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## Synthesis of Pleuromutilin

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## Abstract

Synthesis of a potent inhibitor of bacterial protein synthesis, pleuromutilin, is described. Assembly of the critical cyclooctane fragment relies on an oxidative ring-expansion, and complete stereochemical relay in the synthetic sequence is enabled by the judicious choice of tactics. The requisite connectivity pattern of the perhydroindanone motif is rapidly established in a sequence of cycloaddition and radical cyclization events. Application of this strategy allows for preparation of the target natural product in 16 steps from commercially available material.

Pleuromutilin (1) is a terpenoid produced by basidiomycetes from the *Clitopilus* genus and discovered during the search for new fungal metabolites with antibacterial properties (Figure 1).<sup>1</sup> This natural product exhibits potent activity against Gram-positive pathogens, which results from binding to the bacterial ribosome at a highly conserved peptidyl transferase center.<sup>2</sup> The unique mechanism of action minimizes the appearance of resistant strains and ensures a lack of cross-resistance with other inhibitors of protein synthesis.<sup>3,4</sup> Spectroscopic and degradation studies revealed a distinctive tricyclic motif of mutilin (2), which is found as a glycolate at C14 in pleuromutilin (1).<sup>5–7</sup> The ester functionality is required for biological activity and has provided a crucial handle for structural modifications during the development of semisynthetic mutilin antibiotics.<sup>8</sup> Initial efforts led to the identification of tiamulin and valnemulin for veterinary applications and retapamulin (3) for treatment of skin infections.<sup>9</sup> Recent approval of lefamulin (4) for the treatment of community-acquired bacterial pneumonia marked the first successful development of a mutilin antibiotic for systemic administration in humans.<sup>10</sup> Introduction of a substituted thioglycolate at C14 has been a hallmark of all compounds evaluated in clinical settings so far, with the remainder of

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ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.2c04708.

Experimental and computational procedures and characterization data for all new compounds (PDF)

Accession Codes

CCDC 2170014–2170016 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033. Complete contact information is available at: https://pubs.acs.org/10.1021/jacs.2c04708

the scaffold unchanged.<sup>11</sup> Modifications of the terpenoid core present a challenge due to its complex reactivity patterns. Nevertheless, new derivatives of 12-*epi*-mutilin show promise in extending the spectrum of antibacterial activity to Gram-negative pathogens.<sup>12</sup>

The combination of structural features and promising biological activity has prompted multiple investigations into the chemical synthesis of mutilin antibiotics.<sup>13,14</sup> These efforts led to identification of several strategic bonds within the terpenoid core and ultimately resulted in a diverse collection of synthetic sequences to access the target compounds. Here, we demonstrate a new approach to these antibacterial agents that permits rapid construction of the unique tricyclic framework with robust stereochemical relay throughout the synthetic sequence. Application of our strategy has secured access to mutilin (2) and pleuromutilin (1) in 15 and 16 steps, respectively, from commercially available material.

Our approach to the terpenoid core of mutilin antibiotics relied on ring-expansion of a six-membered carbocyclic fragment to access the desired cyclooctane motif (Figure 1). To accomplish the required insertion of one isoprene unit, we envisioned an annulation to form the C11–C12 and C13–C14 bonds in a fused ring system, which was expected to undergo oxidative cleavage of the C11–C14 bond and furnish the desired functionalization pattern. We also anticipated that a rapid assembly of the requisite precursor to ring-expansion would be enabled by a cycloaddition to form the C4–C5 and C9–C10 bonds and a radical cyclization to form the C5–C6 bond. Notably, the presence of ketones at C3 and C11 in the structure of intermediates was expected to allow for epimerization of critical stereocenters at C4 and C10 when found necessary for the success of the synthesis.<sup>15</sup>

Our synthesis began with Diels-Alder cycloaddition of cyclopentenone 5 and siloxydiene 6 (Scheme 1). We identified a combination of methylaluminum triflimide and a silyl triflate as the only competent catalytic system among the evaluated Lewis acids.<sup>16,17</sup> Tetrahydroindanone 7 was formed as a major product, along with the minor diastereomer resulting from *endo* approach in the cycloaddition.<sup>18</sup> The observed *exo* selectivity is noteworthy, because the configuration at C10 proved critical to successful installation of the quaternary center at C12, whereas the stereocenter at C4 was ablated at the later stages (see below). Oxidation of enol ether 7 with CAN produced unsaturated ketone 8.19 Subjection of diene 8 to the conditions of iron-catalyzed hydrogen atom transfer (HAT) resulted in formation of tricyclic motif 9, presumably through selective generation of a secondary alkyl radical from the monosubstituted alkene and subsequent Giese addition to the pendant unsaturated ketone.<sup>20–23</sup> Notably, construction of the C5–C6 bond proceeded with exceptional stereocontrol, delivering the desired scaffold as a single diastereomer in synthetically useful yield.<sup>24</sup> We found diene  $\mathbf{8}$  to be the optimal precursor en route to the tricyclic motif. Substrates containing an additional electron-withdrawing group at C14 underwent cyclization with diminished diastereoselectivity, while introduction of other substituents at the same position resulted in significant reduction of efficiency.

Elaboration of tricyclic diketone **9** began with conversion of the cyclopentanone to the corresponding silyl enol ether and subsequent alkoxycarbonylation of the remaining ketone at the less substituted position (Scheme 2).<sup>25</sup> The choice of enoxysilane as a protecting group for the carbonyl at C3 enabled functionalization at C4 at a later point in the synthesis

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and also established an optimal steric environment for alkylation of an enolate at C14. The ester at C14 proved critical to implementation of the planned ring-expansion and access to the desired functionalization pattern in the cyclooctane motif. Thus, introduction of the fused cyclobutane fragment was accomplished upon propargylation of an enolate of keto ester 10 with iodide 11 and reductive cyclization of the corresponding alkyne 12 to form allylic alcohol 13.<sup>26</sup> We found that subjection of cyclobutanol 13 to basic conditions established an equilibrium with the corresponding retro-aldol product, which could be trapped as the isolable ketene silvl acetal 14 under carefully controlled conditions.<sup>27</sup> Further addition of base directly to the reaction mixture produced an extended enolate of unsaturated ketone 14, and subsequent alkylation with methyl iodide established the desired quaternary center at C12. Notably, the alkylation proceeded in a highly diastereoselective manner, which we attribute to the effect of the stereochemical configuration at C10 on the conformation of the intermediate enolate. Our computational studies with related enol 21 and its diastereomer 22, which is epimeric at C10, suggest opposing conformational preferences in these cyclooctene motifs (Figure 2). The computed structure 21 indicates a more sterically accessible face of the enol that is consistent with the stereochemical outcome observed in our alkylation of unsaturated ketone 14, whereas epimeric structure 22 predicts undesired direction of electrophilic attack.<sup>28</sup> These results imply that the configuration at C10 obtained in the exo-selective cycloaddition may play an important role in the desired stereochemical relay during our assembly of the quaternary center at C12.

To our surprise, the ketene silyl acetal motif of intermediate **15** (see Scheme 2) proved resistant to oxidation, and efforts to install a carbonyl group at C14 led primarily to functionalization of the vinyl substituent. Furthermore, the corresponding enolate, generated upon treatment of ketene silyl acetal **15** with a source of fluoride, underwent rapid aldol cyclization to reestablish the cyclobutanol moiety, which precluded derivatization with added electrophilic reagents. In contrast to the aldol motif of intermediate **13**, the newly formed cyclobutanol containing a quaternary center at C12 did not engage in the corresponding retro-aldol reaction, but was subject to a different fragmentation pattern wherein cleavage of the C11–C12 and C13–C14 bonds was observed. For example, treatment of closely related hydroxy ester **23** with a strong base resulted in rapid extrusion of isoprene to generate the corresponding keto ester **24** after workup (eq 1).<sup>29,30</sup> We



eventually found that the carboxyl group at C14 offered a suitable handle en route to the requisite functionalization pattern of the cyclooctane fragment. Thus, formation of trione **26** 

upon photocatalyzed oxidation of acid **25** in the presence of acridinium salt **27** provided an encouraging starting point for optimization (entry 1, Table 1).<sup>31,32</sup> Ultimately, application of iridium complex **17** as a catalyst secured access to product **26** in good yield under optimized conditions (entry 2), where inclusion of TEMPO was critical to achieving high efficiency (cf. entry 3).<sup>33–35</sup> This transformation likely involves trapping of the intermediate cyclobutyl radical with oxygen or TEMPO and a Criegee- or Grob-type fragmentation to eventually generate the diketone motif.<sup>36</sup> Silver(I)-based reagents proved incompetent in the desired oxidation (e.g., entry 4) and instead appeared to preferentially engage the cyclobutanol motif in a cleavage of the C11–C12 bond.<sup>37</sup> Application of lead(IV)-based oxidants was similarly unsuccessful (e.g., entry 5).<sup>38</sup>

Implementation of our oxidative fragmentation tactics in the synthesis involved treatment of silyl ether **15** with a source of fluoride in the presence of hexachloroethane followed by photocatalyzed oxidation of the intermediate carboxylic acid **16** under the optimized conditions.<sup>24</sup> Introduction of a chlorine substituent at C4 in triketone **18** provided an important handle for correcting the stereochemical configuration at C10, which proved otherwise challenging in our investigations. Crystallographic studies of triketone **26** revealed antiperiplanar arrangement of the carbonyl at C11 and the adjacent C–H bond in the solid state, which may translate to similar conformational preferences in solution and hinder the requisite deprotonation (see eq 2).<sup>24</sup> Indeed, attempted epimerization of substrate **26** invariably resulted in facile retro-Michael fragmentation of the cyclooctanedione motif to deliver tetrahydroindanone **28**.<sup>39</sup> Variants of substrate **26** containing a protected cyclopentanone moiety were resistant to enolization at C10 even upon exposure to forcing conditions, as judged by the lack of deuterium incorporation



in the presence of isotopically labeled protic solvents. In contrast, subjection of chloroketone **18** to a base in DMSO effected Kornblum oxidation at C2, likely through the intermediacy of the corresponding oxyallyl cation, and enabled in situ epimerization to the desired configuration at C10.<sup>40</sup> We speculate that formation of the *a*-hydroxy-a, $\beta$ -unsaturated ketone retards retro-Michael fragmentation of intermediate **19** due to increased stabilization of the corresponding enolate at C4. Introduction of an sp<sup>2</sup>-hybridized center at C4 may also lead to greater conformational flexibility in the cyclooctanone motif, thereby facilitating

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epimerization at C10. Removal of the extraneous carbonyl at C2 was accomplished upon treatment with zinc, and triketone **20** was obtained with the desired configuration at C4.<sup>41–43</sup>

We found that exposure of penultimate intermediate **20** to sodium in ethanol under carefully controlled conditions resulted in reduction of ketones at C11 and C14, delivering the target compound, mutilin (**2**), in a highly chemo- and stereoselective manner.<sup>44</sup> The choice of solvent proved critical to success of this transformation, and diminished efficiency and chemoselectivity were observed with other alcohols and in the presence of aprotic cosolvents. Acylation of mutilin (**2**) with acetoxyacetic acid followed by addition of base to the reaction mixture produced pleuromutilin (**1**), completing the synthesis in 16 steps from commercially available material.<sup>45</sup>

In summary, we disclose a new synthetic approach to mutilin antibiotics that relies on ring-expansion of a readily accessible intermediate to establish the requisite tricyclic motif. Our strategy takes advantage of a robust stereochemical relay, which is enabled by the judicious choice of tactics and has led to identification of new reactivity patterns associated with the unique terpenoid core. Other salient features of our approach include an oxidative decarboxylation cascade to access the cyclooctane fragment and a cycloaddition–radical cyclization sequence to rapidly construct the ring-expansion precursor. Application of our strategy to the synthesis of mutilin and pleuromutilin delivers the target compounds in 15 and 16 steps, respectively, from commercially available starting material. Reports of modifications within the terpenoid motif that impart a promising, extended spectrum antibacterial activity to mutilin derivatives suggest that lessons learned during this study may contribute to the development of new leads in the arsenal of antibiotics.

### **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

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- (28). In the case of enols 21 and 22, the lowest energy conformers that could be expected to offer opposite facial selectivity during electrophilic attack at C12 of the ones depicted in Figure 2 were computed to be 4.67 and 1.68 kcal/mol higher in energy, respectively See the Supporting Information for details.
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pleuromutilin (1), inhibitor of bacterial protein synthesis



retapamulin (3), approved for topical treatment of infections



lefamulin (4), approved for systemic treatment of infections





annulation– oxidative fragmentation



cycloadditionradical cyclization

## Figure 1.

Representative mutilin antibiotics and our approach to the shared tricyclic motif.





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## Figure 2.

Computed lowest energy conformations of enols **21** and **22**. Computations were performed at the  $\omega$ B97X-D/6-31G(d) level of theory.



**Scheme 1.** Synthesis of Tricyclic Diketone 9



**Scheme 2.** Synthesis of Pleuromutilin (1)

#### Table 1.

Oxidative Decarboxylation en Route to the Desired Functionalization Pattern of the Cyclooctane Fragment



 $^{a}$ Based on internal standard and determined by  $^{1}$ H NMR analysis.