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Fully coupled wellbore-reservoir simulation of supercritical CO<sub>2</sub> injection from fossil fuel power plant for heat mining from geothermal reservoirs

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1 **Fully Coupled Wellbore-Reservoir Simulation of**  
2 **supercritical CO<sub>2</sub> Injection from Fossil fuel power plant for**  
3 **Heat Mining from Geothermal Reservoirs**

4  
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12

13 **Abstract**

14 The concept of injecting **supercritical CO<sub>2</sub> (sCO<sub>2</sub>)** into a geothermal reservoir was  
15 computationally investigated to obtain an insight into the performance of such  
16 system in terms of the benefit of using CO<sub>2</sub> captured from fossil fuel power plants  
17 for geothermal heat mining. A fully coupled wellbore-reservoir system was  
18 simulated considering the flow of pure sCO<sub>2</sub> in an injection well, interaction of sCO<sub>2</sub>  
19 and water in a permeable reservoir, initially filled with water, and the flow of the  
20 two-phase mixture of sCO<sub>2</sub> and water in a production well. A base case simulation  
21 was performed. Results of this simulation indicate that this CO<sub>2</sub> application is  
22 capable of providing a good source of renewable energy. It was found that for the  
23 reservoir section used in this study (0.08 km<sup>3</sup>) about 8-9 MW<sub>th</sub> could be extracted  
24 from the geothermal resource in a steady state fashion, for a lifetime of the wells of  
25 30 years. This is approximately equivalent to 100 MW<sub>th</sub>/km<sup>3</sup>. A sensitivity analysis  
26 of the coupled wellbore-reservoir system provided information on the impact of  
27 certain parameters on the performance of the integrated system. Mass flow rate  
28 and temperature of injected CO<sub>2</sub>, and reservoir permeability have a first order  
29 impact on the pressure management of the reservoir and the amount of heat  
30 mining from the CO<sub>2</sub>-based geothermal reservoir. Additionally, CO<sub>2</sub> injection  
31 temperature has a large effect on the thermosiphon characteristic of this type of  
32 systems.

33

34Keywords: *Supercritical carbon dioxide, geothermal heat mining, Mexico*

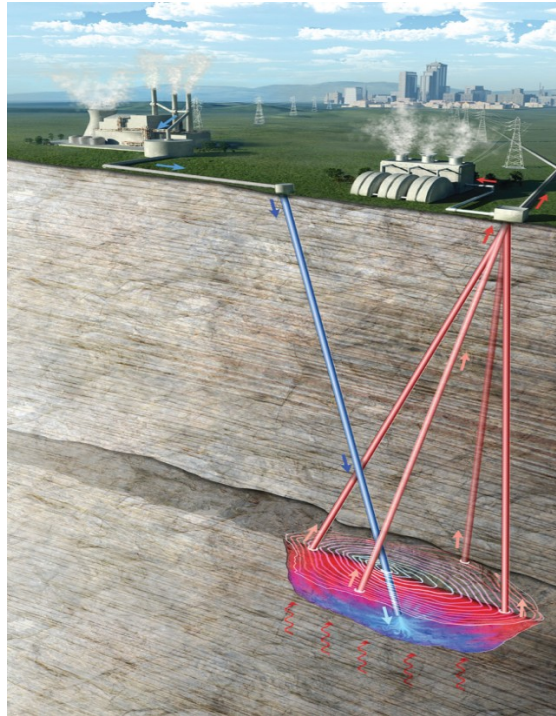
35*geothermal power generation.*

36

## 37**1. Introduction**

38The increased concern with rising atmospheric carbon dioxide (CO<sub>2</sub>), primarily from  
39anthropogenic fossil fuel combustion, has motivated research and development of  
40alternatives to reduce CO<sub>2</sub> emissions from power and industrial plants. CO<sub>2</sub> capture  
41and sequestration in deep saline aquifers has been considered in a series of studies  
42as a means for controlling this major greenhouse gas. One of the options considers  
43injection of supercritical carbon dioxide (sCO<sub>2</sub>) as a working fluid in a naturally high-  
44permeability hydrothermal reservoir [1-4]. In addition to the benefit of sequestering  
45CO<sub>2</sub> in the reservoir, sCO<sub>2</sub> can be used to mine geothermal energy for utilization  
46above the ground. sCO<sub>2</sub>, despite having a smaller mass heat capacity than water;  
47under typical geothermal formation underground conditions, has on average 40% of  
48the viscosity of water and a lower density than water. With those properties,  
49injection of sCO<sub>2</sub> would result in an increased mass flow rate across an equivalent  
50geologic reservoir and, additionally, an augmented buoyancy drive. The injected  
51sCO<sub>2</sub> would form a large subsurface CO<sub>2</sub> plume that would permanently sequester  
52CO<sub>2</sub> underground and also absorb heat from the geothermal reservoir for  
53subsequent utilization at the surface, such as enhanced power generation. An  
54artistic illustration of the concept of sCO<sub>2</sub> injection from fossil fuel power plant for  
55heat mining from geothermal reservoirs is presented in Figure 1.

56



57

58 Figure 1: Illustration of sCO<sub>2</sub> Injection into a Geothermal Reservoir for Heat Mining  
59 and Above-the-Ground Geothermal Energy Utilization.

60

61 This approach is of interest to Mexico, since it combines CO<sub>2</sub> capture and  
62 sequestration, geothermal energy extraction and enhanced electric power  
63 generation. The Mexican government is committed to reduce its carbon footprint  
64 and it has set targets to cut national Green House Gas (GHG) emissions by 22%  
65 below baseline in 2030, equivalent to an increase of emissions by 56% above 1990  
66 levels. Mexico releases approximately 709 million tons of CO<sub>2</sub> annually into the  
67 atmosphere (the world's 12<sup>th</sup> largest carbon emitter), with 30% of this inventory  
68 coming from the electricity generating sector [5]. It is expected that in a future  
69 CO<sub>2</sub>-constrained world, relatively pure CO<sub>2</sub> would be available in large quantities  
70 from Mexican fossil fuel power plant and other energy intensive industrial facilities,  
71 such as chemical process facilities and cement plants. Mexico is conscious of the  
72 need to grow its economy, with the associated need to expand its current power  
73 generating capacity, and its electricity sector currently heavily relies on fossil  
74 energy sources (approximately 75% of the total installed capacity). However,  
75 meeting the forecasted future electricity demand with fossil fuels could increase  
76 Mexico's CO<sub>2</sub> emissions by 230%. Adoption of green technologies by Mexico, such

77as solar, wind, hydro, biomass, and geothermal energy is a necessity to significantly  
78mitigate the global warming impact of an increased power generation base. It is  
79recognized that one of the largest renewable energy sources available to Mexico is  
80geothermal energy. The IGA (International Geothermal Association) has reported  
81that Mexico has estimated geothermal reserves of approximately 8,000 MW<sub>e</sub>,  
82second in the world only to Indonesia. Mexico has a total of eight geothermal power  
83plants, already installed and in construction, totaling a current installed geothermal  
84capacity of 953 MW<sub>e</sub> (fourth in the world) [6]. Additionally, more than 1,000  
85potential geothermal sites have been identified, with a large concentration of  
86medium- and low-enthalpy reservoirs, encompassing Mexico's volcanic region [7].

87

88The concept of using CO<sub>2</sub> as a working fluid to recover heat from geothermal  
89reservoirs has received a good deal of attention of lately. This as-yet-unproven  
90concept relies on replacing water with sCO<sub>2</sub>, which, research results have suggested  
91would be a better working fluid than native reservoir water or brine for geothermal  
92energy extraction [8-17]. Due to its thermofluid behavior under supercritical  
93conditions, a geothermal system utilizing sCO<sub>2</sub> as the subsurface heat exchange  
94fluid, in a naturally porous or fracture permeability-enhanced geologic formation,  
95would provide improved heat extraction for low temperature geothermal resources  
96at shallower subsurfaces below the bedrock. It has been suggested that CO<sub>2</sub>-based  
97geothermal systems could operate at up to 1.5 times the electricity-production  
98efficiency of conventional water-based systems [19]. Additionally, the transport and  
99solubility properties of sCO<sub>2</sub> would also help reduce contamination, scaling and  
100degradation of power equipment found in steam-based geothermal systems.

101

102There is a body of literature presenting variations of the application of sCO<sub>2</sub> for heat  
103mining from geothermal resources. The majority of these studies investigate the  
104feasibility of the concept, framing the results on the impact of reservoir nature,  
105properties and geometry on CO<sub>2</sub> plume formation within the reservoir and its  
106associated heat extraction. One of these sCO<sub>2</sub>-based concepts consists of injecting  
107CO<sub>2</sub> into dry rock or hydrothermal (wet rock) geological formations, where the CO<sub>2</sub>  
108fracture/fill/displaces the native reservoir fluid, mines geothermal heat and is piped  
109back to the surface for electricity production or other applications. Part of the  
110injected CO<sub>2</sub> can be geologically stored. Brown first [8] presented the use of sCO<sub>2</sub> in

111hot dry rock reservoirs in 2000. The work by Brown was based on field testing and  
112demonstrations carried out at the Fenton Hill test site in the Jemez Mountains of  
113North-Central New Mexico. The study by Brown concluded that for a 500 m deep hot  
114dry rock reservoir with an injection pressure of 300 bar, about 100,000 tons of CO<sub>2</sub>  
115per year could be sequestered, in addition to about 50,000 tons of CO<sub>2</sub> available for  
116closed-loop circulation.

117

118Pruess [9, 10] presented the concept of CO<sub>2</sub>-based Enhanced Geothermal Systems  
119(EGS). In this system, sCO<sub>2</sub> would be injected into the hot impermeable rock;  
120opening additional fractures in the reservoir and forming a region prone for CO<sub>2</sub>  
121diffusion in the reservoir. After initial formation of a two phase CO<sub>2</sub>-water mixture in  
122the reservoir, the passage of time will lead to the creation of a reservoir of pure  
123sCO<sub>2</sub>, circulating in closed-loop, while extracting heat and sequestering some CO<sub>2</sub> in  
124the surrounding rock mass. In the simulations performed by Pruess [9], a reservoir  
125thickness of 305 m was used, with reservoir rock temperature of 200°C, injection  
126temperature of 20°C, fracture spacing of 50 m, permeable volume fraction 10%,  
127negligible rock permeability and 50 md fracture permeability, 50% porosity in the  
128permeable domain, and variable reservoir pressures. A five-spot well configuration  
129was modelled, with a two-dimensional and five-point grid of 1,000 m side. All  
130simulations were performed using the TOUGH2 [10] code, augmented with the  
131ECO2N fluid property module. All simulations were performed under CO<sub>2</sub>-only or  
132H<sub>2</sub>O-only systems, with no consideration to mixtures of both fluids, and maintaining  
133the injection and production bottom-well pressures constant. The simulations  
134performed by Pruess [11] conclude that heat extractions from EGS systems can be  
13550 to 100% larger with sCO<sub>2</sub> than with water. The differences become smaller with  
136time, due to the more rapid thermal depletion when using CO<sub>2</sub>. Mass flow rates in  
137the CO<sub>2</sub> system are also larger than for water by factors as high as 3.5. Additional  
138data from the study by Pruess [11] of CO<sub>2</sub>-based EGS estimate that typical fluid loss  
139rates (sequestration rates) would be in the range of 5%, further suggesting about 1  
140kg/sec/MW of sequestered CO<sub>2</sub>. Further work by Spycher [12] on EGS systems  
141concluded that the production of a free aqueous phase form in an EGS operated  
142with CO<sub>2</sub> will occur only after a limited number of years. Spycher added that it is  
143typical to expect a useful life of geothermal reservoirs of about 25-30 years.

144

145 Randolph [14] introduced the concept of CO<sub>2</sub>-Plume Geothermal (CPG) in which  
146s CO<sub>2</sub> is used as the working fluid in a high-permeability, high-porosity geologic  
147reservoir (typically hydrothermal or saline reservoirs) that is overlain by a low-  
148permeability cap rock. The CO<sub>2</sub> displaces the native formation brine in the  
149reservoir, heats up and then is ready for electricity generation at the surface. The  
150sizes of these wet rock reservoirs are typically much larger than those of  
151hydrofractured reservoirs. The simulations performed by Randolph and Saar [14,  
15215] are an extension of the work by Pruess [9] in which the same five-well  
153arrangement and geometry was used and resolved using the TOUGH2 code with the  
154ECO2N fluid property module. Different values of domain permeability were tried  
155by Randolph, with an average value of  $5 \times 10^{-14}$  m<sup>2</sup>. Other simulation parameters  
156include a 20% domain porosity, reservoir pressure of 250 bar, and two reservoir  
157assumed depths and corresponding temperatures of 4 km-150°C and 1 km- 100°C.  
158Heat extraction rates estimated by Randolph for a 25-year average were of 62.6  
159MW for the deep reservoir and 64.1 MW for the shallow reservoir. The results for  
160the CPG systems show that the heat extraction decreases with time as the heat is  
161depleted and the temperature at the production wells decreases with time. The  
162work of Randolph also compares CPG CO<sub>2</sub>-based systems vs. CPG H<sub>2</sub>O-based  
163systems. Cases run at different combinations of initial reservoir pressure and  
164temperature show that for an average 25-year reservoir lifetime, the heat  
165extraction rates for CO<sub>2</sub> are between 2.3 to 3.0 times larger than for the H<sub>2</sub>O-based  
166cases. The corresponding heat extraction ratios of CO<sub>2</sub> to H<sub>2</sub>O are in the range from  
1674.9 to 5.5 [14, 15].

168

169Additional references have reported studies on CO<sub>2</sub> utilization for heat mining  
170considering the geothermal reservoir only. Salimi and Wolf [16] presented another  
171concept for CO<sub>2</sub> utilization in geothermal sites. This concept involves co-injection of  
172CO<sub>2</sub> and water, to prevent drying out and over-pressurizing the reservoirs. Another  
173advantage of this concept is related to the dissolved phase of CO<sub>2</sub> in water, which  
174would avoid confinement of CO<sub>2</sub> to the upper part of the reservoir, decreasing  
175leakage via the cap rock. Self-developed model results were presented that  
176indicate that at CO<sub>2</sub> mole fractions below 0.10, cumulative heat extraction from  
177such system can be as high as 1,000 TJ for 30 years, for a reservoir with dimensions  
178of 250 m thick, 1,600 m long, mean porosity of 0.17, permeability of 21.6 mD and

179initial temperature of 80°C. Buschneck et al. [17] introduced a hybrid two-stage  
180approach to sequester CO<sub>2</sub> and produce geothermal energy in saline, sedimentary  
181formations. In this concept, first brine is extracted from the reservoir to provide  
182pressure relief for CO<sub>2</sub> injection; then, when CO<sub>2</sub> is injected and it reaches the  
183production wells, co-produced treated brine and CO<sub>2</sub> become the working fluids for  
184energy recovery. Three-dimensional model results, using the NUFT code, for  
185reservoirs with temperatures in the 100°C range, report heat extraction rates with  
186this approach as large as 100 MW/m<sup>2</sup>, with combined production flow rates as high  
187as 280 kg/s. A recent study presented by Zhang, et al [18], confirms that sCO<sub>2</sub> has  
188good mobility and heat capacity, and it can be used as an alternative to water for  
189heat recovery from geothermal reservoirs. In the work of Zhang, et al [18] different  
190types of geothermal resources for China were assessed to screen reservoirs suitable  
191for heat mining and geological storage by CO<sub>2</sub> injection, in terms of geological  
192properties, heat characteristics, storage applicability, and development prospects.  
193Reservoir simulations were conducted to analyze the heat extracting capacity and  
194storage efficiency of CO<sub>2</sub> using a simple calculation method. The assessment results  
195show that the recoverable geothermal potential by CO<sub>2</sub> injection in China is around  
1961.55x10<sup>21</sup> J, using Hot Dry Rock (HDR) as the main geothermal resource contributor.  
197The corresponding CO<sub>2</sub> storage capacity is up to 3.53x10<sup>14</sup> kg with the deep saline  
198aquifers accounting for more than 50% of total. It was concluded in this study that  
199CO<sub>2</sub> injection for geothermal production is a more attractive option than pure CO<sub>2</sub>  
200storage due to its higher economic benefits in spite of that many technological and  
201economic issues still need to be solved. Finally, a study by the authors of this paper  
202[19] reports an assessment of the feasibility of using sCO<sub>2</sub> for heat mining for  
203twenty-one geothermal sites in Mexico. This represents the totality of fully  
204characterized geothermal sites in Mexico. The power generation estimate for all the  
205sites is of the order of 1,160 MW<sub>e</sub>, representing a 51.4% additional power  
206generation with sCO<sub>2</sub> than with water. Additionally, it was found from the work  
207reported in Reference 18 that the sum of the Mexican sCO<sub>2</sub>-based systems would be  
208able to sequester, over an expected 30-year life of the reservoirs, approximately 72  
209million tons of CO<sub>2</sub>. Simulations were carried out using the TOUGH2 code, while  
210neglecting the wellbore flows, and with the evaluation of the CO<sub>2</sub> or water heat  
211mining performance, exclusively depending on reservoir properties and behavior.

212



213This paper reports results of a study focused on exploring the injection of CO<sub>2</sub> from  
214fossil fuel power plant for heat mining from geothermal reservoirs in a fully coupled  
215wellbore-reservoir system. The system was simulated in an integrated fashion that  
216considers the flow of pure sCO<sub>2</sub> in an injection well, interaction of sCO<sub>2</sub> and water in  
217a permeable reservoir, initially filled with water, and the flow of the two-phase  
218mixture of sCO<sub>2</sub> and water in a production well. A research version of the  
219T2Well/ECO2N software was used in the simulations. A symmetry section of a five-  
220spot well arrangement was modeled, using a 3-D orthogonal coordinate system.  
221Reservoir properties, fluid flow conditions and well arrangement representative of  
222similar systems in Mexico and of the CO<sub>2</sub> sequestration industry were used in the  
223simulations. The results provide an insight into the behavior of the coupled  
224wellbore-reservoir configuration for the CO<sub>2</sub>/water mixture. A parametric analysis,  
225performed with the integrated model, also provides an indication of a strategy for  
226management of the heat mining process.

227

## 228**2. Modeling Approach**

### 2292.1 Conceptual Model and Gridding

230The coupled wellbore-geothermal reservoir model considered in this study is  
231shown in Figures 2 and 3. The model is a symmetrical box of the five-spot well  
232configuration pattern [9]. For the production wells at the edges of the modeling  
233domain (Figure 3), it is assumed there are hard rock boundaries, so zero flux  
234boundaries were applied. For the box computational domain in the middle, zero flux  
235boundaries were applied due to symmetrical flow pattern. Researchers have  
236reported modeling results for half of the box shown in Figure 3, divided by the  
237diagonal line between the injection and production wellbores [9, 20]. Thus, only 1/8  
238of the basic configuration pattern is considered in the model domain. In this study, a  
2391/4 model of the basic pattern and the wellbore was modeled. By modeling a box,  
240the default mesh generator in the solution solver used in this study was able to be  
241applied directly.

242

243The base case dimensions are shown in Table 1 (see also Figure 2). Reservoir  
244characteristic parameters for the base case simulation are shown in Table 2. The  
245general reservoir parameters were based on measured data from some typical

246reservoirs in Mexico [19]. Initial reservoir conditions as well as injection conditions  
 247are included in Table 3, where the initial pressure is increased by 10 bar for every  
 248100 m of depth. The heat and mass flux were both assumed to be zero on all  
 249reservoir boundaries. Injection wellhead conditions were set at 30 kg/s, 35°C.  
 250Production wellhead conditions were set at 200 bar. Along the length of the  
 251wellbore on top of the reservoir, a linear temperature distribution was assumed,  
 252ranging from 35 to 225°C.

253

254A mesh grid of 30×30 cells along the horizontal direction was used, with a finer  
 255mesh used near the two wellbores. The depth of the reservoir was equally  
 256discretized into 10 layers. This mesh sizing is quite typical for reservoir  
 257discretization. For the wellbores, equal cell length of 50 m was used, resulting in 30  
 258cells for the wellbore part above the reservoir. Together there are 9,060 cells.

259

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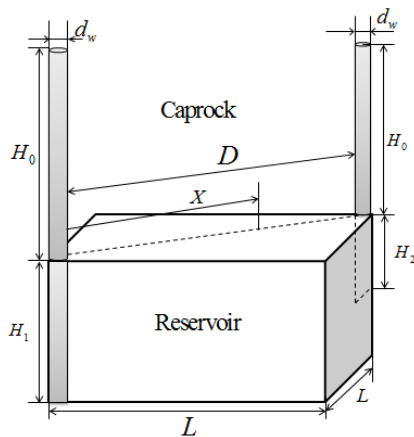


Figure 2. Diagram of Coupled Wellbores and Reservoir (see Variable Definition in Table 1)

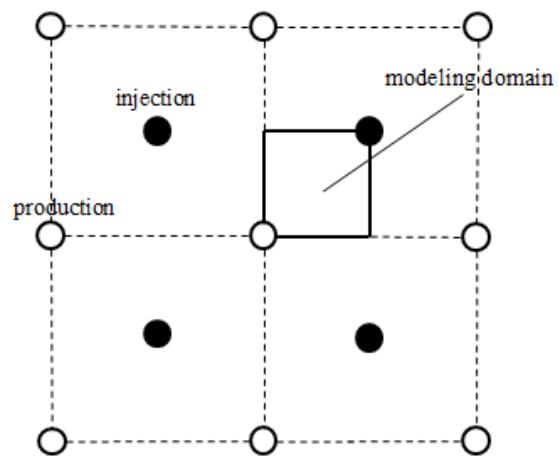


Figure 3. Plane View of the Five-Spot Well Pattern

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263

264

Table 1. Model Dimensions

---

Wellbore Length Through the Caprock Portion,  $H_0$

---

Injection Wellbore Depth in the Reservoir Portion, $H_1$
Production Wellbore Depth in the Reservoir Portion, $H_2$
Length of Reservoir Side, L
Diameter of Injection and Production Wellbores, $d_w$
Diagonal Distance Between the Injection and Production Wellbore, D
Coordinate or Diagonal Distance with Origin at the Injection Well, X

Table 2. Reservoir Characteristic Parameters

Permeability	30 mD
Porosity	0.1
Rock grain density	2600 kg/m <sup>3</sup>
Rock specific heat	920 J/(kg*°C)
Rock thermal conductivity	2.51 W/(m*°C)
Parameters for Relative Permeability [20]	0.01
Residual Gas Saturation	0.65
$m_{VG}$	0.05
Residual Liquid Saturation	1.00
Saturated Liquid Saturation	
Parameters for Capillary Pressure [20]	0.03
Residual Liquid Saturation	0.4118
$m_{VG}$	6.06E-5 Pa <sup>-1</sup>
Alpha	6.40E+7 Pa
Maximum Capillary Pressure	1.00
Saturated Liquid Saturation	

265

266

267

Table 3. Initial Reservoir and Injection Conditions

Reservoir fluid	All water	Injection fluid	CO <sub>2</sub>
Initial Temperature	225°C	Injection temperature	35°C
Initial Pressure	150-200 bar	Injection mass flowrate	30 kg/s

268

269

## 2702.2 Simulator

271The fully coupled wellbore-reservoir simulator T2Well was used in this study.  
272Detailed mathematical descriptions and validations of T2Well are available in [21-  
27324]. This software is commercially available for temperatures up to 100°C. A high  
274temperature research version, provided by Lawrence Berkeley National Laboratory,  
275was used in this study, which is capable to provide CO<sub>2</sub>-brine mixture property  
276values up to 300°C. T2Well incorporates the widely used reservoir-simulation code,  
277TOUGH2 [10]. The equation of state model used was the ECO2N V2.0 [25], which  
278includes a comprehensive description of the thermodynamic and thermo-physical  
279properties of the H<sub>2</sub>O-NaCl-CO<sub>2</sub> system in the pressure and temperature ranges for  
280typical geothermal systems. In T2Well, the wellbore and reservoir are two different  
281subdomains where flow is controlled by different physics. Specifically, viscous flow  
282in the wellbore is governed by the 1-D momentum equation, while 3-D flow in the  
283porous reservoir is governed by a multiphase version of the Darcy's Law. The mass  
284and energy balances are solved together for the two domains, thus capturing the  
285coupled behavior between a wellbore and a reservoir.

286

## 2873. Results and Discussion

### 2883.1 Base Case

289This section presents simulation results for the base case. Figure 4 shows the  
290pressure behavior in the reservoir at different distances, on a diagonal, from the  
291injection to the production well. Simulations were run for a period of 30 years. The  
292pressure traces in Figure 4 correspond to results obtained from the simulations for  
293the top discretized layer beneath the caprock in the reservoir. Initially, the pressure  
294in the reservoir quickly increases over the entire reservoir from the initially 150-200  
295bar to reach a peak around 445 bar. Then the pressure in the reservoir gradually  
296decreases as the gas phase breakthrough the production well (Figure 5). Also a  
297large pressure gradient in the reservoir around the production wellbore can be  
298observed, as indicated by the bottom two pressure traces in Figure 4. The pressure  
299at the production wellhead was set at 200 bar (this boundary condition may be too  
300high during the water production period in a real case; although, this pressure level  
301would be more realistic for a CO<sub>2</sub> breakthrough), and during the first few years,

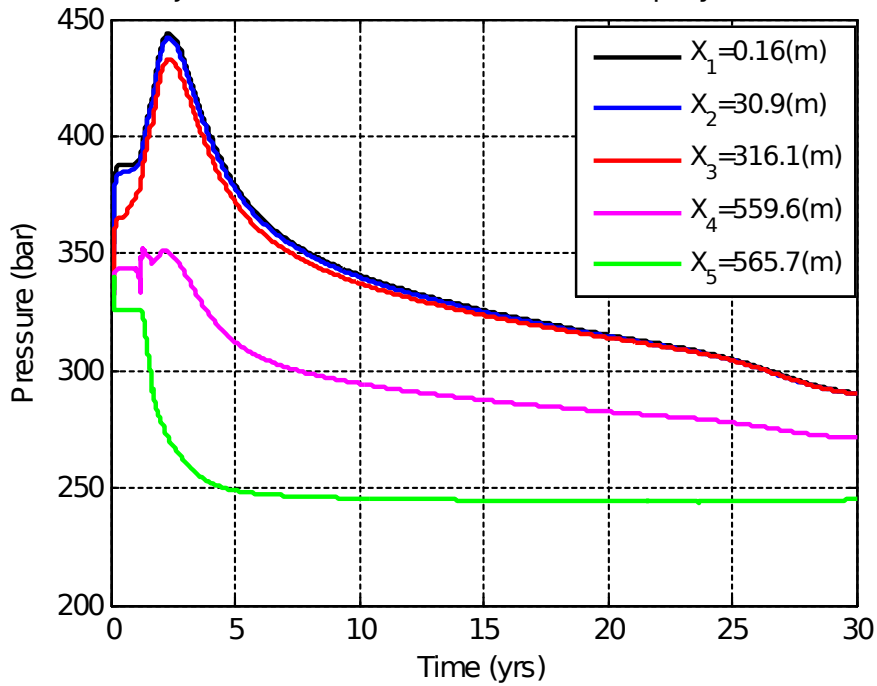
302when the production flow is mostly pure liquid water, the production mass flowrate  
303of the liquid phase goes up to 50 kg/s, and then it drastically decreases to reach a  
304steady flowrate of the order of 30 kg/s (close to the injection flow rate). At the  
305steady production flow conditions, more than 80% of the flow mass fraction is CO<sub>2</sub>,  
306as shown in Figure 5. The CO<sub>2</sub> mass fraction is the flowing fraction (i.e., the ratio of  
307CO<sub>2</sub> mass flow rate over the total mass flow rate. However, during the first few  
308years of injection, there is a pressure buildup in the reservoir due to the large mass  
309flowrate of cold CO<sub>2</sub> being heated in the reservoir. This leads to a large CO<sub>2</sub>  
310compression inside the reservoir, faster than the available reservoir volume can be  
311emptied by the amount of water leaving the reservoir. Pressure spikes inside the  
312reservoir are not desirable since they increase the risk of caprock fracture and  
313potential CO<sub>2</sub> leakage to the surface.

314

315 When the production is CO<sub>2</sub> dominated, the compression effect is greatly  
316reduced, the pressure peak disappears and the flowrate in the production well  
317reaches steady state. This pressure peak behavior is quite unique for CO<sub>2</sub> injection  
318in a reservoir initially filled with water, which will not happen if the reservoir would  
319be initially filled with CO<sub>2</sub>. In a real application, there would be limitations in  
320compressor work that would prevent injection of CO<sub>2</sub> at the target 30 kg/s for the  
321reservoir characteristics used in this study. Strategies would need to be developed  
322to prevent the pressure peak from occurring; i.e., during the first few years of CO<sub>2</sub>  
323injection, a smaller CO<sub>2</sub> injection flowrate could be used, until the CO<sub>2</sub> plume fully  
324develops and CO<sub>2</sub> production dominates in the production well. Another option  
325would be to open the production wellhead to a lower pressure. Then the injection  
326rate could be increased to larger values. Other possible option would be to add  
327reservoir pressure relief through water or brine production wells. This option has  
328been suggested in other references [17], with further reference to water/brine  
329utilization in other production processes, like desalination.

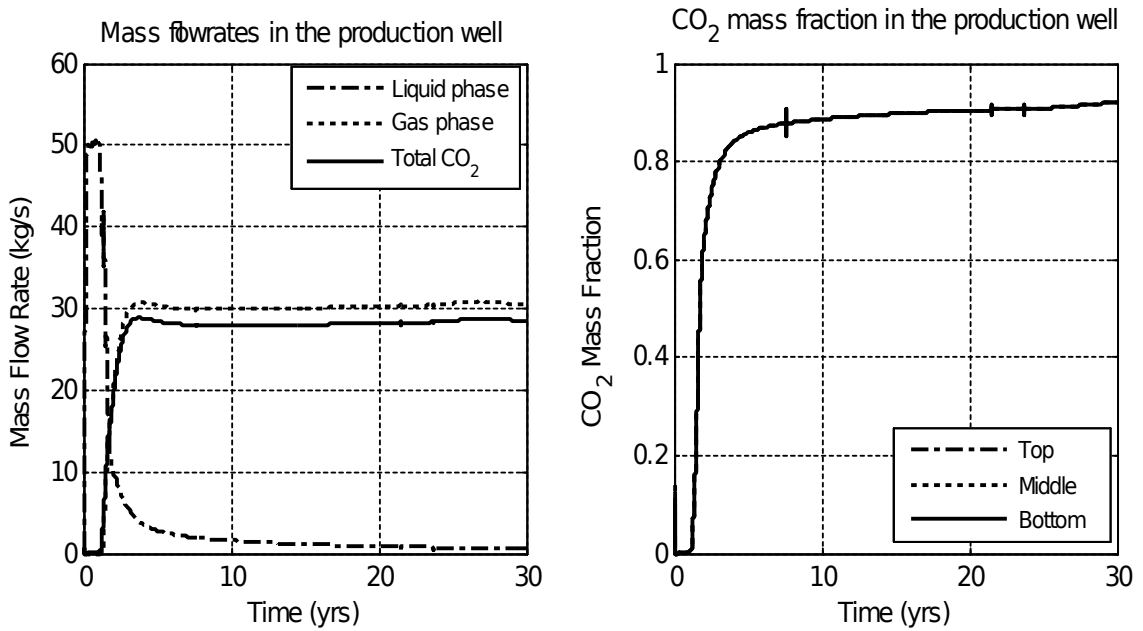
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Pressure trajectories at different locations in the top layer of the reservoir



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332  
333  
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Figure 4. Pressure Traces in the Top Layer of the Reservoir



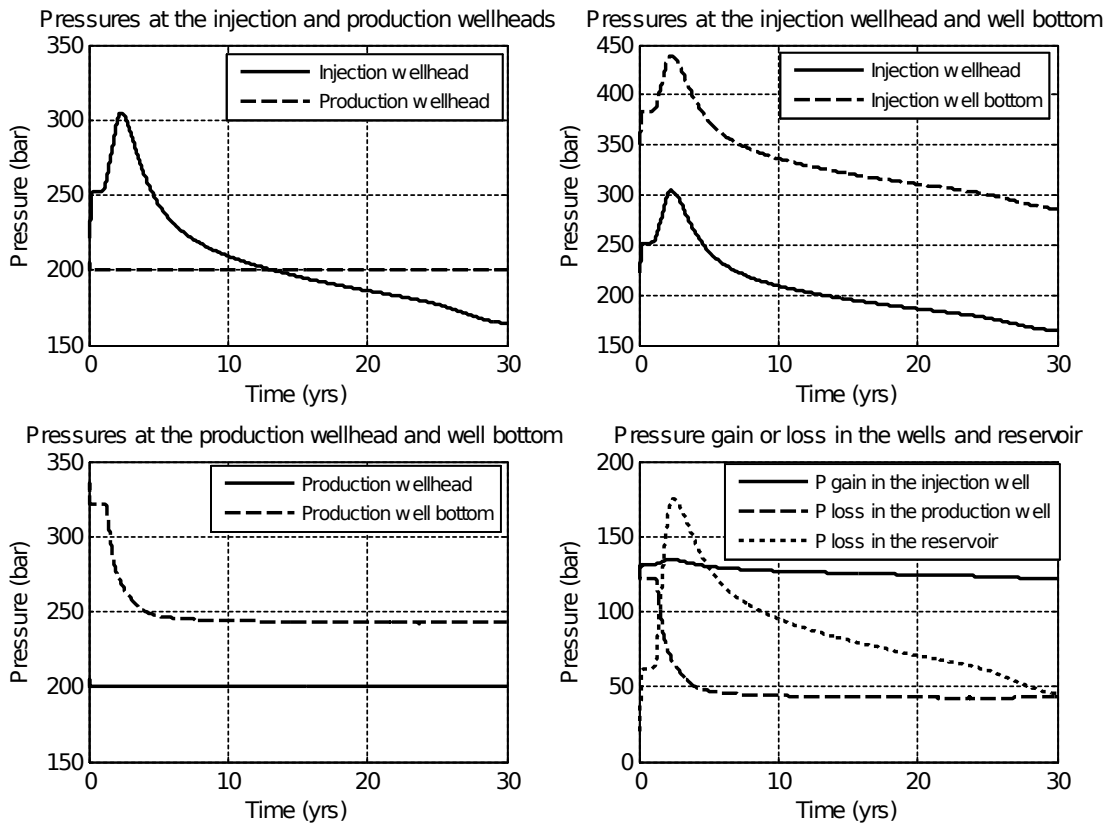
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Figure 5. Mass Flowrates (left) and CO<sub>2</sub> Mass Fraction (right) in the Production Well

340 Figure 6 shows pressures at the injection and production wellhead and well bottom,  
341 as well as the pressure gain or loss in the wells and reservoir. Injection pressure

342gain is defined as the pressure differential from injection wellhead to well bottom.  
 343Production well pressure loss is defined as the pressure differential from production  
 344well bottom to wellhead, while reservoir pressure loss is defined as the change in  
 345pressure in the reservoir from injection well bottom to production well bottom. It  
 346can be seen in Figure 6 that the pressure peak in the reservoir is reflected in the  
 347pressure trace at the injection wellhead. The density difference between the  
 348injection well and the production well is the main cause for the development of the  
 349thermosiphon phenomenon in which the injection wellhead pressure is lower than  
 350the production wellhead pressure. As shown in Figure 6, this happens after about  
 35113 years when the pressure loss through the reservoir significantly drops because of  
 352well development of CO<sub>2</sub> plume in the reservoir.

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Figure 6. Pressures in Injection and Production Wells

358Figure 7 shows temperature plots at the production wellhead and bottom. The  
 359temperature drop between these two locations is almost constant from well bottom

360to head when the production is CO<sub>2</sub> dominated. At about 26 years after the start of  
361injection, the temperature in the reservoir begins to drop and the output  
362temperature at the production wellhead begins to decline. This provides an  
363indication that an optimal well configuration and heat mining strategy are required  
364to exploit geothermal resources with CO<sub>2</sub> in a cost-efficient manner. Figure 8 shows  
365the net energy flow produced from the reservoir. The energy flow is based on the  
366following calculation:

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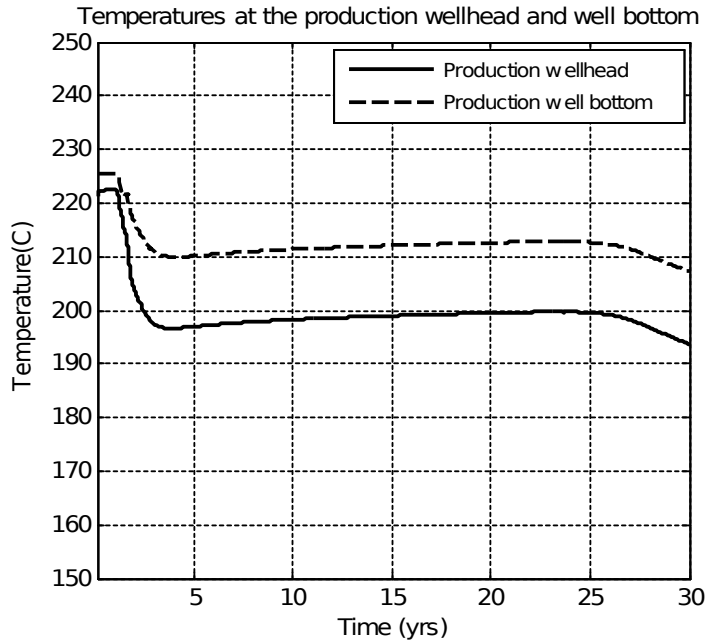
368 
$$\dot{Q} = \dot{m}_{pro}^{CO_2} h(P_{pro}, T_{pro}) - \dot{m}_{inj}^{CO_2} h(P_{inj}, T_{inj}) \quad (3.1)$$

369

370where  $\dot{Q}$  is the energy flow from the production well hot stream,  $\dot{m}_{inj}^{CO_2}$  is the injected  
371CO<sub>2</sub> mass flowrate,  $h(P_{inj}, T_{inj})$  is CO<sub>2</sub> enthalpy at the injection wellhead,  $\dot{m}_{pro}^{CO_2}$  is the  
372pure CO<sub>2</sub> mass flow rate at the production well, and  $h(P_{pro}, T_{pro})$  is CO<sub>2</sub> enthalpy at  
373the production wellhead. The fluid generated at the production well will be a rich  
374CO<sub>2</sub> mixture with some entrained water, with a steady state mass fraction of water  
375of less than 10%. The MW calculation assumes that water is separated before  
376further CO<sub>2</sub> heat utilization. It is estimated that for the reservoir used in this study,  
377about 8-9 MW<sub>th</sub> could be extracted from the geothermal resource for a reservoir  
378section with a volume of 0.08 km<sup>3</sup>, or an approximate equivalent 100 MW<sub>th</sub>/km<sup>3</sup>.  
379During the first two years, the net energy flow is less than zero; this is because  
380there is almost no CO<sub>2</sub> production during the beginning years, while the hot water  
381being produced was not considered to be used to produce energy. For the base  
382case, the total amount of CO<sub>2</sub> being sequestered in the 30-year time lapse is 3.3 Mt  
383in the box domain, or 41.2 kg/m<sup>3</sup>. The total amount of water being extracted was  
3845.4 Mt, or 68.1 kg/m<sup>3</sup>. As the density of water is heavier, more water was  
385extracted.

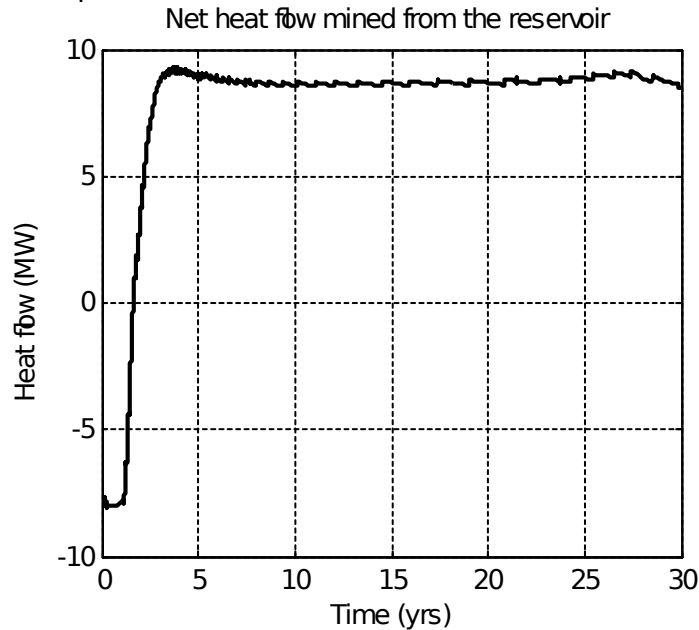
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Figure 7. Temperatures at the Production Wellhead and Well Bottom



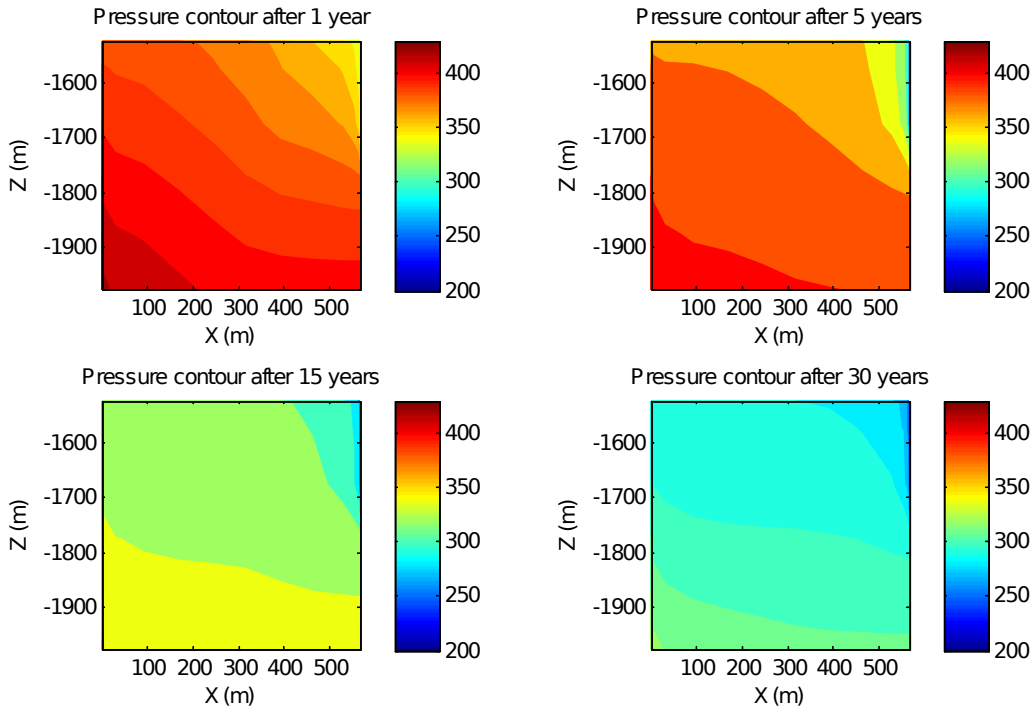
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Figure 8. Net Power produced from the Reservoir

393 Figures 9 to 11 show contours of pressure, temperature and CO<sub>2</sub> saturation in a  
394 diagonal vertical plane between the two wellbores. Results are presented for 1, 5,  
395 15 and 30 years. The highest pressure in the reservoir is located at the injection  
396 well bottom, while the lowest pressure is located at the production well bottom.  
397 The overall the pressure level in the reservoir decreases throughout the 30-year  
398 period. From Figure 10, it can be seen that the temperature in the reservoir drops

399 after 30 years. This is consistent with the contours of Figure 11, which show the  
 400 gradual propagation of CO<sub>2</sub> inside the modeling domain. Figure 11 shows that CO<sub>2</sub>  
 401 concentrates at the bottom of the reservoir, propagating from the injection to the  
 402 production well bottom, which is also consistent with pressure gradients shown in  
 403 Figure 9. In this study it was found that this CO<sub>2</sub> behavior greatly depends on the  
 404 permeability of the reservoir (The CO<sub>2</sub> plume would stay at the top of the reservoir  
 405 if the reservoir permeability was 15mD.). Figure 11 also shows that by Year 30, the  
 406 CO<sub>2</sub> gas plume has diffused to almost occupy the entire reservoir volume. It should  
 407 be noted that the CO<sub>2</sub> saturation values in close to the production well bottom are  
 408 impacted by the resolution in the computational results achieved by the model grid  
 409 size. The results in Figure 11 also suggest that well arrangement, in this CO<sub>2</sub>  
 410 utilization application, would be subject of optimization, in order to provide for a  
 411 system that supplies consistent geothermal energy for the life of the wells. The well  
 412 arrangement should also be subject to constraints oriented to prevent reservoir  
 413 over-pressurization and rapid achievement of CO<sub>2</sub> purity and maximum achievable  
 414 temperature at the production wellhead.

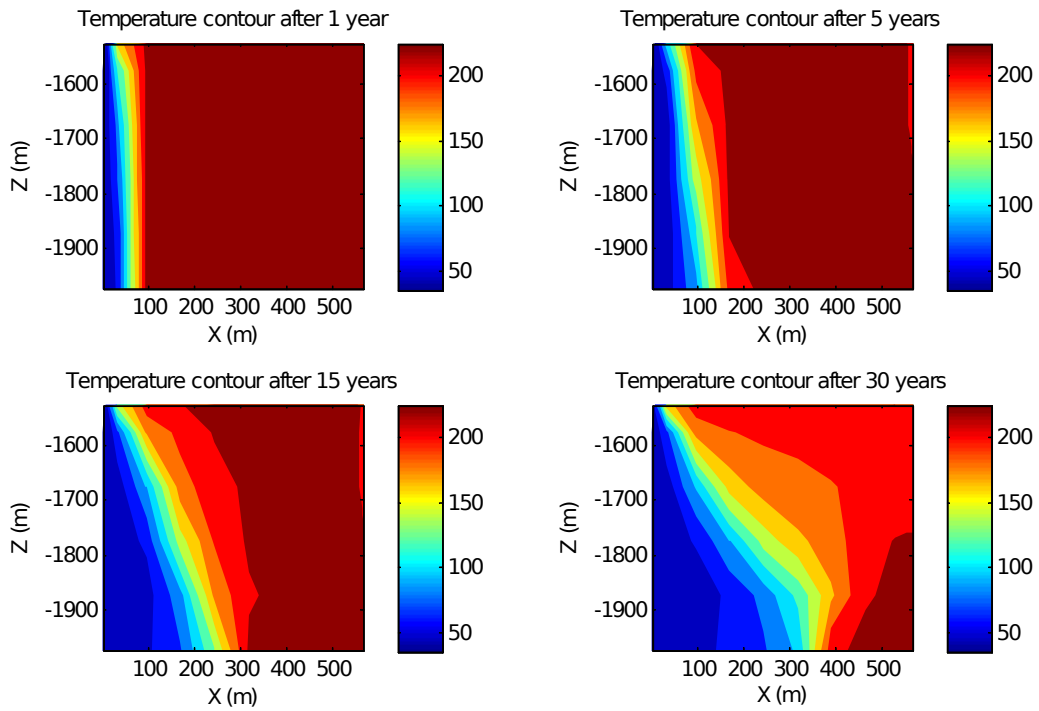
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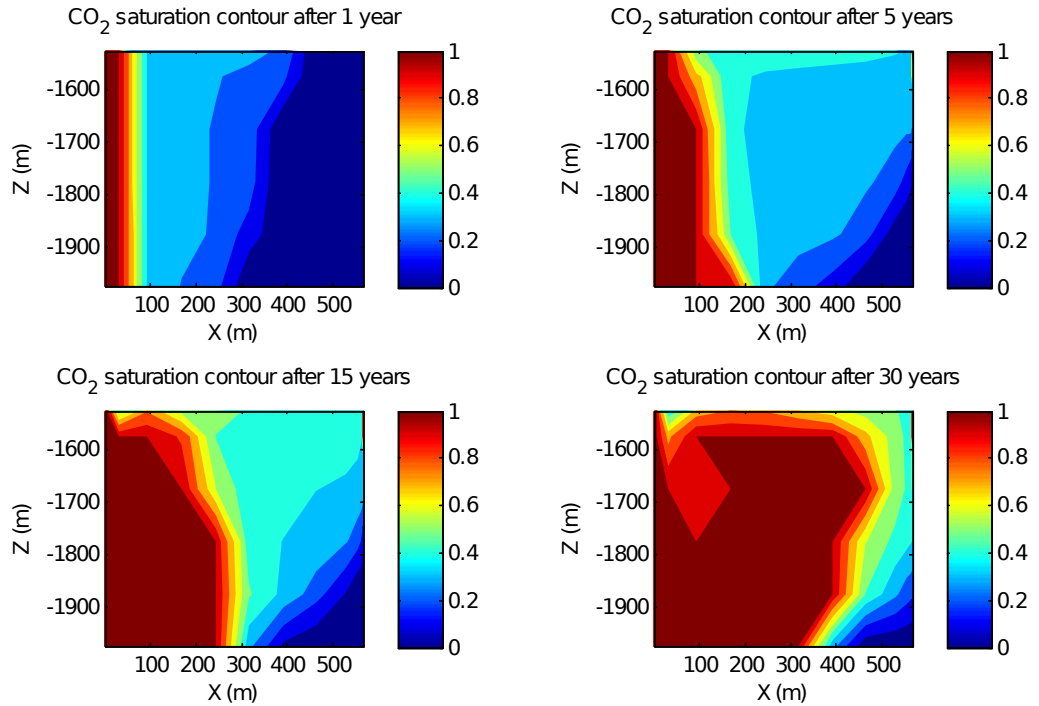
Figure 9. Pressure Contours at the Vertical Diagonal Plane between the two wellbores

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Figure 10. Temperature Contours at the Vertical Diagonal Plane between the two wellbores



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427 Figure 11. CO<sub>2</sub> saturation Contours at the Vertical Diagonal Plane between the two  
428 wellbores

429

### 4303.2 Sensitivity Analysis

431A sensitivity analysis of seven input parameters used in the simulation was  
432performed (see Table 4). Figure 12 shows results of the sensitivity analysis  
433corresponding to variations of the pressure specified in the model formulation at the  
434production wellhead boundary condition. Results are presented in terms of the  
435pressure differential between the injection and production wellhead, CO<sub>2</sub> production  
436temperature, net energy flow (see definition in Section 3.1 Eqn. 3.1) and CO<sub>2</sub> mass  
437fraction (see Figure 12). The specified pressure at the production wellhead has little  
438impact on the thermosiphon effect (production pressure larger than injection  
439pressure) of CO<sub>2</sub> utilization in geothermal reservoirs. Similarly, the impact of the  
440production pressure is negligible on the net energy flow (this is expected because  
441the CO<sub>2</sub> injection rate is fixed) and production stream CO<sub>2</sub> mass fraction. However,  
442the impact of the specified pressure at the production wellhead on the output  
443temperature at the production wellhead is significant. The higher the production  
444pressure, the higher the production temperature. This is mainly because CO<sub>2</sub> is  
445compressible fluid so that the temperature is proportional to the pressure for the  
446same given specific enthalpy.

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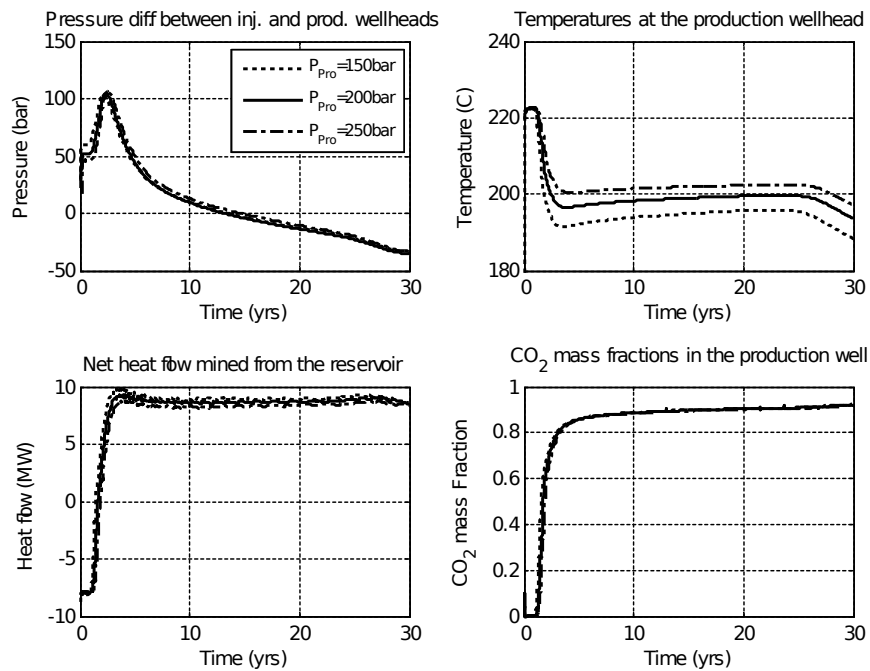
Table 4. Parameters for Sensitivity Analysis

#	Specified pressure at the production wellhead, $P_{Pro}$
1	
#	CO <sub>2</sub> injection temperature, $T_{inj}$
2	
#	Reservoir permeability, $P_m$
3	
#	Injection flowrate, $F_{inj}$
4	
#	Reservoir length, $L$
5	
#	Extension depth of the production wellbore into the 6 reservoir, $H_2$
6	
#	Diameter of the production wellbore, $d_w$
7	

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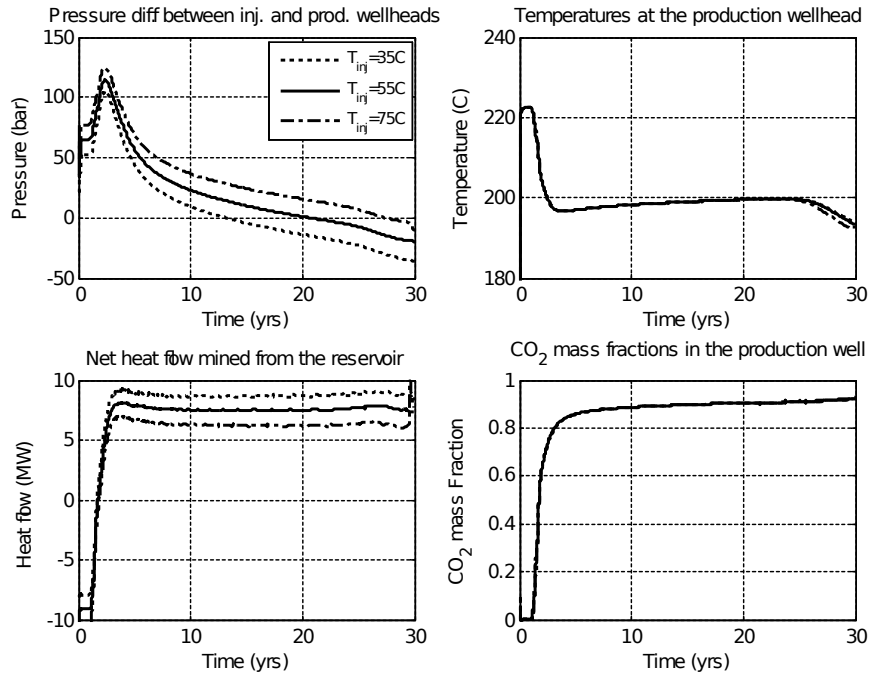
450

451 Figure 13 shows sensitivity analysis results for injection of CO<sub>2</sub> at temperatures from  
452 35°C to 75°C. The output temperature at the production wellhead and  
453 corresponding CO<sub>2</sub> mass fraction at the production well are all unaffected by the  
454 changes in injection temperature over this range. The thermosiphon effect is  
455 slightly affected by this change in injection temperature, because the colder CO<sub>2</sub>  
456 has larger density which results in less injection wellhead pressure needed to  
457 maintain the same injection well bottom pressure to drive the CO<sub>2</sub> through the  
458 reservoir to the production well whose wellhead pressure is fixed. This effect was  
459 already reported by Reference 19. However, it should be noticed that at the 75°C  
460 injection temperature, the thermosiphon effect appears only when close to the end  
461 of 30 year operation. **It was also found that a higher injection temperature would**  
462 **lead to a lower net energy flow. This is mainly because of higher CO<sub>2</sub> enthalpy at a**  
463 **higher injection temperature.**



464

465 Figure 12. Sensitivity Analysis Results for Specified Pressure at the Production  
466 Wellhead

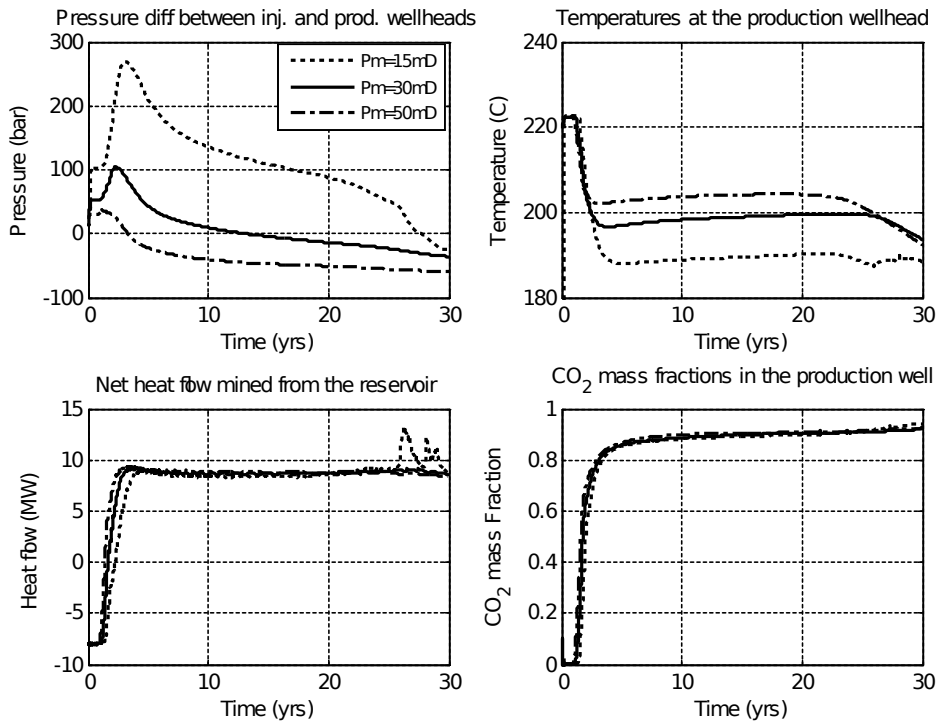


467

468 Figure 13. Sensitivity Analysis Results for Different Injection Temperature

469 Figure 14 shows sensitivity analysis results corresponding to different reservoir  
 470 permeabilities. A range of permeabilities between 15 to 50 mD was used in the  
 471 simulations. The reservoir permeability has a great impact on the pressure  
 472 spike seen in previous results at the beginning of CO<sub>2</sub> plume formation. The pressure  
 473 peak almost disappears when the permeability is 50md. Additionally, it was found  
 474 that when the reservoir permeability is large, higher output temperatures (with the  
 475 same production pressure) can be achieved the production wellhead. However, the  
 476 impact of reservoir permeability on net energy flow is negligible because of the  
 477 small enthalpy variation of CO<sub>2</sub> around 200°C. Figure 15 shows the impact of  
 478 injection flowrate on the parameters of interest. It can be seen from Figure 15 that  
 479 there is a relationship between injection flowrate of CO<sub>2</sub> and the magnitude of the  
 480 pressure spike in the reservoir, further indicating that for certain reservoir  
 481 characteristics and dimensions, there may be a maximum injection flowrate below  
 482 which prevention of pressure spikes is possible. As expected, the CO<sub>2</sub> injection flow  
 483 rate directly controls the production flow rate since a stable injection-production  
 484 state is established after a few years of injection (see Figure 5); thus, resulting in a  
 485 steady net production energy flow. However, larger injection flow rates could result  
 486 in a faster depletion of the geothermal energy reserve in the reservoir.

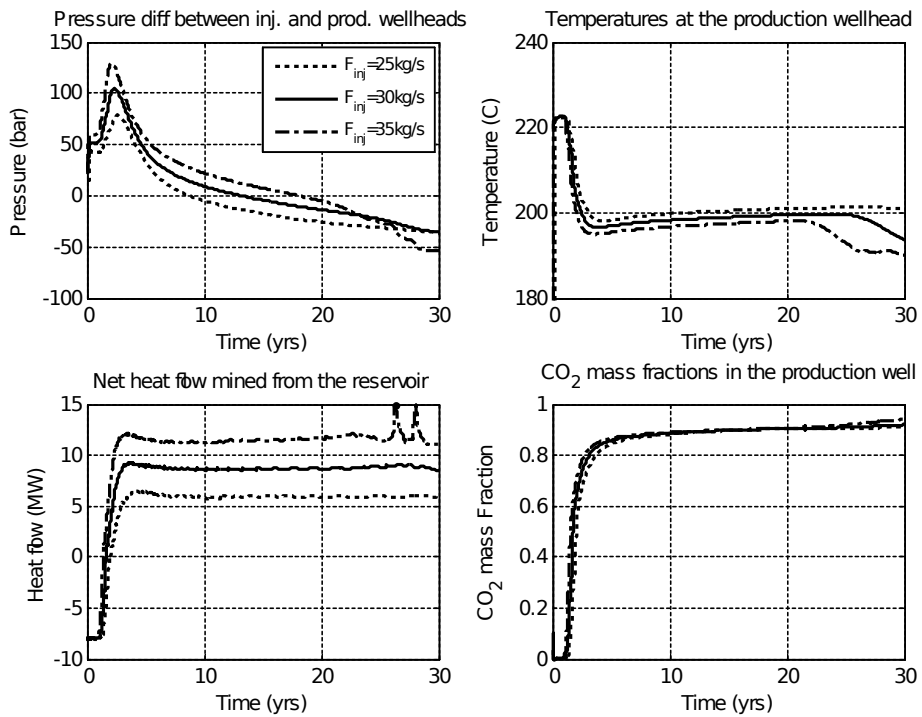
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Figure 14. Sensitivity Analysis Results for Different Reservoir Permeability



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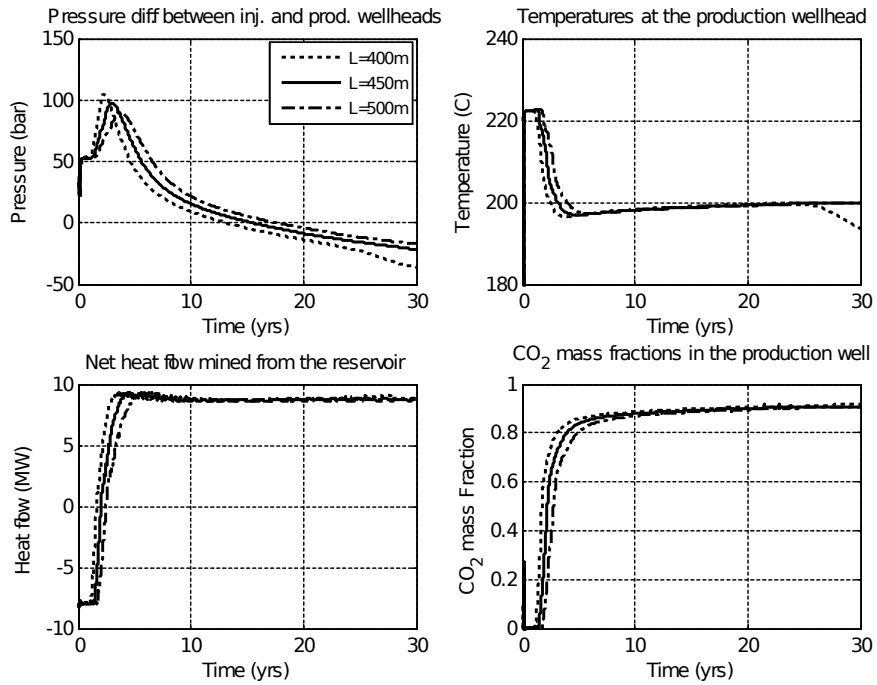
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Figure 15. Sensitivity Analysis Results for Different Injection Flowrate

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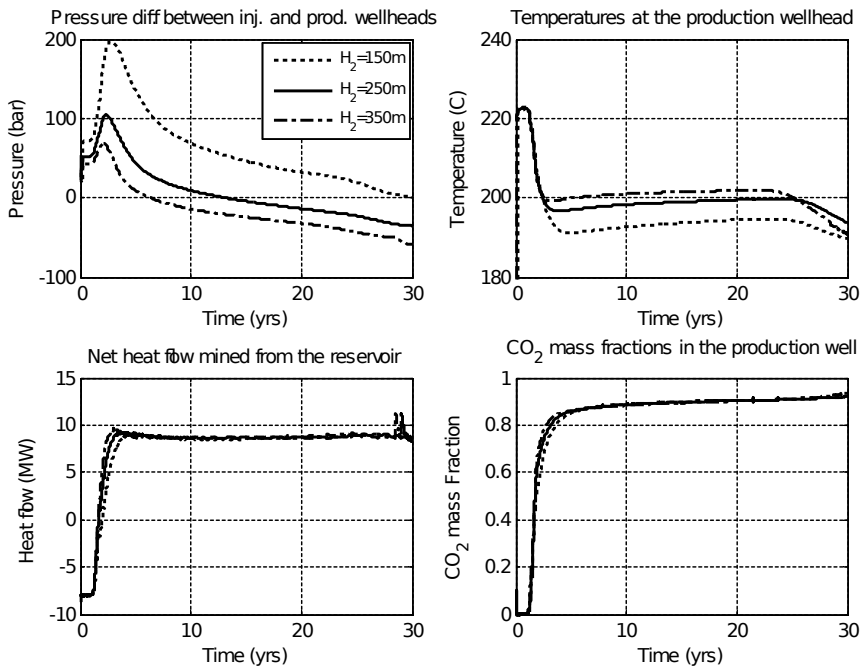
493 Figures 16 to 18 show sensitivity analysis results as function of production well  
494 dimensions and configuration. It can be seen from Figure 16 that larger distances  
495 between the locations of the injection and production wells will ensure that the  
496 output temperature at the production wellhead will not decline toward the end of  
497 the site production life for the same injection flowrate of 30 kg/s. A larger well  
498 spacing also contributes to a smaller peak of reservoir pressure. These are all  
499 because the larger well spacing means larger capacity of the reservoir involved. As  
500 expected, the impact of the distance between both injection and production wells  
501 has a negligible impact of produced net energy flow and mass fraction except the  
502 early transient period. Figure 17 shows sensitivity analysis results on the impact of  
503 extending the depth of the production wellbore into the reservoir. A shallower  
504 production wellbore resulting in larger initial pressure spikes because of its smaller  
505 well-reservoir interface area (i.e., larger resistance to flow. The impact of this  
506 parameter on the produced CO<sub>2</sub> conditions (net energy flow, CO<sub>2</sub> mass fraction and  
507 production temperature) is negligible. It is important to notice that the estimation  
508 of the effect of the arrangement of the wellbore in the reservoirs would not be  
509 possible without a simulation like the one of this study, which couples the wellbores  
510 and reservoir in an integrated fashion. Lastly, Figure 18 shows the impact of the  
511 production wellbore diameter on performance parameters. In the figures, a larger  
512 bore diameter has twice the section area than a smaller diameter wellbore. It was  
513 found that this parameter plays no significant role in the operation of the integrated  
514 system with fixed production pressure and fixed injection rate, except the pressure  
515 difference between injection and production wells.





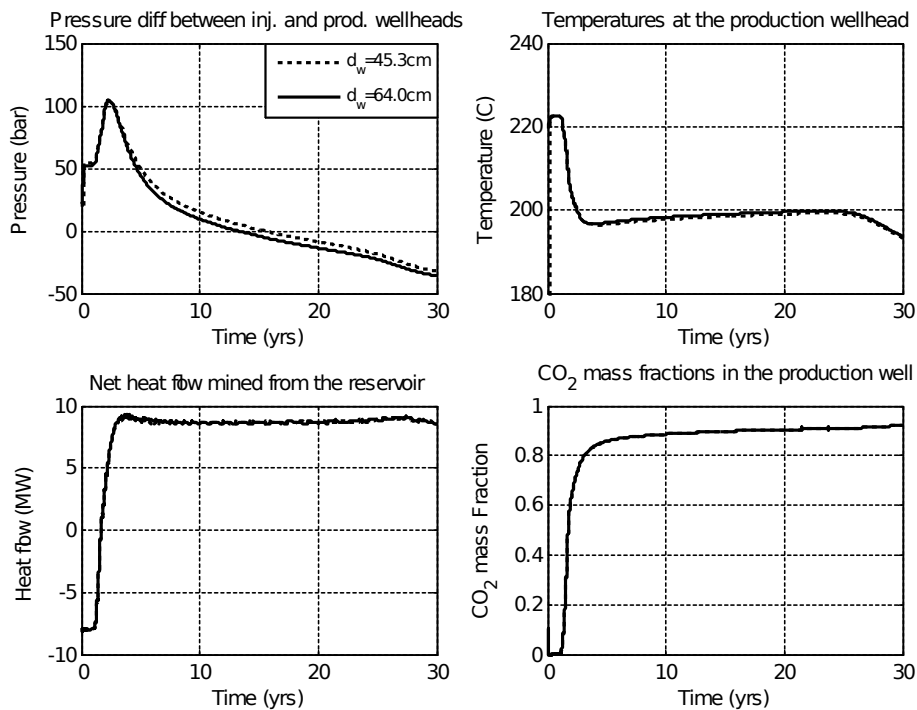
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517 Figure 16. Sensitivity Analysis Results for Different Distance Between Injection and  
518 Production Wells



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520 Figure 17. Sensitivity Analysis Results for Different Extension Depth of the  
521 Production Wellbore into Reservoir



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Figure 18. Sensitivity Analysis Results for Radius of the Production Wellbore

#### 5254. Conclusions

526The concept of injecting  $\text{sCO}_2$  into a geothermal reservoir was investigated  
527computationally to obtain an insight into the performance of such system in terms  
528of the benefit of using  $\text{CO}_2$  captured from fossil fuel power plants for geothermal  
529heat mining. This approach is of interest to Mexico, since it combines  $\text{CO}_2$  capture  
530and sequestration, geothermal energy extraction and enhanced electric power  
531generation. The Mexican government is committed to reduce its carbon footprint  
532and it has set targets to cut national GHG emissions by 22% below baseline in 2030.  
533A fully coupled wellbore-reservoir system was simulated using a research version of  
534the Lawrence Berkeley National Laboratory's T2Well/ECO2N software. The system  
535was simulated in an integrated fashion, considering the flow of pure  $\text{sCO}_2$  in an  
536injection well, interaction of  $\text{sCO}_2$  and water in a permeable reservoir, initially filled  
537with water, and the flow of the two-phase mixture of  $\text{sCO}_2$  and water in a production  
538well. Reservoir properties, fluid flow conditions and well arrangement  
539representative of similar systems in Mexico and of the  $\text{CO}_2$  sequestration industry  
540were used in the simulations.

541

542A base case simulation was first performed. Results of this simulation indicate that,  
543despite a pressure peak that can develop during the initial stage of CO<sub>2</sub> injection,  
544this CO<sub>2</sub> application is capable of providing a good source of renewable energy. It  
545was found that for the reservoir section used in this study (0.08 km<sup>3</sup>) about 8-9 MW<sub>th</sub>  
546could be extracted from the geothermal resource in a steady state fashion, for a  
547lifetime of the wells of 30 years. This is approximately equivalent to 100 MW<sub>th</sub>/km<sup>3</sup>.  
548A sensitivity analysis of the coupled wellbore-reservoir system provided information  
549on the impact of certain parameters, such as injection flow rate and temperature,  
550and well configuration on the performance of the integrated system. It is found that  
551the mass flow rate and temperature of injected CO<sub>2</sub>, and reservoir permeability  
552have a first order impact on the pressure management of the reservoir. Injection  
553flow rate and temperature have additional impact on the amount of heat mining  
554from the CO<sub>2</sub>-based geothermal reservoir. Additionally, CO<sub>2</sub> injection temperature  
555has a large effect on the thermosiphon characteristic of this type of systems, where  
556the pressure differential between production and injection wellhead pressure is  
557positive. From the sensitivity analysis it was evident that system optimization is  
558warranted on these integrated wellbore-reservoir arrangements, to provide a set of  
559optimal operating conditions and well configuration that will result in a cost-  
560effective geothermal resource exploitation.

561

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563

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572code for high temperature simulations.

573

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