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BIS [TRIS (CYCLOPENTADIENYL)URANIUM (IV) [1,2-BIS (DIMETHYLDIPHOSPHINO)ETHANE] [(C5H5)3U]2. (CH3)2PCH2CH2P(CH3)2

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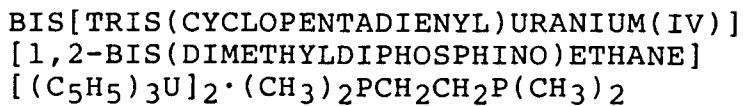
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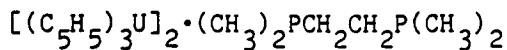


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Bis[tris(cyclopentadienyl)uranium(IV)][1,2-bis(dimethyldiphosphino)ethane]

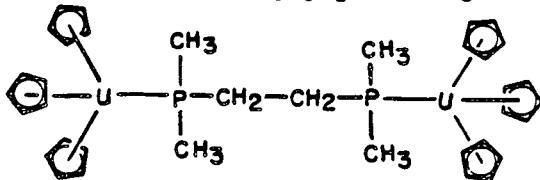


By Allan Zalkin, John G. Brennan & Richard A. Andersen

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Abstract. $[(C_5H_5)_3U]_2 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$, $M_r = 1016.78$, triclinic, $\bar{P\bar{1}}$, $a = 14.467(8)$, $b = 14.819(6)$, $c = 8.378(3)$ Å, $\alpha = 102.72(3)^\circ$, $\beta = 90.03(3)^\circ$, $\gamma = 96.69(3)^\circ$, $V = 1739.5 \text{ \AA}^3$, $Z = 2$, $D_x = 1.941 \text{ g cm}^{-3}$, $\lambda(\text{MoK}\alpha) = 0.71073 \text{ \AA}$, $\mu = 89.3 \text{ cm}^{-1}$, $F(000) = 952$, $T = 296 \text{ K}$, $R = 0.035$ for 1987 unique reflections with $F^2 > 2\sigma(F^2)$ of 3241 total unique data. The phosphorus atom at each end of the bis(dimethylphosphino)ethane ligand is coordinated to a trivalent uranium atom of a triscyclopentadienyluranium fragment; a center of inversion is located at the midpoint of the CH_2CH_2 bridge of the $(CH_3)_2PCH_2CH_2P(CH_3)_2$ ligand. Distances (Å) are: ave U-C(Cp) 2.77 ± 0.03 ; ave U-P 3.022 ± 0.002 ; ave U-Cp(ring) 2.518 ± 0.011 Å.

Introduction. The title complex, $[(C_5H_5)_3U]_2 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$,



was prepared during the course of studies on the coordination chemistry of tetra- and tri-valent uranium complexes with bidentate amine, phosphine, and arsine ligands. The crystal structures of the tetravalent complexes with bidentate ligands, $UCl_4[(CH_3)_2NCH_2CH_2N(CH_3)_2]_2$ (Zalkin, Edwards, Zhang & Andersen, 1986), $U(OC_6H_5)_4[(CH_3)_2PCH_2CH_2P(CH_3)_2]_2$ (Edwards, Andersen & Zalkin, 1981), $UCl_4[1,2-(CH_3)_2AsC_6H_4]_2$ (Edwards, Andersen & Gellert, 1986), and $U(CH_3)(CH_2C_6H_5)_3 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$ (Edwards, Andersen & Zalkin, 1984) have been described. In all of the above cases the ligands are bidentate, chelating ligands towards the uranium atom. In the title compound the $(CH_3)_2PCH_2CH_2P(CH_3)_2$ ligand is not acting as a bidentate, chelating ligand but it is acting as a bridging monodentate ligand towards each trivalent uranium atom.

Experimental. The compound was prepared from $(C_5H_5)_3U \cdot OC_4H_8$ and $(CH_3)_2PCH_2CH_2P(CH_3)_2$. Crystals suitable for X-ray diffraction studies were grown from a mixture of toluene and hexane (Brennan, 1985). Dark red, air sensitive crystals were sealed inside quartz capillaries under argon. Crystal 0.06 x 0.15 x 0.21 mm with 10 faces; modified Picker automatic diffractometer, graphite monochromator; cell dimensions from 49 reflections, $20^\circ < 2\theta < 36^\circ$; analytical absorption correction, range 1.75 to 4.04; max. $(sin\theta)/\lambda = 0.48 \text{ \AA}$; $h -13 \text{ to } 13, k -13 \text{ to } 13, l -8 \text{ to } 8$; three standard reflections, average decay 3%, intensities adjusted accordingly; 6454 data, 3242 unique, $R_{int} = 0.05$; structure solved by Patterson and Fourier methods; refined on F, f' & f'' terms included, 355 parameters;

calculated positional coordinates of the cyclopentadienyl hydrogen atoms with isotropic thermal parameters were included but not refined; anisotropic thermal parameters for non-hydrogen and non-disordered atoms; large thermal anisotropies and unsatisfactory distances in the dimethylphosphinoethane ligand suggested disorder; difference electron density maps were used to locate alternate atomic positions for the dimethylphosphinoethane carbon atoms; each of the carbon atoms in the ligand were placed in two positions and refined as half atoms with isotropic thermal parameters; distance restraints (Waser, 1963) imposed on disordered atoms, C-P 1.85 ± 0.02 , C-C 1.54 ± 0.02 , C-C(next neighbor) 3.03 ± 0.04 Å; $R = 0.035$ for 1987 reflections for which $F^2 > 2\sigma(F^2)$; $R = 0.086$ all data; $wR = 0.040$; $S = 1.0$; $w = 4F_0^2 / [\sigma^2(F_0^2)^2 + (0.06F_0^2)^2]$; max (shift/ σ) < 0.04; no extinction correction; max. and min. of ΔF synthesis 2.4 and -1.2 e Å; atomic f, f' & f" for neutral U, Cl, P, and C, and spherical bonded H from International Tables (1974); local unpublished programs and ORTEP (Johnson, 1965).

Atomic parameters are listed in Table 1,* and distances and angles are listed in Table 2. Fig.1 shows the molecule and numbering scheme.

* Lists of structure factors, anisotropic thermal parameters, calculated hydrogen positions, distances and angles, and least-squares planes have been deposited with the British Library Lending Division as Supplementary Publication No. (15 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Discussion. The structure consists of a $(C_5H_5)_3U$ fragment coordinated to the phosphorus atoms at each end of the $(CH_3)_2PCH_2CH_2P(CH_3)_2$ ligand. The

phosphine ligand bridges the two $(C_5H_5)_3U$ fragments so that $(CH_3)_2PCH_2CH_2P(CH_3)_2$ acts as a monodentate ligand towards each uranium center. There are two crystallographically independent, though chemically equivalent, molecules in the unit cell across the inversion centers at $(0,0,0)$ and $(1/2,1/2,0)$.

The geometry of the complex is unusual since $(CH_3)_2PCH_2CH_2P(CH_3)_2$ usually acts as a bidentate ligand towards a given metal center, rather than as a bridging ligand as found in the title compound. The reason for this structural change presumably is steric, since in a hypothetical molecule, $(C_5H_5)_3U \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$ in which the phosphine acts as a bidentate ligand, the coordination number of the uranium is one greater than found in the title compound. Inspection of the ORTEP diagram in Figure 1 clearly shows that the molecule is economically packed and that increasing the coordination number by one will require a substantial reorganization of the coordination sphere about uranium so that the intramolecular ligand-ligand repulsions are minimized. The coordination geometry about each uranium atom is similar to that found in $(CH_3C_5H_4)_3U \cdot P(CH_3)_3$ (Brennan & Zalkin, 1985), in which the U-P distance is $2.972(6)$ Å. In the title compound the averaged U-P distance is 3.022 ± 0.002 Å. The other bond parameters in these two molecules also are similar. The averaged U-C($CH_3C_5H_4$) distance in $(CH_3C_5H_4)_3U \cdot P(CH_3)_3$ is 2.79 ± 0.04 Å and the U-ring centroid distance is 2.52 ± 0.01 Å, the (ring centroid)-U-(ring centroid) angle is $112 \pm 7^\circ$ and the (ring centroid)-U-P angle is $106 \pm 9^\circ$. In the title complex, the averaged U-C(C_5H_5) distance is 2.77 ± 0.03 Å, the U-(ring centroid) distance is 2.52 ± 0.01 Å, the (ring centroid)-U-(ring centroid) angle is $118 \pm 1^\circ$ and the (ring centroid)-U-P angle is $98 \pm 1^\circ$. Given the large uncertainty in the bond parameters and the range in individual parameters, the

coordination geometries about the trivalent uranium atoms in $[(C_5H_5)_3U]_2 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$ and in $(CH_3C_5H_4)_3U \cdot P(CH_3)_3$ are identical or nearly so.

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Table 1. Atomic Parameters

$$B_{eq} = \sum_i \sum_j B_{ij} a_i^* a_j^* a_i \cdot a_j / 3, \text{ \AA}^2.$$

molecule 1

Atom	x	y	z	B/Beq
U(1)	-0.03723(5)	0.28914(6)	0.54961(9)	3.19
P(1)	-0.1054(4)	0.0982(4)	0.6049(7)	4.67
C(1)	0.1520(18)	0.350(3)	0.603(4)	7.85
C(2)	0.1105(21)	0.3966(17)	0.736(4)	5.76
C(3)	0.0786(16)	0.3304(28)	0.8264(27)	6.05
C(4)	0.1012(23)	0.2433(20)	0.741(4)	6.29
C(5)	0.1462(21)	0.2575(23)	0.605(4)	6.24
C(6)	0.0309(25)	0.288(4)	0.244(3)	8.27
C(7)	0.0261(27)	0.2003(28)	0.248(3)	5.95
C(8)	-0.064(3)	0.1709(20)	0.2382(27)	5.55
C(9)	-0.1162(19)	0.243(4)	0.2391(28)	7.40
C(10)	-0.054(4)	0.3199(23)	0.240(3)	6.76
C(11)	-0.153(3)	0.354(3)	0.801(4)	7.23
C(12)	-0.2067(27)	0.302(3)	0.698(7)	11.69
C(13)	-0.2138(29)	0.345(5)	0.568(5)	14.60
C(14)	-0.145(4)	0.426(3)	0.615(5)	10.00
C(15)	-0.1178(20)	0.4281(24)	0.757(6)	7.50
C(31)	-0.0226(23)	0.0081(24)	0.5841(23)	4.7(10)
C(32)	-0.2149(20)	0.041(3)	0.488(5)	5.0(12)
C(33)	-0.140(3)	0.108(3)	0.8241(26)	5.9(14)
C(31')	-0.0531(7)	-0.0060(17)	0.486(4)	3.7(9)
C(32')	-0.2332(14)	0.067(3)	0.545(5)	6.0(15)
C(33')	-0.1021(29)	0.0856(29)	0.8223(24)	3.8(10)

Table 1. Atomic Parameters (continued)

molecule 2

Atom	x	y	z	B/Beq
U(2)	0.45618(5)	0.21069(6)	0.02519(9)	3.34
P(2)	0.3991(4)	0.4051(4)	0.1139(7)	4.87
C(16)	0.5210(28)	0.1505(28)	-0.286(3)	7.48
C(17)	0.5171(22)	0.2443(24)	-0.2770(27)	5.75
C(18)	0.4277(25)	0.2602(20)	-0.2729(27)	4.97
C(19)	0.3734(19)	0.1765(23)	-0.287(3)	5.48
C(20)	0.4306(29)	0.1103(22)	-0.294(3)	6.87
C(21)	0.5700(23)	0.260(4)	0.300(4)	9.10
C(22)	0.5969(21)	0.176(4)	0.220(6)	9.20
C(23)	0.6414(23)	0.1925(27)	0.085(5)	8.48
C(24)	0.6404(16)	0.2826(26)	0.086(3)	5.60
C(25)	0.6014(23)	0.3267(23)	0.217(5)	7.04
C(26)	0.295(3)	0.2077(23)	0.205(8)	10.59
C(27)	0.2724(22)	0.150(4)	0.068(6)	9.00
C(28)	0.3096(27)	0.0762(27)	0.024(4)	7.30
C(29)	0.3649(23)	0.0727(27)	0.154(6)	8.25
C(30)	0.3524(28)	0.151(3)	0.2671(29)	8.25
C(34)	0.4905(24)	0.5015(23)	0.0905(14)	5.6(11)
C(35)	0.2916(20)	0.425(3)	0.008(5)	4.8(11)
C(36)	0.373(3)	0.433(4)	0.3377(25)	7.4(17)
C(34')	0.4473(7)	0.4971(24)	0.004(5)	5.7(12)
C(35')	0.2710(13)	0.4003(29)	0.061(5)	4.6(11)
C(36')	0.407(3)	0.465(3)	0.3361(24)	5.7(13)

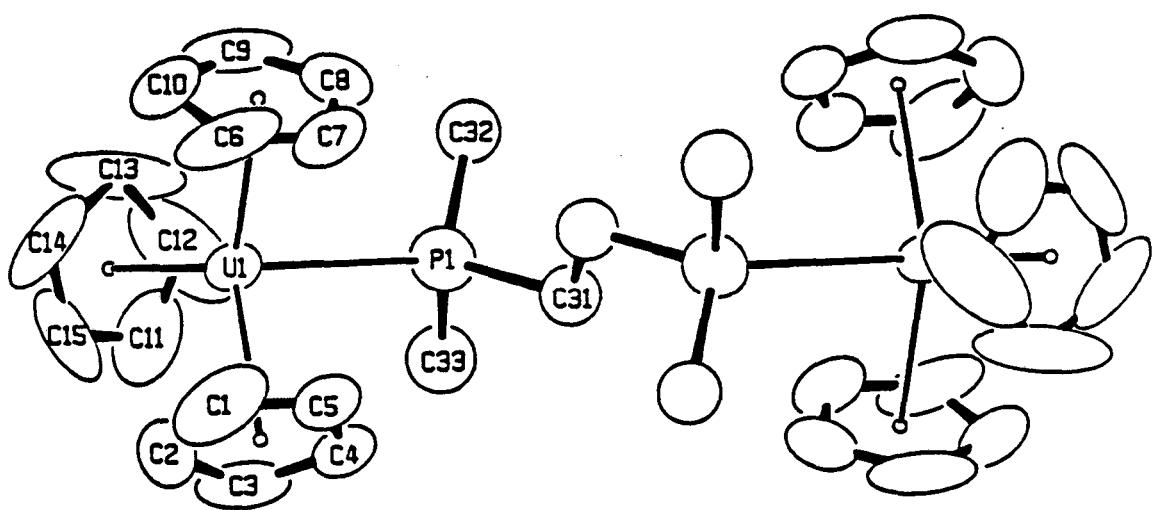
Primed atoms indicate the alternate disorder sites.

Table 2. Selected distances(Å) and angles(°)

U(1)	- P(1)	3.020(06)	U(2)	- P(2)	3.024(06)
U(1)	- C(1)	2.784(28)	U(2)	- C(16)	2.766(25)
U(1)	- C(2)	2.765(25)	U(2)	- C(17)	2.809(20)
U(1)	- C(3)	2.774(22)	U(2)	- C(18)	2.797(23)
U(1)	- C(4)	2.801(25)	U(2)	- C(19)	2.795(26)
U(1)	- C(5)	2.803(27)	U(2)	- C(20)	2.759(27)
U(1)	- C(6)	2.738(25)	U(2)	- C(21)	2.742(30)
U(1)	- C(7)	2.787(26)	U(2)	- C(22)	2.771(27)
U(1)	- C(8)	2.803(24)	U(2)	- C(23)	2.780(27)
U(1)	- C(9)	2.752(24)	U(2)	- C(24)	2.755(24)
U(1)	- C(10)	2.744(23)	U(2)	- C(25)	2.810(30)
U(1)	- C(11)	2.768(21)	U(2)	- C(26)	2.779(28)
U(1)	- C(12)	2.758(33)	U(2)	- C(27)	2.754(28)
U(1)	- C(13)	2.772(30)	U(2)	- C(28)	2.734(26)
U(1)	- C(14)	2.666(27)	U(2)	- C(29)	2.727(25)
U(1)	- C(15)	2.758(23)	U(2)	- C(30)	2.763(22)
U(1)	- Cp(1)	2.527	U(2)	- Cp(4)	2.533
U(1)	- Cp(2)	2.516	U(2)	- Cp(5)	2.523
U(1)	- Cp(3)	2.505	U(2)	- Cp(6)	2.506
Cp(1) - U(1) - Cp(2)	117.31	Cp(4) - U(2) - Cp(5)	119.42		
Cp(1) - U(1) - Cp(3)	117.61	Cp(4) - U(2) - Cp(6)	116.12		
Cp(2) - U(1) - Cp(3)	119.77	Cp(5) - U(2) - Cp(6)	119.49		

^a Cp(1), Cp(2), Cp(3), Cp(4), Cp(5) and Cp(6) represent the centroids of cyclopentadienyl atoms C(1)-C(5), C(6)-C(10), C(11)-C(15), C(16)-C(20), C(21)-C(25) and C(26)-C30 respectively.

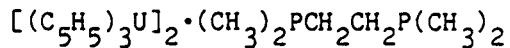
Fig.1. ORTEP (Johnson, 1965) drawing of molecule 1; 50% probability ellipsoids.



Supplemental Material

for

Bis[tris(cyclopentadienyl)uranium(IV)][1,2-bis(dimethyldiphosphino)ethane]

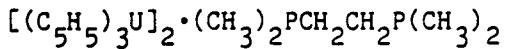


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Abstract. $[(C_5H_5)_3U]_2 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$, $M_r = 1016.78$, triclinic, $\bar{P}1$, $a = 14.467(8)$, $b = 14.819(6)$, $c = 8.378(3)$ Å, $\alpha = 102.72(3)^\circ$, $\beta = 90.03(3)^\circ$, $\gamma = 96.69(3)^\circ$, $V = 1739.5$ Å³, $Z = 2$, $D_x = 1.941$ g cm⁻³, $\lambda(\text{MoK}\alpha) = 0.71073$ Å, $\mu = 89.3$ cm⁻¹, $F(000) = 952$, $T = 296$ K, $R = 0.035$ for 1987 unique reflections with $F^2 > 2\sigma(F^2)$ of 3241 total unique data. The phosphorus atom at each end of the bis(dimethylphosphino)ethane ligand is coordinated to a trivalent uranium atom of a triscyclopentadienyluranium fragment; a center of inversion is located at the midpoint of the CH₂CH₂ bridge of the (CH₃)₂PCH₂CH₂P(CH₃)₂ ligand. Distances (Å) are: ave U-C(Cp) 2.77 ± 0.03; ave U-P 3.022 ± 0.002; ave U-Cp(ring) 2.518 ± 0.011 Å.

Supplemental Table 1. Anisotropic Thermal Parameters (\AA^2)^a



Atom	B_{11}	B_{22}	B_{33}	B_{12}	B_{13}	B_{23}
U(1)	2.33(5)	4.36(6)	3.05(4)	0.70(4)	0.31(4)	1.04(4)
U(2)	2.55(5)	4.84(6)	2.80(4)	0.54(4)	0.45(4)	1.14(4)
P(1)	3.8(3)	5.4(4)	4.7(3)	0.60(27)	1.20(25)	0.85(27)
P(2)	3.2(3)	5.7(4)	5.4(3)	0.51(27)	0.34(25)	0.62(27)
C(1)	3.2(14)	13.3(30)	9.0(23)	1.1(17)	0.9(13)	6.6(22)
C(2)	5.6(17)	3.5(14)	7.9(18)	-1.1(12)	-1.3(14)	1.5(14)
C(3)	2.5(12)	11.3(24)	3.9(13)	0.3(15)	-0.8(10)	1.1(16)
C(4)	7.6(20)	4.8(17)	6.8(18)	-1.5(14)	-4.1(15)	2.9(15)
C(5)	6.9(19)	6.2(19)	5.8(17)	1.8(15)	-2.4(14)	1.1(15)
C(6)	6.8(22)	15.7(35)	4.2(15)	1.7(24)	0.2(13)	6.1(20)
C(7)	6.6(22)	8.3(22)	3.7(13)	3.5(19)	0.0(13)	1.7(14)
C(8)	8.6(23)	5.5(18)	2.0(11)	0.4(18)	-0.1(13)	-0.1(10)
C(9)	4.2(17)	15.5(33)	2.0(11)	2.5(22)	-0.3(11)	0.1(17)
C(10)	11.3(28)	7.4(20)	2.9(12)	3.3(21)	-0.3(15)	2.6(13)
C(11)	8.1(24)	9.0(25)	6.4(18)	6.6(21)	5.2(18)	2.5(19)
C(12)	4.8(21)	10.9(28)	14.9(36)	-5.2(20)	2.0(22)	-3.8(29)
C(13)	4.8(21)	28.5(64)	7.0(22)	11.0(31)	-3.9(19)	-8.1(31)
C(14)	14.3(36)	14.0(34)	6.1(20)	11.9(30)	4.7(21)	6.3(22)
C(15)	5.4(17)	3.7(16)	11.1(27)	0.8(14)	2.4(18)	-3.4(17)
C(16)	9.5(25)	9.7(24)	6.3(16)	6.8(22)	3.7(16)	5.4(17)
C(17)	4.8(17)	10.2(24)	2.6(11)	-1.0(16)	0.5(10)	3.1(13)
C(18)	6.9(19)	4.9(16)	2.9(11)	1.0(15)	-0.4(11)	0.4(11)
C(19)	6.4(17)	5.8(17)	5.2(14)	2.2(17)	1.1(12)	2.6(13)
C(20)	8.4(23)	7.1(21)	4.5(14)	0.1(19)	1.7(15)	0.3(13)
C(21)	5.5(21)	17.7(40)	4.5(17)	4.1(25)	-1.3(14)	2.0(21)
C(22)	1.8(15)	16.2(37)	13.1(30)	-0.4(18)	-1.6(16)	11.5(30)
C(23)	5.7(19)	6.5(21)	12.4(29)	5.3(17)	-2.8(18)	-2.1(19)
C(24)	2.6(12)	10.5(24)	3.2(13)	2.0(14)	0.2(10)	-0.1(14)
C(25)	5.5(19)	9.3(24)	6.4(20)	1.3(17)	-2.2(15)	1.5(18)
C(26)	7.0(25)	3.4(18)	20.4(44)	1.7(18)	10.0(28)	-0.1(24)
C(27)	4.1(17)	6.9(26)	16.0(37)	-1.5(19)	2.7(21)	3.7(24)
C(28)	7.7(22)	6.1(20)	6.3(17)	-4.1(17)	0.6(16)	0.2(16)
C(29)	7.2(20)	8.2(23)	12.2(27)	2.2(17)	2.6(21)	7.8(22)
C(30)	9.5(25)	10.4(26)	2.6(12)	-5.1(21)	2.9(14)	-0.1(15)

^a The anisotropic temperature factor has the form:

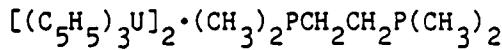
$$\exp[-0.25(B_{11}h^2a^*{}^2 + 2B_{12}hka^*b^* + \dots)].$$

Supplemental Table 2. Estimated Hydrogen Positional and Isotropic Thermal
Parameters (\AA^2)^a $[(\text{C}_5\text{H}_5)_3\text{U}]_2 \cdot (\text{CH}_3)_2\text{PCH}_2\text{CH}_2\text{P}(\text{CH}_3)_2$

Atom	x	y	z	B
H(1)	0.1791	0.3842	0.5249	10.0
H(2)	0.1021	0.4635	0.7733	10.0
H(3)	0.0459	0.3353	0.9297	10.0
H(4)	0.0894	0.181	0.7649	10.0
H(5)	0.1725	0.214	0.5165	10.0
H(6)	0.0927	0.325	0.2441	10.0
H(7)	0.0742	0.1619	0.2475	10.0
H(8)	-0.0924	0.106	0.2386	10.0
H(9)	-0.1828	0.2414	0.2310	10.0
H(10)	-0.0664	0.3840	0.2433	10.0
H(11)	-0.1426	0.3284	0.9036	10.0
H(12)	-0.2512	0.2438	0.6746	10.0
H(13)	-0.2444	0.345	0.4656	10.0
H(14)	-0.1170	0.4825	0.5674	10.0
H(15)	-0.073	0.4742	0.8458	10.0
H(16)	0.5799	0.1243	-0.2853	10.0
H(17)	0.5693	0.2944	-0.2706	10.0
H(18)	0.4037	0.3206	-0.2612	10.0
H(19)	0.3059	0.1643	-0.2926	10.0
H(20)	0.4158	0.0425	-0.3063	10.0
H(21)	0.5397	0.2634	0.4027	10.0
H(22)	0.5888	0.1138	0.2413	10.0
H(23)	0.6676	0.1514	-0.0072	10.0
H(24)	0.6669	0.3167	0.0014	10.0
H(25)	0.592	0.3945	0.2586	10.0
H(26)	0.2812	0.2624	0.2703	10.0
H(27)	0.2254	0.1767	-0.0065	10.0
H(28)	0.296	0.0302	-0.0824	10.0
H(29)	0.4003	0.0166	0.1516	10.0
H(30)	0.3855	0.1654	0.3782	10.0

^a The isotropic temperature factor has the form $\exp[-B(\sin\theta/\lambda)^2]$.

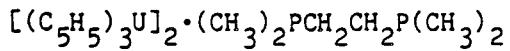
Supplemental Table 3. Additional Distances(Å)



P(1)	- C(33')	1.873(17)	P(2)	- C(36')	1.874(18)
P(1)	- C(31)	1.874(17)	P(2)	- C(35)	1.874(18)
P(1)	- C(32)	1.876(18)	P(2)	- C(34)	1.874(18)
P(1)	- C(31')	1.881(17)	P(2)	- C(36)	1.876(18)
P(1)	- C(33)	1.883(18)	P(2)	- C(34')	1.877(18)
P(1)	- C(32')	1.893(18)	P(2)	- C(35')	1.894(18)
C(1)	- C(2)	1.356(36)	C(12)	- C(13)	1.387(65)
C(1)	- C(5)	1.360(39)	C(13)	- C(14)	1.444(61)
C(2)	- C(1)	1.356(36)	C(14)	- C(15)	1.249(42)
C(2)	- C(3)	1.405(33)	C(16)	- C(20)	1.367(40)
C(3)	- C(4)	1.406(34)	C(16)	- C(17)	1.383(38)
C(4)	- C(5)	1.362(35)	C(17)	- C(18)	1.342(32)
C(6)	- C(7)	1.298(43)	C(23)	- C(24)	1.336(38)
C(6)	- C(10)	1.374(40)	C(24)	- C(25)	1.314(34)
C(7)	- C(8)	1.328(39)	C(26)	- C(27)	1.287(52)
C(8)	- C(9)	1.375(38)	C(26)	- C(30)	1.420(49)
C(10)	- C(10)	1.363(39)	C(27)	- C(28)	1.261(43)
C(11)	- C(12)	1.231(45)	C(28)	- C(29)	1.364(41)
C(11)	- C(15)	1.280(39)	C(29)	- C(30)	1.355(42)
C(31)	- C(31')	0.900(33)	C(34)	- C(34')	0.943(34)
C(31)	- C(31)	1.537(19)	C(34)	- C(34)	1.534(20)
C(31')	- C(31')	1.536(19)	C(34')	- C(34')	1.521(20)
C(32)	- C(32')	0.627(57)	C(35)	- C(35')	0.682(50)
C(33)	- C(33')	0.668(52)	C(36)	- C(36')	0.641(65)
H(1)	- C(1)	0.972(00)	H(16)	- C(16)	0.977(00)
H(2)	- C(2)	0.994(00)	H(17)	- C(17)	0.988(00)
H(3)	- C(3)	0.980(00)	H(18)	- C(18)	0.982(00)
H(4)	- C(4)	0.981(00)	H(19)	- C(19)	0.972(00)
H(5)	- C(5)	0.977(00)	H(20)	- C(20)	0.985(00)
H(6)	- C(6)	0.996(00)	H(21)	- C(21)	0.961(00)
H(7)	- C(7)	0.947(00)	H(22)	- C(22)	0.971(00)
H(8)	- C(8)	0.999(00)	H(23)	- C(23)	0.978(00)
H(9)	- C(9)	0.963(00)	H(24)	- C(24)	1.006(00)
H(10)	- C(10)	0.980(00)	H(25)	- C(25)	1.013(00)
H(11)	- C(11)	1.030(00)	H(26)	- C(26)	0.915(00)
H(12)	- C(12)	1.000(00)	H(27)	- C(27)	1.085(00)
H(13)	- C(13)	0.965(00)	H(28)	- C(28)	1.001(00)
H(14)	- C(14)	1.045(00)	H(29)	- C(29)	1.023(00)
H(15)	- C(15)	1.053(00)	H(30)	- C(30)	1.015(00)

Primed labels indicate alternate disorder atoms.

Supplemental Table 4. Additional Angles (°)



C(31)	-P(1)	-C(32)	107.1(16)	C(34)	-P(2)	-C(35)	105.7(15)
C(31)	-P(1)	-C(33)	100.8(14)	C(34)	-P(2)	-C(36)	104.0(16)
C(31')	-P(1)	-C(32')	102.2(14)	C(34')	-P(2)	-C(35')	99.1(13)
C(31')	-P(1)	-C(33')	104.5(14)	C(34')	-P(2)	-C(36')	104.8(16)
C(32)	-P(1)	-C(33)	102.8(16)	C(35)	-P(2)	-C(36)	105.1(16)
C(32')	-P(1)	-C(33')	102.9(15)	C(35')	-P(2)	-C(36')	102.8(15)
C(2)	-C(1)	-C(5)	110.3(25)	C(17)	-C(16)	-C(20)	106.0(26)
C(1)	-C(2)	-C(3)	106.2(25)	C(16)	-C(17)	-C(18)	108.9(27)
C(2)	-C(3)	-C(4)	107.8(22)	C(17)	-C(18)	-C(19)	108.0(25)
C(3)	-C(4)	-C(5)	106.8(24)	C(18)	-C(19)	-C(20)	107.8(27)
C(1)	-C(5)	-C(4)	108.9(27)	C(16)	-C(20)	-C(19)	109.3(28)
C(7)	-C(6)	-C(10)	114.2(34)	C(22)	-C(21)	-C(25)	108.9(31)
C(6)	-C(7)	-C(8)	104.5(30)	C(21)	-C(22)	-C(23)	106.0(33)
C(7)	-C(8)	-C(9)	111.2(31)	C(22)	-C(23)	-C(24)	107.5(30)
C(8)	-C(9)	-C(10)	106.5(28)	C(23)	-C(24)	-C(25)	111.8(29)
C(6)	-C(10)	-C(9)	103.6(28)	C(21)	-C(25)	-C(24)	105.6(32)
C(12)	-C(11)	-C(15)	113.5(41)	C(27)	-C(26)	-C(30)	96.8(28)
C(11)	-C(12)	-C(13)	107.5(44)	C(26)	-C(27)	-C(28)	121.4(45)
C(12)	-C(13)	-C(14)	101.8(27)	C(27)	-C(28)	-C(29)	105.2(33)
C(13)	-C(14)	-C(15)	106.8(36)	C(28)	-C(29)	-C(30)	104.1(29)
C(11)	-C(15)	-C(14)	109.9(37)	C(26)	-C(30)	-C(29)	112.0(32)
P(1)	-C(31)	-C(31)	111.6(17)	P(2)	-C(34)	-C(34)	111.1(18)
P(1)	-C(31')	-C(31')	111.1(17)	P(2)	-C(34')	-C(34')	111.2(18)

Supplemental Table 5. Least-squares Planes
 $[(C_5H_5)_2U]_2 \cdot (CH_3)_2PCH_2CH_2P(CH_3)_2$

Equation of plane No. 1

$$12.445 a - 0.625 b + 4.026 c = 4.093$$

Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(1)	0.0075(596)	
C(2)	-0.0045(457)	
C(3)	0.0054(522)	
C(4)	-0.0019(475)	
C(5)	-0.0012(452)	
not in the plane		
U(1)	-2.525(17)	

Equation of plane No. 4

$$-0.119 a - 1.071 b + 8.284 c = -2.604$$

Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(16)	-0.0103(525)	
C(17)	-0.0135(461)	
C(18)	0.0134(450)	
C(19)	-0.0081(442)	
C(20)	0.0010(523)	
not in the plane		
U(2)	2.533(17)	

Equation of plane No. 2

$$-0.539 a + 0.052 b + 8.165 c = 2.009$$

Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(6)	-0.0160(673)	
C(7)	0.0150(538)	
C(8)	-0.0202(608)	
C(9)	0.0187(719)	
C(10)	-0.0005(679)	
not in the plane		
U(1)	2.514(23)	

Equation of plane No. 5

$$12.380 a - 0.196 b + 4.001 c = 8.233$$

Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(21)	-0.0265(733)	
C(22)	0.0215(673)	
C(23)	0.0104(518)	
C(24)	-0.0174(499)	
C(25)	0.0183(520)	
not in the plane		
U(2)	2.514(23)	

Equation of plane No. 3:

$$-11.810 a + 7.481 b + 2.540 c = 6.495$$

Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(11)	-0.0049(641)	
C(12)	-0.0197(785)	
C(13)	0.0563(954)	
C(14)	-0.0314(684)	
C(15)	0.0215(625)	
not in the plane		
U(1)	-2.497(27)	

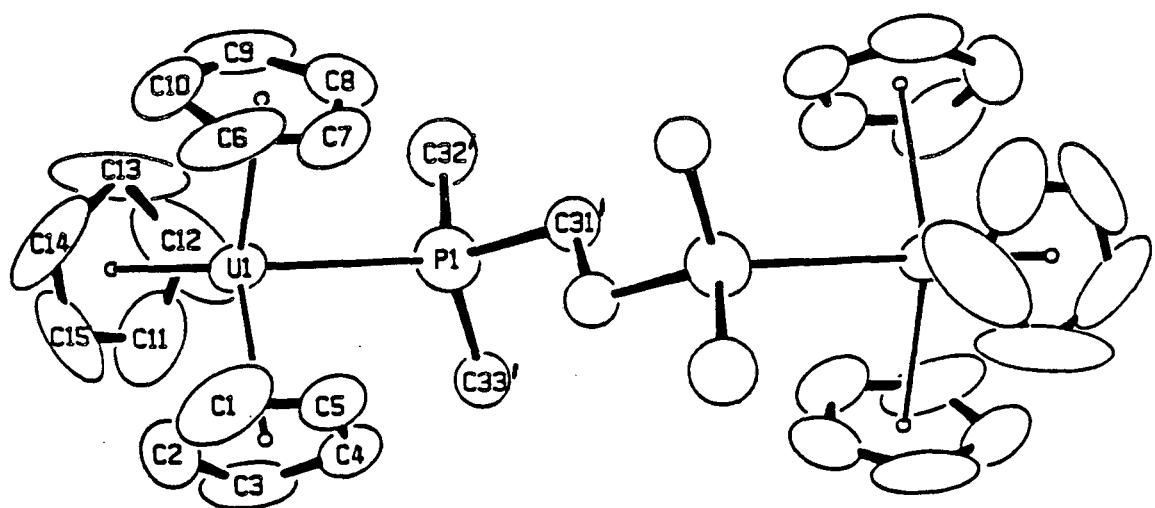
Equation of plane No. 6

$$10.439 a + 7.258 b - 4.193 c = 3.679$$

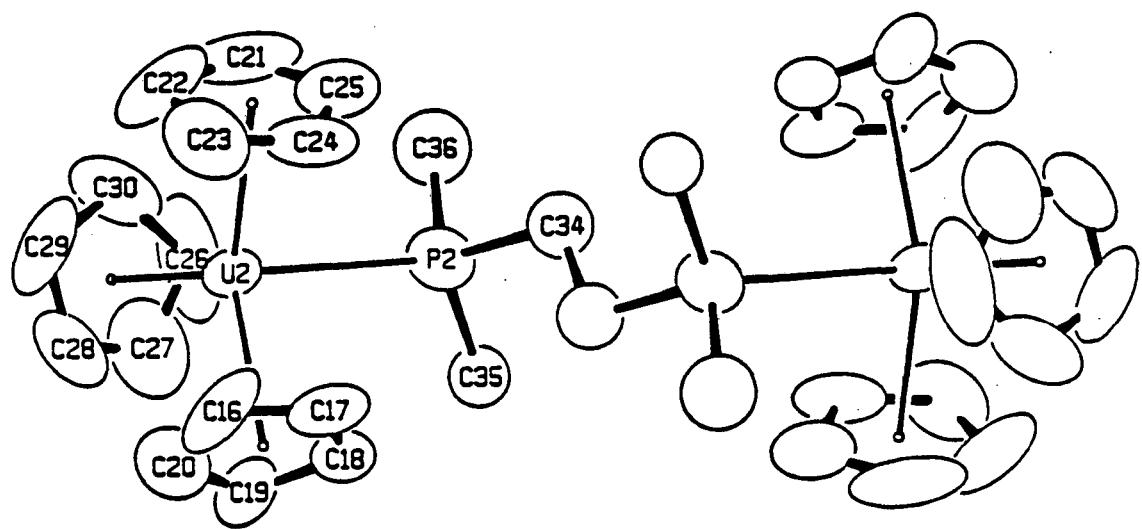
Distance to the plane from atoms
 in the plane

atom	d(Å)	$\sigma(d)$
C(26)	0.0484(804)	
C(27)	-0.0289(661)	
C(28)	0.0058(529)	
C(29)	0.0123(662)	
C(30)	-0.0246(662)	
not in the plane		
U(2)	2.507(23)	

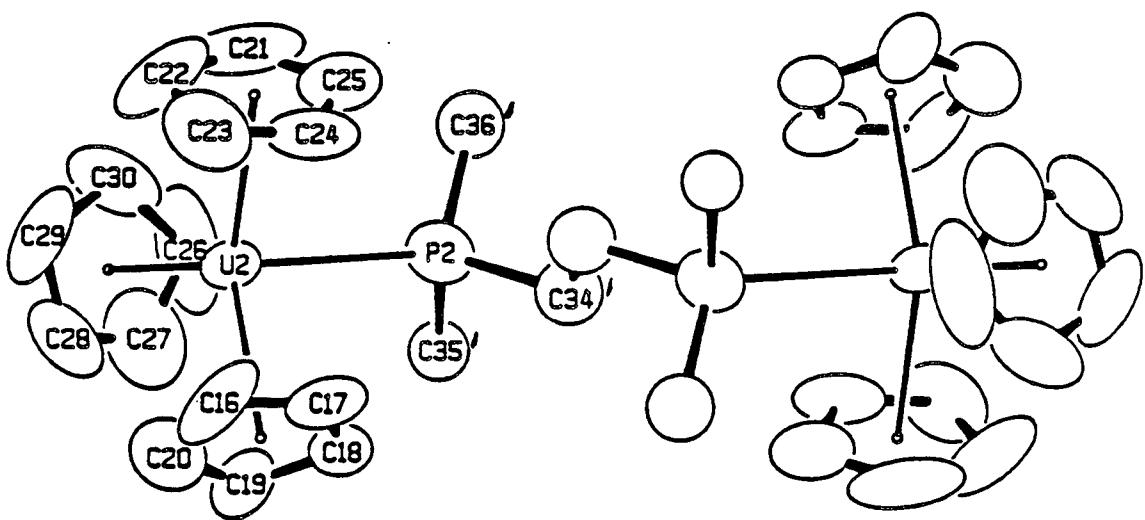
Supplemental Fig. 1. ORTEP drawing of molecule 1 using alternate disorder configuration (primed atoms); 50% probability ellipsoids.



Supplemental Fig. 2. ORTEP drawing of molecule 2; 50% probability ellipsoids.



Supplemental Fig. 3. ORTEP drawing of molecule 2 using alternate disorder configuration (primed atoms); 50% probability ellipsoids.



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