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Title

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Permalink

https://escholarship.org/uc/item/6q34t9dw

Journal

Microscopy and Microanalysis, 28(S1)

ISSN

1431-9276

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Publication Date

2022-08-01

DOI

10.1017/s1431927622009850

Peer reviewed

Atomic Scale Crystal Field Mapping of Polar Vortices in Oxide Superlattices

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Polar vortices in oxide superlattices can be utilized as potential candidates for data storage applications due to their unique polarization topologies.[1] The structure and dipole arrangement in polar vortices has been studied via X-ray scattering techniques, (scanning) transmission electron microscopy ((S)/TEM) and computational calculations.[1] However, the fundamental correlation between the atomic structure and the electronic structure (which is manifested in the chemical bonding) has heretofore not been explored. The hybridization between nominally empty *d* orbitals on the B-site with the occupied O 2p orbitals favors the condensation of a polar (ferroelectric) state in ABO_3 perovskite oxides.[2] The complex, continuously rotating local polarization texture of the vortices, in turn, can result in especially intricate *d*-orbital interactions. Soft X-ray spectroscopy can probe these interactions at the transition metal *L*-edge, but these techniques do not have the spatial resolution to resolve variations within one vortex (~5 nm region). Electron energy loss spectroscopy (EELS) in the STEM mode uses inelastically scattered electrons to probe the core-shell excitations (empty density of states) of transition metals at atomic resolution.

We studied the crystal field of the Ti *L*-edge in polar vortices formed in [(PbTiO₃)₁₆/(SrTiO₃)₁₆]₈ (PTO/STO) superlattices with a combination of high-resolution monochromated STEM-EELS mapping using a state-of-the-art direct electron detector and spectrometer (Gatan Continuum with a K3 detector), first-principles calculations, and crystal field multiplet theory. Changes in the crystal field of the Ti⁴⁺ cations in the PTO/STO superlattices are mapped as the spontaneous displacement of Ti^{4+} (and its corresponding 3*d* orbitals) rotates within the vortices.[3]

Figure 1(a) shows the zoomed-in schematic of a polar vortex where the continuous rotation of polarization affects the corresponding orbital hybridization. We first mapped out the presence of vortices using displacement vector mapping of the A sites in HAADF-STEM images (Figure 1 b,c). The atomic resolution EELS map was used to identify the Ti atoms (Figure 1d). The orbital hybridization within a vortex structure was experimentally identified by fine-structure of Ti *L*-edge spectra from different areas in STO, vortex edge and core (areas marked in Figure 1b). Within the PTO layer, the e_g peak in the Ti *L*-edge spectra has a negative shift as we move from vortex edge to core (Figure 1f). We fitted two gaussians (e'_g and e''_g) to e_g peak to understand this variation. We could map out the vortex core regions using the e''_g peak in the Ti *L*-edge spectra (Figure 1e). To examine the origin of subtle differences between Ti *L* edge spectra at the vortex core and edge, we employed crystal field multiplet EELS calculations in combination with density of states (PDOS) calculations. Figure 2a shows that the c/a ratio changes as the



polarization rotates. We used the e_g , t_{2g} and crystal field splitting parameters at different points in the oxide superlattices as inputs to calculate crystal field multiplet EELS spectra. We find that with the variation of strain (position 1 and 2), orbital rotation (position 1 and 3) and polarization magnitude (position 5 and 4) affects the local e_g and t_{2g} splitting and crystal field splitting of Ti 3d-O 2p orbitals, creating a pseudo Jahn Teller effect (Figure 2b). Mapping of the Ti 3d orbital rotation and hybridization at high resolution will serve as a stepping-stone to understand the microscopic consequences of physical phenomena such as chirality and negative permittivity that have been reported in such polar textures [1][4].



Figure 1: (a) Schematic representing the rotation of TiO₆ octahedra within one vortex domain. (b) Simultaneously acquired HAADF-STEM image. (c) The corresponding A site displacement vector map displaying the presence of the vortices in white circles. (d) Atomic resolution Ti L edge EELS map. (e) Intensity of the e''_g peak of the Ti- L_3 edge obtained via Gaussian fitting after binning by position. (f) De-noised Ti L edge spectra of STO (green solid box), PTO at the vortex core (blue box), and edge (orange box).



Figure 2: (a) Theoretical local polarization profile of polydomain structures in PTO/STO superlattices. Background gray scale represents the local tetragonality. (b) Comparison of multiplet calculated EEL spectra using parameters in crystal field, e_g and t_{2g} splitting as inputs and experimental EEL spectra for the five Ti₄₊ positions.

References:

- [1] A. K. Yadav et. al., Nature **530**, 198 (2016).
- [2] R. E. Cohen, Origin of Ferroelectricity in Perovskite Oxides, Nature 358, 136 (1992).

- [3] S. Susarla et. al. Nat. Commun. 12, 6273 (2021).
- [4] The electron microscopy experiments were performed at the Molecular Foundry, Lawrence Berkeley National Laboratory, which is supported by the U.S. Department of Energy under contract no. DE-AC02-05CH11231. S.S. is supported by the DOE BES program on Quantum Materials.