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ABSTRACT

Tables of Q-values for nuclear reactions have been calculated from the currently available mass data on the nuclides. The calculations were carried out on all stable and naturally occurring radioactive isotopes from a Z of 2 to 92, as well as on a few common long-lived isotopes. The tabulated reactions involve the following:

Isotope	Incoming Particles	Outgoing Particles
$5 \leq Z \leq 22$	γ , n, p, d, t, He^3 , He^4 , Li^6 , Li^7 and C^{12}	γ , n, p, d, t, He^3 , He^4 , He^6 , Li^6 , Li^7 , Li^8 , Li^9 , Be^7 , Be^9 , Be^{10} , B^8 , B^{10} and B^{11}
$2 \leq Z \leq 4$ and $23 \leq Z \leq 92$	γ , n, p, d, t, He^3 , He^4 , Li^6 , Li^7 and C^{12}	γ , n, p, d, t, He^3 , He^4 , He^6 and Li^6

The computations were carried out using the FORTRAN IV program SHADRACH on an IBM-7094 computer. This program, which will be made available on request, contains all the stored mass information and permits the calculation of Q-value matrices for any set of incident and exit particles.

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I. INTRODUCTION

With the current advances in particle detection equipment and the advent of modern accelerators, some experimental nuclear physicists are expanding their research to include more complex nuclear reactions. The purpose of this report is to provide nuclear reaction Q-value information that includes some of these complex multi-nucleon transfer reactions and which utilizes the improved atomic mass information that has become available in the past few years.

In compiling the mass data for this report we relied largely on recent compilations by other authors^{1,2,3)}, although eighty-six of these mass assignments have been revised in accordance with new experimental measurements and twenty-three new isotopes have been added. These additions and revisions are listed in Table I and discussed in Section II.

The choice of targets and the size of the Q-value matrices have been somewhat limited by space considerations. Matrices have therefore been calculated for all stable and naturally occurring radioactive isotopes (according to the Chart of the Nuclides, General Electric, 1966) and for a few common long-lived isotopes of $Z \leq 20$; the Table of Contents gives a full list of the target nuclides. The two-body reactions chosen and their inherent Q-values and errors are given in Table II. A more detailed description of the Q-value tables on pages 1 to 344 is given in Section III.

It was realized that our selection of both targets and reactions would be inadequate to cover the needs of many experimenters. For this reason the FORTRAN IV program SHADRACH will be made available on request. With this code it is possible to calculate any desired set of nuclear reactions on any series of targets. Furthermore the program includes a convenient means of maintaining an up-to-date mass table. A more complete discussion is given in Section IV.

II. ATOMIC MASSES

Three compilations formed the basis for the atomic mass data used in this report. That of Lauritsen and Ajzenberg-Selove ¹⁾ was used from mass 5 to 10. The 1964 Atomic Mass Table of Mattauch, Thiele and Wapstra ²⁾ was used for masses 1 through 4 and 11 through 205. For the masses over 205 we used a recent revision of the 1964 Mass Table ³⁾. This revision includes the recent experimental data in that mass region and corrects the masses of all the nuclides decaying by an α - β decay chain into Bi²¹³. The previously reported mass excesses of these nuclides have been found to be in error ³⁾. This new information from Wapstra is derived in the same manner as in the earlier mass table ²⁾.

An attempt was made in the mass 1 to 205 region to include more recent experimental information than was available in the above compilations. Table I gives a list of changes or additions to references 1 and 2. Those mass excesses listed as estimates are based on indirect experimental measurements. With these exceptions only direct experimental determinations of masses were used. It should be noted that while Table I includes no mass predictions, references 2 and 3 do include mass values derived from estimated α - or β -decay energies, and these values are contained in the following tables. Token errors of 1 MeV have been assigned to such values ²⁾. Q-values listed in the matrices as having errors of 1 MeV or

Table I. Additions and revisions to mass data.

Z	ISOTOPE	MASS EXCESS (MeV)	ERROR (MeV)	REF. NO.	Z	ISOTOPE	MASS EXCESS (MeV)	ERROR (MeV)	REF. NO.
2	He ⁸	31.650	0.120	4	31	Ga ⁶⁷	-66.862	0.010	23
4	* Be ¹²	25.100	1.000	5		Ga ⁷⁵	-68.530	0.200	31
6	C ⁹	28.990	0.070	6	32	Ge ⁷¹	-69.8992	0.0046	23
7	N ¹²	17.349	0.009	7		Ge ⁷⁸	-71.760	0.200	32
	N ¹⁸	13.100	0.400	8	33	As ⁸¹	-72.600	0.200	31
8	O ¹³	23.110	0.070	6	34	Se ⁷⁵	-72.1664	0.0037	23
	F ¹⁶	10.686	0.040	9		Se ⁸³	-75.440	0.200	33
9	F ²²	4.500	0.060	10	35	Br ⁷⁵	-69.159	0.020	34
10	Ne ¹⁷	16.510	0.200	11		Br ⁸⁶	-75.660	0.300	35
11	Na ²⁰	6.980	0.080	12		Br ⁸⁷	-74.600	0.500	35
12	* Mg ²⁰	16.400	1.500	13, (Na ²⁰)	36	Kr ⁷⁴	-61.310	1.000	36
	* Mg ²¹	10.900	0.200	14		Kr ⁷⁵	-64.060	1.000	(Br ⁷⁵)
	Mg ²²	-0.380	0.050	4		Kr ⁹⁰	-74.806	0.100	(Sr ⁹⁰)
13	Al ²⁴	-0.070	0.060	79	37	Rb ⁸³	-79.160	1.000	37
14	* Si ²⁵	4.000	0.200	14		Rb ⁸⁶	-82.725	0.007	20
	Si ³²	-24.090	0.007	15		Rb ⁹⁰	-79.366	0.100	(Sr ⁹⁰)
15	F ²⁸	-7.120	0.060	79	38	Sr ⁸³	-76.950	1.000	38, (Rb ⁸³)
16	* S ²⁹	-2.900	0.200	16		Sr ⁹⁰	-85.956	0.005	20, (Y ⁹⁰)
	S ³⁰	-13.957	0.025	17		Sr ⁹¹	-83.678	0.012	(Y ⁹¹)
18	Ar ³³	-9.600	0.200	27, 18		Sr ⁹³	-79.420	0.100	(Zr ⁹³)
19	K ³⁷	-24.7996	0.0031	19	39	Y ⁹⁰	-86.502	0.004	39, 40, 20
	K ⁴²	-35.016	0.012	20		Y ⁹¹	-86.348	0.007	40
	K ⁴⁴	-36.210	0.200	21		Y ⁹³	-84.223	0.021	(Zr ⁹³)
20	Ca ³⁷	-13.240	0.050	22	40	Zr ⁹³	-87.113	0.006	41
	Ca ⁴¹	-35.125	0.004	23		Zr ⁹⁸	-82.010	1.000	(Nb ⁹⁸)
	* Ca ⁵⁰	-41.100	1.000	24	41	Nb ⁹²	-86.455	0.010	42
21	Sc ⁴⁰	-20.380	0.060	25		Nb ⁹⁸	-83.510	0.200	43
	Sc ⁴¹	-28.630	0.011	(Ca ⁴¹)	42	Mo ⁹⁰	-80.173	0.013	44
	Sc ⁴²	-32.109	0.004	26	46	Pd ¹⁰¹	-85.404	0.023	45
22	* Ti ⁴¹	-15.830	0.100	27, (Ca ⁴¹)	48	cd ¹⁰⁹	-88.549	0.008	46
23	V ⁴⁶	-37.069	0.003	26	49	In ¹⁰⁸	-84.730	0.050	47
	V ⁴⁹	-47.9565	0.0022	23		In ¹⁰⁹	-86.530	0.013	(cd ¹⁰⁹)
	V ⁵³	-51.780	0.050	28		In ¹¹⁴	-88.585	0.008	39, 20
25	Mn ⁵⁰	-42.618	0.005	26		In ¹²⁰	-85.500	0.600	48
	Mn ⁵³	-54.6828	0.0034	23	52	Tc ¹¹⁵	-82.500	0.700	49
26	Fe ⁵⁵	-57.4728	0.0034	23	53	I ¹¹⁸	-80.700	1.100	50
27	Co ⁵⁷	-59.3389	0.0046	23		I ¹³⁰	-86.890	0.011	51
28	Ni ⁵⁹	-61.1599	0.0039	23	54	Xe ¹²⁷	-88.441	0.023	52
	Ni ⁶⁷	-63.200	0.300	29	55	Cs ¹²⁷	-86.341	0.060	(Xe ¹²⁷)
30	Zn ⁷²	-68.144	0.009	30		Cs ¹³⁸	-83.660	0.077	53
					57	La ¹³⁵	-86.820	0.220	54

* Estimated value

Table I. (cont.)

Z	ISOTOPE	MASS EXCESS (MeV)	ERROR (MeV)	REF. NO.	Z	ISOTOPE	MASS EXCESS (MeV)	ERROR (MeV)	REF. NO.
59	Pr ¹³⁸	-82.930	0.054	55	71	Lu ¹⁷⁰	-57.120	0.060	68
	Pr ¹³⁹	-85.048	0.054	56		Lu ¹⁷⁹	-48.920	0.090	53
	Pr ¹⁴⁰	-84.785	0.025	57	72	Hf ¹⁷⁵	-54.700	0.060	69
	Pr ¹⁴⁷	-75.480	0.200	58	73	Ta ¹⁷⁷	-51.562	0.070	53
	Pr ¹⁴⁸	-72.930	0.400	58		Ta ¹⁸²	-46.344	0.038	70
60	Nd ¹⁴⁰	-84.485	1.000	(Pr ¹⁴⁰)	75	Re ¹⁸⁹	-37.825	0.080	71
	Nd ¹⁴⁷	-78.178	0.017	78	76	Os ¹⁹⁴	-32.375	0.023	72
61	Pm ¹⁴⁴	-81.335	0.056	60	77	Ir ¹⁹²	-34.733	0.060	73
63	Eu ¹⁴⁴	-75.660	0.030	59		Ir ¹⁹⁴	-32.472	0.023	74
	Eu ¹⁵⁷	-69.419	0.035	61		Ir ¹⁹⁶	-29.233	0.024	75
	Eu ¹⁵⁹	-66.016	0.050	62	79	Au ¹⁹⁵	-32.548	0.017	76
65	Tb ¹⁶⁰	-67.846	0.019	63,64		Au ¹⁹⁸	-29.594	0.006	77
69	Tm ¹⁶⁵	-62.870	0.035	65		Au ¹⁹⁹	-29.085	0.007	77
	Tm ¹⁶⁶	-61.876	0.034	66					
	Tm ¹⁶⁷	-62.121	0.030	67					

greater have in general relied on at least one of these estimates. All values are atomic mass excesses in MeV on the C¹² scale.

In compiling Table I we have chosen to use the experimental data of the reference cited. Where there are multiple references to an isotope, the first is to be considered the prime reference used in determining the listed mass excess. No attempt was made to average experimental measurements. In cases where an isotope is listed as a reference, a change in the mass of the reference isotope has affected the mass of the reported nuclide. This occurs when masses are related through β-decay chains. The error on the mass excesses shown in Table I is either the experimental error or - primarily for β-decay chain data - the rms value obtained using the experimental error of the decay energy and the error in the mass of the daughter nuclide. Often the β-decay energy between nuclides is known more accurately than the mass of the daughter; this is indicated by reporting the mass excess of the parent to more significant figures than the error placed on it would warrant.

III. DESCRIPTION AND USE OF TABLES

A matrix of the tabulated reactions is shown in Table II together with the inherent mass difference and error for each. The (X,γ) row of this Table provides the mass excess and error of the incident particles while the (γ,X) column (with a sign change) gives the mass excess and error of the outgoing particles.

On pages 1 through 344 the Q-values for the various reactions on 293 isotopes are tabulated in matrix form. The general notation for the reaction is:

Target (incoming, outgoing) Product.

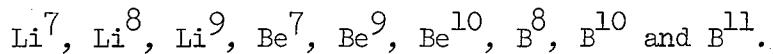
The ten incoming particles are the same on all targets:

γ, n, p, d, t, He³, He⁴, Li⁶, Li⁷ and C¹².

Nine outgoing particles are also common to all targets:

γ, n, p, d, t, He³, He⁴, He⁶ and Li⁶.

In addition for targets of $Z \leq 22$, reactions involving the following outgoing particles are also tabulated:



Across the top of each matrix is written the symbol and mass number of the target nucleus, its mass excess and error. Beneath the target information is a list of the incoming particles. The outgoing particles are listed along the left hand margin. In the box corresponding to the intersection of the "incoming" column and the "outgoing" row there are, in general, four pieces of information pertaining to that reaction as follows:

Q-value of the reaction (MeV)

Error in Q-value (MeV)

Symbol and mass number of product nucleus

Mass excess of product nucleus (MeV).

The following special cases are, however, somewhat different:

- (1) In elastic scattering reactions only the symbol and mass number of the target appears.
- (2) Some capture reactions in the light elements are redundant when the composite nucleus is also a tabulated "outgoing" particle (e.g. $\text{He}^3(n,\alpha)\gamma$). In this case the word GAMMA appears in place of the product nucleus and no mass excess is given.
- (3) In a reaction which does not conserve nucleons or when the product nucleus consists solely of multiple protons or neutrons, the words NO REACTION appear.
- (4) In reactions where the mass of the product nucleus is not known, the words MASS XX UNKNOWN are written, where XX stands for the symbol and mass number of the nuclide in question.

In the very light elements we also suppress entirely those rows of the matrix in which the exit particle corresponds identically with the target nucleus, a case which reduces to simple elastic scattering.

Table II. Atomic mass differences and uncertainties (all in MeV).

IN OUT	GAMMA	N	P	D	T	He^3	He^4	Li^6	Li^7	C ¹²
GAMMA	—	8.0714 ±.0001	7.2890 ±.0001	13.1359 ±.0001	14.9499 ±.0002	14.9313 ±.0002	2.4248 ±.0004	14.0884 ±.0011	14.9073 ±.0011	0.0000 ±.0000
N	- 8.0714 ±.0001	—	- 0.7824 ±.0001	5.0645 ±.0001	6.8785 ±.0002	6.8599 ±.0002	- 5.6467 ±.0004	6.0170 ±.0011	6.8359 ±.0011	- 8.0714 ±.0001
P	- 7.2890 ±.0001	0.7824 ±.0001	—	5.8469 ±.0001	7.6610 ±.0002	7.6423 ±.0002	- 4.8642 ±.0004	6.7994 ±.0011	7.6183 ±.0011	- 7.2890 ±.0001
D	- 13.1359 ±.0001	- 5.0645 ±.0001	- 5.8469 ±.0001	—	1.8140 ±.0002	1.7954 ±.0002	- 10.7112 ±.0004	0.9525 ±.0011	1.7714 ±.0011	- 13.1359 ±.0001
T	- 14.9499 ±.0002	- 6.8785 ±.0002	- 7.6610 ±.0002	- 1.8140 ±.0002	—	- 0.0186 ±.0003	- 12.5252 ±.0004	- 0.8615 ±.0011	- 0.0426 ±.0011	- 14.9499 ±.0002
He^3	- 14.9313 ±.0002	- 6.8599 ±.0002	- 7.6423 ±.0002	- 1.7954 ±.0002	0.0186 ±.0003	—	- 12.5066 ±.0004	- 0.8429 ±.0011	- 0.0240 ±.0011	- 14.9313 ±.0002
He^4	- 2.4248 ±.0004	5.6467 ±.0004	4.8642 ±.0004	10.7112 ±.0004	12.5252 ±.0004	12.5066 ±.0004	—	11.6636 ±.0012	12.4825 ±.0012	- 2.4248 ±.0004
He^6	- 17.5982 ±.0040	- 9.5268 ±.0040	- 10.3092 ±.0040	- 4.4623 ±.0040	- 2.6483 ±.0040	- 2.6669 ±.0040	- 15.1734 ±.0040	- 3.5098 ±.0041	- 2.6909 ±.0041	- 17.5982 ±.0040
Li^6	- 14.0884 ±.0011	- 6.0170 ±.0011	- 6.7994 ±.0011	- 0.9525 ±.0011	0.8615 ±.0011	0.8429 ±.0011	- 11.6636 ±.0012	—	0.8189 ±.0016	- 14.0884 ±.0011
Li^7	- 14.9073 ±.0011	- 6.8359 ±.0011	- 7.6183 ±.0011	- 1.7714 ±.0011	0.0426 ±.0011	0.0240 ±.0011	- 12.4825 ±.0012	- 0.8189 ±.0016	—	- 14.9073 ±.0011
Li^8	- 20.9462 ±.0015	- 12.8748 ±.0015	- 13.6572 ±.0015	- 7.8103 ±.0015	- 5.9962 ±.0015	- 6.0149 ±.0015	- 18.5214 ±.0016	- 6.8578 ±.0019	- 6.0389 ±.0019	- 20.9462 ±.0015
Li^9	- 24.9650 ±.0200	- 16.8936 ±.0200	- 17.6760 ±.0200	- 11.8291 ±.0200	- 10.0150 ±.0200	- 10.0337 ±.0200	- 22.5402 ±.0200	- 10.8766 ±.0200	- 10.0577 ±.0200	- 24.9650 ±.0200
Be^7	- 15.7689 ±.0011	- 7.6975 ±.0011	- 8.4799 ±.0011	- 2.6330 ±.0011	- 0.8190 ±.0011	- 0.8376 ±.0011	- 13.3441 ±.0012	- 1.6805 ±.0016	- 0.8616 ±.0016	- 15.7689 ±.0011
Be^9	- 11.3505 ±.0009	- 3.2791 ±.0009	- 4.0615 ±.0009	1.7854 ±.0009	3.5994 ±.0009	3.5808 ±.0009	- 8.9257 ±.0010	2.7379 ±.0014	3.5568 ±.0014	- 11.3505 ±.0009
Be^{10}	- 12.6070 ±.0022	- 4.5356 ±.0022	- 5.3180 ±.0022	0.5289 ±.0022	2.3429 ±.0022	2.3243 ±.0022	- 10.1822 ±.0022	1.4814 ±.0025	2.3003 ±.0025	- 12.6070 ±.0022
B ⁸	- 22.9231 ±.0015	- 14.8517 ±.0015	- 15.6341 ±.0015	- 9.7872 ±.0015	- 7.9732 ±.0015	- 7.9918 ±.0015	- 20.4983 ±.0016	- 8.8347 ±.0019	- 8.0158 ±.0019	- 22.9231 ±.0015
B ¹⁰	- 12.0522 ±.0005	- 3.9808 ±.0005	- 4.7632 ±.0005	1.0837 ±.0005	2.8978 ±.0005	2.8791 ±.0005	- 9.6274 ±.0006	2.0362 ±.0012	2.8551 ±.0012	- 12.0522 ±.0005
B ¹¹	- 8.6677 ±.0003	- 0.5962 ±.0003	- 1.3787 ±.0003	4.4682 ±.0003	6.2823 ±.0004	6.2637 ±.0004	- 6.2429 ±.0005	5.4207 ±.0011	6.2396 ±.0011	- 8.6677 ±.0003

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	He	2		S ₃₃	60
3	Li ₇ ⁶	3		S ₃₄	62
	Li ₇	4		S ₃₆	64
4	* Be ₉ ⁷	5	17	Cl ₃₅	66
	Be ₁₀	6		Cl ₃₆	68
	* Be	7		Cl ₃₇	70
5	B ₁₁ ¹⁰	8	18	Ar ₃₆	72
	B ₁₁	10		Ar ₃₈	74
6	C ₁₃ ¹²	12	19	Ar ₄₀	76
	C ₁₄	14		* K ₄₀ ³⁹	78
	* C	16		* K ₄₁	80
7	N ₁₅ ¹⁴	18	20	K	82
	N ₁₅	20		Ca ₄₁ ⁴⁰	84
8	O ₁₇ ¹⁶	22		* Ca ₄₂	86
	O ₁₈	24		Ca ₄₃	88
	O	26		Ca ₄₄	90
9	F ¹⁹	28		Ca ₄₆	92
10	Ne ₂₁ ²⁰	30	21	Ca ₄₈	94
	Ne ₂₂	32		Ca	96
	Ne	34		Sc ⁴⁵	98
11	* Na ₂₃ ²²	36			
	Na	38	22	Ti ₄₇ ⁴⁶	100
12	Mg ₂₅ ²⁴	40		Ti ₄₈ ⁴⁷	102
	Mg ₂₆	42		Ti ₄₉ ⁴⁸	104
	Mg	44	23	Ti ₅₀ ⁴⁹	106
13	* Al ₂₇ ²⁶	46		Ti	108
	Al	48	24	* V ₅₁ ⁵⁰	110
14	Si ₂₉ ²⁸	50		V	111
	Si ₃₀	52		Cr ₅₂ ⁵⁰	112
	Si	54		Cr ₅₃ ⁵²	113
15	P ³¹	56	25	Cr ₅₄ ⁵³	114
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			26	Mn ⁵⁵	116
				Fe ₅₇ ⁵⁴	117
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				Fe	119

* Naturally occurring or otherwise available radioactive isotope.

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27	Co ⁵⁹ 121	38	Sr ⁸⁴ Sr ⁸⁶ Sr ⁸⁷ Sr ⁸⁸ Sr 158 159 160 161
28	Ni ⁵⁸ Ni ⁶⁰ Ni ⁶¹ Ni ⁶² Ni ⁶⁴ Ni 122 123 124 125 126	39	Y ⁸⁹ 162
29	Cu ⁶³ Cu ⁶⁵ Cu 127 128	40	Zr ⁹⁰ Zr ⁹¹ Zr ⁹² Zr ⁹⁴ Zr ⁹⁶ Zr 163 164 165 166 167
30	Zn ⁶⁴ Zn ⁶⁶ Zn ⁶⁷ Zn ⁶⁸ Zn ⁷⁰ Zn 129 130 131 132 133	41	Nb ⁹³ 168
31	Ga ⁶⁹ Ga ⁷¹ 134 135	42	Mo ⁹² Mo ⁹⁴ Mo ⁹⁵ Mo ⁹⁶ Mo ⁹⁷ Mo ⁹⁸ Mo ¹⁰⁰ Mo 169 170 171 172 173 174 175
32	Ge ⁷⁰ Ge ⁷² Ge ⁷³ Ge ⁷⁴ Ge ⁷⁶ Ge 136 137 138 139 140	44	Ru ⁹⁶ Ru ⁹⁸ Ru ⁹⁹ Ru ¹⁰⁰ Ru ¹⁰¹ Ru ¹⁰² Ru ¹⁰⁴ Ru 176 177 178 179 180 181 182
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34	Se ⁷⁴ Se ⁷⁶ Se ⁷⁷ Se ⁷⁸ Se ⁸⁰ Se ⁸² Se 142 143 144 145 146 147	46	Pd ¹⁰² Pd ¹⁰⁴ Pd ¹⁰⁵ Pd ¹⁰⁶ Pd ¹⁰⁸ Pd ¹¹⁰ Pd 184 185 186 187 188 189
35	Br ⁷⁹ Br ⁸¹ 148 149	47	Ag ¹⁰⁷ Ag ¹⁰⁹ 190 191
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*Naturally occurring or otherwise available radioactive isotope.

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	Cd ¹¹¹	195		Ba ¹³⁴	235
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	Cd ¹¹⁴	198		Ba ¹³⁷	238
	Cd ¹¹⁶	199		Ba ¹³⁸	239
49	In ¹¹³	200			
	* In ¹¹⁵	201	57	* La ¹³⁸	240
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50	Sn ¹¹²	202	58	Ce ¹³⁶	242
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				Nd ¹⁵⁰	253
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* Naturally occurring or otherwise available radioactive isotope.

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* Naturally occurring or otherwise available radioactive isotope.

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