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Non-adiabatic few-femtosecond dynamics of excited methanol with super-resolution time-resolved photoelectron spectroscopy

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We report a novel approach to tracking nuclear motion and non-adiabatic dynamics in molecular systems, on sub-10 fs timescales, by combining a detailed analysis of the onset and decay times of the features in time-resolved photoelectron spectroscopy experiments with electronic structure calculations. The approach is applied to photoexcitation of methanol at 156 nm, followed by delayed photoionization at 260 nm, using 20 fs laser pulses. Energy-resolved signal onset times are extracted from the measured photoelectron spectra to achieve high temporal resolution, beyond the pump and probe pulse duration. When combined with ab initio calculations of selected cuts through the excited state potential energy surfaces, this information allows the dynamics of the transient excited molecule, which include multiple nuclear and electronic degrees of freedom, to be tracked on their intrinsic femtosecond timescale. Within 15 fs of photoexcitation, we observe nuclear motion on the initially excited bound state, through a region of non-adiabatic coupling, and along different dissociative coordinates.

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I. INTRODUCTION

Molecules excited by light can evolve by coupled nuclear and electronic motion over ultrafast timescales¹. In many photochemical reactions, the products and their relative yields depend sensitively on the details of these coupled molecular dynamics and processes that occur within a few femtoseconds. For example, near a conical intersection, where the evolving molecular geometry finds two or more electronic states of equal potential energy, non-adiabatic transitions can occur, enabling important relaxation mechanisms, such as competing isomerization or dissociation pathways, and non-radiative electronic relaxation^{2,3}. Considerable interest in investigating such dynamics on their natural femtosecond timescales has driven recent developments of time-resolved spectroscopic methods employing few-cycle infrared lasers and broadband attosecond pulses⁴⁻⁷.

Time-resolved photoelectron spectroscopy (TRPES) has arguably become the gold standard to study ultrafast non-adiabatic dynamics in neutral excited molecules⁸. A measurement of the instantaneous photoelectron binding energy of the excited wavepacket can be parsed into contributions from each state involved in the electronic relaxation process^{9,10}, allowing kinetic models to be constructed. Photoelectron spectra are also sensitive to excited vibrational wavepacket dynamics due to the dependence of the binding energy on the evolving molecular geometry¹¹⁻¹³. Signatures of such dynamics have been observed in TRPES as shifts^{11,14-19} in the measured photoelectron kinetic energy as the probe pulse delay is increased. Such shifts are a direct result of the wavepacket motion on the excited state potential energy surface (PES), due to the configuration dependent ionization potential.

A few techniques are available to exceed the apparent instrument resolution limit of an experiment. For example, ultrafast excited state decay lifetimes may be extracted²⁰ by measuring a slightly broadened or delayed pump-probe cross-correlation. We draw an analogy between the present work and hyperspectral super-resolution imaging, where multiple images are captured with sub-pixel shifts at various wavelengths²¹. These images can be combined to obtain a hyperspectral image with spatial resolution beyond conventional hyperspectral imaging.^{21,22} In the present experiments, photoelectron energy dependent spectra obtained with sub-pulse duration pump-probe time-delay steps spectra reveal nuclear motion, with a temporal resolution beyond the instrument response, if the characteristic onset and decay times of spectral features are appropriately mapped as functions of photoelectron kinetic energy.¹⁸

We temporally and spectrally resolve photodissociation dynamics of methanol following excitation to the $2^1A''$ (S_2) electronic state at 156 nm, in a demonstration of super-resolution TRPES to identify and follow the photochemical reaction. Using a 260 nm single-photon probe, we measure dynamical spectral features in the photoelec-

tron spectrum as a function of the pump-probe time delay, and quantify them to extract signatures of the nuclear motion that emerges in the first 10 fs following excitation. Measurements of the photoelectron kinetic energy E_{e^-} allow the determination of the transient binding energy $\mathcal{E}(\vec{\mathbf{R}})$ of the system,

$$\begin{aligned}\mathcal{E}_i(\vec{\mathbf{R}}) &= \hbar\omega_{pump} + \hbar\omega_{probe} - E_{e^-} \\ &= \hbar\omega_{pump} + E_c(\vec{\mathbf{R}}) - E_i(\vec{\mathbf{R}}),\end{aligned}\quad (1)$$

where the potential energies of the cation $E_c(\vec{\mathbf{R}})$ and the neutral excited state $E_i(\vec{\mathbf{R}})$ depend on the nuclear coordinates $\vec{\mathbf{R}}$, and $E_c(\vec{\mathbf{R}}) - E_i(\vec{\mathbf{R}})$ is the coordinate-dependent ionization potential. The pump and probe photon energies are $\hbar\omega_{pump}$ and $\hbar\omega_{probe}$, respectively. When interpreted with the aid of electronic structure calculations, which assist in reducing the dimensionality of the dynamics through the identification of a characteristic reaction coordinate, this technique enables the nuclear wavepacket to be tracked as it evolves on the PES.

Methanol plays an important role in atmospheric chemistry, fuels and energy transport^{23,24}, and the excited state dynamics of methanol are highly relevant to each of these applications. Direct investigation of excited state dynamics of methanol in the time domain has been challenging, in part due to the high electronic excitation energies, the ultrafast internuclear motion on the excited states, and the relative complexity of the molecule involving two fundamental functional groups. Methanol can therefore be employed as a model polyatomic system to develop valence-sensitive time-resolved experimental techniques and excited state electronic structure theory to address each of these challenges.

In previous studies, several neutral dissociation channels were observed following photoexcitation of the $2^1A''$ (S_2) state in methanol^{25,26}. Among these, the dominant two-body breakup processes produce either $\text{CH}_3\text{O}+\text{H}$ or CH_3+OH , while $\text{CH}_2\text{OH}+\text{H}$ and $\text{CH}_2\text{O}+\text{H}_2$ are also produced with lower yields. The S_2 excited state is bound along the O-H and C-O coordinates, while the $1^1A''$ (S_1) state is dissociative in the same coordinates²⁷. Previous calculations revealed avoided crossings between the S_1 and S_2 states along each of these coordinates²⁷, which suggested that the observed dissociation occurs after a non-adiabatic transition to the lower of these two $1^1A''$ states^{25,26}. Analysis of the fragment kinetic energy distributions for the O-H and C-O dissociation channels revealed that most of the available energy is channeled into translational kinetic energy, which also supports the possible mechanism of prompt fragmentation on the S_1 PES, which connects asymptotically to the $\text{CH}_3\text{O} + \text{H}$ and $\text{CH}_3 + \text{OH}$ thermodynamic limits^{25,26,28-30}.

In the valence isoelectronic system CH_3SH , the S_1 and S_2 surfaces were previously explored by neutral fragment velocity map imaging³¹ and ab initio theory^{2,32,33}. In CH_3SH , a conical intersection between the S_2 and S_1 PESs, involving C-S bond stretch and S-H motion, enables dissociation. In the present investigation we ex-

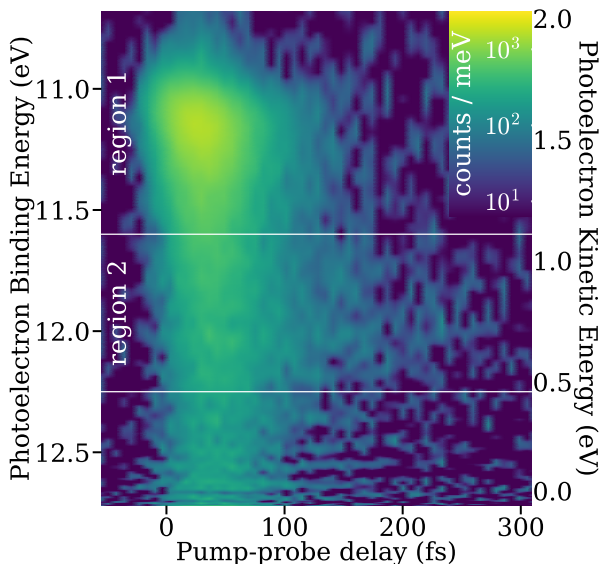


FIG. 1. Time-resolved photoelectron spectrum from ionization at 260 nm following excitation at 156 nm, taken in 6.57 fs steps, with the background due to multiphoton ionization from the pump or probe alone removed by subtracting the averaged spectra taken when the probe pulse arrives between 50 and 150 fs early. For positive time delays, the probe pulse arrives after the pump pulse. White lines partition the spectrum into regions that are discussed in the text.

explore the possibility of an analogous conical intersection between the S_2 and S_1 PESs in methanol, and address whether the nuclear dynamics following excitation to the S_2 state are mediated by S_1 , or a previously ignored $1^1A'$ (S_3) state.

II. RESULTS AND DISCUSSION

A. Signal Onset Times and Resolving a Fast Dissociative Reaction Pathway.

The experimental time-resolved photoelectron spectra, for excitation at 156 nm and ionization by a delayed 260 nm probe pulse, are shown in Fig. 1. The pump and probe pulses, each having 20 fs duration, are generated by focusing 780 nm laser pulses in argon to generate high order harmonics through the non-linear process of high harmonic generation. The high harmonic beam is split into two parallel beams containing the 3rd and the 5th harmonics by using filters and a split mirror setup. One half of the split mirror is mounted on a linear translation stage to vary the pump-probe time delay (see Methods). The overall photoelectron yield decays within ~ 200 fs. We observe features in the photoelectron spectra that shift as the time delay is increased, with a peak first appearing near the binding energy of 10.85 eV (the ionization potential, corresponding to a photoelectron kinetic energy of 1.87 eV) and extending towards higher binding

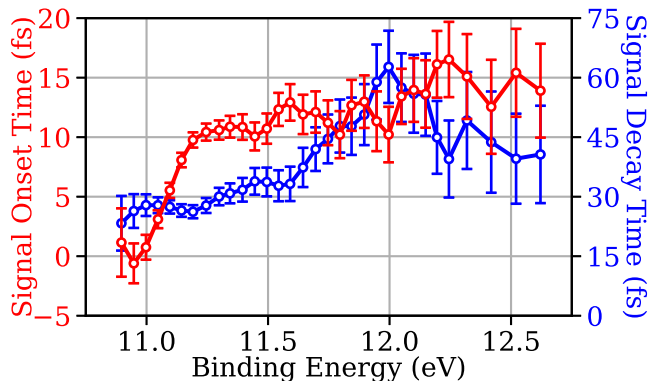


FIG. 2. Retrieved photoelectron signal onset times $t_0(\mathcal{E})$ (red) and decay times $\tau(\mathcal{E})$ (blue) for various binding energy slices extracted from the data of Fig. 1, with the error bars representing one standard deviation in the uncertainty of the fit parameters. The slices are 50 meV wide, except above 12.25 eV, where each slice is 100 meV wide.

energies (lower kinetic energies) with longer time delays. The probe pulse intensity and photon energy are carefully selected to limit access to excited electronic states of the cation (see Methods), so that the TRPES features are due to ultrafast wavepacket motion on the neutral excited states and the topology of the ground state PES of the cation.

To quantify the measured photoelectron spectra, we use a global fitting algorithm that fits each energy-slice, $I(\mathcal{E}, t)$, with a function parameterized by the energy-dependent signal onset time $t_0(\mathcal{E})$ and exponential decay time $\tau(\mathcal{E})$, convolved with a global Gaussian function g of width σ :

$$I(\mathcal{E}, t) = A(\mathcal{E}) \left[H(t - t_0(\mathcal{E})) \times \exp\left(-\frac{t - t_0(\mathcal{E})}{\tau(\mathcal{E})}\right) \right] \otimes g(t - t_0(\mathcal{E}); \sigma), \quad (2)$$

where $H(t)$ is the Heaviside step function¹⁸. More information on this fitting procedure is available in the supplementary materials. The retrieved signal onset times, $t_0(\mathcal{E})$, and signal decay times, $\tau(\mathcal{E})$, are shown in Fig. 2. Eq. 2 adds an energy-dependent framework to the TRPES technique to allow access to dynamics with time resolution beyond the duration of pump and probe pulses.

The retrieved onset times can be interpreted as the time it takes for the wavepacket to arrive at a region of the PES with a specified binding energy. **Following photoexcitation, the excited state wavepacket rapidly moves away from Franck-Condon geometries to a region of increased binding energy.** Over the first ~ 10 fs, the binding energy increases smoothly at 0.05 eV/fs. Above 11.3 eV, there is an abrupt change in this rate to 0.3 eV/fs, which continues for the remainder of the measurable binding energies up to 12.72 eV, suggesting neutral dissociation. Thus, by analyzing the time-dependence of the excited state binding energies (see onset times of Fig. 2), the

initial motion of the excited molecule can be accessed on timescales significantly shorter than the ~ 20 fs pulse widths of the pump and probe pulses.

In methanol, there are $3N - 6 = 12$ nuclear degrees of freedom. While this number is small relative to other alcohols, examining excited state dynamics in methanol is challenging due to the high dimensionality of its PESs. In the present work, we use equation-of-motion coupled-cluster theory (see Methods) to examine cuts of the PESs of the involved excited states. We define two 1-dimensional coordinates to significantly reduce the dimensionality of the system. To explain the measured photoelectron spectral shifts, we explored the simultaneous stretching of the C–O bond and the opening of the C–O–H bond angle. A reaction coordinate, representative of the general shape of this surface, was identified and is shown in Fig. 3. The initial excited wavepacket will advance on the S_2 PES along this coordinate until it reaches a crossing point between this surface and that of S_3 ($1^1A'$). At and in the neighborhood of this geometry, internal conversion to S_3 becomes possible. Starting at this geometry on S_3 , the same fragments, specifically those due to C–O and O–H cleavage, can be reached as previously considered^{25,26}. The S_3 and S_1 surfaces become degenerate at linear C–O–H geometries, however crossing to S_1 is not required for dissociation, in contrast to the valence isoelectronic system CH_3SH , and previous assumptions about methanol.

The calculations along this reaction coordinate agree well with the features of the signal onset time measurement. The abrupt increase in the rate of change of the measured binding energy discussed above is explained by the transition from the initially excited S_2 to S_3 , where the binding energy depends more strongly on the nuclear geometry. The energetic location of this transition near $\mathcal{E}(\vec{\mathbf{R}}) = 11.3$ eV is also in agreement with the calculated crossing geometry of these two states, as seen in Fig. 3. After the internal conversion to the dissociative state, the other coordinates begin to play a role. Notably, the O–H bond breaking on S_3 will lead to measured binding energy shifts faster than the other dissociation channels³⁴.

The measured signal decay times of Fig. 2 characterize the timescale for the excited state wavepacket to exit a region of the PES at a specified binding energy. For binding energies associated with S_2 , the decay times are between 25 and 30 fs, which is consistent with the ~ 26 fs lifetime that can be extracted from the ~ 100 meV wide vibrational structure seen in the S_2 absorption spectrum at 156 nm^{26,35,36}. In this case, the measurements reveal the time needed to completely depopulate the Franck-Condon region, augmenting the information provided by the signal onset times.

The previous studies of this excited state of methanol consider only the crossing between the two A'' states S_1 and S_2 ^{25,27}. A calculation on CH_3SH ³³ suggests that an analogous conical intersection could be arrived at largely through C–S (analogous to C–O) bond stretch. We performed a search for a conical intersection between S_1 and

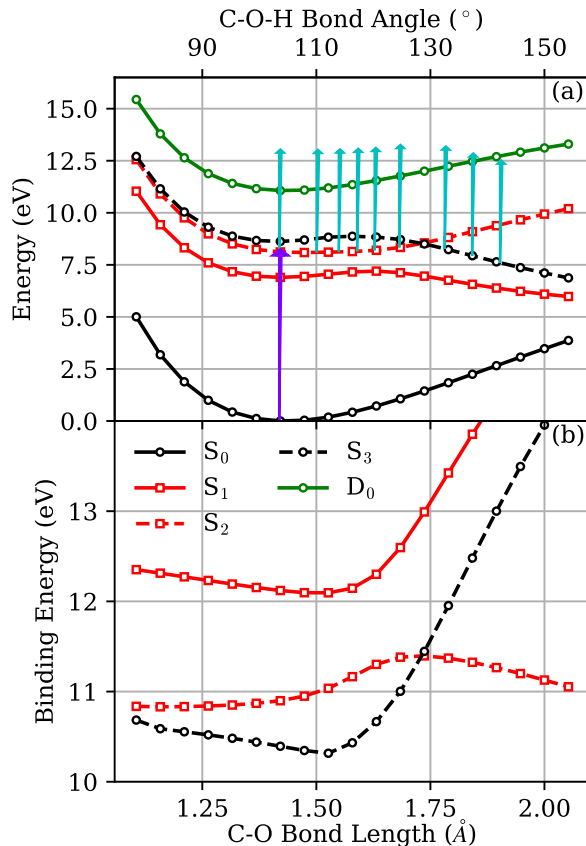


FIG. 3. Calculated cuts (a) of the potential energy surfaces and (b) the associated photoelectron binding energies for the first two A' states (black lines, S_0 and S_3), first two A'' states (red lines, S_1 and S_2), and cation ground state (green line, D_0) as the C–O bond distance and C–O–H bond angle are simultaneously stretched and opened, with arrows indicating the initial excitation (magenta) and time-delayed ionization (cyan). The ionization is shown in 2 fs steps starting at pump-probe overlap, using the experimental binding energy dependent onset times and the binding energy calculations shown here, and making the approximation that all of the measured binding energy shifts are due to motion along this coordinate. See the supplementary materials for further discussion.

S_2 in methanol, and only found one in a region of the PES that is energetically inaccessible for the present experiments. The importance of the A' state S_3 in the dissociation of methanol underlines the weak correspondence in the excited state dynamics of the two isoelectronic molecules. The previously ignored S_3 state appears to be responsible for the rapid relaxation within 15 fs observed in the present experiments, and is also consistent with previous measurements.^{25,27}

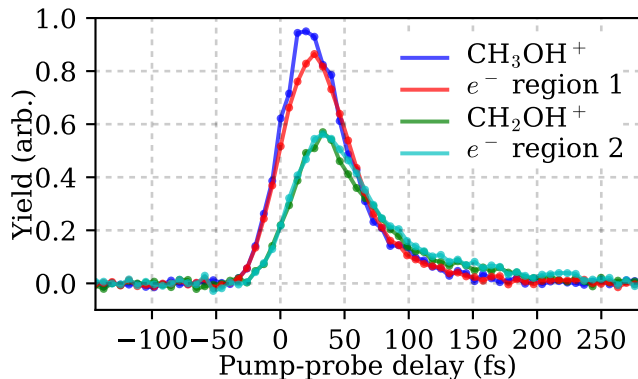


FIG. 4. Pump-probe time dependence of the yield of the various ion and photoelectron channels. In this analysis, only the total ion and photoelectron yields were normalized. The two electron regions correspond to 10.85 to 11.6 eV (red, fast decay) and 11.6 to 12.25 eV (cyan, slow decay), as labeled in Fig. 1.

B. Correlating Delayed-Onset Photoelectrons with Methyl Deprotonation.

Excited methanol fragments through multiple dissociation channels following photoexcitation. Both the $\text{CH}_2\text{OH}^+ + \text{H}$ channel and the $\text{CH}_2\text{O}^+ + \text{H}_2$ channel, with appearance energies of 11.6 eV and 12.45 eV, respectively, are energetically accessible for the photon energies of the present experiments. We measured the time-dependent ion yields for the species associated with these H and H_2 elimination channels along with the bound ionic methanol channel using the same experimental apparatus and the same 156 nm pump, 260 nm probe scheme. The results are shown in Fig. 4. A fast rise in the CH_3OH^+ yield near t_0 with a fast decay, and a slightly delayed rise in the CH_2OH^+ yield with a slower decay, are observed. Hydrogen loss in the latter channel was confirmed to occur exclusively from the methyl site of methanol by performing the same experiment with its deuterated isotopologues CD_3OH and CH_3OD . There was no significant yield of CH_2O^+ measured in the present experiments.

Fitting the time-dependent ion yields with the ansatz of Eq. 2 gives a 7 fs delay for the fragment signal relative to that of the parent ion and a 30 fs and 45 fs $1/e$ lifetime for the parent and fragmented ion, respectively. Although these ion yield measurements were not made in coincidence with the photoelectron measurements, the faster CH_3OH^+ yield correlates well with that of photoelectrons with binding energies between the ionization potential and 11.6 eV (region 1 in Fig. 1), as shown in Fig. 4, and the slower CH_2OH^+ yield correlates with that of photoelectrons between 11.6 eV and 12.25 eV (region 2 in Fig. 1), where the signal takes longer to decay (see also Fig. 2)³⁷. Note that the time dependence in each ion yield and the correlated photoelectron yield exhibits earlier (later) onset times followed by shorter (longer) decay

TABLE I. Fitted exponential decay times, in femtoseconds, of the ion and electron channels presented in Fig. 4 and in the text, for methanol and its deuterated isotopologues, with uncertainties representing one standard deviation. The two electron regions are illustrated in Fig. 1. NM indicates a channel that was not measured.

Channel	CH_3OH	CH_3OD	CD_3OH
Parent ion	30 ± 3	43 ± 6	36 ± 6
e^- region 1	29 ± 2	39 ± 9	NM
Fragment ion	45 ± 4	91 ± 8	56 ± 6
e^- region 2	48 ± 4	97 ± 16	NM

times. The fitted signal decay lifetimes for the fragment and parent ion yields, along with the corresponding photoelectron yields from each of the binding energy bands described above, are summarized in Table I for each target species. Although C–H fragmentation occurs on the methyl site of the molecule, the signal decay timescales were greatly affected by hydroxyl deuteration and less so by fully deuterating the methyl functional group, indicating that hydroxyl motion in the excited state plays an important role in facilitating the methyl hydrogen elimination, which was previously determined to occur on the cationic states²⁵. Deuteration did not strongly affect the appearance time of the fragment ion, relative to the parent ion, nor the photoelectron signal onset times, indicating that stretching of the C–O bond, rather than hydrogen motion, limits the rate of the initial dynamics.

To gain insight into the dynamics leading to methyl hydrogen loss in the present experiments, we turn to the computed 2-dimensional cuts, along the C–H stretch and C–O–H angle coordinates, of the S_2 and S_3 excited states and the lowest cation state D_0 , which are shown in Fig. 5. The cuts reveal a path for a fraction of the cation wavepacket on D_0 to dissociate by C–H bond cleavage. The pathway is reached following ionization by the probe photon for C–O–H angles $\gtrsim 140^\circ$, which are accessed through a conical intersection between the S_2 and S_3 states. A low barrier (< 400 meV) on S_2 to angle opening is consistent with the delayed onset of the CH_2OH^+ fragment ion and the correlated lower-energy photoelectrons in the present measurements on CH_3OH . Such a barrier also provides a possible explanation for the longer CH_2OD^+ fragment ion and photoelectron decay times observed in the present measurements on deuterated methanol CH_3OD . Significant potential energy barriers around 1.7 Å on both of the S_2 and S_3 cuts may inhibit direct C–H dissociation by 156 nm photoexcitation, as observed in previous experimental investigations²⁵, however motion along other coordinates reduce the barriers (see the Supplementary Material for details).

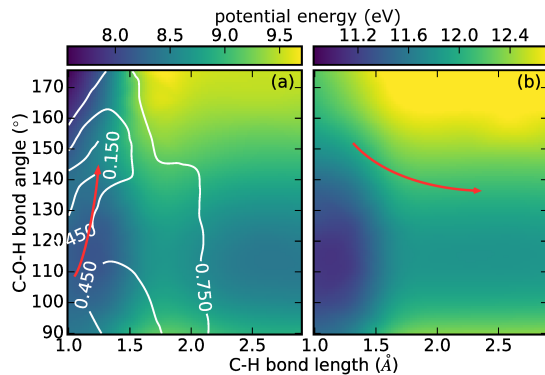


FIG. 5. 2-dimensional cuts of (a) the S_2 and S_3 excited state PES, where the lower of the two surfaces is illustrated by the color map (energy scale in eV) and the difference $|S_3-S_2|$ by white contours, and (b) the cationic state D_0 PES (energy scale in eV). The excited state wavepacket in the Franck-Condon region of S_3 experiences a potential favoring C–O–H angle opening, towards a conical intersection between S_3 and S_2 , as indicated by the red arrow on the left panel (a). A C–H dissociation path on the cation surface D_0 , indicated by a red arrow on the right panel (b), may be follow ionization of the excited neutral system by a single probe photon, for open C–O–H angles $\gtrsim 140^\circ$ and pump-probe time delays of up to 200 fs.

III. CONCLUSIONS

We have measured and described the ultrafast dynamics of methanol after 156 nm photoexcitation, and demonstrated sensitivity to the coupled electronic and nuclear motion immediately following electronic excitation, through super-resolution TRPES. By calculating cuts through the potential energy surfaces of the electronic excited states, we find mechanisms to explain the measured dissociation processes, invoking a previously unexplored A' electronic state. This state is accessed non-adiabatically through a conical intersection, and dissociates rapidly along the C–O bond. The non-adiabatic transition to the dissociative S_3 state occurs within 15 fs of excitation. C–H dissociation was observed to be delayed relative to the C–O fragmentation process in the experiments and was determined to occur on the ground electronic state of the methanol cation, following significant C–O–H angle opening, facilitated by a conical intersection between the neutral states S_3 and S_2 .

Comparison of the time-resolved measurements with the computed PES cuts enables the identification of reaction coordinates and a reduction in dimensionality of the dynamics. Resolving nuclear motion in higher dimensions through this procedure could be possible in further experiments involving photoionization to multiple electronic continua, given that the valence electron binding energies have unique dependencies on molecular configuration. These approaches could enable a more detailed understanding of non-adiabatic dynamics in electronically excited polyatomic molecules, and may also be

useful in studying highly coupled electronic nuclear motion, even on sub-10 fs timescales^{38,39}.

IV. METHODS

A. Experimental Details

The experimental setup is described in detail elsewhere^{18,40}. Briefly, light from a 25 mJ, 25 fs, 1 kHz Ti:sapphire laser system generates vacuum and extreme ultraviolet light via high harmonic generation at a 6 m focus in 25 Torr of argon in a 10 cm-long gas cell. Using a set of D-shaped filters and a split mirror interferometer, the output frequency comb is separated spectrally and temporally into a pump arm at the fifth harmonic (156 nm, 7.95 eV) and a probe arm at the third harmonic (260 nm, 4.77 eV) of the fundamental infrared light. The experiment was conducted with the pulse energy splitting between pump and probe beams at the interferometer chosen to optimize the rate of counts at the pump-probe time overlap relative to the rate when the probe arrived early, but was repeated at a different splitting to dismiss the presence of unwanted nonlinear effects. The two beams are focused into a velocity map imaging spectrometer, which records the kinetic energy spectra of either the electrons or ions resulting from the photoionization of target molecules. A set of microchannel plates and a CCD camera are used to record the projected electron and ion momentum distributions, and the pBASEX algorithm⁴¹ is used to recover the kinetic energy distributions. For ions, a fast voltage pulse applied to the detector, at specific times relative to the ionizing laser pulse, allowed different ion fragments to be isolated by their mass-to-charge ratio. Methanol vapor from a container held at room temperature flows through a variable leak valve and enters the interaction region through a long 100 μm -diameter capillary.

B. Theoretical Calculations.

We used equation-of-motion coupled-cluster theory with single and double excitations (EOM-CCSD)^{42,43} and an aug-cc-pVTZ basis set⁴⁴ to calculate the PES of the ground and first few excited states of neutral methanol. The ground state of the cation was calculated using RHF-RCCSD^{45,46}. Since methanol's excited states have significant Rydberg character, it is necessary to use augmented functions in the basis set. At geometries with stretched methanol bonds, we used unrestricted reference functions (UHF-EOM-CCSD)^{47–51} to ensure that the EOM-CCSD results were reliable and the presence of multireference correlation did not significantly change the results. The EOM-CCSD calculations were performed using the MOLPRO electronic structure suite^{52,53}, while those with an unrestricted reference were

performed using the NWChem package^{47,48}. Calculations of the two-dimensional potentials in Fig. 5 were performed by optimizing the energy on the cation surface with respect to the other internal coordinates. The geometry optimizations used multireference Rayleigh-Schrödinger perturbation theory^{54–57}, using an active space of 3 electrons in 5 orbitals. Loose geometry convergence parameters were used for some of the geometries far from equilibrium. A single-point energy calculation at the EOM-CCSD and RHF-RCCSD levels for the neutral molecule and cation, respectively, was performed on the optimized geometries.

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- ¹A. H. Zewail, “Laser femtochemistry,” *Science* **242**, 1645–1653 (1988).
- ²D. R. Yarkony, “Diabolical conical intersections,” *Reviews of Modern Physics* **68**, 985–1013 (1996).
- ³B. G. Levine and T. J. Martínez, “Isomerization Through Conical Intersections,” *Annual Review of Physical Chemistry* **58**, 613–634 (2007).
- ⁴T. Ergler, A. Rudenko, B. Feuerstein, K. Zrost, C. D. Schröter, R. Moshhammer, and J. Ullrich, “Spatiotemporal imaging of ultrafast molecular motion: Collapse and revival of the D_2^+ nuclear wave packet,” *Physical Review Letters* **97**, 193001 (2006).
- ⁵Y. Nabekawa, Y. Furukawa, T. Okino, A. A. Eilanlou, E. J. Takahashi, K. Yamanouchi, and K. Midorikawa, “Sub-10-fs control of dissociation pathways in the hydrogen molecular ion with a few-pulse attosecond pulse train,” *Nature Communications* **7**, 12835 (2016).
- ⁶P. Sándor, V. Tagliamonti, A. Zhao, T. Rozgonyi, M. Ruckebauer, P. Marquetand, and T. Weinacht, “Strong field molecular ionization in the impulsive limit: Freezing vibrations with short pulses,” *Physical Review Letters* **116**, 063002 (2016).
- ⁷M. C. E. Galbraith, C. T. L. Smeenk, G. Reitsma, A. Marciniak, V. Despré, J. Mikosch, N. Zhavoronkov, M. J. J. Vrakking, O. Kornilov, and F. Lépine, “XUV-induced reactions in benzene on sub-10 fs timescale: nonadiabatic relaxation and proton migration,” *Physical Chemistry Chemical Physics* **19**, 19822–19828 (2017).
- ⁸A. Stolow, A. E. Bragg, and D. M. Neumark, “Femtosecond time-resolved photoelectron spectroscopy,” *Chemical Reviews* **104**, 1719–1758 (2004).
- ⁹H. Satzger, D. Townsend, and A. Stolow, “Reassignment of the low lying cationic states in gas phase adenine and 9-methyl adenine,” *Chemical Physics Letters* **430**, 144–148 (2006).
- ¹⁰C. Z. Bisgaard, O. J. Clarkin, G. Wu, A. M. D. Lee, O. Geßner, C. C. Hayden, and A. Stolow, “Time-resolved molecular frame dynamics of fixed-in-space CS_2 molecules,” *Science* **323**, 1464–1468 (2009).
- ¹¹P. Hockett, E. Ripani, A. Rytwinski, and A. Stolow, “Probing ultrafast dynamics with time-resolved multi-dimensional coincidence imaging: butadiene,” *Journal of Modern Optics* **60**, 1409–1425 (2013).
- ¹²H. Marciniak and S. Lochbrunner, “On the interpretation of decay associated spectra in the presence of time dependent spectral shifts,” *Chemical Physics Letters* **609**, 184–188 (2014).
- ¹³M. H. Elkins, H. L. Williams, and D. M. Neumark, “Isotope effect on hydrated electron relaxation dynamics studied with time-resolved liquid jet photoelectron spectroscopy,” *The Journal of Chemical Physics* **144**, 184503 (2016).
- ¹⁴Y.-I. Suzuki, T. Horio, T. Fuji, and T. Suzuki, “Time-resolved photoelectron imaging of $S_2 \rightarrow S_1$ internal conversion in benzene and toluene,” *The Journal of Chemical Physics* **134**, 184313 (2011).
- ¹⁵T. Fuji, Y.-I. Suzuki, T. Horio, and T. Suzuki, “Excited-state dynamics of CS_2 studied by photoelectron imaging with a time resolution of 22 fs,” *Chemistry – An Asian Journal* **6**, 3028–3034 (2011).
- ¹⁶T. Horio, Y.-i. Suzuki, and T. Suzuki, “Ultrafast photodynamics of pyrazine in the vacuum ultraviolet region studied by time-resolved photoelectron imaging using 7.8-eV pulses,” *The Journal of Chemical Physics* **145**, 044307 (2016).
- ¹⁷T. Kobayashi, T. Horio, and T. Suzuki, “Ultrafast deactivation of the $\pi\pi^*(V)$ state of ethylene studied using sub-20 fs time-resolved photoelectron imaging,” *The Journal of Physical Chemistry A* **119**, 9518–9523 (2015).
- ¹⁸E. G. Champenois, N. H. Shivaram, T. W. Wright, C.-S. Yang, A. Belkacem, and J. P. Cryan, “Involvement of a low-lying Rydberg state in the ultrafast relaxation dynamics of ethylene,” *The Journal of Chemical Physics* **144**, 014303 (2016).
- ¹⁹T. Horio, R. Spesyvtsev, K. Nagashima, R. A. Ingle, Y.-i. Suzuki, and T. Suzuki, “Full observation of ultrafast cascaded radiationless transitions from $S_2(\pi\pi^*)$ state of pyrazine using vacuum ultraviolet photoelectron imaging,” *The Journal of Chemical Physics* **145**, 044306 (2016).
- ²⁰P. Farmanara, O. Steinkellner, M. T. Wick, M. Wittmann, G. Korn, V. Stert, and W. Radloff, “Ultrafast internal conversion and photodissociation of molecules excited by femtosecond 155 nm laser pulses,” *The Journal of Chemical Physics* **111**, 6264–6270 (1999).
- ²¹P. Ghamisi, N. Yokoya, J. Li, W. Liao, S. Liu, J. Plaza, B. Rasti, and A. Plaza, “Advances in Hyperspectral Image and Signal Processing: A Comprehensive Overview of the State of the Art,” *IEEE Geoscience and Remote Sensing Magazine* **5**, 37–78 (2017).
- ²²Q. Dai, E. Pouyet, O. Cossairt, M. Walton, and A. K. Katsaggelos, “Spatial-Spectral Representation for X-Ray Fluorescence Image Super-Resolution,” *IEEE Transactions on Computational Imaging* **3**, 432–444 (2017).
- ²³G. A. Olah, “Beyond Oil and Gas: The Methanol Economy,” *Angewandte Chemie International Edition* **44**, 2636–2639 (2005).
- ²⁴J. Kothandaraman, A. Goeppert, M. Czaun, G. A. Olah, and G. K. S. Prakash, “Conversion of CO_2 from air into methanol using a polyamine and a homogeneous ruthenium catalyst,” *Journal of the American Chemical Society* **138**, 778–781 (2016).
- ²⁵S. Harich, J. J. Lin, Y. T. Lee, and X. Yang, “Photodissociation Dynamics of Methanol at 157 nm,” *The Journal of Physical Chemistry A* **103**, 10324–10332 (1999).
- ²⁶B.-M. Cheng, M. Bahou, W.-C. Chen, C.-h. Yui, Y.-P. Lee, and L. C. Lee, “Experimental and theoretical studies on vacuum ultraviolet absorption cross sections and photodissociation of CH_3OH , CH_3OD , CD_3OH , and CD_3OD ,” *The Journal of Chemical Physics* **117**, 1633–1640 (2002).
- ²⁷R. J. Buenker, G. Olbrich, H. P. Schuchmann, B. L. Schuermann, and C. Von Sonntag, “Photolysis of methanol at 185 nm. Quantum-mechanical calculations and product study,” *Journal of the American Chemical Society* **106**, 4362–4368 (1984).
- ²⁸K.-j. Yuan, Y. Cheng, F.-y. Wang, and X.-m. Yang, “Photodissociation Dynamics of Methanol and Ethanol at 157 nm,” *Chinese Journal of Chemical Physics* **21**, 301 (2008).
- ²⁹J. G. Philis, “Resonance-enhanced multiphoton ionization spectra of jet-cooled methanol and ethanol,” *Chemical Physics Letters* **449**, 291–295 (2007).

- ³⁰Z. Chen, A. T. J. B. Eppink, B. Jiang, G. C. Groenenboom, X. Yang, and D. H. Parker, "Product pair correlation in CH_3OH photodissociation at 157 nm: the $\text{OH} + \text{CH}_3$ channel," *Physical Chemistry Chemical Physics* **13**, 2350–2355 (2011).
- ³¹J. G. Izquierdo, G. A. Amaral, F. Ausfelder, F. J. Aoiz, and L. Bañares, "Velocity map imaging study of the photodissociation of CH_3SH : Internal energy distribution of the SH fragment," *ChemPhysChem* **7**, 1682–1686 (2006).
- ³²J. E. Stevens, H. W. Jang, L. J. Butler, and J. C. Light, "An adiabatic model for the photodissociation of CH_3SH in the first ultraviolet absorption band," *The Journal of Chemical Physics* **102**, 7059–7069 (1995).
- ³³D. R. Yarkony, "On the role of conical intersections of two potential energy surfaces of the same symmetry in photodissociation. I. $\text{CH}_3\text{SH} \rightarrow \text{CH}_3\text{S} + \text{H}$ and $\text{CH}_3 + \text{SH}$," *The Journal of Chemical Physics* **100**, 3639–3644 (1994).
- ³⁴C. C. Marston, K. Weide, R. Schinke, and H. U. Suter, "Product selectivity of vibrationally mediated photofragmentation of methanol," *The Journal of Chemical Physics* **98**, 4718–4727 (1993).
- ³⁵E. Sominska and A. Gedanken, "The absorption spectrum of a supersonically expanded beam of methanol in the vacuum ultraviolet region," *Journal of Molecular Spectroscopy* **175**, 234–238 (1996).
- ³⁶K. Varela, L. R. Hargreaves, K. Ralps, M. A. Khakoo, C. Winstead, V. McKoy, T. N. Rescigno, and A. E. Orel, "Excitation of the 4 lowest electronic transitions in methanol by low-energy electrons," *Journal of Physics B: Atomic, Molecular and Optical Physics* **48**, 115208 (2015).
- ³⁷W. K. Peters, D. E. Couch, B. Mignolet, X. Shi, Q. L. Nguyen, R. C. Fortenberry, H. B. Schlegel, F. Remacle, H. C. Kapteyn, M. M. Murnane, and W. Li, "Ultrafast 25-fs relaxation in highly excited states of methyl azide mediated by strong nonadiabatic coupling," *Proceedings of the National Academy of Sciences*, 201712566 (2017).
- ³⁸F. Calegari, D. Ayuso, A. Trabattori, L. Belshaw, S. D. Camillis, S. Anumula, F. Frassetto, L. Poletto, A. Palacios, P. Decleva, J. B. Greenwood, F. Martín, and M. Nisoli, "Ultrafast electron dynamics in phenylalanine initiated by attosecond pulses," *Science* **346**, 336–339 (2014).
- ³⁹M. Vacher, M. J. Bearpark, M. A. Robb, and J. P. Malhado, "Electron dynamics upon ionization of polyatomic molecules: Coupling to quantum nuclear motion and decoherence," *Physical Review Letters* **118**, 083001 (2017).
- ⁴⁰N. Shivaram, E. G. Champenois, J. P. Cryan, T. Wright, T. Wingard, and A. Belkacem, "Focal overlap gating in velocity map imaging to achieve high signal-to-noise ratio in photo-ion pump-probe experiments," *Applied Physics Letters* **109**, 254101 (2016).
- ⁴¹G. A. Garcia, L. Nahon, and I. Powis, "Two-dimensional charged particle image inversion using a polar basis function expansion," *Review of Scientific Instruments* **75**, 4989–4996 (2004).
- ⁴²T. Korona and H.-J. Werner, "Local treatment of electron excitations in the EOM-CCSD method," *The Journal of Chemical Physics* **118**, 3006–3019 (2003).
- ⁴³H. J. Monkhorst, "Calculation of properties with the coupled-cluster method," *International Journal of Quantum Chemistry* **12**, 421–432 (1977).
- ⁴⁴T. H. Dunning, "Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen," *The Journal of Chemical Physics* **90**, 1007–1023 (1989).
- ⁴⁵P. J. Knowles, C. Hampel, and H. Werner, "Coupled cluster theory for high spin, open shell reference wave functions," *The Journal of Chemical Physics* **99**, 5219–5227 (1993).
- ⁴⁶P. J. Knowles, C. Hampel, and H.-J. Werner, "Erratum: "Coupled cluster theory for high spin, open shell reference wave functions" [*J. Chem. Phys.* **99**, 5219 (1993)]," *The Journal of Chemical Physics* **112**, 3106–3107 (2000).
- ⁴⁷M. Valiev, E. J. Bylaska, N. Govind, K. Kowalski, T. P. Straatsma, H. J. J. Van Dam, D. Wang, J. Nieplocha, E. Apra, T. L. Windus, and W. A. de Jong, "NWChem: A comprehensive and scalable open-source solution for large scale molecular simulations," *Computer Physics Communications* **181**, 1477–1489 (2010).
- ⁴⁸S. Hirata, "Symbolic algebra in quantum chemistry," *Theoretical Chemistry Accounts* **116**, 2–17 (2006).
- ⁴⁹D. C. Comeau and R. J. Bartlett, "The equation-of-motion coupled-cluster method. Applications to open- and closed-shell reference states," *Chemical Physics Letters* **207**, 414–423 (1993).
- ⁵⁰R. J. Bartlett and G. D. Purvis, "Many-body perturbation theory, coupled-pair many-electron theory, and the importance of quadruple excitations for the correlation problem," *International Journal of Quantum Chemistry* **14**, 561–581 (1978).
- ⁵¹R. J. Bartlett and G. D. P. III, "Molecular applications of coupled cluster and many-body perturbation methods," *Physica Scripta* **21**, 255 (1980).
- ⁵²H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz, P. Celani, T. Korona, R. Lindh, A. Mitrushenkov, G. Rauhut, K. R. Shamasundar, T. B. Adler, R. D. Amos, A. Bernhardtson, A. Berning, D. L. Cooper, M. J. O. Deegan, A. J. Dobbyn, F. Eckert, E. Goll, C. Hampel, A. Hesselmann, G. Hetzer, T. Hrenar, G. Jansen, C. Köppl, Y. Liu, A. W. Lloyd, R. A. Mata, A. J. May, S. J. McNicholas, W. Meyer, M. E. Mura, A. Nicklass, D. P. O'Neill, P. Palmieri, D. Peng, K. Pflüger, R. Pitzer, M. Reiher, T. Shiozaki, H. Stoll, A. J. Stone, R. Tarroni, T. Thorsteinsson, and M. Wang, "MOLPRO, version 2012.1, a package of ab initio programs," (2012), see <http://www.molpro.net/>.
- ⁵³H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, and M. Schütz, "Molpro: a general-purpose quantum chemistry program package," *Wiley Interdisciplinary Reviews: Computational Molecular Science* **2**, 242–253 (2012).
- ⁵⁴H.-J. Werner, "Third-order multireference perturbation theory The CASPT3 method," *Molecular Physics* **89**, 645–661 (1996).
- ⁵⁵P. Celani and H.-J. Werner, "Multireference perturbation theory for large restricted and selected active space reference wave functions," *The Journal of Chemical Physics* **112**, 5546–5557 (2000).
- ⁵⁶H. Werner and P. J. Knowles, "A second order multiconfiguration SCF procedure with optimum convergence," *The Journal of Chemical Physics* **82**, 5053–5063 (1985).
- ⁵⁷P. J. Knowles and H.-J. Werner, "An efficient second-order MC SCF method for long configuration expansions," *Chemical Physics Letters* **115**, 259–267 (1985).