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CARRIER-FREE RADIOISOTOPES FROM CYCLOTRON TARGETS IV. PREPARATION AND ISOLATION OF ${\rm Mn}^{54}$ AND ${\rm Co}^{56}$,57,58

Roy D. Maxwell, Jeanne D. Gile, Warren M. Garrison, and Joseph G. Hamilton

August 17, 1949

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CARRIER-FREE RADIOISOTOPES FROM CYCLOTRON TARGETS IV. PREPARATION AND ISOLATION OF Mn54 AND Co56,57,581

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Carrier free radioisotopes of manganese and cobalt of useful half-life may be obtained only by cyclotron bombardment. The present paper describes chemical methods of isolating Mn^{54} and Co^{56} ,57,58 from an iron target bombarded with 19 MeV deuterons in the 60-inch cyclotron at the Crocker Laboratory. At this energy Mn^{54} and Co^{56} ,57,58 are produced in a thick target by the nuclear reactions³: $\text{Fe}^{56}(\text{d},\alpha)\text{Mn}^{54}$, $\text{Fe}^{56}(\text{d},2n)\text{Co}^{56}$, $\text{Fe}^{56}(\text{d},n)\text{Co}^{57}$, $\text{Fe}^{57}(\text{d},2n)\text{Co}^{57}$, $\text{Fe}^{57}(\text{d},n)\text{Co}^{58}$. Other possible shorter-lived radioisotopes of manganese and cobalt were allowed to decay out prior to the separation.

The iron 4 target was silver soldered to a water cooled copper probe and bombarded for 2000 μa -hr at an average beam intensity of approximately 50 μa . The bombarded surface was dissolved off in 6 N HCl.

All but approximately 10 mg. of the target iron was extracted with ether. The aqueous phase was neutralized with NH $_{\downarrow}$ OH, a few drops of bromine were added, and Fe(OH) $_{3}$ was precipitated with 15 N NH $_{\downarrow}$ OH in excess. Under these conditions carrier free radio manganese was quantitatively carried, presumably in the

- (1) This document is based on work performed under the auspices of the Atomic Energy Commission.
- (2) Lieutenant Colonel, U. S. Army, now stationed at Walter Reed Hospital, Washington, D. C.
- (3) G. T. Seaborg and I. Perlman, Rev. Mod. Phys. 20, 585 (1948).
- (4) The iron was obtained from Johnson Mathey & Co. Manganese and cobalt were not detected by spectrographic analysis.

quadrivalent state. Over 98% of the radio cobalt remained in the supernatant⁵. The Fe(OH)₃ plus radio-manganese was reprecipitated twice with Br₂-NH₄OH in the presence of 1 mg of cobalt hold-back carrier. Three additional precipitations removed traces of carrier cobalt. The Fe(OH)₃ was washed, dissolved in 6 N HCl and iron was extracted with ether. The solution of carrier free radio-manganese in 6 N HCl was evaporated to dryness after the addition of 5 ml of isotonic saline solution. Addition of 5 ml of distilled water gave an isotonic solution of carrier-free Mn⁵⁴ for subsequent biological experiments.

The activity was identified by the 310 day half-life and by the 0.85 Mev gamma ray reported for Mn⁵⁴. As further identification, a small amount of activity added to a solution containing carrier amounts of Mn, Fe and Co was quantitatively recovered in the Mn fraction from a chemical separation.⁷

The supermatant from the first Fe(OH)₃ precipitation containing carrier-free radio-cobalt, NH₄Cl and NH₄OH was evaporated to dryness and treated with 16 N HNO₃ to destroy ammonium salts. 12 N HCl was added to remove HNO₃ and the solution was evaporated to dryness on 40 mg of NaCl. The activity dissolved quantitatively in 5 ml of water.

The radio-cobalt showed the 72-day half-life of ${\rm Co}^{56,58}$; after 200 days the longer period of ${\rm Co}^{57}$ was apparent. In addition, chemical separation with added Mn, Fe and Co carriers identified the activity as cobalt. Mass absorption measurements showed the 1.2 MeV beta reported⁸ for ${\rm Co}^{56}$. The combined gamma radiation had a half-thickness of approximately 10 gm/cm² in lead.

We are grateful to Professor G. T. Seaborg for reading this manuscript, to the staff of the 60-inch cyclotron for bombardments, and to Mrs. Alberta Mozley and Mrs. Helen Haydon for technical assistance.

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