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Characteristics of secondhand cannabis smoke from common smoking methods:  
Calibration factor, emission rate, and particle removal rate

Tongke Zhao, Kai-Chung Cheng, Wayne R. Ott, Lance Wallace, Lynn M. Hildemann



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Tongke Zhao: Methodology, Formal analysis, Investigation, Writing - Original Draft, Visualization

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Wayne R Ott: Methodology, Formal analysis, Investigation, Data Curation, Writing - Review & Editing

Lance Wallace: Methodology, Writing - Review & Editing

Lynn M Hildemann: Writing - Review & Editing, Supervision, Funding acquisition

# 1           **Characteristics of Secondhand Cannabis Smoke from Common** 2           **Smoking Methods: Calibration Factor, Emission Rate, and Particle** 3           **Removal Rate**

4           Tongke Zhao, Kai-Chung Cheng, Wayne R Ott, Lance Wallace, Lynn M Hildemann

5                           Civil and Environmental Engineering, Stanford University,

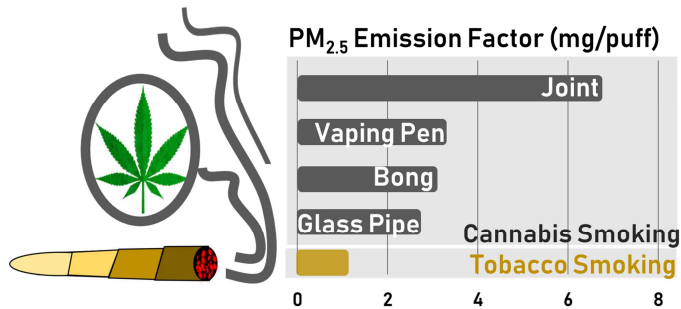
6   Stanford, California 94305, United States

## 7           **Abstract**

8           The widespread legalization of recreational cannabis use raises growing concerns about  
9 exposure to secondhand cannabis smoke (SCS). However, few studies are characterizing fine  
10 particulate matter (PM<sub>2.5</sub>) exposure from SCS. Here, we determined PM<sub>2.5</sub> calibration factors (CFs)  
11 for SCS from four common cannabis consuming methods—glass pipe smoking, joint smoking,  
12 bong smoking, and cannabis pen vaping—for widely used optical monitors (SidePak™ AM510,  
13 TSI Inc., Shoreview, MN) by comparing the monitors with gravimetric mass measurements. We  
14 furthermore investigated the emission rate, particle size distribution, and particle removal rate of  
15 SCS. The CFs of SidePak PM<sub>2.5</sub> monitors measuring the four types of SCS were 0.31 (SE=0.02),  
16 0.39 (SE=0.02), 0.40 (SE=0.01), and 0.44 (SE=0.03), respectively. The arithmetic mean emission  
17 rates of the four SCS sources were ~2-6 times that of secondhand tobacco smoke (STS) on a per-  
18 puff basis. The fresh SCS (1-min after smoking) showed a bimodal size distribution—one mode  
19 located at ~380-420 nm and another at ~800-840 nm. Under low-ventilation conditions, the indoor  
20 removal rates of SCS from cannabis vaping were slightly higher (i.e., 0.61-0.77/h) than those for the  
21 other three sources (i.e., 0.35-0.53/h).

22 **Key Words:** *cannabis; marijuana; particulate matter; calibration factor; size distribution; emission*  
 23 *rate; removal rate; PM<sub>2.5</sub>; aerosols;*

## 24 Graphical Abstract



25

## 26 1. Introduction

27 With the expanding worldwide legality of adult recreational cannabis use (Shi et al., 2019), the  
 28 popularity of consuming cannabis continues to grow. As in June of 2019, eleven states in the United  
 29 States had legalized recreational cannabis use, and Canada and Uruguay have legalized cannabis  
 30 nationwide (Wikipedia, 2019). Studies showed that in legalized states, cannabis use increased in  
 31 both over and under 21-year-old people (Kerr et al., 2017; Parnes et al., 2019; Parnes et al., 2018).  
 32 Also, the legalization of cannabis use stimulates the growth of cannabis dispensaries, which tends to  
 33 increase the consuming population (Dills et al., 2016).

34 When smoking or vaping cannabis, the high-temperature combustion or vaporization process  
 35 produces high levels of fine particulate matter (PM<sub>2.5</sub>) emissions, which is well-known to cause  
 36 respiratory and cardiovascular diseases (Dominici et al., 2006). This makes secondhand cannabis  
 37 smoke (SCS) an emerging air quality and health concern (Gates et al., 2014). The delta-9-  
 38 tetrahydrocannabinol (THC) in cannabis, which may cause psychosis-like effects (D'Souza et al.,  
 39 2004), is emitted into the air, leading to passive exposure (Balducci et al., 2009). It has been found  
 40 that children living in homes with indoor cannabis smoking had 83% higher odds of adverse health

41 outcomes compared to children in homes with no indoor cannabis smoking (Posis et al., 2019).

42 Cannabis smoking methods are varied—including glass pipe, joint, and bong smoking, as well  
43 as vaping—making the characteristics and exposure assessment of SCS complex. Glass pipe  
44 smoking is a combustion particle source. The smoker lights up cannabis buds in the bowl of a glass  
45 pipe and gives a puff. The joint is a pre-rolled cigarette filled with loose cannabis leaves. Unlike  
46 regular commercial cigarettes, which are usually the same length and circumference, joints vary in  
47 size. A bong, also known as a water pipe, burns cannabis buds in the bowl and filters the smoke by  
48 water in the glass reservoir before the smoke is inhaled. A cannabis vaping pen is similar to an e-  
49 cigarette, the emission process of which involves heating and subsequent evaporation and  
50 condensation of cannabis vape liquid (Giroud et al., 2015). Unlike the e-liquid in an e-cigarette,  
51 which contains mostly volatile compounds like propylene glycol or glycerol (Zhao et al., 2017;  
52 Zhao et al., 2016), the cannabis vape liquid usually contains THC and cannabinoid (CBD)  
53 concentrates, vitamin E acetate, triglyceride oils, terpenes and relatively small amounts of propylene  
54 glycol and glycerol (Blount et al., 2020; Giroud et al., 2015), which are more viscous and less  
55 volatile than e-liquid.

56 There is little research about the characteristics of or exposure to fine particulate matter ( $PM_{2.5}$ )  
57 in SCS. It is known that optical  $PM_{2.5}$  monitor output depends on the optical and physical  
58 characteristics of particles, including density, refractive index, and size distribution, which vary  
59 with  $PM_{2.5}$  sources (Hinds, 2012). It is crucial that validated calibration factors (CFs) be determined  
60 for different aerosol mixtures of interest for a research-grade monitor, which then can produce  
61 accurate measurements of  $PM_{2.5}$  mass concentrations. Health-based air quality standards are written  
62 in units of  $\mu\text{g}/\text{m}^3$ , increasing the importance of precise measurements of  $PM_{2.5}$  mass concentrations  
63 for evaluating health risks. The SidePak (AM510, TSI Inc., Shoreview, MN) is one of the most  
64 popular research-grade, small, and portable  $PM_{2.5}$  real-time monitors on the market. The CFs of

65 SidePaks have been developed and published for aerosols from a variety of particle sources (e.g.,  
66 water pipes, cigarettes, stick incense, fireplace emissions, wildfires, candles, burning toast, etc.)  
67 (Dacunto et al., 2013; Jiang et al., 2011; Travers et al., 2018). Moreover, a number of published  
68 studies have used SidePak monitors to measure  $PM_{2.5}$  mass concentrations from STS, taking  
69 advantage of the instrument's portability, 8-hour battery life, data logging capability, and validated  
70 CFs (Acevedo & Bolton et al., 2014; Cho et al., 2014; Klepeis et al., 2003; Klepeis et al., 2016; Ott et  
71 al., 2017; Repace et al., 2011). No previously published studies have estimated the SidePak CFs for  
72  $PM_{2.5}$  from cannabis sources. Therefore, to accurately assess the  $PM_{2.5}$  exposure from SCS, we first  
73 focused on the CFs of SCS from four cannabis smoking methods. Using the results presented in this  
74 paper, we can add cannabis joints, bongs, glass pipe, and vaping to the long list of sources for which  
75 SidePak CFs have been developed for estimating mass concentrations.

76 Since gravimetric filter sampling is accurate but time-consuming, we also used a Piezobalance  
77 (also known as a piezoelectric microbalance, Model 8511, TSI Inc., Shoreview, MN), which is a  
78 real-time, portable gravimetric PM measurement device. A Piezobalance fitted with a  $PM_{2.5}$   
79 impactor collects  $PM_{2.5}$  on a crystal, which oscillates in an electric circuit at a certain resonant  
80 frequency. As the mass of the crystal changes when  $PM_{2.5}$  is deposited on it, the resonant frequency  
81 changes and this change is proportional to the mass of deposited  $PM_{2.5}$  (Sem et al., 1977). In this  
82 study, we investigate the feasibility of using the Piezobalance to measure the  $PM_{2.5}$  concentrations  
83 of SCS accurately.

84 This study, for the first time, characterizes the CFs, emission rates (*mg/puff*), particle size  
85 distributions, and removal rates of SCS from four cannabis smoking methods: glass pipe smoking,  
86 joint smoking, bong smoking, and cannabis vaping. We compare the emission rates of SCS with  
87 that of secondhand tobacco smoke (STS). The particle size distribution determines the deposition in  
88 human respiratory tracts and affects the removal on room surfaces (Nazaroff and Klepeis, 2003).

89 Thus, these results can serve as input for assessing exposure to SCS and for studying the impacts  
 90 and fate of SCS indoors.





## 91 2. Method and Materials

### 92 2.1 Participant

93 A habitual user of cannabis and tobacco, who consumes cannabis in multiple ways, was  
 94 recruited to help generate SCS. The study protocol was accepted by the participant, and a signed  
 95 consent form was obtained before the experiments. The cannabis materials and consuming devices  
 96 used in this study were provided by the participant. The study protocol was approved by the  
 97 Institutional Review Board at Stanford University.

### 98 2.2 Cannabis Devices and Materials

99 **Table 1.** Pictures of the smoking devices and the THC and CBD proportions in the corresponding  
 100 cannabis materials used in this study.

Device	Picture	Cannabis Materials*	
		THC Proportion (%)	CBD Proportion (%)
Glass Pipe		16.9	0.08
Joint		5.86	16.57
Bong		16.9	0.08
Vaping Pen		3.51	69.8

101 \*Proportions as reported on the packaging labels.

102 Four cannabis consuming methods were investigated in this study: glass pipe smoking, joint

103 smoking, bong smoking, and cannabis pen vaping. Pictures of the devices and the THC and CBD  
104 proportions in the corresponding cannabis materials are shown in **Table 1**. The glass pipe was 10  
105 cm in length and was equipped with a 2 cm diameter bowl. On each use, the habitual smoker put ~  
106 0.2 g of cannabis buds into the bowl before lighting it. The mass of cannabis buds was weighed on a  
107 microbalance (Model XPR6UD5, Mettler Toledo Inc., Columbus, OH). The joint was 64 mm in  
108 length, and its larger end was 8 mm in diameter while its smaller end, which was the mouthpiece,  
109 was 4 mm in diameter. The weight of each joint was ~0.8 g. The bong was 18 cm tall, and its  
110 reservoir was filled with ~100 mL water. On each bong use, the smoker placed ~ 0.2 g of cannabis  
111 buds into the bong's 2 cm diameter bowl. The battery-powered vaping pen was 8.7 cm long and 1.1  
112 cm in diameter, and it was attached to a cannabis cartridge labeled "18:1 CBD:THC". On each use,  
113 the participant selected the *High Heat* setting and pre-heated the vaping pen for about five seconds  
114 prior to using it. To compare the emission rates and removal rates of SCS with those of secondhand  
115 tobacco smoke (STS), we also included cigarette smoking in this study. The cigarettes were regular  
116 Marlboro cigarettes.

### 117 **2.3 Real-time Particle Measurements**

118 We used six SidePak monitors and two Piezobalances to measure the real-time  $PM_{2.5}$   
119 concentrations with a data logging time interval of 1-min. Before each experiment, we cleaned and  
120 greased the impactor inside each SidePak as recommended in the manual, and we zeroed each  
121 SidePak by attaching its inlet to a high-efficiency air filter. Before each experiment, the crystal in  
122 each Piezobalance was cleaned by applying a detergent to a sponge that was placed in direct contact  
123 with the crystal followed by a second sponge with distilled water to clean off the detergent. The  
124 sampling flow rate was verified to be 1.0 L/min by a flow meter (Sensidyne, Clearwater, FL, USA).

125 The particle number concentrations measured by an optical particle sizer (OPS 3330, TSI Inc.,  
126 Shoreview, MN) ranged from 0.3 to 10  $\mu\text{m}$ . The data logging interval was 1 min. We captured the



127 particle size distributions at 1, 30, and 90 mins after the smoking activity to examine the dynamics  
128 of SCS. The number of modes for each particle size distribution was determined via visual  
129 inspection. Then, we divided the spectrum into one or two sub-groups according to the number of  
130 modes. Each sub-group was fitted with a lognormal distribution by using Sigmaplot 12.5 lognormal  
131 three-parameter fitting procedure (Zhu et al., 2002), yielding the count median diameter (CMD) and  
132 geometric standard deviation (GSD) for each particle size distribution mode.

#### 133 **2.4 Gravimetric Sampling Method**

134 Two custom-built cyclones (John and Reischl, 1980) were used to provide a 50% cut-size of  
135 2.5  $\mu\text{m}$  and to collect particles less than or equal to 2.5  $\mu\text{m}$  on 47 mm PTFE filters (2  $\mu\text{m}$  pore size,  
136 Pall Corp., Ann Arbor, MI). Each pump's sampling flow rate was ~18-20 L/min as designated by  
137 each cyclone and calibrated by a flow meter (Gilian Instrument Corp., West Caldwell, NJ). The  
138 filters were conditioned for 24-48 hours in an environmental chamber with a constant temperature  
139 of 24 °C and relative humidity (RH) within 30-40%, allowing equilibrium of semi-volatiles. Filters  
140 then weighed by a microbalance (Model XPR6UD5, Mettler Toledo Inc., Columbus, OH) with a  
141 readability of 0.5  $\mu\text{g}$  before and after each sampling. Each filter was usually measured three times,  
142 and the absolute error was no more than 3  $\mu\text{g}$ . The range of the sampling  $\text{PM}_{2.5}$  concentrations was  
143 from 50 to 1500  $\mu\text{g}/\text{m}^3$ . The sampling time was pre-estimated to aim for a collected mass of more  
144 than 100  $\mu\text{g}$  on the filter. During a typical ~2.5 hours decay period in each experiment, we collected  
145 two or three filter samples. Therefore, when we sampled at a higher concentration range, the  
146 sampling time length was shorter. Certainly, when we sampled at a lower concentration range, the  
147 sampling time length was longer. An example of a sampling period, including time lengths,  
148 concentrations ranges, and mass collected on the filter, is shown in the **Supporting Information**  
149 **Table S1**. Finally, the collected mass on the filter, sampling time length, and pump flow rate were  
150 used to determine the mean  $\text{PM}_{2.5}$  mass concentration during a sampling period.

## 151 2.5 Experimental Setting and Protocol

152 The study was conducted in a car that was used as an air chamber and parked in an attached  
153 garage in Redwood City, CA. The car was a Honda Element 2006, with no carpet covering. The  
154 second-row seats were removed to provide more space. The volume of the cabin was measured as  
155  $6.5 \text{ m}^3$  ( $SD=0.3 \text{ m}^3$ ,  $n=5$ ) by using sulfur hexafluoride ( $\text{SF}_6$ ) as the tracer gas. The measurement  
156 method with a diagram of the setup can be found in the **Supporting Information** and its **Figure S1**.  
157 The experiments were conducted from Dec. 2018 to Mar. 2019, when the indoor temperature and  
158 RH were within the range of 15 to 22 °C and 30 to 70%, respectively.

159 In the car chamber, six SidePak monitors were placed on the backseat behind the driver's seat.  
160 The Piezobalances, the OPS 3330, and the gravimetric sampling instruments were placed outside  
161 the car, and each was connected to the car with *Tygon*<sup>TM</sup> tubing. The intake ends of the tubing were  
162 collocated with the SidePak sampling tubing inside the car. A small battery-powered fan with an  
163 11-cm diameter blade was used inside the car to help mix the air.

164 Before each experiment, the car's doors and windows, as well as the garage's doors were  
165 opened to flush the chamber with ambient air. After that, the doors and windows were closed again,  
166 and we started the instruments and measured the background concentrations for at least 10 minutes.  
167 Then the smoker entered the car and sat on the driver's seat. The smoker smoked cannabis by one of  
168 the four methods and gave one to three puffs *ad libitum*. As observed, during a puff, the inhaling  
169 time was ~ 2 seconds, and the exhaling time was ~2 to 5 seconds. After that, the smoker exited the  
170 car promptly, and the car doors remained closed. We extinguished the cannabis combustion with  
171 water immediately to minimize the potential of increasing the  $\text{PM}_{2.5}$  background in the garage.  
172 During the experiments, the garage doors remained closed.

173 By recruiting an experienced smoker to consume cannabis and tobacco *ad libitum*, this work is

174 intended to characterize SCS in a realistic situation. We did not provide the participant with any  
175 cannabis materials, smoking devices, or research cigarettes. Nor did we introduce smoking sensors  
176 to measure the puff topography and the heating temperature as our participant believed they would  
177 interfere with his/her consuming behaviors.

178 SidePak's data showed the background is not larger than  $5 \mu\text{g}/\text{m}^3$ . As Jiang et al.(2011)  
179 showed, the calibration factor of SidePaks measuring ambient particles varied from 0.66 to 0.98.  
180 Therefore, we can reasonably assume that the actual  $\text{PM}_{2.5}$  background concentration is lower than  
181  $5 \mu\text{g}/\text{m}^3$ .

182 In total, we conducted 28 experiments, including eight on glass pipe smoking, five on joint  
183 smoking, six on bong smoking, six on cannabis pen vaping, and three on tobacco cigarette smoking.

## 184 **2.6 Data Processing**

185 When calculating CFs for SidePaks, we compared the arithmetic mean of six SidePak-  
186 measured concentrations over a certain period of sampling time (i.e. 10-90 min) with the arithmetic  
187 mean of two masses on the filters. Then we plotted the gravimetric concentrations on the  $y$ -axis, the  
188 SidePak readings on the  $x$ -axis, and a regression line was forced through zero. The slope is the CF.  
189 The agreement among SidePak monitors was evaluated by their absolute percent error (APE) and  
190 the method can be found in Jiang et al., (2011). A CF of each SidePak was also obtained by  
191 comparing each SidePak's measured concentrations with the gravimetric concentrations by the  
192 same method. We also used the same method to determine CFs for the Piezobalances.

193 In this study, we determined the emission rate as the mass of emitted  $\text{PM}_{2.5}$  per puff ( $\text{mg}/\text{puff}$ ).  
194 Examples of SCS  $\text{PM}_{2.5}$  concentration decay curves from four cannabis sources and cigarette  
195 smoking are shown in the **Supporting Information Figure S2**. The  $\text{PM}_{2.5}$  time-series  
196 concentrations can be fit by an exponential decay equation with high  $R^2$  values ( $\geq 0.98$ ), which

197 indicated the well-mixed state of SCS in the car. As shown in Equation (1),  $g$  is the emission rate  
198 ( $mg/puff$ ),  $V$  ( $m^3$ ) is the mixing volume of the car chamber, and  $n$  is the number of puffs per each  
199 experiment,  $C_{max}$  ( $mg/m^3$ ) is the maximum  $PM_{2.5}$  concentration at the end of each emission period  
200 and is calculated by extrapolating the exponential decay line back to the time when the emission  
201 ended (Jiang et al., 2011; Ott et al., 2008).

$$g = \frac{C_{max} \times V}{n} \quad (1)$$

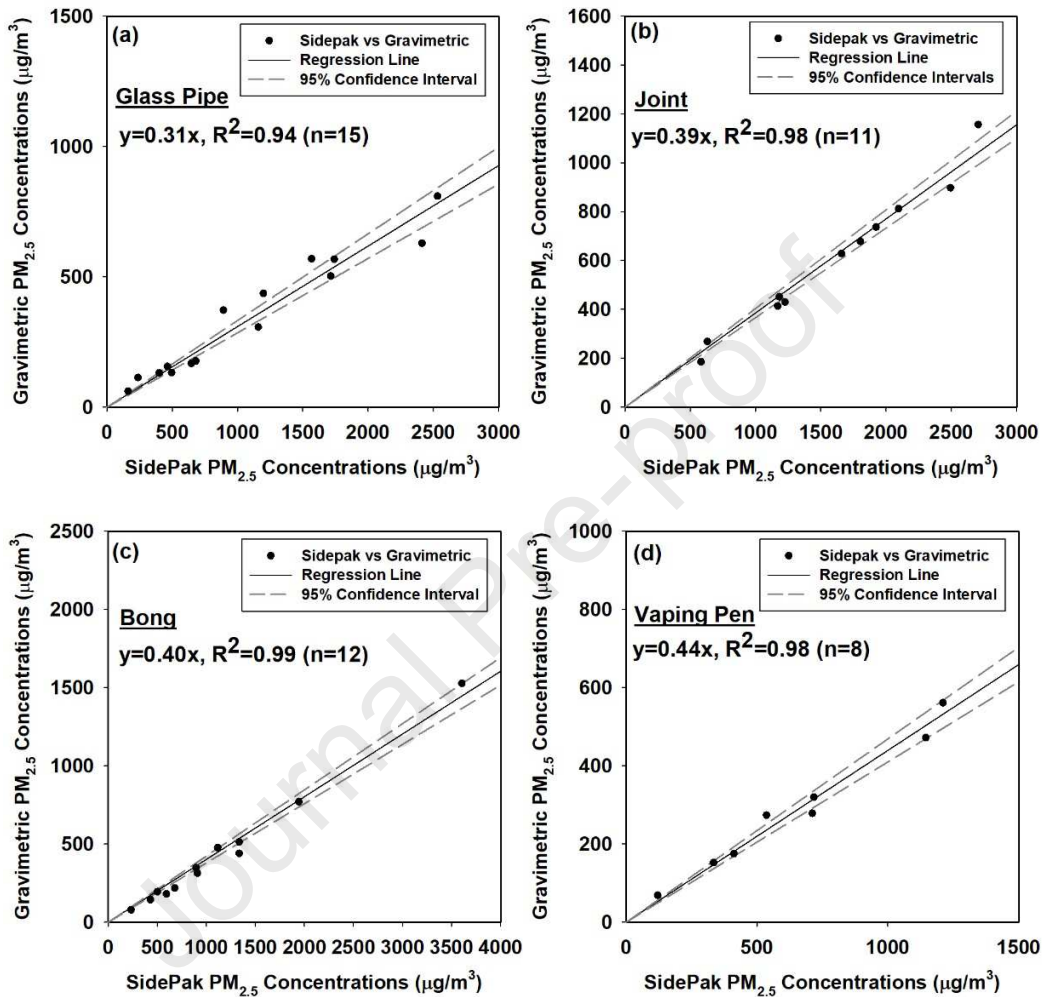
202 The particle removal rate is derived by subtracting air exchange rate (AER,  $h^{-1}$ ) from the  
203 particle concentration exponential decay rate ( $h^{-1}$ ) in each experiment. The removal rate in this  
204 study represents the removal due to both the particle surface deposition and evaporation. In this  
205 study, we measured AERs in 10 experiments to obtain the removal rates for each secondhand  
206 smoke source, including two samples each for the glass pipe, joints, bongs, vaping pens, and  
207 tobacco cigarettes. The AER was measured by releasing pure carbon dioxide ( $CO_2$ ) into the car  
208 (Sherman, 1990) after the participant smoked and left the car, with peak  $CO_2$  concentrations  
209 reaching approximately 3000 ppm—about six times of the background levels. During this  
210 experimental phase, all investigators left the garage to avoid increasing the  $CO_2$  background  
211 concentrations. During those sampling periods, we just collected one filter sample each time.

212

213

214

215

216 **3. Results and Discussion**217 **3.1 Calibration Factors**

218  
219  
220 **Figure 1.** Comparison of gravimetric sampling with SidePak measured PM<sub>2.5</sub> concentrations for SCS from (a)  
221 glass pipe, (b) joint, (c) bong, and (d) vaping pen.

222 **Figures 1a, 1b, 1c, and 1d** show that the SidePak's CFs for SCS from glass pipe smoking,  
223 joint smoking, bong smoking and cannabis e-vaping are 0.31 (SE=0.02), 0.39 (SE=0.02), 0.40  
224 (SE=0.01), and 0.44 (SE=0.03), respectively. The CFs for SCS are larger than that for STS  
225 (CF=0.29) (Jiang et al., 2011). It is worth noting that the cannabis buds used in the study when  
226 smoking the glass pipe and the bong were the same.

227 The CFs are highly influenced by the physical characteristics of particles, including the size  
228 distribution, shape, density, and refractive index (Hinds, 2012). The CFs for bong and joint  
229 smoking were similar. The  $PM_{2.5}$  from cannabis vaping had the largest CF, consistent with our  
230 previous finding that CFs for non-combustion heating processes (such as cooking on an electric  
231 stove) are larger than for combustion emissions (Dacunto et al., 2014).

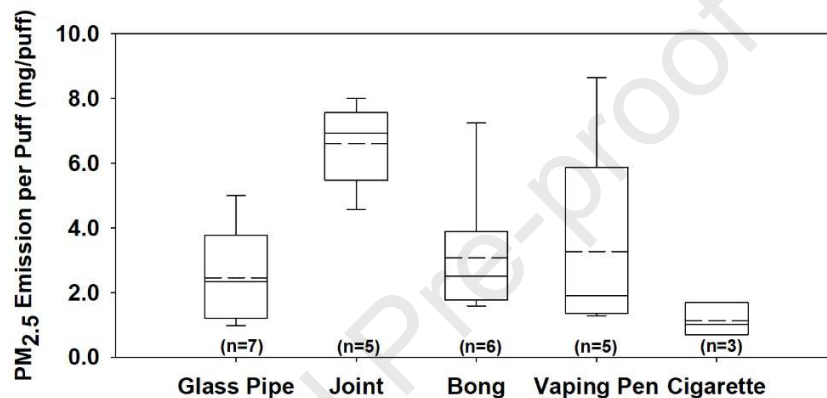
232 Zhao et al. (2017) measured the real-time  $PM_{2.5}$  concentrations of secondhand e-cigarette  
233 vaping aerosols from active human use in indoor environments. That study found that e-cigarette  
234 vaping aerosols went through quick evaporation, and the concentrations did not have a constant  
235 first-order decay rate. In contrast, we observed relatively constant decay rates in SCS from cannabis  
236 vaping in the car chamber (**Supporting Information Figure S2**).

237 Since six SidePak monitors were employed in this study, a CF for each monitor was also  
238 investigated (**Supporting Information Table S2**). The *R*-squared values for each linear regression  
239 result were all  $\geq 0.94$ . We found discrepancies between a CF derived from an arithmetic mean of six  
240 SidePak readings and a CF for an individual SidePak (i.e., CF ranges for glass pipe: 0.27-0.37, joint:  
241 0.33-0.40, bong: 0.35-0.45, and vaping pen: 0.38-0.50), although our SidePaks are factory-  
242 calibrated yearly. We found the APEs of all SidePaks were within 10%, except for one that was  
243 16%. These results suggest it is desirable to derive a CF for each  $PM_{2.5}$  optical monitor to ensure  
244 accurate exposure measurements.

245 The CFs for the Piezobalance measuring  $PM_{2.5}$  in SCS from glass pipe smoking, joint  
246 smoking, bong smoking, and vaping are 1.06 (SE=0.05), 1.15 (SE=0.06), 1.03 (SE=0.02), and 0.86  
247 (SE=0.04), respectively (as shown in **Supporting Information Figures S3a, S3b, S3c, and S3d**).  
248 The results are consistent with a previous study showing that the Piezobalance's results agree within  
249 15% of gravimetric sampling mass for a variety of aerosols, including polymer beads, road dust,  
250 pollen, and tobacco smoke (Sem et al., 1977). The CF (1.06) of SCS from the glass pipe is almost

251 the same as that from a bong (1.03). This is different from the SidePak CFs for these two sources  
 252 (Glass pipe: 0.31; Bong: 0.40 in **Figure 1**)—unlike SidePaks, the Piezobalance measurements do  
 253 not vary with aerosol optical properties. Since gravimetric sampling is usually time-consuming and  
 254 requires high-precision measurements of the mass, the Piezobalance can be used as a quick  
 255 reference method.

### 256 3.2 Emission rates



257 **Figure 2.** The emission rates (*mg/puff*) of glass pipe, joint, bong, vaping pen, and tobacco cigarette. (In each  
 258 boxplot, the dashed line represents the arithmetic mean value, and the solid line represents the median value.  
 259 The upper whisker represents the maximum value, while the lower whisker represents the minimum value.)

261 Although it is typical to express emission rates of STS as the emitted  $PM_{2.5}$  mass per tobacco  
 262 cigarette (*mg/cigarette*) or the emitted  $PM_{2.5}$  mass per minute (*mg/min*), these methodologies cannot  
 263 be generalized to SCS from various smoking methods. Glass pipe and bong smoking use cannabis  
 264 buds, and the quantities of each depend on the smokers' preferences. Joints vary in size, as do  
 265 cannabis vaping pens. Moreover, cannabis smoking often follows unique smoking patterns—  
 266 cannabis glass pipe and bong smokers usually smoke for only one or two puffs at a time. The puff  
 267 intervals vary and are difficult to estimate. Thus, we report the emission rate as the emitted  $PM_{2.5}$   
 268 mass per puff (*mg/puff*).

269 As shown in **Figure 2**, the arithmetic mean emission rates of SCS from glass pipe smoking,

270 joint smoking, bong smoking and vaping are 2.5 (SD=1.6), 6.6 (SD=3.1), 3.1 (SD=2.1), and 3.3  
271 (SD=3.1) mg/puff, while our measurement of the arithmetic mean emission rate of STS from  
272 cigarette smoking is only 1.1 (SD=0.5) mg/puff. The PM<sub>2.5</sub> emission rates (*mg/puff*) of the four SCS  
273 methods are ~2-6 times the emission rate of STS. Nazaroff and Klepeis (2003) summarized 17  
274 studies and reported a median emission rate for cigarette smoking as 12.7 mg/cigarette (including  
275 both machine and human smoking), and these studies showed that a typical commercial cigarette is  
276 sustainable for 6-16 puffs (Klepeis et al., 2003; Loeffroth et al., 1989; Singer et al., 2002). Thus, the  
277 emission rate of cigarette smoking can be estimated as 0.8-2 mg/puff, placing our STS data within  
278 this range.

279 We also observed that the cannabis buds often go out very quickly after each puff, so the SCS  
280 from a glass pipe and bong smoking might be mainly exhaled mainstream smoke. In addition, the  
281 SCS from vaping is entirely exhaled mainstream aerosol. Like STS from tobacco smoke, SCS from  
282 joint smoking is mostly a mixture of smoldering smoke between puffs (sidestream) and exhaled  
283 mainstream smoke, which might be the reason that it had the highest emission rate of the four  
284 cannabis sources tested.

### 285 3.3 Size Distributions

286 The size distributions of SCS from the four common smoking methods (i.e., glass pipe smoking,  
287 joint smoking, bong smoking, and cannabis pen vaping) at 1, 30, and 90 mins after each smoking  
288 activity are shown in the **Supporting Information Figures S4a, S4b, S4c, and S4d**, respectively.  
289 These data for glass pipe smoking, joint smoking, and bong smoking are based on a single puff,  
290 while the vaping experiment is based on three puffs. Because the background concentration was not  
291 significant compared to the SCS concentration, the size distribution of the background particles was  
292 not subtracted. The results showed that the number concentration of fresh SCS (1-minute after  
293 smoking) is bimodal—one mode located at ~380-420 nm and a smaller mode at ~800-840 nm. The



294 CMDs and GSDs for each mode are provided in **Supporting Information Table S3**. Our  
 295 observation of a ~380-420 nm mode is similar to Hiller et al. (1984), in which a puffing machine  
 296 was used to smoke a cannabis joint (85 mm in length and 25 mm in circumference) with the smoke  
 297 quickly diluted 126,000-fold. Their results, measured by a single particle aerodynamic relaxation  
 298 time analyzer, showed the mode ranged from 350 to 430 nm.

### 299 **3.4 Removal Rates**

300 **Table 2.** The removal rates of secondhand smoke from four methods of cannabis smoking and  
 301 cigarette smoking.

Smoking Types	Experiment No.	Removal Rate* (/h)
Cannabis Smoking		
Glass Pipe	1	0.41
	2	0.40
Joint	1	0.49
	2	0.38
Bong	1	0.35
	2	0.53
Vaping Pen	1	0.61
	2	0.77
Cigarette Smoking		
	1	0.44
	2	0.41

302 \* The removal rate represents the removal due to both the particle surface deposition and evaporation. The  
 303 air exchange rate has been subtracted from the total decay rate.

304 The AER in the stationary car with closed windows and doors during ten experiments was  
 305 0.47-0.59/h, mainly governed by the forced airflows due to the gravimetric sampling (18-20 L/min

306 each), as well as the Piezobalances and OPS sampling (1 L/min each). As shown in **Table 2**,  
307 excluding ventilation, the removal rates of SCS from a glass pipe, joint, bong, and cigarette  
308 smoking were similar, ranging from 0.35 to 0.53/h. The SEs were all  $< 0.001$ . These removal rates  
309 were affected by the increased air velocity and therefore higher deposition rates produced by the  
310 sampling pumps (Lai and Nazaroff, 2000). The removal rates of SCS from the vaping pen were  
311 higher (0.61-0.77/h) than other sources. This might be partly due to the evaporation of glycols in  
312 cannabis vape liquid, even though as mentioned earlier, the evaporation of particles from cannabis  
313 vaping is not expected to be as significant as those from e-cigarette vaping. We observed that the  
314 removal rates showed a 20%-33% variation between the two runs for bong smoking, joint smoking,  
315 and vaping. The removal of particles was caused by several dynamics, and it can be affected by  
316 turbulent intensity, velocity gradient, etc. (Lai, 2002; Lai and Nazaroff, 2000; Thatcher et al., 2002).  
317 We cannot conclude a specific reason for the variances through this study. Other studies have  
318 investigated removal rates of STS in a car chamber, mostly under high ventilation conditions (Liu  
319 and Zhu, 2010; Ott et al., 2008). For low-ventilation conditions, Xu et al. (1994) reported that the  
320 arithmetic mean removal rate (subtracting the AER from an overall decay rate) of STS was 0.172/h  
321 for an AER of 0.02/h in a 36.5 m<sup>3</sup> room. These results could provide input data for modeling SCS  
322 dynamics in in-cabin and indoor environments.

### 323 **3.5 Limitations of the Study**

324 The large variety of cannabis materials available on the market—various sizes of smoking  
325 devices and materials as well as different manufacturing methods—may result in differences in  
326 emission rates. Since SCS from glass pipe smoking, bong smoking, and vaping consist mostly of  
327 exhaled mainstream smoke, the emission rates will be highly dependent on the puffing  
328 topographies. As one of the very first studies focusing on secondhand cannabis emissions, this study  
329 did not include a large number of samples. Therefore, various puffing topographies—recruiting a

330 larger group of participants or employing various machine-puffing simulations—should be taken  
331 into consideration in the future. The particle size distribution of SCS may also be affected by the  
332 cannabis materials and puffing topographies (Zhao et al., 2016).

333 Jiang et al. (2011) found that the CFs of SidePaks for a given emission source were robust.  
334 They derived similar CFs when conducting controlled chamber experiments (mean = 0.29, SD  
335 =0.02) by using Marlboro cigarettes with a human puffing, and when measuring CFs in casinos with  
336 various kinds of cigarettes and human puffing (mean = 0.33, SD = 0.04). They also used 16 SidePak  
337 monitors in two sets of controlled chamber experiments 15 months apart, and they found the mean  
338 CF for STS differed by only 3% in the two studies. Lee et al. (2008) used Marlboro cigarettes in  
339 eight experiments comparing a SidePak monitor with gravimetric filter measurements, reporting  
340 almost the same CF value (i.e., 0.295) as Jiang et al.,(2011) (i.e., 0.29). Dacunto et al., (2013)  
341 reported a similar CF value (mean = 0.32, SD = 0.01) for the Marlboro cigarette. Since CF values  
342 will depend on the sizes and chemical composition of the particulate emissions, they should remain  
343 fairly robust for a given source type. However, there may be minor changes in the CF with the age  
344 of the emissions, due to gradual changes in size distribution and perhaps the composition of the  
345 aerosol.

346 Emission rates can be computed as the emitted  $PM_{2.5}$  mass normalized by the mass of  
347 consumed materials ( $mg/g$ ), which offers a generalized method to compare emission rates. However,  
348 in this study, we did not collect data on the consumed mass of cannabis materials. Use of an  
349 emission rate based on  $mg/puff$  allowed very different types of sources to be compared with each  
350 other, such as a cannabis joint with a vaping pen, or a cigarette with a cannabis water bong. For a  
351 cannabis joint, a better measure might be the emitted mass per unit length consumed ( $mg/mm$ ),  
352 which can be compared with tobacco emission rates.

353 Particle size distributions below 300 nm of SCS from different cannabis smoking and vaping

354 methods were not measured but could be valuable for future research. Anderson et al. (1989) found  
355 a mode with arithmetic mean CMD of ~ 100 nm when measuring the quickly diluted cannabis joint  
356 smoke using an electrical aerosol analyzer.

#### 357 **4. Conclusions**

358 This study, for the very first time, characterized SCS from four cannabis consuming methods  
359 and investigated PM<sub>2.5</sub> calibration factors, emission rates, size distributions, and removal rates. For  
360 SCS from four common sources, the calibration factors for the optical monitors ranged from 0.31 to  
361 0.44. Moreover, the CFs for the Piezobalances ranged from 0.86 to 1.15. Due to different smoking  
362 patterns between cannabis and tobacco smoking, we generalized the emission rate as emitted PM<sub>2.5</sub>  
363 mass per puff (*mg/puff*). The results showed the emission rates of SCS were ~2-6 times that of STS.  
364 Fresh SCS (1 min after smoking) was observed to have a bimodal distribution—one mode at ~380-  
365 420 nm and a smaller mode at ~800-840 nm. Under low-ventilation conditions, the indoor removal  
366 rates of SCS from the vaping pen were higher (i.e., 0.61-0.77/h) than those for the glass pipe, joint,  
367 bong, and cigarette smoking (which ranged from 0.35 to 0.53/h).

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# **Characteristics of Secondhand Cannabis Smoke from Common Smoking Methods: Calibration Factor, Emission Rate, and Particle Removal Rate**

Tongke Zhao, Kai-Chung Cheng, Wayne R Ott, Lance Wallace, Lynn M Hildemann

Civil and Environmental Engineering, Stanford University,

Stanford, California 94305, United States

## **Highlights:**

- Emission rates of secondhand cannabis smoke were 2-6 times of tobacco smoke;
- Fresh secondhand cannabis showed a bimodal size distribution;
- The indoor removal rates of secondhand cannabis vaping were slightly higher;



**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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