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Lignin depolymerization by fungal secretomes and a microbial sink[†]

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In Nature, powerful oxidative enzymes secreted by white rot fungi and some bacteria catalyze lignin depolymerization and some microbes are able to catabolize the resulting aromatic compounds as carbon and energy sources. Taken together, these two processes offer a potential route for microbial valorization of lignin. However, many challenges remain in realizing this concept, including that oxidative enzymes responsible for lignin depolymerization also catalyze polymerization of low molecular weight (LMW) lignin. Here, multiple basidiomycete secretomes were screened for ligninolytic enzyme activities in the presence of a residual lignin solid stream from a corn stover biorefinery, dubbed DMR-EH (Deacetylation, Mechanical Refining, and Enzymatic Hydrolysis) lignin. Two selected fungal secretomes, with high levels of laccases and peroxidases, were utilized for DMR-EH lignin depolymerization assays. The secretome from Pleurotus eryngii, which exhibited the highest laccase activity, reduced the lignin average molecular weight (M_w) by 63% and 75% at pH 7 compared to the M_w of the control treated at the same conditions and the initial DMR-EH lignin, respectively, and was applied in further depolymerization assays as a function of time. As repolymerization was observed after 3 days of incubation, an aromatic-catabolic microbe (Pseudomonas putida KT2440) was incubated with the fungal secretome and DMR-EH lignin. These experiments demonstrated that the presence of the bacterium enhances lignin depolymerization, likely due to bacterial catabolism of LMW lignin, which may partially prevent repolymerization. In addition, proteomics was also applied to the P. eryngii secretome to identify the enzymes present in the fungal cocktail utilized for the depolymerization assays, which highlighted a significant number of glucose/ methanol/choline (GMC) oxidoreductases and laccases. Overall, this study demonstrates that ligninolytic enzymes can be used to partially depolymerize a solid, high lignin content biorefinery stream and that the presence of an aromatic-catabolic bacterium as a "microbial sink" improves the extent of enzymatic lignin depolymerization.

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Introduction

Biological depolymerization of lignin has been studied extensively for the last several decades, built upon the discovery of several classes of powerful oxidative enzymes secreted by white rot fungi. These enzymes include laccases, oxidases such as aryl-alcohol oxidases (AAO) and glyoxal oxidases, and peroxidases such as manganese peroxidases (MnP), lignin peroxidases (LiP), versatile peroxidases (VP), and dye-decolorizing peroxidases (DyP).^{1,2} Most of these enzymes exhibit wide substrate specificity and utilize radical intermediates for lignin depolymerization. More recently, the discovery of enzymes that are specific to particular linkages and stereochemistry has further broadened the scope of the enzymatic systems that



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microbes employ to depolymerize lignin.^{3–5} Additionally, the realization that some bacteria can also secrete many of the same types of enzymes has indicated that biological lignin depolymerization is not limited to fungi.^{6,7} Despite these efforts, the full extent of how ligninolytic enzymes work synergistically to depolymerize lignin, however, still remains somewhat unclear, which in turn limits our ability to rationally employ these powerful enzymes for industrial lignin depolymerization.⁸ Several recent studies coupling proteomic analysis of fungal secretomes with detailed lignin characterization are providing deeper insights into how these interactions occur in Nature.^{9,10}

Using enzymes to depolymerize lignin would ideally enable both the ability to remove lignin from biomass - a common objective in overcoming recalcitrance - and also produce aromatic compounds that could be isolated and converted to bioproducts - an emerging objective in biorefineries to improve overall process economics.¹¹ Despite significant promise, very few studies have effectively demonstrated robust lignin depolymerization using enzyme cocktails or isolated ligninolytic enzymes.¹²⁻¹⁵ Challenges in this vein are many, including the fact that most oxidative enzymes used to depolymerize lignin are also capable of polymerizing aromatic compounds.^{12,16} Moreover, obtaining lignin that maintains the natural linkages and molecular weight of the macromolecule in the plant cell wall is challenging, which can lead to results that do not reflect natural lignin breakdown processes. Indeed, many commercial lignins, such as Kraft lignin, are often heavily processed so that many natural linkages, including abundant ether bonds, have been cleaved and more refractory C-C bonds retained.^{17,18} These factors, and others, have led to significant differences in the literature, making it challenging to compare between studies wherein enzymes are examined for lignin depolymerization.

Perhaps given the limited understanding of how ligninolytic enzymes function on natural substrates, most applications to date for these enzymes (e.g. laccases) are focused on their use as biocatalysts to bleach lignin-rich substrates for pulping processes,19-22 as a pretreatment in bioethanol production,^{13,23} as a proposed detoxification tool in ethanol fermentation via polymerization of phenolic compounds,²⁴ or in wastewater treatment.^{25,26} However, in emerging biorefinery applications for lignin valorization, it would be ideal to depolymerize lignin to low molecular weight (LMW) compounds for the purpose of recovering and upgrading to bioproducts.^{8,27,28} To effectively upgrade lignin to a single compound will require processes able to overcome the inherent heterogeneity of lignin. It has been recently proposed that aromatic catabolic microbes may be a potential solution for this problem via upper pathways that are able to convert heterogeneous slates of aromatic compounds into single intermediates, such as protocatechuate or catechol, and subsequently to central carbon metabolism.^{8,27} A related hypothesis to how this concept occurs in Nature is that extracellular oxidative enzymes secreted by white rot fungi and bacteria produce a pool of available aromatic carbon, and, in response, microbes evolved pathways to utilize these compounds as carbon and energy sources.⁷ These microbes thus operate as a "microbial sink" which could prevent extracellular repolymerization of LMW aromatic compounds (Fig. 1).

The aim of the current study is to examine how natural cocktails of ligninolytic enzymes depolymerize lignin in a solid, biorefinery substrate, and how the presence or absence of a microbe affects lignin depolymerization. The substrate of interest in this study originates from an emerging pretreatment that employs Deacetylation - mild alkaline treatment followed by Mechanical Refining and Enzymatic Hydrolysis (DMR-EH).^{29,30} This process generates high vields of hemicellulose and cellulose-derived sugars when used in combination with modern (hemi)cellulase cocktails, and is predicted to be cost competitive with standard deconstruction processes.³¹ Given that this substrate has not experienced high temperatures or harsh chemical treatment that targets ether bonds, we hypothesized that the resulting lignin from this process will maintain many of the linkages present in native corn stover. In the current work, DMR-EH lignin is comprehensively characterized and used with 12 different fungal cultures to induce the production of ligninolytic enzymes. From these assays, a laccase-rich and a peroxidaserich secretome were selected to examine in vitro depolymerization of DMR-EH lignin. For the laccase-rich cocktail from Pleurotus eryngii, which exhibited the highest extent of depolymerization, a proteomic analysis was also conducted to identify the enzymes responsible for the observed activity. Lastly, to test the hypothesis that the presence of a microbe can increase the extent of lignin depolymerization, the P. eryngii secretome and Pseudomonas putida KT2440 were combined, and the results indicate enhanced lignin depolymerization with the microbe present, demonstrating the concept of "microbial sink".



Fig. 1 Microbial sink concept. The presence of bacteria with ligninolytic enzymes, which are able to depolymerize high molecular weight (HMW) lignin, may act as a sink to avoid repolymerization of low molecular weight (LMW) aromatic compounds and enhance biological lignin depolymerization.

Results

Characterization of a biorefinery stream with native lignin-like properties

In the current study, the lignin-rich solid stream from the DMR-EH process was used as the solid substrate. DMR-EH lignin was characterized and compared to native corn stover *via* several methods. A full compositional analysis revealed that the lignin content in the residual DMR-EH solids is 60.0 wt% compared to the lignin in native corn stover at 14.8 wt% (Table 1). Residual polysaccharides from cellulose

Table 1Compositional analysis data of DMR-EH from corn stover. Theamounts are shown as wt% on a dry weight basis

	Corn stover (%)	DMR-EH
Lignin	14.80	60.00
Ash	2.54	2.18
Glucan	36.48	9.24
Xylan	30.18	9.36
Galactan	1.76	1.04
Arabinan	3.52	1.62
Acetate	2.71	0.72
Proteins	NA	6.07
Total mass balance	91.99	90.23

NA: not analyzed.

and hemicellulose were the second main component of this substrate, comprising 21.3 wt%, significantly lower than that in native corn stover (77%). These results demonstrate how DMR pretreatment generates a highly digestible substrate, enabling effective sugar recovery.

Gel permeation chromatography (GPC) was conducted to measure the molecular weight distribution, shown in Fig. 2A. It is noteworthy that considering both sample preparation and GPC method, which employs a diode array detector with absorbance at 260 ± 40 nm, only lignin should be observed in this analysis (see Materials and methods). The GPC profile reveals that DMR-EH exhibits a large amount of high molecular weight (HMW) lignin (>~1300 Da), with an average molecular weight (M_w) of 9200 Da. When analyzing GPC data, three additional parameters are also reported: the number average molecular weight (M_n) , peak molecular weight (M_p) and polydispersity (PD). The $M_{\rm p}$ of DMR-EH lignin is 2100 Da. The $M_{\rm p}$ is 3900 Da, indicating that the most abundant molecule is 3900 Da, and the PD value reflects the distribution of molecular weight. High values for PD (e.g., >3), as the observed in the current substrate (PD 4.4), indicates that the molecular weight distribution of DMR-EH lignin is broad. These four parameters $M_{\rm w}$, $M_{\rm n}$, $M_{\rm p}$, and PD will be presented in this study.

A 2D-nuclear magnetic resonance (NMR) spectrum of DMR-EH was also collected to obtain chemical information



Fig. 2 Characterization of DMR-EH lignin. (A) GPC chromatogram of DMR-EH. LMW = low molecular weight; HMW = high molecular weight. M_{p} , M_{n} , M_{w} , and PD values correspond to DMR-EH lignin. (B) Structures identified in lignin. H: *p*-hydroxyphenyl units, G: guaiacyl units; S: syringyl units; A: β-O-4 alkyl aryl ethers; PCA: *p*-coumaric acid; FA: ferulic acid. (C) Expanded HSQC NMR spectrum of DMR-EH lignin side chain region (δ_C/δ_H 100–160/5.5–9.0 ppm). Solvent: d_6 -DMSO : d_5 -pyridine (4 : 1, v/v).

related to the lignin composition and interunit linkages and remaining carbohydrates. The Heteronuclear Single Quantum Coherence (HSQC) NMR spectrum consists of three regions: the saturated non-oxygenated aliphatic region ($\delta_{\rm C}/\delta_{\rm H}$ 10-40/ 0.5-3.0 ppm), the lignin side chain and carbohydrate region $(\delta_{\rm C}/\delta_{\rm H}$ 50–110/2.8–5.7 ppm), and the aromatic region $(\delta_{\rm C}/\delta_{\rm H})$ 100-165/6.3-8.0 ppm). The expanded spectra for the side chain and aromatic regions are shown in Fig. 2C and D, respectively. The typical lignin units and linkages in herbaceous biomass are shown in Fig. 2B (H: p-hydroxyphenyl units, G: guaiacyl units; S: syringyl units, A: β-O-4 alkyl aryl ethers; PCA: p-coumaric acid; FA: ferulic acid). Signals were assigned from previously reported HSQC data.³²⁻³⁵ In the HSQC spectrum in the side chain region (Fig. 2C), signals at $\delta_{\rm C}/\delta_{\rm H}$ 70–73/ 4.8-5.3 and 81-83/4.3-4.5 ppm are assigned to the C α /H α and $C\beta/H\beta$ positions in β -O-4 structures. A methoxyl peak appears at $\delta_{\rm C}/\delta_{\rm H}$ 54–57/3.1–4.5 ppm. Anomeric carbon peaks in the carbohydrates are present at $\delta_{\rm C}/\delta_{\rm H}$ 92–110/4.1–5.6 ppm. Broad peaks in the region of $\delta_{\rm C}/\delta_{\rm H}$ 58–80/2.9–4.6 ppm are also from carbohydrates. Few correlation peaks of other interunit linkages (resinol, phenylcoumaran, spirodienone) were detected, meaning that the DMR-EH lignin exhibits mainly β-O-4 structures. In the HSQC spectrum of the DMR-EH lignin aromatic region (Fig. 2D), peaks of H, G, S, PCA, and FA units were observed. Peaks in main units of H, G, and S appeared as follows: the ¹³C–¹H correlation for S2/6 was at $\delta_{\rm C}/\delta_{\rm H}$ 104.1/6.7, G2 at $\delta_{\rm C}/\delta_{\rm H}$ 110.6/7.0, and G5/6 at 113-120/6.6-7.1 ppm, respectively. The correlations of C2/6 and C β in PCA units were found at $\delta_{\rm C}/\delta_{\rm H}$ 130/7.48 and 114.0/6.4 ppm, respectively. The resonance from C β in FA is minimal at $\delta_{\rm C}/\delta_{\rm H}$ 115/6.38 ppm. The signals from the Ca positions in PCA and FA overlapped at $\delta_{\rm C}/\delta_{\rm H}$ 144.7/7.6–7.4.

Screening of oxidoreductase activity in the secretome of 12 white-rot fungi

Oxidoreductases are the main oxidative enzymes involved in lignin depolymerization by fungi.¹ As such, the activities of diverse oxidoreductases were evaluated in the secretomes of twelve white-rot fungi grown in submerged cultures in the presence of DMR-EH lignin. The fungi selected for this study were Bjerkandera sp., Bjerkandera adusta, Cerioporopsis subvermispora, Irpex lacteus, Panus tigrinus, Phanerochaete chrysosporium, Phellinus robustus, Pleurotus eryngii, Pleurotus ostreatus, Polyporus alveolaris, Stereum hirsutum, and Trametes versicolor. Spectrophotometric (visible and UV) assays were used to evaluate the activity of the following enzymes: laccases, AAO, and various peroxidases such as LiP, MnP, VP, and DyP. Overall, the fungi exhibit substantial differences in both enzyme diversity and time-dependent activity profiles in the assay conditions. Fig. 3 shows the results from the six species that displayed the highest laccase and/or peroxidase activities from the twelve fungi (Fig. S1[†]). P. eryngii, for instance, produced the highest levels of laccase, up to 7 U mL⁻¹, between 7 and 9 days of incubation (Fig. 3D). AAO was also detected in the same secretome, peaking at 7 days and reaching 145 mU mL^{-1} . Conversely, Bjerkandera sp., mainly produced peroxidases



Fig. 3 Detection of oxidoreductases in the secretome of six white-rot fungi. Enzymes activities are presented for the fungi (A) *Bjerkandera* sp., (B) *P. tigrinus*, (C) *P. robustus*, (D) *P. eryngii*, (E) *P. ostreatus*, and (F) *T. versicolor*. Oxidoreductase activity was followed by the oxidation of different substrates (see Materials and methods). It is noteworthy that the oxidation of the same substrate can be the result of the action of different enzymes. In brief, the substrates and pH utilized for the enzyme screening were: 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulfonic acid) (ABTS) at pH 5 for laccases; veratryl alcohol at pH 5 for AAO; MnSO₄ with H₂O₂ at pH 5 for both VP and DyP; and ABTS with MnSO₄ and H₂O₂ at pH 5 for VP, DyP, and short-MnP.³⁸ Laccase activity was subtracted from peroxidase activity when both enzymes were present and ABTS was used as substrate.

throughout the incubation, reaching values up to 1.5 U mL^{-1} (Fig. 3A). To ascertain the type of peroxidases secreted by this fungus, enzyme assays using different dyes were employed. Discoloration was positive for an azo-dye, but not an anthraquinone dye, suggesting that it is not a DyP-type peroxidase but most likely a VP, although anthraquinone discoloration has been also recently reported for this type of enzyme.³⁶ AAO was also detected in the *Bjerkandera* sp. secretome, peaking at 5 days at 65 mU mL⁻¹. *P. tigrinus* (Fig. 3B), *P. robustus* (Fig. 3C), *P. ostreatus*, and *T. versicolor* (Fig. 3E) produced both laccases and peroxidases throughout the incubation, but at levels lower than reported above. As such, *P. eryngii* and *Bjerkandera* sp. were selected to produce larger volumes of secretome to perform depolymerization assays on DMR-EH lignin.

Depolymerization of DMR-EH lignin with fungal secretomes containing high laccase and peroxidase activity

Lignin depolymerization assays were performed at pH 4.5 and 7. The former pH was selected because it is close to the optimal for standard ligninolytic enzyme function³⁷ and the latter because lignin is well known to be more soluble at higher pH. Moreover, pH 7 was also chosen as it is close to the optimal for typical bacterial growth, in preparation for the microbial sink experiments reported below. Laccases from *P. eryngii* exhibit approximately 3-fold lower activity at pH 7 than 4.5. Peroxidase activity from the *Bjerkandera* sp. secretome was only detected at pH 4.5, and as such depolymerization assays were only performed at pH 4.5 (with H₂O₂ and Mn²⁺) for this cocktail. Different enzymes dosages were also tested, expressed as U of enzyme (at the corresponding pH and using ABTS as substrate) per g of DMR-EH.

GPC was utilized to analyze lignin depolymerization in the insoluble fraction of DMR-EH after 3 and 6 days of enzyme treatment (Fig. 4, Table S1†). The control samples (non-enzymatically treated and incubated during 6 days in the same conditions than the enzyme treated samples) exhibit different molecular weights depending on the pH; namely the lignin M_w is approximately 50% higher at pH 4.5 (12 000 Da) (Fig. 4A) than at pH 7 (6100 Da) (Fig. 4B). GPC parameters from both controls at pH 4.5 in the presence (Fig. 4C and D) and the absence of H_2O_2 and Mn^{2+} (Fig. 4A) were similar.

For the enzyme-treated samples, generally, higher enzyme dosages produced greater decreases of lignin Mw. The lone exception to this is the treatment from the Bjerkandera sp. secretome, which increased M_w at the lowest enzyme dosage compared to the control. Peroxidases from Bjerkandera sp. (80 U g⁻¹) decreased lignin $M_{\rm w}$ by 54% at 6 days, while *P. eryngii* laccases (100 U g⁻¹) decreased lignin $M_{\rm w}$ by 55% and 63% at pH 4.5 and 7, respectively, after 3 days of incubation. However, laccases and peroxidases combined at pH 4.5 at the highest dosages decreased $M_{\rm w}$ by 84%, reaching the lowest lignin $M_{\rm w}$ in the current study (2000 Da) (Fig. 4D). The most dramatic changes, in terms of release of LMW lignin species, were observed in treatments with the P. eryngii secretome at pH 7 and 30 U g^{-1} at both 3 and 6 days but also when utilizing mixed secretomes at pH 4.5. It is also notable that treatments at 3 days produced lower lignin $M_{\rm w}$ than at 6 days when the P. eryngii secretome was present - most likely due to repolymerization. Based on the results in Fig. 4, we selected the treatments at pH 7 with the P. eryngii secretome for further examination of lignin depolymerization since (1) these treatments produce the second lowest lignin Mw, releasing considerable LMW lignin compounds, (2) laccases do not require addition of co-factors such as H_2O_2 and Mn^{2+} , and (3) pH ~7



Fig. 4 GPC results from DMR-EH lignin before and after the enzyme treatments with the fungal secretomes at 3 and 6 days. Graphs show the results from treatments with the *P. eryngii* secretome at (A) pH 4.5 and (B) pH 7, treatments with (C) *Bjerkandera* sp. secretome at pH 4.5, and treatments with (D) both fungal secretomes at pH 4.5. Enzyme treatments on DMR-EH lignin were conducted over 3 (dashed lines) and 6 (solid lines) days at 30 °C and 150 rpm. H_2O_2 and Mn^{2+} were added periodically to treatments containing peroxidases and to the corresponding controls (CTL). Enzyme dosages are expressed as U of enzyme per gram of DMR-EH lignin. Laccase (LA) and peroxidase (PE) activities were calculated following the oxidation of ABTS at the corresponding pH in the absence or the presence of Mn^{2+} and H_2O_2 , respectively. The asterisk (*) highlights 3-day treatments.

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was required for further complementary bacterial assays. Moreover, due to results found at 3 days of treatment compared to 6 days, the incubation time was shortened.

DMR-EH lignin depolymerization at pH 7 as a function of time with the *P. eryngii* secretome

Subsequently, an experiment was performed over 4 days with the P. eryngii secretome to understand how depolymerization occurs as a function of time at pH 7. DMR-EH lignin was treated with similar laccase dosages from the previous section $(3, 10, \text{ and } 30 \text{ U g}^{-1})$ and the insoluble fraction analyzed by GPC every 24 h. Different to the previous experiment, DMR-EH lignin was autoclaved before the enzyme treatments since, for further experiments in the presence of bacteria, substrate sterilization is required; this is the main difference between the experiments shown in Fig. 4 and 5 and is likely responsible for the differences in molecular weight changes. In addition, for comparison purposes, a parallel experiment was conducted with a dialyzed secretome to remove possible aromatics, growth media, or fungal metabolites remaining in the secretome that could interfere in the depolymerization assays or later in the bacterial treatments.

Fig. 5(A–C) and Table S2[†] shows the GPC profiles and other GPC parameters from this experiment. There are not substantial differences among the treatments using non-dialyzed and dialyzed secretomes (Fig. 5B and C). In all three cases, the depolymerization trends exhibit an early release of LMW lignin species and a corresponding reduction of HMW lignin until the third day of treatment, followed by an increase in

HMW lignin at 4 days. These trends were observed in all the treatments, but are more evident when using the highest laccase dosage (Fig. 5D), especially in the release of LMW lignin species (<1000 Da). Fig. 5D illustrates the $M_{\rm w}$ of DMR-EH lignin and presents all the parameters obtained from GPC in those treatments with the non-dialyzed secretome. Higher enzyme dosages resulted in a higher extent of depolymerization. The lignin $M_{\rm w}$ decreased compared to the control (at 4 days and with inactivated secretome) by 25, 37, and 45% when using 3, 10, and 30 U of laccase per g of substrate at 3 days, respectively. It is noteworthy that the control using boiled (inactivated) secretome gave similar profiles (data not shown) and numbers (Fig. 5D) than the control with buffer alone, suggesting that there are not critical interactions between the substrate and the enzymes that could interfere during the processing and analysis of the samples, and thus the results.

To further understand depolymerization and repolymerization, changes in the DMR-EH lignin structure after the enzyme treatment were evaluated by 2D-NMR. Fig. 6 and 7 show the HSQC NMR spectra of the DMR-EH, DMR-EH control (after autoclaving), and the treated DMR-EH at 2, 3 and 4 days. Expanded HSQC spectra in the lignin side chain and carbohydrate region are shown in Fig. 6. The correlation peaks of C α and C β in the β -O-4 structure appear at δ_C/δ_H 71/5.05 and 82/4.3 ppm, respectively. Both peaks decreased in 2 and 3 days, but the C α peak intensity did not change in 4 days. This implies that β -O-4 structures were broken gradually by laccases releasing phenolic compounds, and then cleavage of



Fig. 5 GPC profiles of DMR-EH lignin as a function of time from *P. eryngii* secretome treatments at pH 7 over 4 days of incubation. Graphs show the DMR-EH lignin M_w profile from treatments with (A) 3 U laccase (LA) per g from non-dialyzed secretome, (B) 30 U LA per g from non-dialyzed secretome, (C) 30 U LA per g from dialyzed secretome, and (D) DMR-EH lignin M_w by using different laccase dosages from non-dialyzed secretomes. Green continuous arrows highlight the highest depolymerization extent at 3 days of treatment. Red discontinuous arrows indicate the increase in lignin M_w possibly *via* repolymerization. The asterisk (*) at 4 days corresponds to the control that contains boiled secretome.



Fig. 6 Expanded HSQC NMR spectra in lignin side chain and carbohydrate region (δ_C/δ_H 35–110/2.2–5.6 ppm) of solid residues before and after enzyme treatments for 2–4 days with non-dialyzed secretome. (A) DMR-EH control (autoclaved) without enzyme, (B, C, D) 2, 3, and 4 days of treatment with *P. eryngii* secretome containing 30 U laccase per g. The C β peak at δ_C/δ_H 82/4.3 ppm decreased, but does not disappear after 2, 3 and 4 days. The peak shows if the lower contour level in HSQC spectrum is selected. However, under the lower contour level, the noisy peaks also show and methoxyl peak overlaps with the sugar peaks. In order to keep important peaks clear, the HSQC NMR spectra with a high contour level was selected.



Fig. 7 Expanded HSQC NMR spectra in aromatic region (δ_C/δ_H 100–160/6.0–9.0 ppm) of solid residues before and after enzyme treatments for 2–4 days with non-dialyzed secretome. (A) DMR-EH control (autoclaved) without enzyme, (B, C, D) 2, 3, and 4 days of treatment with *P. eryngii* secretome containing 30 U laccase per g.

β-O-4 leveled-off after 3 days. The methoxyl peak at δ_C/δ_H 54–57/3.1–4.5 ppm also decreased, indicating that demethoxylation occurred, likely catalyzed by laccase.

C1 anomeric peaks in sugar at $\delta_{\rm C}/\delta_{\rm H}$ 93/5.05, 98/4.45 and 108/5.5 ppm disappeared in the control, suggesting that 4-O-Me-glucuronic acid units and 2-O-Ac- β -D-xylopyranose units were solubilized without the secretome, potentially during autoclaving. The peak intensities of C6 in cellulose at $\delta_{\rm C}/\delta_{\rm H}$ 60/3.4 ppm and xylan peaks at the region of $\delta_{\rm C}/\delta_{\rm H}$ 58–80/ 2.9–4.6 ppm slightly decreased for longer treatments. Anomeric peaks in cellulose and xylan at 102/4.4 ppm exhibit a similar trend to xylan peaks. These results show that carbohydrates in DMR-EH lignin were slightly depolymerized by the fungal secretome (as corroborated by the presence of some carbohydrateactive enzymes shown below *via* proteomics).

Expanded HSQC spectra in the aromatic region are shown in Fig. 7. The types of subunits (S, G, PCA, and FA) in Fig. 7 are the same as in Fig. 2B. All of the correlation peaks in the DMR-EH control are similar to those in DMR-EH lignin, demonstrating that the lignin aromatic structure was not significantly altered by autoclaving and the incubation of the substrate without the secretome. Peaks of S2/6 and G2/5/6 decreased in 3 days, but slightly increased at 4 days. The peak for PCA2/6 also decreased somewhat at 2 days and did not change at 3 and 4 days. The FA peak decreased considerably at 2, 3, and 4 days. These results suggest that the repolymerized portion at 4 days probably consisted of G and S units, and that PCA was slightly reduced although FA was almost fully absent. It is noteworthy that Cα-OH is often oxidized by laccase and subsequently the $C\alpha$ - $C\beta$ bond is cleaved to generate aldehyde, carboxylic acid, and quinone structures. However, there are no correlation peaks of aromatic-C2/6 in Ca-oxidized S-type lignin unit at $\delta_{\rm C}/\delta_{\rm H}$ 106.1/7.31. Also, no peaks of aldehyde and carboxylic acid were detected in the ¹³C-NMR spectrum. This result implies that generated intermediates of aldehydes and quinones repolymerized, which is in agreement with an increase in $M_{\rm w}$ in 4 days.

In conclusion, these data suggest how the laccase-rich *P. eryngii* secretome depolymerizes lignin, but also how a repolymerization effect takes place at the end of these treatments in the assay conditions. On the basis of these results, we selected 3 days and 30 U g^{-1} as optimum treatment length and laccase dosage, respectively, to obtain the highest depolymerization extent in further experiments.

Testing the "microbial sink" hypothesis: incubation of an aromatic catabolic bacterium and the *P. eryngii* secretome to enhance DMR-EH lignin depolymerization

The results in Fig. 5 suggest that LMW lignin is being repolymerized during the end of the enzyme treatments. As such, the next experiment was designed to combine an aromatic catabolic bacteria and fungal enzymes to potentially avoid repolymerization of monomeric species and/or increase the lignin depolymerization degree. We dub this hypothesis the "microbial sink", wherein a microbe catabolizes aromatic compounds while ligninolytic enzymes depolymerize lignin. To test this concept, it is essential to monitor changes in lignin (*e.g.*, GPC) and microbial growth (OD_{600}). When mixed physically together, bacteria often stick to biomass, thus potentially causing interference and leading to poor results for both GPC and OD_{600} measurements. As such, bacteria were cultivated in a dialysis membrane with DMR-EH lignin and fungal enzymes added external to the membrane. The dialysis membrane cutoff was <3.5 kDa, such that only small molecules and no proteins could pass through.

Control experiments were first conducted to test the resistance of the cellulose-based dialysis membranes to the fungal secretome as cellulases in either the secretome or secreted by *P. putida* could compromise the membrane integrity. In addition, several cellulase assays were performed in the secretome, and no activity was detected in the assay conditions. Simultaneously in this control experiment, we demonstrated that LMW compounds were able to pass through the membrane to serve as a carbon and energy source for *P. putida* by adding glucose outside the membrane as the sole carbon and energy source; in this experiment, the bacteria grew to an OD_{600} of 6 and the membrane conserved its integrity.

After demonstrating both the resistance and permeability of the membrane, we conducted experiments with (1) DMR-EH and boiled (inactivated) fungal secretome, (2) DMR-EH and active fungal secretome, (3) DMR-EH, inactivated fungal secretome, and membranes with bacteria, and (4) DMR-EH, active fungal secretome, and membranes with bacteria (Fig. 8A). The secretome used for these experiments was dialyzed to avoid interference with the fungal growth media. Bacterial growth was analyzed at the end point of the experiment (3 days). Bacterial ODs were higher in the samples containing the active secretome (OD = 5.2 ± 0.2) (Fig. 8B) than those ones with the inactivated one (OD = 4.3 ± 0.6). In addition, Fig. 8B also shows that the membranes appear clean after the cultivations, suggesting no major issues with lignin adsorption to the membranes. Fig. 8C shows the Mw of DMR-EH lignin in the insoluble and soluble fraction after each treatment. The M_w



Fig. 8 Incubation of DMR-EH lignin, the *P. eryngii* secretome, and *P. putida* KT2440 to demonstrate the microbial sink concept. (A) DMR-EH with fungal secretome and bacteria. (B) Bacterial growth after the treatment represented in (A). (C) DMR-EH lignin M_w obtained from the different treatments in the insoluble and soluble fractions. (D) GPC profiles of DMR-EH lignin from the insoluble fraction after the different treatments. (E) LC-MS/ MS from the soluble fraction at the end point of the different treatments. DMR = DMR-EH with inactivated fungal secretome; DMR + S = DMR-EH with fungal secretome; DMR + B = DMR-EH with bacteria and inactivated secretome; DMR + S + B = DMR-EH with bacteria and fungal secretome; 4-HBA: 4-hydroxybenzoic acid; 4-HB: 4-hydroxybenzaldehyde.

decreased significantly after adding the fungal secretome, 38% in the solid and 78% in the liquid fraction. However, when adding the membrane system with both P. putida KT2440 and the secretome (DMR + B + S), the decrease was more pronounced for the solid fraction at 55%, but not for the liquid fraction (58%). These results demonstrate that if bacteria are present, the lignin depolymerization extent increases; however, in the liquid fraction, the lower apparent extent of M_w decrease is likely a result of the catabolism of LMW species (the microbial sink) thus leading to an increase in the apparent M_w of that fraction due to bacterial removal of catabolized species. A minor reduction of the $M_{\rm w}$ of insoluble lignin was also observed when having only bacteria (~12%) (Fig. 8C); however, the GPC profile shows (Fig. 8D) how the LMW lignin is not released in that case, as it does in the presence of ligninolytic enzymes. Table S3[†] shows different GPC parameters from these results. It is also worth noting here that the solubilization degree of DMR-EH was similar after all the treatments $(\sim 40\%)$ as well as the lignin content in the insoluble fraction $(\sim 41\%)$ (data not shown). These results show that although substrate solubilization (into water) is not enhanced after the enzyme and/or bacterial treatment, depolymerization of HMW lignin is prevalent (Fig. 8C).

To determine if bacteria are consuming aromatic compounds, LC-MS/MS was performed in the soluble fraction from all the treatments, and seven aromatic species were abundant (Fig. 8E). It is evident that some monomeric aromatics, such as PCA, FA, 4-hydroxybenzoate, and syringate, decrease and/or disappear in the presence of the secretome alone and in the presence of bacteria alone. This observation demonstrates how the enzymes in the secretome can act on these monomeric compounds, but also how the bacterium consumes the aromatic compounds, respectively. In addition, vanillin and vanillate appear primarily in the treatments where bacteria are present and not in the control (DMR-EH). These aromatic compounds are intermediates in the catabolism of FA with vanillate being a typical bottleneck in this pathway in P. putida $KT2440.^{27}$ The presence of vanillin in the DMR + S + B soluble fraction demonstrates how catabolism may be more rapid than the secretome enzymatic action on the LMW aromatic compounds present in the soluble fraction. Lastly, 4-hydroxybenzaldehyde (4-HB), which is present in the control, is also metabolized by P. putida KT2440; however, it is not a substrate for ligninolytic enzymes (Fig. 8E). Overall, the presence of a microbial sink appears to enhance the overall extent of lignin depolymerization.

Proteomic analysis of P. eryngii secretome

The current work demonstrates how the *P. eryngii* secretome is able to depolymerize lignin from the solid DMR-EH stream. Among the enzymes analyzed, there may be isoenzymes that exhibit similar activities or enzymes not detected at the assay conditions. Consequently, to more deeply characterize the enzyme cocktail utilized in the current study, a proteomic analysis was performed. Over 400 proteins in the *P. eryngii* secretome were classified after BLAST searches³⁹ against a

combination of Uniprot and NCBI (fungal kingdom) to obtain the best hit (defined as the match with the lowest *E*-value and highest percent identity) and non-hypothetical match. The first non-hypothetical match was the one considered to classify the proteins (Fig. 9). To determine the relative abundance of the proteins, the sum of the peptide spectrum matches (PSMs) belonging to each protein were utilized. Only those proteins with \geq 2 peptides detected were considered in the analysis.

Proteins were first organized by function and abundance (Fig. 9A). The oxidoreductase group was the most abundant (38%) despite containing a fewer number of proteins than the Glycoside Hydrolases (GHs) (42 and 82 proteins, respectively). GHs (23%) are the second most abundant group followed by proteinases (11%) and esterases (5%) (Table S4[†]). As oxidoreductases are the main proteins involved in lignin depolymerization, a deeper analysis of this group is also presented (Fig. 9B, Table S4[†]). Glucose/methanol/choline (GMC) oxidoreductases (41%), AAOs (14%) (which are also GMC-oxidoreductases), and laccases (23%) were the most abundant oxidoreductases, comprising 78% of the total. It is also noteworthy that peroxidases (e.g., DyP-type peroxidases, 4%) were quite limited in number in the 9-day fungal secretome, which matches with the low measured peroxidase activities (Fig. 3D). The presence of a signal peptide cleavage site, which predicts if a protein could be secreted, was also analyzed in these 42 oxidoreductases. Only 8 proteins did not contain a signal peptide, namely one DyP, the two reductases, the three FMNlinked oxidoreductases, one glyoxal oxidase, and one AAO



Fig. 9 Proteomic analysis of the secretome of *P. eryngii* utilized for DMR-EH lignin depolymerization assays. (A) Functional classification of the proteins detected in the fungal secretome. (B) Classification of the oxidoreductases in the secretome. The abundance of each group, calculated by the sum of the PSMs belonging to each protein, is presented as a percentage. The numbers in parentheses at the right side of each protein group indicate the number of different proteins observed.

(Table S4[†]). Although proteins with one unique peptide were not considered for the analysis, it is noted that a single peptide was also found for an unidentified heme-peroxidase and a short MnP (data not shown).

Regarding the composition of the second major functional group, GHs, proteins such as cerato-platanins, α -L-arabinofuranosidases, glucoamylases, GH family 79 (which contains proteins such as β -glucuronidases), and 1,3- β -glucosyltransferases (Table S4†) were the most abundant. It is worth mentioning that in this group, three hits for the GH Family 61 were also found. Although they are included in GH category, according to their current characterized function,⁴⁰ GH61s are lytic polysaccharide monooxygenases. Overall, the oxidoreductases are the most prevalent group of enzymes in this organism which aligns well with its ability to selectively depolymerize lignin in biomass.⁴¹

Discussion

Selective and efficient lignin depolymerization remains one of the main challenges in biomass conversion. Currently, lignin upgrading to value-added compounds, such as polyhydroxyalkanoates, fatty acids, or muconic acid, has been described, but primarily from LMW lignin species.^{7,27,28,42} Although advances have been made in the lignin depolymerization field by bacteria,^{4,7,43} carbon conversion remains low. Fungal ligninolytic enzymes, on the other hand, are effective at depolymerizing lignin and could be a powerful complement to enhance bacterial lignin conversion. However, these types of enzymes are also involved in lignin polymerization,^{12,16} which could limit monomer yields. Thus, from another perspective, bacteria could be also considered a natural "microbial sink" to minimize action of ligninolytic enzymes on LMW compounds.

In the current study, a biorefinery stream, DMR-EH lignin, was utilized to demonstrate how fungal ligninolytic cocktails can depolymerize lignin and to test how bacteria can enhance the resulting depolymerization. Typically, most in vitro studies utilizing oxidoreductases to depolymerize lignin employ commercially available single enzymes, not secretomes. The content of a "ligninolytic cocktail" is highly dependent on the fungus, as demonstrated in Fig. 3, but also on the culture conditions.^{9,44} Here, a standard medium for white-rot fungal cultivation was used, but supplemented with autoclaved DMR-EH since lignocellulosic substrates induce production of ligninolytic9 and cellulolytic enzymes.10 The fungi P. eryngii and Bjerkandera sp. were chosen from an initial screening with the highest laccase and peroxidase activity, respectively. Moreover, a specific harvesting time was selected that exhibited primarily one enzyme activity or the other (especially in P. eryngii). The secretion of these enzymes by these fungi has been reported previously^{45,46} and trends, in particular for laccase production, are similar to those found earlier.45 However, the secretomes from these fungi have never been utilized for lignin depolymerization. In fact, the few studies reporting the use of fungal secretomes are more related with the production of mechanical pulps and the improvement of their properties.⁴⁷ In addition, in the pulping field, delignification is measured by the decrease of the kappa number;⁴⁸ however, with this type of analysis, it is difficult to know if lignin depolymerization is occurring or if it is just a grafting effect mediated by laccases that modifies lignin and enhances solubility.¹²

Regarding substrate conditioning, DMR-EH was subjected to different processing as required prior to each enzyme treatment (i.e. with and without autoclaving, different pH, and different buffers). These variations changed the $M_{\rm w}$ of lignin. For instance, a M_w of 6100 Da (Fig. 4B), 5500 Da (Fig. 5D), and 8000 Da (Fig. 8C), all at pH 7, were observed for the insoluble (non-enzyme treated) fraction of non-autoclaved, autoclaved with malonate buffer, and autoclaved with phosphate buffer samples, respectively. Moreover, when pH 4.5 in non-autoclaved DMR-EH was used, a M_w of 12 000 Da (Fig. 4A, C and D) was measured. These results show that lignin solubilization is highly dependent on pH, with the M_w of DMR-EH lignin at pH 7 being ~50% lower than at pH 4.5. Conversely, during autoclaving, lignin can also likely undergo additional polymerization, increasing the initial M_w. As such, the pH and DMR-EH conditioning are critical for the current experiments.

Fungal ligninolytic enzymes generally have low optimal pH (<5), in particular peroxidases; on the other hand, the aromatic-catabolizing bacteria (P. putida) used in the current study grows optimally at pH 7. As such, we performed depolymerization assays at pH 4.5 and 7. Although the degree of depolymerization was maximum when using mixed P. eryngii and Bjerkandera secretome at the highest enzyme dosage at pH 4.5 (around 84%) (Fig. 4D), the final lignin M_w was similar to that obtained in treatments with the P. eryngii secretome (30 U laccase per g) at pH 7 (Fig. 3B) at 2300 Da. This M_w roughly corresponds to oligomers of 10 units of aromatics, a considerable reduction of the $M_{\rm w}$ of the solid fraction when compared with the initial DMR-EH substrate $M_{\rm w}$ (10 000 Da) (Fig. 2A). Around 50% Mw reduction was reached just with the pH adjustment, which increased to 75% upon addition of the secretome. Considering these results, the laccase-rich secretome was selected over the peroxidase-rich enzyme cocktail for reasons mentioned above. Nevertheless, laccases are not typically able to oxidize non-phenolic compounds if mediators are not present. However, aromatic compounds such as sinapic acid, PCA, and FA are natural mediators that increase the ability of laccases to oxidize more recalcitrant compounds.49,50 As presented in Fig. 8E, PCA and FA are the main aromatic compounds detected in the supernatant of DMR-EH lignin, and thus, laccases could use these as mediators to enhance their substrate oxidation abilities.

Compiling results from the published work using ligninolytic enzymes on lignin-rich substrates, it is apparent that depolymerization is highly variable, making direct comparisons difficult, considering the diversity of substrates and the different lignin analysis methodologies employed. For pure peroxidase treatments, for example, a horseradish peroxidase (HRP),⁵¹ fungal LiPs and MnPs from *P. chrysosporium*,⁵² and VP from *P. eryngii*⁵³ have been tested and demonstrated some

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degree of lignin depolymerization on different substrates such as synthetic lignin, insoluble isolated poplar lignin, and lignosulfonates, respectively. Recently, a laccase from Myceliophthora thermophila and Trametes villosa were utilized to delignify eucalyptus.^{13,22,23} In this case, $C\alpha$ -C β cleavage was demonstrated, but the molecular weight changes were not reported. Instead, in this case, a reduction in the Klason lignin content was evident. Nevertheless, different to our work, those enzyme treatments were complemented with peroxide, which also could enhance lignin degradation. Laccase dosage has been also presented as another important parameter to improve the depolymerization degree (Fig. 4 and 5). Chen et al.⁵⁴ utilized laccase in synergy with cellulases demonstrating lignin decomposition and increasing sugar yields at a dosage of 4000 U g⁻¹ of material. The highest dosage utilized in the current study is 100 U g^{-1} at pH 4.5, thus complicating direct comparisons.

Many studies utilizing pure laccases to oxidize substrates also supplement mediators (natural or synthetic) to enhance lignin depolymerization.55 However, it is not still clear if these mediators can also increase lignin polymerization.⁵⁶ Both depolymerization and repolymerization processes are natural effects produced via free radical generation by oxidoreductases.¹² In the time course experiment shown in Fig. 5, repolymerization was observed from 3 to 4 days at all the enzyme dosages and using dialyzed and non-dialyzed secretomes. Hofrichter et al. reported that a MnP from Phebia radiata polymerized and depolymerized according to the initial nature of the substrate, LMW and HMW lignin, respectively.⁵⁷ As this work is performed with an entire secretome and not a single enzyme, it is challenging to pinpoint the main factor responsible for this effect. The nature of the substrate is likely to partially determine these responses, although the literature is not clear. For instance, HRP and laccase treatments in milled wood lignin solubilized in NaOH mainly resulted in depolymerization;⁵⁸ however, some efforts in industrial lignin (from a paper factory and also solubilized in NaOH) with semipurified laccase from Fusarium proliferatum exhibited considerable polymerization, but when ABTS was used as a mediator, depolymerization was also observed.59

The search for a "radical sequestration" agent could be an alternative to decrease the degree of repolymerization. Dordick et al.⁵¹ reported that peroxidases depolymerize lignin only in organic media, but not water, highlighting that solvents might quench free radicals that lead to repolymerization. Also, the presence of enzymes such as AAO (synonym: veratryl alcohol oxidase) from P. ostreatus have been described to prevent polymerization in substrates oxidized by laccases by reducing aromatic radicals.^{9,60} These type of oxidases are also involved in providing H₂O₂ to peroxidases. In the secretome utilized here, AAOs are present (Fig. 3D and 9B), which could have a similar effect and, indirectly, improve depolymerization. However, the main radical sequestration in the current study is likely to be the microbial sink. Results presented in Fig. 8 demonstrate how the treatment with both the P. eryngii secretome and bacteria increase the depolymerization degree

compared to those treatments alone. The GPC chromatogram of the solid fraction (which shows a M_w reduction greater than 50% compared to control), LC-MS/MS of soluble fraction (showing the presence of bacterial aromatic intermediates that are not present in the control), and bacterial growth into the membranes (higher when the secretome is also present), all support the idea of the bacteria taking up aromatic compounds and improving depolymerization of DMR-EH lignin. A similar study was performed utilizing the secretome from the fungus P. chrysosporium (with high peroxidase content instead of laccases) and the bacterium Xanthomonas sp. by using different $M_{\rm w}$ lignins.⁶¹ However, in this case, after adding the secretome, lignin mineralization by the bacteria was lower likely due to an increase in the $M_{\rm w}$ of lignin and thus, being less available for the bacterium. Another recent study also utilized a pure laccase cocktail from T. versicolor and the bacterium Rhodococcus opacus during Kraft lignin degradation; however, lignin Mw (analyzed by GPC), also increased in all cases.42

As highlighted above, lignin depolymerization by P. eryngii secretome cannot be only attributed to the presence of laccases but the synergy of all the proteins secreted. Thus, to obtain a deeper characterization of the enzyme cocktail utilized, a proteomics study of the fungal secretome was also conducted. As expected, due to the selective nature of this fungus for depolymerizing lignin, the major functional group was the oxidoreductases (Fig. 9A). Apart from laccases, GMC oxidoreductases were also a major enzyme family present in the secretome. Considering previous enzymatic studies on P. eryngii, the activity of the GMC enzymes in this organism are hypothesized to possess an AAO activity.62 As expected, GHs were also detected in the secretome. Considering that typical biorefinery streams would contain polysaccharides and lignin, the use of complete fungal secretomes could have the advantage of enhancing substrate degradation and thus improving final carbon availability and conversion. A proteomic study has been recently performed in the fungal counterpart P. ostreatus.9 In that study, 508 proteins were detected after 21 d cultivation. Here, we are presenting a fungal cocktail of over 400 proteins detected after 9 days of incubation in the presence of DMR-EH. The number of proteins reported in these studies highlights how complex lignocellulose degradation is in Nature and how potentially difficult relying on a single enzyme will be for efficient lignin depolymerization.

Conclusion

As demonstrated in this work, fungal secretomes containing high laccase and peroxidase activity are able to depolymerize a solid lignin-rich substrate with native lignin like properties. In addition, a new concept dubbed the "microbial sink" is presented which is based – as it could occur in Nature – on complementing fungal ligninolytic enzymes and an aromaticcatabolizing bacterium. This "natural synergy" might act as a sequestration mechanism to avoid the repolymerization of

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LMW lignin *via* enzyme redox reactions, thus improving lignin depolymerization. Despite partial depolymerization of solid lignin using a solely biological approach, higher lignin conversions will be required for realistic lignin valorization processes. In future work, we will report chemo-catalytic solubilization of the DMR-EH residual lignin, which may enable higher extents of microbial lignin conversion.

Materials and methods

Substrate preparation

Corn stover was deacetylated and mechanically disc refined at 1 ton per day pilot plant scale according to Chen et al.⁶³ Enzymatic digestions were carried out using NREL's 1900-L DI paddle mixer. The 1900-L DI mixer was cleaned and sterilized at ~120 °C-130 °C for 1 h prior to adding 268 kg (110 kg dry) of DMR corn stover and the ~30 wt% slurry was pasteurized at 80 °C for 2 h. The pH of the slurry was adjusted to 5.0. The remainder of the water was added and the pH adjusted to 5.0 at 50 °C prior to adding the cellulase and hemicellulase enzyme preparations to give a final total solids loading of ~20 wt%. The enzyme loading was 50 mg Novozymes CTec3 per g of glucan and 10 mg HTec3 per g xylan in the DMR residues. The temperature was maintained at 48-50 °C for 3 days. The liquor was drained and washed to remove residual components. The solids were further enriched in lignin by repeating the enzymatic hydrolysis to remove more of the glucan and xylan in similar conditions along with adding 1 μ g L⁻¹ of Lactrol (Phibro) to inhibit microbial contamination. The solids were washed three times with water and centrifuged at 8000g for 30 min. The supernatants were then decanted and the remaining solids were lyophilized for storage and further characterization.

Substrate characterization

Full compositional analysis. Compositional analysis was performed according to the procedure in NREL LAP/TP-510-42618⁶⁴ to determine lignin, sugar, and other organic compound concentrations in the substrate.

Gel permeation chromatography (GPC) analysis. DMR-EH residue (20 mg) was acetylated in a mixture of pyridine (0.5 mL) and acetic anhydride (0.5 mL) at 40 °C for 24 h with stirring. The reaction was terminated by addition of methanol (0.2 mL). The acetylation solvents were then evaporated from the samples at 40 °C under a stream of nitrogen gas. Samples were further dried in a vacuum oven at 40 °C overnight. The dried, acetylated samples were suspended in tetrahydrofuran (THF, Baker HPLC grade), and then filtered through 0.45 µm nylon membrane syringe filters. Most of lignin in the samples was soluble in THF as measured by ¹³C solid-state NMR of the insoluble material, but polysaccharides were not fully acetylated and not dissolved in THF. GPC analysis was performed using an Agilent HPLC with 3 GPC columns (Polymer Laboratories, 300×7.5 mm) packed with polystyrene-divinyl benzene copolymer gel (10 µm beads) having nominal pore

diameters of 10^4 , 10^3 , and 50 Å. The eluent was THF and the flow rate 1.0 mL min⁻¹. An injection volume of 25 μ L was used. The GPC was attached to a diode array detector measuring absorbance at 260 nm (band width 80 nm). Retention time was converted into molecular weight (MW) by applying a calibration curve established using polystyrene standards. The highest response value in the *y*-axis in each GPC chromatogram was normalized to 100.

NMR. HSQC NMR spectra were acquired for DMR-EH. DMR-EH (0.2 g) was ball milled by a planetary ball milled using a Retsch PM100 mill fitted with a 50 mL ZrO₂ grinding jar and 10 × 10 mm ball bearings, set at 600 rpm. Ball milled DMR-EH (80 mg) was suspended into d_6 -DMSO/ d_5 -pyridine (4:1, v/v) in a NMR tube. The mixture was sonicated for 5 h until the gel became homogeneous.³²

Spectra were acquired at 40 °C on a Bruker Avance III 600 MHz spectrometer at 11.7 T using a TCI cryoprobe. Spectra were acquired with 1024 points and a sweep width of 15 ppm in the F2 (¹H) dimension and 512 points and SW = 220 ppm in the F1 (¹³C) dimension. Peak assignment was performed according to literature.

Analysis of aromatics by LC-MS/MS. Mass spectrometry was used in the last experiment of the current study to analyze aromatics from the soluble fraction. For this purpose, 14.5 mg of freeze-dried supernatant from 8 different treatments was reconstituted in 1 mL methanol. Analysis of samples was performed on an Agilent 1100 LC system equipped with a diode array detector (DAD) and an Ion Trap SL MS (Agilent Technologies, Palo Alto, CA) with in-line electrospray ionization (ESI) as detailed in the ESI.† Vanillin, 4-hydroxybenzaldehyde, PCA, vanillic acid, syringic acid, 4-hydroxybenzoic acid, and FA were the only aromatics detected from an instrument specific in-house database of 70 aromatic compounds. Standards were prepared in the range of 1–100 μ g mL⁻¹ and run at the same conditions as the samples.

Fungal strains and cultivation conditions

The strains of basidiomycetes used in the present study were obtained from Centraalbureau voor Schimmelcultures (CBS; Baarn, The Netherlands) and Instituto Jaime Ferrán de Microbiología (IJFM; Centro Investigaciones Biológicas, Madrid, Spain). The fungal species consisted of Bjerkandera sp. IJFM A660, Bjerkandera adusta CBS 230.93, Cerioporopsis subvermispora IJFM A718, Irpex lacteus IJFM A792, Panus tigrinus IJFM A768, Phanerochaete chrysosporium CBS 481.73, Phellinus robustus IJFM A788, Pleurotus eryngii IJFM A582, Pleurotus ostreatus CBS 411.71, Polyporus alveolaris IJFM A794, Stereum hirsutum IJFM A793, and Trametes versicolor IJFM A136. Strains were maintained on 2% malt extract agar (w/v) and preserved at 4 °C. With the aim of reviving the fungi before starting the experiments, fungi were individually cultured at 28 °C for 7 days on potato dextrose agar (39 g L⁻¹) plates. After this incubation time, four agar plugs of about 1 cm² were cut from actively growing mycelium and inoculated into 250 mL Erlenmeyer baffled flasks with 50 mL of modified Czapek-Dox media (final pH 5.9), which contained 10 g L^{-1}

glucose, 2 g L⁻¹ ammonium tartrate, 1 g L⁻¹ KH₂PO₄, 0.5 g L⁻¹ MgSO₄ × 7·H₂O, 0.5 g L⁻¹ KCl, 1 g L⁻¹ yeast extract, 5 g L⁻¹ peptone, 0.1 g L⁻¹ B₄O₇Na₂ × 10·H₂O and 1 mL L⁻¹ of trace elements which contained (×1000) 16 mg L⁻¹ MnSO₄, 27 mg L⁻¹ Fe₂SO₄, 74 mg L⁻¹ CuSO₄ × 5·H₂O, and 73 mg L⁻¹ ZnSO₄ × 7·H₂O. After 6 days of incubation at 28 °C and 180 rpm, each culture was aseptically blended (Bead Beater) for 30 seconds to homogenize mycelia size, and 2.5 mL (per 50 mL of media) were used to inoculate those flasks for the enzyme screening and secretome production.

Enzyme screening in fungal secretomes

To determine the best fungal species producing ligninolytic enzymes, an initial screening was performed in the secretome of twelve basidiomycetes (detailed above) while growing in the presence of DMR-EH lignin. For this purpose, triplicates of 250 mL flasks containing 50 mL of modified Czapek-Dox growth media and 1% DMR-EH were utilized to grow each strain. Fungi were incubated at 28 °C and 180 rpm during 15 days. Samples (1 mL) were taken every two days since the third day during the whole incubation time. Samples were then centrifuged during 10 minutes at 14 000 rpm and the supernatant transferred to another tube for further enzyme assays.

Enzyme assays

Different UV/colorimetric enzyme assays were performed in the fungal supernatants to evaluate activities such as laccase, AAO, MnP, DyP, LiP, and other peroxidases such as VP. Laccase activity was determined following ABTS oxidation to its cation radical (ε_{418} = 36 000 cm⁻¹ M⁻¹) in 100 mM tartrate buffer at pH 5. AAO activity was assayed by veratraldehyde formation from 5 mM veratryl alcohol in 100 mM sodium tartrate buffer at pH 5 (ε_{310} = 9300 cm⁻¹ M⁻¹). MnP and VP activity can be measured by the formation of Mn^{3+} -tartrate complex (ε_{238} = 6500 cm⁻¹ M⁻¹) during the oxidation of 0.1 mM MnSO₄ in 100 mM tartrate buffer at pH 5. LiP activity was assayed by veratraldehyde formation from 5 mM veratryl alcohol in 100 mM sodium tartrate buffer at pH 3 (ε_{310} = 9300 cm⁻¹ M⁻¹). Other peroxidase activities such as DyP and short-MnP could be also followed by the oxidation of ABTS in the absence and the presence of 0.1 mM MnSO4 at pH5 respectively. VP can be detected using ABTS as well with and without Mn²⁺. In all cases, peroxidase activity assays were conducted in the presence of 0.1 mM H₂O₂ excluding LiP, which was activated by 0.4 mM H₂O₂. Peroxidase activities evaluated through ABTS were all corrected by subtracting laccase activity. The discoloration of reactive blue 19 (anthraquinone dye) at 50 μ M (ε_{595} = 10 000 cm⁻¹ M⁻¹) and reactive black 5 at 25 μ M (ϵ_{598} = 30 000 cm⁻¹ M⁻¹) in sodium tartrate buffer at pH 4 were also used to ascertain the type of peroxidase, being VP able to discolor reactive black 5 and DyP both of them in the presence of 0.1 mM H_2O_2 . One enzymatic activity unit was defined as the amount of enzyme that oxidizes 1 µmol of substrate in 1 min.

After the enzyme screening, both laccases from *P. eryngii* and peroxidases from *Bjerkandera* sp. were tracked by ABTS oxidation at pH 4.5 and 7 in 100 mM sodium malonate.

Peroxidases were tested in the absence and the presence of 0.1 mM MnSO_4 and activated by $0.1 \text{ mM H}_2\text{O}_2$. The buffer was selected based on the buffering capacity at that pH range and its manganese chelating properties. Laccases from *P. eryngii* were evaluated in the last experiment in phosphate buffer at pH 7 instead of sodium malonate since that was the main buffer during the depolymerization assay. Laccase activity in both buffers was similar.

The secretome of *P. eryngii* was also evaluated for cellulase activity by measuring the activity on amorphous and crystalline cellulose.⁶⁵ 500 μ g of secretome proteins were mixed in a 1% slurry of cellulose substrate and incubated at 28 °C for 5 days in a rotisserie incubator. The mixture was then run on an HPLC and no measurable levels of cellobiose, glucose, or xylose were detected.

Secretome production and processing

To produce larger secretome volumes from the selected fungi, 1 L flasks containing 200 mL of modified Czapek-Dox media and 1% DMR-EH were selected to grow the fungi. Cultures were incubated at 28 °C and 180 rpm, enzymes evaluated every day, and secretomes collected after 9 days. To process the secretome, cultures were filtered through Miracloth to remove the fungi, then through 1.1 μ M pore size glass fiber filters, and lastly through 0.2 μ M pore size PES filters (Thermo ScientificTM NalgeneTM Rapid-FlowTM) in sterile conditions.

For the last experiments with the *P. eryngii* secretome, part of the secretome was dialyzed in a 350 mL pressurized ultracentrifugation system (Amicon) with 10 kDa membranes. In detail, at least 1 L of secretome broth was concentrated down to approximately 40 mL and then washed several times with saline buffer until the dialysis factor was higher than 100. The dialysis buffers depended on the subsequent experiment. For the depolymerization assays over time with the *P. eryngii* secretome, 20 mM sodium malonate at pH 7 was used. For the experiment where bacteria were present, 20 mM phosphate at pH 7 (Na₂HPO₄ and KH₂PO₄ at a ratio of 1:2.2) was utilized. The volume of the dialyzed secretomes was adjusted to achieve the same laccase activity values than in the original, nondialyzed secretome.

Enzyme depolymerization assays of DMR-EH

DMR-EH treatment with the secretome of *P. eryngii* and *Bjerkandera* sp. at pH 4.5 and pH 7. Before starting the depolymerization assays, laccase and peroxidase activity was measured at pH 4.5 and 7 using ABTS as substrate, and peroxidase activity calculated in the presence of Mn^{2+} and H_2O_2 as previously described. Laccases from *P. eryngii* were active at both pHs but the peroxidases from *Bjerkandera* only at pH 4.5. Thus, treatments with the *P. eryngii* secretome were carried out at both pHs, but treatments containing peroxidases from the *Bjerkandera* secretome only at 4.5. The pH was tracked during the incubation time, and no changes were observed. Regarding enzyme stability, laccase and peroxidase activity remained at the end of the treatments.

Different volumes from both fungal secretomes were utilized to achieve different laccase and peroxidase dosages at the different pHs. The enzyme added in each treatment was expressed as total U per gram of DMR-EH. We evaluated laccase at dosages of 10, 50, and 100 U g^{-1} at pH 4.5 and 3, 15, 30 U g^{-1} at pH 7 and peroxidase from *Bjerkandera* sp. at dosages of 8, 40, 80 U g⁻¹ at pH 4.5. Secretomes were added to 250 mL flasks containing 0.5 g DMR-EH, 10 mL sodium malonate (250 mM) at pH 4.5 or 7, and 150 µL of tetracycline $(250 \ \mu g \ mL^{-1})$ to avoid bacterial contamination. Sterile distilled water was also added to achieve a total volume of 50 mL. When the secretomes of both fungi were complemented, half volume of each secretome was utilized. In addition, in those treatments using Bjerkandera's secretome, 0.1 mM MnSO4 and 0.5 mM H₂O₂ were also included, the latter added at 0, 4, 8, 12, 22, 28, 34, 48, 54, 60, 72, 78, 84, 96, 104, 120, 128 h. The same H₂O₂ concentration was selected across each enzyme loading since the exact amount of peroxidase (mg mL⁻¹) in the sample and the abiotic effect of H₂O₂ on the lignin were unknown. Control treatments without enzyme addition were carried out in parallel at both pHs and in the presence and the absence of MnSO4 and H2O2 and analyzed after 6 days of incubation. Treatments were performed at 30 °C, 150 rpm during 3 and 6 days, although at 3 days only those treatments utilizing the highest enzymes dosages were sampled. After incubation, the flask content was centrifuged at 5000 rpm for 20 min. The supernatant was collected and the solid fraction weighted after drying at 60 °C for 48 h. This solid fraction was then utilized for GPC analysis.

DMR-EH treatment with the secretome of P. eryngii at pH 7 over time. Considering the results obtained in the previous experiment, the treatment with the P. eryngii secretome at pH 7 was the selected condition to obtain DMR-EH lignin with the lowest molecular weight. For a time-dependent characterization of depolymerization, three different enzyme dosages were utilized (3, 10, 30 U laccase per g DMR-EH) and GPC analysis were performed from samples collected over time. The secretome utilized for this experiment was both dialyzed and non-dialyzed. Secretomes were added to 250 mL flasks containing 1 g of autoclaved DMR-EH (120 °C, 20 min) in the presence of 10 mL sodium malonate (250 mM) at pH 7. Sterile distilled water was also added to achieve a total volume of 50 mL. Control treatments, without enzymes and with inactivated boiled secretome (utilizing the same volume of that treatment corresponding to the highest laccase dosage) were also carried out and analyzed at 1 and 4 days of incubation. Treatments were incubated at 30 °C and 150 rpm and samples collected after 1, 2, 3, and 4 days of treatment. At these time points, the flask contents were centrifuged at 5000 rpm for 20 min. The supernatant was collected and the solid fraction weighted after drying at 60 °C for 48 h. This solid fraction was then utilized for GPC analysis. All treatments were performed in duplicates.

DMR-EH treatment with the secretome of *P. eryngii* at pH7 in the presence and the absence of *P. putida* KT2440. The aim of this experiment was to evaluate if bacteria enhance lignin depolymerization in enzyme treatments with the P. eryngii secretome. For these experiments, the secretome was dialyzed in phosphate buffer (pH 7) as detailed previously and half of the volume boiled (20 min) to inactivate the enzymes. The treatments performed in this last experiment were (1) a control with DMR-EH and boiled secretome, (2) DMR-EH plus secretome (30 U laccase per g of DMR-EH), (3) DMR-EH plus boiled secretome and bacteria, and (4) DMR-EH plus secretome (30 U laccase per g DMR-EH) and bacteria. For these experiments, 1 g of DMR-EH was autoclaved (120 °C, 20 min) in the presence of 10 mL phosphate buffer (20 mM) at pH 7 in a 0.5 L beaker. Then, 10 mL of modified M9 media (×10), which is the growth media for the bacteria (pH 7), was also added and contained per L: 6.78 g Na₂HPO₄, 3 g KH₂PO₄, 0.5 g NaCl, and 15 mM (NH₄)₂SO₄. Then, 2 mL of 1 M MgSO₄, 100 µL of 1 M CaCl₂, and 1 mL of ion traces (CuSO₄, MnSO₄, FeSO₄, and $ZnSO_4$, 100 μ M each) per L of treatment were added. The same phosphate buffer (20 mM) was finally added to achieve a total volume of 50 mL in all the treatments.

To grow and separate the bacteria, P. putida KT2440 (ATCC 47054), from the DMR-EH/enzyme mixture, a membrane system was employed. Dialysis membranes (cassettes) of 3.5 kDa cutoff (Thermo Scientific) were UV-sterilized for 20 minutes and inoculated with 2.5 mL of P. putida culture at an initial optical density (OD) of 0.1 at 600 nm. To prepare the bacterial inoculum, P. putida KT2440 was grown in LB at 30 °C and 220 rpm overnight. Cells were then centrifuged, washed, and diluted in modified M9 (×1) to reach the desired initial OD. Prior the experiments on lignin, the resistance of the cellulose-based dialysis membranes to the fungal secretome as well as the permeability of the membranes to LMW compounds was tested. For this purpose, inoculated membranes were incubated 3 days at 30 °C and 180 rpm in the presence of the secretome and only glucose (5 g L^{-1}) as carbon source in M9. For the experiments on lignin, inoculated membranes were then included in the corresponding beakers and treatments also incubated at 30 °C and 180 rpm during 3 days. Glucose was not added in these experiments. After the incubation, membranes were taken out and cells extracted with a syringe to quantify the final OD in a volume of 2.5 mL. The beaker contents were centrifuged at 5000 rpm for 20 minutes. Solids were dried for further weight, GPC, and compositional analysis. The supernatant was freeze-dried for GPC and LC-MS/MS analysis. All treatments were performed in duplicates.

Proteomic analysis in the secretome of P. eryngii

For proteomic analyses, 50 mL of the *P. eryngii* secretome from 9 days of incubation (used for the depolymerization assays) were taken from 2 biological replicates and flash-frozen. Then, samples were treated and analyzed as described in the ESI.† In brief, the volumes of the secretomes from *P. eryngii* were initially reduced prior to isolating the proteins with a chloro-form-methanol extraction.⁶⁶ Then, the proteins were tryptically digested following a previously published protocol.⁶⁷ After this step, protein concentrations were normalized and

analyzed by LC-MS using a Velos Pro Orbitrap mass spectrometer (Thermo Scientific, San Jose, CA) outfitted with an electrospray ionization (ESI) interface coupled to custom built constant flow high-performance liquid chromatography (HPLC) system.^{68,69} The MS/MS spectra from 15 LC-MS/MS datasets were converted to ASCII text (.dta format) using DeconMSn⁷⁰ and the data files were then interrogated *via* target-decoy approach⁷¹ using MSGFPlus.⁷² Protein sequences for each reported entry were subjected to BLAST analysis (version 2.2.28) using a combined collection of 2784 909 fungal proteins reported in the Uniprot knowledgebase and NCBI. SignalP 4.0⁷³ was finally utilized to predict the presence of signal peptide cleavage sites in the detected proteins.

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