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Atomic-Scale Imaging of Polymers and Precision Molecular Weight Analysis

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Supporting Information

ABSTRACT: Polymer design requires fine control over syntheses and a thorough understanding of their macromolecular structure. Herein, near-atomic level imaging of polymers is achieved, enabling the precise determination of one of the most important macromolecular characteristics: molecular weight. By judiciously designing and synthesizing different linear metal(loid)-rich homopolymers, subnanoscale polymer imaging is achieved through annular dark field-scanning transmission electron microscopy (ADF-STEM), owing to the incorporation of high *Z* atoms in the side chain of the monomeric units. The molecular weight of these polymers can be precisely determined by detecting and counting their metal(loid) atoms upon ADF-STEM imaging, at sample concentrations as low as 10 $\mu\text{g}\cdot\text{mL}^{-1}$. Notably, a commonly used C, H, and O-containing polymer (*i.e.*, poly(methyl acrylate)) that was thus far inaccessible at the atomic scale is derivatized to allow for subnano-level imaging, thus expanding the scope of our approach toward the atomic-level visualization of commodity polymers.

The design of soft matter with predefined properties necessitates the (sub)nanoscale analysis of polymers, tailored with precision for significant performance.^{1,2} A fundamental characteristic of polymers is their molecular weight (MW). The leading techniques for MW determination are size exclusion chromatography (SEC), ¹H nuclear magnetic resonance (¹H NMR), and high-resolution mass spectrometry (HR-MS) (Scheme 1). Although well-established, these techniques possess significant limitations when complex systems are targeted, including organometallic^{3,4} and conjugated polymers,^{5,6} or complex architectures.^{7,8} For instance, SEC requires the combination of suitable solvents, columns and MW standards,⁹ ¹H NMR requires distinctive end-groups, while topologically complex polymers, with high dispersity and MW, are not suitable for HR-MS.^{10,11}

Acknowledging those challenges, Junkers and colleagues developed a universal approach to determine polymer MW through diffusion-ordered NMR spectroscopy (DOSY), overcoming calibration and solvent implications,^{12–14} while Haddleton and Lester developed a facile strategy for MW online monitoring through DOSY.^{15,16} However, for polymers with compositional complexity and aggregation behavior in most solvents, solid-state MW analysis is necessary. Costantini and colleagues reported on the MW determination of conjugated polymers, through scanning tunnelling microscopy (STM) and vacuum electrospray deposition (ESD),⁶ while another powerful example is the work from Matyjaszewski and Sheiko who achieved in-depth analysis of high MW cylindrical brushes using atomic force microscopy (AFM).^{17–19} Although STM and AFM have provided valuable insights into the understanding of macromolecular characteristics, they are

limited to the imaging of conjugated polymers or polymers with very high MW and branching.

Within the scope of visual understanding of polymers, electron microscopy (EM) techniques including (cryogenic) transmission electron microscopy, (Cryo-) TEM, and liquid-cell electron microscopy (LC-EM) have revolutionized the field of polymer imaging. Exemplary are the studies from Patterson,^{1,20–26} Gianneschi,^{27–33} Sommerdijk,^{34,35} and de Jonge,^{36–40} among others. However, the subnano level imaging of nonconjugated synthetic polymers has been largely inaccessible. Apart from their structural complexity, their elemental composition is mostly limited to C, O, H, and N, exhibiting similarity with most TEM support grids. Consequently, the low contrast obtained during conventional TEM does not allow for precise subnano level imaging. Additionally, their light element composition renders them challenging to detect through atomic-level EM methods, such as annular dark field (ADF) scanning transmission electron microscopy (STEM), where contrast depends on the atomic number.^{1,41–44}

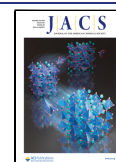
Our vision was to overcome those challenges and approach atomic-level analysis of polymers through ADF-STEM, as well as to visualize their MW (Scheme 1), by strategically designing the synthesis of metalloid-rich homopolymers bearing one arsenic (As) atom per monomer unit. For that purpose, free

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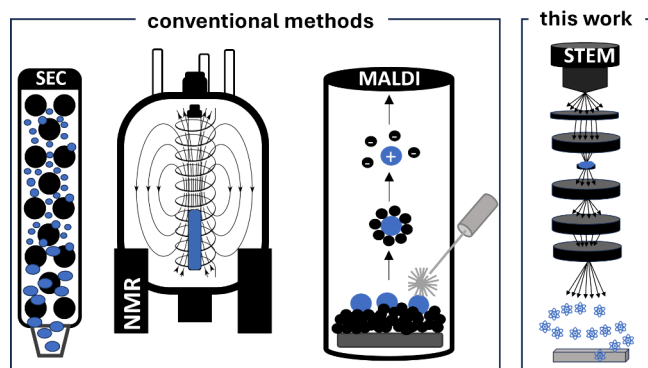
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Scheme 1. Schematic Illustration of the Methods Traditionally Used to Determine the MW of Polymers and Our Approach through ADF-STEM

Molecular Weight of Polymers



radical polymerization (FRP) and reversible deactivation radical polymerization (*i.e.*, reversible addition–fragmentation chain transfer polymerization, RAFT) were employed to generate polymers with various MW and \bar{D} values. To expand to another polymer family and metal functionality, ferrocenylmethyl methacrylate was used to generate an Fe-rich poly(methyl methacrylate). Finally, to render widely used polymers visible on the atomic level, a poly(methyl acrylate) (PMA) was derivatized with ferrocenecarboxylic acid, and its MW and \bar{D} were calculated through imaging.

Initially, an As-acrylamide monomer was synthesized according to the literature^{45,46} and used to generate an As-polyacrylamide (PAsAm) through FRP (PAsAm_{FRP}, Figure S1, SI). A highly dilute (50 $\mu\text{g}\cdot\text{mL}^{-1}$ in 0.1 M NaOH) solution of the purified homopolymer was prepared and placed under vacuum prior to imaging (SI). To gain a first understanding of the As signal, we employed ADF-STEM through a double aberration corrected JEOL ARM200F microscope, operated at 200 kV. At 3 million times magnification ($\times 3\text{M}$), bright nanoclusters were evident (Figure S2), while at $\times 8\text{M}$ and $\times 12\text{M}$ magnification, their structure was elucidated, depicting the randomly coiled polymer chains consisting of As atoms (appearing as bright spots, Figure S2). To enhance sample stability and mitigate contamination, “beam shower” was applied prior to imaging at high magnifications.⁴⁷ Although the organic content is sensitive and prone to beam damage under the applied conditions,⁴⁶ the metalloid-rich chains remained intact throughout imaging. An advantage of this approach is

that any potential damage to the organic components of the polymers by the electron beam will not affect the results of the MW analysis, as they depend only on the beam-stable metals.

Having achieved the detection of single chains and their As atoms, we sought to visualize the MW distribution of the polymers. Three well-defined PAsAms with targeted $DP_n = 50, 20,$ and 10 were synthesized through RAFT polymerization (SI, Figures S3–S5), while aqueous-SEC and DOSY NMR were employed to determine the MW of the homopolymers after purification (Table 1). The ADF-STEM of PAsAm₅₀,

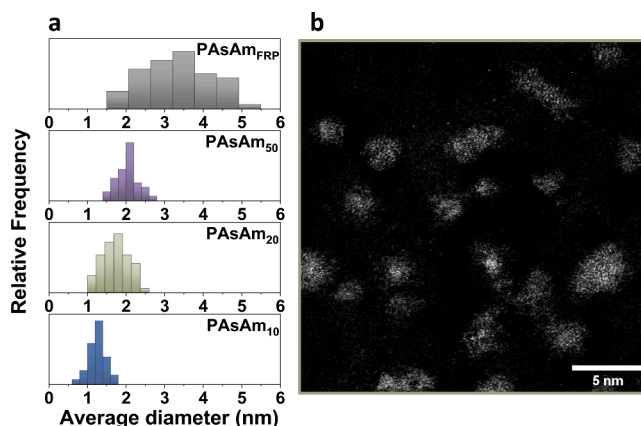


Figure 1. (a) Histograms showing the distribution of chain diameter for the four polymers and (b) high-resolution ADF-STEM image of PAsAm₂₀ (scale bar: 5 nm).

PAsAm₂₀ (Figure 1b), and PAsAm₁₀ at $\times 8\text{M}$ magnification revealed polymer chains smaller than in the case of PAsAm_{FRP} (Figures 1a, 2c–f, and S6–S8). Owing to the different average chain length of the imaged polymers, the chain diameter increased with the increase in MW (Figure 1a), while the low \bar{D} polymers exhibited narrow diameter distribution, compared to PAsAm_{FRP}. Importantly, when the As-monomer was imaged under the same EM conditions, only individually scattered single As atoms were detected (Figure S9).

To determine the polymer MW and \bar{D} , the intensities of single chains were calculated upon subtraction of their background, while the intensity of single As atoms was used as the calibrant, assuming a linear relationship between the integrated ADF intensity of single atoms and very small nanoclusters when kinematic diffraction effects dominate the signals collected by ADF-STEM imaging (Figure S10).^{48–50} The same process was repeated for each sample individually in the same session. The integrated As atoms’ intensity allowed

Table 1. Molecular Weight and \bar{D} Values from SEC, DOSY, and STEM for the Different Metal(loid)-rich Homopolymers

Polymer	$M_{n,SEC}^a$	$DP_{n,SEC}^a$	$M_{w,SEC}^a$	\bar{D}_{SEC}^a	MW_{DOSY}^b	$DP_{n,DOSY}^b$	MW_{STEM}^c	$DP_{n,STEM}^c$	$\bar{D}_{STEM}^{c,d}$
PAsAm ₁₀	5,700	21	6,300	1.10	2,500	9	2,300	8	1.30
PAsAm ₂₀	7,900	29	8,700	1.10	6,900	25	7,500	28	1.20
PAsAm ₅₀	12,100	45	13,900	1.15	9,600	35	11,000	40	1.10
PAsAm _{FRP}	218,000	N/A	509,000	2.30	101,700	375	113,000 ^h	417 ^h	1.60
PFeMMA ₁₀	5,900 ^e	21 ^e	8,600 ^e	1.4 ^{ef}	3,700 ^g	14 ^g	3,500	13	1.40
	7,100 ^f	25 ^f	10,200 ^f						

^aAqueous-SEC, average molecular weight values expressed as MW equivalents relative to PEG/PEO standards, ^bin D₂O/NaOH using an 80 MHz benchtop NMR, calculated through MaDDOSY,¹⁵ ^c MW_{STEM} expressed in $\text{g}\cdot\text{mol}^{-1}$, conditions: 200 kV at $\times 8\text{M}$ magnification (and $\times 10\text{M}$ for PAsAm₁₀ and PFeMMA₁₀), ^dcalculated based on the literature,⁵¹ ^e CHCl_3 -SEC, average molecular weight values expressed as MW equivalents to PS or ^fPMMA standards, ^gin CDCl₃ using an 80 MHz benchtop NMR, ^haverage of the broad main distribution from 13,000 to 80,000 $\text{g}\cdot\text{mol}^{-1}$ ($DP_n = 48–295$) and chains reaching up to 240,000 $\text{g}\cdot\text{mol}^{-1}$ ($DP_n \sim 885$).

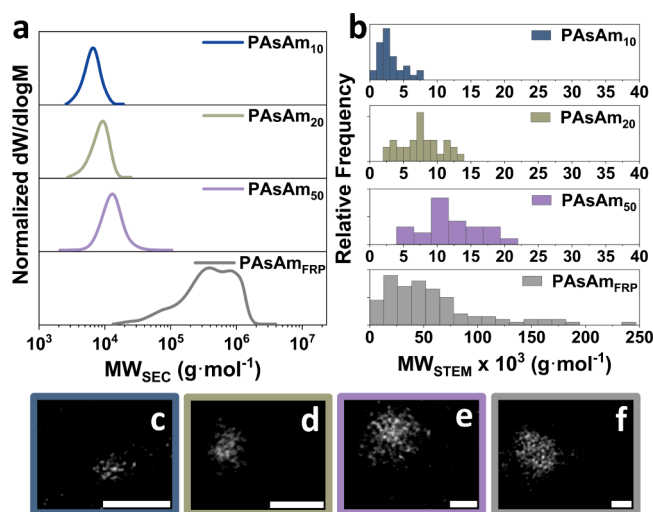


Figure 2. (a) Aqueous-SEC traces for PAsAm₁₀, PAsAm₂₀, PAsAm₅₀ and PAsAm_{FRP}. (b) MW_{STEM} distributions for the different polymers and ADF-STEM images showing segmented individual polymer chains for (c) PAsAm₁₀, (d) PAsAm₂₀, (e) PAsAm₅₀, and (f) PAsAm_{FRP} (scale bars: 1 nm, images were smoothed post imaging).

for determination of the DP_n for PAsAm₅₀, PAsAm₂₀, and PAsAm₁₀ through atom counting (Table 1, SI).

In other words, the number of As atoms in each chain corresponded to DP_n , which was used to calculate the corresponding MW_{STEM}. The D_{STEM} of the polymers was estimated according to the literature, based on standard deviation (σ), and the relation between D and σ (SI).⁵¹

For the three well-defined polymers, MW_{STEM} was comparable with MW_{SEC} while there was particularly good agreement between STEM and DOSY for PAsAm₁₀ and PAsAm₂₀ (Table 1; Figures 2a–b, S4, S5, S7, and S8). Importantly, in contrast with DOSY, STEM can provide a distribution of MW, representative of the nonidentical chain lengths in a synthetic polymer sample. For PAsAm₁₀, $M_{n,SEC}$ was significantly higher than MW_{DOSY} because low MWs necessitate better separation for higher accuracy. The D_{STEM} results for PAsAm₂₀ and PAsAm₅₀ exhibited proximity to D_{SEC} with both approaches resulting in $D_{STEM} \leq 1.2$.

To push the limits of our system, we attempted to calculate the MW_{STEM} of PAsAm_{FRP}. As expected, the STEM results showed a nonsymmetrical distribution of As atoms per chain, with a predominant broad MW_{STEM} distribution from 13,000 to ~80,000 g·mol⁻¹, along with the presence of high MW species up to ~240,000 g·mol⁻¹, with average MW_{STEM} = 113,000 g·mol⁻¹ (Figure 2b, Table 1). The D_{STEM} was 1.60, and although lower than the corresponding D_{SEC} , it illustrated the broad MW distribution of PAsAm_{FRP}. Samples with such high DP_n heterogeneity are highly challenging to quantitatively analyze with accuracy from single ADF-STEM images, since the very high MW chains might exhibit similarities with aggregated species; thus, careful interpretation of the images is necessary.^{44,52} In general, highly pure polymer samples, careful sample preparation (*i.e.*, suitable support grids),^{53,54} and thorough pretreatment (*i.e.*, vacuum drying, beam shower,^{47,55} SI) are essential requirements, especially when sensitive samples are used.⁵³

To expand the scope of metal functionality and polymer type, we synthesized an Fe-rich polymethacrylate (PFerMMA₁₀) (Figures 3a and S11, SI). As in the case of PAsAm, the

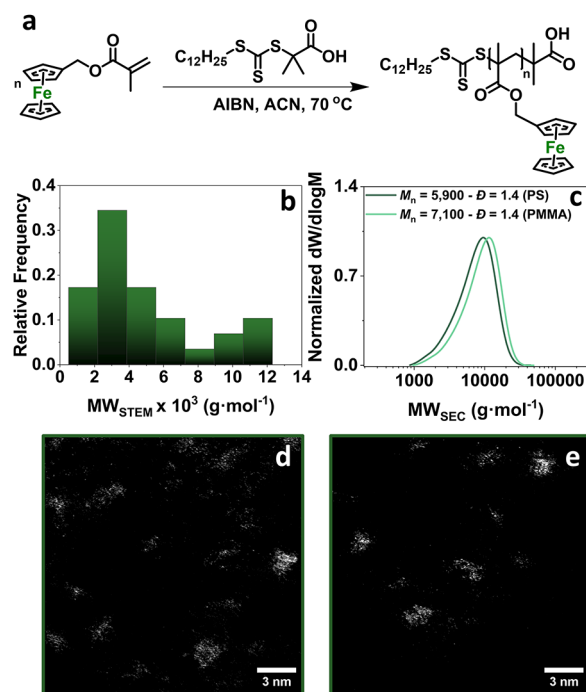


Figure 3. (a) Reaction scheme for the synthesis of PFerMMA₁₀, (b) MW_{STEM} histogram for PFerMMA₁₀, (c) SEC traces of PFerMMA₁₀ and (d, e) ADF-STEM images for PFerMMA₁₀ (scale bar: 3 nm).

MW_{STEM} (Figure 3b,d,e) was comparable to MW_{DOSY} while lower than $M_{n,SEC}$ (Table 1, Figure 3b,c). The SEC analysis of PFerMMA₁₀ exhibited distinct deviations when PMMA and PS standards were used (Figure 3c), highlighting the limitations of SEC when samples deviate from the calibrant. The range of MW_{STEM} (~1,000–12,000 g·mol⁻¹) with the existence of a second smaller population with MW ~9,500–12,000 g·mol⁻¹, was depicted in the obtained $D_{STEM} = 1.40$ (Table 1). Therefore, the calculation of MW_{STEM} could be successfully achieved both for metallo- and metal-containing acrylamide and methacrylate homopolymers, while their DP_n heterogeneity could be estimated through D_{STEM} calculation.

Finally, we were interested in applying our approach to widely used C-, H-, and O-containing polymers, without using specially designed monomers. Thus, a PMA₂₀ was synthesized (Figures 4a and S12, SI) and subsequently amidated using 4-amino-1-butanol, according to a literature procedure.⁵⁶ The amidation of PMA₂₀ to poly(hydroxybutyl acrylamide, PHBA_m) was quantitative, with a full shift of the PMA methyl protons as verified by ¹H NMR (Figure S14d), and full shift of the 1730 cm⁻¹ peak (C=O, PMA) along with the formation of the 1635 cm⁻¹ (C=O, amide) and 1543 cm⁻¹ (N–H) PHBA_m peaks, as verified by FT-IR (Figure S13). The shift toward higher MW was verified by THF-SEC (Figure 4b). The obtained –OH functional polymer was further functionalized through DCC/DMAP coupling with ferrocene (Fc) carboxylic acid (Figure 4a, SI), leading to derivatization of the parent PMA into an Fe-containing polyacrylamide. THF-SEC showed a clear shift toward higher MW (Figures 4b and S14); FT-IR verified the appearance of the C=O band (1700 cm⁻¹) attributed to the Fc-ester (Figure S13), while ¹H NMR confirmed the incorporation of the Fc moieties in the polymer (Figure S14d). ADF-STEM (Figure S15) revealed a predominant MW_{STEM} distribution at 3,000–4,300 g·mol⁻¹

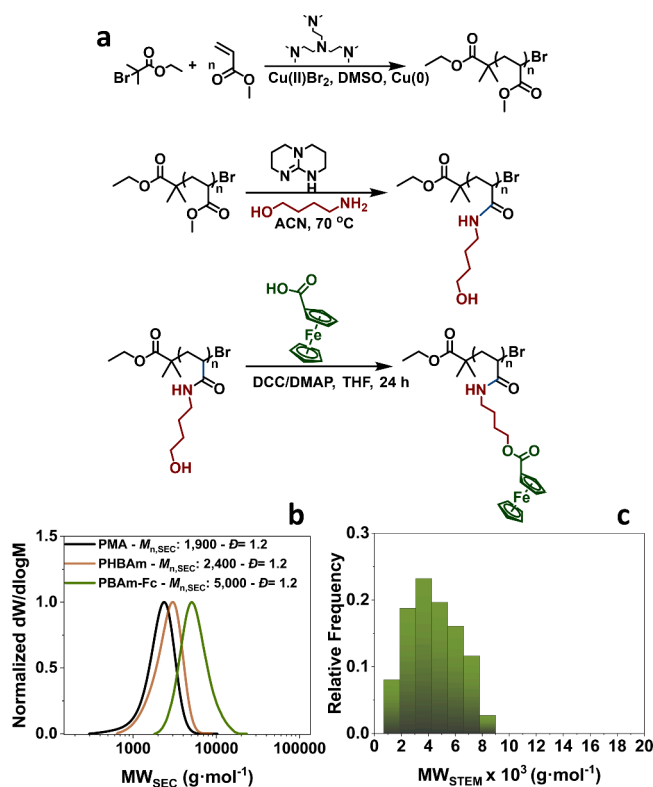


Figure 4. (a) Reaction scheme for the derivatization of PMA. (b) SEC traces of the parent PMA, the amidated derivative PHBAm, and the final PBAm-Fc, and (c) MW_{STEM} histogram of the functional PBAm-Fc.

(vs $M_{n,SEC} = 5,000$, Figure 4b,c), indicating that on average ~ 11 monomer units per chain had been functionalized (vs ~ 14 from SEC). Therefore, the achievement of near-atomic level imaging of a commonly used polymer through derivatization has critical potential to serve as a promising strategy to visualize materials that had thus far been unobtainable. To the best of our knowledge, this is the first example of near-atomic level imaging of such a widely used polymer. Owing to the various synthetic tools available, we envisage that the modification of other commonly used polymers (*i.e.*, polystyrene, polyolefins) through different derivatization approaches (*i.e.*, Diels–Alder, click chemistry)^{57–64} will expand the scope of this approach and establish it as a platform for advanced polymer imaging.

Our work presents the first approach toward atomic level imaging of synthetic polymers and MW determination through atom counting. By combining metal(loid)-containing monomers, different polymerization approaches, and atom counting through ADF-STEM, fundamental polymer characteristics were determined in the subnano scale. Additionally, the subnano level imaging of a widely used polymer (*i.e.*, PMA) was achieved upon derivatization. Our combinatorial approach sets the ground for atomic level analysis of polymer fundamentals that could not be imaged with such precision before and facilitates the profound understanding of their structure–property relationships.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.4c13812>.

Materials, instrumentation details, methods, synthetic procedures, NMR, SEC, FT-IR, UV–vis spectra, and ADF-STEM images (PDF)

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Notes

The authors declare no competing financial interest.

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