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Haasch, Richard T
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Single Crystal Perovskites Analyzed Using X-ray Photoelectron Spectroscopy: 1. SrTiO₃(001)

Richard T. Haasch^{a)}

Department of Materials Science and Engineering and Materials Research Laboratory, University of Illinois, Urbana-Champaign

Eric Breckenfeld

Department of Materials Science and Engineering and Materials Research Laboratory, University of Illinois, Urbana-Champaign; and Naval Research Laboratory, Washington, DC

Lane W. Martin

Department of Materials Science and Engineering and Materials Research Laboratory, University of Illinois, Urbana-Champaign; Department of Materials Science and Engineering, University of California, Berkeley; and Materials Science Division, Lawrence Berkeley National Laboratory

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X-ray photoelectron spectroscopy (XPS) was used to analyze a commercially available SrTiO₃(001) bulk single crystal. XP spectra were obtained using incident monochromatic Al K_α radiation at 0.83401 nm. A survey spectrum together with O 1s, Ti 2p, C 1s, Sr 3p, Sr 3d, Ti 3p, Sr 4p, Ti 3p, Sr 4p, and O 2s core level spectra and the valence band are presented. The spectra indicate the principle core level photoelectron and Auger electron signals and show only minor carbon contamination. Making use of the O 1s, Ti 2p, Sr 3d lines and neglecting the components related to surface contaminants, XPS quantitative analysis reveals an altered stoichiometry of the air-exposed crystal surface of Sr_{1.16}TiO_{2.09}. © 2014 American Vacuum Society.

[<http://dx.doi.org/10.1116/11.20140901>]

Keywords: strontium titanium oxide; perovskite

INTRODUCTION

Transition metal oxides present an impressive variety of functionality which is not available in more traditional systems such as group IV and III-V semiconductors or elemental metals. Among the many possible functionalities are, for instance, ferroelectricity (Ref. 1) and magnetism (Ref. 2), colossal magnetoresistance (Ref. 3), and high temperature superconductivity (Ref. 4), with transport character ranging from insulating to semiconducting to metallic. Furthermore, these properties are extremely sensitive to perturbations from chemistry, structural defects, strain and many other effects and this, in turn, provides the materials engineer a number of routes by which to engineer new functionalities in this class of materials (Ref. 5). While even simple oxide systems, such as binary oxides, exhibit a broad diversity of properties, it is the ternary systems which have received the most attention in recent years. In particular, materials possessing the perovskite structure (with chemical formula ABO₃) have been observed to exhibit an incredible variety of functionality and phenomena. Advances in thin film epitaxy, particularly pulsed laser deposition, RF magnetron sputtering, and molecular beam epitaxy, have enabled researchers to carefully tune material properties using epitaxial strain. Such approaches have provided an opportunity to apply large biaxial strains (as much as several percent in some cases) to nanoscale films of various materials which would lead to cracks in bulk materials under similar values of hydrostatic strain (Ref. 6).

^{a)}Author to whom correspondence should be addressed.

SPECIMEN DESCRIPTION (ACCESSION #01310)

Host Material: Single crystal SrTiO₃

CAS Registry #: 1206-05-9

Host Material Characteristics: homogeneous; solid; single crystal; dielectric; inorganic compound

Chemical Name: Strontium titanium oxide

Source: Crystec, GmbH. Grown by the Verneuil method.

Host Composition: SrTiO₃

Form: single crystal

Structure: cubic, perovskite structure, $a = 0.3905$ nm (Ref. 7)

History & Significance: SrTiO₃ has been widely studied for its interesting properties and as a common substrate for epitaxial film growth. The static dielectric permittivity (ϵ_r) of SrTiO₃ is ~ 300 at room temperature and rapidly increases upon cooling to values in excess of 20,000 in the quantum paraelectric state below 10K (Ref. 8). Likewise, SrTiO₃ has drawn attention as a candidate material for thermoelectrics based on its large carrier effective mass and resulting large thermopower (Ref. 9). More generally, SrTiO₃ is one of the most widely studied perovskite oxides and is highly susceptible to donor-doping by cationic substitution, oxygen vacancies, and field effects that result in a dramatic range of transport properties ranging from insulating to metallic to superconducting states (Ref. 10). In order to gain an increased understanding of the surfaces and hetero-interfaces of perovskite-based materials, a SrTiO₃(001) bulk single crystal was analyzed using X-ray photoelectron spectroscopy.

Accession #: 01310

Technique: XPS

Host Material: Single crystal SrTiO₃

Instrument: Kratos Axis Ultra

Major Elements in Spectra: Sr, Ti, O

Minor Elements in Spectra: C

Published Spectra: 7

Spectra in Electronic Record: 7

Spectral Category: comparison

As Received Condition: as grown

Analyzed Region: same as host material

Ex Situ Preparation/Mounting: Samples were cleaned ultrasonically for 5 min each in Formula 409[®], methyl alcohol, and deionized water. Samples were mounted onto the sample holder using double-sided carbon tape (Pella product number 16074).

In Situ Preparation: None

Pre-Analysis Beam Exposure: Less than 2 min; no x-ray degradation effects observed

Charge Control: low energy flood gun/magnetic immersion lens combination, filament current = 2.1 A, charge balance = 2.1 V, filament bias = 2 V

Temp. During Analysis: 300 K

Pressure During Analysis: $<3 \times 10^{-7}$ Pa

INSTRUMENT DESCRIPTION

Manufacturer and Model: Kratos Axis Ultra

Analyzer Type: spherical sector

Detector: channeltron electron multiplier

Number of Detector Elements: 8

INSTRUMENT PARAMETERS COMMON TO ALL SPECTRA

■ Spectrometer

Analyzer Mode: constant pass energy

Throughput ($T = E^M$): $N = 0$

Excitation Source Window: not specified

Excitation Source: Al K_{α} , monochromatic

Source Energy: 1486.6 eV

Source Strength: 120 W

Source Beam Size: $2000 \mu\text{m} \times 2000 \mu\text{m}$

Signal Mode: multichannel direct

■ Geometry

Incident Angle: 54°

Source to Analyzer Angle: 54°

Emission Angle: 0°

Specimen Azimuthal Angle: 45°

Acceptance Angle from Analyzer Axis: 0°

Analyzer Angular Acceptance Width: $40^\circ \times 40^\circ$

DATA ANALYSIS METHOD

Energy Scale Correction: The binding energy scale was referenced to C 1s = 285.0 eV.

Recommended Energy Scale Shift: +3.929 eV for high-resolution spectra

Peak Shape and Background Method: Background: Custom three parameter Tougaard background (Ref. 11), U 4 Tougaard (B, C, D, $T_0 = 0$) (Ref. 12), was used. O 1s: $B = 270 \text{ eV}^2$, $C = 300 \text{ eV}^2$, $D = 100 \text{ eV}^2$. Ti 2p: $B = 230 \text{ eV}^2$, $C = 300 \text{ eV}^2$, $D = 100 \text{ eV}^2$. C 1s: $B = 165 \text{ eV}^2$, $C = 260 \text{ eV}^2$, $D = 300 \text{ eV}^2$. Sr 3d: $B = 230 \text{ eV}^2$, $C = 300 \text{ eV}^2$, $D = 100 \text{ eV}^2$.

Quantitation Method: Quantification was done using region and component definitions with CasaXPS version 2.3.15. Sensitivity factors supplied by Kratos Analytical. Errors are given as ± 1 standard deviation. Standard deviations are calculated by CasaXPS using a Monte Carlo method for determining the error distribution for the computed areas.

ACKNOWLEDGMENTS

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REFERENCES

1. M. Dawber, K. M. Rabe, and J. F. Scott, *Rev. Mod. Phys.* **77**, 1083 (2005).
2. S. A. Chambers, T. C. Droubay, C. M. Wang, K. M. Rosso, S. M. Heald, D. A. Schwartz, K. R. Kittilstved, and D. R. Gamelin, *Mater. Today* **9**, 28 (2006).
3. A. P. Ramirez, *J. Phys.: Condens. Matter* **9**, 8171 (1997).
4. J.-M. G. Chen Langlois, Y. Guo, and W. A. Goddard III, *Proc. Natl. Acad. Sci. U.S.A.* **86**, 3447 (1989).
5. D. G. Schlom, L.-Q. Chen, C.-B. Eom, K. M. Rabe, S. K. Streiffner, and J.-M. Triscone, *Annu. Rev. Mater. Res.* **37**, 589 (2007).
6. L. W. Martin and D. G. Schlom, *Curr. Opin. Solid State Mater. Sci.* **16**, 199 (2012).
7. R. W. G. Wyckoff, *Crystal Structures*, 2nd ed. (Wiley, New York, 1963), Vol. 2, p. 390.
8. K. A. Muller and H. Burkard, *Phys. Rev. B* **19**, 3593 (1979).
9. H. P. R. Frederikse, W. R. Thurber, and W. R. Hosler, *Phys. Rev.* **134**, A442 (1964).
10. O. N. Tufte and P. W. Chapman, *Phys. Rev. B* **155**, 796 (1967).
11. S. Tougaard, *Surf. Interface Anal.* **25**, 137 (1997).
12. N. Fairley and A. Carrick, *The Casa Cookbook Part 1: Recipes for XPS Data Processing* (Acolyte Science, Cheshire, UK, 2005), pp. 147–671.
13. B. Reihl, J. G. Bednorz, K. A. Miller, Y. Jugnet, G. Landgren, and J. F. Morar, *Phys. Rev. B* **30**, 803 (1984).

SPECTRAL FEATURES TABLE

Spectrum ID #	Element/ Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (eV × cts/s)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
01310-02	O 1s	529.5	1.07	16808.6	0.780	35.45	SrTiO ₃
01310-02 ^a	O 1s	531.7	2.07	7456.7	0.780	15.73	carbonate
01310-03	Ti 2p	18117.1	2.001	14.59	...
01310-03	Ti 2p _{3/2}	458.4	0.98	SrTiO ₃
01310-03	Ti 2p _{1/2}	464.1	1.72	SrTiO ₃
01310-04 ^a	C 1s	285.0	1.20	2426.4	0.278	14.36	hydrocarbon
01310-04 ^a	C 1s	288.9	1.53	418.5	0.278	2.48	carbonate
01310-04	Sr 3p _{3/2}	268.7	2.11	77.8	0.278	0.46	SrTiO ₃
01310-04	Sr 3p _{1/2}	279.1	2.11	SrTiO ₃
01310-05	Sr 3d	18616.8	1.843	16.93	...
01310-05	Sr 3d _{5/2}	132.9	0.85	SrTiO ₃
01310-05	Sr 3d _{3/2}	134.7	0.85	SrTiO ₃
01310-06	Ti 3p	37.6	1.83	SrTiO ₃
01310-07	Sr 4p _{3/2}	18.7	1.26	SrTiO ₃
01310-07	Sr 4p _{1/2}	19.8	1.26	SrTiO ₃
01310-07	O 2s	21.4	2.13	SrTiO ₃
01310-07	O 2s	24.2	3.78	carbonate
01310-07 ^b	valence band	7.2	1.33	SrTiO ₃
01310-07 ^c	valence band	6.0	1.10	SrTiO ₃
01310-07 ^d	valence band maximum	3.1	1.41	SrTiO ₃

^a Result of exposure to air

^b O 2p and Ti 3d (Ref. 13)

^c O 2p (Ref. 13)

^d The position of VBM was estimated by subtracting 1/2 of the full width at half maximum (FWHM) from the position of the maximum intensity at the VBM.

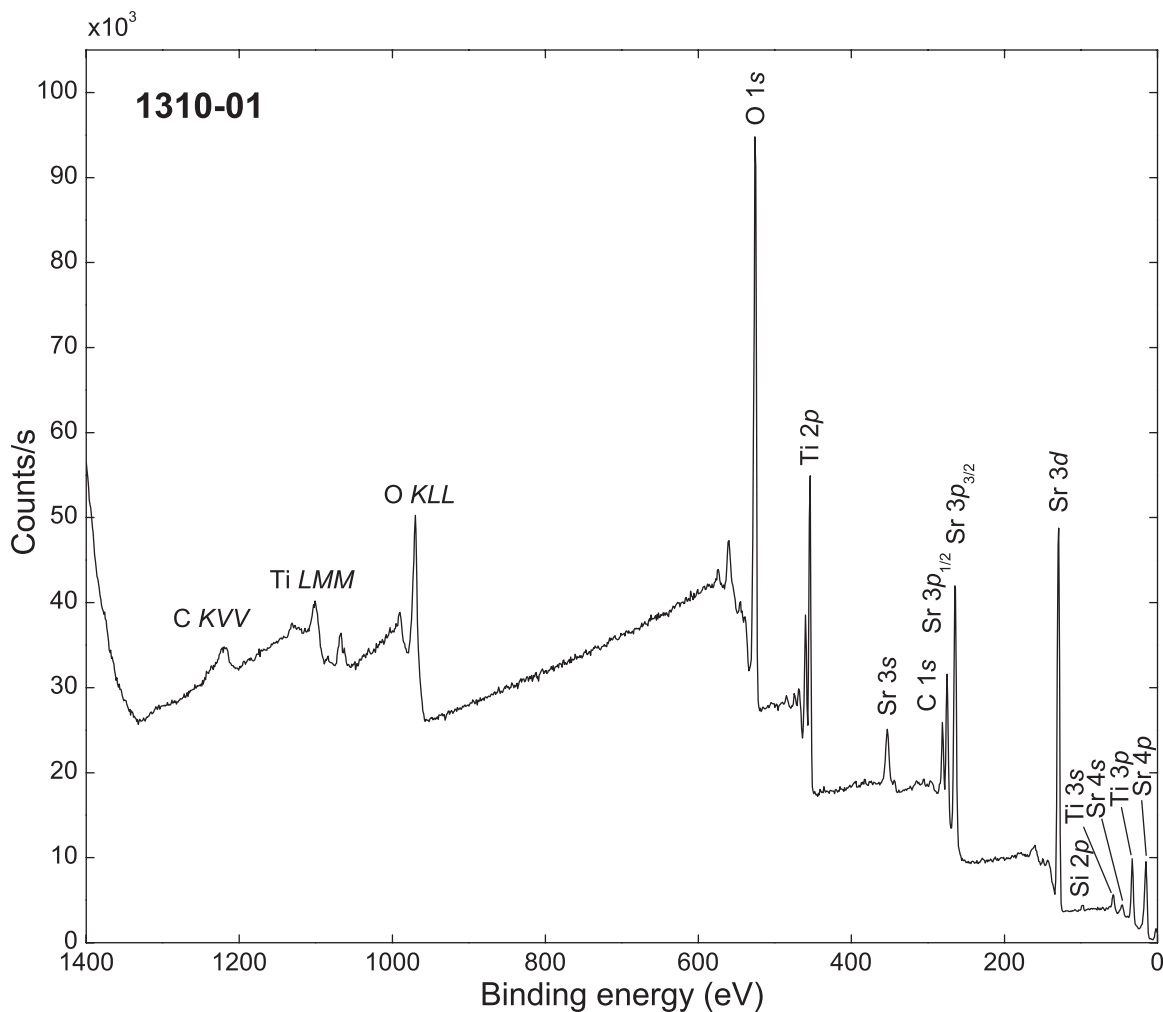
ANALYZER CALIBRATION TABLE

Spectrum ID #	Element/ Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (eV × cts/s)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
	Au 4f _{7/2}	84.0	0.72	151917.9
	Ag 3d _{5/2}	368.2	0.58	230506.2
	Cu 2p _{3/2}	932.6	0.88	410979.8

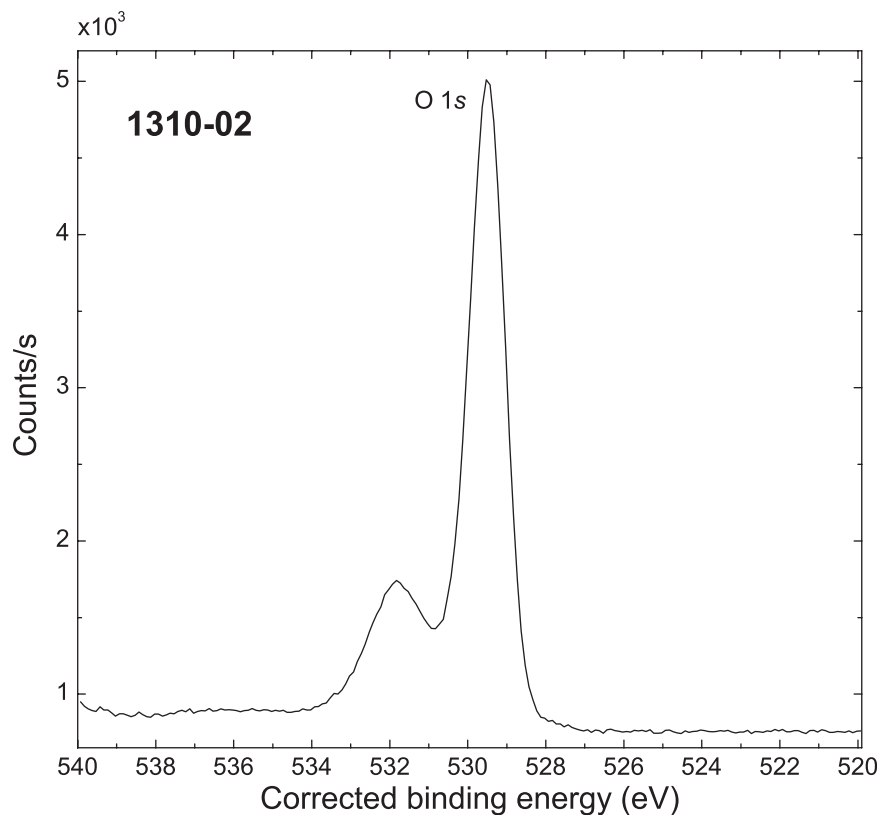
GUIDE TO FIGURES

Spectrum (Accession) #	Spectral Region	Voltage Shift[*]	Multiplier	Baseline	Comment #
1310-01	survey	0	1	0	
1310-02	O 1s	-3.929	1	0	
1310-03	Ti 2p	-3.929	1	0	
1310-04	C 1s, Sr 3p	-3.929	1	0	
1310-05	Sr 3d	-3.929	1	0	
1310-06	Ti 3p	-3.929	1	0	
1310-07	Sr 4p, O 2s, valence band	-3.929	1	0	

^{*} Voltage shift of the archived (as-measured) spectrum relative to the printed figure. The figure reflects the recommended energy scale correction due to a calibration correction, sample charging, flood gun, or other phenomenon.

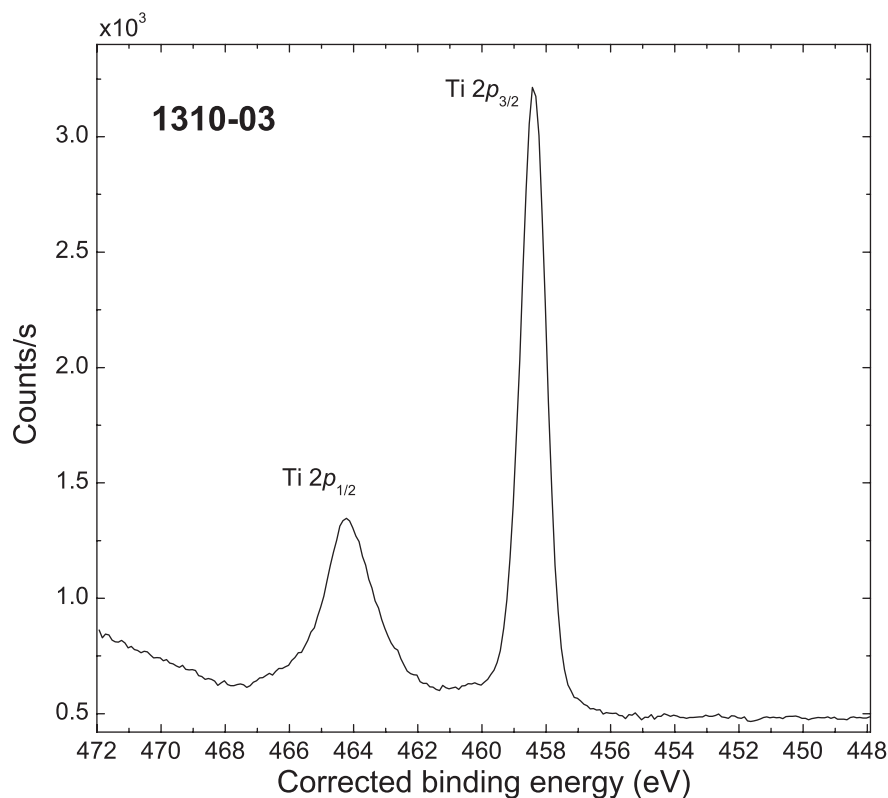


Accession #	01310-01
Host Material	Single crystal SrTiO ₃
Technique	XPS
Spectral Region	survey
Instrument	Kratos Axis Ultra
Excitation Source	Al K ₂ monochromatic
Source Energy	1486.6 eV
Source Strength	120 W
Source Size	2 mm × 2 mm
Analyzer Type	spherical sector
Incident Angle	54°
Emission Angle	0°
Analyzer Pass Energy:	160 eV
Analyzer Resolution	2.4 eV
Total Signal Accumulation Time	560 s
Total Elapsed Time	1120 s
Number of Scans	5
Effective Detector Width	33.6 eV



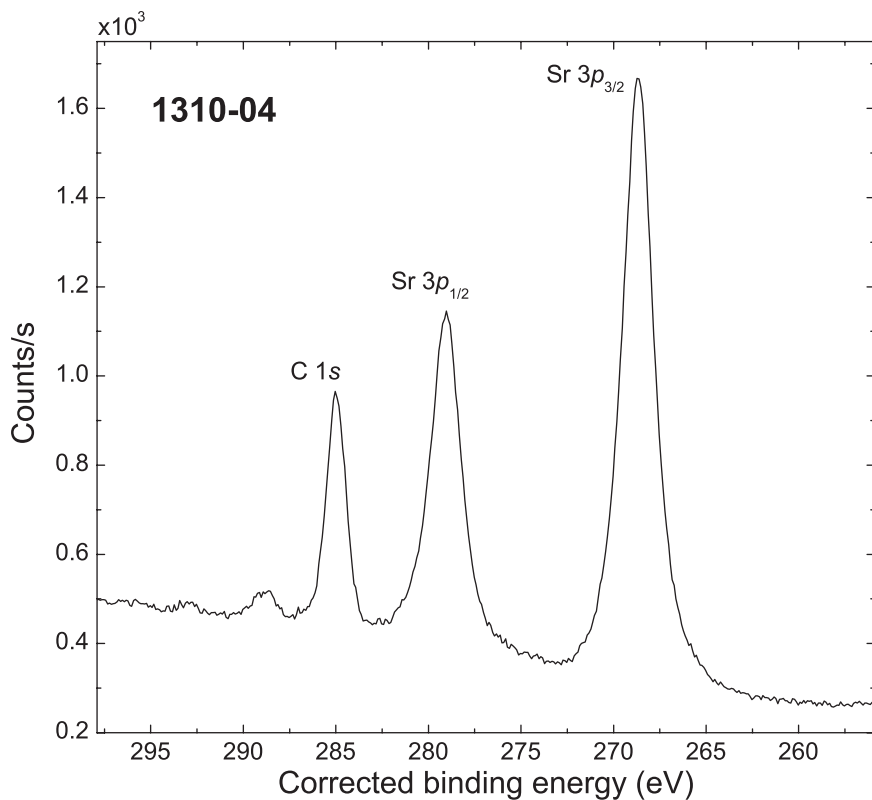
■ **Accession #:** 01310-02
 ■ **Host Material:** Single crystal SrTiO₃
 ■ **Technique:** XPS
 ■ **Spectral Region:** O 1s

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 1508 s
 Total Elapsed Time: 4147 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV



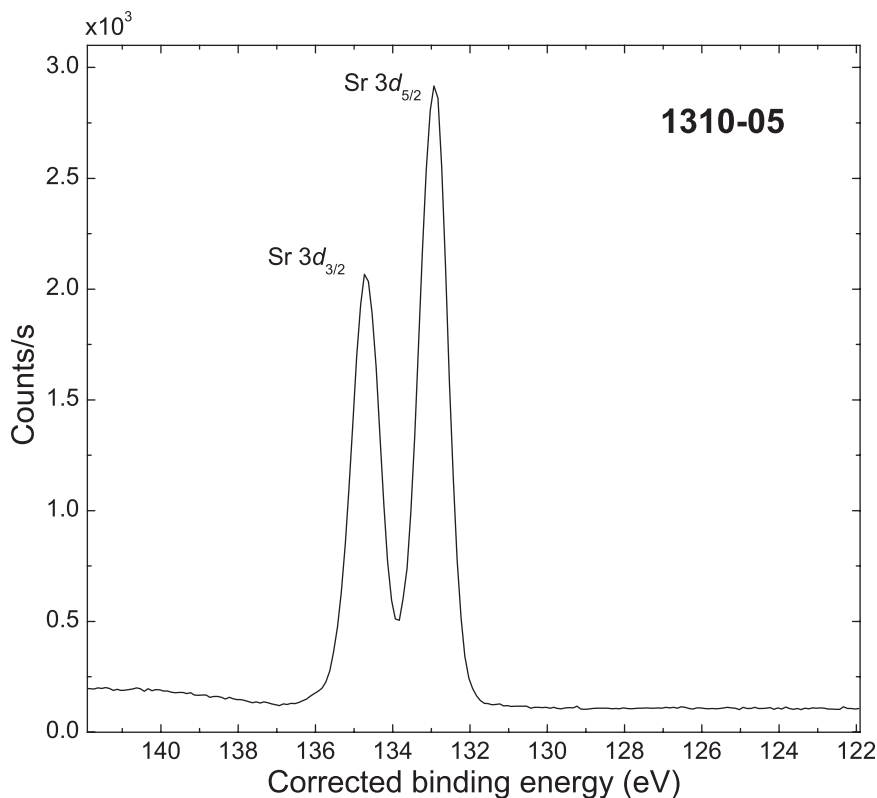
■ **Accession #:** 01310-03
 ■ **Host Material:** Single crystal SrTiO₃
 ■ **Technique:** XPS
 ■ **Spectral Region:** Ti 2p

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 1808 s
 Total Elapsed Time: 4972 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV



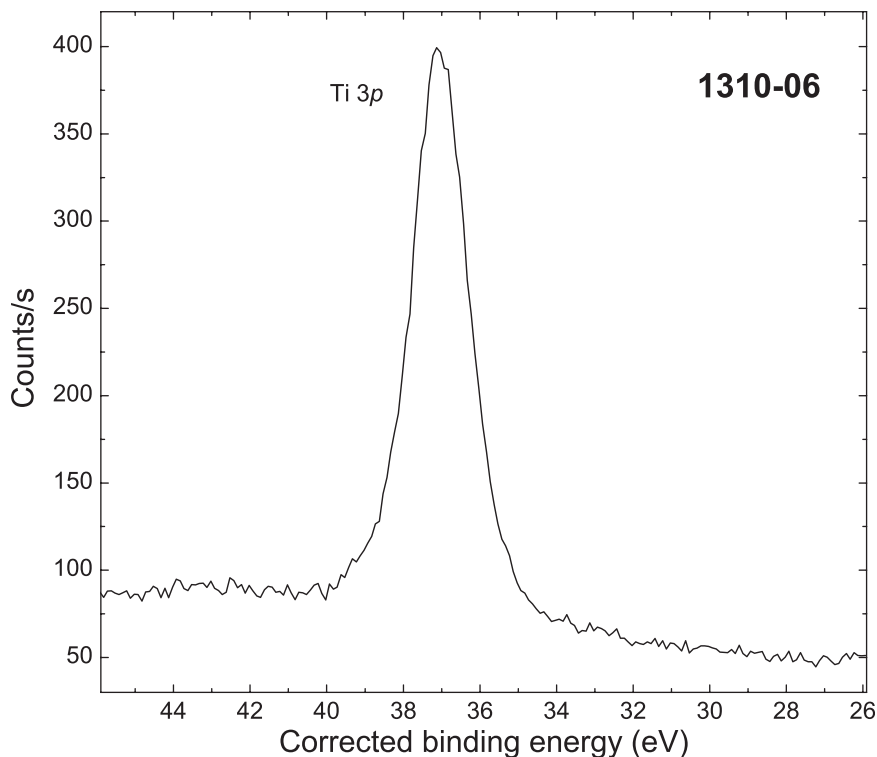
- Accession #: 01310-04
- Host Material: Single crystal SrTiO₃
- Technique: XPS
- Spectral Region: C 1s; Sr 3p

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 4210 s
 Total Elapsed Time: 11577.5 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV



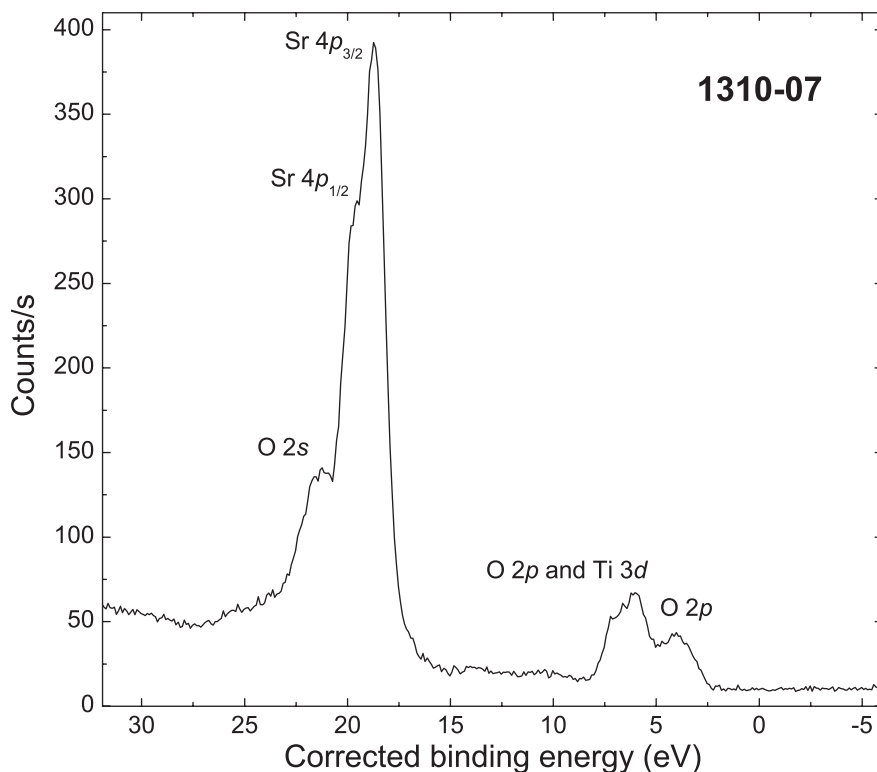
- Accession #: 01310-05
- Host Material: Single crystal SrTiO₃
- Technique: XPS
- Spectral Region: Sr 3d

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 1508 s
 Total Elapsed Time: 4147 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV



- Accession #: 01310-06
- Host Material: Single crystal SrTiO₃
- Technique: XPS
- Spectral Region: Ti 3p

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 2010 s
 Total Elapsed Time: 5527.5 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV



- Accession #: 01310-07
- Host Material: Single crystal SrTiO₃
- Technique: XPS
- Spectral Region: Sr 4p; O 2s; valence band

Instrument: Kratos Axis Ultra
 Excitation Source: Al K_α monochromatic
 Source Energy: 1486.6 eV
 Source Strength: 120 W
 Source Size: 2 mm × 2 mm
 Analyzer Type: spherical sector
 Incident Angle: 54°
 Emission Angle: 0°
 Analyzer Pass Energy: 20 eV
 Analyzer Resolution: 0.3 eV
 Total Signal Accumulation Time: 4762 s
 Total Elapsed Time: 13095.5 s
 Number of Scans: 25
 Effective Detector Width: 4.2 eV