# Lawrence Berkeley National Laboratory

**LBL Publications** 

## Title

Summary of the Research Progress Meeting of April 6, 1950

## Permalink

https://escholarship.org/uc/item/7xt9b4mg

## Author

Kramer, Henry P

## **Publication Date**

1950-05-01

## **Copyright Information**

This work is made available under the terms of a Creative Commons Attribution License, available at <a href="https://creativecommons.org/licenses/by/4.0/">https://creativecommons.org/licenses/by/4.0/</a>

DECLASSIFIED

UCRL\_676

0

UCRL. 67

 $\mathcal{O}$ 

C.2



Radiation Laboratory

TWO-WEEK LOAN COPY

This is a Library Circulating Copy which may be borrowed for two weeks. For a personal retention copy, call Tech. Info. Division, Ext. 5545

# BERKELEY, CALIFORNIA

## UNIVERSITY OF CALIFORNIA Radiation Laboratory

Cover Sheet Do not remove

INDEX NO	UCRL 676
This document	contains <u>8</u> pages
This is copy 6	<b>2</b> of <u>7</u> 2Series <u>A</u>

DECLASSIFIED to \_INFO. DIVISION



Classification

Each person who receives this document must sign the cover sheet in the space below

Route to	Noted by	Date	Route to	Noted by	Date
· .		s .			
•	· · ·				
	· · ·				
	· · · · · · · · · · · · · · · · · · ·	-			
				· · · · · · · · · · · · · · · · · · ·	
		· · · ·			
	•				

UCRL-676

Chemistry-Transuranic elements

# DECLASSIFIED

UNIVERSITY OF CALIFORNIA

Radiation Laboratory

Contract No. W=7405-eng-48

CLASSIFICATION CANCELLED BY AUTHORITY OF THE DECLASSIFICATION ERAN H USAEC OF THE DIATE SIGNATURE OF THE DATE PAR ON MAKING THE CHANGE

SUMMARY OF THE RESEARCH PROGRESS MEETING

of April 6, 1950

Henry P. Kramer

May 11, 1950

ON mation affecting the States. its contents in is prohibited es under any and may applicable Fe

Some of the results reported in this document may be of a preliminary or incomplete nature. It is the request of the Radiation Laboratory that the document not be circulated off the project nor the results quoted without permission.

Berkeley, California

FIFN

UCRL-676 Chemistry-Transuranic Elements

Nos.

#### -2-

### Standa

Standard Distribution	Copy
Argonne National Laboratory	1-10
Atomic Energy Commission, Washington	11-12
Brookhaven National Laboratory	13-16
Carbide and Carbon Chemicals Corporation (K-25 Plant)	17-20
General Electric Company, Richland	21–26
Hanford Operations Office	27
Iowa State College	28
Knolls Atomic Power Laboratory	29 <b>-3</b> 2
Los Alamos Scientific Laboratory	33-35
Mound Laboratory	36-38
Naval Radiological Defense Laboratory	39
NEPA Project	40
New York Operations Office	. 41-42
Oak Ridge National Laboratory	43-50
Patent Branch, Washington	51
Technical Information Branch, ORE	52-66
UCLA Medical Research Laboratory (Warren)	67
University of California Radiation Laboratory	68-70
University of Rochester	71-72

Total

72

Information Division Radiation Laboratory University of California Berkeley, California

UCRL 676 Unclassified Distribution

-3-

#### SUMMARY OF THE RESEARCH PROGRESS MEETING

### of April 6, 1950

Henry P. Kramer

### The New Element Berkelium. S. Thompson.

The speaker, A. Ghiorso, and G.T. Seaborg.have discovered an isotope of element 97. They received help from Professor B. B. Cunningham, K. Hulet, the Health Chemistry group under the direction of N. Garden and the crew of the 60-inch cyclotron.

The work that culminated in this discovery was motivated by interest in the systematics of  $\checkmark$  and  $\beta$  decay, in actinide chemistry, and in this connection, in the extension of the analogy between rare earths and actinide elements. The possibility of discovering more stable configurations of neutrons and protons and thus extending knowledge of the variation of stability with atomic and mass numbers was a further spur to the research.

By extrapolating the analogy between the rare earth elements and the existing actinides to the element with atomic number 97, a guess was made that the chemical properties of 97 should in the main correspond to those of terbium of the lanthanides. One can understand the analogy in properties by considering that in the rare earth series the 4f shell is being filled whereas in the actinides the 5f shell is being completed. For example, 96 Cm has 7 electrons in the 5f shell. Its analogue 64 Gd has 7 electrons in the 4f shell. Element 97 has 8 electrons in the 5f shell while its expected analogue 65 Tb has the same number of electrons in the 4f shell. Several methods for the preparation of 97 were tried. It is conveivable that each of the following three nuclear reaction sequences may result in the production of isotopes of 97:

> $Am^{241}(a, {}^{2n})97^{243}$  $Cm^{242}(d,n) 97^{243}$

-4-

 $Am^{241}(n,\gamma)Am^{242}(n,\gamma)Am^{243}(n,\gamma)Am^{244}(\beta \operatorname{decay})Cm^{244}(n,\gamma)Cm^{245}(n,\gamma)Cm^{246}(n,\gamma)Cm^{246}(n,\gamma)Cm^{247}(\beta \operatorname{decay})97^{247}.$ 

Sufficient quantities of americium were obtained by neutron bombardment of  $Pu^{239}$  resulting in the production of  $Pu^{241}$  which transmutes by beta decay to  $Am^{241}$ .

At present only the methods involving cyclotron bombardment have met with success. The bombardment in the cyclotron presented the problem of containing the sample in a target holder constructed in such a way that no radioactivity might leave it to contaminate the cyclotron. To keep the target from melting, a water cooling system was incorporated in the target holder. The bombardments were carried out with 35 Mev alpha particles accelerated in the 60-inch cyclotron.

Before doing the experiments, estimates were made of the nuclear properties of some isotopes of element 97. The alpha energy and half-life were estimated by extrapolating the graphical compilation of alpha energy versus mass number and atomic number, which has been published by Seaborg, Perlman and Ghiorso, and by closing energy cycles. The energy obtained from the closed cycles was used to estimate the half-life for electron-capture using an empirical relationship established for the heavy isotopes.

Americium was bombarded as the oxide. After the bombardment it was dissolved in strong HNO<sub>3</sub> and precipitated as the hydroxide. The hydroxide was dissolved and oxidized to the +VI oxidation state. The portion of

**UCRL 676** 

Americium in the +III state was then precipitated as the fluoride, converted to the hydroxide and dissolved in HClO<sub>4</sub>. By oxidizing about 99% of the americium to its +VI state the americium fraction remaining in the +III state and containing 97 was reduced in bulk.

-5-

The rare earths were eliminated by adsorbing the mixture of radioactivities on Dowex 50 resin, transferring the resin to a column packed with the same resin and eluting with 13M HCl which removes actinides faster than rare earths. The actinide fraction was run through another Dowex 50 resin column with ammonium citrate for elution. The temperature was maintained at  $87^{\circ}$ C in order to reduce the time of separation. Element 97 was obtained in the neighborhood of the fortieth drop in a position relative to Cm and Am corresponding to the position of its rare earths analogue, terbium, relative to Gd and Eu.

The activity was placed in the pulse analyzer and three alpha groups were found of energies 6.20 Mev, 6.55 Mev, and 6.72 Mev. The half-life for  $\lambda$  - decay was~l year. The electromagnetic radiation was examined in an x-ray proportional counter. In this way the branching ratio, i.e., the ratio between the number of decays by alpha emission and the number of decays by electron capture was determined to be 0.001. The half-life of 97<sup>243</sup> was 4.6 hours.

The electron-capture-decay daughter of  $97^{243}$ ,  $\text{Cm}^{243}$ , was observed to grow into the sample.  $\text{Cm}^{243}$  emits alpha particles of energy of 5.84 Mev. The -decay daughter of  $97^{243}$ ,  $\text{Am}^{239}$  was also observed.

To establish the chemical properties of 97 a number of experiments have been carried out. They show among other things that the oxidation potential of the couple  $97(+3) \rightarrow 97(+4)$  is of approximately the same magnitude as that of the couple  $Ce(+3) \rightarrow Ce(+4)$ , namely ~ 1.6 volts.

-6-

## Photon Coincidences from Proton Bombardment. B. Moyer.

The high energy x rays that have been observed by Moyer and York have been tentatively explained as arising from the decay of a neutral meson into two photons. In order to test this hypothesis measurements have been instituted with a view to observing two  $\gamma$  ray photons in coincidence at the 184-inch cyclotron. For this purpose the equipment designed by J. Steinberger for use in similar measurements of **y** rays produced by the action of the high energy x-ray beam of the synchrotron was used. The apparatus is sketched in Fig. 1. Its dominant characteristic is its extremely short resolution time of 1 microsecond. The gain in resolution that is to be had by the use of the high-speed coincidence-anti-coincidence apparatus is off-set to a certain extent by the small solid angle subtended by the crystals. This factor reduces the counting rate which is critically low in the small intensity deflected proton beam. The background is strongly target-dependent, consisting in the main of neutrons converted to protons in the converter and is primarily directed forward. In order to reduce the background counting rate, therefore, the counter telescopes were tilted somewhat backward of the target plane.

Because of the low counting rates the statistics are as yet very sparse and are of a very preliminary nature.

Converter I	Converter II	Accidentals Measured Predicted		Counts/Min Total Count		Net Count Measured Predicted			
Pb	РЪ	.17 0.09	.1	.15	0.41	.05	•24	.10	.72
РЪ	none	3 count <b>s</b> in 22 min.	ı	,	·				
none	none	2 counts in 51 min.	1						
Cu	Cu	.13 .07			0.26	.07	.13	.10	

The large difference between actual and predicted count is thought to be due in part to the scattering of electrons arising from converted  $\gamma$  -rays.

Information Division mkm/5-11-50



Apparatus for Measuring  $\gamma$  -rays from  $\pi$  <sup>o</sup> Meson Decay.

Fig. 1

.

n.

ŝ

MALE 67

JECLASSIFIED

.