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Spectroscopic evidence of spin-state excitation in d-electron correlated semiconductor FeSb₂

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Iron antimonide (FeSb₂) has been investigated for decades due to its puzzling electronic properties. It undergoes the temperature-controlled transition from an insulator to an ill-defined metal, with a cross-over from diamagnetism to paramagnetism. Extensive efforts have been made to uncover the underlying mechanism, but a consensus has yet to be reached. While macroscopic transport and magnetic measurements can be explained by different theoretical proposals, the essential spectroscopic evidence required to distinguish the physical origin is missing. In this paper, through the use of X-ray absorption spectroscopy and atomic multiplet simulations, we have observed the mixed spin states of 3d⁶ configuration in FeSb₂. Furthermore, we reveal that the enhancement of the conductivity, whether induced by temperature or doping, is characterized by populating the high-spin state from the low-spin state. Our work constitutes vital spectroscopic evidence that the electrical/magnetic transition in FeSb₂ is directly associated with the spin-state excitation.

X-ray spectroscopy | spin-state excitation | electronic correlation

Some compounds with partially filled *d*-orbitals, such as transition metal oxide or chalcogenide LaCoO₃, NiS_{2-x}Se_x, VO₂, and so on, are insulating at low temperatures, but with temperature-enhanced electrical conductivity driven by correlations arising from charge, orbital, spin, and even the intertwined degrees of freedom (1, 2). Among those insulators/semiconductors with partially filled *d*-shell, FeSb₂ has an insulating and diamagnetic ground state at low temperatures, manifesting the record-breaking colossal thermoelectric effect and puzzling resistance anomaly. When the temperature ramps up, the electrical resistivity of FeSb₂ is gradually diminished by more than five orders of magnitude, with the cross-over from diamagnetism to paramagnetism (3–6). Such exotic properties make people recall the metal-insulator transition (MIT-like), but lacking a clear transition point from the temperature-dependent resistivity (*R*-*T*) curve. Various theoretical approaches have been proposed to unveil the underlying mechanism of the transition from the insulating phase to the ill-defined “metallic phase” in FeSb₂. For instance, the insulating and diamagnetic nature of FeSb₂ (and correlated insulator FeSi) has been discussed as a consequence of the Kondo singlet ground state, and the closure of the gap is attributed to the melting of the Kondo singlet with increasing temperature (7–12). Some recent studies even claim that FeSb₂ and FeSi are probably topological Kondo insulators, resembling SmB₆ (13, 14). The spin-state excitation (SSE) mechanism has also been tentatively proposed to explain its MIT-like behaviors, in analogy to LaCoO₃, which is well known for its SSE-induced insulator-to-metal transition (4, 15–18). The temperature-controlled excitation from 3*d*-orbital low-spin (LS) state to the high-spin (HS) state would lead to the presence of delocalized electrons near the Fermi level, which is responsible for the itinerant and paramagnetism. For instance, High-field Mössbauer spectroscopy of hyperfine interaction within FeSb₂ also hinted at possibly two different spin components from a single valence condition (19). Unfortunately, theoretical models of totally different microscopic basis can not exclusively address the MIT-like phenomena of FeSb₂ just from the aspects of the macroscopic transport and magnetic measurements (3–9). Due to the lack of explicit and direct spectroscopic evidence, the physical origin of the MIT-like behaviors in FeSb₂ remains an open question.

In this work, we present a systematic X-ray absorption spectroscopy (XAS) study on FeSb₂ and slightly Ru-doped FeSb₂. With the help of atomic multiplet simulations (AMS), we revealed that Fe ions are in 3d⁶ configuration with mixed LS (*S* = 0) and HS (*S* = 2) states. The insulating LS state dominates the ground state at low temperatures. The enhancement of the conductivity, driven by either temperature or doping effect, is

Significance

Turning a metal into an insulator (and vice versa) by temperature is an important research field in physics and is essential for advancing information technology. Iron antimonide (FeSb₂) manifests mysterious MIT (metal-insulator transition)-like behaviors and record-breaking colossal thermoelectricity. This paper reports a comprehensive X-ray absorption spectroscopy study on FeSb₂, proving that Fe ions are in 3d⁶ configuration with mixed insulating low-spin (LS) state and metallic high-spin (HS) state. The profound enhancement of the conductivity, driven by either temperature or substitution, is consistently associated with populating in the HS state from the LS state. This work provides crucial evidence to support that electronic/magnetic transitions in FeSb₂ are directly associated with spin-state excitation.

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The authors declare no competing interest.

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consistently associated with populating into the metallic HS state from the LS state. This direct observation provides compelling evidence that the MIT-like behaviors of FeSb₂ are closely correlated with the SSE within the framework of the HS/LS energy hierarchy.

Results

FeSb₂ crystalizes to a marcasite lattice (Fig. 1A) with the orthorhombic structure (space group *pnmm*). Under the CEF of distorted octahedral symmetry, Fe-3d orbitals form *e_g* and *t_{2g}* orbitals, and the energy degeneracy is further removed under the *C_{2h}* symmetry, as sketched in Fig. 1B. Early work claimed Fe is in a 3d⁴ configuration, occupying the lowest *d_{xz}*/*d_{yz}* orbitals (20, 21). Later, density-functional calculations and X-ray diffraction experiments suggested a divalent state of Fe ions due to the network of strong covalency (22–24). XAS is a powerful tool for studying the valence state of Fe ions. Fig. 2A presents the experimental (at 20 K) and simulated Fe *L*-edge XAS curves. Fig. 2B shows the second derivative curves of Fig. 2A to enhance the peak features for easy reading. The experimental data agree well with the simulated curves considering an LS 3d⁶ configuration but obviously deviate from the 3d⁵ or 3d⁴ outer shell configurations (SI Appendix, Supplementary Note 1). In the LS 3d⁶ configuration, the *t_{2g}* orbitals are fully occupied, as sketched in Fig. 2C, Left, resulting in a zero spin moment (*S* = 0) and a multiplet energy gap. This coincides with the nonmagnetic insulating ground state of FeSb₂.

Although a fairly good agreement between the experimental and simulated results, the simulated XAS curve with 100% LS state does not fully reproduce the experimental XAS curve, as it fails to capture the prominent peaks at $h\nu = 705.7$ eV ($h\nu = 718.7$ eV) in the pre-edge region of the Fe-*L*₃ (Fe-*L*₂) main peak. Such pronounced pre-edge multiplet features are rarely seen in the XAS of 3d⁶ LS state with *O_h* CEF symmetry [e.g., iron-based superconductors (25, 26)] but were discovered in the previous XAS studies on FeSb₂ and other Fe-compounds with the HS state (e.g., FeO) (27–30). They were also tentatively claimed to result from a mixed-valence state without experimental confirmation (13). However, it is very unlikely since mixing with a 3d⁵ or 3d⁴ configuration would shift the XAS peaks to higher energy (SI Appendix, Supplementary Note 1) (28). Here, we found the 3d⁶ HS state is very crucial to reconciling the XAS results. The HS state (Fig. 2C, Right) leaves vacancies in *t_{2g}* orbitals, so electrons are allowed to be excited from the core-level to *t_{2g}* orbitals, requiring less energy than the excitation to *e_g* orbitals in the LS state. Thus the 3d⁶ HS state gives rise to the XAS peaks shifting to lower energy (blue lines in Fig. 2A and B). Considering a mixed spin state (~15% HS and ~85% LS) in simulations leads to pre-edge peaks in both *L*₃- and *L*₂-edges

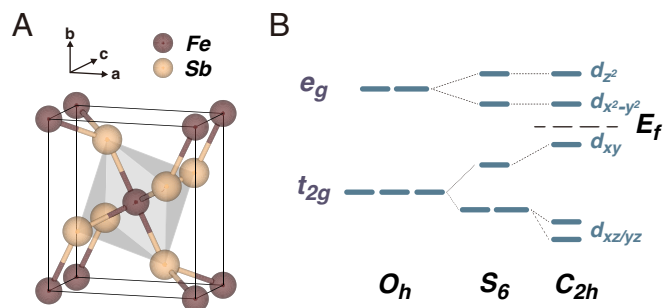


Fig. 1. FeSb₂ lattice structure and the energy levels of Fe-3d orbitals. (A) The unit cell of FeSb₂ lattice. (B) The energy levels of Fe-3d orbitals result from the CEF in a Marcasite lattice with *C_{2h}* symmetry (20). The Fermi level was noted as *E_f*.

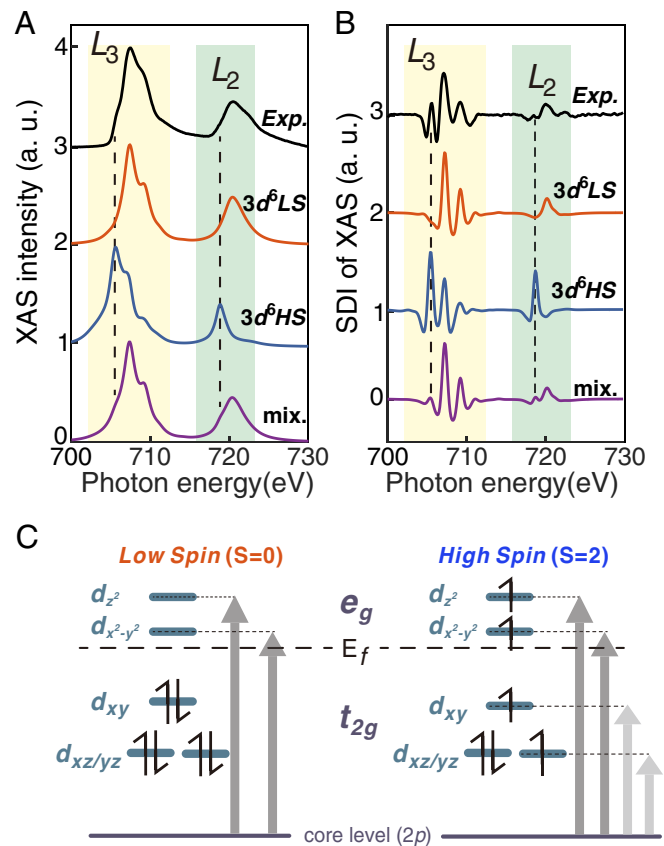


Fig. 2. The *L*-edge XAS and corresponding simulation of FeSb₂ and the spin excitation diagram. (A) The comparison of Fe *L*-edge XAS curves between experiments (black, *T* = 20 K) and the simulations of 3d⁶ LS state (red), HS state (blue), and the mixed spin state (purple, with 15% HS). (B) The second derivative images (SDI) of the experimental and simulated XAS curves. The vertical dashed lines mark the XAS peak ($h\nu = 705.7$ eV and $h\nu = 718.7$ eV) originating from the HS state. (C) The CEF split energy diagram of the Fe-3d orbitals in FeSb₂. All the *t_{2g}* orbitals are occupied for the LS state, forming an insulating state. For the HS state, *t_{2g}* and *e_g* orbitals are partially occupied, creating a metallic phase. The vertical arrows note the allowed possible transitions from the core levels to the vacant orbital. *d_{xz}*/*d_{yz}* orbitals are sketched here with degeneracy for simplicity. Please note that all the simulated and experimental XAS curves in this paper are normalized with the integrated XAS intensity from 700 eV to 730 eV. All XAS curves present in this figure is taken from (010) facet.

(solid purple curves in Fig. 2A and B), significantly improving the agreement between the simulations and the experiments. Notably, a stable intermediate spin state (*S* = 1) would also partially occupy *e_g* orbitals and leave vacancies in *t_{2g}* orbitals. However, it requires a special CEF environment (such as Fe-molecular under square planar symmetry) (28), which is not the case in FeSb₂. Therefore, the presence of pre-edge peaks and the overall features of XAS support that FeSb₂ possesses a 3d⁶ configuration with a mixture of HS and LS states.

To investigate the correlation between the mixed spin state and the MIT-like transition, we performed temperature-dependent XAS measurements from *T* = 20 K to *T* = 300 K. The XAS curves of different temperatures appear quite similar (Fig. 3A), but the temperature-dependent contrast becomes apparent (Fig. 3B) after subtracting the XAS curve of 20 K from other XAS curves. Within the *L*₃-edge region, a prominent feature with positive intensity is observed at around $h\nu = 705.7$ eV, corresponding to the energy position of the HS state in Fig. 2A. This positive feature is incremental with the temperature. Meanwhile, two dips with negative intensities were observed at about $h\nu = 707.5$ eV and $h\nu = 709.4$ eV at high temperatures, matching the energy positions of two prominent peaks of the LS state in Fig. 2A. The appearance of dips means the decrease

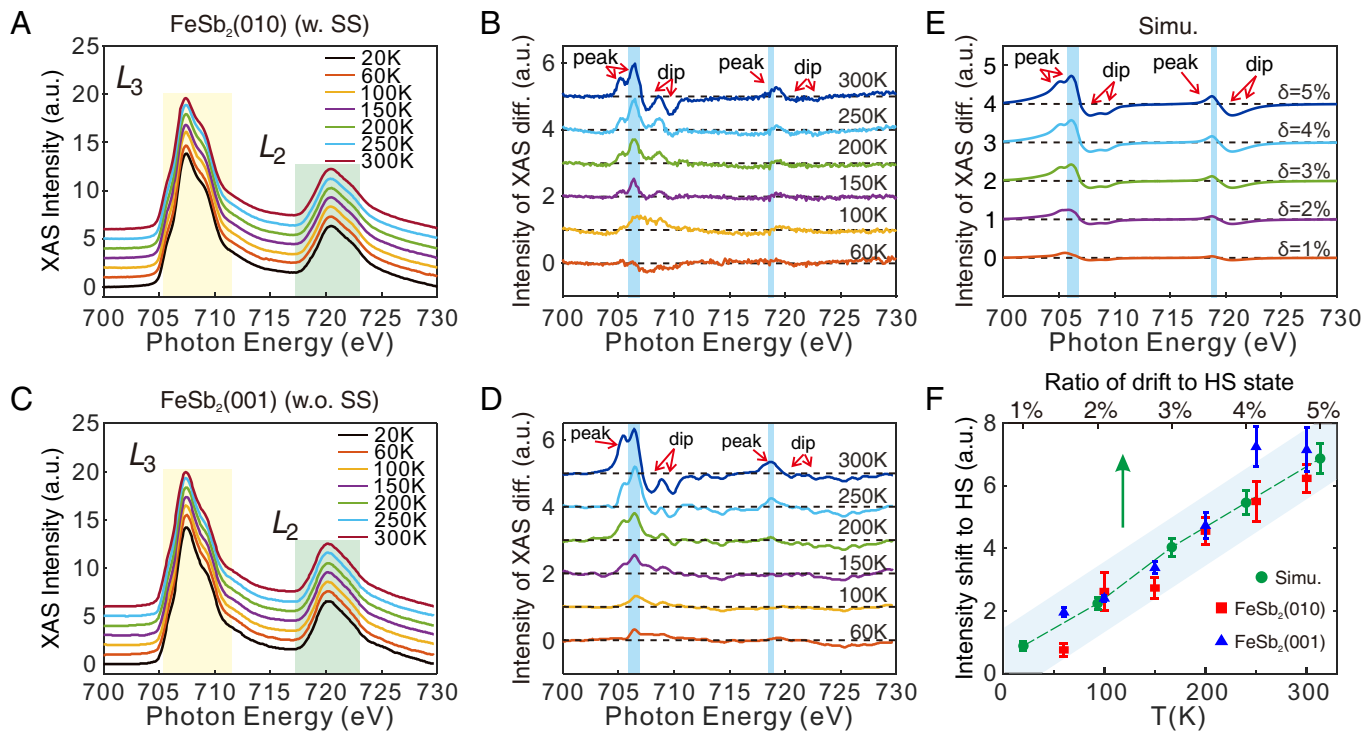


Fig. 3. Temperature-dependent XAS measurements and simulations. (A) The XAS curves of $\text{FeSb}_2(010)$ at different temperatures (from 20 K to 300 K). The XAS was measured during the warming-up process to eliminate the errors in reading the temperature. (B) The temperature-dependent differentiated $\text{FeSb}_2(010)$ XAS curves after subtracting the XAS curve of $T = 20$ K ($I_{(T=20K)}$). (C) Separated T-dependent XAS was measured on $\text{FeSb}_2(001)$, which is without electronic SS. (D) The T-dependent differentiated $\text{FeSb}_2(001)$ XAS after subtracting the XAS curve of $T = 20$ K ($I_{(T=20K)}$). (E) The simulated differentiated XAS curves after subtracting the XAS curve with 15% HS state. Red arrows note the peaks and dips. More simulation details can be found in *SI Appendix, Supplementary Note V*. (F) The XAS intensities shifted from the LS state to the HS state extracted from experiments (red/blue) and simulations (green) (Left axis), and the corresponding ratio of HS state populated from low temperature to 300 K (Top axis). The shifted intensities are sampled by integrating the intensity of blue-shaded ranges in L_3 -edge (705.8 eV to 706.8 eV) and L_2 -edges (718.5 eV to 719 eV) of the differentiated XAS from (B), (D) and (E).

in the LS state. The gain of the HS state's intensity is simultaneously along with the loss of intensity in the energy regime of two dips, indicating the drift from the LS state to the HS state. A similar behavior exists in the L_2 -edge, although the signals of peaks and dips are weaker. We notice that magnetic circular dichroism (XMCD) measurements on an unspecified sample's surface in ref. 27 are incompatible with the magnetic susceptibility, indicating possibly a bulk-surface separation. If so, these XAS peaks might come from the electronic surface states (SS), which are identified at the surface of $\text{FeSb}_2(010)$ and (110) facets (13, 24, 31). To rule out the possibility that what we observed is an effect resulting from the electronic SS resident on $\text{FeSb}_2(010)$ facet (13, 24), we repeated the same measurements on the cleaved surface of (001) facet (Fig. 3 C and D), which is identified to be free from any SS by high-resolution angle-resolved photoemission spectroscopy (ARPES, see *SI Appendix, Supplementary Note II*). The highly identical results from the (001) and (010) facets suggest that the drift of spin states is not induced by SS. The seeming mismatch between the magnetic susceptibility and the XMCD in ref. 27 is interesting, but would possibly be addressed if more experimental information and analysis were provided in that work. Furthermore, temperature-cycled XAS measurements confirm the changes of XAS as temperature dependence is reversible (*SI Appendix, Supplementary Note III*). The T-dependent XAS measurements were performed on a separate crystal (namely, sample #2) to show the same behaviors, indicating it is an intrinsic property of FeSb_2 (*SI Appendix, Supplementary Note IV*).

Moreover, we simulated the XAS curves by increasing the ratio of the HS state to catch the fine distinctions of T-dependent XAS (see more details in *SI Appendix, Supplementary Note V*). Fig. 3E presents the simulated results, which were differentiated between the XAS of increasing HS state ratio (from 16 to 20%) and the

base XAS with an HS state of 15%. Consistent with experiments, the simulated differentiated XAS curves also show peak-dip features at the characteristic energies in both L_3 and L_2 -edges (Fig. 3E). A quantitative assessment of how much the ratio of LS state drifted to the HS state from 20 K to 300 K is then evaluated. By integrating the intensity of the peaks of experimental data (blue-shaded ranges, Fig. 3 B and D), we can trace the change of HS state's intensity as a function of temperature (red squares and blue triangles in Fig. 3F). In the meantime, using the simulated data in Fig. 3E, one can also extract the corresponding intensity of the HS state (blue-shaded ranges) as a function of the HS state ratio change, and compare it directly with the experimental results (Fig. 3F). The increase of the spectral weight of the HS state from 20 K to 300 K can be fitted well by raising the ratio of the HS state approximately from 15 to 20%. Yet, one needs to be very cautious when interpreting the quantitative assessment (5% increment of HS) due to the limitation of the AMS. Nevertheless, the coordinative results from both experiment and simulation successfully depict the picture of SSE from LS to HS during the transition from insulating to conducting state.

Despite the temperature, the substitution of Fe-ions could also tune the $3d^6$ spin state and the corresponding electrical properties of FeSb_2 . Slightly substituting Fe with ruthenium (Ru) would not alter the crystal structure since RuSb_2 is isostructural to FeSb_2 . As shown in Fig. 4A, the 5% Ru-doped sample retains the MIT-like behavior (see magnetic susceptibilities shown in *SI Appendix, Supplementary Note VI*). The absolute resistivity becomes larger (Inset of Fig. 4A) in doped samples, which is possibly due to doping-induced disorders. In contrast, the resistance ratio value ($R_{(2K)}/R_{(300K)}$) was reduced by about ~ 4 times after Ru doping (Fig. 4A). The resistance anomaly at $T < 5$ K was found in both

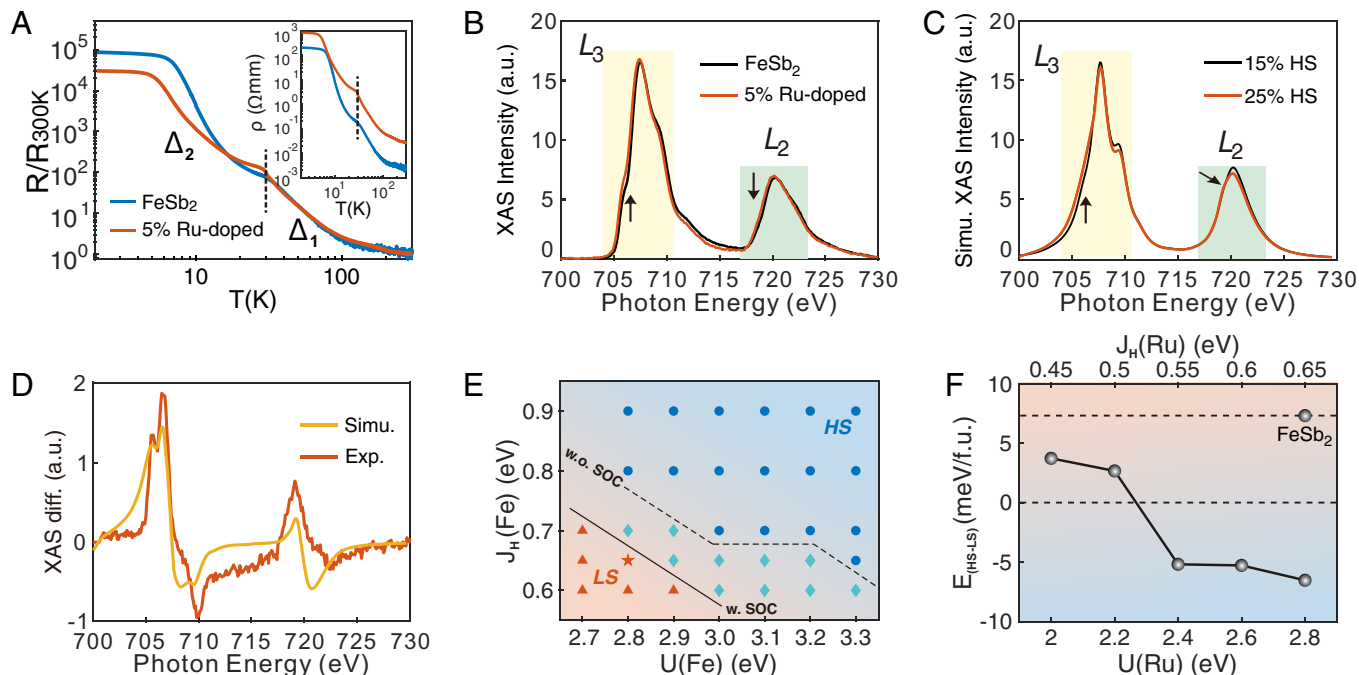


Fig. 4. Electrical transport, XAS measurements, and first-principles calculations on the Ru-doped FeSb₂. (A) The normalized resistance ($R/R_{300\text{K}}$) of FeSb₂ (blue) and Fe_{0.95}Ru_{0.05}Sb₂ (red) as a function of temperature. The vertical dashed line at $T \sim 30$ K marks the temperature value where the system transits from a larger energy gap (Δ_1) to a smaller energy gap (Δ_2). The *Inset* shows the resistivity. (B) The XAS curves of FeSb₂ (black) and Ru-doped FeSb₂ (red) at $T = 20$ K. (C) The comparison of simulated XAS curves of different HS state ratios [15% (black), 25% (red)]. Black arrows in (B) and (C) indicate the photon energy where the XAS intensity changes. (D) The comparison of differentiated XAS ($XAS_{\text{Ru-doped}} - XAS_{\text{pristine}}$, red) with simulated differentiated XAS. The simulated differentiated XAS was generated by subtracting the XAS curve of 25% HS state from the XAS curve of 15% HS state. (E) The spin-state phase diagram of FeSb₂ in the parameter space of Fe's U and J_H . The blue dots (red triangles) indicate that the ground state is the HS (LS) state regardless of whether the SOC effect is taken into account. The cyan diamonds indicate a transition region where the ground state is the LS state without SOC but the HS state with SOC. The red star indicates the selected U and J_H when calculating Ru-doping. (F) The energy difference between the HS and LS states ($E_{\text{HS-LS}}$) changes with the Ru-dopants' U/J_H values. The J_H/U strength is roughly fixed as ~ 0.23 .

samples, which is possibly a result of the in-gap SS identified in FeSb₂, resembling the similar feature in SmB₆ (13, 24, 31, 32). Meanwhile, the R - T curves of both samples share the same kink feature at $T \sim 30$ K to separate two regimes, dominated by different energy gaps (Δ_1 and Δ_2) (3–7, 13, 32, 33). The larger gap Δ_1 (above 30 K) is widely accepted as the intrinsic multiplet gap of Fe $3d$ -orbital, while the physical origin of the smaller one Δ_2 (below 30 K) is still controversial. It was claimed that Δ_2 might present the excitation of impurity states within the CEF gap (34, 35). However, the indirect excitation nature supports it could be an intrinsically correlated gap (33, 36). In our experiments, both gaps of each sample were extracted by fitting the R - T curves by using

the Arrhenius equation $R \sim Ae^{\frac{\Delta}{2k_B T}}$, and both gaps are shrunken after the Ru-doping (SI Appendix, Supplementary Note VI). The multiplet gap Δ_1 is about 33.4 meV for FeSb₂ but 30.6 meV after doping with Ru. The gap Δ_2 is about 10.4 meV for FeSb₂ and 6.3 meV for Ru-doped FeSb₂. The reduction of Δ_1 after 5% Ru-doping indicates a smaller $3d^6$ multiplet gap between the t_{2g} and e_g states. Since the CEF splitting is a key factor in tuning the competition between the LS and HS states, the reduced CEF gap would favor the HS state. Therefore, one would anticipate a gain in the HS state in the XAS measurement on Ru-doped samples. The XAS of Ru-doped FeSb₂ shows an apparent increase in intensity at the characteristic energy position of the HS state (Fig. 4B), which can be reproduced by simulation with a larger portion (25%) of HS state (Fig. 4C). Meanwhile, increment of 10% HS (from 15 to 25%) also very well fits the differentiated XAS curve in Fig. 4D. Although the increased HS state ratio with the shrunken multiplet gap in Ru-doped samples is in line with the SSE narrative, the exact

reason why the HS ground state is substantially enhanced by Ru-dopants needs to be well understood quantitatively.

Therefore, to explore the associated energy scales and the underlying mechanism of how 5% Ru-dopants greatly boost HS state occupations, we performed first-principles calculations. A phase diagram (Fig. 4E) is obtained in the parameter space of the Hubbard U and Hund's coupling J_H . It is expected that the LS state (red-shaded) is preferred in the low U - J_H region, while the HS (blue-shaded) occupies the high U - J_H region. An important message is that the spin-orbit coupling (SOC) promotes the HS state, by pushing the LS-HS phase boundary to the low U - J_H direction. According to this phase diagram and experimental results, we adopted $U = 2.8$ eV/ $J_H = 0.65$ eV as a proper set of parameters to describe FeSb₂, which leads to the LS state near the phase boundary and a semiconductor gap (Δ_1) of ~ 50 meV. The energy difference between the HS and LS states is rather tiny as $E_{\text{HS-LS}} \sim 7.5$ meV/f.u. Then, the spin ground state of Ru-doped FeSb₂ is also surveyed as the function of Ru-dopants' U - J_H . Since the SOC can promote the HS state (Fig. 4E) and Ru's SOC is larger than Fe's, it is reasonable to expect the SOC-driven HS tendency. As shown in Fig. 4F, the $E_{\text{HS-LS}}$ value gradually shrinks with low U -values and the ground state can even flip to HS with higher U -values. By considering the larger extension of the Ru $4d$ -orbitals, it is well accepted that the U - J_H of $4d$ orbitals should be notably smaller than the $3d$ ones. Consequently, the smaller U/J_H , e.g., $U = 2.2$ eV/ $J_H = 0.5$ eV, should reasonably fit the LS ground state. Furthermore, the significantly dampened $E_{\text{HS-LS}} \sim 2.5$ meV/f.u. is comparable to the small thermal activation energy of $T \sim 20$ K, which explains why Ru-dopants greatly enhance the HS state occupation.

Although the SSE narrative is based on a framework nearly degenerate HS-LS state, Our experiments and calculations suggest that CEF energy and Hund's rule repulsion are not the sole factors in determining which spin state (LS or HS) dominates the ground state (24, 37). The SOC effect which favors the HS state, must also be considered. For a given SOC strength, a dedicated balance would be reached when the CEF splitting competes with a strong Hund's correlation. It is known that Hund's coupling is weakly dependent on temperature (38). CEF energy is inversely related to the lattice volume and usually slightly weakens under the lattice thermally expansion. When the energy difference between the LS and HS ground states is rather small (or nearly degenerate, namely), thermally activation energy could stand out to tune the occupation of HS/LS states and the metallic/insulating state.

Meanwhile, the physical origin of the mixed LS-HS ground state in FeSb₂ is an interesting question. The spin-state transition identified in the other compounds like LaCoO₃ is inevitably derived from the change of local chemical environment like the bond-state and lattice structure (39). The lattice distortion of FeSb₂ has been reported (7, 40), but with a clear difference when compared to LaCoO₃. The change of bond length within LaCoO₃ is continuous to even T ~ 1,000 K (39). However, the detected lattice distortion in FeSb₂ is more complicated. Despite thermally induced lattice expansion, an abrupt lattice distortion happened at T ~ 100 K in FeSb₂ by Raman spectroscopy, which is likely related to the lattice symmetry along the a-axis (40). It was argued that the symmetry changes at T ~ 100 K would trigger the spin state excitation (7). However, the photoemission results suggest a continuous change of electronic band structure from ~20 K to ~200 K (13, 24), which is in line with our XAS result. Previous XAS/RIXS work of LaCoO₃ shows that the HS state is populated to more than 50% at T ~ 300 K, which is profoundly larger than the ratio of 5% extracted by AMS in FeSb₂. Here, we'd like to reiterate the limits of AMS in quantitatively accessing the spin-state ratio in FeSb₂. Furthermore, the excitation of spin-unpaired electrons has been detected by high-field electron paramagnetic resonance spectroscopy in LaCoO₃ (41), but was absent in FeSb₂ (*SI Appendix, Supplementary Note VII*). Therefore, FeSb₂ should not be viewed as a straightforward analog of LaCoO₃, and more research is needed to determine whether the observed LS-HS excitation is the only mechanism involved in modulating the electric/magnetic properties.

Conclusion

To conclude, our comprehensive study on FeSb₂ and the analysis of its spin state reveal that FeSb₂ has a 3d⁶ mixed spin state with equilibrium between S = 0 and S = 2. The enhanced conductivity either tuned by temperature or doping is strongly associated with the drift from the insulating LS state to the metallic HS state. Our work supports the SSE mechanism, instead of the Kondo correlation, which plays an important role in tuning the electrical properties in FeSb₂. Finally, it is noteworthy that figuring out the orbital occupancy/spin state is the first step to fully understanding the mysterious properties of FeSb₂. For instance, the record-breaking colossal thermopower is deeply rooted in the electronic structure near the Fermi level (40, 42). The determined low-energy electronic structure could also help realize the altermagnetism in Co-doped FeSb₂ (43). Meanwhile, The SSE mechanism of MIT-like behaviors in FeSb₂ will have substantial implications for other mysterious iron-based correlated materials like FeSi or FeGa₃ (12, 44) that exhibit similar macroscopic properties as FeSb₂.

Materials and Methods

The single crystals of FeSb₂ and Ru-doped FeSb₂ were grown by a self-flux method, the same as ref. 5. They are cleaved in the ultrahigh vacuum with a base pressure better than 2×10^{-9} torr to obtain clean (010) and (001) surface for spectroscopic study. The soft X-ray absorption (Fe L-edge) spectroscopy measurements were performed in the Beamlines MCD-A and MCD-B (Soochow Beamline for Energy Materials) at National Radiation Source Lab (Hefei). The X-ray was with π polarization, and the data were collected with the total electron yield mode. AMS of XAS were performed as in refs. 25, 45, and 46, describing $2p^6 3d^n \rightarrow 2p^5 3d^{n+1}$ (Fe L-edge) X-ray absorption in the dipole approximation. Hartree-Fock parameters were obtained from the Cowan code (47), and full diagonalization of the multiplet Hamiltonian was performed using LAPACK drivers (48). The *d*-shell interaction (F_{dd}) was renormalized as 80% and the crystal field splitting energy (CEF, 10Dq value) was set to 2.0 eV to generate a 3d⁶ LS ground state. Meanwhile, the 10Dq was lowered to 1.8 eV to gain pure 3d⁶ HS ground states. All the other parameters remain intact and identical between the HS and LS state simulations to exclude the uncertainties induced by tuning other parameters. However, we notice that HS/LS states would generate different core-hole energies. Thus, the simulations of divalent Fe HS state and LS state are toy models based on the large CEF and relatively smaller CEF scenarios (with the cross-over 10Dq = 1.87 eV) to illustrate only the framework of SSE.

The first-principles calculations based on density functional theory (DFT) were performed with the projector augmented wave pseudopotentials as implemented in the Vienna ab initio Simulation Package (49, 50). The plane-wave cutoff energy was 500 eV, and the Fe's 4s3d3p electrons were treated as valence states. The *k*-point grid of $6 \times 5 \times 10$ was adopted for the smallest primitive cell. A criterion of 0.01 eV/Å was used for the Hellman-Feynman forces during the structural relaxation and the energy convergence criterion is 10^{-6} eV. The Perdew-Burke-Ernzerhof parametrization of generalized gradient approximation was used for the exchange-correlation function (51). In order to describe the strongly correlated *d*-electrons from the iron element, the rotationally invariant DFT+*U* without exchange splitting introduced by Liechtenstein (LDAUTYPE=4) was taken into consideration (52), where the effective on-site Coulomb parameter (*U*) and exchange parameter (*J_H*) can be adjusted. The Ru doping was done in a $2 \times 2 \times 2$ supercell with a concentration of 6.25%, close to the 5% doping ratio in the experiment.

The ARPES measurements in *SI Appendix, Supplementary Note II* note were performed at the MERLIN ARPES endstation beamline 4.0.3 at the Advanced Light Source and BLO3U of Shanghai Synchrotron Radiation Facility. The temperature was maintained at T = 20 K, and the overall energy and resolution were 15 ~ 20 meV. The base pressure is better than 1×10^{-10} Torr. All ARPES data were taken within 10 h after cleavage, and band structure was stable on this timescale.

Data, Materials, and Software Availability. All study data are included in the article and/or *SI Appendix*.

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