂, and L-Cbz-Cys-NH₂ were individually subjected (as triplicates) to reactions with mono phenoxy derivative **5** (0.065 mmol, 1.3 eq) and Cs_2CO_3 (0.065 mmol, 1.3 eq) in DMF (0.1 M, 0.5 ml) at 0 °C. Aliquots (10 µl) were removed periodically, quenched in an HPLC vial containing 1M AcOH in MeCN, and analyzed by HPLC-UV. Time-course peak area data (254 nm) was used to construct the corresponding plots (below). L-Cbz-Cys-NH² was observed to be fully consumed within 15 seconds.

	Mean ($n = 3$) Normalized % Area (254 nm)		
Time (min)	L -Cbz-His-NH ₂	Product (2.10)	
0	100	0	
1	51.3	48.7	
2	34.3	65.7	
5	17.7	82.0	
10	10.0	90.3	
15	6.7	93.3	
30	3.3	96.7	
45	1.9	98.1	
60	1.5	98.5	

Figure S1. Rate determination for the reaction of L-Cbz-His-NH² with mono phenoxy derivative **2.5**.

	Mean ($n = 3$) Normalized % Area (254 nm)		
Time (min)	L -Cbz-Tyr-NH ₂	Product (2.8)	
0	99.99	0.01	
	69.5	30.5	
2	55.2	44.8	
5	35.0	65.0	
10	26.9	73.1	
15	21.6	78.4	
30	14.9	85.1	
45	11.1	89.0	
60	7.6	92.4	

Figure S2. Rate determination for the reaction of L-Cbz-Tyr-NH² with mono phenoxy derivative **2.5**.

	Mean ($n = 3$) Normalized % Area (254 nm)		
Time (min)	L-Cbz-Ser-NH ₂	Product (2.11 a/b)	
1	90.80	9.23	
2	87.30	12.70	
5	80.57	19.40	
10	69.97	30.03	
15	63.87	36.13	
30	47.10	52.83	
45	37.70	62.07	
60	31.00	69.03	
90	24.40	75.63	
120	21.33	78.70	

Figure S3. Rate determination for the reaction of L-Cbz-Ser-NH² with mono phenoxy derivative **2.5**.

tert-Butyl ((*S***)-1-(((***S***)-1-((4-hydroxyphenethyl)amino)-1-oxo-3-phenylpropan-2-yl)amino)-1 oxopropan-2-yl)carbamate (S2.2)**

Tyramine·HCl (1.70 g, 9.80 mmol), iPr2EtN (3.41ml, 19.6 mmol) and HBTU (3.72 g, 9.80 mmol) were added to a solution of Boc-L-Phe-OH (2.00 g, 7.54 mmol) in DMF (30 ml). The resulting solution was allowed to stir for 1 hr at rt. The reaction mixture was added H_2O and extracted with EtOAc, washed with 1N HCl, sat. NaHCO₃ and brine. The organic layer was dried over Na₂SO₄, filtered and the solvent was removed by rotary evaporation. The residue was dissolved in DCM (30 ml) and added TFA (15 ml). The resulting solution was allowed to stir for 1 hr at rt and concentrated by rotary evaporation to give a crude solid (3.11 g, 7.54 mmol analytical). The residue (1.72 g, analytical 4.16 mmol) was dissolved in DMF (40 ml) and treated with Boc-L-Ala-OH (1.02 g, 5.41 mmol), iPr_2E tN (2.17 ml, 12.5 mmol) and then by HBTU (2.05 g, 5.41 mmol). After stirring 1 hr, added H_2O and extracted with EtOAc. The combined organic phase was washed sequentially with 1N HCl, sat. NaHCO₃ and brine. Dried over $Na₂SO₄$ and concentrated. Purification by column chromatography (SiO₂, gradient 30-40% EtOAc/hexane) afforded the title compound (1.54 g, 3.37 mmol, 81% from Boc-L-Phe-OH) as a white foam. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.13 (s, 1H), 7.93 (t, *J* = 4.9 Hz, 1H), 7.73 (d, *J* = 8.4 Hz, 1H), 7.22-7.20 (m, 2H), 7.16-7.13 (m, 3H), 6.97-6.94 (m, 1H), 6.93 (d, *J* = 8.4, 2H), 6.63 (d, *J* = 8.4 Hz, 2H), 4.42-4.38 (m, 1H), 3.87-3.84 (m, 1H), 3.22-3.17 (m, 1H), 3.14-3.09 (m, 1H), 2.90 (dd, *J* = 13.6, 5.1 Hz 1H), 2.77 (dd, *J* = 13.6, 8.8 Hz, 1H), 2.53-2.48 (m, 2H), 1.35 (s, 9H), 1.04 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.3, 170.5, 155.6, 155.1, 137.6, 129.4, 129.3, 129.2, 128.0, 126.2, 115.1, 78.2, 53.6, 50.1, 40.6, 37.7, 34.2, 28.2, 18.0. HRMS m/z $[M + H]^{+}$ calc'd for C₂₅H₃₃N₃O₅Na 478.2318, found 478.2343.

*N***-((***S***)-1-(((***S***)-1-((4-Hydroxyphenethyl)amino)-1-oxo-3-phenylpropan-2-yl)amino)-1 oxopropan-2-yl)pent-4-ynamide (S2.3)**

TFA (10 ml) was added to a solution of **S2.2** (1.75 g, 3.84 mmol) in DCM (20 ml) and stirred for 1hr at rt. The reaction mixture was concentrated by rotary evaporation to give a crude solid **2.12** (1.80 g, 3.84 mmol). The residue **2.12** (250 mg, 0.532 mmol, analytical) was dissolved in DMF (4 ml) and added 4-pentinoic acid (78 mg, 0.798 mmol), HOBt (108 mg, 0.798 mmol), $iPr₂EtN$ (277 μl, 1.60 mmol) and EDC·HCl (153 mg, 0.798 mmol). The mixture was stirred for 3 hrs and diluted with EtOAc. The organic phase was washed with 1N HClaq, sat NaHCO $_3$ and brine, then dried over Na₂SO₄ and concentrated. The residue was triturated with Et₂O, centrifuged, and the resulting solid was collected to give the title compound (220 mg, 0.505 mmol, 95%) as a pale yellow solid, used for next step without further purification. ¹H NMR (DMSO-*d*6, 400 MHz) δ 9.15 (s, 1H), 8.07 (d, *J* = 7.1 Hz, 1H), 7.88 (t, *J* = 5.6 Hz, 1H), 7.83 (d, *J* = 8.3 Hz, 1H), 7.26-7.21 (m, 2H), 7.20-7.14 (m, 3H), 6.96 (d, *J* = 8.4 Hz, 2H), 6.66 (d, *J* = 8.4 Hz, 2H), 4.43-4.36 (m, 1H), 4.25- 4.16 (m, 1H), 3.26-3.10 (m, 2H), 2.95 (dd, *J* = 13.8, 5.2 Hz, 1H), 2.82-2.74 (m, 2H), 2.57-2.51 (m, 1H), 2.37-2.26 (m, 4H), 1.10 (d, *J* = 7.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.9, 170.5, 170.4, 155.6, 137.8, 129.5, 129.4, 129.2, 128.0, 126.2, 115.1, 83.7, 71.4, 53.8, 48.4, 40.6, 37.5, 34.2, 33.9, 17.9, 14.0. HRMS m/z [M + Na]⁺ calc'd for C₂₅H₂₉N₃O₄Na 458.2056, found 458.2083.

*N***-((***S***)-1-(((***S***)-1-((4-((2-(But-3-yn-1-yloxy)-3,3,4,4,5,5-hexafluorocyclopent-1-en-1 yl)oxy)phenethyl)amino)-1-oxo-3-phenylpropan-2-yl)amino)-1-oxopropan-2-yl)pent-4 ynamide (2.13)**

6 (95 mg, 0.361 mmol) and Cs_2CO_3 (110 mg, 0.361 mmol) were added to a solution of **S3** (105 mg, 0.241 mmol) in DMF (2 ml). The resulting solution was allowed to stir for 3 hrs at rt. The reaction mixture was diluted with EtOAc, washed with brine, dried over Na_2SO_4 , filtered and the solvent was removed by rotary evaporation to give a crude oil. Purification by column chromatography (SiO₂, EtOAc only) afforded the title compound (86 mg, 0.127 mmol, 53%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.07 (d, *J* = 7.0 Hz, 1H), 7.95 (t, *J* = 5.5 Hz, 1H), 7.85 (d, *J* = 8.3 Hz, 1H), 7.32-7.07 (m, 9H), 4.43 (m, 1H), 4.33 (t, *J* = 6.2 Hz, 1H), 4.20 (qn, *J* = 7.0 Hz, 1H), 2.95-2.89 (m, 2H), 2.80-2.74 (m, 2H), 2.69-2.64 (m, 2H), 2.54-2.51 (m, 2H), 2.36-2.27 (m, 4H), 1.10 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.9, 170.6, 170.4, 153.7, 140.9- 140.4 (m,1C), 137.8, 130.5-130.2 (m, 1C), 136.1, 130.4, 129.1, 128.0, 126.2, 116.5, 114.6-106.4 (m, 3C), 83.7, 79.8, 73.0, 71.3, 70.8, 53.8, 48.4, 40.1, 37.5, 34.0, 33.9, 19.1, 17.9, 14.0. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -110.2 (s, 2F), -114.1 (s, 2F), -129.1-(-129.2) (m, 2F). HRMS m/z [M + H ⁺ calc'd for C₃₄H₃₃F₆N₃O₅H 678.2403, found 678.2400.

Macrocyclic Product (2.14)

 $Cu(OAc)₂·H₂O$ (6.7 mg, 0.0369 mmol), NiCl₂ (4.8 mg, 0.0369 mmol), Et₃N (15 µl, 0.111 mmol) and pyridine (15 μl, 0.185 mmol) were added to a solution of **13** (25 mg, 0.0369 mmol) in EtOH (4 ml). The reaction mixture was vigorously stirred for 3 hrs under open air at 50 °C. After cooling to rt, diluted with EtOAc, washed with 1N HCl ag, sat. NaHCO $_3$ and brine. The organic layer was dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. Purification by column chromatography (SiO2, gradient 80-100% EtOAc/hexane) afforded the title compound (13 mg, 0.0192 mmol, 52%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.20 (d, *J* = 6.0 Hz, 1H), 7.81 (d, *J* = 8.4 Hz, 1H), 7.75 (t, *J* = 5.5 Hz, 1H), 7.31 (d, *J* = 8.6 Hz, 2H), 7.28-7.22 (m, 1H), 7.23 (d, *J* = 7.5 Hz, 2H), 7.19-7.15 (m, 1H), 7.12 (d, *J* = 7.2 Hz, 2H), 7.10 (d, *J* = 8.6 Hz, 1H), 4.31-4.25 (m, 1H), 4.23-4.16 (m, 2H), 4.02 (qn, *J* = 6.7 Hz, 1H), 3.39-3.27 (m, 2H), 2.95 (dd, *J* = 13.9, 4.7 Hz, 1H), 2.79-2.73 (m, 3H), 2.61-2.28 (m, 6H), 1.03 (d, *J* = 7.1 Hz, 3H). 13C NMR (DMSO*d*6, 126 MHz) δ 171.9. 171.1, 170.5, 153.3, 138.9-138.5 (m, 1C), 138.7, 138.2, 136.5, 130.5, 129.2-129.0, 129.0, 128.0, 126.2, 116.7, 115.5-105.5 (m, 3C), 78.3, 72.9. 70.4, 66.7, 65.0, 54.2, 49.3, 36.8, 33.9, 32.8, 19.7, 17.1, 14.6.¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -111.0 (d, *J* = 27.0 Hz, 2F), -112.3-(-112.4) (m, 2F), -128.7 (s, 2F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{34}H_{31}F_6N_3O_5H$ 676.2246, found 676.2241.

Macrocyclic Product (2.16)

TFA (10 ml) was added to a solution of **S2.2** (1.75 g, 3.84 mmol) in DCM (20 ml) and stirred for 1hr at rt. The reaction mixture was concentrated by rotary evaporation to give a crude solid **2.12** (1.80 g, 3.84 mmol). The residue **2.12** (62 mg, 0.131 mmol, analytical) was dissolved in DMF (1.3 ml) and added 2.7 (43 mg, 0.144 mmol), cooled to 0 $^{\circ}$ C, and then added iPr₂EtN (34 ul, 0.197 mmol). The reaction mixture was stirred for 2 hrs at rt, then added H_2O , extracted with EtOAc, washed with 1N HCl, sat. NaHCO₃ and brine. The organic layer was dried over Na₂SO₄, filtered and the solvent was removed by rotary evaporation to give crude **2.15**. The residue **2.15** was

dissolved in DMF (13 ml) and added $Cs₂CO₃$ (64 mg, 0.196 mmol) at 0 °C, stirred for 1 hr at rt.. The reaction mixture was added H_2O , extracted with EtOAc. The combined extract was washed with brine (x5), dried over $Na₂SO₄$ and concentrated. Purification by column chromatography (SiO2, gradient 1-5% MeOH/CHCl3) afforded the title compound (38 mg, 0.0619 mmol, 47% from **S2**) as a white solid. ¹H NMR (DMSO-*d*6, 400 MHz) δ 8.14 (d, *J* = 4.4 Hz, 1H), 7.53 (d, *J* = 8.1 Hz, 1H), 7.32, (t, *J* = 5.4 Hz, 1H), 7.26-7.24 (m, 2H), 7.18-7.14 (m, 5H), 7.02 (d, *J* = 8.6 Hz, 2H), 4.31- 4.27 (m, 1H), 4.17-4.11 (t, *J* = 6.8 Hz, 2H), 3.73-3.71 (m, 1H), 3.60-3.57 (m, 1H), 3.19 (dd, *J* = 14.3, 4.0 Hz, 1H), 3.05-2.96 (m, 1H), 2.99 (dd, *J* = 14.3, 10.0 Hz, 1H), 2.80-2.75 (m, 1H), 2.68- 2.63 (m, 1H), 2.02-1.89 (m, 2H), 1.43-1.28 (m, 2H), 1.00 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ172.6, 172.3, 171.0, 153.2, 138.8-138.3 (m, 1C), 138.3, 136.7, 130.6, 129.6-129.1 (m, 1C), 128.8, 128.1, 126.2, 116.2, 115.6-105.7 (m, 3C), 72.9, 53.7, 50.5, 40.6, 35.8, 33.7, 30.5, 24.5, 16.9. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -111.8 (d, *J* = 54.6 Hz, 2F), -112.5 (d, *J* = 71.9 Hz, 2F), -128.6-(-128.8) (m, 2F). HRMS m/z $[M + H]^+$ calc'd for $C_{28}H_{30}F_6N_3O_5H$ 614.2084, found 614.2062.

tert-Butyl ((*S***)-1-(((***S***)-1-((4-hydroxybutyl)amino)-3-(naphthalen-2-yl)-1-oxopropan-2 yl)amino)-1-oxopropan-2-yl)carbamate (2.17)**

K-Oxyma (2.58 g, 14.3 mmol), EDC HCl (2.73 g, 14.3 mmol), $iPr₂EtN$ (3.31ml, 19.0 mmol) and 4amino-1-butanol (1.76 g, 19.0 mmol) were added to a solution of Boc-L-2-Nal-OH (3.00 g, 9.51 mmol) in DMF (60 ml). The resulting solution was allowed to stir for 2 hrs at rt. The reaction mixture was added H_2O and extracted with EtOAc, washed with 1N HCl, sat. NaHCO₃ and brine. The organic layer was dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. The residue was dissolved in DCM (20 ml) and added TFA (20 ml). The resulting solution was allowed to stir for 1 hr at rt and concentrated by rotary evaporation to give a crude oil (8.78 g, 9.51 mmol analytical). The residue (6.51 g, analytical 7.05 mmol) was dissolved in DMF (50 ml) and treated with Boc-L-Ala-OH (1.73 g, 9.17 mmol), K-Oxyma (1.91 g, 10.6 mmol), iPr2EtN (2.45 ml, 14.1 mmol) and then by EDC·HCl (2.30 g, 12.0 mmol). After stirring 3 hrs, diluted with EtOAc, washed sequentially with 1N HCl, sat. NaHCO₃ and brine. Dried over Na₂SO₄ and concentrated. Purification by column chromatography (SiO₂, gradient 40-60% EtOAc/hexane) afforded the title compound (2.97 g, 6.49 mmol, 92% from Boc-L-2-Nal-OH) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.89-7.86 (m, 1H), 7.84-7.81 (m, 1H), 7.80-7.76 (m, 3H), 7.65 (s, 1H), 7.47-7.40 (m, 2H), 7.34 (d, *J* = 8.2 Hz, 1H), 6.98 (d, *J* = 7.1 Hz, 1H), 4.50 (q, *J* = 5.9 Hz, 1H), 4.32 (t, *J* = 5.1 Hz, 1H), 3.85 (t, *J* = 7.1, 1H), 3.31-3.26 (m, 2H), 3.11-2.93 (m, 4H), 1.38-1.23 (m, 4H), 1.33 (s, 9H), 1.03 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ172.3, 170.3, 155.1, 135.3, 132.9, 131.8, 127.9, 127.5, 127.4, 127.4, 127.4, 125.9, 125.3, 78.2, 60.3, 53.6, 50.2, 38.5, 38.0, 29.7, 28.1, 25.6, 17.9. HRMS m/z $[M + H]^+$ calc'd for $C_{25}H_{35}N_3O_5H$ 458.2655, found 458.2661.

tert-Butyl ((*S***)-1-(((***S***)-3-(naphthalen-2-yl)-1-oxo-1-((4-((perfluorocyclopent-1-en-1 yl)oxy)butyl)amino)propan-2-yl)amino)-1-oxopropan-2-yl)carbamate (2.18)**

A solution of **2.17** (569 mg, 1.24 mmol) in DMF (12 ml) was cooled to 0 °C and added octafluorocyclopentene (1M in MeCN, 3.73 ml, 3.73 mmol) and Et3N (691 μl, 4.96 mmol). The resulting solution was allowed to stir for 3 hr at rt. The reaction mixture was diluted with EtOAc. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. Purification by column chromatography $(SiO₂,$ gradient 50-80% EtOAc/hexane) afforded the title compound (420 g, 0.647 mmol, 52%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 7.98-7.93 (m, 1H), 7.87-7.77 (m, 4H), 7.67 (s, 1H), 7.49-7.41 (m, 2H), 7.37 (d, *J* = 8.3 Hz, 1H), 7.00 (d, *J* = 7.0 Hz, 1H), 4.53 (q, *J* = 6.8 Hz, 1H), 4.40-4.30 (m, 2H), 3.92-3.86 (m, 1H), 3.17-3.08 (m, 2H), 3.06-2.92 (m, 2H), 1.54-1.15 (m, 4H), 1.35 (s, 9H), 1.07 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.4, 170.4, 155.2, 135.7-135.2 (m, 1C), 135.2,

134.3-131.6 (m, 1C), 132.9, 131.8, 127.8, 127.5, 127.4, 127.4 (2C), 125.8, 125.3, 113.8-106.2 (m, 3C), 78.2, 73.6 (d, 4.1 Hz, 1C), 53.7, 50.1, 38.0, 37.8, 28.1 (3C), 25.9, 24.8, 17.9. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ;.-112.5-(-112.9) (m, 2F), -114.6-(-115.0) (m, 2F), -128.1-(--128.5) (m, 2F), $-161.0-(-161.5)$ (m, 1F). HRMS m/z [M + H]⁺ calc'd for C₃₀H₃₄F₇N₃O₅H 650.2465, found 650.2494.

(*S***)-2-Amino-3-(1***H***-imidazol-4-yl)-***N***-((***S***)-1-(((***S***)-3-(naphthalen-2-yl)-1-oxo-1-((4- ((perfluorocyclopent-1-en-1-yl)oxy)butyl)amino)propan-2-yl)amino)-1-oxopropan-2 yl)propanamide (2.19)**

TFA (1 ml) was added to a solution of **2.18** (324 g, 0.498 mmol) in DCM (3 ml) and stirred for 1hr at rt. The reaction mixture was concentrated by rotary evaporation to give a crude solid (331 mg, 0.498 mmol). The residue (55 mg, 0.0831 mmol, analytical) was dissolved in DMF (1 ml) and added Boc-L-His-OH (25 mg, 0.0997 mmol), K-Oxyma (16 mg, 0.0914 mmol), Et₃N (35 μl, 0.249 mmol) and EDC·HCl (24 mg, 0.125 mmol). The resulting solution was allowed to stir for 5 hrs at rt. The reaction mixture was cooled to 0 $^{\circ}$ C, added H₂O and then extracted with EtOAc. The organic layer was washed with brine (x5), dried over $Na₂SO₄$ and concentrated. The residue was dissolved in DCM (2 ml), added TFA (1 ml) and stirred for 1hr at rt. The reaction mixture was concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (45-75% MeCN/H2O) to give the title compound (34 mg, 0.0372 mmol, 45% from **18**) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 14.43 (brs, 1H), 8.91 (s, 1H), 8.68 (d, *J* = 2.3 Hz, 1H), 8.53-8.20 (m, 2H), 8.48 (d, *J* = 7.9 Hz, 1H), 8.03 (t, *J* = 5.2 Hz, 1H), 7.85-7.78 (m, 3H), 7.70 (s, 1H), 7.50-7.36 (m, 4H), 4.60-4.52 (m, 1H), 4.41-4.24 (m, 3H), 4.12 (t, *J* = 6.1 Hz, 1H), 3.20-2.88 (m, 6H), 1.50- 1.26 (m, 4H), 1.23 (d, *J* = 7.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 170.2, 167.0, 158.4 (q, *J* = 31.6 Hz, 2CF3**C**O2H), 135.7-135.2 (m, 1C), 135.1, 134.5, 134.3-131.4 (m, 1C), 132.9, 131.8, 127.9, 127.6, 127.5, 127.4, 127.4, 125.9, 125.4, 120.7-106.1 (m, 3C and 2**C**F3CO2H), 73.6 (d, *J* = 4.1 Hz, 1C), 54.4, 51.2, 48.7, 38.1, 37.7, 26.5, 25.8, 24.8, 18.0. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -73.3 (s, 6F), -112.7 (s, 2F), -114.9 (s, 2F), -128.3 (s, 2F). HRMS m/z [M $+ H$ ⁺ calc'd for C₃₁H₃₃F₇N₆O₄H 687.2530, found 687.2524.

Macrocyclic Product (2.20)

Cs2CO³ (14 mg, 0.0437 mmol) was added at 0 °C to a solution of **2.19** (10 mg, 0.0109 mmol) in DMF (1 ml) and stirred for 2 hrs at rt,. The reaction mixture was added H₂O at 0 °C, extracted with EtOAc. The combined extract was washed with brine (x5), dried over $Na₂SO₄$ and concentrated. Purification by column chromatography (SiO₂, gradient 5-15% MeOH/CHCl₃) afforded the title compound (3.8 mg, 0.00425 mmol, 39%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.64 (d, *J* = 6.5 Hz, 1H), 8.35 (d, *J* = 7.8 Hz, 8.19-8.12 (m, 3H), 8.07 (s, 1H), 7.87-7.76 (m, 4H), 7.67 (s, 1H), 7.44 (qn, *J* = 6.7 Hz, 2H), 7.40-7.36 (m, 2H), 4.57-4.51 (m, 1H), 4.33-4.16 (m, 3H), 4.14- 4.09 (m, 1H), 3.37-3.22 (m, 2H), 3.19-3.13 (m, 1H), 3.08-3.02 (m, 1H), 2.94-2.88 (m, 1H), 1.75- 1.67 (m, 2H), 1.55-1.48 (m, 2H), 1.11 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.6, 170.5, 167.7, 158.3 (q, *J* = 35.8 Hz, CF3**C**O2H), 143.8-143.0 (m, 1C), 138.2, 135.6, 135.6, 132.9, 131.8, 127.9, 127.5, 127.4, 127.3, 125.9, 125.4, 118.6-107.3 (m, 3C), 117.7, 115.8 (q, *J* = 292.8 Hz, **C**F3CO2H), 112.1-111.4 (m, 1C), 73.0, 54.4, 51.5, 49.2, 37.9, 36.8, 28.5, 25.9, 24.5, 17.6. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -74.2 (s, 6F), -106.7 and -107.6 (AB quartet, *J* = 244 Hz, 2F), -110.8 and -111.3 (AB quartet, $J = 256$ Hz, 2F), -127.5 (s, 2F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{31}H_{32}F_6N_6O_4H$ 667.2468, found 667.2442.

Ac-CFAC-NH² (2.21)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.19 (d, *J* = 7.0 Hz, 1H), 8.09 (d, *J* = 8.0 Hz, 1H), 8.00 (d, *J* = 7.9 Hz, 1H), 7.92 (d, *J* = 8.0 Hz, 1H), 7.37 (s, 1H), 7.27-7.15 (m, 6H), 4.53-4.47 (m, 1H), 4.35-4.25 (m, 3H), 3.06 (dd, *J* = 14.0, 4.0 Hz), 2.84-2.66

(m, 4H), 2.61-2.54 (m, 1H), 2.24 (t, *J* = 8.5 Hz, 1H), 2.23 (t, *J* = 8.4 Hz, 1H), 1.83 (s, 3H), 1.24 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.0, 171.3, 170.8, 169.9, 169.5, 137.7, 129.2, 128.0, 126.3, 55.0, 54.6, 53.9, 48.5, 37.1, 26.1, 26.1, 22.5, 17.8. HRMS m/z [M + Na] ⁺ calc'd for C20H29N5O5S2Na 506.1508, found 506.1528.

Macrocyclic Product (2.22)

To a solution of **2.21** (56 mg, 0.116 mmol) in DMF (12 ml) was added Et3N (40 μl, 0.290 mmol) and octafluorocyclopentene (1M in MeCN, 174 μl, 0.174 mmol) over 3 minutes at 0 °C. The resulting solution was allowed to stir for 2 hr at 0 °C. The reaction mixture was diluted with EtOAc. The organic layer was washed with brine (5 times), dried over $Na₂SO₄$, filtered and the solvent was removed by rotary evaporation. Purification by preparative reverse-phase HPLC (55-90% MeCN/H₂O) afforded the title compound (39 mg, 0.0595 mmol, 51%) as a white powder. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.57 (d, *J* = 8.6 Hz, 1H), 8.42 (d, *J* = 8.8 Hz, 1H), 8.29 (d, *J* = 8.1 Hz, 1H), 7.69 (d, *J* = 7.2 Hz, 1H), 7.49 (s, 1H), 7.39 (s, 1H), 7.28-7.24 (m, 2H), 7.23-7.16 (m, 3H), 4.63- 4.58 (m, 1H), 4.51-4.45 (m, 1H), 4.40-4.32 (m, 2H), 3.63 (dd, *J* = 13.9, 3.3 Hz, 1H), 3.36 (dd, *J* = 12.6, 8.4 Hz, 1H), 3.26 (dd, *J* = 12.6, 4.3 Hz, 1H), 3.02 (dd, *J* = 13.9, 10.7 Hz, 1H), 2.96 (dd, *J* = 13.9, 4.7 Hz, 1H), 2.90 (dd, *J* = 13.9, 10.0 Hz, 1H), 1.80 (s, 3H), 1.22 (d, *J* = 6.6 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.9, 170.6, 170.3, 169.0, 168.2, 138.2-137.7 (m, 1C), 137.3, 136.0- 135.4 (m, 1C), 129.1, 128.2, 126.5, 118.6-108.2 (m, 3C), 55.7, 51.0, 51.0, 48.3, 37.2, 34.0, 31.8, 22.2, 18.7. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -100.0 and -100.9 (AB quartet, *J* = 68.4 Hz, 2F), - 108.2 and -109.0 (AB quartet, 192.8 Hz, 2F), -127.9 (s, 2F). HRMS m/z [M + H]⁺ calc'd for C25H27F6N5O5S2H 656.1436 found 656.1415.

Ac-Y(2-Nal)AC-NH² (2.23)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.13 (br s, 1H), 8.25 (d, *J* = 6.9 Hz, 1H), 8.09 (d, *J* = 8.0, 1H), 7.95 (d, *J* = 8.1, 1H), 7.93 (d, *J* = 8.3 Hz, 1H), 7.86- 7.78 (m, 3H), 7.74 (s, 1H), 7.49-7.41 (m, 3H), 7.39 (s, 1H), 7.22 (s, 1H), 6.92 (d, *J* = 8.3 Hz, 2H), 6.58 (d, *J* = 8.3 Hz, 2H), 1.62 (td, *J* = 8.8, 4.3 Hz, 1H), 4.37-4.28 (m, 3H), 3.24 (dd, *J* = 13.9, 4.1 Hz, 1H), 2.99 (dd, *J* = 13.9, 9.4 Hz, 1H), 2.85-2.69 (m, 3H), 2.56-2.50 (m, 1H), 2.24 (t, *J* = 8.4 Hz, 1H), 1.67 (s, 3H), 1.26 (d, *J* = 7.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.0, 171.5, 171.3, 170.8, 169.1, 155.7, 135.4. 133.0, 131.8, 129.9, 128.0, 127.6, 127.5, 127.4, 127.4, 125.9, 125.4, 114.8, 54.6, 54.3, 53.7, 48.6, 37.5, 36.5, 26.1, 22.4, 17.8. HRMS m/z [M + H]⁺ calc'd for $C_{30}H_{35}N_5O_6$ SH 594.2386, found 594.2383.

Macrocyclic Product (2.24)

Synthesized according to General Procedure C. The reaction mixture was triturated by adding H₂O and MeOH instead of diluted with EtOAc, and isolated by centrifugation. The peptide pellet was repeated washed with EtOAc to give the title compound (24 mg, 0.0313 mmol, 58%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.52 (d, *J* = 8.5 Hz, 1H), 8.48 (d, *J* = 9.0 Hz, 1H), 8.08 (d, *J* = 8.4 Hz), 7.86 (d, *J* = 7.8 Hz, 1H), 7.81 (d, *J* = 7.8 Hz, 1H), 7.78 (d, *J* = 8.4 Hz, 1H), 7.63 (s, 1H), 7.57 (s, 1H), 7.54 (d, *J* = 7.7 Hz, 1H), 7.50-7.43 (m, 2H), 7.33 (s, 1H), 7.29 (d, *J* = 8.4 Hz, 1H), 7.19 (d, *J* = 8.5 Hz, 2H), 7.11 (d, *J* = 8.5 Hz, 2H), 4.76-4.70 (m, 1H), 4.46 (q, *J* = 8.0 Hz, 1H), 4.38 (td, *J* = 8.9, 4.9 Hz, 1H), 4.28 (qn, *J* = 6.9 Hz, 1H), 3.21 (dd, *J* = 13.4, 7.6 Hz, 1H), 3.11-2.92 (m, 4H), 2.71 (dd, *J* = 13.9, 3.2 Hz, 1H), 1.61 (s, 3H), 1.12 (d, *J* = 6.8 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.1, 170.5, 170.1, 169.5, 168.5, 151.5, 148.6-148.1 (m, 1C), 135.2, 134.8, 132.9, 131.9, 130.1, 127.8, 127.8, 127.6, 127.5, 127.4, 125.8, 125.4, 122.6-122.1 (m, 1C), 118.5, 116.8-107.4 (m, 3C), 54.9, 52.0, 51.8, 47.9, 38.1, 36.0, 32.7, 22.2, 19.1. 19F NMR (DMSO*d*6, 282 MHz) δ -103.6 (d, *J* = 249.7 Hz, 1F), -107.9 (d, *J* = 257.5 Hz, 1F), -109.2 (d, *J* = 249.7 Hz,

1F), -114.5 (d, *J* = 257.5 Hz, 1F), -128.2 and -129.6 (AB quartet, *J* = 235.4 Hz, 2F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{35}H_{33}F_{6}N_{5}O_{6}SH$ 766.2134, found 766.2152.

Ac-YFAH-NH2·TFA (2.25)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 14.13 (br s, 2H), 9.16 (br s, 1H), 8.95 (s, 1H), 8.16 (d, *J* = 6.6 Hz, 1H), 8.09 (d, *J* = 8.2 Hz, 1H), 8.01 (s, *J* = 8.3 Hz, 1H), 7.99 (d, *J* = 8.3 Hz, 1H), 7.37-7.14 (m, 8H), 6.94 (d, *J* = 8.5 Hz, 2H), 6.60 (d, *J* = 8.5 Hz, 2H), 4.52-4.46 (m, 2H), 4.35-4.29 (m, 1H), 4.21 (qn, *J* = 6.9 Hz, 1H), 3.12 (dd, *J* = 15.3, 5.3 Hz, 1H), 3.04 (dd, *J* = 14.1, 4.1 Hz, 1H), 2.94 (dd, *J* = 15.3 Hz, 1H), 2.81-2.73 (m, 2H), 2.56-2.51 (m, 1H), 1.73 (s, 3H), 1.20 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 171.6, 171.6, 171.1, 169.4, 158.0 (TFA), 155.7, 137.7, 133.7, 130.0, 129.4, 129.2, 128.0, 127.9, 126.3, 116.8, 114.8, 54.4, 53.7, 51.4, 48.7, 37.1, 36.4, 26.8, 22.5, 17.5. HRMS m/z $[M + H]^{+}$ calc'd for C₂₉H₃₅N₇O₆H 578.2727, found 578.2745.

Macrocyclic Product (2.26)

Synthesized according to General Procedure C. Purification by preparative reverse-phase HPLC (50-85% MeCN/H2O) afforded the title compound (37 mg, 0.0494 mmol, 35%) as a white powder. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.64 (d, *J* = 6.9 Hz, 1H), 8.03 (d, *J* = 8.0 Hz, 1H), 7.96 (d, *J* = 5.3 Hz, 1H), 7.40 (s, 1H), 7.33 (s, 1H), 7.29-7.24 (m, 3H), 7.23-7.17 (m, 3H), 7.12 (s, 1H), 7.06 (s, 1H), 6.93 (d, *J* = 8.6 Hz, 2H), 6.74 (d, *J* = 8.6 Hz, 2H), 4.63-4.58 (m, 1H), 4.30-4.16 (m, 3H), 3.02- 2.97 (m, 2H), 2.94-2.87 (m, 2H), 2.85-2.72 (m, 2H), 1.82 (s, 3H), 1.31 (d, *J* = 7.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.1, 171.8, 171.2, 170.4, 168.9, 150.2, 142.0-141.5 (m, 1C), 139.3, 137.9, 137.2, 134.5, 130.9, 129.0, 128.2, 126.5, 116.9, 115.2, 114.5-113.9 (m, 1C), 115.9- 116.8 (m, 3C), 56.0, 52.3, 52.1, 48.7, 37.1, 36.7, 30.1, 22.3, 18.4. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -107.1 and -108.9 (AB quartet, *J* = 246.6 Hz, 2F), -114.1 and –115.7 (AB quartet, *J* = 254.3 Hz, 2F), -128.4 (s, 2F). HRMS m/z [M + H]⁺ calc'd for C₃₄H₃₃F₆N₇O₆H 750.2475, found 750.2439.

Ac-SFAC-NH² (2.27)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.16 (d, *J* = 6.9 Hz, 1H), 8.06 (d, *J* = 7.9 Hz, 1H), 7.92 (d, *J* = 7.6 Hz, 7.88 (d, *J* = 8.0 Hz, 1H), 7.35 (br s, 1H), 7.29-7.18 (m, 6H), 4.99 (br s, 1H), 4.54-4.48 (m, 1H), 4.36-4.24 (m, 3H), 3.49 (d, *J* = 5.9 Hz, 2H), 3.10 (dd, *J* = 14, 3.9 Hz, 1H), 2.87-2.80 (m, 2H), 2.78-2.71 (m, 1H), 2.27 (t, *J* = 8.4 Hz, 1H), 1.85 (s, 3H), 1.25 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.0, 171.3, 170.9, 170.4, 169.5, 137.7, 129.2, 128.0, 126.2, 61.7, 55.1, 54.7, 53.9, 48.6, 36.9, 26.1, 22.5, 17.6. HRMS m/z $[M + Na]^{+}$ calc'd for $C_{20}H_{29}N_{5}O_{6}S$ Na 490.1736, found 490.1759.

Macrocyclic Product (2.28)

Synthesized according to General Procedure C. After work-up and concentration, the residue was triturated by with EtOAc, and isolated by centrifugation to give the title compound (41 mg, 0.0641 mmol, 56%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.54 (d, *J* = 8.1 Hz, 1H), 8.33 (d, *J* = 9.1 Hz, 1H), 8.30 (d, *J* = 7.6 Hz, 1H), 7.71 (d, *J* = 7.6 Hz, 1H), 7.45 (s, 1H), 7.34 (s, 1H), 7.31- 7.26 (m, 2H), 7.25-7.19 (m, 3H), 4.66-4.53 (m, 3H), 4.46-4.41 (m, 1H), 4.34-4.28 (m, 2H), 3.46 (dd, *J* = 13.9, 3.0 Hz, 1H), 3.00-2.90 (m, 3H), 1.80 (s, 3H), 1.19 (d, *J* = 6.8 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 170.8, 170.7, 169.4, 168.4, 151.5-151.1 (m, 1C), 137.1, 129.0, 128.3, 126.6, 118.6-107.6 (m, 3C), 111.4-110.8 (m, 1C), 71.4, 56.2, 52.0, 51.4, 48.4, 36.9, 35.1, 22.2, 18.1. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -100.0 (d, *J* = 245.1 Hz, 1F), -107.2 (d, *J* = 245.1 Hz, 1H), -110.2 (d, *J* = 255.7 Hz, 1F), -115.7 (d, *J* = 255.7 Hz, 1F), -128.2 and -129.2 (AB quartet, *J* $= 235.4$ Hz, 2F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{25}H_{27}F_{6}N_{5}O_{6}SH$ 640.1664, found 640.1668.

Ac-YWTC-NH² (2.29)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.81 (d, *J* = 1.4 Hz, 1H), 9.13 (br s, 1H), 8.20 (d, *J* = 7.7 Hz, 1H), 7.98 (d, *J* = 8.3 Hz, 1H), 7.91 (d, *J* = 7.8 Hz, 1H), 7.90 (d, *J* = 7.8 Hz, 1H), 7.60 (d, *J* = 7.9 Hz, 1H), 7.35 (s, 1H), 7.31 (d, *J* = 8.1 Hz, 1H), 7.29 (br s, 1H), 7.17 (d, *J* = 1.8 Hz, 1H), 7.05 (t, *J* = 7.5 Hz, 1H), 7.00-6.94 (m, 3H), 6.60 (d, *J* = 8.4 Hz, 2H), 4.65-4.60 (m, 1H), 4.42-4.34 (m, 2H), 4.30 (dd, *J* = 8.0, 3.9 Hz, 1H), 4.08-4.02 (m, 1H), 3.20 (dd, *J* = 15.0, 4.4 Hz, 1H), 3.00 (dd, *J* = 15.0, 9.3 Hz, 1H), 2.90-2.73 (m, 3H), 2.56 (dd, *J* = 14.0, 10.4 Hz, 1H), 2.32 (t, *J* = 8.5 Hz, 1H), 1.71 (s, 3H), 1.03 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.7, 171.7, 171.4, 169.7, 169.2, 155.7, 136.0, 130.0, 128.1, 127.3, 123.6, 120.8, 118.5, 118.2, 114.8, 111.2, 110.0, 66.5, 58.1, 54.7, 54.2, 53.4, 36.6, 27.2, 26.0, 22.4, 19.3. HRMS m/z $[M + H]^{+}$ calc'd for $C_{29}H_{36}N_6O_7S$ 613.2444, found 613.2441.

Macrocyclic Product (2.30)

A flask was charged with linear precursor **2.29** (24 mg, 0.0392 mmol) and DMF (6 ml). To the mixture octafluorocyclopentene (1M in MeCN, 196 μl, 0.196 mmol) and Et₃N (8 μl, 0.0588 mmol) were added at 0 °C. The mixture was stirred for 5 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was dropped to a suspension of $Cs₂CO₃$ (13 mg, 0.0392 mmol) in DMF (2 ml) at 0 °C over 3 min and stirred for 1 hr at 0 °C. The mixture was quenched with AcOH (7µl, 0.0588 mmol) and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (55%-75% MeCN/H₂O) afforded the title compound (13 mg, 0.0166 mmol, 42%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.85 (d, *J* = 1.9 Hz, 1H), 8.57 (d, *J* = 9.2 Hz, 1H), 8.38 (d, *J* = 8.8 Hz, 1H), 8.25 (d, *J* = 8.4 Hz, 1H), 7.57 (s, 1H), 7.51 (d, *J* = 8.1 Hz, 1H), 7.46 (s, 1H), 7.45 (d, *J* = 8.0 Hz, 1H), 7.31 (d, *J* = 8.1 Hz, 1H), 7.20 (d, *J* = 8.6 Hz, 2H), 7.07 (d, *J* = 2.2 Hz, 1H), 7.06-7.02 (m, 3H), 6.96-6.92 (m, 1H), 5.51 (d, *J* = 4.7 Hz, 1H), 4.88-4.82 (m, 1H), 4.48-4.42 (m, 1H), 4.37-4.31 (m, 2H), 3.87-3.81 (m, 1H), 3.54 (dd, *J* = 14.0, 5.2 Hz, 1H), 3.08 (dd, *J* = 14.0, 9.6 Hz, 1H), 3.05- 2.93 (m, 3H), 2.76 (dd,*J* = 14.4, 3.7 Hz, 1H), 1.83 (s, 3H), 0.81 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.1, 171.0, 169.4, 168.8, 168.3, 151.5, 148.8-148.2 (m, 1C), 136.1, 135.2, 130.0, 127.2, 124.3-123.7 (m, 1C), 120.8, 118.3, 118.1, 118.0, 115.8-109.6 (m, 3C), 111.2, 109.3, 79.2, 66.3, 56.8, 54.6, 52.2, 51.6, 35.9, 32.6, 28.4, 22.4, 17.9. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -101.7 (d, *J* = 249.5 Hz, 1F), -107.0 (d, *J* = 254.4 Hz, 1F), -110.1 (d, *J* = 248.3 Hz, 1F), - 116.2 (d, *J* = 254.4 Hz, 1F), -128.4 and -129.5 (AB quartet, *J* = 235.3 Hz, 2F). HRMS m/z [M + H]⁺ calc'd for $C_{34}H_{34}F_6N_6O_7SH$ 785.2192, found 785.2200.

Ac-YAC-NH² (2.31)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.15 (s, 1H), 8.26 (d, *J* = 6.9 Hz, 1H), 8.04 (d, *J* = 8.1 Hz, 1H), 7.83 (d, *J* = 8.0 Hz, 1H), 7.35 (s, 1H), 7.21 (s, 1H), 7.03 (d, *J* = 8.3 Hz, 2H), 6.63 (d, *J* = 8.3 Hz, 2H), 4.44-4.36 (m, 1H), 4.33-4.28 (m, 1H), 4.27- 4.19 (qn, *J* = 6.9 Hz, 1H), 2.94-2.86 (m, 1H), 2.84-2.77 (m, 1H), 2.77-2.70 (m, 1H), 2.60 (dd, *J* = 13.6, 10.4 Hz, 1H), 2.24 (t, *J* = 8.4 Hz, 1H), 1.75 (s, 3H), 1.24 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 171.7, 171.3, 169.3, 155.7, 130.1, 128.1, 114.8, 54.5, 54.3, 48.6, 36.6, 26.1, 22.5, 17.6. HRMS m/z [M + Na]⁺ calc'd for C₁₇H₂₄N₄O₅SNa 419.1365, found 419.1385

Macrocyclic Product (2.32)

Synthesized according to General Procedure C. Purification by trituration with DMF and CHCl₃ afforded the title compound (33 mg, 0.0580 mmol, 46%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.21 (d, *J* = 7.3 Hz, 1H), 7.95 (d, *J* = 8.2 Hz, 1H), 7.69 (d, *J* = 8.6 Hz, 1H), 7.43 (brs, 1H), 7.39-7.18 (brm, 2H), 7.31 (brs, 1H), 7.09-7.01 (m, 2H), 4.61-4.56 (m, 1H), 4.39 (td, *J* = 8.9, 3.5 Hz, 1H), 4.30 (qn, *J* = 7.1 Hz, 1H), 3.28 (dd, *J* = 12.0, 3.6 Hz, 1H), 3.10 (dd, *J* = 12.0, 9.5 Hz, 1H),

2.94-2.89 (m, 1H), 2.84-2.77 (m, 1H), 1.87 (s, 3H), 1.08 (d, *J* = 6.8 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.3, 170.2, 169.5, 169.2, 168.8, 151.1, 147.6-146.9 (m, 1C), 134.8, 130.7, 118.9- 118.3 (m, 1C), 118.2, 116.8-108.9 (m, 3C), 54.0, 50.6, 47.5, 37.5, 34.3, 22.5, 18.3. ¹⁹F NMR (DMSO-d6, 282 MHz) δ -102.8 and -103.9 (AB quartet, *J* = 245.0 Hz, 2F), -115.0 and -115.5 (AB quartet, J = 192.0 Hz, 2F), -128.3 and -129.0 (AB quartet, J = 177.1 Hz, 2F). HRMS m/z [M + Na]⁺ calc'd for $C_{22}H_{22}F_6N_4O_5S$ Na 591.1113, found 591.1141.

Ac-YNTQC-NH² (2.33)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.15 (brs, 1H), 8.42 (d, *J* = 7.4 Hz, 1H), 8.08 (d, *J* = 8.3 Hz, 1H), 8.04 (d, *J* = 7.6 Hz,1H), 7.79 (d, *J* = 8.2 Hz, 1H), 7.71 (d, *J* = 7.7 Hz, 1H), 7.50 (s, 1H), 7.28-7.19 (m, 3H), 7.04-6.99 (m, 1H), 7.03 (d, *J* = 8.4 Hz, 2H), 6.79 (brs, 1H), 6.62 (d, *J* = 8.4 Hz, 2H), 4.95 (brs, 1H), 4.60 (q, *J* = 6.9 Hz, 1H), 4.45-4.39 (m, 1H), 4.29-4.24 (m, 1H), 4.24-4.17 (m, 1H), 4.13 (dd, *J* = 7.7, 3.5 Hz, 1H), 4.10-4.03 (m, 1H), 2.89-2.77 (m, 2H), 2.74-2.44 (m, 4H), 2.27 (t, *J* = 8.5 Hz, 1H), 2.19-2.09 (m, 2H), 2.01-1.93 (m, 1H), 1.87-1.78 (m, 1H), 1.75 (s, 3H), 1.06 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.9, 171.9, 171.9, 171.5, 171.4, 171.3, 170.3, 169.4, 155.7, 130.1, 128.0, 114.8, 66.3, 58.6, 55.0, 54.3, 52.8, 49.9, 36.8, 36.7, 31.6, 27.3, 26.0, 22.4, 19.7. HRMS m/z [M + H] ⁺ calc'd for $C_{27}H_{40}N_8O_{10}SH$ 669.2666, found 669.2653.

Macrocyclic Product (2.34)

A flask was charged with linear precursor **33** (40 mg, 0.0598 mmol) and DMF (6 mL). The mixture was added octafluorocyclopentene (1M in MeCN, 299 μl, 0.299 mmol) and Et₃N (12.5μl, 0.0897 mmol) at 0 °C. The mixture was stirred for 30 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added $Cs₂CO₃$ (19 mg, 0.0598 mmol) at 0 °C and stirred for 1 hr at rt. The mixture was added AcOH (17 μl, 0.299 mmol) and concentrated by rotary evaporation. Purification by preparative reversephase HPLC (30%-70% MeCN/H₂O) afforded the title compound (25 mg, 0.0297 mmol, 50%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.56 (d, *J* = 5.3 Hz, 1H), 8.43 (d, *J* = 6.3 Hz, 1H), 7.75 (d, *J* = 7.0 Hz, 1H), 7.68 (d, *J* = 8.0 Hz, 1H), 7.52 (s, 1H), 7.44 (d, *J* = 6.6 Hz, 1H), 7.30 (brs, 1H), 7.28 (d, *J* = 8.6 Hz, 2H), 7.19 (brs, 1H), 7.11 (d, *J* = 8.6 Hz, 2H), 7.07 (brs, 2H), 6.77 (brs, 1H), 4.92 (brs, 1H), 4.39-4.33 (m, 2H), 4.27-4.21 (m, 1H), 4.05-3.97 (m, 2H), 3.93 (dd, *J* = 6.6, 5.2 Hz, 1H), 3.25 (dd, *J* = 12.7 4.3 Hz, 1H), 3.02-2.89 (m, 3H), 2.61-2.51 (m, 2H), 2.18-2.10 (m, 1H), 2.10-2.01 (m, 1H), 1.94 (s, 3H), 1.90-1.82 (m, 1H), 1.79-1.70 (m, 1H), 1.05 (d, *J* = 6.3 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.6, 171.9, 171.9, 171.7, 171.6, 170.7, 170.5, 170.3, 151.5, 148.5-147.9 (m, 1C), 135.0, 130.9, 119.2-118.6 (m, 1C), 118.9, 118.6-106.7 (m, 3C), 66.0, 60.3, 55.9, 53.5, 53.0, 51.3, 36.8, 35.7, 34.0, 31.4, 26.9, 22.5, 19.6. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -103.5 and -104.6 (AB quartet, *J* = 250.8 Hz, 2F), -113.7 and -114.8 (AB quartet, *J* = 252.8 Hz, 2F), -128.7 (brd, J = 5.2 Hz, 2F). HRMS m/z [M + Na]⁺ calc'd for C₃₂H₃₈F₆N₈O₁₀SNa 863.2233, found 863.2222.

Ac-YQY-NH² (2.35)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.14 (brs, 2H), 8.22 (d, *J* = 7.5 Hz, 1H), 8.06 (d, *J* = 8.0 Hz, 1H), 7.73 (d, *J* = 8.1 Hz, 1H), 7.30 (s, 1H), 7.23 (s, 1H), 7.07 (s, 1H), 7.03 (d, *J* = 8.5 Hz, 2H), 6.99 (d, *J* = 8.5 Hz, 2H), 6.78 (s, 1H), 6.63 (d, *J* = 8.5 Hz, 2H), 6.62 (d, *J* = 8.5 Hz, 2H), 4.40-4.34 (m, 1H), 4.34-4.28 (m, 1H), 4.18-4.12 (m, 1H), 2.91- 2.82 (m, 2H), 2.72 (dd, *J* = 8.3, 14.0 Hz, 1H), 2.60 (dd, *J* =10.5, 14.0 Hz, 1H), 2.12-1.99 (m, 2H), 1.88-1.80 (m, 1H), 1.76 (s, 3H), 1.77-1.67 (m, 1H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.9, 172.8, 171.9, 170.8, 169.5, 155.8, 155.7, 130.1, 130.0, 128.1, 127.7, 114.9, 114.8, 54.6, 54.0, 52.6, 36.7, 36.5, 31.4, 27.6, 22.5. HRMS m/z [M + Na]⁺ calc'd for C₂₅H₃₁N₅O₇Na 536.2121, found 536.2083.
Macrocyclic Product (2.36)

A flask was charged with linear precursor **2.35** (45 mg, 0.0870 mmol) and DMF (8.7 mL). The mixture was added $Cs₂CO₃$ (43 mg, 0.131 mmol) and octafluorocyclopentene (1M in MeCN, 131 μl, 0.0756 mmol) at rt. The mixture was stirred for 1 h. The mixture was added AcOH (25 μl, 0.435 mmol) and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (45%-75% MeCN/H2O) afforded the title compound (37 mg, 0.0539 mmol, 62%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.25 (d, *J* = 7.7 Hz, 1H), 8.19 (d, *J* = 8.5 Hz, 1H), 7.70 (d, *J* = 7.2 Hz, 1H), 7.56 (s, 1H), 7.28 (s, 1H), 7.24 (d, *J* = 8.6 Hz, 2H), 7.17 (s, 1H), 6.96 (d, *J* = 8.6 Hz, 2H), 6.78 (s, 1H), 6.69 (d, *J* = 8.6 Hz, 2H), 6.45 (d, *J* = 8.6 Hz, 2H), 4.54 (td, *J* = 7.0, 3.3 Hz, 1H), 4.35 (td, *J* = 8.6, 3.5 Hz, 1H), 4.31-4.25 (m, 1H), 2.95 (dd, *J* = 13.6, 7.2 Hz, 1H), 2.87-2.80 (m, 3H), 2.16-2.05 (m, 2H), 2.00-1.91 (m, 1H), 1.95 (s, 3H), 1.80 (m, 1H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 173.9, 173.7, 173.0, 170.8, 169.2, 169.0, 152.5, 152.1, 135.3, 134.1-132.9 (m, 2C), 131.2, 131.0, 116.6-106.9 (m, 3C), 116.2, 116.0, 53.9, 52.8, 52.1, 36.6, 36.5, 31.4, 27.5, 22.7. ¹⁹F NMR (DMSO-*d*6, 376 MHz) δ -112.6 and -113.4 (AB quartet, *J* = 249.6 Hz, 2F), -113.3 and -113.4 (AB quartet, J = 250.3 Hz, 2F), -128.5 (s, 2F). HRMS m/z [M + Na]⁺ calc'd for C₃₀H₂₉F₆N₅O₇Na 708.1869, found 708.1904.

Cyclic Product (2.38)

37[S2] (100 mg, 329 mmol) was added to a 100-mL round bottom flask equipped with a stir bar, diluted with DMF (28 mL), and allowed to cool to 0 °C. Then was added Et_3N (69 uL, 494 mmol) followed by addition of octafluorocyclopentene (3M in MeCN, 110 uL, 494 mol). The mixture was then allowed to stir for 30 min at 0 °C Then the mixture was added water (10 mL), extracted with EtOAc (3 x 10 mL). The organic layer was washed with brine (3 x 10 mL), dried over Na₂SO₄, filtered, and the volatiles removed under rotary evaporation to afford crude compound **38**. Purification by preparative reverse-phase HPLC (40-80% MeCN/H2O) afforded the title compound **38** (103 mg, 329 mmol, 71%) as a white-powder. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.24 (d, *J* = 4.8 Hz, 1H), 7.79 (d, *J* = 8.1 Hz, 1H), 7.55 (s, 1H), δ 7.22-7.17 (m, 4H), 7.15-7.14 (m, 2H), 4.51-4.47 (m, 2H), 3.32-3.29 (m, 1H), 2.98 (dd, *J* = 13.8, 4.6 Hz, 1H), 2.85 (dd, *J* = 13.8, 8.1 Hz, 1H), 2.79 (dd, *J* = 12.9, 3.1 Hz, 1H). ¹³C NMR (DMSO-*d*⁶ 126 MHz) δ 172.1, 168.2, 137.2, 137.0-136.6 (m, 1C), 129.2, 128.0, 126.3, 118.5 (m, 1C), 116.7-116.4 (m, 1C), 115.0-114.6 (m, 1C), 113.1-112.7 (m, 1C), 110.9-110.6 (m 1C), 108.8-108.4 (m, 1C), 96.9 (m, 1C), 53.7, 52.8, 37.7, 24.7. ¹⁹F NMR (DMSO-*d*6, 282 MHz) -100.6 (dd, *J* = 234.0, 12.2 Hz, 1F), -101.7 (dd, *J* = 225.6, 28.2 Hz, 1F), - 111.2 and -112.2 (AB quartet, *J* = 249.4 Hz, 2F), -125.4 and -128.0 (AB quartet *J* = 234.4, 2F). HRMS m/z $[M + Na]+$ calc'd for $C_{17}H_{15}F_6N_3O_2S$ Na 462.0687, found 462.0675.

H-C(2-Nal)AC-NH2·TFA (2.39)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.68 (d, *J* = 7.9 Hz, 1H), 8.55 (d, *J* = 6.9 Hz, 1H), 8.23-8.02 (br m, 3H), 8.00 (d, *J* = 7.9 Hz, 1H), 7.89-7.80 (m, 4H), 7.54-7.41 (m, 4H), 7.23 (s, 1H), 4.77-4.70 (m, 1H), 4.37-4.30 (m, 2H), 3.97-3.91 (m, 1H), 3.33-3.26 (m, 1H), 3.07-2.93 (m, 2H), 2.91-2.78 (m, 2H), 2.76-2.69 (m, 1H), 2.23 (br s, 1H), 1.28 (d, *J* = 7.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.0, 171.3, 170.7, 166.9, 135.3, 133.0, 131.9, 127.7, 127.6, 127.6, 127.5, 127.4, 126.0, 125.5, 54.5, 54.3, 53.6, 48.6, 37.3, 26.2, 25.5, 17.9. HRMS m/z $[M + H]$ ⁺ calc'd for C₂₂H₂₉N₅O₄S₂H 492.1739, found 492.1787.

Cyclic Product (2.40)

To a solution of **2.39** (70 mg, 0.116 mmol) in DMF (10 ml) was added Et3N (40 μl, 0.29 mmol) and octafluorocyclopentene (1M in MeCN, 173 μl, 0.173 mmol) over 3 minutes at 0 °C. The resulting solution was allowed to stir for 2 hr at 0 $^{\circ}$ C. The reaction mixture was diluted with EtOAc. The organic layer was washed with brine (5 times), dried over Na₂SO₄, filtered and the solvent was

removed by rotary evaporation. Purification by preparative reverse-phase HPLC (65-80% MeCN/H₂O) afforded the title compound (35 mg, 0.0450 mmol, 39%) as a white powder. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.41 (d, *J* = 7.1 Hz, 1H), 8.14 (d, *J* = 4.6 Hz, 1H), 8.04 (d, *J* = 4.3 Hz, 1H), 8.02 (d, *J* = 4.1 Hz, 1H), 7.85-7.82 (m, 1H), 7.81-7.78 (m, 1H), 7.75 (d, *J* = 8.5 Hz, 1H), 7.71 (s, 1H), 7.48-7.42 (m, 3H), 7.38 (dd, *J* = 8.4, 1.4 Hz, 1H), 7.23 (s, 1H), 4.69 (dt, *J* = 8.6, 4.0 Hz, 1H), 4.44-4.40 (m, 1H), 4.40-4.30 (m, 2H), 3.26-3.20 (m, 2H), 2.99 (dd, *J* = 13.9, 9.1 Hz, 1H), 2.86- 2.79 (m, 2H), 2.76-2.69 (m, 1H), 2.23 (t, *J* = 8.4 Hz, 1H), 1.27 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 171.3, 170.4, 168.4, 137.1-136.7 (m, 1C), 135.0, 133.0, 131.9, 127.8, 127.6, 127.4, 127.4, 125.9, 125.4, 118.7-108.5 (m, 3C), 96.6-96.1 (m, 1C), 54.6, 53.9, 52.4, 48.5, 37.9, 26.2, 25.0, 18.1. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -100.4 (dd, *J* = 240.4, 12.4 Hz, 1F), -101.7 (dd, *J* = 238.9, 15.1 Hz, 1F), -111.3 and -112.4 (AB quartet, *J* = 248.6 Hz, 2F), - 125.4 and -128.1 (AB quartet, $J = 235.8$ Hz, 2F). HRMS m/z $[M + H]$ ⁺ calc'd for $C_{27}H_{27}F_6N_5O_4S_2H$ 664.1487, found 664.1529.

Macrocyclic Product (2.41)

A flask was charged with linear precursor **2.22** (35 mg, 0.0724 mmol) and DMF (7 mL). The mixture was added octafluorocyclopentene (1M in MeCN, 109 μl, 0.109 mmol) and Et₃N (25 μl, 0.181 mmol) at 0 $^{\circ}$ C. The mixture was stirred for 30 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added PhOH (13.6 mg, 0.145 mmol) and $Cs₂CO₃$ (47 mg, 0.145 mmol) at 0 °C and stirred for 4 hr at rt. Then the mixture was added PhOH (13.6 mg, 0.145 mmol) and $Cs₂CO₃$ (47 mg, 0.145 mmol) and stirred for 2 hr at rt. The mixture was added AcOH (25 μl, 0.434 mmol) at 0 °C and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (45%-80% MeCN/H2O) afforded the title compound (11 mg, 0.0155 mmol, 21%) as an off-white solid. ¹H NMR (DMSO*d*6, 500 MHz) δ 9.46 (s, 1H), 8.53 (d, *J* = 8.6 Hz, 1H), 8.29 (d, *J* = 8.3 Hz, 1H), 8.03 (d, *J* = 8.8 Hz, 1H), 7.39 (t, *J* = 7.9 Hz, 2H), 7.29-7.12 (m, 8H), 7.04 (d, *J* = 9.4 Hz, 1H), 4.75 (dt, *J* = 9.4, 2.5 Hz, 1H), 4.58-4.51 (m, 1H), 4.49-4.41 (m, 1H), 4.29-4.20 (m, 1H), 3.62-3.53 (m, 1H), 2.98-2.88 (m, 2H), 2.88-2.75 (m, 3H), 1.72 (s, 3H), 1.20 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.3, 171.0, 169.2, 168.6, 168.1, 153.5, 142.7-142.0 (m, 1C), 136.9, 133.7-133.4 (m, 1C), 129.9, 129.1, 128.2, 126.5, 124.5, 118.8-113.7 (m, 2C), 116.4, 68.6-67.9 (m, 1C), 55.4, 51.7, 48.2, 46.4, 37.3, 30.7, 30.5, 22.3, 17.1. ¹⁹F NMR (DMSO-*d*6, 376 MHz) δ -103.0 (dd, *J* = 252.2, 13.8 Hz, 1F), -114.0 (dd, *J* = 223.6, 12.6 Hz, 1F), 119.6 (dd, *J* = 253.5, 11.9 Hz, 1F), -127.5 (dd, *J* = 223.0, 11.9 Hz, 1F). HRMS m/z [M + Na]⁺ calc'd for $C_{31}H_{31}F_4N_5O_6S_2N$ a 732.1550, found 732.1545.

Ac-YACFAC-NH² (2.42)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.14 (brs, 1H), 8.23 (d, *J* = 7.0 Hz, 1H), 8.19 (d, *J*= 7.1 Hz, 1H), 8.09 (d, *J* = 8.0 Hz, 1H), 8.01 (d, *J* = 8.3 Hz, 1H), 7.92 (d, *J* = 8.1 Hz, 1H), 7.89 (d, *J* = 8.1 Hz, 1H), 7.38 (brs, 1H), 7.27-7.15 (m, 6H), 7.02 (d, *J* = 8.4 Hz, 2H), 6.63 (d, *J* = 8.4 Hz, 2H), 4.55-4.49 (m, 1H), 4.42-4.21 (m, 5H), 3.06 (dd, *J* = 14.1, 4.0 Hz, 1H), 2.89-2.64 (m, 6H), 2.58 (dd, *J* = 14.0 Hz, 1H), 2.25 (dd, *J* = 9.2, 8.0 Hz, 1H), 2.23 (dd, *J* = 8.8, 8.0 Hz, 1H), 1.74 (s, 3H), 1.24 (d, *J* = 7.1 Hz, 3H), 1.18 (d, *J* = 7.1 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 172.1, 172.0, 171.5, 171.3, 170.8, 169.5, 169.2, 155.7, 137.6, 130.1, 129.2, 128.1, 128.1, 126.3, 114.8, 54.8, 54.6, 54.2, 53.9, 48.6, 48.2, 37.2, 36.7, 26.3, 26.1, 22.5, 17.8, 17.8. HRMS m/z [M + Na]⁺ calc'd for $C_{32}H_{43}N_7O_8S_2Na$ 740.2512, found 740.2492.

Macrocyclic Product (2.43)

A flask was charged with linear precursor **2.42** (36 mg, 0.0504 mmol) and DMF (10 mL). The mixture was added octafluorocyclopentene (1M in MeCN, 76 μl, 0.0756 mmol) and Et₃N (18 μl, 0.126 mmol) at 0 °C under argon. The mixture was stirred for 5 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added $Cs₂CO₃$ (66 mg, 0.202 mmol) at 0 °C and stirred for 4 hr at rt. The mixture was added AcOH (17 μl, 0.302 mmol) and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (40%-60% MeCN/H2O) afforded the title compound (12 mg, 0.0141 mmol, 28%) as an off-white solid. Major isomer: ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.48 (s, 1H), 8.53 (d, *J* = 8.1 Hz, 1H), 8.37 (d, *J* = 8.0 Hz, 1H), 8.08 (d, *J* = 7.8 Hz, 1H), 8.06 (d, *J* = 7.4 Hz, 1H), 7.52 (d, *J* = 8.5 Hz, 1H), 7.30 (d, *J* = 8.9 Hz, 1H), 7.28-7.12 (m, 6H), 7.09-6.87 (m, 3H), 4.70 (dt, *J* = 8.9, 2.7 Hz, 1H), 4.68-4.61 (m, 1H), 4.54-4.47 (m, 2H), 4.23 (dt, *J* = 8.1, 7.0 Hz, 1H), 4.06-3.99 (m, 1H), 3.54 (dd, *J* = 14.6, 2.6 Hz, 1H), 3.39-3.33 (m, 1H), 2.99 (dd, *J* = 13.3, 5.5 Hz, 1H), 2.94-2.85 (m, 3H), 2.83 (dd, *J* = 14.6, 2.1 Hz, 1H), 2.61 (dd, *J* = 13.1, 10.9 Hz, 1H), 1.86 (s, 3H), 1.18 (d, *J* = 7.2 Hz, 3H), 0.96 (d, *J* = 6.8 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.5, 171.4, 171.1, 168.8, 168.7, 168.1, 167.6, 151.9, 140.3-139.8 (m, 1C), 136.8, 134.6-134.3 (m, 1C), 132.3, 130.4, 129.0, 128.3, 126.6, 116.6-113.8 (m, 2C), 114.8, 69.0-68.2 (m, 1C), 54.8, 53.5, 50.5, 49.7, 46.7, 46.7, 37.7, 37.3, 33.5, 30.3, 22.5, 19.5, 16.8. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -103.5 (dd, *J* = 250.4, 13.1 Hz, 1F), -114.3 (dd, *J* = 223.0, 12.7 Hz, 1F), 119.5 (dd, *J* = 250.4, 13.6 Hz, 1F), -127.0 (d, $J = 223.0$ Hz, 1F). HRMS m/z $[M + Na]^+$ calc'd for $C_{37}H_{39}F_4N_7O_8S_2Na$ 872.2136, found 872.2136.

Ac-YNCTFC-NH² (2.44)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.15 (brs, 1H), 8.41 (d, *J* = 7.6 Hz, 1H), 8.09 (d, *J* = 8.0 Hz, 1H), 8.04 (d, *J* = 8.2 Hz, 1H), 7.96 (d, *J* = 7.8 Hz, 1H), 7.91 (d, *J* = 8.0 Hz, 1H), 7.86 (d, *J* = 8.1 Hz, 1H), 7.47 (brs, 1H), 7.26-7.14 (m, 7H), 7.03 (d, *J* = 8.5 Hz, 2H), 7.00 (brs, 1H), 6.63 (d, *J* = 8.5 Hz, 2H), 4.57-4.50 (m, 2H), 4.49-4.44 (m, 1H), 4.41-4.36 (m, 1H), 4.31-4.26 (m, 1H), 4.14 (dd, *J* = 8.0, 4.8 Hz, 1H), 3.95-3.90 (m, 1H), 3.07 (dd, *J* = 14, 5.1 Hz, 1H), 2.90-2.82 (m, 2H), 2.81-2.66 (m, 4H), 2.63-2.55 (m, 2H), 2.50-2.43 (m, 1H), 2.29 (t, *J* = 8.6 Hz, 1H), 2.22 (t, *J* = 8.5 Hz, 1H), 1.75 (s, 3H), 0.97 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.8, 171.8, 171.3, 171.0, 170.7, 169.8, 169.7, 169.5, 155.7, 137.5, 130.0, 129.2, 128.1, 128.0, 126.3, 114.8, 66.5, 58.7, 55.0, 55.0, 54.5, 54.0, 49.9, 37.0, 36.7, 36.6, 26.1, 26.0, 22.5, 19.5. HRMS m/z $[M + H]^+$ calc'd for $C_{34}H_{46}N_8O_{10}S_2H$ 791.2856, found 791.2849.

Intermediate 2.45 and 2.46

A flask was charged with linear precursor **2.44** (100 mg, 0.126 mmol) and DMF (25 mL). The mixture was added octafluorocyclopentene (1M in MeCN, 190 μl, 0.190 mmol) and Et₃N (44 μl, 0.315 mmol) at 0 °C under argon. The mixture was stirred for 5 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added $Cs₂CO₃$ (328 mg, 1.01 mmol) at rt and stirred for 2 hr. The mixture was added AcOH (108 μl, 1.89 mmol) and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (36%-57% MeCN/H2O) afforded **2.45** (9 mg, 0.00935 mmol, 7.4%) as an off-white solid., **2.46** (18 mg, 0.0191 mmol, 15%) as an off-white solid and **2.47** (23 mg, 0.0249 mmol, 20%) as an off-white solid. **2.45:** ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.11 (s, 1H), 8.70 (d, *J* = 7.8 Hz, 1H), 8.47 (d, *J* = 7.6 Hz, 1H), 8.43 (d, *J* = 9.2 Hz, 1H), 8.06 (d, *J* = 8.4 Hz, 1H), 7.89 (d, *J* = 7.2 Hz, 1H), 7.38 (s, 1H), 7.31 (s, 1H), 7.28 (d, *J* = 7.3 Hz, 1H), 7.25-7.08 (m, 5H), 6.97 (d, *J* = 8.3 Hz, 2H), 6.91 (s, 1H), 6.84 (s, 1H), 6.59 (d, *J* = 8.3 Hz, 2H), 5.07 (brs, 1H), 4.85-4.80 (m, 1H), 4.61- 4.55 (m, 2H), 4.42-4.36 (m, 1H), 4.36-4.30 (m, 1H), 4.05-3.99 (m, 2H), 3.58-3.42 (m, 3H), 2.94 (d, *J* = 6.1 Hz, 2H), 2.87 (dd, *J* = 14.3, 11.5 Hz, 1H), 2.79 (dd, *J* = 14.0, 2.8 Hz, 1H), 2.60 (dd, *J* = 15.7, 6.6 Hz, 1H), 2.55-2.51 (m, 1H), 2.35 (dd, *J* = 15.7, 6.9 Hz, 1H), 1.70 (s, 3H), 0.98 (d, *J* = 5.8 Hz, 3H). ¹³C NMR (DMSO-*d*₆, 126 MHz) δ 172.1, 171.3, 170.7, 170.4, 170.2, 169.6, 169.1, 168.4, 155.6, 140.5-139.9 (m, 1C), 136.8, 135.5-134.8 (m, 1C), 129.9, 129.5, 128.2, 128.1, 126.4, 114.7, 65.8, 60.5, 54.0, 53.9, 51.0, 50.4, 49.3, 38.5, 36.9, 36.3, 34.2, 32.9, 22.5, 22.4, 20.6. ¹⁹F NMR (DMSO-*d*6, 376 MHz) δ -98.4 and -98.9 (AB quartet, *J* = 180.2 Hz, 2F), -110.4 and -111.0 (AB

quartet, J = 177.4 Hz, 2F), -127.3 and -128.8 (AB quartet, J = 234.6 Hz, 2F). HRMS m/z [M + Na]⁺ calc'd for C39H44F6N8O10S2Na 985.2424, found 985.2465. **2.46:** ¹H NMR (DMSO-*d*6, 400 MHz) δ 9.43 (s, 1H), 9.16 (brs, 1H), 8.36 (d, *J* = 8.2 Hz, 1H), 8.31 (d, *J* = 7.7 Hz, 1H), 8.27 (d, *J* = 9.0 Hz, 1H), 8.04 (d, *J* = 8.0 Hz, 1H), 7.93 (d, *J* = 7.5 Hz, 1H), 7.49 (brs, 1H), 7.28-7.17 (m, 7H), 7.03 (brs, 1H), 7.02 (d, *J* = 8.5 Hz, 2H), 6.63 (d, *J* = 8.5 Hz, 2H), 4.98 (brs, 1H), 4.89-4.83 (m, 1H), 4.78- 4.72 (m, 1H), 4.57 (td, *J* = 9.7, 4.0 Hz, 1H), 4.49 (q, *J* = 7.0 Hz, 1H), 4.41-4.33 (m, 1H), 3.84 (t, *J* = 6.9 Hz, 1H), 3.58-3.48 (m, 3H), 3.28 (dd, *J* = 13.9, 4.0 Hz, 1H), 2.98 (dd, *J* = 13.5, 6.0 Hz, 1H), 2.89-2.79 (m, 2H), 2.75 (dd, *J* = 13.9, 10.6 Hz, 1H), 2.64-2.54 (m, 2H), 2.45 (dd, *J* = 15.4, 6.5 Hz, 1H), 1.75 (s, 3H), 0.63 (d, *J* = 6.2 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.8, 171.7, 171.1, 170.5, 170.2, 169.5, 169.1, 167.9, 155.7, 148.8-146.1 (m, 1C), 138.0, 130.0, 129.2, 128.2, 128.1, 126.3, 118.6-108.2 (m, 3C), 114.8, 68.7-68.2 (m, 1C), 65.9, 61.7, 54.4, 53.9, 52.5, 49.7, 46.6, 37.3, 36.9, 36.5, 31.4, 30.5, 22.4, 20.1. ¹⁹F NMR (DMSO-*d*6, 376 MHz) δ -109.0 (dt, *J* = 250.9, 14.1 Hz, 1F), -113.8 (dd, *J* = 221.8, 10.0 Hz, 1F), -118.6 (dt, *J* = 250.9, 12.1 Hz, 1F), -123.6 (dd, *J* = 221.8, 11.3 Hz, 1F), -135.3 (t, *J* = 14.1 Hz, 1F). HRMS m/z [M + Na] ⁺ calc'd for $C_{39}H_{43}F_5N_8O_{10}S_2N_8965.2361$, found 965.2401.

Macrocyclic Product (2.47)

A flask was charged with linear precursor **2.44** (60 mg, 0.0759 mmol) and DMF (15 mL). The mixture was added octafluorocyclopentene (1M in MeCN, 114 μl, 0.114 mmol) and Et₃N (26 μl, 0.190 mmol) at 0 °C under argon. The mixture was stirred for 5 minutes and then excess amount of octafluorocyclopentene and MeCN were removed by rotary evaporation. The mixture was added $Cs₂CO₃$ (99mg, 0.304 mmol) at 0 °C and stirred for 2 hr at rt. The mixture was added $Cs₂CO₃$ (99 mg, 0.304 mmol) and stirred for 4 hr at rt. The mixture was added AcOH (65 µl, 1.14 mmol) and concentrated by rotary evaporation. Purification by preparative reverse-phase HPLC (40%-55% MeCN/H2O) afforded the title compound (29 mg, 0.0311 mmol, 41%) as an off-white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.51 (s, 1H), 8.33 (d, *J* = 9.5 Hz, 1H), 8.12 (d, *J* = 6.9 Hz, 1H), 8.05 (d, *J* = 8.2 Hz, 1H), 7.98 (d, *J* = 6.8 Hz, 1H), 7.69 (d, *J* = 7.7 Hz, 1H), 7.42 (s, 1H), 7.36 (d, *J* = 8.6 Hz, 1H), 7.30-7.17 (m, 6H), 7.06-6.84 (m, 3H), 7.03 (s, 1H), 4.98 (brs, 1H), 4.84-4.79 (m, 1H), 4.72 (dd, *J* = 14.2, 6.8 Hz, 1H), 4.60-4.55 (m, 1H), 4.55-4.48 (m, 1H), 4.41-1.34 (m, 1H), 3.74 (t, *J* = 6.9 Hz, 1H), 3.61 (d, *J* = 14.0 Hz, 1H), 3.48-3.41 (m, 2H), 3.22 (dd, *J* = 14.0, 3.7 Hz, 1H), 2.97 (dd, *J* = 13.2, 5.1 Hz, 1H), 2.89-2.82 (m, 3H), 2.70 (dd, *J* = 13.2, 10.1 Hz, 1H), 2.38 (dd, *J* = 15.6, 8.5 Hz, 1H), 2.32 (dd, *J* = 15.6, 5.0 Hz, 1H), 1.87 (s, 3H), 0.51 (d, *J* = 6.0 Hz, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.7, 170.8, 170.3, 169.5, 169.1, 168.8, 168.5, 168.0, 152.4, 139.2- 138.8 (m, 1C), 137.8, 135.2-134.9 (m, 1C), 132.1, 130.5, 129.1, 128.2, 126.4, 118.7-111.5 (m, 2C), 114.9, 68.6-68.1 (m, 1C), 65.9, 62.1, 54.8, 53.6, 50.4, 48.4, 46.5, 38.4, 37.3, 37.2, 32.7, 30.6, 22.4, 20.0. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -103.2 and -119.2 (AB quartet, *J* = 252.2 Hz, 2F), - 113.8 and -127.0 (AB quartet, $J = 223.0$ Hz, 2F). HRMS m/z $[M + Na]^{+}$ calc'd for C39H42F4N8O10S2Na 945.2299, found 945.2327.

Ac-SFC-NH² (S2.4)

Synthesized according to General Procedure A. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.14 (d, *J* = 7.8 Hz, 1H), 7.98 (d, *J* = 7.7 Hz, 1H), 7.96 (d, *J* = 6.6 Hz, 1H), 7.26-7.21 (m, 5H), 7.20-7.16 (m, 2H), 4.48 (ddd, *J* = 9.2, 8.0, 4.6 Hz, 1H), 4.29 (td, *J* = 8.0, 4.9 Hz, 1H), 4.25 (dd, *J* = 7.4, 6.3 Hz, 1H), 3.48 (d, *J* = 6.3 Hz, 2H), 3.09 (dd, *J* = 14.0, 4.6 Hz, 1H), 2.87 (dd, *J* = 13.5, 8.9 Hz, 1H), 2.80 (ddd, *J* = 13.6, 9.3, 5.0 Hz, 1H), 2.70 (dt, *J* = 13.6, 7.7 Hz, 1H), 2.27 (dd, *J* = 9.3, 7.7 Hz, 1H), 1.83 (s, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.4, 170.8, 170.5, 169.6, 137.8, 129.2, 128.1, 126.3, 61.6, 55.2, 55.1, 54.3, 36.7, 26.0, 22.5. HRMS m/z [M + Na]⁺ calc'd for C₁₇H₂₄N₄O₅SNa 419.1365, found 419.1362.

Macrocyclic Product (2.48)

Synthesized according to General Procedure C. Purification by preparative reverse-phase HPLC (45%-65% MeCN/H2O) afforded the title compound (24 mg, 0.0422 mmol, 28%) as a white solid. ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.77 (d, *J* = 9.0 Hz, 1H), 8.23 (d, *J* = 7.7 Hz, 1H), 8.04 (d, *J* = 8.9 Hz, 1H), 7.31 (brs, 1H), 7.29-7.17 (m, 5H), 6.98 (brs, 1H), 4.76-4.67 (m, 2H), 4.65-4.59 (m, 1H), 4.43-4.33 (m, 2H), 3.32 (dd, *J* = 14.0, 4.3 Hz, 1H), 3.13 (dd, *J* = 14.0, 5.4 Hz, 1H), 2.98 (dd, *J* = 14.0, 9.5 Hz, 1H), 2.82 (d, *J* = 14.0, 9.8 Hz, 1H), 1.86 (s, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 171.3, 171.1, 169.5, 168.1, 154.9-154.2 (m, 1C), 137.9, 129.1, 128.2, 126.4, 117.7-108.4 (m, 4C), 72.7, 54.1, 51.4, 50.4, 36.1, 35.5, 22.3. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -102.7 and -108.2 (AB quartet, *J* = 244.0 Hz, 2F), -107.5 and -113.6 (AB quartet, *J* = 261.0 Hz, 2F), -127.4 and -128.1 (AB quartet, $J = 200.0$ Hz, 2F). HRMS m/z $[M + Na]^+$ calc'd for $C_{22}H_{22}F_6N_4O_5SNa$ 591.1113, found 591.1077.

Compound 2.49

¹H NMR (DMSO-*d*6, 500 MHz) δ 8.68 (d, *J* = 7.9 Hz, 1H), 8.45 (d, *J* = 7.6 Hz, 1H), 8.28-8.18 (m, 3H), 7.42 (brs, 1H), 7.33-7.16 (m, 6H), 4.66-4.57 (m, 1H), 4.38-4.24 (m, 2H), 4.19 (dd, *J* = 7.0, 14.5 Hz, 1H), 4.11-4.02 (m, 1H), 3.12 (dd, *J* = 14.0, 4.3 Hz, 1H), 2.89-2.76 (m, 3H), 2.00 (s, 3H). ¹³C NMR (DMSO-*d*6, 126 MHz) δ 179.9-179.2 (m, 2C), 171.6, 170.1, 170.0, 165.6, 137.6, 129.2, 128.2, 126.4, 112.8-108.1 (m, 2C), 107.0, 62.1, 54.6, 52.8, 51.3, 37.2, 34.8, 20.5. ¹⁹F NMR (DMSO-d₆, 282 MHz) δ -124.9 (s, 4F). HRMS m/z [M + Na]⁺ calc'd for C₂₂H₂₄F₄N₄O₇SNa 587.1199, found 587.1186.

Water Sensitivity Experiments

A flask was charged with linear precursor **2.44** (19 mg each, 0.0240 mmol) and DMF/H2O (a; 5 ml of DMF without H₂O. b; 5ml of DMF + 1 eq. H₂O. c; 4 ml of DMF + 1 ml of H₂O). To the mixture octafluorocyclopentene (1M in MeCN, 36 μl, 0.036 mmol) and Et₃N (8.4 μl, 0.060 mmol) were added at 0 °C. The mixture was stirred for 5 minutes.

%B
4
55
100
100

Figure S2.4. Method for the determination of aqueous stability in the reaction of **44** to **45**.

pH Sensitivity Experiments

Compound **2.22** was subjected to aq. NaOH and aq. HCl in MeOH separately to probe stability in basic/acidic conditions, respectively. Under basic conditions, decomposition of **2.22** was observed after one hour. Whereas under acidic conditions, **2.22** remained stable over 24 hours. These results were determined based on HPLC traces obtained at various time points as outlined below.

 $T = 1 h (NaOH)$

Figure S2.5. Determination of basic pH stability of **2.22** by HPLC over 1 hour.

Figure S2.6. Rate determination for the reaction of L-Cbz-His-NH² with mono phenoxy derivative **5**.

Probing the YXCXXC Consensus Sequence for Macrobicycle Synthesis

The following amino acids were reacted with OFCP to probe the consensus sequence YXCXXC. In all cases tentative HPLC-MS data suggested the formation of intermediates derived from three substitutions; however, the fourth substitution, as seen similarily in compounds **2.43** and **2.47**, proved to be much slower and provided complex product mixtures compared to the originally outlined polycyclizations. In the cases of linear peptides Ac-NYCAFC-NH₂ and Ac-YCNTFC-NH₂, the difficulty involved in the fourth substitution likely stems from the adjacent cysteine and tyrosine residues preventing the formation of a strained bicyclic system. Regarding linear peptide Ac-CNYTFC-NH2, the bulkiness of tyrosine as well as the sterically hindering spirocycle intermediate likely disallows the fourth substitution based on this unfavorable approach from its position within the ring.

Polymacrocycle (Fourth Substitution)

Figure S2.7. Probing the YXCXXC consensus sequence for macrobicycle syntheses using scrambled variations of Ac-YNCTFC-NH2

 \times

Crystal Structure Analysis

Deposition Number 1946197 contains the supplementary crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service [www.ccdc.cam.ac.uk/structures.](http://www.ccdc.cam.ac.uk/structures)

Figure S2.8. Structure was solved in space group P2(1) with ORTEP showing 50% probability. Both residual DMSO and water molecules can be observed within the unit cell. Water atoms can be seen localized between the two asymetric units of the of the unit cell and may be providing additional stabilization effects through hydrogen bonding.

NMR Spectra

COSY spectrum of compound 2.11a (DMSO-*d***6)**

HMBC spectrum of compound 2.11a (DMSO-*d***6)**

Compound 2.11a (DMSO-*d***6)**

ppm $\overline{\text{HSQC}}$ 20 40 60 80 100 $\alpha=\frac{1}{2}$. α 120 J. -140 -160 -180 —
5 10 $\overline{7}$ 11 9 $\bf8$ 6 $\pmb{4}$ 3 $\mathbf 2$ $\mathbf{1}$ ppm **HMBC spectrum of compound 2.11b (DMSO-***d***6)**ıllı ppm HMBC ${\bf 20}$ $\delta \hat{\theta}$ 40 60 80 -100 -120 -140 - 160 -180 $\overline{11}$ 10 $\overline{9}$ $\ddot{\mathbf{8}}$ $\overline{7}$ $\overline{\bf{5}}$ $\ddot{4}$ $\overline{\mathbf{3}}$ $\overline{2}$ ppm $\bf 6$ $\mathbf{1}$

HSQC spectrum of compound 2.11b (DMSO-*d***6)**

Compound 2.11b (DMSO-*d***6)**

F NMR of compound 2.16 (DMSO-*d***6, 282 MHz)**

C NMR of compound 2.17 (DMSO-*d***6, 126 MHz)**

www <u>mull</u> ΛA A MA ppm ${\bf 20}$ $\ddot{\mathbf{0}}$ 40 e al 60 \bullet 80 100 120 140 - 160 180 $\dot{\bm{9}}$ $\dot{\mathbf{8}}$ $\overline{7}$ $\overline{\bf{5}}$ $\frac{1}{4}$ $\dot{3}$ $\frac{1}{2}$ ppm $\ddot{\bf{6}}$ $\mathbf{1}$ **HSQC spectrum of compound 2.20 (DMSO-***d***6)**www Mad ppm ${\bf 20}$ \bullet a b $000 - 6$ 40 60 80 -100 -120 -688 \mathbf{J} -140 - 160 180 $\mathbf{9}$ $\overline{7}$ $\ddot{\bf{6}}$ $\overline{\bf{5}}$ $\frac{1}{4}$ $\overline{\mathbf{3}}$ $\overline{2}$ \mathbf{i} $\bf{8}$ ppm

HMBC spectrum of compound 2.20 (DMSO-*d***6)**

Compound 2.22 (DMSO-*d***6)**

HMBC spectrum of compound 2.26 (DMSO-*d***6)**

Compound 2.28 (DMSO-*d***6)**

Compound 2.34 (DMSO-*d***6)**

HMBC spectrum of compound 2.38 (DMSO-*d***6)**

HMBC spectrum of compound 2.40 (DMSO-*d***6)**

HSQC spectrum of compound 2.43 (DMSO-*d***6)**

Compound 2.43 (DMSO-*d***6)**

HSQC spectrum of compound 2.47 (DMSO-*d***6)**

Compound 2.47 (DMSO-*d***6)**

References

- [S1] N. Hada, Y. Shida, N. Negishi, F. Schweizer, T. Takeda, *Chem. Pharm. Bull*., **2009**, *57*, 1081-1088.
- [S2] H. Albrich, H. Vahrenkamp, *Chem. Ber*. **1994**, *127*, 1223- 1233

Chapter 3 Appendix Material

General Methods

See Chapter 2 Appendix Material

General procedure A – Peptide Synthesis:

Peptides were prepared using standard solution phase techniques and the Boc protection strategy, or standard solid phase peptide synthesis (SPPS) using the Fmoc protection strategy. Boc deprotection was performed using 4 M HCl in dioxane or 1:1 TFA in DCM. Cbz deprotection was performed with palladium on carbon in methanol under an atmosphere of hydrogen. Prior to removal of the final protecting group, peptides were purified by column chromatography on $SiO₂$ eluted with 2-12% MeOH in CHCl3. Peptide identities were verified by HPLC-ESI-MS.Using SPPS peptides were synthesized manually using standard Fmoc solid phase synthesis protocols on Rink Amide MBHA resin (200-400 mesh, 0.73 mmol/g, 1% DVB) on $0.50 - 1.50$ mmol scale using a fritted glass reaction vessel. Fmoc-deprotection was achieved with 5% piperazine and 2% DBU in DMF (2 x 15 min). The reaction vessel was washed with DMF (3x) and CH_2Cl_2 (2x). The vessel was then charged with the appropriate Fmoc-amino acid (3 eq.) and HBTU (3 eq.) followed by DMF (10-20 ml) and $iPr₂NEt$ (10 eq). The resin was agitated for 45 min, drained, and washed with DMF (3X). After all coupling were completed the resin was cleaved with TFA/ water/TIPS (90:5:5) for 1.5 hrs. The cleaved resin was removed by filtration into cold $Et₂O$. The peptide was precipitated with Et₂O and isolated by centrifugation. The peptide pellet was repeatedly triturated with $Et₂O$ and isolated by centrifugation to ensure complete removal of cleavage reagents. Peptides were typically used as crude moving forward, but purified by preparative RP-HPLC if necessary.

General Procedure B – One-pot polycylization

A flask was charged with linear precursor (1 eq) and DMF (5 mM in substrate). The mixture was added OFCP (1M in MeCN, 1.5 eq) and Et_3N (2.5 eq) at 0 °C. The mixture was stirred for 30 minutes (until HPLC indicated completion) and then excess amount of OFCP and MeCN were removed by rotary evaporation. The mixture was then cooled again to 0 °C then added Cs_2CO_3 (4 eq) or Me3SiOK (4 eq) and at let stir at rt until completion observed by HPLC. After completion the mixture was quenched with AcOH (10 eq) and concentrated under reduced pressure to afford crude product. Purification performed by RP-HPLC to give desired product.

Experimental Procedures

Compound 3.10

Synthesized according to General Procedure B using crude peptide AcHFCASC-NH² to give **3.10** (27%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.38 (s, 1H), 8.42 (d, J = 8.9 Hz, 1H), 8.36 – 8.28 (m, 3H), 7.93 (t, J = 8.2, 1H), 7.31 (d, J = 9.2 Hz, 2H), 7.28 – 7.20 (m, 7H), 7.20 – 7.16 (m, 1H), 4.85 $-$ 4.80 (m, 1H), 4.63 – 4.53 (m, 3H), 4.31 (quint, J = 4.5 Hz, 1H), 4.20 (quint, J = 7.4 Hz, 1H), 3.82 $(dd, J = 10.85, 5.1 Hz, 1H), 3.52 (dd, J = 10.85, 3.80 Hz, 1H), 3.34 (dd, J = 12.6, 6.7 Hz, 1H),$ 2.92 – 2.86 (m, 2H), 2.85 – 2.79 (m, 1H), 2.79 – 2.74 (m, 1H) 1.82 (s, 3H), 1.24 (d, J = 7.2 Hz, 3H). ¹³C NMR (DMSO-d6, 126 MHz) δ 172.9, 171.4, 169.7, 169.6, 168.8, 167.8, 167.7, 137.3, 133.8, 129.2, 128.0, 126.3, 61.2, 55.3, 53.5, 51.1, 51.0, 50.3, 48.6, 48.4, 46.4, 36.8, 33.7, 30.5, 30.1, 22.5, 17.6. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -103.2 – (-104.7) (m, 1F), -114.0 – (-116.4) (m, 2F), -136 – (-125.2) (m, 2F). HRMS m/z [M + H]⁺ calc'd for $C_{34}H_{37}F_4N_9O_8S_2H 840.2221$, found 840.2233.

Compound 3.11

Synthesized according to General Procedure B using crude peptide AcHFCASC-NH² to give **3.11** (42%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.10 (d, J = 7.7, 1H), 8.84 (s, 1H), 8.41 (d, J = 8.7 Hz, 1H), 8.07 - 8.01 (m, 2H), 7.63 (d, J = 8.1 Hz, 1H), 7.54 (d, J = 7.65 Hz, 1H), 7.38 (s, 1H), 7.28 - 7.17 (m, 6H), 4.75 - 4.70 (m, 1H), 4.68 - 4.62 (m, 1H), 4.62 - 4.56 (m, 2H) 4.52 (quint, J = 7.2, 1H), 4.43 - 4.36 (m, 1H), 3.80 - 3.73 (m, 1H), 3.57 - 3.39 (m, 3H), 3.47 - 3.41 (m, 1H), 3.16 - 3.05 (m, 2H), 3.03 - 2.94 (m, 2H), 2.94 - 2.81 (m, 3H), 1.77 (s, 3H), 1.31 (d, J = 7.3 Hz, 3H). ¹³C NMR (DMSO-d6, 126 MHz) δ 170.6, 170.3, 170.1, 170.0, 168.9, 167.9, 167.5, 141.3 - 140.9 (m, 1C), 139.0, 137.5, 135.6, 132.8 - 132.6 (m, 1C), 129.0, 128.2, 126.4, 117.6 - 115.8 (m, 2C), 117.6 - 115.8 (m, 2C), 114.5, 70.4 - 69.7 (m, 1C), 61.3, 55.1, 53.8, 51.6, 51.1, 48.5, 47.1, 38.6, 37.8, 30.8, 29.8, 22.5, 16.0. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -98.8 and -120.7 (AB quartet, *J* = 254.1 Hz, 2F), -117.2 and -127.3 (AB quartet, *J* = 222.4 Hz, 2F). HRMS m/z [M + H]⁺ calc'd for C44H37F4N9O8S2H 840.2221, found 840.2213.

Compound 3.12

Synthesized according to General Procedure B using crude peptide AcYGHCAC-NH² to give **3.12** (20%).

Compound 3.13

Synthesized according to General Procedure B using crude peptide AcYGSHAC-NH² to give **3.13** (40%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.93 (s, 1H), 8.58 (d, J = 7.0 Hz, 1H), 8.42 (d, J = 6.6 Hz, 2H), 8.34 (s, 1H), 7.78 (d, J = 7.8 Hz, 1H), 7.54 (s, 1H), 7.19 (s, 1H), 7.15 - 7.10 (m, 2H), 7.10 - 7.00 (m, 2H), 6.67 - 6.58 (m, 1H), 4.71 - 4.64 (m, 1H), 4.58 - 4.51 (m, 1H), 4.27 (q, J = 6.4 Hz, 1H), 4.17 (q, J = 7.0 Hz, 1H), 4.09 (quint, J = 7.1 Hz, 1H), 3.95 - 3.87 (m, 1H); 3.72 - 3.56 (m, 3H),, 3.45 (d, J = 13.1 Hz, 1H), 2.90 - 2.83 (m, 2H); 2.82 - 2.73 (m, 1H), 2.68 - 2.58 (m, 2H), 1.79 (s, 1H), 1.24 (d, J = 7.2 Hz, 3H). ¹³C NMR (DMSO-d6, 126 MHz) δ 171.6, 171.0, 169.7, 169.7, 169.3, 169.0, 167.2, 151.0, 143.1 - 142.5 (m, 1C), 139.4, 136.8, 135.7, 130.8, 128.2 - 127.8 (m, 1C), 119.7 - 113.0 (m, 2C), 119.7 - 113.0 (m, 2C), 119.2, 118.09, 67.3 - 66.4 (m, 1C), 60.9, 55.8, 54.7, 50.8, 49.3, 46.3, 42.8, 36.7, 30.1, 29.8, 22.3, 17.3. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -105.9 and -123.8 (AB quartet, *J* = 257.6 Hz, 2F), -117.1 and -127.2 (AB quartet, *J* = 223.6 Hz, 2F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{33}H_{35}F_{4}N_{9}O_{9}SH 810.2293$, found 810.2294.

Compound 3.14

Synthesized according to General Procedure B using crude peptide AcYGSdHAC-NH₂ to give **3.14** (37%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.25 (s, 1H), 8.75 (d, J = 6.6 Hz, 1H), 8.44 (d, J = 6.8 Hz, 1H), 8.23 (t, J = 5.1 Hz, 1H), 7.33 – 7.28 (m, 1H), 7.26 (d, J = 5.76 Hz, 1H), 7.09 – 7.05 $(m, 2H)$, 7.93 – 6.98 $(m, 4H)$, 6.19 $(d, J = 8.6 Hz, 1H)$, 4.66 – 4.61 $(m, 1H)$, 4.58 – 4.53 $(m, 1H)$, $4.41 - 4.36$ (m, 1H), 4.08 (quint, $J = 3.3$ Hz, 1H), 4.0 (quint, $J = 6.9$ Hz, 1H), 3.91 (dd, $J = 16.2$, 5.2 Hz, 1H), 3.85 (dd, J = 10.5, 3.8 Hz, 1H), 3.71 – 3.66 (m, 1H), 3.56 (d, J = 5.2 Hz, 1H), 3.54 – 3.48 (m, 1H), 3.06 (dd, J = 15.4, 3.5 Hz, 1H), 2.96 (dd, J = 13.4, 3.78 Hz, 1H), 2.81 – 2.75 (m, 3H), 1.93 (s, 3H), 1.13 (d, J = 7.3 Hz, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 171.2, 170.7, 170.6, 170.1, 169.4, 167.0, 153.8, 139.6, 134.0, 131.2, 128.3, 127.6, 116.8, 115.3, 67.0, 61.5, 56.3, 52.9, 52.5, 50.0, 46.0, 43.0, 37.8, 30.2, 28.1, 25.1, 22.7, 16.7. HRMS m/z [M + H]⁺ calc'd for $C_{33}H_{35}F_4N_9O_9SH$ 810.2293, found 810.2293.

Compound 3.15

Synthesized according to general procedure B using crude peptide AcYGTMQVSHAC-NH₂ to give **3.15** (35%). See NMR Spectra Appendix section below for crude NMR spectra of this compound. HRMS m/z $[M + H]^+$ calc'd for $C_{52}H_{69}F_4N_{14}O_{15}S_2H$ 1269.4444, found 1269.4454.

Compound 3.16

Synthesized according to general procedure B using crude peptide AcD(C)VHSAY-NH₂ to give **3.16**.

Compound 3.17

Synthesized according to general procedure B using crude peptide AcYGAE(C)H-NH₂ to give **3.17** (8%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.26 (s, 1H), 8.60 (s, 1H), 8.49 (d, J = 8.1 Hz, 1H), 8.37 (d, J = 7.7, 1H), 7.7 (s, 1H), 7.65 (d, J = 10.1 Hz, 1H), 7.18 (s, 1H), 6.99 (d, J = 8.7 Hz, 2H); 6.91 (d, J = 8.7, 2H), 6.70 (d, J = 6.2, 1H), 6.62 (s, 1H), 5.24 (quint, J = 7.1 Hz, 1H), 4.82 - 4.73 $(m, 1H)$, 4.79 - 4.72 $(m, 1H)$, 4.36 - 4.26 $(m, 1H)$, 4.15 $(dd, J = 17.3, 7.4, 1H)$, 4.03 - 3.97 $(m, 1H)$, 3.53 (dd, J = 17.3, 3.4, 1H), 3.16 (s, 3H), 3.12 - 2.98 (m, 1H), 3.04 - 2.99 (m, 1H), 2.90 - 2.82 (m, 1H), 2.85 - 2.79 (m, 2H), 2.77 (s, 3H), 2.53 - 2.47 (m, 1H), 2.02 - 1.96 (m, 1H), 1.72 - 1.66 (m, 1H), 1.96 - 1.91 (m, 1H), 1.71 - 1.67 (m, 1H), 1.84 (s, 3H), 1.22 (d, J = 6.70, 3H). ¹³C NMR (DMSOd6, 126 MHz) δ 173.8, 171.2, 171.0, 170.8, 170.0, 169.7, 168.9, 167.4, 153.2, 146.5 - 144.6 (m, 1C), 137.6, 136.9, 135.1, 130.2, 129.5 - 128.2 (m, 1C), 118.4, 118.4, 118.1 - 117.0 (m, 2C), 118.1 - 117.0 (m, 2C). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -111.1 –(-112.2) (m, 1F), -115.8 – (-119.6) (m, 3F). HRMS m/z $[M + H]^+$ calc'd for $C_{37}H_{42}F_4N_{10}O_9SH$ 879.2871, found 879.2868.

Compound 3.18

Synthesized according to general procedure B using crude peptide Ac-CWSC-NH² to give **3.18** (35%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.88 (s, 1H), 8.61 (s, 1H), 7.82 (s, 1H), 7.64 – 7.54 (m, 1H), 7.54 – 7.47 (m, 1H), 7.46 – 7.38 (m, 1H), 7.38 – 7.28 (m, 1H), 7.26 – 7.16 (m, 1H),7.11 – 7.02 (m, 1H), 7.02 – 6.93 (m, 1H), 6.75 (bs, 1H), 5.22 – 5.11 (m, 1H), 5.07 – 4.99 (m, 1H), 4.55 – 4.45 (m, 1H), 4.37 – 4.25 (m, 1H), 4.15 – 4.00 (m, 1H), 3.95 – 3.87 (m, 1H), 3.72 – 3.55 (m, 2H), $3.52 - 3.43$ (m, 1H), $3.26 - 3.13$ (m, 3H), $2.74 - 2.65$ (m, 1H), $2.02 - 1.93$ (m, 1H). ¹³C NMR (DMSO-d6, 126 MHz) δ 173.2, 171.1, 170.8, 169.8, 169. 4, 136.1, 127.2, 123.5, 121.0, 118.4, 118.0, 111.4, 109.4, 64.9, 60.3, 57.4, 56.8, 51.7, 36.5, 34.4, 26.4, 24.3. HRMS m/z [M + H]⁺ calc'd for $C_{27}H_{28}F_5N_6O_6S_2H$ 691.1432, found 691.1431.

Compoud 3.19

Synthesized according to general procedure B using crude peptide Ac-CWS-homo-C-NH $_2$ to give **3.18** (35%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.89 (s, 1H), 9.47 (s, 1H), 8.58 (s, 1H), 7.51 (d, J $= 8.6$ Hz, 1H), 7.32 (t, J = 8.55 Hz, 2H), 7.26 (s, 1H), 7.17 (s, 1H), 7.06 (t, J = 7.4 Hz, 1H), 6.84 $(s, 1H)$, 6.88 (d, J = 5.2 Hz, 1H), 5.02 (s, 1H), 4.33 – 4.26 (m, 2H), 4.12 – 4.07 (m, 1H), 3.83 – 3.78 (m, 1H), 3.74 – 3.61 (m, 3H), 1.95 (s, 3H), 1.83 (s, 1H). ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ - 108.8 (dt, J = 252.1, 14.8 Hz, 1F), -112.6 (d, J = 218.1 Hz, 1F), -116.2 (d, J = 219.0 Hz, 1F), - 120.1 – (-121.2) (m, 1F), -136.4 (d, J = 14.8 Hz, 1F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{28}H_{29}F_5N_6O_6S_2H$ 705.1588, found 705.1590.

Compound 3.20

Synthesized according to general procedure B using crude peptide AcCWS-*d-*C-NH² to give **3.20** (27%).¹H NMR (DMSO-*d*6, 500 MHz) δ 10.87 (s, 1H), 9.25 (s, 1H), 8.49 (d, J = 8.49 Hz, 1H), 8.40 (d, J = 8.4 Hz, 1H), 8.29 (d, J = 7.2 Hz, 1H), 7.45 (d, J = 7.9 Hz, 1H), 7.32 (d, J = 8.1 Hz, 1H), 7.29 (d, J = 9.0 Hz, 1H), 7.10 – 7.03 (m, 2H), 6.96 (t, J = 7.6 Hz, 1H), 5.07 (t, J = 5.3 Hz, 1H), 4.67 – 4.59 (m, 2H), 4.53 – 4.45 (m, 1H), 4.23 – 4.18 (m, 1H), 3.72 (quint, J = 5.7 Hz, 1H), 3.55 –

3.49 (m, 1H), $3.42 - 3.37$ (m, 1H), 3.27 (dd, $J = 11.0$ Hz,, 4.5 Hz, 1H), 3.17 (d, $J = 5.15$ Hz, 1H), $3.12 - 3.05$ (m, 1H), $3.04 - 2.96$ (m, 1H), $2.75 - 2.67$ (m, 1H), 1.81 (s, 1H). ¹³C NMR (DMSO-d6, 126 MHz) δ 172.5, 169.5, 169.0, 168.0, 167.4, 136.1, 127.1, 123.7, 120.9, 118.4, 117.9, 111.3, 109.1, 60.8, 57.0, 54.8, 51.1, 46.4, 32.2, 30.5, 28.7, 22.2. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -106.4 (dt, J = 250.8, 14.7 Hz, 1F), -114.9 (dd, J = 222.0, 10.2 Hz, 1F), -121.5 (dt, J = 250.2, 11.7 Hz, 1F), -125.6 – (-126.6) (m, 1F), -142.4 (t, J = 14.3 Hz, 1F). HRMS m/z [M + H]⁺ calc'd for C27H27F5N6O6S2H 691.1432 found 691.1432.

Compound 3.21

Synthesized according to general procedure B using crude peptide AcCSC-NH² to give **3.21** (40%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.09 (d, J = 8.0 Hz, 1H), 8.90 (bs, 1H), 8.29 (d, J = 7.2 Hz, 1H), 7.76 (d, J = 9.8 Hz, 1H), 5.09 – 5.02 (m, 1H), 4.71 – 4.63 (m, 1H), 4.45 – 4.37 (m, 1H), 3.83 (dd, J = 11.2, 5.6 Hz, 1H), 3.68 (dd, J = 11.1, 3.0 Hz, 1H), 3.64 – 3.57 (m, 1H), 3.53 (dd, J = 11.2, 4.0 Hz, 1H), 3.48 (d, J = 13.7 Hz, 1H), 3.43 – 3.36 (m, 1H), 2.87 (dd, J = 13.9, 3.4 Hz, 1H), 1.92 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 169.5, 168.7, 168.0, 167.5, 150.6 – 150.1 (m, 1C), 127.5 – 125.8 (m, 3C), 68.8 - 68.6 (m, 1C), 60.7, 60.3, 55.5, 51.1, 46.6, 31.3, 29.2, 22.4. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -105.7 (dt, J = 255.1, 15.9 Hz, 1F), -120.2 (dd, J = 221.5, 6.99 Hz, 1F), - 123.9 (dt, J = 255.4, 11.1 Hz, 1F), -125.0 (dd, J = 220.6, 13.3 Hz, 1F), -131.8 – (-132.1) (m, 1F). HRMS m/z $[M + H]^{+}$ calc'd for $C_{16}H_{17}F_5N_4O_5S_2H$ 505.0639 found 505.0640.

Compound 3.22

A flask was charged with compound **3.18**, diluted with DMF (50 mM), and cooled to 0 °C. To the flask was added beta-D-thioglucose sodium salt (1 equiv) and stirred for 1 h. After completion the reaction was concentrated under reduced pressured and purified by RP-HPLC to give **3.22** (>95%).¹H NMR (DMSO-*d*6, 500 MHz) δ 10.89 (s, 1H), 9.19 (bs, 1H), 9.11 – 8.84 (m, 2H), 8.26 $(d, J = 8.6 \text{ Hz}, 1\text{H})$, 7.50 $(d, J = 7.8 \text{ Hz}, 1\text{H})$, 7.33 $(d, J = 8.1 \text{ Hz}, 1\text{H})$, 7.19 $(d, J = 9.3 \text{ Hz}, 1\text{H})$, 7.06 (t, J = 7.26 Hz, 1H), 6.98 (t, J = 7.2 Hz, 1H), 4.79 (d, J = 9.2 Hz, 1H), 4.74 (d, J = 9.6 Hz, 1H), 4.66 (q, J = 7.6 Hz, 1H), 4.48 (q, J = 7.6 Hz, 1H), 4.38 – 4.31 (m, 1H), 3.85 – 3.77 (m, 1H), 3.64 – 3.54 (m, 3H), 3.53 – 3.43 (m, 3H), 3.30 – 3.21 (m, 3H), 3.11 – 2.98 (m, 3H), 1.83 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 172.1, 169.9, 169.5, 168.8, 167.6, 136.1, 127.1, 123.8, 120.9, 118.4, 118.0, 111.4, 109.3, 84.6, 81.3, 78.0, 73.5, 69.4, 61.9, 60.7, 55.7, 55.3, 51.5, 46.1, 33.4, 30.7, 27.8, 22.5. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -98.6 and -112.7 (AB quartet, J = 248.8 Hz, 2F), -111.3 and -127.0 (AB quartet, $J = 223.6$ Hz, 2F). HRMS m/z $[M + H]^{+}$ calc'd for C33H38F4N6O11S3H 867.1775 found 867.1776.

Compound 3.23

A flask was charged with compound **3.18**, diluted with DMF (50 mM), and cooled to 0 °C. To the flask was added coniferyl carbonate **3.27** (2 equiv), KOSiMe₃ (2 equiv), and stirred until completion observed by HPLC. After completion the reaction was quenched with AcOH (5 equiv) and concentrated under reduced pressured and purified by RP-HPLC to give **3.22** (53%). ¹H NMR (DMSO-*d*6, 500 MHz) δ10.87 (s, 1H), 9.30 (s, 1H), 8.62 (d, J = 9.3 Hz, 1H), 8.32 (t, J = 9.6 Hz, $2H$), 7.49 (1H, J = 7.9 Hz, 1H), 7.33 (d, J = 7.9 Hz, 2H), 7.27 (d, J = 1.8 Hz, 1H), 7.16 (J = 9.9 Hz, 1H), 7.11 (1H, J = 8.7Hz, 1H), 7.08 – 7.03 (m, 2H), 6.98 (t, J = 7.1 Hz, 1H) 6.62 (1H, J = 15.9 Hz, 1H), 6.31 (dt, J = 15.9, 6.3 Hz, 1H), 4.87 (dt, J = 9.9, 2.6 Hz, 1H), 4.73 (d, J = 5.7 Hz, 2H), 4.66 – 4.53 (m, 2H), 4.46 – 4.44 (m, 1H), 3.89 (d, J = 6.6 Hz, 2H), 3.77 (s, 3H), 3.63 (d, J = 12.1 Hz, 1H), $3.55 - 3.49$ (m, 1H), 3.38 (bs, 1H), $3.12 - 2.98$ (m, 3H), 2.93 (t, J = 12.2, 1H), $2.75 - 2.68$ (m, 1H), 1.95 – 1.85 (m, 2H), 1.73 (s, 3H), 1.46 – 1.41 (m, 1H), 0.89 (d, J = 6.6 Hz, 6H). ¹³C NMR (DMSOd6, 126 MHz) 172.7, 169.7, 169.5, 169.1, 168.5, 155.0, 150.3, 142.2, 136.6, 132.9, 129.3, 127.4, 125.4, 124.1, 122.8, 121.4, 118.9, 118.3, 117.8, 113.6, 111.8, 109.6, 73.8, 68.2, 61.9, 61.2, 56.2,

56.0, 54.9, 52.5, 46.3, 30.5, 29.7, 28.4, 27.7, 22.7, 19.2. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -103.1 and -119.0 (AB quartet, $J = 252.3$ Hz, $2F$), -114.2 and -127.7 (AB quartet, $J = 222.7$ Hz, $2F$). HRMS m/z $[M + H]^+$ calc'd for $C_{42}H_{46}F_4N_6O_{11}S_2H$ 951.2680 found 951.2686.

Compound 3.24

A flask was charged with compound **3.18**, diluted with DMF (50 mM), and cooled to 0 °C. To the flask was added NaN₃ (1 equiv) and stirred for 1 h. After completion the reaction was concentrated under reduced pressured and purified by RP-HPLC to give 3.24 (>95%). ¹H NMR (DMSO- d_6 , 500 MHz) δ 10.88 (s, 1H), 9.16 (s, 1H), 8.52 (d, 1H), 8.35 (t, J = 10.0, 2H), 7.49 (d, J = 7.8 Hz, 1H), 7.33 (d, J = 8.1 Hz, 1H), 7.12 (d, J = 9.5 Hz, 1H), 7.10 - 7.03 (m, 2H), 6.98 (t, J = 7.4 Hz, 1H), 4.97 (bs, 1H), 4.81 (d, J = 9.2 Hz, 1H), 4.64 – 4.53 (m, 2H), 4.38 – 4.30 (m, 1H), 3.89 – 3.80 (m, 1H), 3.59 (d, J = 13.2 Hz, 1H), 3.54 – 3.46 (m, 1H), 3.23 – 3.13 (m, 2H), 3.09 – 3.02 (m, 2H), 1.79 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 172.1, 169.4, 169.1, 168.6, 167.8, 136.1, 127.1, 123.7, 120.9, 118.4, 117.9, 111.4, 109.2, 61.3, 55.3, 54.8, 51.9, 45.9, 32.1, 27.8, 22.4, 21.1. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -102.8 and -115.7 (AB quartet, J = 250.6 Hz, 2F), -113.5 and -126.5 (AB quartet, $J = 224.7$ Hz, 2F). HRMS m/z $[M + H]$ ⁺ calc'd for $C_{27}H_{27}F_4N_9O_6S_2H$ 714.1540 found 714.1542.

Compound 3.25

A flask was charged with compound **3.24**, diluted with EtOH (50 mM) and flushed with argon. Pd/C (10 wt %) was then added, flushed with hydrogen gas, and warmed to 35 °C. After completion the reaction was filtered with celite and concentrated under reduced pressure. Purification by RP-HPLC gave **3.25** (87%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.88 (s, 1H), 8.63 $(s, 1H)$, 8.54 (d, J = 8.6 Hz, 1H), 8.41 (d, J = 9.2 Hz, 1H), 8.25 (d, J = 7.7 Hz, 1H), 7.50 (d, J = 7.5 Hz, 1H), 7.33 (d, J = 7.75 Hz, 1H), 7.16 – 7.03 (m, 3H), 6.98 (t, J = 7.0 Hz, 1H), 6.55 (s, 2H), 4.74 $(d, J = 9.2 \text{ Hz}, 1\text{H})$, 4.70 – 4.58 (m, 2H), 4.41 – 4.32 (m, 1H), 3.51 – 3.42 (m, 2H), 3.17 (s, 1H), 3.06 (bs, 2H), 2.70 – 2.64 (m, 1H), 2.62 – 2.55 (m, 1H), 1.80 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 172.2, 169.1, 168.9, 168.6, 168.0, 136.1, 127.2, 123.7, 120.9, 118.4, 118.0, 111.4, 109.1, 61.4, 55.1, 54.7, 51.9, 45.7, 34.4, 31.3, 30.4, 29.8, 28.3, 22.4, 20.7. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ –104.3 – (-106.0) (m, 2F), -111.5 – (-113.7) (m, 2F), -120.6 – (-122.5) (m, 2F), -126.7 – (-128.6) (m, 2F). HRMS m/z $[M + H]^+$ calc'd for $C_{27}H_{29}F_4N_7O_6S_2H$ 688.1635 found 688.1637.

Compound 3.26

A flask was charged with compound **3.24**, propargyl alcohol (1 equiv), phenylenediamine (15 mol %), sodium ascorbate (10 mol%), $CuSO₄•5H2O$ (5 mol%) in 2:3 H₂O:tBuOH (0.2 M). The reaction was let stir for 12 h until observed completion by HPLC. After completion the reaction was concentrated under reduced pressure and purified by RP-HPLC gave **3.26** (75%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.89 (s, 1H), 9.48 (s, 1H), 8.44 (d, J = 9.1 Hz, 1H), 8.32 (d, J = 8.32 Hz, 2H), 8.27 (d, J = 8.6 Hz, 1H), 7.49 (d, J = 7.9 Hz, 1H), 7.32 (d, J = 8.1 Hz, 1H), 7.22 (d, J = 9.3 Hz, 1H), 7.10 (d, J = 2.2 Hz, 1H), 7.06 (t, J = 7.3 Hz, 1H), 6.97 (t, J = 7.4 Hz, 1H), 4.87 (dt, J = 9.3, 2.5 Hz, 1H), 4.61 (s, 2H), 4.47 (g, J = 7.7 Hz, 2H), 4.37 (quint, J = 4.2 Hz, 2H), 3.84 (dd, J = 10.8, 4.4 Hz, 1H), 3.72 – 3.65 (m, 1H), 3.47 (dd, J = 10.8, 3.5 Hz, 1H), 3.26 (dd, J = 12.5, 8.1 Hz, 1H), 3.11 – 2.98 (m, 2H), 2.87 – 3.83 (m, 1H), 2.36 (dd, J = 12.5, 7.0 Hz, 1H), 1.74 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) 171.9, 169.6, 169.3, 168.6, 167.6, 149.1, 136.1, 127.1, 123.8, 123.1, 121.0, 118.4, 118.0, 111.4, 109.0, 61.4, 55.6, 55.0, 54.5, 51.6, 46.1, 32.6, 30.8, 27.6, 22.3. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -101.7 – (-103.2) (m, 1F), -113.1 – (-118.0) (m, 2F), -124.0 – (- 125.3) (m, 1F). HRMS m/z $[M + H]^+$ calc'd for $C_{30}H_{31}F_4N_9O_7S_2H$ 770.1802 found 770.1806.

3-((tert-butyldimethylsilyl)oxy)-4-methoxybenzaldehyde (S3.1)

To a flame-dried flask with 3-hydroxy-4-methoxybenzaldehyde was added DMF (1 M) and Pr_2 EtN (1.2 equiv). The reaction was cooled to 0 \degree C and added TBDMS-CI (1.1 equiv) slowly then stirred for 18 h at rt. The reaction was diluted with H_2O then extracted with EtOAc. The combined organic phases were washed with brine (x3), dried over NaSO⁴ concentrated in vacuo. The residue was purified by flash column chromatography (0-15% EtOAc/Hexanes) to give **S3.1** (97%) as a clear oil. ¹H NMR (CDCl₃, 300 MHz) δ 9.81 (s, 1H), 7.47 (dd, J = 8.3, 2.0 Hz, 1H), 7.36 (d, J = 2.0 Hz, 1H), 6.94 (d, J = 8.3 Hz, 1H), 3.88 (s, 3H), 1.00 (s, 9H), 0.16 (s, 6H). ¹³C NMR (CDCl₃, 126 MHz) δ 191.0, 156.7, 145.7, 130.3, 126.4, 120.1, 111.3, 55.6, 25.7, 18.5, -4.6 HRMS m/z [M + H]⁺ calc'd for C14H22O3SiH 267.1416, found 267.1424.

ethyl (E)-3-(3-((tert-butyldimethylsilyl)oxy)-4-methoxyphenyl)acrylate (S3.2)

To a stirred solution of **S3.1** in DCM (30 mM) was added ethyl 2-(triphenyl-l5 phosphaneylidene)acetate (1.5 equiv) at room temperature under inert atmosphere. The reaction mixture was stirred at room temperature for 24 h. Solvent was removed under reduced pressure and residue was purified by flash column chromatography (0-15% EtOAc/Hexanes) to give **S.3.2** (70%). ¹H NMR (CDCl3, 500 MHz) δ 7.58 (d, J = 15.9 Hz, 1H), 7.09 (dd, J = 8.2, 2.1 Hz, 1H), 7.05 $(d, J = 2.1$ Hz, 1H), 6.83 $(d, J = 8.2$ Hz, 1H), 6.25 $(d, J = 15.9$ Hz, 1H), 4.25 $(q, J = 7.12$ Hz, 2H), 3.83 (s, 1H), 1.33 (t, J = 7.00 Hz, 3H), 1.00 (s, 3H), 0.16 (s, 6H). ¹³C NMR (CDCl₃, 126 MHz) δ 167.5, 153.2, 145.3, 144.6, 127.6, 123.2, 119.8, 115.9, 111.8, 60.5, 55.6, 25.8, 18.6, 14.5, -4.5. HRMS m/z $[M + H]^{+}$ calc'd for $C_{18}H_{28}O_{4}$ SiH 337.1835, found 337.1836.

(E)-3-(3-((tert-butyldimethylsilyl)oxy)-4-methoxyphenyl)prop-2-en-1-ol (S3.3)

To a stirred solution of **S3.2** in DCM (0.2 M) was added DIBAL-H (3 equiv) at RT under inert atmosphere. The reaction mixture was stirred at RT for 24 h. After completion the reaction was concentrated under reduced pressure and the resulting residue was purified by flash column chromatography (0 – 20% EtOAc/Hex) to give **S3.3** (62%).¹H NMR (CDCl₃, 500 MHz) δ 6.95 – 6.90 (m, 2H), 6.79 (d, J = 8.9 Hz, 1H), 6.49 (d, J = 15.9 Hz, 1H), 6.20 (dt, J = 15.6, 5.9 Hz, 1H), 4.29 (dd, J = 6.1, 1.4 Hz, 2H), 3.80 (s, 1H), 1.00 (s, 9H), 0.16 (s, 6H). ¹³C NMR (CDCl₃, 126 MHz) δ 151.0, 145.2, 131.3, 129.9, 126.4, 120.6, 118.8, 112.0, 64.1, 55.6, 25.9, 18.6, -4.5. HRMS m/z $[M + H]^{+}$ calc'd for $C_{16}H_{27}O_3Si$ 295.1729, found 295.1735.

(E)-3-(3-((tert-butyldimethylsilyl)oxy)-4-methoxyphenyl)allyl isobutyl carbonate (S3.4)

To a stirred solution of **S3.3** and N-methylmorpholine (4.5 equiv) in DCM (0.5 M) was cooled to - 5 °C under argon. Isobutyl chloroformate (2.1 equiv) was added dropwise and continued to stir at -5 °C until completion. After completion as observed by TLC (1 h), the reaction was quenched with NaHCO₃, extracted with EtOAc, washed with NaHCO₃ (x2), brine, dried with NaSO₄, and concentrated under reduced pressure. The crude product was pushed through a plug of silica (50% EtOAc/Hex) and pushed forward without further purification.

(E)-3-(3-hydroxy-4-methoxyphenyl)allyl isobutyl carbonate (3.27)

To a stirred solution of $S3.4$ in DMF/H₂O (10:1) was added Cs_2CO_3 (0.5 equiv) and allowed to stir at RT until complete by TLC. The reaction was then diluted with EtOAc, washed with bring (x2), dried by NaSO4, and concentrated under reduced pressure. The crude product was purified by flash column chromatography (0 – 30% EtOAc/Hex) to give **3.27** as a clear oil (75%). ¹H NMR $(CDC₁₃, 500 MHz)$ δ 7.01 (d, J = 2.0 Hz, 1H), 6.86 (dd, J = 8.3, 1.8, 1H), 6.79 (d, J = 8.4, 1H), 6.58 $(d, J = 15.7 \text{ Hz}, 1\text{ H}), 6.15 \text{ (dt, } J = 15.8, 6.6 \text{ Hz}, 1\text{ H}), 5.62 \text{ (s, 1H)}, 4.75 \text{ (d, } J = 6.6 \text{ Hz}, 2\text{ H}), 3.94 \text{ (d, } J = 15.7 \text{ Hz})$ $J = 6.7$ Hz, 2H), 3.88 (s, 1H), 1.98 (sept, $J = 6.7$ Hz, 1H), 0.95 (d, $J = 6.7$ Hz, 6H). ¹³C NMR (CDCl3, 126 MHz) δ 155.4, 146.9, 145.8, 134.7, 129.9, 120.9, 119.5, 112.3, 110.6, 74.3, 68.6, 56.1, 27.9, 19.0. HRMS m/z $[M + H]$ ⁺ calc'd for C₁₅H₂₀O₅H 281.1389, found 281.1394.

Compound S3.5

To a flask was added ethyl 2-aminothiazole-5-carboxylate, $NaNO₂$ (1.1 equiv), diluted with MeCN (1 equiv) and H₂O (1 equiv), and allowed to stir at 0 °C. Then CuBr (1.1 equiv) in solution of HBr (1 equiv 1.23 M) was added dropwise at 0 °C. The reaction mixture was then allowed to gradually warm up to 23 °C and allowed to stir for 16 h. The mixture was then diluted with ice/DCM and then transferred to an erlenmeyer flask. The solution was kept cold and then gradually quenched with solid NaHCO₃ and then transferred to a separatory funnel. The aqueous layer was then extracted with DCM (3 times), organic layers combined, washed with brine, dried over $Na₂SO₄$, filtered, and concentrated to afford crude product. Purification by FCC $(0 - 2.5\%$ EtOAc/Hex) afforded desired product **S3.5** as a faint yellow oil (62%). ¹H NMR (DMSO-*d*6, 500 MHz) δ8.31 (s, 1H), 4.32 (q, 2H), 1.29 (t, 3H) ¹³C-NMR (500 MHz, DMSO-d6) δ 159.47, 148.17, 142.19, 132.75, 61.94, 14.04. HRMS [M+1] calc'd for C6H6BrNO2SH 236.9375383, found 236.93706.

Compound S3.6

To a flask equipped with a reflux condenser was added **S3.5**, NaSH (2 equiv, 60% wt), then diluted with EtOH (1 equiv, 0.45 M). The reaction was left overnight, then allowed to cool to 23 °C and poured onto an iced solution of aq. 1N HCl. The mixture was then extracted with EtOAc (x4), organic layers combined, dried over MgSO4, filtered, and concentrated to afford crude product as a white solid. Purification by FCC (15 – 30% EtOAc/Hex) afforded desired product **S3.6** as a white-solid (99%). ¹H-NMR (500 MHz, DMSO-d6) δ 13.80 (s, 1H), 8.08 (s, 1H), 4.24 (q, 2H), 1.25 (t, 3H). ¹³C-NMR (500 MHz, DMSO-d6) δ 190.93, 158.87, 136.83, 118.59, 61.43, 14.08. HRMS $[M+1]$ calc'd for $C_6H_6NO_2S_2H$ 189.9996, found 189.9998.

Compound 3.30

To a flask was added **S3.6** and diluted with anhydrous THF (0.14 M) and cooled to -78 °C. DIBAL-H (3 equiv, 1M) was then added dropwise at allowed to stir at -78 °C for 20 min then warmed to 0 °C and stirred for an additional 2 min. After 2 h, Rochelle's salt was added and allowed to stir at RT for 1 h. The reaction mixture was then filtered through celite and NaSO₄ and concentrated under reduced pressure to afford crude product. Purification by FCC (35 – 75% EtOAc/Hex) afforded desired product **3.30** as a white solid (90%). ¹H-NMR (500 MHz, DMSO-d6) δ 12.97 (s, 1H), 7.16 (s, 1H), 5.46 (s, 1H), 4.37 (s, 2H) ¹³C-NMR (500 MHz, DMSO-d6) δ 188.59, 132.59, 125.19, 55.88. HRMS [M+1] calc'd for C₄H₅NOS₂H 147.9891, found 147.9894.

Compound 3.31

To a stirred solution of crude peptide AcYACFAC-NH₂ in DMF (50 mM) was added Et₃N (3 equiv) and OFCP (1.5 equiv) at 0 °C. After stirring for 30 minutes the reaction was concentrated under reduced pressure. The crude product was then diluted with 16% DMF in THF (50 mM) and added compound **3.30** (1.1 equiv) followed by Me₃SiOK (3.5 equiv) at 0 °C. The mixture was then allowed to stir at RT until HPLC indicated full conversion. The reaction mixture was then cooled to 0 °C and quenched with AcOH (10 equiv), concentrated under reduced pressure, and purified by FCC 10 – 15% MeOH/CHCl³ to give **3.31** (18%).

Compound 3.32

To a stirred solution of 3.31 in MeNO₂ (5 mM) was added TFA (1 equiv) dropwise. The reaction was then heated to 80 °C for 12 h. After completion the reaction was quenched with NaHCO₃ (xs), concentrated under reduced pressure then purified by RP-HPLC to afford **3.32** (52%). ¹H-

NMR (500 MHz, DMSO-d6) δ 9.5 (s, 1H), 8.47 (d, J = 8.2 Hz, 1H), 8.40 (d, J = 8.5 Hz, 1H), 8.17 (d, J = 7.9 Hz, 1H), 8.08 (d, J = 7.6 Hz, 1H), 7.95 (s, 2H), 7.65 (d, J = 7.6 Hz, 1H), 7.31 – 7.25 (m, 3H), 7.21 (q, J = 7.9 Hz, 3H), 7.16 (d, J = 7.5 Hz, 2H), 7.12 (d, J = 12.1 Hz, 2H), 6.86 (d, J = 7.2 Hz, 2H), 6.68 (d, J = 8.1 Hz, 2H), 4.86 (d, J = 5.7 Hz, 1H), 4.70 (d, J = 8.0 Hz, 1H), 4.62 – 4.50 (m, 2H), 4.44 (g, J = 7.5 Hz, 1H), 4.25 – 4.14 (m, 2H), 4.03 (d, J = 15.7 Hz, 1H), 3.90 (d, J = 15.7 Hz, 1H), 3.76 – 3.66 (m, 1H), 3.61 – 3.49 (m, 2H), 1.86 (s, 3H), 1.25 – 1.22 (m, 3H). 1.01 (d, $J = 6.8$ Hz, 3H). ¹³C-NMR (500 MHz, DMSO-d6) δ 171.6, 171.3, 171.0, 169.3, 169.0, 168.8, 167.6, 153.2, 152.8, 143.3, 141.3, 136.6, 130.6, 129.4, 129.0, 128.4, 127.8, 126.6, 124.4, 124.4, 114.8, 55.4, 53.6, 53.6, 52.1, 48.4, 47.4, 46.5, 37.2, 36.1, 34.3, 32.9, 32.3, 22.5, 19.1, 17.3. HRMS [M+1] calc'd for $C_{41}H_{42}F_{4}N_8O_8S_4H$ 979.2023, found 979.2025.

Compound 3.33

3.33 was made based on procedures for compounds **3.31** and **3.32** to give a 25% yield over 2 steps. See NMR Spectra Appendix section below for crude NMR spectra of this compound. HRMS [M+1] calc'd for C₂₉H₂₈F₄N₆O₆S₄H 761.0968, found 761.0970.

Compound 3.34

A flask was charged with compound crude peptide AcYGHAC-NH² diluted with DMF (10 mM), Et₃N (1.5 equiv), and cooled to 0 °C. To the flask was added OFCP (3 equiv) dropwise and let stir for 30 min. Then the flask was concentrated under reduced pressure. The crude intermediate macrocycle was then diluted with a DMF/THF mixture (1:4 30 mM) a cooled to 0 °C. To the flask was added 3.27 (4 equiv), Cs₂CO₃ (6 equiv), and let stir until completion observed by HPLC. After completion the reaction was quenched with AcOH (15 equiv), concentrated under reduced pressured and purified by RP-HPLC to give **3.34** (20%).

Compound 3.35

To a flask was added 3.34 and diluted with dry MeNO₂ (5 mM). Tf₂NH (5 equiv) in dry MeNO₂ (5 mM) was then added under argon. The reaction mixture was then allowed to stir at RT for 1 h or until completion observed by HPLC. Afterwards, the reaction mixture was quenched with iPr_{2} Net (10 equiv) and concentrated under reduced pressure to afford crude product. Purification by RP-HPLC afforded pure **3.35** (45%) as a TFA salt. ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.15 (bs, 1H), 9.02 $(s, 1H)$, 8.65 (d, J = 7.5 Hz, 1H), 8.40 (t, J = 6.2 Hz, 1H), 8.05 (d, J = 7.7 Hz, 1H), 7.89 (d, J = 6.6 Hz, 1H), 7.45 – 7.33 (m, 1H), 7.06 – 7.0 (m, 2H), 6.96 (s, 1H), 6.90 – 6.83 (m, 3H), 6.70 (d, J = 8.1 Hz, 1H), 6.54 (d, J = 9.3 Hz, 1H), 6.18 (dt, J = 15.8, 6.3 Hz), 6.07 (d, J = 15.8 Hz, 1H), 4.61 – 4.53 (m, 2H), 4.35 (g, J = 7.0, 1H), 4.14 (quint, J = 7.4 Hz, 1H), 4.01 (dd, J = 15.9, 7.0 Hz), 3.80 $(s, 3H)$, 3.42 – 3.33 (m, 4H), 2.94 – 2.89 (m, 1H), 2.77 – 2.71 (m, 3H), 1.84 (s, 3H), 1.21 (d, J = 7.3 Hz, 3H). HRMS [M+1] calc'd for C₄₀H₄₀F₄N₈OSH 885.2653, found 885.2684.

Compound 3.36

A flask was charged with compound crude peptide AcWCASC-NH2, diluted with DMF (10 mM), added Et₃N (1.5 equiv), and cooled to 0 °C. To the flask was added OFCP (3 equiv) dropwise and let stir for 30 min. Then the flask was concentrated under reduced pressure. The crude intermediate macrocycle was then diluted with a DMF/THF mixture (1:4 30 mM) a cooled to 0 °C. To the flask was added 3.27 (4 equiv), Cs₂CO₃ (6 equiv), and let stir until completion observed by HPLC. After completion the reaction was quenched with AcOH (15 equiv), concentrated under reduced pressured and purified by RP-HPLC. The acyclic intermediate was then diluted with dry MeNO₂ (5 mM). To the flask was added ScOTf₃ (2 equiv) and let stir until completion observed by HPLC. Upon completion, the mixture was concentrated under reduced pressure and purified by RP-HPLC to obtain **3.36** (35%) and two other regioisomeric products for a 70% combined yield.

¹H NMR (DMSO-*d*6, 500 MHz) δ 10.9 (s, 1H), 9.22 (s, 1H), 8.24 (d, J = 7.2 Hz, 2H), 7.94 (d, J = 8.8 Hz, 1H), 7.60 (d, J = 7.6 Hz, 1H), 7.50 (s, 1H), 7.26 – 7.07 (m, 6H), 6.96 (dt, J = 29.7, 7.3 Hz, 2H), $6.64 - 6.45$ (m, 2H), 4.86 (d, $J = 5.7$ Hz, 1H), 4.80 (d, $J = 9.6$ Hz, 1H), $4.64 - 4.54$ (m, 2H), $4.38 - 4.33$ (m, 1H), 4.30 (t, J = 7.8 Hz, 1H), 3.82 (s, 3H), $2.44 - 2.16$ (m, 2H), $3.13 - 3.06$ (m, 4H), $2.90 - 2.80$ (m, 1H), $2.79 - 2.70$ (m, 1H), 2.67 (d, $J = 13.9$ Hz, 1H), 1.70 (s, 3H), 1.24 (d, $J =$ 7.2 Hz, 3H). ¹³C-NMR (500 MHz, DMSO-d6) δ 173.2, 172.0, 169.3, 169.2, 168.1, 168.0, 148.9, 141.8, 137.4, 134.7, 133.1, 130.2, 130.0, 129.2, 128.9, 128.2, 126.8, 125.3, 124.6, 120.2, 118.3, 118.0, 117.2, 113.2, 110.6, 108.4, 74.2, 61.3, 55.9, 54.6, 54.2,52.7, 50.3, 32.9, 31.1, 30.7, 26.8, 22.8, 17.9. HRMS [M+1] calc'd for $C_{40}H_{41}F_{4}N_{7}O_{9}SH$ 904.2422, found 904.2423.

Compound 3.38

To a stirred solution of crude peptide AcAsp(CSY)-Gly-Ser-Ala-Cys-NH₂ in dry DMF (10 mM) was added Et₃N (3 equiv), then OFCP (1.5 equiv) dropwise at 0 °C. The reaction was allowed to stir at 0 °C for 30 min until completion of macrocycle intermediate formation. The reaction was then concentrated under reduced pressure to remove excess Et₃N, OFCP, and MeCN. The mixture was then cooled to 0 °C and added Me₃SiOK (2 equiv) and allowed to stir at RT for 30 min. Then cysteamine (1.1 equiv) was added and let stir for 15 minutes. After completion as observed by HPLC, the reaction was quenched with AcOH (10 equiv) and concentrated under reduced pressure to afford crude product. Purification by FCC (5 -10% MeOH/CHCl3.) afforded acyclic intermediate macrocycle. This material was added to a flask diluted with 2:8 MeCN:NaOAc/AcOH (aq) (1 mM) then added NCS in MeCN (1 equiv, 10 mM) in two 0.5 equiv portions. After completion as observed by HPLC the reaction was concentrated under reduced pressure. Purification by RP-HPLC afforded **3.38**. Material was pushed forward.

Compound 3.39

To a stirred solution of 3.38 in dry DMF (5 mM) was added iPr₂Net (3.5 equiv) and allowed to cool to 0 °C. Then T3P (1.5 equiv) was added and the reaction mixture was allowed to stir at 0 °C for 30 minutes. Upon completion as observed by HPLC, the mixture was quenched with AcOH (5 equiv). The reaction was then concentrated under reduced pressure to afford an orange oil as crude product. Purification by FCC (5 – 10% MeOH/CHCl3) afforded **3.39** (50%; 30% from linear peptide SM). ¹H NMR (DMSO-*d*6, 500 MHz) δ 8.6 (J = 4.1 Hz, 1H), 8.06 (d, J = 7.7 Hz, 1H), 7.92 $(d, J = 7.7$ Hz, 1H), $7.81 - 7.73$ (m, 1H), $7.56 - 7.52$ (s, 1H), 7.50 (d, $J = 7.6$ Hz, 1H), 7.41 (s, 1H), $7.31 - 7.27$ (m, 2H), $7.27 - 7.19$ (m, 4H), 4.95 (d, $J = 7.3$ Hz, 1H), 4.83 (d, $J = 9.8$ Hz, 1H), $4.56 - 7.3$ 4.48 (m, 1H), 4.26 – 4.14 (m, 5H), 4.10 – 4.04 (m, 2H), 3.37 – 3.26 (m, 3H), 3.18 – 3.07 (m, 2H), 3.04 – 2.97 (m, 2H), 1.80 (s, 3H), 1.16 (d, J = 6.7 Hz, 3H). ¹³C-NMR (500 MHz, DMSO-d6) δ 171.4, 171.2, 170.9, 168.9, 168.6, 168.3, 167.8, 162.3, 136.9, 129.0, 128.3, 126.6, 57.3, 56.4, 52.6, 50.4, 48.5, 41.6, 41.1, 36.1, 35.8, 32.0, 30.8, 30.3, 22.5, 16.0. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -102.7 – (104.2) (m, 1F), -107.0 – (-108.3) (m, 1F), -114.6 – (-115.9) (m, 1F), -114.4 – (-120.8) (m, 1F). HRMS [M+1] calc'd for $C_{33}H_{38}F_4N_8O_9S_2H$ 831.2218, found 831.2220.

Compound 3.40

To a stirred solution of resorcinol (2 equiv) in acetone (0.5 M) was added benzyl chloride (1 equiv) and K_2CO_3 (1 equiv) then allowed to reflux overnight. After completion as observed by TLC the reaction was filtered and concentrated under reduced pressure then purified by FCC (10 – 40% EtOAc/Hex) to afford 3.40 (65%). ¹H and ¹³C NMR were identical to literature reported spectra.

Compound 3.41

To a stirred solution of **3.40** in DCM (25 mM) was added NBS (1 equiv) at 0 °C. The reaction was stirred at RT for 2 h, then washed with water, separated the organic layer, dried over $Na₂SO₄$ and concentrated under reduced pressure. Purification by FCC (2 – 10% EtOAc/Hex) afforded **3.41**. ¹H and ¹³C NMR were identical to literature reported spectra.

Compound 3.42S1

A stirred solution of **3.41** and HMDS (2 equiv) in THF (2.9 M) was refluxed for 90 min. The solvent was evaporated under reduced pressure, and the residue was subjected to vacuum to remove excess NH₃ and unreacted HMDS. The crude product was then dissolved in THF (0.78 M), cooled to -100 °C and n-BuLi (2 equiv) was added dropwise. The mixture was stirred for 20 min while the temperature reached -80 °C. Then the mixture was cooled again to -100 °C, and added Tf_2O (2 equiv) dropwise while continuing to stir for 20 min while the temperature reached -80 °C. Cold sat. aq. NaHCO₃ was added, the phases separated and extracted with Et₂O. The combined organic layers were dried with $Na₂SO₄$, filtered and concentrated under reduced pressure. Purification by FCC afforded **3.42** (0 – 31%). ¹H NMR (CDCl3, 500 MHz) δ 7.47 – 7.32 (m, 6H), $7.02 - 6.93$ (m, 2H), 5.08 (s, 2H), 0.34 (s, 9H). ¹³C-NMR (500 MHz, CDCl₃) δ 161.1, 155.7, 136.9, 136.1, 128.9, 127.7, 123.4, 114.0, 107.2, 70.6, -0.7.

Compound 3.43

Original Route

3.42 was dissolved in MeOH (40 mM). Raney Ni slurry was weighed out (~1g), added to the reaction mixture, flushed with argon and allowed to stir at RT. After 1 h the reaction showed completion. The mixture was then filtered with celite and concentrated under reduced pressure. Purification by FCC (100% DCM) to give **3.43** as a clear oil (40%).

*Optimized Route*S2

To a stirred solution of 4-bromoresorcinol in THF (0.53 M) under inert atmosphere was added HMDS (2.2 equiv). The resulting solution was heated to reflux for 12 h, then concentrated under reduced pressure. The crude material was dissolved in anhydrous THF (0.4 M) and cooled to -78 °C under inert atmosphere. nBuLi (1.1 equiv) was slowly added dropwise, then Tf_2O (1.5 equiv) was added and stirred for 10 min. The reaction was then quenched with 5% aq. HCl. The resulting mixture was extracted with EtOAc, organic layers combined washed with sat. NaHCO $_3$, dried over Na2SO4, filtered, and concentrated under reduced pressure. Purification by FCC (5% EtOAc/Hex) afforded **3.43** as a clear oil (15 – 45%). ¹H NMR (CDCl₃, 500 MHz) δ 7.37 (d, J = 8.2 Hz, 1H), 6.87 (d, $J = 2.16$ hz, 1H), 6.82 (dd, $J = 8.2$, 2.2 Hz, 1H), 0.33 (s, 9H).

Compound 3.46

To a stirred solution of 3.45 in 20% DMF/THF (50 mM) was added 3.43 (10 equiv) and Me₃SiOK (2 equiv) at 0 °C. The reaction was allowed to stir until complete as observed by HPLC. Upon completion the reaction was quenched with AcOH (5 equiv) and concentrated under reduced pressure. Purification by FCC (0 – 10% MeOH/CHCl3) afforded **3.46**. ¹H NMR (DMSO-*d*6, 300 MHz) δ 10.81 (s, 1H), 8.57 (d, J = 8.2 Hz, 1H), 8.38 (d, J = 9.0 Hz, 1H), 8.29 (d, J = 8.1 Hz, 1H), 8.01 (d, J = 7.6 Hz, 1H), 7.64 (d, J = 8.4 Hz, 1H), 7.53 (d, J = 7.8 Hz, 1H), 7.36 (dd, J = 8.3, 2.2 Hz, 1H), 7.31 (d, J = 8.0 Hz, 1H), 7.22 – 7.13 (m, 2H), 7.13 – 6.91 (m, 3H), 5.06 – 4.94 (m, 1H), 4.89 – 4.79 (m, 1H), 4.68 – 4.56 (m, 1H), 4.52 – 4.42 (m, 1H), 4.41 – 4.32 (m, 1H), 4.32 – 4.19 (m, 1H), 3.92 – 3.80 (m, 1H), 3.68 – 3.56 (m, 1H), 3.56 – 3.45 (m, 1H), 3.23 – 2.93 (m, 1H), 2.92 $-$ 2.71 (m, 1H), 1.74 (s, 3H), 1.23 (d, J = 7.3 Hz, 3H), 0.31 (s, 9H).

Compound 3.47S3

To a solution of 3-bromophenol in THF (0.25 M) was added nBuLi (2.7 eq.) at -78 °C dropwise. The reaction mixture was stirred at this temperature for 30 min, then added TMSCl (3 equiv) and the solution was allowed to reach room temperature. The reaction was quenched with sat. aq. $NH₄Cl$, extracted with EtOAc (x3), dried combined organic layers over $Na₂SO₄$, filtered, and concentrated under reduced pressure. Purification by FCC (100% Hex) afforded **3.47** (98%). ¹H and ¹³C NMR were identical to literature reported spectra.^{S4}

Compound 3.48^{S5}

To a solution of 3.47 in AcOH (1.19 M) was added a solution of Br_2 in AcOH (6.2 M) at 0 °C. After 30 min the reaction mixture was transferred to a separatory funnel and washed with water (x5), extracted with EtOAc, dried over Na₂SO₄ and concentrated under reduced pressure. Purification by FCC (0-5% EtOAc/Hex) afforded **3.48** (44%). ¹H NMR (CDCl3, 500 MHz) δ 7.37 (d, J = 8.52 Hz, 1H), 6.91 (d, J = 3.12 Hz, 1H), 6.69 (dd, J = 8.5, 3.2 Hz, 1H), 4.91 (bs, 1H), 0.34 (s, 9H). ¹³C NMR (CDCl3) δ 154.3, 142.7, 133.9, 123.0, 120.8, 117.9, -0.57.

Compound 3.50

To a stirred solution of 2.5 in DMF (0.1 M) was added 3.48 (1.1 equiv) and Cs₂CO₃ (1.1 equiv) at 0 °C and allowed to stir until completion as observed by TLC. Upon completion, the reaction mixture was quenched with AcOH (5 equiv) concentrated under reduced pressure. Purification by FCC (100% Hex) afforded **3.50** (57%). ¹H NMR (CDCl3, 500 MHz) δ 7.31 (d, J = 8.6 Hz, 1H), 7.32 -7.12 (m, 2H), $7.08 - 7.0$ (m, 1H), $6.69 - 6.60$ (m, 3H), $6.60 - 6.53$ (m, 1H), 0.31 (s, 1H).

Compound 3.51

A stirred solution of 2-Bromo-4-methylphenol and HMDS (1.1 equiv) in THF (2.9 M) was refluxed for 90 min. The solvent was evaporated under reduced pressure, and the residue was subjected to vacuum to remove excess $NH₃$ and unreacted HMDS. The crude product was then dissolved in THF (0.78 M), cooled to -100 °C and n-BuLi (1.1 equiv) was added dropwise. The mixture was stirred for 20 min while the temperature reached -80 °C. Then the mixture was cooled again to - 100 °C, and added Tf₂O (1.5 equiv) dropwise while continuing to stir for 20 min while the temperature reached -80 $^{\circ}$ C. Cold sat. aq. NaHCO₃ was added, the phases separated and extracted with Et₂O. The combined organic layers were dried with $Na₂SO₄$, filtered and concentrated under reduced pressure. Purification by FCC (0 – 5% EtOAc/Hex) afforded **3.51** (65%).¹H and ¹³C NMR were identical to literature reported spectra.

Compound 3.52 and 3.53

To a stirred solution of **3.51** in MeCN (0.1 M) was added pyrrolidine (2 equiv) and CsF (2 equiv) at 0 °C. The reaction was stirred until completion as observed by TLC. Upon completion the mixture was washed with sat. aq. NaHCO₃, extracted with EtOAc, dried over Na₂SO₄, and concentrated under reduced pressure. Purification by (0 – 10% EtOAc/Hex) afforded regioisomers **3.52** and **3.53** (62% combined). ¹H and ¹³C NMR were identical to literature reported spectra.

Compound 3.57

To a stirred solution of crude peptide Cbz-Glu(Cys)-Ala-Histamine in DMF (5 mM) was added Et₃N (2.5 equiv), and OFCP (1.5 equiv) at 0 °C and let stir for 30 min. After initial macrocyclization was complete, the mixture was concentrated under reduced pressure to remove excess Et_3N , OFCP, and MeCN. The reaction mixture was then cooled once again to 0° C and added Me₃SiOK (4 equiv) and let stir at RT. After reaction completion as observed by HPLC the mixtue was quenched with AcOH (10 equiv) and concentrated under reduced pressure. Purification by FCC

(0 - 10% MeOH/CHCl3) afforded **3.57** (41%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 9.96 (bs, 1H), 9.88 (bs, 1H), 8.48 (s, 1H), 8.01 (s, 1H), 7.67 (s, 1H), 7.46 (s, 1H), 7.42 - 7.24 (m, 5H), 7.07 (s, 1H), 5.07 - 4.92 (m, 2H), 4.89 (s, 1H), 4.4 - 4.25 (m, 1H), 4.04 (q, J = 12.3 Hz 1H); 3.09 (d, J = 13.0 Hz, 1H), 3.8 - 3.64 (m, 1H), 2.98 (d, J = 15.0 Hz, 1H), 2.85 - 2.63 (m, 2H), 2.29 - 2.16 (m, 1H); 2.13 - 1.98 (m, 1H), 1.97 - 1.72 (m, 2H), 1.18 - 1.02 (m, 3H). ¹³C-NMR (500 MHz, DMSO-d6) δ 174.2, 171.3, 170.6, 170, 155.3, 140.6, 139.8 - 139.1 (m, 1C), 138.4, 137, 128.4, 127.9, 127.8, 123.3 - 122.9 (m, 1C), 115.5 - 109.5 (m, 2C), 113.4, 65.4, 53.5, 47.5, 46, 35.5, 31.4, 30.9, 29.9, 28.2, 17.9. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -110.7 (dt, J = 251.3, 13.5 hz, 1F), -112.2 (dd, J = 223.7, 8.9 Hz, 1F), -116.9 (dt, J = 251.5, 12.5 Hz, 1F), -119.8 (dd, J = 224.0, 10.5 Hz, 1F), -146.4 (t, J = 13.9 Hz, 1F). HRMS [M+1] calc'd for $C_{29}H_{30}F_5N_7O_6SH$ 700.1977, found 700.1998.

Compound 3.58

To a stirred solution of crude peptide des-aminoTyr-Pro-Cys-Val-Ala-Cys-NH² in dry DMF (5 mM) was added Et₃N (3 equiv), and OFCP (1.5 equiv) at 0 °C and let stir for 30 min. After initial macrocyclization was complete, the mixture was concentrated under reduced pressure to remove excess Et₃N, OFCP, and MeCN. The reaction mixture was then cooled once again to 0 $^{\circ}$ C and added $Cs₂CO₃$ (8 equiv) and let stir at RT. After reaction completion as observed by HPLC the mixture was quenched with AcOH (20 equiv) and concentrated under reduced pressure. Purification by FCC (0 - 10% MeOH/CHCl3) afforded **3.58** (26%).[Note: rotational isomers present] ¹H NMR (DMSO- d_6 , 500 MHz) δ 8.31 (d, J = 9.5 Hz, 1H), 7.90 (d, J = 7.66 Hz, 1H), 7.70 (d, J = 6.9 Hz, 1H), 7.56 (d, J = 9.0 Hz, 1H), 7.42 (d, J = 8.4, 1H), 7.35 (bs, 1H), 7.30 (d, J = 8.3, 1H), 7.22 – 7.14 (m, 1H), 6.97 – 6.90 (m, 1H), 4.81 (d, J= 7.8 Hz, 1H), 4.71 (d, J = 6.6 Hz, 1H), 4.40 $(d, J = 6.7$ Hz, 1H), $4.34 - 4.28$ (m, 1H), $4.25 - 4.18$ (m, 1H), $3.94 - 3.88$ (m, 1H), $3.64 - 3.53$ (m, 2H), 3.51 – 3.41 (m, 1H), 3.07 – 2.97 (m, 1H), 2.86 (d, J = 4.8 Hz, 3H), 2.82 (d, J = 3.8 Hz, 1H), 2.78 (d, J = 14.5 Hz, 1H), 2.75 – 2.68 (m, 1H), 2.22 – 2.15 (m, 1H), 2.08 – 1.93 (m, 2H), 1.88 – 1.82 (m, 1H), 1.70 – 1.63 (m, 2H), 1.31 (d, J = 7.4 Hz, 3H), 0.80 (dd, J = 15.9, 6.5 Hz, 6H). ¹³C-NMR (500 MHz, DMSO-d6) δ 172. 2, 171.9, 171.4, 170.4, 168.5, 167.6, 152.8, 137.8, 131.5, 128.9, 128.4, 127.8, 118.6, 115.6, 60.1, 58.8, 55.1, 52.1, 49.5, 48.4, 47.3, 35.5, 35.3, 35.2, 33.8, 30.4, 29.5, 25.9, 24.3, 19.0, 18.5, 17.9. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -106.2 – (-107.4) (m, 1F), -113.0 – (-114.0) (m, 1F), -119.1 – (-120.8) (m, 2F). HRMS [M+1] calc'd for C34H40F4N6O7S2H 785.2414, found 785.2414.

Compound 3.59

To a stirred solution of Ac-Cys-Thr-Phe-Cys-NHMe in DMF (5 mM) was added Et_3N (2.5 equiv) and OFCP (1.5 equiv) dropwise at 0 $^{\circ}$ C and let stir for 30 min. After initial macrocyclization was complete, the mixture was concentrated under reduced pressure to remove excess $Et₃N$, OFCP, and MeCN. The reaction mixture was then cooled once again to 0 °C and added $Cs₂CO₃$ (4 equiv), phenol (1.5 equiv) and let stir at RT. After reaction completion as observed by HPLC the mixture was quenched with AcOH (10 equiv) and concentrated under reduced pressure. Purification by FCC (0 - 10% MeOH/CHCl3) afforded **3.59**.

Compound S3.7

To a stirred solution of **3.18** in DMF (50 mM) was added Et₃N (2.2 equiv) and 1,4-benzenedithiol (0.5 equiv) at 0 °C and let stir and warm to RT. Upon completion as observed by HPLC, the reaction was quenched with AcOH (5 equiv) and concentrated under reduced pressure. Purification by RP-HPLC afforded the dimerized product **S3.7** (20%). ¹H NMR (DMSO-*d*6, 500 MHz) δ 10.91 (s, 2H), 9.51 (s, 2H), 8.50 (d, J = 9.1 Hz, 2H), 8.39 (d, J = 8.0 Hz, 2H), 8.31 (d, J = 8.5 Hz, 2H), 7.69 (d, J = 19.5 Hz, 2H), 7.50 (d, J = 7.7 Hz, 2H), 7.38 – 7.29 (m, 6H), 7.22 (d, J =

9.0 Hz, 2H), 7.13 (s, 2H), 7.07 (t, J = 7.3 Hz, 2H), 6.98 (t, J = 7.3 Hz, 2H), 5.02 (s, 2H), 4.83 (d, J $= 8.9$ Hz, 2H), 4.66 (q, J = 7.7 Hz, 2H), 4.47 (q, J = 7.4 Hz, 2H), 4.42 – 4.33 (m, 2H), 4.26 – 4.17 (m, 2H), 3.88 – 3.76 (m, 2H), 3.66 (d, 2H), 3.55 – 3.42 (m, 5H), 3.11 – 3.03 (m, 6H), 2.82 (d, J = 14.0 Hz, 2H), 1.79 (s, 6H). ¹³C-NMR (500 MHz, DMSO-d6) δ 171.8, 169.8, 169.5, 168.7, 167.6, 167.0, 136.1, 131.6, 130.4, 129.7, 128.7, 127.1, 123.7, 121.0, 118.5, 118.0, 111.4, 109.1, 79.2, 65.7, 61.6, 55.7, 55.0, 51.5, 46.1, 45.6, 36.0, 35.9, 35.8, 33.7, 33.0, 32.6, 31.9, 31.3, 30.7, 29.0, 28.9, 28.7, 28.7, 28.7, 28.0, 27.6, 26.2, 25.8, 25.5, 25.4, 25.3, 22.5, 22.4, 22.3, 22.1, 19.4, 19.1, 14.0, 14.0, 12.0, 11.2, 8.5. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ -96.8 –(-98.8) (m, 2F), - 111.3 – (- 113.8) (m, 4F), -127.0 – (-128.7) (m, 2F).

Compound S3.11

To a stirred solution of crude peptide AcCysSerCys-NH₂ in DMF (5 mM) was added Et₃N (2.5 equiv) and OFCP (1.5 equiv) at 0 °C. The reaction was allowed to stir until completion as observed by HPLC. Upon completion, the reaction was concentrated under reduced pressure and purified by FCC (0 – 10% MeOH/CHCl3) to give **S3.11** (55%). 1H NMR (DMSO-d6, 500 MHz) δ 8.63 (d, J $= 9.2$ Hz, 1H), 8.22 (d, J = 7.4 Hz, 1H), 8.12 (d, J = 8.9 Hz, 1H), 7.41 (s, 1H), 7.23 (s, 1H), 4.99 $(t, J = 5.4$ Hz, 1H), 4.75 (g, $J = 6.2$ Hz, 1H), 4.63 – 4.48 (m, 2H), 3.67 – 3.50 (m, 2H), 3.45 (dd, J $= 14.0, 4.1$ Hz, 1H), 3.36 – 3.34 (m, 1H), 3.24 (dd, J = 13.9, 8.7 Hz, 1H), 1.89 (s, 3H). ¹³C NMR (DMSO-d6, 126 MHz) δ 171.0, 170.5, 169.2, 168.9, 143.2 – 142.2 (m, 1C), 138.2 – 137.3 (m, 1C), 119.0 – 110.3 (m, 3C), 60.9, 54.5, 51.9, 50.6, 35.5, 34.8, 22.4. ¹⁹F NMR (DMSO-*d*6, 282 MHz) δ $-104.7 - 108.3$ (m, 4F), -128.8 (q, J = 260 Hz, 2F). HRMS m/z [M + H]⁺ calc'd for C16H18F6N4O5S2H 525.0701, found 525.0712.

Compound S3.12

Synthesized according to General Procedure B using crude peptide AcCysValTrpSerCys-NH² and purified by FCC (0 – 10% MeOH/CHCl₃) to give **S3.12** (25%).¹H NMR (DMSO- d_6 , 500 MHz) δ 10.86 (s, 1H), 8.40 (d, J = 8.4 Hz, 1H), 8.29 (d, J = 8.4 Hz, 1H), 8.10 (d, J = 8.5 hz, 1H), 7.95 (d, J = 6.0 Hz, 1H), 7.84 (d, J = 6.4 Hz, 1H), 7.56 (d, J = 7.9 Hz, 1H), 7.33 (d, J = 8.1 Hz, 2H), 7.20 (s, J = 8.1 Hz, 1H), 7.06 (t, J = 7.5 Hz, 1H), 7.0 (t, J = 7.3 Hz, 2H), 4.88 (s, 1H), 4.74 (q, J = 7.7 Hz, 1H), 4.45 – 4.43 (m, 2H), 4.22 – 3.98 (m, 3H), 3.77 – 3.68 (m, 1H), 3.57 – 3.45 (m, 2H), $3.31 - 3.22$ (m, 2H), $3.22 - 3.13$ (m, 4H), $3.07 - 2.99$ (m, 1H), 1.85 (s, 3H), 0.67 (dd, 6H).¹³C NMR (DMSO-d6, 126 MHz) δ 171.4, 171.3, 169.6, 169.4, 169.3, 168.5, 136.1, 127.3, 123.6, 121.0, 118.3, 111.3, 109.6, 60.7, 60.1, 58.8, 55.4, 54.9, 54.4, 52.0, 48.6, 48.4, 33.3, 30.7, 30.3, 29.2, 29.1, 27.6, 22.3, 19.2, 17.3.

Compound S3.13

To a stirred solution of crude peptide AcTyr(OMe)CysAlaSerCys-NH² in DMF (5 mM) was added Et₃N (2.5 equiv) and OFCP (1.5 equiv) at 0 °C. The reaction was allowed to stir until completion as observed by HPLC. Upon completion, the reaction was concentrated under reduced pressure and purified by FCC (0 – 10% MeOH/CHCl₃) to give **S3.13** (42%).¹H NMR (DMSO- d_6 , 500 MHz) δ 8.70 (d, J = 9.4 Hz, 1H), 8.63 (d, J = 7.3 Hz, 1H), 8.25 (d, J = 7.4 Hz, 1H), 8.11 (d, J = 8.32 Hz, 1H), 7.51 (s, 1H), 7.25 (d, J = 7.2 Hz, 1H), 7.24 (s, 1H), 7.14 (d, 2H), 6.80 (d, J = 8.7 Hz, 2H), 5.48 (bs, 1H), 4.73 (q, J = 5.9 Hz, 1H), 4.53 – 4.34 (m, 3H), 4.32 – 4.24 (m, 1H), 4.10 (quint, J = 7.3 Hz, 1H), 3.78 (dd, J = 14.7, 3.0 Hz, 1H), 3.70 (s, 3H), 3.69 – 3.63 (m, 1H), 3.53 – 3.42 (m, 2H), 2.91 – 2.81 (m, 2H), 2.65 (dd, J = 13.8, 9.7 Hz, 1H), 1.73 (s, 3H), 1.27 (d, J = 7.4 Hz, 3H). ¹³C NMR (DMSO-d₆, 126 MHz) δ 172.1, 171.4, 171.0, 169.2, 167.9, 157.8, 130.1, 129.6, 113.4, 61.8, 54.9, 54.0, 50.7, 50.5, 50.3, 36.4, 34.0, 32.2, 30.7, 22.4, 17.3.

Compound S3.19

Synthesized according to General Procedure B using crude peptide AcTrpHisAlaSerCys-NH₂ and purified by FCC (0 – 10% MeOH/CHCl3) to give **S3.13** (23%).¹H NMR (DMSO-*d*6, 500 MHz) δ 10.90 (s, 1H), 9.50 (s, 1H), 8.54 (d, J = 7.4 Hz, 1H), 8.26 (d, J = 6.5 Hz, 1H), 8.03 (s, 1H), 7.99 (d, J = 7.0 Hz, 1H), 7.59 (d, J = 7.6 Hz, 1H), 7.50 (s, 1H), 7.40 – 7.27 (m, 3H), 7.18 (s, 1H), 7.09 -6.93 (m, 3H), 4.97 (s, 1H), 4.58 – 4.45 (m, 2H), 4.45 – 4.35 (m, 1H), 4.13 – 4.04 (m, 1H), 3.98 – 3.90 (m, 1H), 3.77 – 3.65 (m, 2H), 3.22 – 3.12 (m, 3H), 3.01 – 2.92 (m, 1H), 1.84 (s, 3H), 1.23 $(d, J = 6.5, 3H)$. ¹³C NMR (DMSO-d₆, 126 MHz) δ 172.7, 172.1, 171.5, 170.2, 167.9, 140.2, 137.9, 136.1, 127.2, 123.6, 120.9, 118.3, 116.1, 111.4, 111.1, 110.1, 60.4, 56.2, 54.3, 51.6, 50.1, 48.0, 45.6, 31.1, 28.9, 27.1, 22.6, 17.3.
Parallel Artificial Membrane Permeability Assay (PAMPA) Experiment

This assay was performed as outlined in the protocol of the parallel artificial permeability assay kit (BioAssay Systems Cat. # PAMPA-096). Incubation was performed at RT for 18 hours. Data analysis was performed using a Tecan M1000 plate reader and 96-well UV plates purchased from BioAssay Systems (Cat. # P96 UV). UV absorbance was measured from 230 to 500 nm in 10 nm intervals to determine peak absorbance of test compounds and controls. The following equation used to determined Permeability Rate (P_e) :

$$
P_e = C \times -\ln(1 - \frac{OD_A}{OD_F}) \text{ cm/s}
$$

Where OD_A is the absorbance of Acceptor Solution minus Blank, OD_e is the absorbance of Equilibrium Standard minus Blank, and, using an 18 h incubation, $C = 7.72 \times 10^{-6}$.

Figure S3.1. Permeability rates of full library including permeability standards.

Table S3.2. Compound numbers and notebook pages corresponding to wells of 96-well plate

Raw Experimental Absorbance Data

NMR Spectra

HMBC spectrum of compound 3.10 (DMSO-*d***6)**

Polycycle **3.10** (600 MHz, DMSO-d6, 298K)

HMBC spectrum of compound 3.11 (DMSO-*d***6)**

HSQC spectrum of compound 3.11 (DMSO-*d***6)**

COSY spectrum of compound 3.11 (DMSO-*d***6)**

Polycycle **3.13** (600 MHz, DMSO-d6, 298K)

198

200

¹³C NMR of compound 3.15 (DMSO-d6, 126 MHz)

HMBC spectrum of compound 3.13 (DMSO-*d***6)**

HMBC spectrum of compound 3.17 (DMSO-*d***6)**

¹⁹F NMR of compound 3.19 (DMSO-d₆, 282 MHz)

HSQC spectrum of compound 3.20 (DMSO-*d***6)**

222

¹H NMR of compound 3.21 (DMSO-d₆, 500 MHz)

F NMR of compound 3.23 (DMSO-*d***6, 282 MHz)**

F NMR of compound 3.25(DMSO-*d***6, 282 MHz)**

F NMR of compound 3.26 (DMSO-*d***6, 282 MHz)**

H NMR of compound S3.1 (CDCl3, 300 MHz)

H NMR of compound S3.2 (CDCl3, 500 MHz)

H NMR of compound S3.3 (CDCl3, 500 MHz)

H NMR of compound 3.27 (CDCl3, 500 MHz)

HMBC spectrum of compound 3.32 (DMSO-*d***6)**

HSQC spectrum of compound 3.32 (DMSO-*d***6)**

HMBC spectrum of compound 3.33 (DMSO-*d***6)**

HMBC spectrum of compound 3.36 (DMSO-*d***6)**

COSY spectrum of compound 3.36 (DMSO-*d***6)**

¹⁹F NMR of compound 3.39 (DMSO-d₆, 282 MHz)

HMBC spectrum of compound 3.39 (DMSO-*d***6)**

HSQC spectrum of compound 3.39 (DMSO-*d***6)**

¹H NMR of compound 3.43 (CDCl3, 500 MHz)

Compound **3.57** (500 MHz, DMSO-d6, 298K)

275

COSY spectrum of compound 3.57 (DMSO-*d***6)**

HMBC spectrum of compound 3.58 (DMSO-*d***6)**

HSQC spectrum of compound 3.58 (DMSO-*d***6)**

COSY spectrum of compound 3.58 (DMSO-*d***6)**

References

- [S1] Lakshmi, B. V.; Wefelscheid, U. K.; Kazmaier, U. *Synlett*, **2011** *2011*(03), 345-348.
- [S2] Xu, H.; He, J.; Shi, J.; Tan, L.; Qiu, D.; Luo, X.; Li, Y. *J. Am. Chem. Soc.* **2018**, *140*(10), 3555-3559.
- [S3] Guo, P.; Wang, K.; Jin, W. J.; Xie, H.; Qi, L.; Liu, X.Y.; Shu, X. Z. *J. Am. Chem. Soc.* **2020** *143*(1), 513-523.
- [S4] Dichiarante, V.; Salvaneschi, A.; Protti, S.; Dondi, D.; Fagnoni, M.; Albini, A. *J. Am. Chem. Soc.* **2007**, *129*(51), 15919-15926.
- [S5] Tadashi, H. *Yakugaku Zasshi* **1960**, 80, 1399 1404.

Chapter 4 Appendix Material

General Methods

See Chapter 2 Appendix Material.

General Procedure A – Peptide synthesis

Peptides were prepared using standard solution phase techniques and the Boc protection strategy, or standard solid phase synthesis using the Fmoc protection strategy. Couplings were performed with either O-(benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium hexafluorophosphate (HBTU) or O-(benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium tetrafluoroborate (TBTU). Boc deprotection was performed using 4 M HCl in dioxane or 1:1 TFA in DCM. Cbz deprotection was performed with palladium on carbon in methanol under an atmosphere of hydrogen. Fmoc deprotection was performed as described using DBU in the presence of octyl mercaptan. N-Acetylation was performed under coupling conditions with acetic acid. Prior to removal of the final protecting group, peptides were purified by column chromatography on $SiO₂$ eluted with 2-12% MeOH in CHCl₃. Peptide identities were verified by HPLC-ESI-MS.

General Procedure B – Acylation of peptides with G1

The peptide (1.5 eq) was dissolved in N,N-DMF (0.2M), neutralized with Hünig's base (1.5 eq) to free the ammonium salt, where necessary, and treated with **G1**. The reaction was stirred at room temperature until complete by TLC or HPLC, and was worked up by partitioning between EtOAc and saturated NaHCO₃. The organic phase was washed with saturated NaHCO₃ (x2) and brine, then dried over $MgSO_4$ and concentrated. Purification was accomplished by flash chromatography on $SiO₂$ eluted with 0-12% MeOH in CHCl₃.

General Procedure C – Acylation of peptides with G2/G3 then Pictet Spengler Annulation

The peptide (1.5 eq) was dissolved in N,N-DMF (0.2M), neutralized with Hünig's base (1.5 eq) to free the ammonium salt, where necessary, and treated with **G1**. The reaction was stirred at room temperature until complete by TLC or HPLC, and was worked up by partitioning between EtOAc and saturated NaHCO₃. The organic phase was washed with saturated NaHCO₃ (x2) and brine, then dried over $MqSO₄$ and concentrated. Crude linear precursor was dissolved in a 4:1 mixture of AcOH/H2O (0.2M) and sturred until HPLC analysis confirmed reaction completion – typicall 12 hours. The volatiles were removed and the residue was rotovapped from acetonitrile (x3) followed by CHC I_3 (x3) to remove residual AcOH. Purification was accomplished by flash chromatography on $SiO₂$ eluted with 0-12% MeOH in CHCl₃.

General Procedure D – Palladium catalyzed macrocyclization

For G1 and G2 Macrocycles – Pd(PPh3)⁴ as catalyst

The acyclic cinnamyl carbonate was dissolved in dry N,N-DMF (5 mM) and sparged with argon for approximately 15 min. The reaction vessel was briefly opened to introduce $Pd(PPh₃)₄$ (5 mol%) as a solid or stock solution in N,N-DMF, and sparging was continued for approximately 5 min. All reactions proceeded to completion within 60 min at rt, and were halted by passing air through the reaction for several minutes, causing the yellow color of the catalyst to fade. The reaction mixture was then concentrated to dryness, reconstituted in N,N-DMF and purified by preparative RP-HPLC unless otherwise noted.

For G3 Macrocycles – [PdCl(C3H5)]² as catalyst

A flask was charged with Pictet-Spengler product (1 eq.) , Cs₂CO₃ (2 eq.) , and DMF (5 mM in) substrate) and sparged for 30 minutes. In a glove bag, a flame-dried Schlenk tube was charged with $[PdCl(C₃H₅)]₂$ (9 mg) and Xantphos (37 mg). Outside of the glovebag, the Schlenk tube was charged with 9 mL of 1:1 THF/DMF, which had been separately sparged for 1 hour. The catalyst solution was stirred for 5 minutes under Ar and 4 mol% Pd was added to the reaction flask via syringe. Reaction progress was monitored by analytical HPLC-MS. After reaction completion, the reaction was diluted with EtOAc and washed with 3x NH4Cl and 1x brine. The organic layer was dried with MgSO4 and concentrated in vacuo. Purificaiton by preparative HPLCby first dissolving the crude in ~1 mL DMSO.

General Procedure E – Acid promoted macrocyclization

(*Note: Commercial nitromethane should be aged over activated 3 Å molecular sieves for approximately one week. Alternatively, nitromethane may be treated with oven dried Brockmann type 1 neutral alumina overnight.)* The acyclic tert-butyl/iso-butyl carbonate was dissolved or suspended in dry nitromethane (5 mM) under ambient atmosphere, and treated with TFA (5-10%) t room temperature. After consumption of SM the reaction was worked up by partitioning between EtOAc and saturated NaHCO₃. The organic phase was washed with saturated NaHCO₃ (x2) and brine, then dried over MgSO₄ and concentrated. Alternatively, reactions were quenched with Hünig's base (15 eq.) and concentrated. The product mixture was reconstituted in N,N-DMF and purified by preparative RP-HPLC (see individual examples, below). Samples were analyzed by RP-HPLC-UV with detection at 254nm (see individual methods, chromatograms, below).

Full Library Data Set

Experimental Procedures

Peptide S4.1

Synthesized according to general procedure A (solution phase).

Peptide S4.2

Synthesized according to general prodedure A (solution phase).

Peptide S4.3

Synthesized according to general procedure A (solution phase).

Peptide S4.4

Synthesized according to general procedure A (solution phase).

Peptide S4.6

Synthesized according to general procedure A (solution phase).

Peptide S4.7

Synthesized according to general procedure A (solution phase).

Peptide S4.8

Synthesized according to general procedure A (solution phase).

Peptide S4.9 Synthesized according to general procedure A (solution phase).

Peptide S4.10

Synthesized according to general procedure A (solution phase).

Peptide S4.11

Synthesized according to general procedure A (solution phase).

Peptide S4.12

Synthesized according to general procedure A (solution phase).

Peptide S4.13

Synthesized according to general procedure A (solution phase).

Peptide S4.14

Synthesized according to general procedure A (solution phase).

Peptide S4.15

Synthesized according to general procedure A (solution phase).

Peptide S4.16

Synthesized according to general procedure A (solution phase).

Peptide S4.17 Synthesized according to general procedure A (solution phase). ЭH

Peptide S4.18

Synthesized according to general procedure A (solution phase).

OН

Peptide S4.19 Synthesized according to general procedure A (solution phase).

Peptide S4.20

Synthesized according to general procedure A (solution phase).

OН

OН

Peptide S4.21

Synthesized according to general procedure A (solution phase).

Peptide S4.22 Synthesized according to general procedure A (solid phase).

Peptide S4.23

Synthesized according to general procedure A (solid phase).

Acyclic Product S4.24

Synthesized according to general procedure B and **G1** using **S4.1**.

Acyclic Product S4.25

Synthesized according to general procedure B and **G1** using **S4.2**.

Acyclic Product S4.26

Synthesized according to general procedure B and **G1** using **S4.3**.

Acyclic Product S4.27

Synthesized according to general procedure B and **G1** using **S4.4**.

Acyclic Product S4.28

Synthesized according to general procedure B and **G1** using **S4.5**.

Acyclic Product S4.29

Synthesized according to general procedure B and **G1** using **S4.6**.
 EXECUTE:

Acyclic Product S4.30

Synthesized according to general procedure B and **G1** using **S4.7**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.31

Synthesized according to general procedure B and **G1** using **S4.8**.
 EDUCALLY

Acyclic Product S4.32

Synthesized according to general procedure B and **G1** using **S4.9**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO₂

Acyclic Product S4.33

Synthesized according to general procedure B and **G1** using **S4.10**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO₂

Acyclic Product S4.34

Synthesized according to general procedure B and **G1** using **S4.11**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO₂

Acyclic Product S4.35

Synthesized according to general procedure B and **G1** using **S4.12**
 $\text{IBuo}_2\text{CO}_\text{C}$ HO₂ HO

Acyclic Product S4.36

Synthesized according to general procedure B and **G1** using **S4.13**.

Acyclic Product S4.37

Synthesized according to general procedure B and **G1** using **S4.14**.
 $\text{IBuo}_2\text{CO}_\text{S}$ HO₂

Acyclic Product S4.38

Synthesized according to general procedure B and **G1** using **S4.15**
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.39

Synthesized according to general procedure B and **G1** using **S4.16**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.40

Synthesized according to general procedure B and **G1** using **S4.17**.

Synthesized according to general procedure B and **G1** using **S4.18**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.42

Synthesized according to general procedure C and **G2** using **S4.18** (54%, over 2 steps).

IBuO₂CO₂ HO₂

Acyclic Product S4.43

Synthesized according to general procedure C and **G3** using **S4.18** (60%, over 2 steps).

Synthesized according to general procedure B and **G1** using **S4.19**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.45

Synthesized according to general procedure C and **G2** using **S4.19** (45%, over 2 steps).

IBuO₂CO₂ HO₂

Acyclic Product S4.46

Synthesized according to general procedure C and **G3** using **S4.19** (54%, over 2 steps).

Synthesized according to general procedure B and **G1** using **S4.20**.
 $\text{IBu0}_2\text{CO}_\star$ HO₂

Acyclic Product S4.48

Synthesized according to general procedure C and **G2** using **S4.20** (34%, over 2 steps).

IBuO₂CO₂ HO₂

Acyclic Product S4.49

Synthesized according to general procedure C and **G3** using **S4.20** (38%, over 2 steps).

Synthesized according to general procedure B and **G1** using **S4.21**.
 $\text{IBuo}_2\text{CO}_\text{C}$ HO

Acyclic Product S4.51

Synthesized according to general procedure C and **G2** using **S4.21** (75%, over 2 steps).

IBuO₂CO₂ HO₂ HO

Acyclic Product S4.52

Synthesized according to general procedure C.

Synthesized according to general procedure B and **G1** using **S4.22**.

Acyclic Product S4.54

Synthesized according to general procedure B and **G2** using **S4.23**.

Macrocycle 4.6 Synthesized according to general procedure **E** using **S4.24**.

Macrocycle 4.7

Synthesized according to general procedure **E** using **S4.25**.

Macrocycle 4.8

Synthesized according to general procedure **E** using **S4.26**.

Macrocycle 4.9

Synthesized according to general procedure **E** using **S4.27**.

Macrocycle 4.10

Synthesized according to general procedure **E** using **S4.28**.

Macrocycle 4.11 Synthesized according to general procedure **D** using **S4.24**

Macrocycle 4.12

Synthesized according to general procedure **D** using **S4.25**.

Macrocycle 4.13

Synthesized according to general procedure **D** using **S4.26**.

Macrocycle 4.14

Synthesized according to general procedure **D** using **S4.27**

Macrocycle 4.15 Synthesized according to general procedure **D** using **S4.28**.

Macrocycle 4.16

Synthesized according to general procedure **D S4.29**.

Macrocycle 4.17

Synthesized according to general procedure **D** using **S4.30**.

Macrocycle 4.18

Synthesized according to general procedure **D** using **S4.31**.

Macrocycle 4.19

Synthesized according to general procedure **D** using macrocycle **4.8**.

Macrocycle 4.20

Synthesized according to general procedure **D** using **S4.32**.

Macrocycle 4.21

Synthesized according to general procedure **D** using **S4.33**.

Macrocycle 4.22

Synthesized according to general procedure **D** using **S4.34**.

Peparative HPLC method Column: Waters SunfireTM C₁₈, 19x250mm, 5 μ m *Solvent A*: H2O + 0.1%v TFA *Solvent B*: ACN + 0.1%v TFA Flow rate: 18.00 ml/min

Macrocycle 4.23

Synthesized according to general procedure **D** using **S4.35**.

Macrocycle 4.24

Macrocycle 4.25 Synthesized according to general procedure **D** using **S4.36**.

Peparative HPLC method Column: Waters Sunfire[™] C₁₈, 19x250mm, 5µm *Solvent A*: H2O + 0.1%v TFA *Solvent B*: ACN + 0.1%v TFA Flow rate: 18.00 ml/min

Macrocycle 4.26

Synthesized according to general procedure **D** using **S4.37**.

Macrocycle 4.27

Synthesized according to general procedure **D** using **S4.38**.

Macrocycle 4.28

Synthesized according to general procedure **D** using **S4.39**.

Macrocycle 4.29

Synthesized according to general procedure **D** using **S4.40**.

Macrocycle 4.30

Synthesized according to general procedure **D** using **S4.41** (26%, over 2 steps).

Macrocycle 4.31

Synthesized according to general procedure **D** using **S4.42** (55% yield).

Macrocycle 4.32

Synthesized according to general procedure **D** using using **S4.43** (30% yield).

Macrocycle 4.33

Synthesized according to general procedure **D** using **S4.44** (25%, over 2 steps).

Macrocycle 4.34

Synthesized according to general procedure **D** using **S4.45** (32% yield).

Macrocycle 4.35

Synthesized according to general procedure **D** using **S4.46** (28% yield).

SB-I-38 (65-75%)

Macrocycle 4.36

Synthesized according to general procedure **D** using **S4.47** (40%, over 2 steps).

Macrocycle 4.37

Synthesized according to general procedure **D** using **S4.48** (18% yield).

Macrocycle 4.38

Synthesized according to general procedure **D** using **S4.49** (11% yield).

Macrocycle 4.39

Synthesized according to general procedure **D** using **S4.50** (21%, over 2 steps).

Macrocycle 4.40

Synthesized according to general procedure **D** using **S4.51** (24% yield).

Macrocycle 4.41

Synthesized according to general procedure **D** using **S4.52** (15% yield).

Macrocycle 4.42

Macrocycle 4.43

Synthesized according to general procedure **E** using **S4.53**.

Macrocycle 4.44

Synthesized according to general procedure **E** using **S4.54**.

NMR Spectra

Macrocycle **4.30** (600 MHz, DMSO-d6, 298K)

COSY

Macrocycle **4.33** (600MHz, DMSO-d6, 298K)

HMBC

TOCSY

Macrocycle **4.35** (600MHz, DMSO-d6, 298K)

HMBC

COSY

TOCSY

NOESY

