# **Lawrence Berkeley National Laboratory**

# **Recent Work**

## **Title**

MICROSTRUCTURAL FEATURES AFFECTING FRACTURE TOUGHNESS OF HIGH STRENGTH STEELS

# **Permalink**

https://escholarship.org/uc/item/7158n8sk

## **Authors**

Parker, Earl R. Zackay, Victor F.

# **Publication Date**

1974-06-01

To be presented at International United States-Japan Conference on Fracture Mechanics, Tohuku University of Sendai, Sendai, Japan, Aug. 12-16, 1974

> MICROSTRUCTURAL FEATURES AFFECTING FRACTURE TOUGHNESS OF HIGH STRENGTH STEELS

Earl R. Parker and Victor F. Zackay

June 1974

Prepared for the U.S. Atomic Energy Commission under Contract W-7405-ENG-48

# TWO-WEEK LOAN COPY

This is a Library Circulating Copy which may be borrowed for two weeks. For a personal retention copy, call Tech. Info. Division, Ext. 5545



## DISCLAIMER

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.

# MICROSTRUCTURAL FEATURES AFFECTING FRACTURE TOUGHNESS OF HIGH STRENGTH STEELS

bу

## Earl R. Parker and Victor F. Zackay

Inorganic Materials Research Division, Lawrence Berkeley Laboratory and Department of Materials Science and Engineering, College of Engineering; University of California, Berkeley, California

#### ABSTRACT

The fracture toughness of quenched and tempered steels, such as AISI 4340, AISI 4130, and 300M, can be increased by 50 to 100 percent by minor changes in heat treating procedures. Certain microstructural features, particularly blocky ferrite, upper bainite, and twinned martensite plates, are deleterious to fracture toughness. Similarly, the presence of undissolved carbides and sulfide inclusions, which act as crack nuclei, can lower fracture toughness by 25 to 50 percent. Other microstructural constituents, such as lower bainite, autotempered martensite, and retained austenite can enhance fracture toughness. By controlling the amounts and distributions of the microstructural constituents, the fracture toughness values of AISI 4340, AISI 4130, and 300M can be raised to the fracture toughness level of 18Ni maraging steel at equivalent values of yield strength.

#### MICROSTRUCTURAL FEATURES AFFECTING FRACTURE TOUGHNESS OF HIGH STRENGTH STEELS

#### INTRODUCTION

Ultra-high strength steels, such as AISI 4340 heat treated to have a yield strength over 200,000 psi, are relatively brittle when crack-like defects are present. An extensive research program has been underway in the author's laboratory during the past several years, with the objective of identifying the elements of defect structure and microstructure that have major effects upon the plane strain fracture toughness. Some elements of structure enhance fracture toughness, while others degrade this important property. The highlights of the various investigations involved are reported herein.

By minimizing or eliminating those microconstituents that lowered fracture toughness, and increasing those that enhanced it, the fracture toughness could be increased 50 to 100 percent above the values obtained with commercially used heat treatments. The optimization of properties could be attained by either of two means—(1) changing the heat treating procedure, or (2) changing the chemical composition of the alloy.

#### EXPERIMENTAL PROGRAM

In the heat treatment of steel, austenitizing is a critical step.

In general, use of low austenitizing temperatures is preferred because this procedure produces the smallest austenite grain size—and presumably the best combination of mechanical properties. In the present investigation, it was quickly found that low austenitizing temperatures did not provide the maximum fracture toughness (measured in accordance with ASTM

standard procedures). In fact, the reverse was found to be true, with increases of 50 percent or more being obtained when high austenitizing temperatures were used. (However, as is well known, slightly better elongation and reduction in area are obtained in tensile test bars that have been subjected to low austenitizing temperatures). The compositions of the steels used in the present investigation are listed in Table I.

One of the important factors influenced by austenitizing temperature is the presence of undissolved carbides in the microstructure. This was clearly shown by the work of T. Tom[1], who used special steels containing 0.30C-5Mo and 0.41C-5Mo in his investigation. Both strength and fracture toughness were found to increase with increasing austenitizing temperature, as shown in Fig. 1. Metallographic examinations revealed that undissolved carbides were present below about 1050°C in the 0.30C-5Mo steel, but not above this temperature. The plane strain fracture toughness of the steel increased suddenly from 50 ksi-in  $^{1/2}$  to 100 ksi-in  $^{1/2}$ when the austenitizing temperature was raised from 1000°C to 1100°C. A similar, but more gradual, change was found for the 0.41C-5Mo steel. Experiments were made with several other steels of lower alloy content (e.g. 0.34C-1Mo and 0.35C-1Mo-3Ni). In these steels, all carbides were dissolved at 870°C and the fracture toughness values were found to be nearly independent of the austenitizing temperature (or grain size). These results are shown in Fig. 2.

Another important influence of austenitizing temperature is its effect on hardenability. Two factors affecting hardenability are the austenite grain size and the amount of carbon in solution. In steels of

high hardenability, and in which all of the carbides are dissolved at low austenitizing temperatures, the grain size effect is not normally evident. This is so because the hardenability, even with the smaller grain size, is adequate to prevent the formation of ferrite and upper bainite during quenching. Low values of fracture toughness are generally associated with the presence of these transformation products. However, with steels of intermediate hardenability, such as AISI 4130, the effect of grain size (i.e. austenitizing temperature) may be very important[2]. In the present study when AISI 4130 steel was austenitized at 1200°C and ice-brine quenched, the fracture toughness was nearly twice as high as it was for the same steel oil quenched from 870°C. Oil quenching from 1200°C improved the fracture toughness relative to the lower temperature treatment, but less so than the more severe quench. The results are shown in Fig. 3. The tensile mechanical properties of as-quenched AISI 4130, AISI 4330 and AISI 4340 steels of the present investigation are listed in Table II. The microstructural features of AISI 4130 steel differed for the various treatments, with the 870°C oil quench producing a mixture of blocky ferrite and martensite that was partially autotempered. The microstructure of the 1200°C ice-brine quenched material had no blocky ferrite; it consisted entirely of martensite (partially autotempered). A similar correspondence between heat treatment, fracture toughness, and microstructure existed for AISI 4330 steel. Plots of room temperature plane strain fracture toughness vs tempering temperature for two austenitizing treatments are shown in

The use of high austenitizing treatments for steels having high hardenabilities, such as AISI 4340, also resulted in substantially higher  $K_{IC}$  values, even though there were no undissolved carbides, blocky ferrite, or upper bainite present in the microstructure resulting from the use of lower (870°C) quenching temperature. The room temperature fracture toughness of AISI 4340 steel is shown in Fig. 5 as a function of tempering temperature for two austenitizing temperatures. The preferred heat treatment involves a step-quench, i.e. cooling slowly from 1200°C to 870°C before oil quenching. Quenching directly into oil from 1200°C tended to cause cracking, as did ice-brine quenching from 870°C. Step-quenching (1200°>870°C 0Q) nearly doubled the as-quenched (870°C 0Q) fracture toughness.

Optical metallographic examination failed to reveal any significant microstructural difference resulting from the two austenitizing treatments (other than differences in austenite grain size). Transmission electron microscopy was employed to reveal additional small-scale structural differences. It was found that the 1200°C material had a significant amount of retained austenite in the form of 100-200Å thick films between martensite laths, whereas the 870°C material had very few austenite films[5], as shown in Fig. 6. Austenite is tough and crack resistant. The beneficial effect of the higher temperature treatment was therefore attributed mainly to the presence of the austenite films. However, there was another distinct microstructural difference observed in the transmission electron microscope studies. The packets of lath martensite were similar for both austenitizing treatments, but the

material heat treated at 870°C also had twinned martensite plates. Das and Thomas[3] and Thomas[4] have shown that the presence of twinned martensite plates causes the fracture toughness to be lower than with lath martensite alone. This additional difference in microstructure is believed to have been partially responsible for the higher fracture toughness of the step-quenched steel.

Additional investigations of the effects of heat treatments and variations in chemical composition are continuing. Preliminary results indicate that even greater improvements in fracture toughness may be forthcoming.

#### ACKNOWLEDGMENTS

The authors are grateful to Dr. Michael Yokota, Dr. George Lai, Dr. William Wood, Dr. Thomas Tom and Mr. Naga Prakash Babu for use of unpublished data, and to Dr. M. Dilip Bhandarkar for many helpful discussions and critical technical review.

This work was performed partially under the auspices of the United States Atomic Energy Commission through the Lawrence Berkeley Laboratory, Inorganic Materials Research Division, and partially under the auspices of the Army Mechanics and Materials Research Center, Watertown, Mass.

#### REFERENCES

- [1] T. Tom, Microstructural Variables and Fracture Toughness of High Strength Mo and Mo-Ni Steels. D. Eng. Thesis, LBL-1856, University of California, Berkeley, Calif., Sept. 1973.
- [2] W. E. Wood, E. R. Parker, and V. F. Zackay, An Investigation of
  Metallurgical Factors Which Affect the Fracture Toughness of
  Ultra-High Strength Steels, LBL-1474, Lawrence Berkeley Laboratory,
  Berkeley, Calif., May 1973.
- [3] S. K. Das and G. Thomas, Structure and Mechanical Properties of Fe-Ni-Co-C Steels, Trans. ASM 62, 659-676, 1969.
- [4] G. Thomas, Electron Microscopy Investigations of Ferrous Martensite.

  Met. Trans. 2, 2373-2385, 1971.
- [5] G. Y. Lai, W. E. Wood, R. A. Clark, V. F. Zackay, and E. R. Parker,

  Effect of Austenitizing Temperature on the Amount of Retained

  Austenite in AISI 4340 Steel, To be submitted to Met Trans., 1974.

TABLE I Chemical Compositions of Steels

	Compositions, Wt%									
Designation	, C	Cr	Mn	Мо	Ni	P	S	Si	V	
0.30C-5Mo	0.30	_	0.60	5.03	_	0.008	0.005	<0.02		
0.41C-5Mo	0.41	. · <del>-</del> .	0.51	4.93	<del>-</del>	0.007	0.005	<0.02	-	
0.34C-1Mo	0.34	_	0.63	0.95	_	0.008	0.005	<0.02		
0.35C-1Mo-3Ni	0.35	; <del>-</del>	o.a	0.95	3.1	0.007	0.005	<0.02	-	
AISI 4130-A	0.31	0.85	0.57	0.18	0.15	0.008	0.009	0.28	<0.005	
AISI 4130-B	0.33	0.90	0.63	0.18	0.15	0.008	0.009	0.27	<0.005	
AISI 4330	0.28	0.85	1.02	0.40	1.80	0.009	0.005	0.28	0.07	
AISI 4340	0.40	0.72	0.85	0.24	1.73	0.004	0.010	0.22	<0.005	

TABLE II Tensile Properties of As-quenched AISI 4130, AISI 4330 and AISI 4340 Steels

Steel	Austenitizing Temp. (°C) and Quenching Medium	0.2% Offset Yield Strength, ksi	Ultimate Tensile Strength, ksi	Elong.	Red. in Area %
AISI 4130	870, Oil	201	284	11.4	32.9
	1200, Oil	205	276	5.6	9.9
	1200, Ice-brine	215	249	1.6	6.6
AISI 4330	870, 011	234	284	13.1	43
	1200, Ice-brine	210	222	0.6	3.4
AISI 4340	870, 011	231	322	9.0	30.8
	1200→870, Oi1	231	318	3.2	7.8

# FIGURE CAPTIONS

- 1. Plots showing the influence of austenitizing temperature on room temperature fracture toughness ( $K_{IC}$  or  $K_Q$ ), yield strength (Y.S.), and ultimate strength (U.S.) of 0.30C-5Mo and 0.41C-5Mo Steels.
- 2. Plots of room temperature plane strain fracture toughness vs prior austenite grain size (indicated by ASTM grain size number) for as-quenched 0.34C-1Mo and 0.35C-1Mo-3Ni steels.
- 3. Plots of room temperature plane strain fracture toughness vs

  tempering temperature for AISI 4130 steel. Austenitizing temperatures and quenching media are indicated. (IBQLN is an abbreviation for ice-brine quenching followed by refrigeration in liquid nitrogen.)
- 4. Plots of room temperature plane strain fracture toughness vs
  tempering temperature for AISI 4330 steel. Austenitizing temperatures and quenching media are indicated.
- 5. Plots of room temperature plane strain fracture toughness vs tempering temperature for AISI 4340 steel. Austenitizing temperatures and quenching media are indicated.
- 6. Transmission electron micrographs of as-quenched AISI 4340 steel:
  Bright field (a) and dark field of austenite reflection (b) for the
  870°C austenitized specimen, and bright field (c) and dark field of
  austenite reflection (d) for the 1200°C>870°C austenitized specimen.

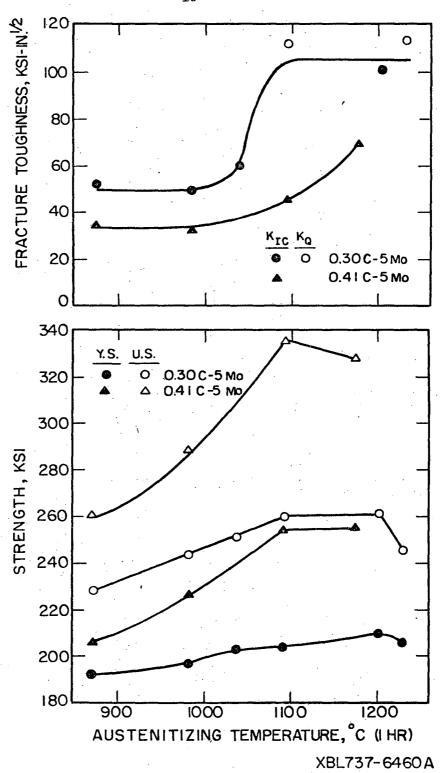


Fig. 1

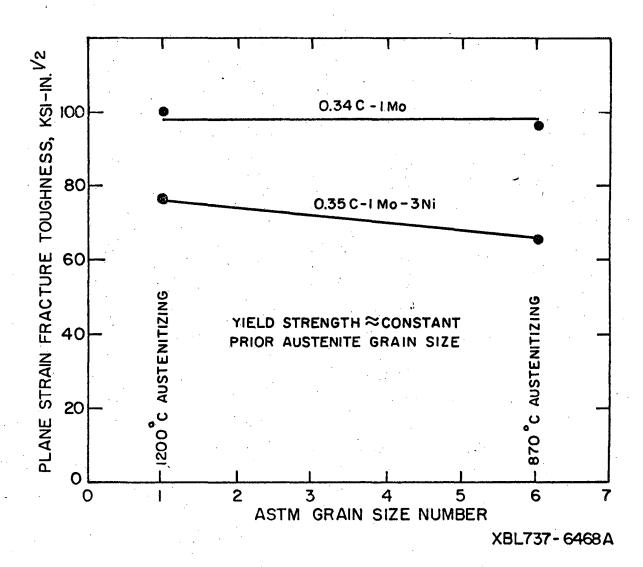


Fig. 2

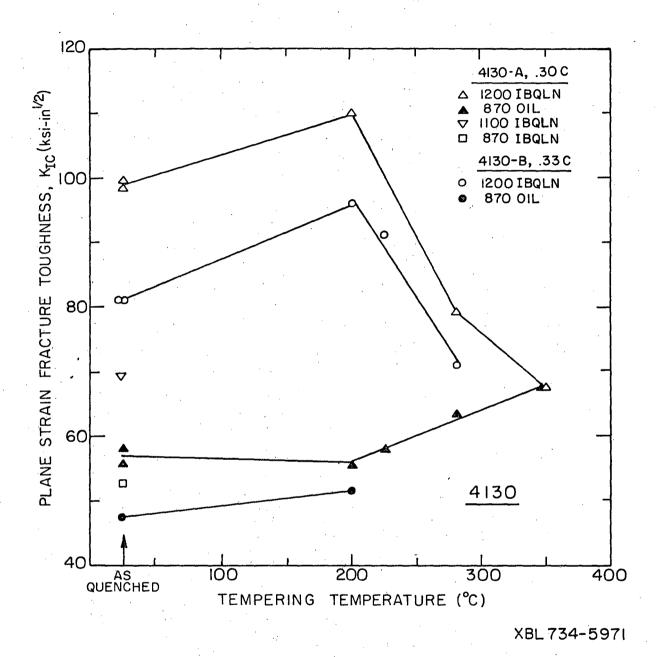
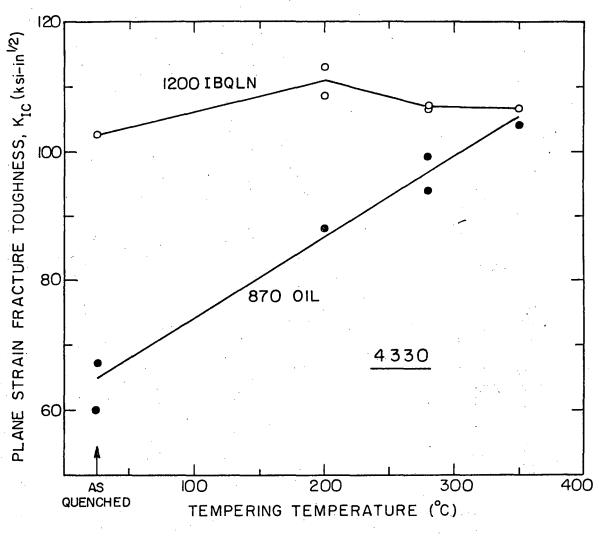


Fig. 3



XBL 734-5973

Fig. 4

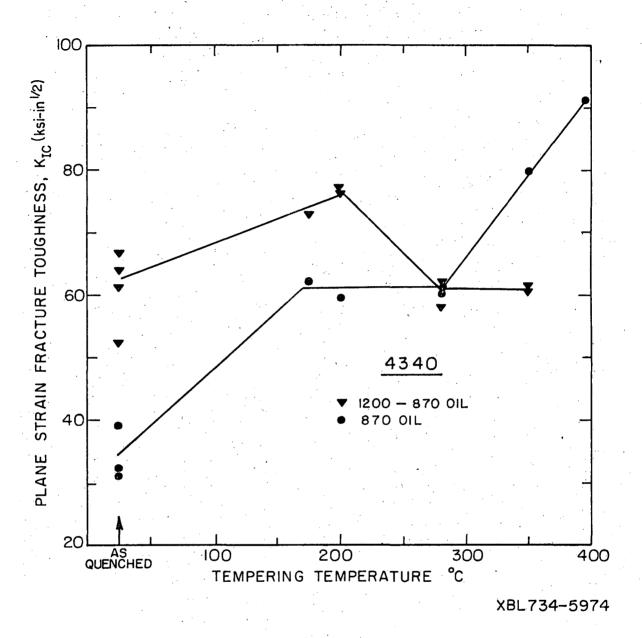


Fig. 5

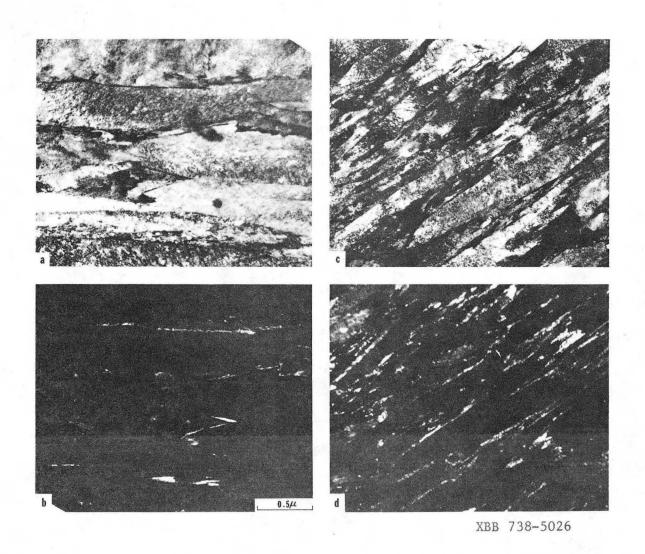


Fig. 6

#### LEGAL NOTICE

This report was prepared as an account of work sponsored by the United States Government. Neither the United States nor the United States Atomic Energy Commission, nor any of their employees, nor any of their contractors, subcontractors, or their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness or usefulness of any information, apparatus, product or process disclosed, or represents that its use would not infringe privately owned rights.

TECHNICAL INFORMATION DIVISION LAWRENCE BERKELEY LABORATORY UNIVERSITY OF CALIFORNIA BERKELEY, CALIFORNIA 94720